

# Degradation & Performance Studies of ALD-Stabilized Nano-Composite SOFC Cathodes DE-FE0031250 – FOA 1735

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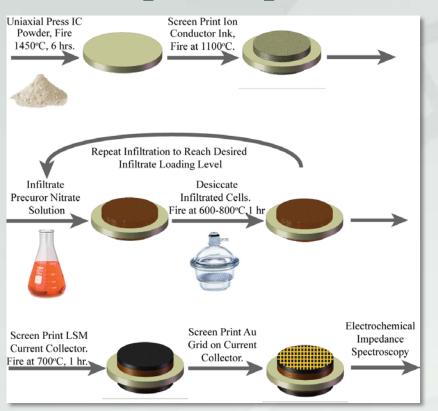
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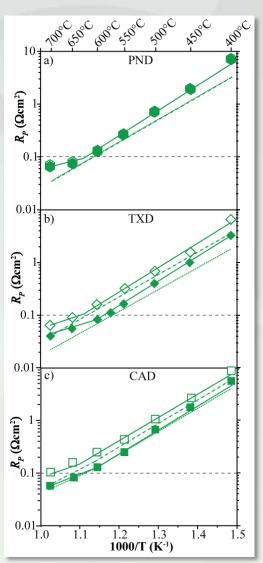
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November 30, 2017

## Previous work by the Nicholas Group has shown that the desiccation of LSCF infiltrate solution leads to improved performance of $0.1\Omega cm^2$ at $\sim 575^{\circ}C$





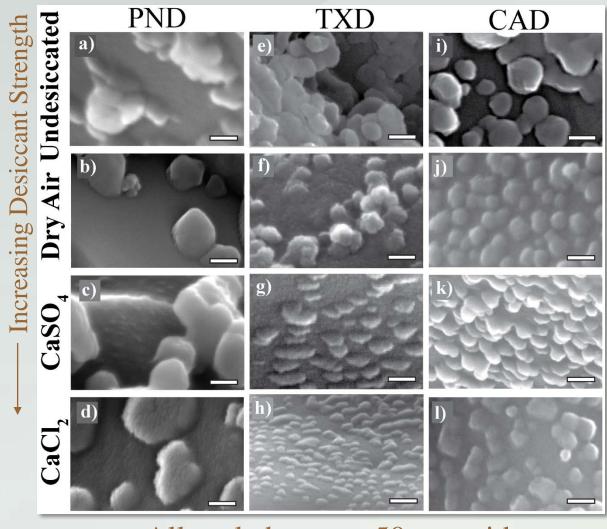
PND= Pure Nitrate Derived

TXD= Triton X-100 Derived

CAD=Citric Acid Derived

Inclined dashed lines are SIMPLE model predictions made with the observed particle size predictions

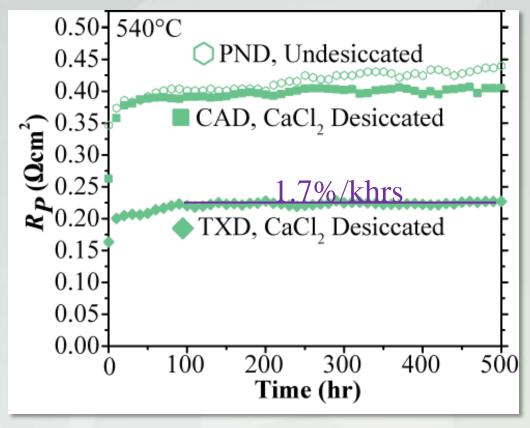
## Precursor Solution Desiccation Reduces Average La<sub>0.6</sub>Sr<sub>0.4</sub>Co<sub>0.8</sub>Fe<sub>0.2</sub>O<sub>3-x</sub> (LSCF) Infiltrate Particle Size



PND=Pure Nitrate
Derived
TXD= Triton X-100
Derived
CAD=Citric Acid
Derived

All scale bars are 50 nm wide

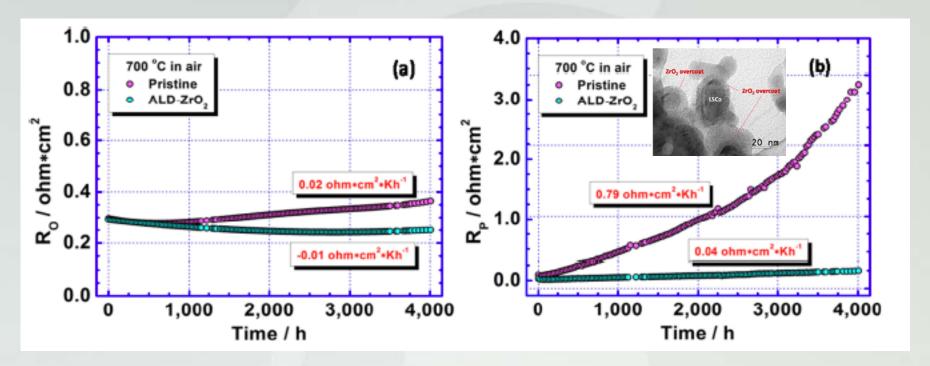
## Unfortunately, uncoated nano-composite cathode stability is a major concern



DOE target degradation rate: 0.2%/khrs

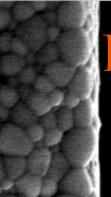
Commercial cells degradation rate: 0.5%-1%/khrs

## Previous Work by the Huang Group has Shown that Electrodes with ALD-ZrO<sub>2</sub>Coated LSC nanoparticles have improved stability and good performance



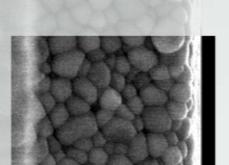
LSC on LSGM ( $La_{0.8}Sr_{0.2}Ga_{0.83}Mg_{0.17}O_{3-\delta}$ ) scaffold (firing nitrates with TX-100 at 800°C for 1h)

➤ The assumed stabilization mechanism is the prevention of LSC nanoparticle coarsening Sr segregation

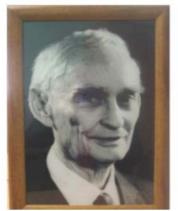


#### RESEARCH OBJECTIVES

- Optimize the ALD overcoat microstructure to maximize the performance and stability of infiltrate size-tailored LSCF-GDC cathodes
- Deconvolute the degradation mechanisms active in sizetailored, ALD-stabilized LSCF-GDC cathodes
- Evaluate the performance and stability of size-tailored, ALDstabilized Co<sub>3</sub>O<sub>4</sub>-GDC cathodes



## Atomic layer deposition (ALD) has been used for thin-film deposition for over 50 years.

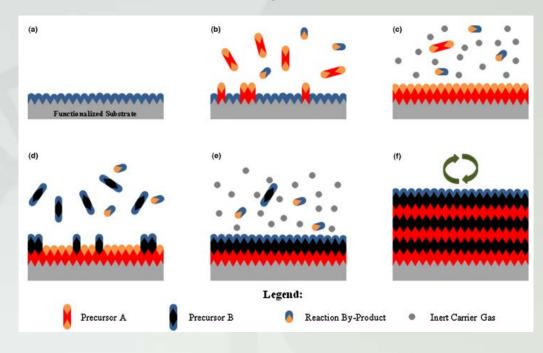




ALD was invented by Valentin Borisovich Aleskovskii (left) and Stanislav Ivanovich Koltsov (right) in 1960s

Huang Group Nanotech Savannah 200 Atomic Layer Deposition System



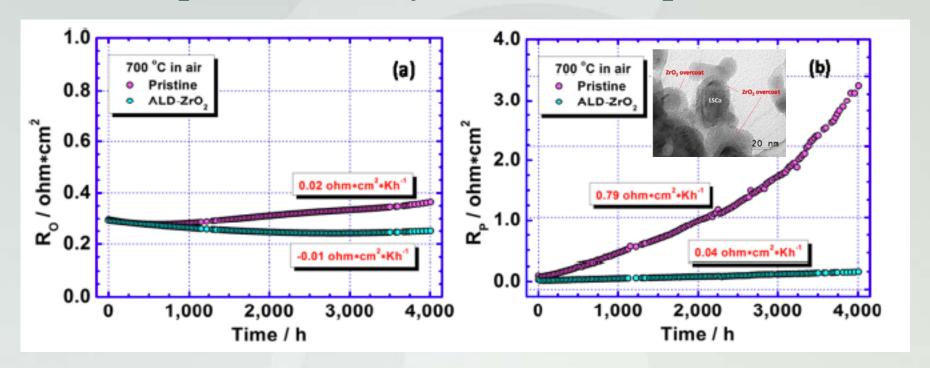


Precursor A deposition—Excess precursor and by-products purged by inert gas—Precursor B deposition—Excess precursor and by-products purged

➤ ALD can form conformal layer structures at modest temperature (<350°C)

Johnson, Richard W., Adam Hultqvist, and Stacey F. Bent. *Materials today* 17.5 (2014).

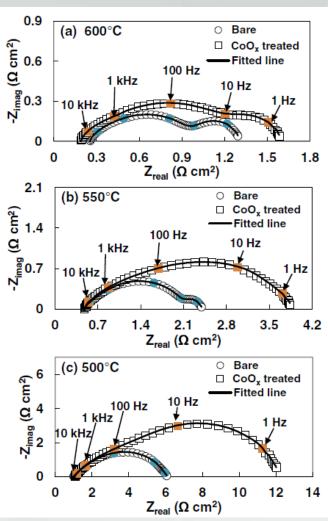
## Previous Work by the Huang Group has Shown that Electrodes with ALD-ZrO<sub>2</sub>Coated LSC nanoparticles have improved stability and mediocre performance



LSC on LSGM ( $La_{0.8}Sr_{0.2}Ga_{0.83}Mg_{0.17}O_{3-\delta}$ ) scaffold (firing nitrates with TX-100 at 800°C for 1h)

➤ The assumed stabilization mechanism is the prevention of LSC nanoparticle coarsening Sr segregation

## There is literature suggesting ALD coatings may hurt cathode performance



- ➤ 1.8nm Co<sub>3</sub>O<sub>4</sub> was coated on LSC thin film (PLD, 2.5um)
- ➤ Co<sub>3</sub>O<sub>4</sub> coated LSC film performed worse than bare cell (Ni-YSZ as cathode and YSZ as electrolyte)
- ➤ They claimed Co<sub>3</sub>O<sub>4</sub> layer blocked the sites on LSC and affected the O<sub>2</sub> adsorption and/or desorption process..

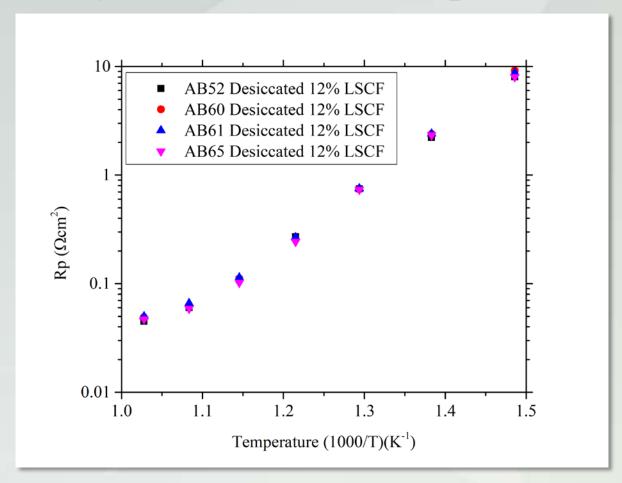
#### **Phase I SOPO Tasks**

- Task 1.1: Produce ALD-coated, infiltrate-size-tailored LSCF-GDC symmetric cells (Huang and Nicholas)
- Task 1.2: Evaluate symmetric cell performance and stability and identify optimal ALD porosity/thickness/deposition conditions (Huang, Validation by Nicholas)
- Task 1.3: Produce ALD-coated infiltrate size tailored LSCF-GDC SOFCs (Huang, Nicholas, Atrex)
- Task 1.4: Perform long-term fuel cell tests to evaluate electrode performance and stability under SOFC operation (CH<sub>4</sub> at 650C, H2 at 450-650C) (Atrex Energy, Inc.)

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## Reproducible ALD-Ready desiccated LSCF-GDC symmetric cells can be produced



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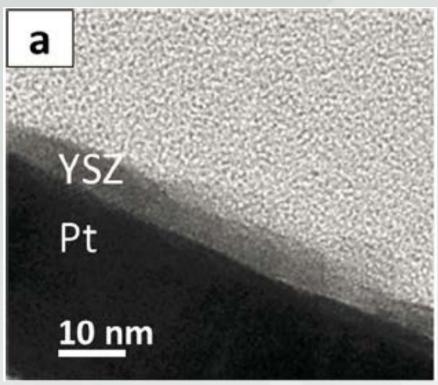
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#### **Atrex Energy's Test Capabilities**

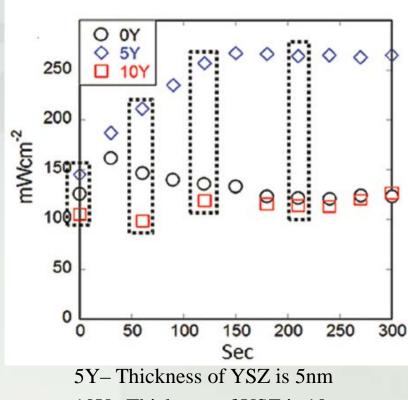


- Capacity to test 35 tubular cells over 9 cell test stands.
- ➤ Independent load control and fuel flow for each cell in every test stand, which typically operated in constant current mode.
- ➤ Typical fuel include hydrogen and methane reformate.
- The cell testing laboratory has natural gas, compressed air, hydrogen, pure methane, propane and reducing cover gas plumbed around a central manifold system.

#### ALD coatings have been reported to change ORR performance

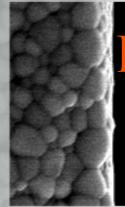


Cross section TEM for YSZ coated Pt



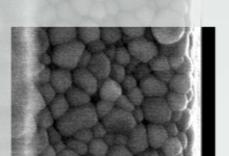
10Y–Thickness of YSZ is 10nm

- > YSZ has been coated on Pt and it has changed the ORR performance of Pt cathode.
- ➤ It's possible that ZrO<sub>2</sub> ALD Overcoats may alter the ORR performance of SOFC infiltrate particles as well



#### RESEARCH OBJECTIVES

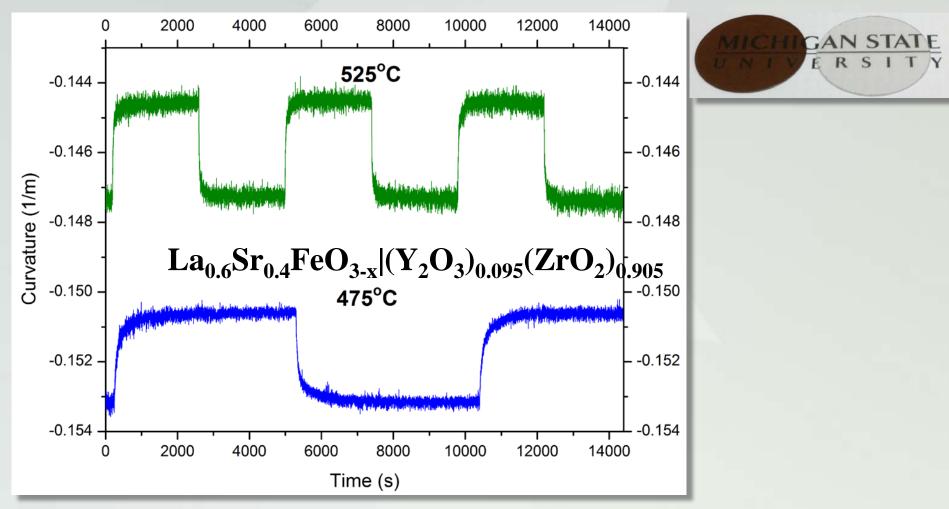
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#### **Phase II SOPO Tasks**

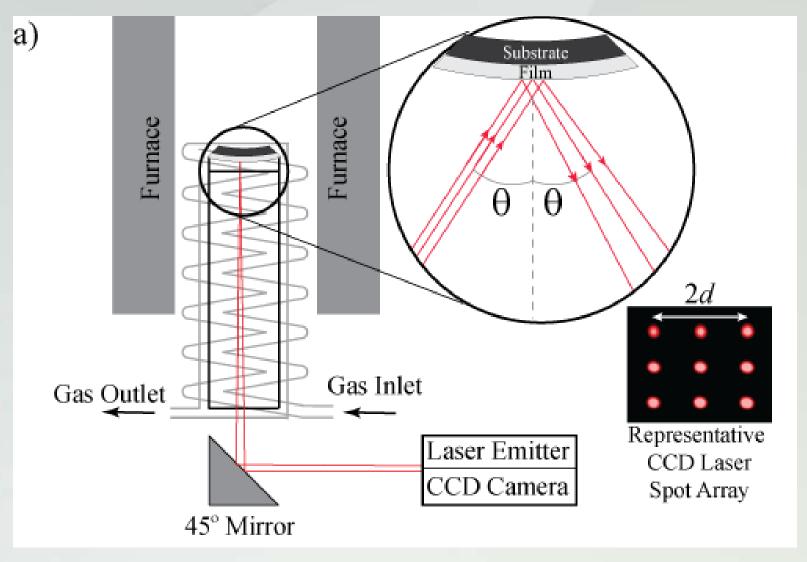
- Task 2.1: Quantify infiltrate particle size coarsening (Huang, Validation by Nicholas)
- Task 2.2: Produce ALD-coated thin film LSCF samples (Nicholas and Huang)
- Task 2.3: Perform LSCF Wafer Curvature  $k_{chem}$  Measurements vs Aging Time
- Task 2.4: Use Simple Models to Relate  $R_P$ ,  $k_{chem}$ , and Infiltrate Particle Size Changes (Nicholas)

#### Representative Curvature Data for a Mechano-Chemically Active Material on an Inert Substrate Responding to a Sudden pO2 Switch from 0.2 to 0.02 atm



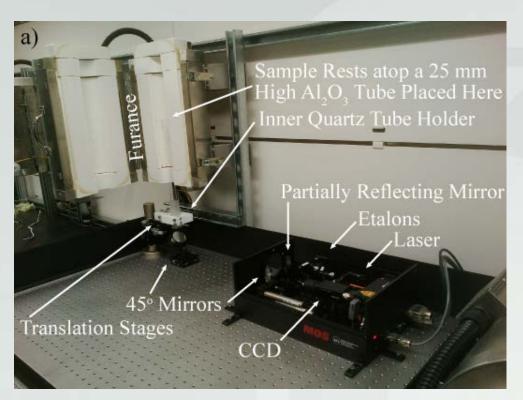
Yang et al. ECS Transactions. 224<sup>th</sup> ECS Meeting - San Francisco, CA (2013). Yang *et al.* Solid State Ionics, 249–250, 123–128 (2013).

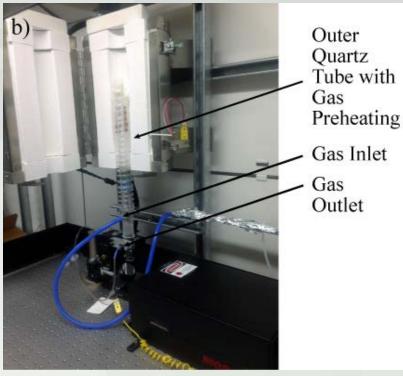
#### In Situ/In Operando Wafer Curvature Measurements



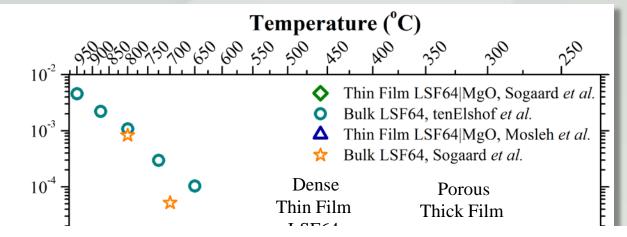
Nicholas, Extreme Mechanics Letters, (2016) In Press. <a href="http://dx.doi.org/10.1016/j.eml.2016.04.006">http://dx.doi.org/10.1016/j.eml.2016.04.006</a>

### The Nicholas Group In Situ Wafer Curvature Measurement Setup

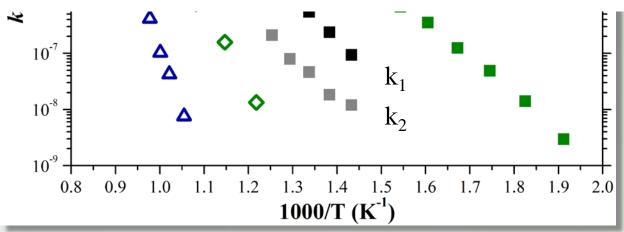




### The KR-Measured k's Have Activation Energies that Agree with Bulk La<sub>0.6</sub>Sr<sub>0.4</sub>FeO<sub>3-x</sub> Values



#### No ALD Coated Oxygen Surface Exchange Measurements Have Ever Been Performed!

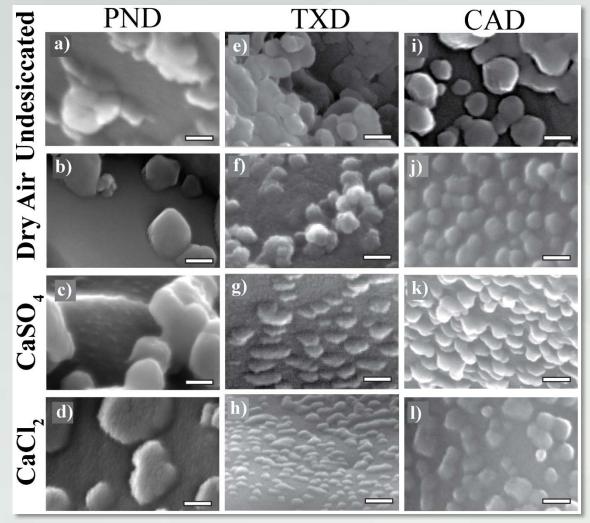


Sogaard et al. J. Solid State Chem, 180, 1489 (2007); Mosleh et al. J. Electrochem. Soc., 156, B441 (2009); Sogaard et al. J. Electroceramics, 27, 134 (2011); ten Elshof, J. Electrochem. Soc., 144, 1060 (1997); Yang et al. Solid State Ionics 249–250 (2013) 123–128; Yang and Nicholas *J. Electrochem. Soc.*, 161, F3025-F3031 (2014).

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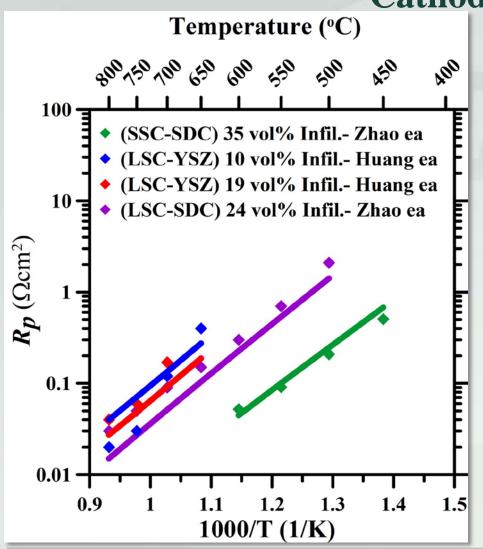
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#### BET and/or SEM infiltrate particle size analysis can be used to track infiltrate coarsening



All scale bars are 50 nm wide

### SIMPLE Model Fits to Literature Nano-Composite Cathodes



$$SSC=Sm_{0.5}Sr_{0.5}CoO_{3-x} \\ SDC=Sm_{0.2}Ce_{0.8}O_{1.9} \\ LSC=La_{0.6}Sr_{0.4}CoO_{3-x} \\ YSZ=Y_{0.08}Zr_{0.92}O_{1.96}$$

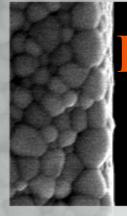
Obtain Custom Rp Predictions for Your Infiltrated Cells Using the SIMPLE Model Calculator at:

https://www.egr.msu.edu/nicholasgroup/simple.php

Zhao et al. Journal of Power Sources, v185, p13 (2008).

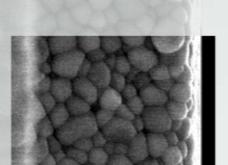
Zhao et al. Materials Research Bulletin, v43, p370 (2008).

Huang et al. Journal of the Electrochemical Society, v151, pA1592 (2004).



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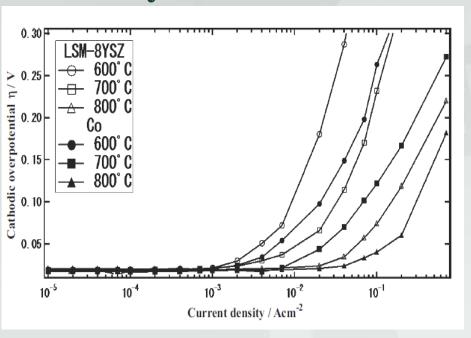
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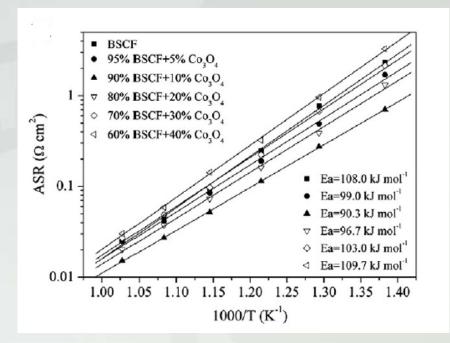


#### **Phase III SOPO Tasks**

- Task 3.1: Produce ALD-coated, infiltrate-size-tailored Co<sub>3</sub>O<sub>4</sub>-GDC electrodes (Nicholas and Huang)
- Task 3.2: Measure the Performance and Stability of ALD-coated infiltrate size tailored Co<sub>3</sub>O<sub>4</sub>-GDC electrodes (Huang, Validated by Nicholas)

## Co<sub>3</sub>O<sub>4</sub> additions are known to improve the performance of many SOFC cathode materials like LSM and BSCF

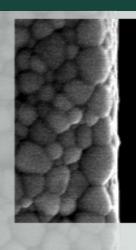


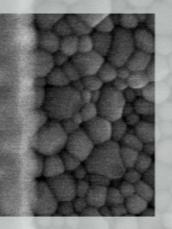


- ➤ LSM performance at different temperatures with and without Co<sub>3</sub>O<sub>4</sub>
- ➤ BSCF performance with different loading level of Co<sub>3</sub>O<sub>4</sub>

#### ➤ Co<sub>3</sub>O<sub>4</sub> can also improve the performance of GDC, LSC and SSC

Table 1 Cathode polarization resistances and maximum power densities of fuel cells with different cathode electrocatalysts at 600 °C						
	GSC	GSC/Co <sub>3</sub> O <sub>4</sub>	LSC	LSC/Co <sub>3</sub> O <sub>4</sub>	SSC	SSC/Co <sub>3</sub> O <sub>4</sub>
OCV/V	0.89	0.887	0.88	0.882	0.913	0.893
$P_{\rm max}/{\rm mW~cm^{-2}}$	497	705	463	821	595	1033
$R/\Omega$ cm <sup>2</sup>	0.252	0.171	0.269	0.136	0.233	0.073







- Performance and stability data on LSCF-GDC and Co<sub>3</sub>O<sub>4</sub>-GDC cathodes ALD stabilized with various ZrO<sub>2</sub> overcoat microstructures
- The world's first k<sub>chem</sub> measurement on an ALD-coated oxygen exchange material
- k<sub>chem</sub> and infiltrate particle size aging data and an analysis of whether the observed trends can explain the measured infiltrated cathode aging behavior
- Documented exchange of PI Nicholas's sizetailored infiltrate processing knowledge and PI Huang's ALD coating knowledge to Atrex Energy