

Water Working Group



RCSP Water Working Group Annual Meeting

August 18, 2015

Ryan J. Klapperich,¹ David V. Nakles,² and Andrea T. McNemar³

¹Energy & Environmental Research Center ²The CETER Group, Inc. ³National Energy Technology Laboratory





- Water Working Group (WWG):
 - Composed of experts from government, academia, and industry who are members of the U.S. Department of Energy (DOE) National Energy Technology Laboratory (NETL) RCSP Program.
- Mission statement:

The Plains CO₂ Reduction Partnership

 Address stakeholder concerns regarding emerging carbon capture and storage (CCS) technology and potential interactions with local and regional water resources.

















The Water and CCS Nexus



Research and Development of the search and Development of the search and Development of the search and the sear

Capture



Compression



Transportation



- Additional resources
 - Water: cooling, process water
 - Power
 - Replace power diverted to capture and compression















Water and Carbon Capture, Compression, and Transportation



- Chemical or physical absorption
 - 15%–30% water increase
 - Additional cooling
 - Solvent/sorbent regeneration
 - Process-specific subprocesses



















World Water and Carbon Capture Centers of the second conters of the second content con



Source: Gerdes, K., and Nichols, C., 2009, Water requirements for existing and emerging thermoelectric plant technologies: DOE/NETL Report 402/080108; U.S. Department of Energy National Energy Technology Laboratory: Morgantown, West Virginia.













Optimizing Carbon Capture Water Use

- Efficient process design
 - Update existing facilities
 - New technologies (e.g., compression)
- Alternative cooling technologies
- Alternative (degraded) water sources



















WORLD-Water and Carbon Storage Centers of Excellent Carbon Storage

Injection



Extraction?



Treatment



- Storage formations almost certainly contain water.
 - Not suitable for drinking water (total dissolved solids [TDS] >10,000 ppm)
 - Potential for additional water resource (produced water associated with carbon storage)



The Plains CO₂ Reduction Partnership











Water and Carbon Storage

- Water resources:
 - Protect drinking water sources.
 - Minimize brine mobilization.
- Proper site characterization and selection along with monitoring, verification, and accounting (MVA) and appropriate regulatory compliance will ensure these goals are met.



















Water and Carbon Storage

- Storage targets and water:
 - Oil/gas reservoirs, coal seams, saline aquifers
- Each represents unique storage environments.
- Water may be extracted from storage activities in order to:
 - Enhance recovery.
 - Increase storage volume.
 - Increase plume management control.











RESEARCH AND DEVELOPMENT OF THE PROGRAMS. OPPORTUNE TECHNOLOGY COMMERCIAL WORLD-CLASS Extracted Vater CENTERS OF EXCEL

- Typically, water extraction is expected to be avoided.
- If extracted, management strategies include:
 - Reinjection.
 - Minimal treatment for beneficial use.
 - Substantial treatment for beneficial use.



















Options for Extracted-Water Management

- Water quality potential for extracted water:
 - TDS range = low to very high.
- Water quality needed for beneficial use:
 - Most uses require lower TDS concentrations.
- Water treatment processes:
 - Desalination methods
 would be used to
 match extracted-water
 quality to water quality
 needs.





Extracted-Water Treatment Technologies

- Managing salinity
 - Reverse osmosis
 - Forward osmosis
 - Membrane filtration
 - ♦ Micro-, ultra-, and nanofiltration
 - Electrodialysis and electrodialysis reversal
 - Thermal treatment



- Crystallizer, mechanical vapor compression/recompression, multieffect distillation, industrial evaporation, solar evaporation, and freeze-thaw/evaporation
- Ion exchange and other deionization (e.g., Higgins Loop[™])













CCS Produced Water – Beneficial Use Potential



RESEARCH AND DEVELOPMENT PROGRAMS, OPPORTUNIS TECHNOLOGY COMMERCIAL WORLD - CLASS CENTERS OF EXCEPTION

RESEARCH AND DEVELOPMENT PROGRAMS, OPPORTUNITIES FOR TECHNOLOGY COMMERCIATIZATION NOAL D - C L A S CENTERS OF EXCELLENCE ENVIRONMENTAL TECHNOLOGY

Imperative to any CCS operation.

Establish baseline conditions for naturally occurring CO_2 levels present in surface water and shallow groundwater aquifers in the vicinity of the carbon storage formation.

Provide a source of data to show that surface environments remain unaffected by fluid or gas migration and to identify the source and quantify of the impact of an out-of-zone migration event should it occur.





Additional Information



Fact Sheets



Presentations



















Regional Carbon Sequestration Partnership Water Working Group



Regional Carbon Sequestration Partnership Water Working Group

Introduction

Members of the U.S. Department of Energy (DOE) Regional Carbon Sequestration Partnerships (RCSPs) have formed the Water Working Group (WWG), a team of experts from government, academia, and industry whose goal is to address stakeholder concerns regarding emerging carbon capture and storage (CCS) technology and its potential interactions with local and regional water resources. Members of the WWG represent different regions of North America, each with its own unique set of challenges surrounding water resources and CCS (Figure 1). The opportunities and challenges at the nexus of CCS and water are being evaluated by the RCSP WWG as various carbon dioxide (CO) capture and storage strategies are assessed.

Carbon Capture and Storage

A majority of CO₂ generated by humans comes from the use of fossil fuels as reliable sources of energy, helping us to maintain our current economy and quality of life. Carbon dioxide emissions can be reduced through energy conservation, increased fossil fuel efficiency, increased utilization of renewable sources of energy and nuclear power, and implementation of CCS. CCS holds the



Figure 1. DOE has organized seven RCSPs to evaluate a variety of CO, storage strategies to determine which is best suited for specific regions of the country.

potential to substantially reduce greenhouse gas emissions to the atmosphere and is most efficient when applied to large utility or industrial sources where high volumes and/or concentrations of CO₂ are emitted. Through the use of specialized processes and equipment, CO₂ is captured, compressed, and transported to sites appropriate for safe long-term geologic storage (Figure 2).



Figure 2. CO₂ is pumped 4800 feet underground at the CO₂ injection site in the Weyburn–Midale Field in Saskatchewan, Canada.

Underground storage entails injecting compressed CO, into deep rock formations that are both physically and chemically stable; have an appropriate amount of porosity (spaces within the rock); and are covered by thick, relatively impermeable (flow-resistant) rock formations that confine the CO₂ at depths typically greater than 1 mile.

Water and CCS

Water is involved in every step of the CCS process (Figure 3). Current capture technologies require additional water supplies at the site of CO, generation, either as a direct result of the capture process, or indirectly through parasitic electrical demand and the associated cooling water requirements for thermoelectric power generation. Within the reservoir itself, the impact of storage activities on appropriately targeted rock formations has been shown to be minimal. CCS activities require great depth, and in most cases, the targeted formations will be separated from potable water resources by hundreds to thousands of feet of rock, including multiple bow-permeability barries. Producing

- Who is the WWG?
- What is the water and CCS nexus?
- What are the opportunities and challenges associated with water and CCS?

















Carbon Capture and Storage: Protecting Freshwater Resources

- How is freshwater protected?
- What steps need to be taken?
- What are the existing regulations?
- What steps are needed for commercial deployment?



Introduction

The overall goal of carbon capture and storage (CCS) is to inject carbon dioxide (CQ) that has been captured from a point source, such as a power plant, into a deep underground storage formation and ensure that it remains there. Maintaining the security of that CQ₃ is crucial to protecting our water resources. This fact sheet identifies the keys to successfully protect water resources during CCS and introduces the evolving regulatory framework set up for that purpose.

CCS and CO₂ Containment

The commercial geologic injection of fluids has been done safely in the United States for decades and currently occurs under the Underground Injection Control Program, which is directed by state and federal regulatory agencies. Each day, large volumes of fluids are injected for waste disposal, enhanced oil recovery (EOR), and liquid hydracrabin and natural gas storage (Table 1). The subsurface systems encountered during CCS (Figure 1) are similar to those encountered during the deep injection activities identified in Table 1.

Keys to Successful Protection of Water Resources

oble 1. Injection Wall Types in the United Statesh

The keys to water resource protection during CCS include detailed site characterization, sound well construction and operation protocols, and comprehensive monitoring and



Figure 1. Illustrative example of CO₂ injection into a deep storage reservoir during CCS (image not to scale).

Well Type/Class	Number of Wells	Comments*	
Oil and Gas-Related Injection Wells (Class II)		Over 2 billion gallons of brine injected/day; 80% associated with EOR, the other 20% at natural gas/oil production facilities.	
Includes Brine Disposal and the Injection of CO ₂ and Other Fluids/Gases for EOR	~150,000	As of March 2010, 11 TCF (560 million metric tons) of CO ₂ ha been consumed by the U.S. CO ₂ EOR industry.	
Natural Gas Storage	400 active storage facilities in the lower 48 states	5000 to 7000 BCF of natural gas in storage/month.	
Liquid Hydrocarbon Storage (Class II)	100	Part of U.S. Strategic Petroleum Reserve.	
Hazardous Waste Disposal (Class I) as Defined by RCRA ^b	120	Generally located at industrial facilities.	
Nonhazardous Industrial Waste Disposal (Class I)	260	Currently operate in 19 states, primarily Texas, Wyoming, Kansas, and Louisiana.	
Municipal Wastewater Disposal (Class I)	160	Primarily in Florida; large diameter and gravity-fed.	















Monitoring, Verification, and Accounting Plans for Protection of Water Resources During the Geologic Storage of Carbon Dioxide



Monitoring, Verification, and Accounting Plans for Protection of Water Resources During the Geologic Storage of Carbon Dioxide

Introduction

Geologic storage (GS) is an evolving strategy being investigated for the long-term management of carbon dioxide (CO₃) emissions. A key component of CO2 GS is the presence of cost-effective and efficient CO2 monitoring, verification, and accounting (MVA) programs designed to demonstrate that each GS site is performing as anticipated, CO₂ is being sequestered, and water resources are being protected (Figure 1). The U.S. Department of Energy's (DOE's) Carbon Storage Program is researching, developing, and demonstrating a wide variety of the technologies for GS, including those needed for MVA. This fact sheet, developed by DOE's Water Working Group, focuses on the monitoring aspects of the MVA framework and provides an overview of the monitoring technologies that are being investigated for the protection of water resources.

MVA Monitoring Framework for GS Projects

An MVA monitoring framework has evolved that captures both the physical and temporal aspects of a typical GS project. This monitoring framework focuses on three distinct vertical zones: atmospheric, near surface, and subsurface, during four distinct periods of operation; prospectation (or baseline), operation, closure, and postclosure (U.S. Department of Energy, 2009) (Figure 2). The framework comprises appropriate monitoring technologies needed to validate GS storage performance and to meet applicable EPA permit requirements. For the protection of water resources, i.e., underground sources of dinking water (USDW) and surface water bodies, these potential pathways include 1) natural leakage from the reservoir through cap rock seals, 2) leakage from the reservoir sthrough cap rock faults and fractures, and 3) leakage from textivities are the first line of defense



Figure 1. Water sampling at a CO₂ GS demonstration site.

for addressing these migration pathways, with the goal of screening out those sites where they pose a measureable risk. However, the potential existence of these pathways should still be addressed on a site-by-site basis along with other potential pathways that may be unique to individual sites.

An MVA plan contains a mixture of monitoring techniques designed to detect the presence or absence of migration along each of these pathways and to provide assurance that storage site integrity is maintained. Should migration along a potential leakage pathway be detected, the MVA plan provides the basis for implementing mitigation strategies (e.g., halting CO, injection or implementing pump and treat scenarios), if required, to prevent and/or reduce the impacts of this migration to water resources.

- How does MVA protect water resources?
- What technologies are available?
- How is MVA carried out at large-scale projects?
- What is required for successful deployment?















Long-Term Protecion of Freshwater Resources Following CO₂ Storage

- How is long-term storage achieved? How do we know freshwater resources are protected in the long-term?
- What are the dominant CO₂ storage mechanisms?
- How is long-term storage demonstrated by RCSP programs?



Long-Term Protection of Freshwater Resources Following CO₂ Storage

Introduction

The subsurface geologic storage of carbon dioxide (CO₂) represents a primary option for achieving the reduction of greenhouse gas emissions to the atmosphere.

Important to the successful commercial deployment of the geologic strappe of CQ₀ are 1) good site selection and 2) implementation of conventional, as well as innovative, monitoring methods to ensure that active carbon capture and storage (CC2) operations are performing properly and are protective of human health and the environment. Equally important to commercialization is being able to provide assumances to the general public that impacts to human health and the environment will not occur.

This fact sheet identifies the primary physical and chemical mechanisms that are being relied upon to ensure the long-term containment of CO_2 in the storage reservoir following injection.

Primary Mechanisms for the Subsurface Storage of CO_2

Target rock formations for geologic storage, such as depicted oil and gas reservoirs or deep saine formations, are much deeper than any usable groundwater and are separated from that groundwater by thick barriers of Impervious rock (Figure 1), Generally, these formations have already proved their effectiveness in containing CO₂ by keeping highly safty safity water separate from usable groundwater for millions of years.

Following injection into the subsurface, several physical and chemical mechanisms actively store CO₂ and have the potential to ensure that the subsurface movement of CO₂ does not occur beyond the boundaries of the storage system. These 'trapping mechanism' include structural/stratigraphic hydrodynamic, mineral, residual-phase, and solubility trapping. Each of these mechanisms' binefly described below.

Structural/Stratigraphic Trapping. In a structural/stratigraphic trap, (O₂ is physically trapped at the top of an anticline or in a tilted fault block. It is kept from further upward movement by the sealing rock (or cap rock).

 $\label{eq:Hydrodynamic Trapping.} Hydrodynamic trapping results when the travel time of CO_2 in low-permeability storage aquifers is on the$



Figure 1. Geologic storage of CO₂ occurs deep below the surface and is separated from freshwater resources by thousands of feet of rock and impermeable barriers (modified from Peck and others, 2012).

order of thousands to millions of years. Factors that have a substantial influence on the length of the migration pathway include 1) stratigraphic heterogenetites (eg., siltstones, shales, and coals in the reservoir nock), 2) geochemical reactions, and 3) temperature.

Mineral Trapping. When dissolved CD, reacts with the reservoir rock, carbonate minerals can form and precipitate, trapping the CD, in stable chemical forms. The rate and extent of these reactions depend on the composition of the reservoir rock, the temperature and pressure of the reservoir, the chemical composition of the water, the water-rock contact area, and the rate of fluid flow through the rock. Mineral trapping is considered the most secure stage of CD, trapping.

Residual-Phase Trapping. When free-phase CO₂ migrates, it forms a plume. At the tail of this plume, the concentration of CO₂ decreases, and it becomes trapped in the pore spaces between the rock by capillary pressure from the water, which stops its movement. Over time, this residually trapped CO₂ can dissolve into the formation water, promoting even more secure mineral tapoling.

















WORLD-CLASS WWGWebSite Centers of Excelled

www.netl.doe.gov/research/coal/carbon-storage/wwg

NETL HOM	About Events Library Careers	
RCSP WATER WORK		
Infrastructure Global NATCARB/Atlas Program Plan Project Portfolio Publications Carbon Storage Newsletter FAQs Contacts	MARCHART REAL AND A CONTRACT OF A CONTRAC	
	[The following will be accessible in the left navigation of all pages when released live] • WWG Home • CCS • Water and CCS • Challenges & Opportunities • Links • About Us	
The Plains CO, SECAR	WWG PUBLICATIONS	BIG SKY CARBON

RESEARCH AND DEVELOPMENT OF THE SEARCH AND DEVELOPMENT OF THE SEAR

- The WWG is focused on developing a final capstone document summarizing the "state of the science" at the nexus of water and CCS. Expected to be completed fall 2016.
- Developing a Special Issue of the *International Journal of Greenhouse Gas Control* centered on these issues, with participation from authors beyond the WWG. Expected to be published spring 2016.

















For More Information, Contact:

E-Mail: wwg@undeerc.org

Ryan J. Klapperich, Research Scientist Energy & Environmental Research Center rklapperich@undeerc.org Telephone No. (701) 777-5340

Andrea T. McNemar, Project Manager National Energy Technology Laboratory Andrea.McNemar@NETL.DOE.GOV Telephone No. (304) 285-2024



















RESEARCH AND DEVELOPMENT PROGRAMS, OPPORTUNITIES FOR TECHNOLOGY COMMERCIALIZATION WORLD-CLASSING / O CENTERS OF EXCELLENCE ENVIRONMENTAL FECTION OF OCCUS

Acknowledgment

This material is based upon work supported by the U.S. Department of Energy National Energy Technology Laboratory under Award No. DE-FC26-05NT42592.

Disclaimer

This presentation was prepared as an account of work sponsored by an agency of the United States Government. Neither the United States Government, nor any agency thereof, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.













