

Zeolite Membrane Reactor for Pre-Combustion Carbon Dioxide Capture

*Lie Meng and Jerry Y.S. Lin**

Arizona State University

DOE Award:

DE-FE0026435

2018 NETL CO₂ Capture Technology Project Review Meeting

August 14, 2018, Pittsburgh, Penn



Overview

Timeline

- Project start date:
Oct. 1, 2015
- Project end date:
Jan. 31, 2019
- Budget Periods:
I: 10/1/2015-7/30/2017
II: 8/1/2017-1/31/2019

Budget

- Total project funding
 - DOE **\$2,760,797**
 - Cost-share: **\$689,963**
 - Total: **\$3,450,760**

Research Area

2B2: Bench-Scale Pre-Combustion CO₂ Capture Development and Testing

Partners

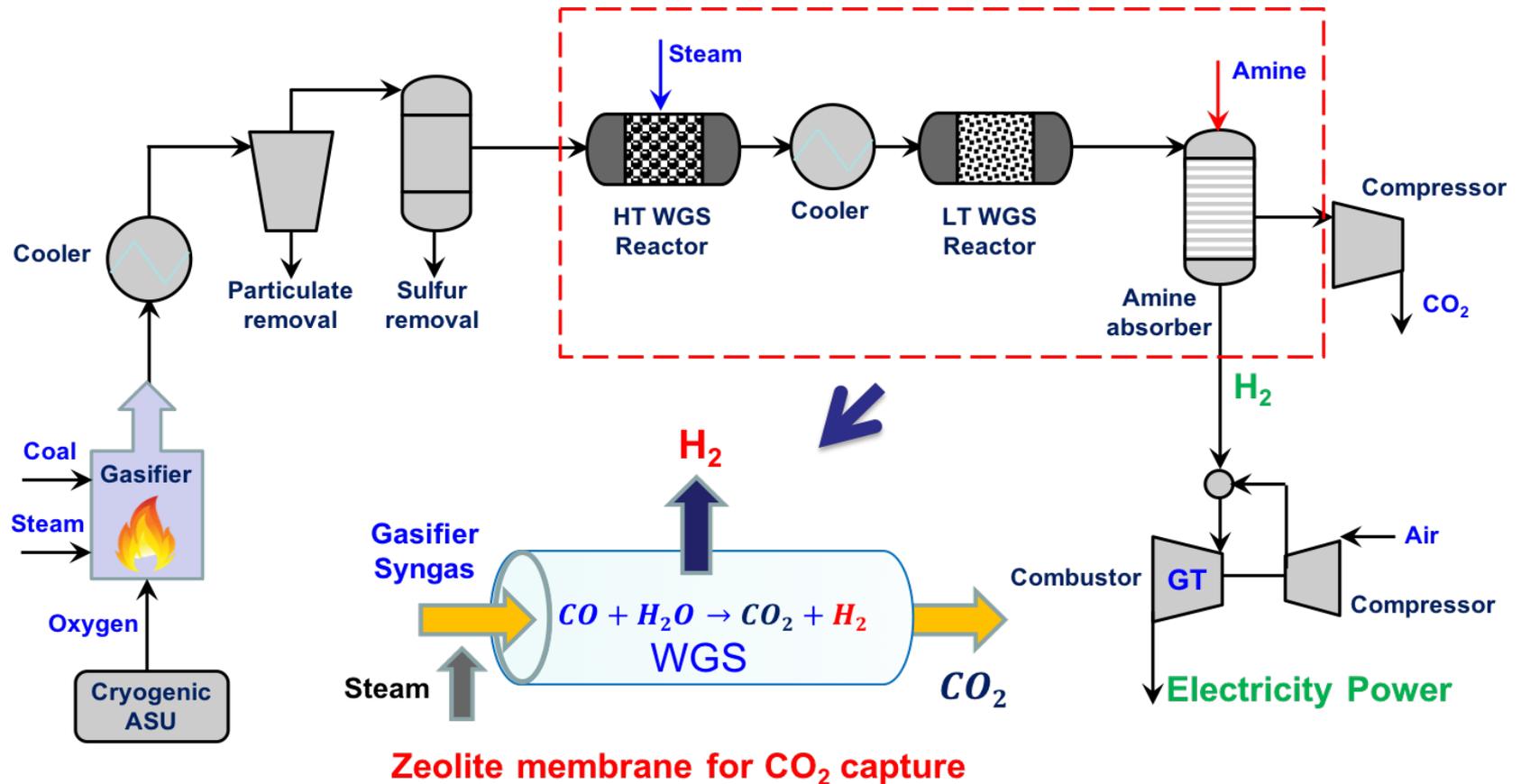
- Arizona State University (ASU)
- University of Cincinnati (UC)
- Media and Process Technology, Inc (MPT)
- Nexant, Inc.
- University of Kentucky Applied Energy Research Center

Project Objectives

To demonstrate a bench-scale zeolite membrane reactor (ZMR) for WGS reaction of coal gasification gas for hydrogen production for integration with IGCC power plant.

To evaluate the performance and cost-effectiveness of this new membrane reactor process for use in 550 MW coal-burning IGCC plant with CO₂ capture.

Zeolite Membrane Reactor for Water-Gas Shift Reaction for CO₂ Capture

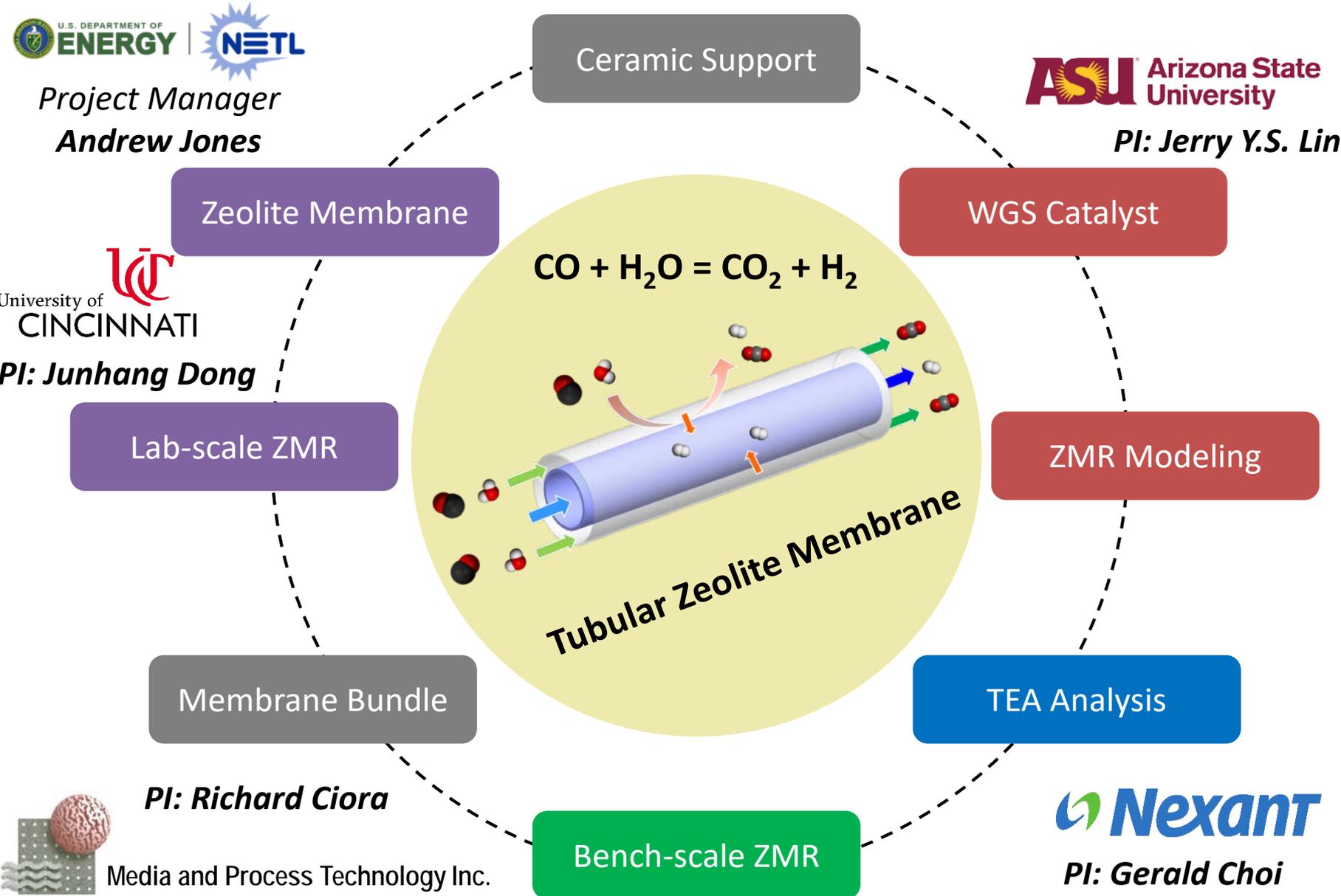


Zeolite Membrane Requirements:

- Operate at 350-550°C
- Chemically stable in H₂S, thermally stable at ~500°C
- H₂ permeance > 1x10⁻⁷ mol/(m².s.Pa) (>300 GPU) with H₂/CO₂ selectivity > 10

DOE Project: Zeolite Membrane Reactor for Pre-Combustion CO₂ Capture

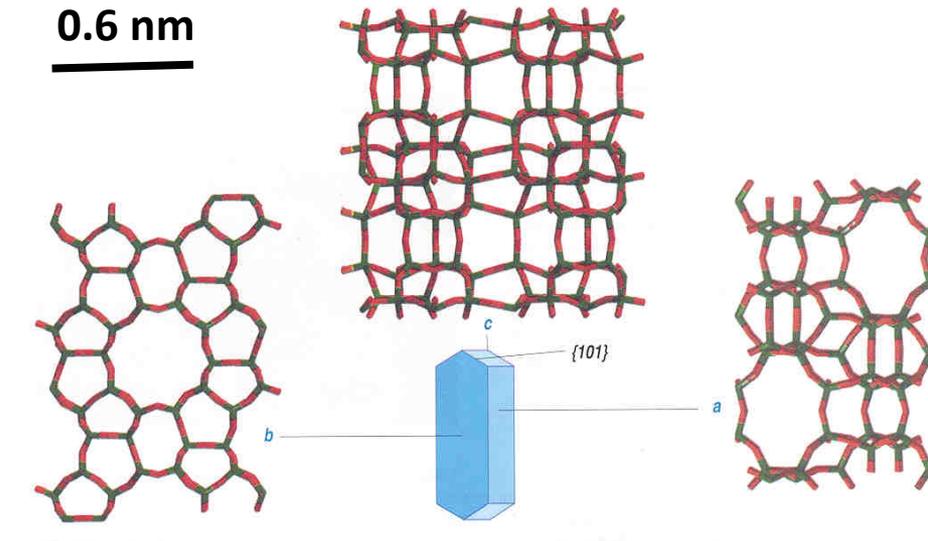
Task description



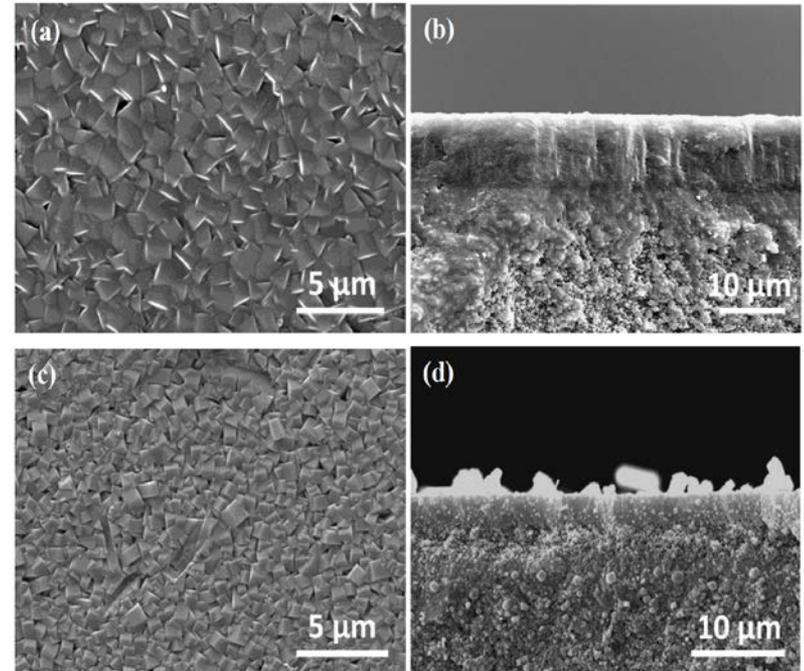
MFI-type Zeolite

Structure and property

MFI-type Zeolite (Silicalite-1 or ZSM-5)



- Molecular sieving at high temperatures
- Highly chemically and thermally stable (up to 700°C)

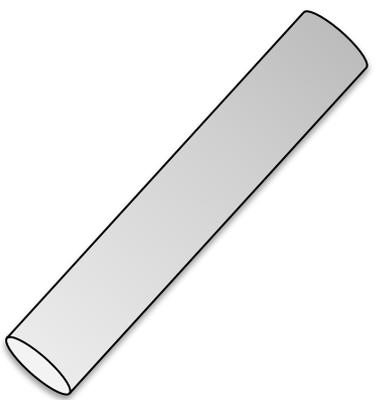
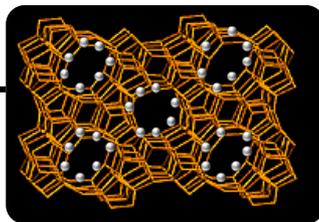


Surface and cross-section SEM images of (a, b) templated synthesized random oriented MFI membrane, and (c, d) template-free synthesized random oriented MFI membranes (from Lin lab)

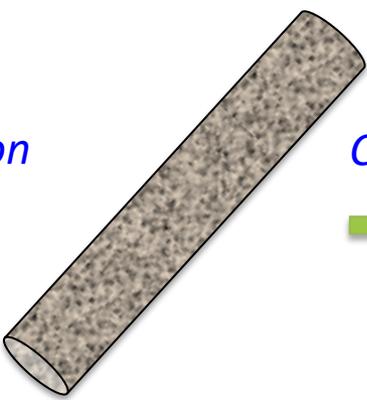
- Tailorable structure

Tubular MFI-type Zeolite Membranes

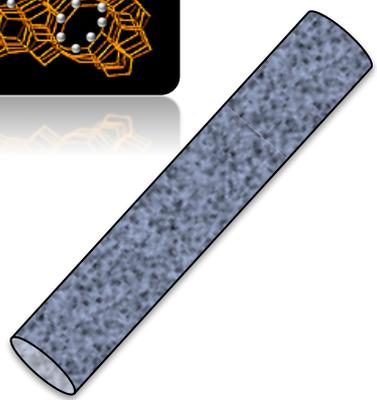
Membrane preparation and property



in-situ crystallization



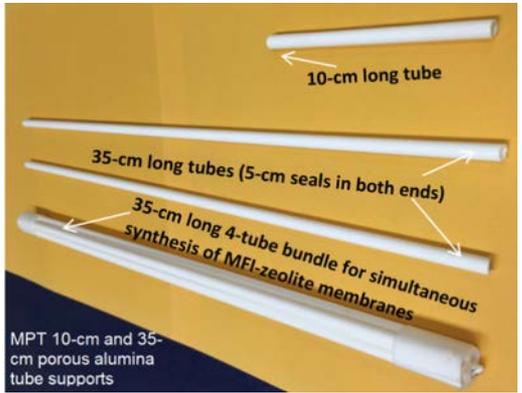
*CCD modification**



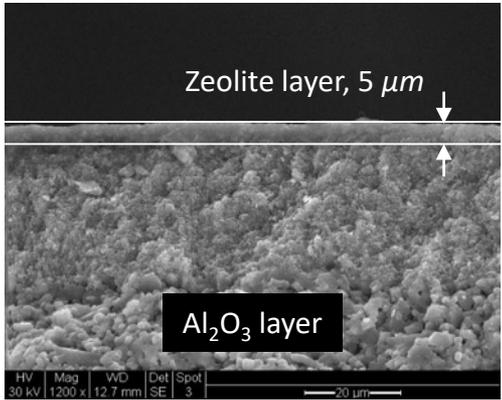
Al₂O₃ tubular support

MFI zeolite membrane

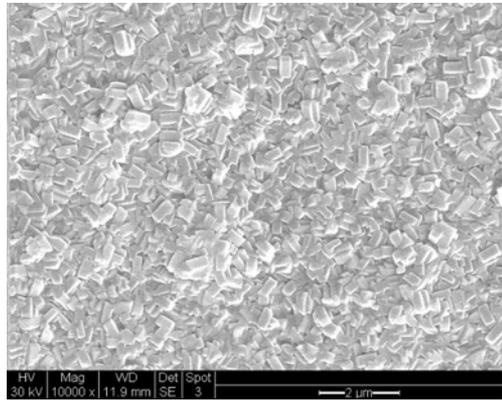
Modified zeolite membrane



OD = 5.7 mm; ID = 4.7 mm
Pore size < 100 nm



25°C: H₂/CO₂ = 0.1-0.4
450°C: H₂/CO₂ = 4.0-5.0



25°C: H₂/CO₂ = 1.5-3.0
450°C: H₂/CO₂ = 10-45

*Catalytic Cracking Decomposition of Methyl-diethoxysilane (MDES)

Scope of work

- 1) **Scaling up ZMRs from lab-scale to bench-scale for combined WGS reaction and H₂ separation**
- 2) **Conducting a bench-scale study using these ZMRs for hydrogen production for IGCC with CO₂ capture.**

Goal is to demonstrate effective production of H₂ and CO₂ capture by the **bench-scale** zeolite membrane reactor from a coal gasification syngas at temperatures of 400-550°C and pressures of 20-30 atm:

- Bench-scale zeolite membrane reactor: **21 zeolite membrane tubes** of 3.5 ID, 5.7 OD and **25-cm long** (active)
- A system producing H₂ at rate of about 2 kg/day, equivalent to a 2 kW_{th} IGCC power plant

General Approach to Scaling up WGS-ZMR

Single-tube zeolite membrane reactor: study WGS up to 30 atm by experiments and modeling



Intermediate-scale zeolite membrane reactor: 3-7 tube membrane module for WGS reaction



Bench-scale zeolite membrane reactor: 21 tube membrane module for WGS reaction at UK-CBTL

Zeolite membrane reactor in IGCC with CO₂ capture - process design and techno-economic analysis

Progress and Accomplishments

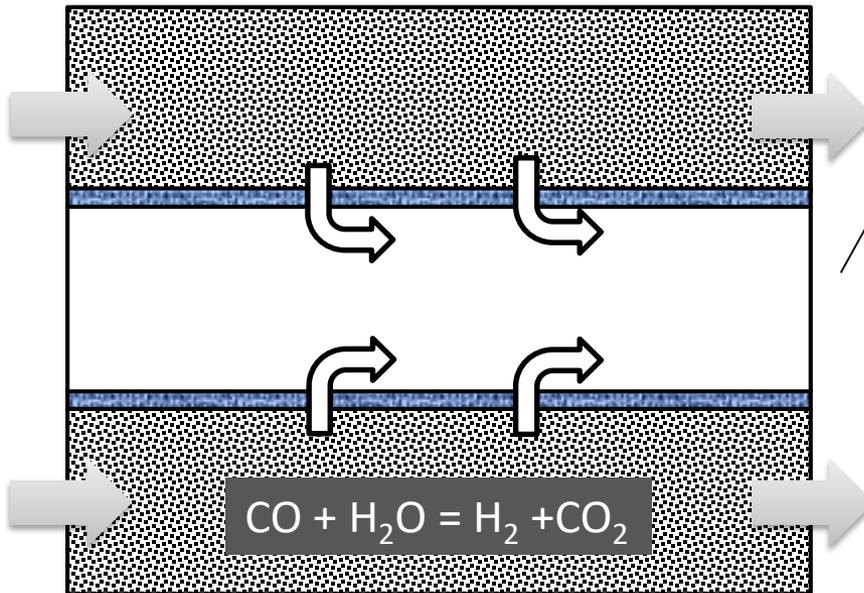
- **Modeling and Analysis of WGS in Bench-Scale Zeolite Membrane Modules (Task 10.0)**
- **Fabrication of Large Quality Tubule Supports (Task 11.0)**
- **Preparation of Large Quantity MFI Zeolite Tubule Membranes for Bench-Scale Module (Task 12.0)**
- **Design and Fabrication of Bench-Scale Zeolite Membrane Housing (Task 13.0)**
- **Building Bench-Scale Zeolite Membrane Reactors (Task 14.0)**
- **Testing WGS Reaction in Bench-Scale Membrane Reactor (Task 15.0)**
- **Process Design, Techno-Economic and EH&S Analyses (Task 16.0)**

Task 10: Modeling and Analysis of WGS in Bench-Scale Zeolite Membrane Modules

Research target for ZMR performance

Feed stream

syngas mixture



Permeate stream

H₂ recovery:

$$R_{\text{H}_2} = \frac{F_{\text{H}_2, \text{perm}}}{F_{\text{H}_2, \text{reten}} + F_{\text{H}_2, \text{perm}}} > 92\%$$

H₂ purity:

$$G_{\text{H}_2} = \frac{F_{\text{H}_2, \text{perm}}}{F_{\text{total, perm}}}$$

Retentate stream

CO₂ capture:

$$R_{\text{CO}_2} = \frac{F_{\text{CO}_2, \text{reten}}}{F_{\text{CO}_2, \text{reten}} + F_{\text{CO}_2, \text{perm}}} > 90\%$$

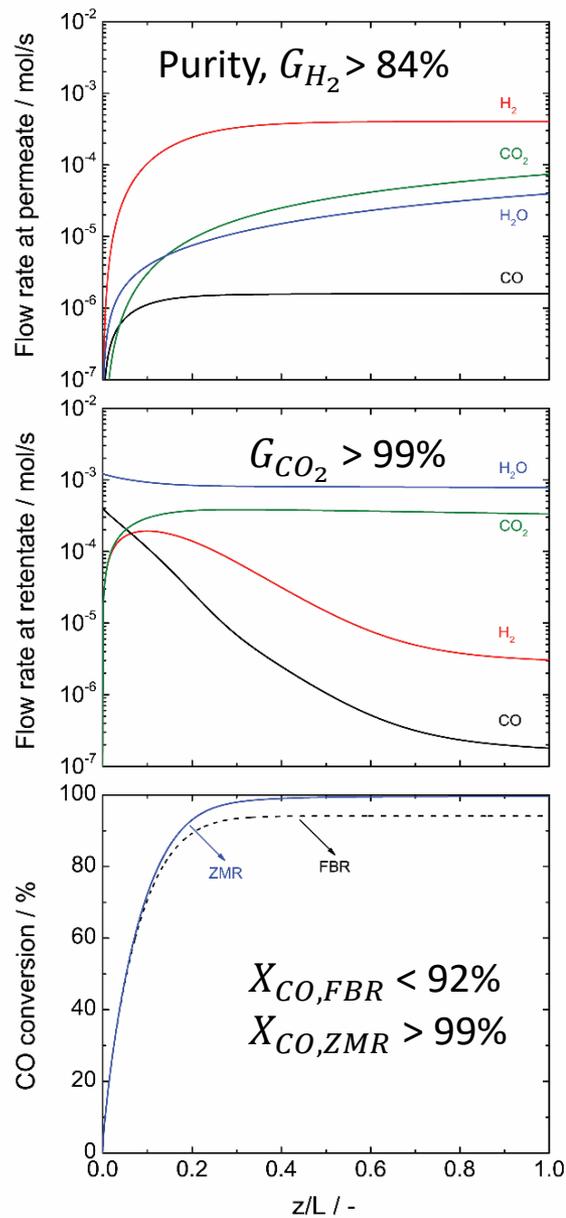
CO₂ purity:

$$G_{\text{CO}_2} = \frac{F_{\text{CO}_2, \text{reten}}}{F_{\text{total, reten}}} > 95\%$$

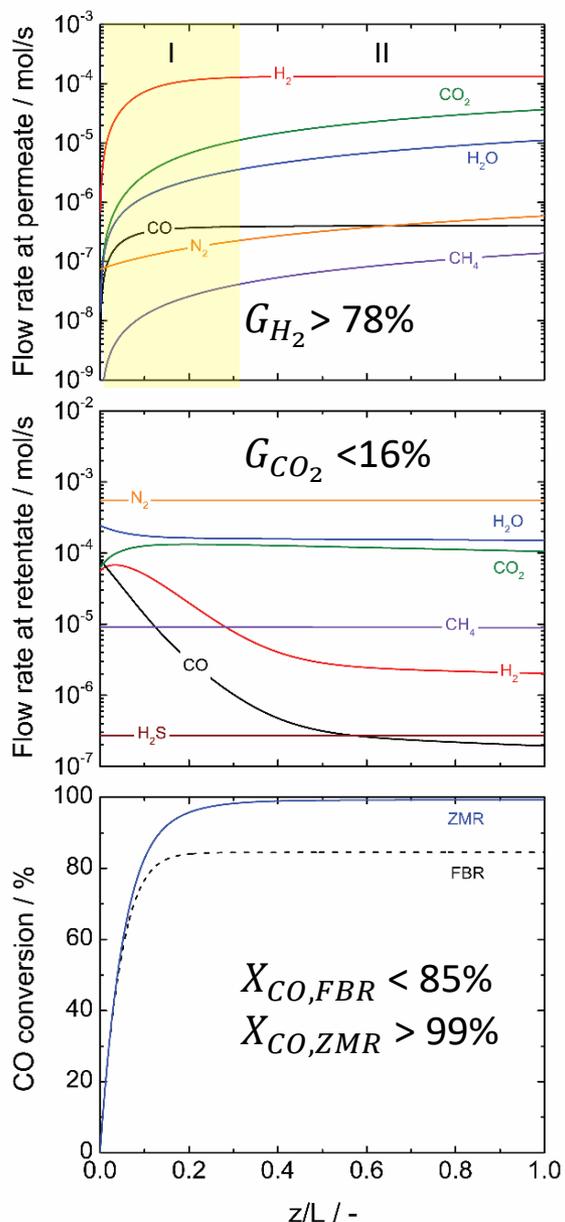
CO conversion:

$$X_{\text{CO}} = \frac{F_{\text{CO, feed}} - F_{\text{CO, reten}} - F_{\text{CO, perm}}}{F_{\text{CO, feed}}} > 99\%$$

(1) Simulated gas



(2) Air-blown gasifier syngas



(3) O₂-blown gasifier syngas

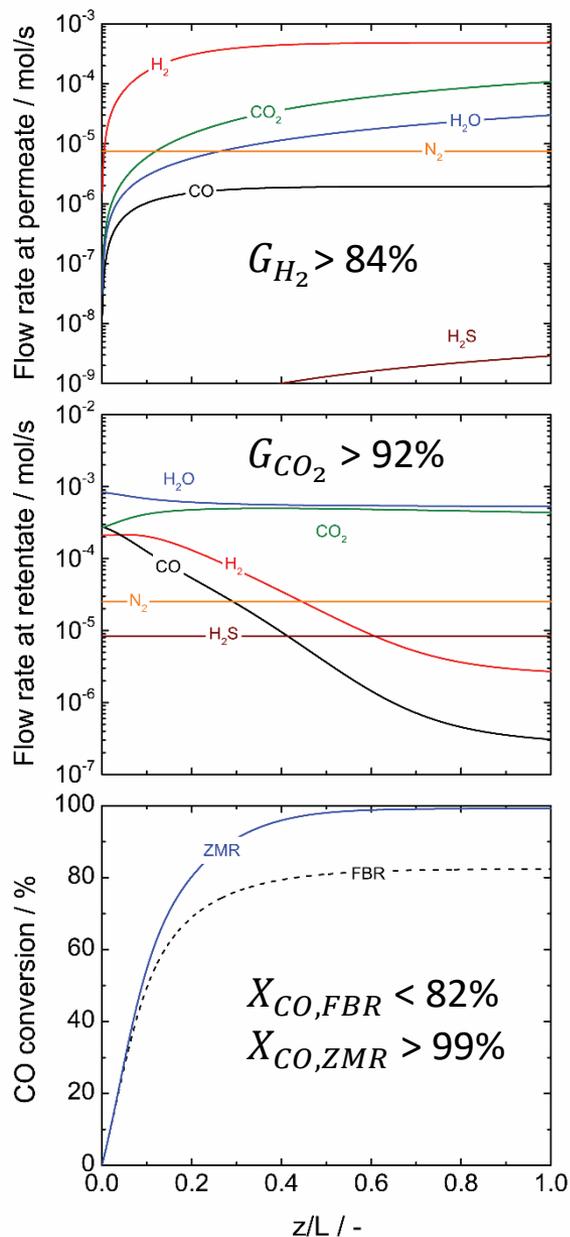
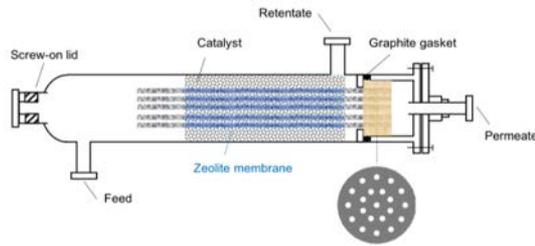
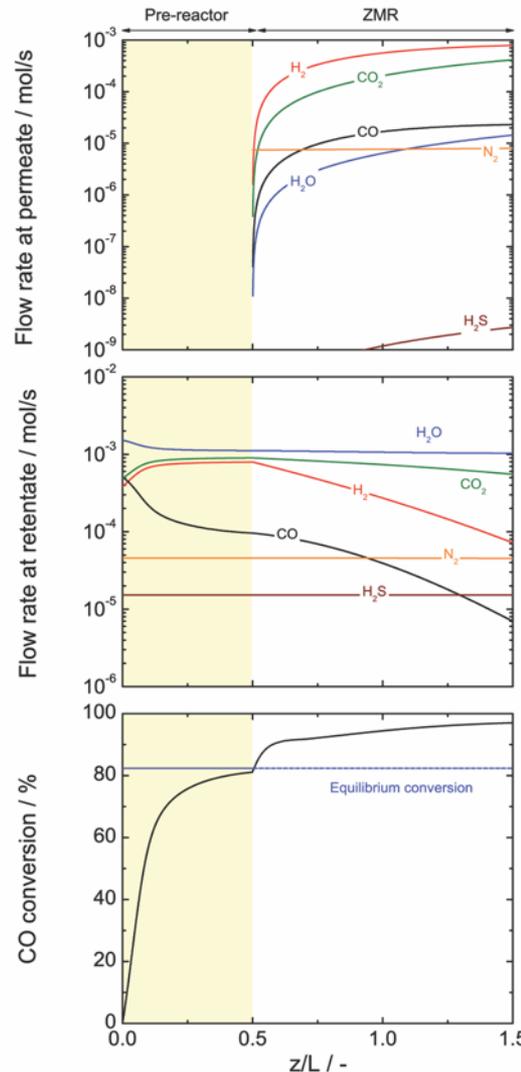
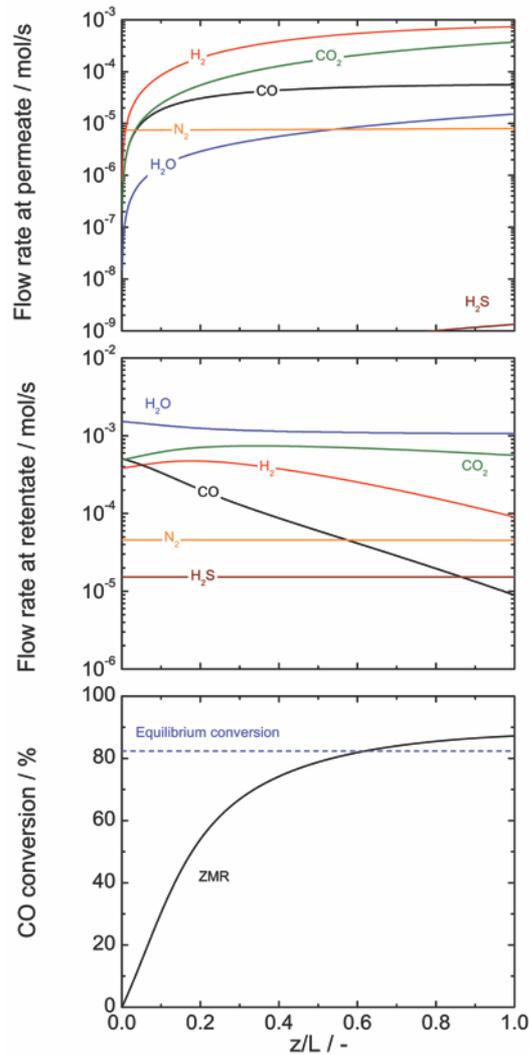
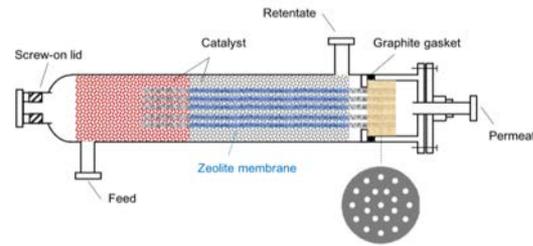


Figure 1. Profile of CO conversion and gas flow rates in zeolite membrane reactor and fixed-bed reactor.

Zeolite Membrane Reactor



Pre-reactor Assisted Zeolite Membrane Reactor



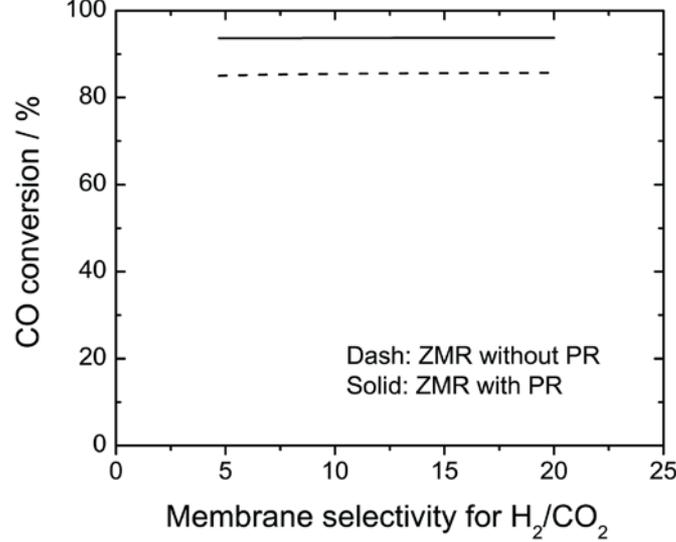
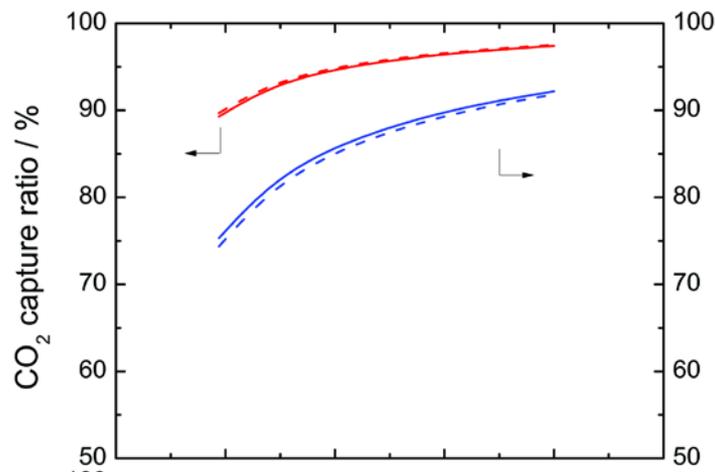
- CO conversion: 87.2% → 97.0%.
- Pre-reactor effect: equilibrium conversion a higher H₂ partial pressure for but a lower CO partial pressure.
- Catalyst packing UC: 30% Simulation: 50%

Figure 2. Profile of CO conversion and gas flow rates in a zeolite membrane reactor with/without a pre-reactor at 450°C. Pressure: 30 bar; Steam/CO mole ratio: 3.0.

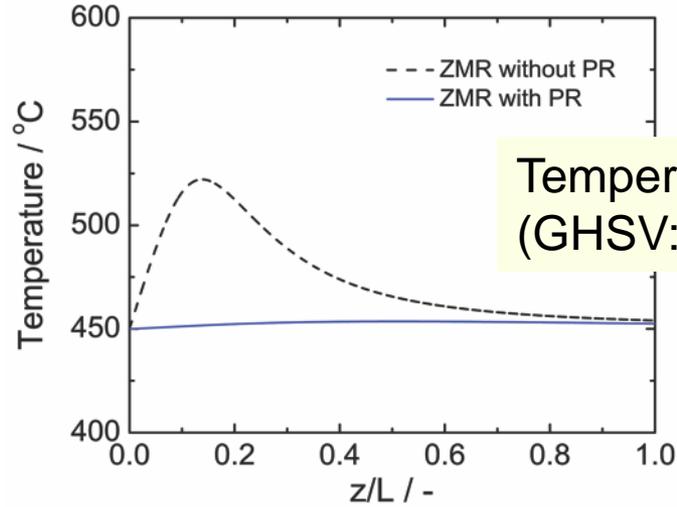
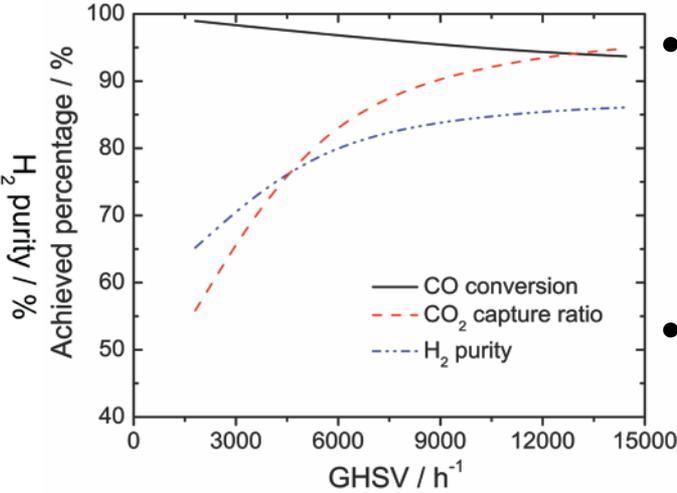
Task 10: Modeling and Analysis of WGS in Bench-Scale Zeolite Membrane Modules

Performance improvement with pre-reactor (PR)

Effect of H_2/CO_2 selectivity
(GHSV: 14446.5 h^{-1})



Effect of GHSV
(H_2/CO_2 Selectivity: 10)



Temperature profile
(GHSV: 5417.5 h^{-1})

- PR enhances the CO conversion but less enhancement in the CO_2 capture and H_2 purity.
- Increasing GHSVs improves the CO_2 capture ratio significantly .

Inlet temperature: 450°C ; Pressure: 30 bar; Steam/CO mole ratio: 3.0.

Task 11: Fabrication of a Large Quantity of Tubular Supports

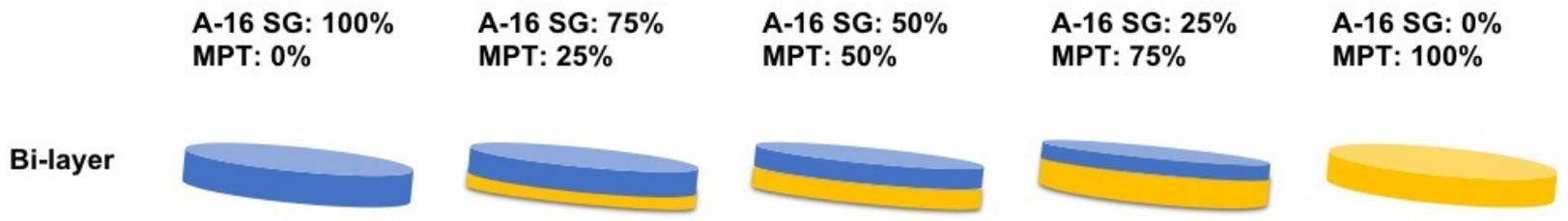
Tubular Substrate Supply

Over 250 tubular substrates of various types and formulations have been prepared

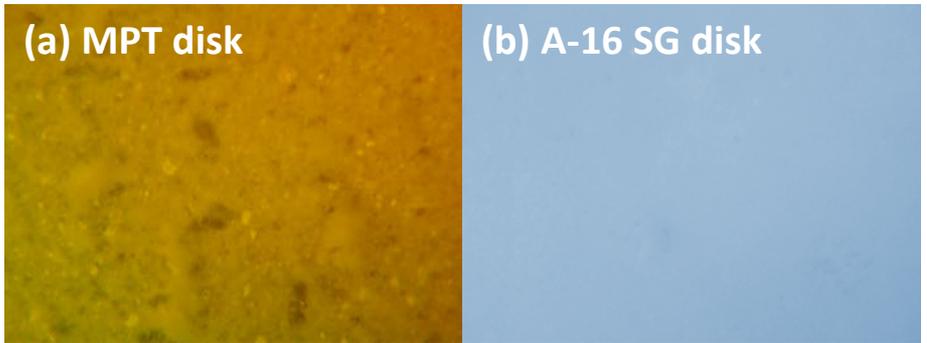
UC-#	Batch-#	Date shipped	Items	Qty.	Description
B5	MPT-4	8/29/2017	1	10	96% Alumina body, 5.7x3.5mm(ODxID), 35cm, outside coated 0.05 µm, 5cm glass glaze on both open ends
B6	MPT-5	9/6/2017	1	4	99.9% Alumina body, 5.7x3.5mm(ODxID), 35cm, outside coated 0.05 µm, 5cm glass glaze on both open ends (These are test pieces for 99.9% Alumina body.)
			2	4	96% Alumina body, 5.7x1.9mm(ODxID), 35cm, outside coated 0.05 µm, 5cm glass glaze on both open ends (These are thick-wall substrate.)
			3	3	96% Alumina body, 5.7x3.5mm(ODxID), 35cm, outside coated 0.05 µm, 5cm glass glaze on the sealed end, 25cm glass on the open end (For 99.9% Alumina body.)
			4	9	99.9% Alumina body, 5.7x3.5mm(ODxID), 10cm, outside coated 0.05 µm, NO glass end seal (These are test pieces for 99.9% Aluminum body.)
			5	6	96% Alumina body, 5.7x1.9mm(ODxID), 10cm, outside coated 0.05 µm, NO glass end seal (These are thick-wall substrate.)
B7	MPT-6	9/21/2017	1	11	96% Alumina body, 5.7x1.9mm(ODxID), 10cm, outside coated 0.05 µm, NO glass end seal (thick-wall substrate)
			2	12	96% Alumina body, 5.7x1.9mm(ODxID), 35cm, outside coated 0.05 µm, 5cm glass glaze on both open ends (thick-wall substrate)
			3	10	96% Alumina body, 5.7x2.9mm(ODxID), 35cm, outside coated 0.05 µm, 5cm glass glaze on both open ends (thick-wall substrate)
B8	MPT-7	10/17/2017	1	38	96% Alumina body, 5.7x2.9mm(ODxID), 10cm, outside coated 0.05 µm, NO glass end seal
			2	10	96% Alumina body, 5.7x2.9mm(ODxID), 35cm, outside coated 0.05 µm 5cm glass glaze on both open ends
			3	9	96% Alumina body, 5.7x2.9mm(ODxID), 35cm, outside coated 0.05 µm 5cm glass glaze on the open and tipped end
B9	MPT-8	12/7/2017	1	21	96% Alumina body, 5.7x2.9mm(ODxID), 35cm, outside coated 0.05 µm (25cm long), 5cm glass glaze on tipped end, 5cm glass glaze on open end
			2	5	96% Alumina body, 5.7x3.5mm(ODxID), 35cm, outside coated 0.05 µm (5cm long), 5cm glass glaze on tipped end, 25cm glass glaze on open end
			3	5	96% Alumina body, 5.7x3.5mm(ODxID), 35cm, outside coated 0.05 µm (10cm long), 5cm glass glaze on tipped end, 20cm glass glaze on open end
B10	MPT-9	1/9/2018	1	18	96% Alumina body, 5.7x2.9mm(ODxID), 35cm, outside coated 0.05 µm (25cm long), 5cm glass glaze on tipped end, 5cm glass glaze on open end
B11	MPT-10	3/8/2018	1	4	99-3070, 5.7x3.5mm (ODxID), 3x10in, 1x7in, no top layer/glass glaze
			2	1	99-115, 5.7x3.5mm (ODxID), 1x5in, no top layer/glass glaze
			3	1	99-114, 5.7x3.5mm (ODxID), 1x5in, no top layer/glass glaze
			4	23	96%, 5.7x2.9mm (ODxID), 35cm, 5cm glass glaze on the tipped & open ends
B12	MPT-11	3/22/2018	1	2	99-3070, 5.7x3.5mm (ODxID), 35cm long, no glass end seal
B13	MPT-12	5/17/2018	1	3	99-3070, 5.7x3.5mm (ODxID), 2x6.5in, 1x7.5in, no top layer/glass glaze, fired at 1650°C
B14	MPT-13	6/14/2018	1	5	96%-body, 5.7x2.9mm (ODxID), 35cm long, outside coated 0.05 µm, 5cm glass glaze on the open & tipped ends
		UC	2	1	96%-body, 5.7x2.9mm(ODxID), 35cm long, outside coated 0.05 µm (10cm), 5cm glass glaze on the tipped end, 20cm glass on the open end
			3	2	96%-body, 5.7x3.5mm(ODxID), 35cm long, outside coated 0.05 µm (10cm), 5cm glass glaze on the tipped end, 20cm glass on the open end
			4	1	96%-body, 5.7x2.9mm(ODxID), 35cm long, outside coated 0.05 µm (5cm), 5cm glass glaze on the tipped end, 25cm glass on the open end
			5	2	96%-body, 5.7x3.5mm(ODxID), 35cm long, outside coated 0.05 µm (5cm), 5cm glass glaze on the tipped end, 25cm glass on the open end
			6	15	96%-body, 5.7x3.5mm(ODxID), 35cm long, outside coated 0.05 µm (25cm), 5cm glass glaze on two open ends
			7	200 grams	MPT-114-99% powders, for discs
		6/14/2018 ASU	1	200 grams	MPT-114-99% powders, for discs (Shipped to Lie Meng, ASU)
B15	MPT-14	7/2/2018	1	23	96%-body, 5.7x3.5mm (ODxID), 35cm long, outside coated 0.05 µm, 5cm glass glaze on the open & tipped ends
			Total =	258	

Task 12: Preparation of Large Quantity Zeolite Membranes for Bench-Scale Module

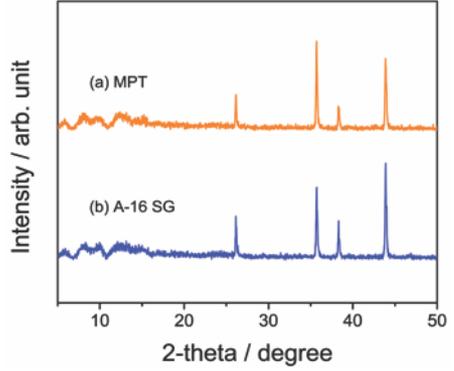
Alumina leaching in the synthesis of MFI zeolite membranes



microscope digital images



XRD patterns



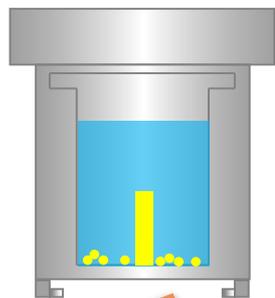
Task 12: Preparation of Large Quantity Zeolite Membranes for Bench-Scale Module

Alumina leaching in the synthesis of MFI zeolite membranes

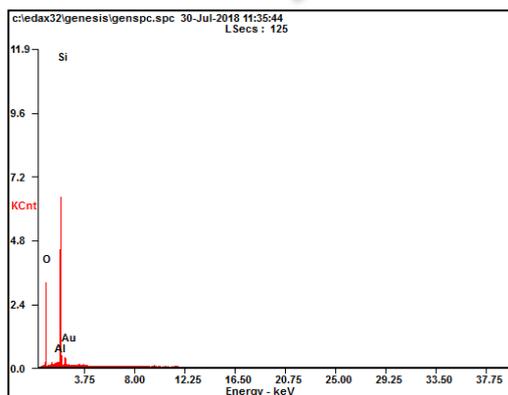
in-situ crystallization at 180°C,

3 hours using 0.055 SiO₂: 0.0058

NaOH: 0.017 TPAOH: 0.92 H₂O



EDAX



High Al content was found in the MFI zeolite formed in the synthesis of zeolite membranes supported by disks pressed with MPT powder.

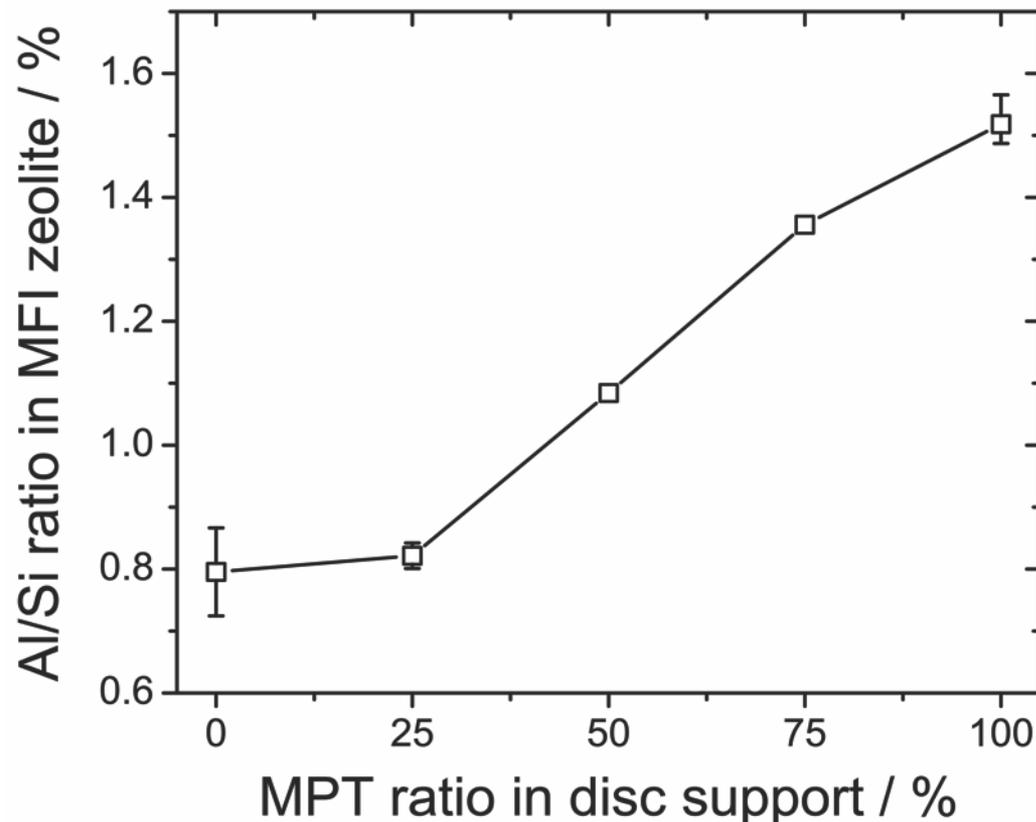
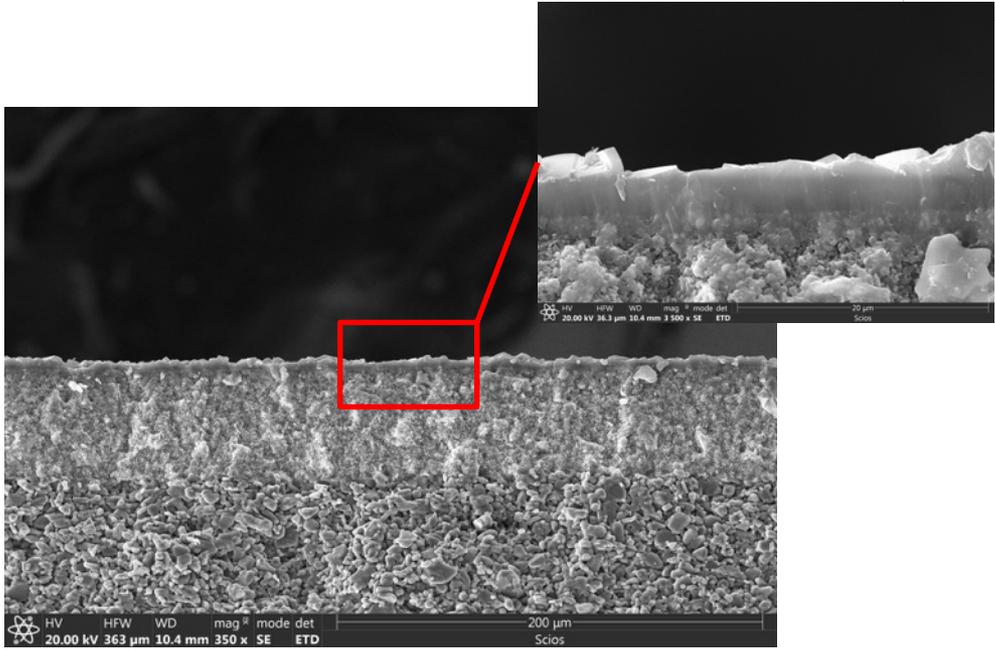
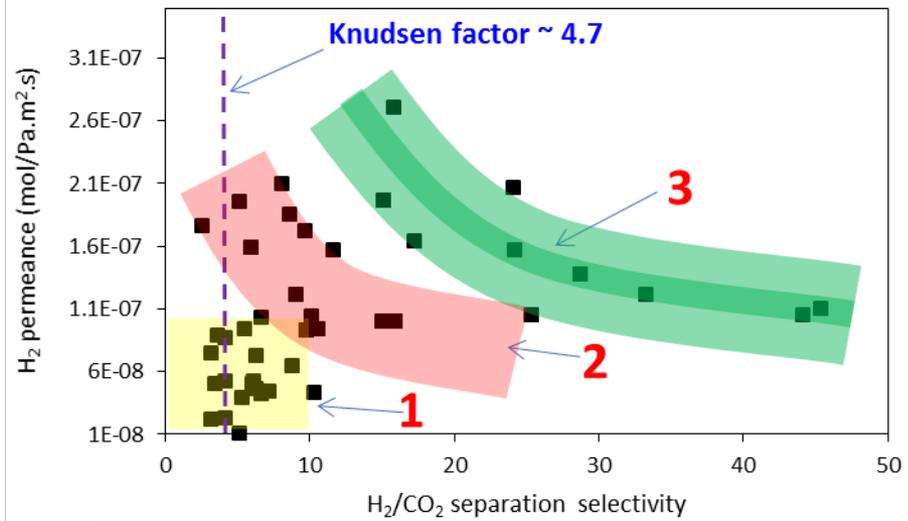
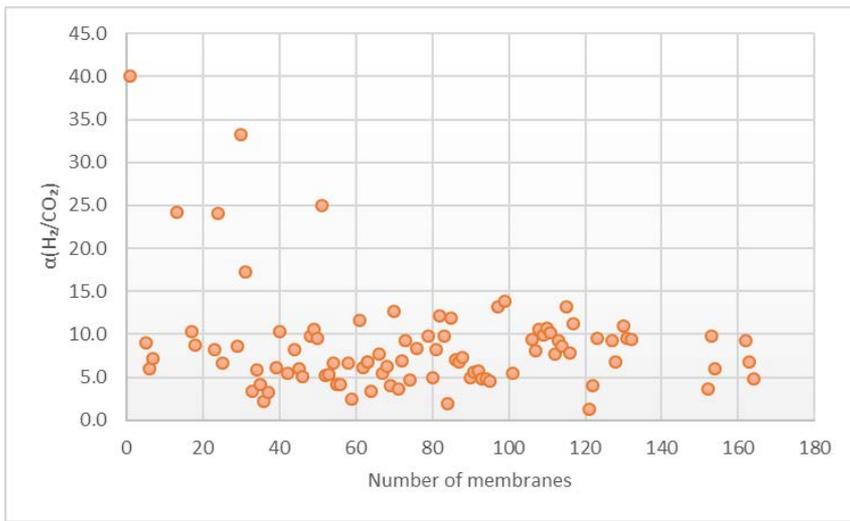


Figure 4. Al/Si ratio in powders collected at the bottom of autoclaves for in-situ synthesis of MFI zeolite membranes supported by varied disks.

Task 12: Preparation of Large Quantity Zeolite Membranes for Bench-Scale Module

Quality of modified tubular MFI zeolite membranes



Crystal size
600 nm

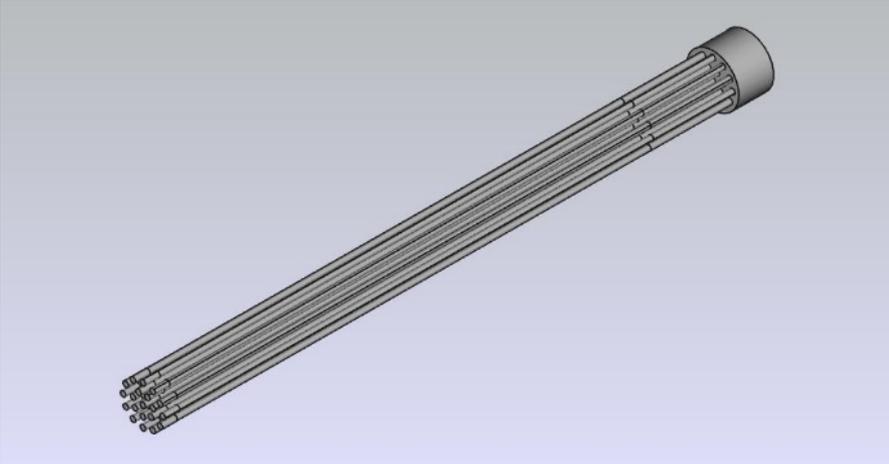
Layer thickness
5-8 μ m

Excellent
surface
coverage

75%
membrane
reproducibility

Task 13: Design and Fabrication of Bench-Scale Zeolite Membrane Module Housing

On-going work: design of bench-scale membrane module



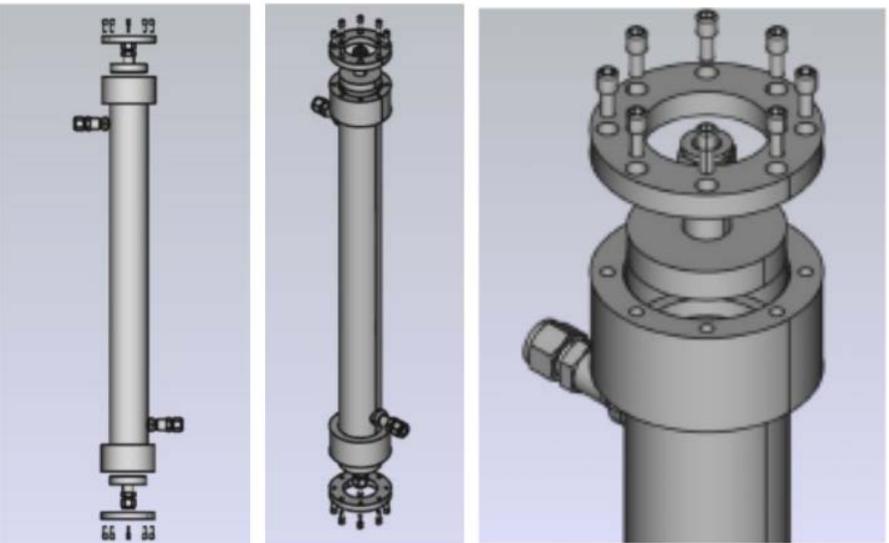
21-Tubule Membrane Bundles

- ✓ Bench scale testing
- ✓ H₂ production rate > 30 L/min
- ✓ Mesh number > 1,000,000

Zeolite Membrane Reactor

Pending to be optimized

- Module configuration
- Catalyst loading
- Operation conditions



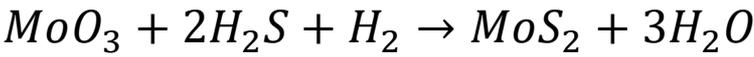
Module material: SS316

Task 14: Building Bench-Scale Zeolite Membrane Reactors

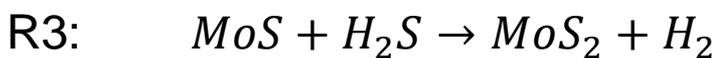
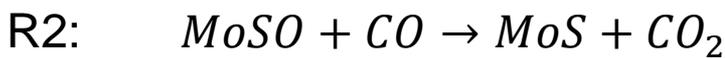
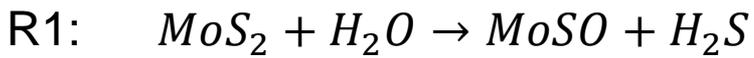
Fabrication and evaluation of WGS catalyst for bench-scale WGS reaction

Kinetic model for the Co-Mo catalyst in the high-temperature WGS

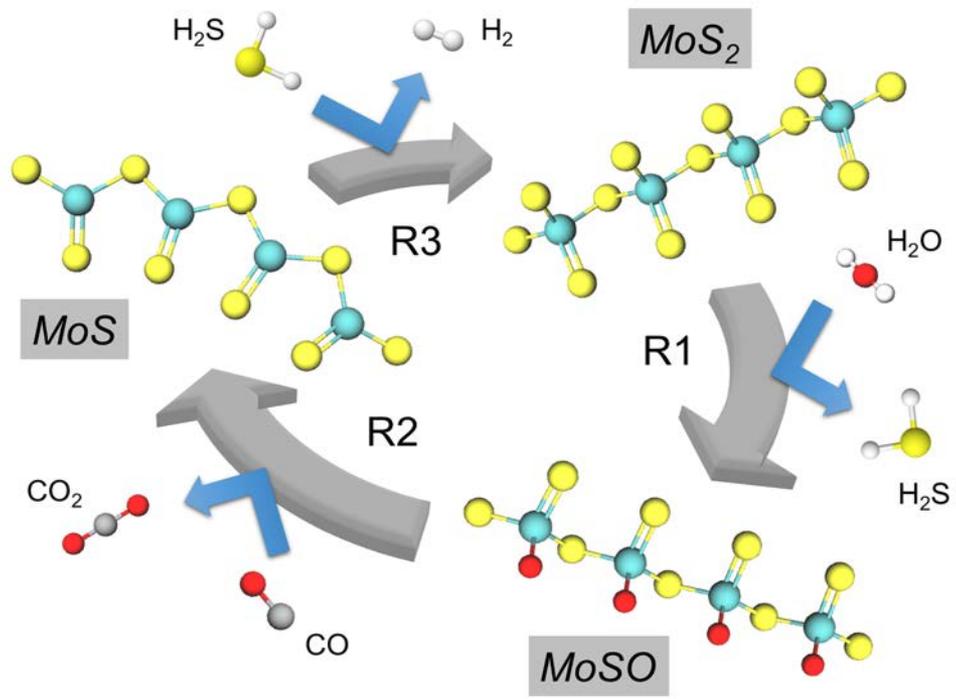
Sulfidation (pre-treatment):



WGS:



Reaction pathway of WGS over Co-Mo catalysts



The power-law model

$$r_{CO} = k_0 \exp\left(\frac{-E_a}{RT}\right) P_{CO}^a P_{H_2O}^b P_{CO_2}^c P_{H_2}^d (1 - \beta)$$

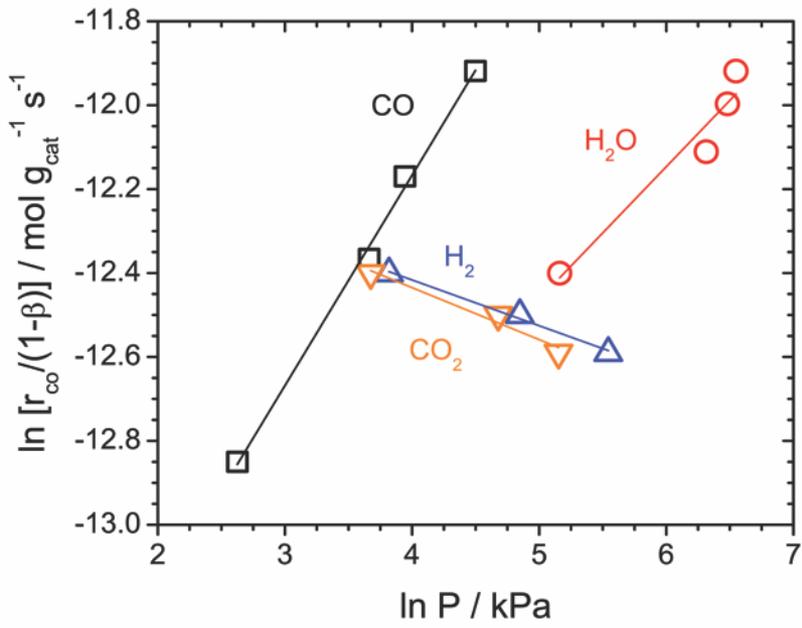
$$\beta = \frac{1}{K} \frac{P_{CO_2} P_{H_2}}{P_{CO} P_{H_2O}}$$

Task 14: Building Bench-Scale Zeolite Membrane Reactors

Fabrication and evaluation of WGS catalyst for bench-scale WGS reaction

Determination of power-law model reaction orders

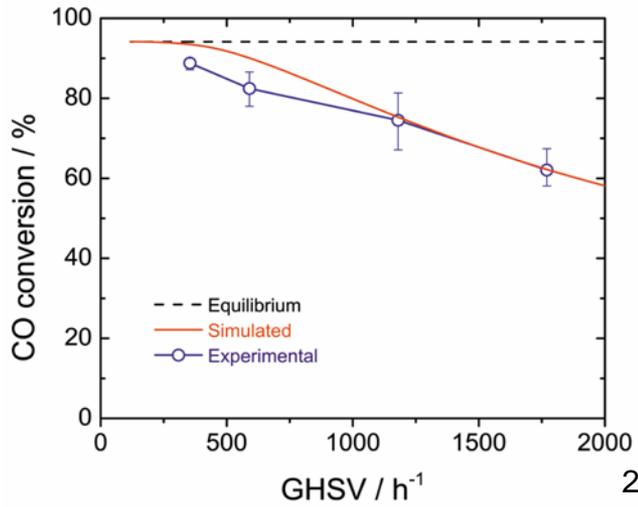
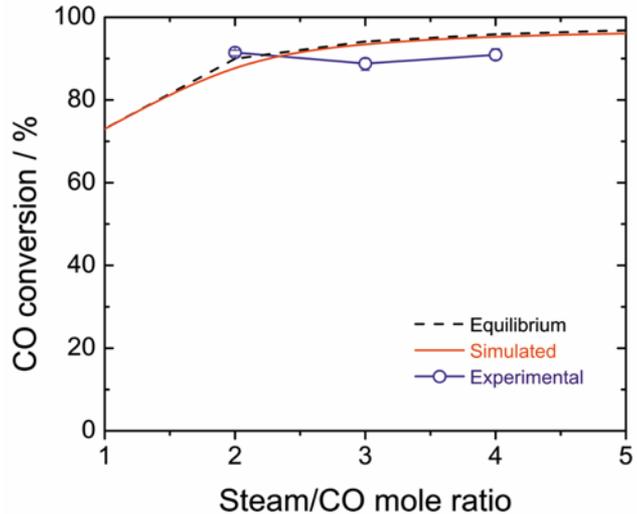
Log-log plots for the effect of CO, H₂O, CO₂, and H₂ partial pressure on reaction rates over Co-Mo catalyst.



$$r_{CO} = 1.58 \times 10^{-4} \exp\left(\frac{-37.99}{RT}\right) P_{CO}^{0.50} P_{H_2O}^{0.32} P_{CO_2}^{-0.12} P_{H_2}^{-0.11} \left(1 - \frac{1}{K} \frac{P_{CO_2} P_{H_2}}{P_{CO} P_{H_2O}}\right)$$



CO conversion in a WGS-FBR (450°C. Pressure: 10 bar)



Task 14: Building Bench-Scale Zeolite Membrane Reactors

Assembling and Testing Bench Scale Zeolite Membrane Reactor

Thermal Stability Testing of Mock 21-tube Membrane Bundle

Testing Conditions: $T = 450^{\circ}\text{C}$; $P = 300$ to 350 psig

21-Tube Candle Filter Bundle

Ceramic Tube Sheet and Ceramic/Glass Potting

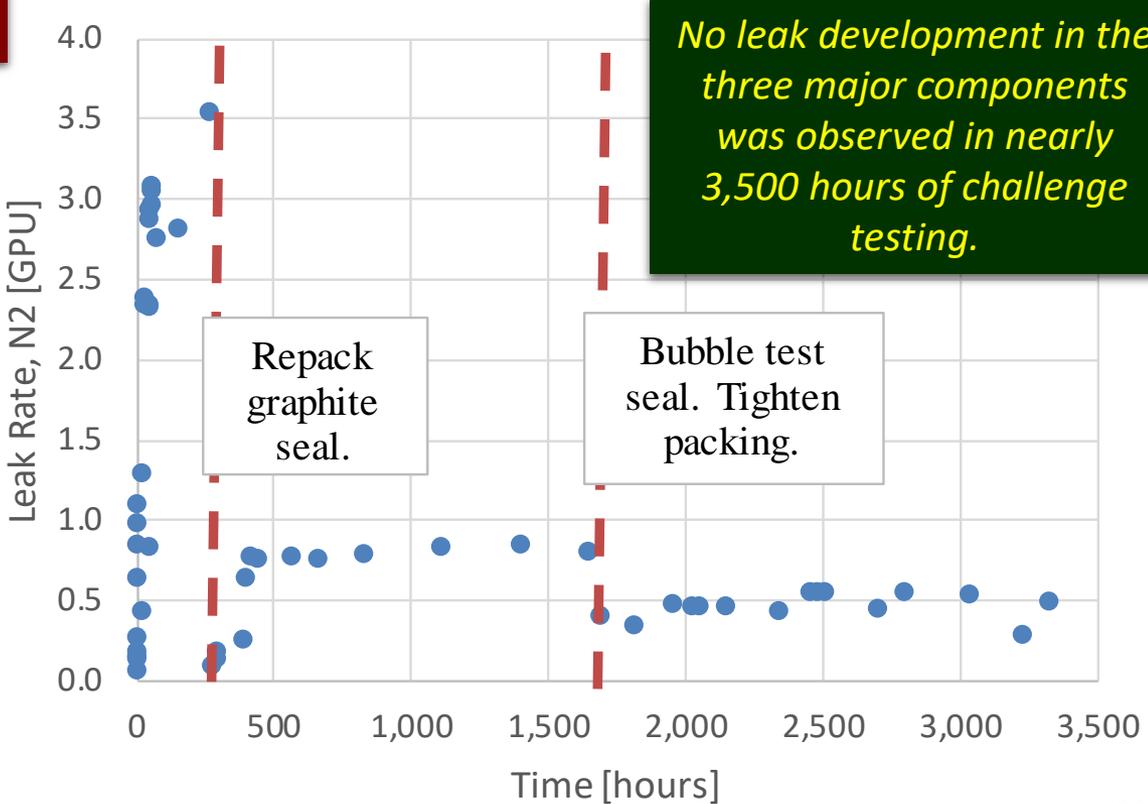
Impermeable Ceramic Tip



High Temperature Test Housing



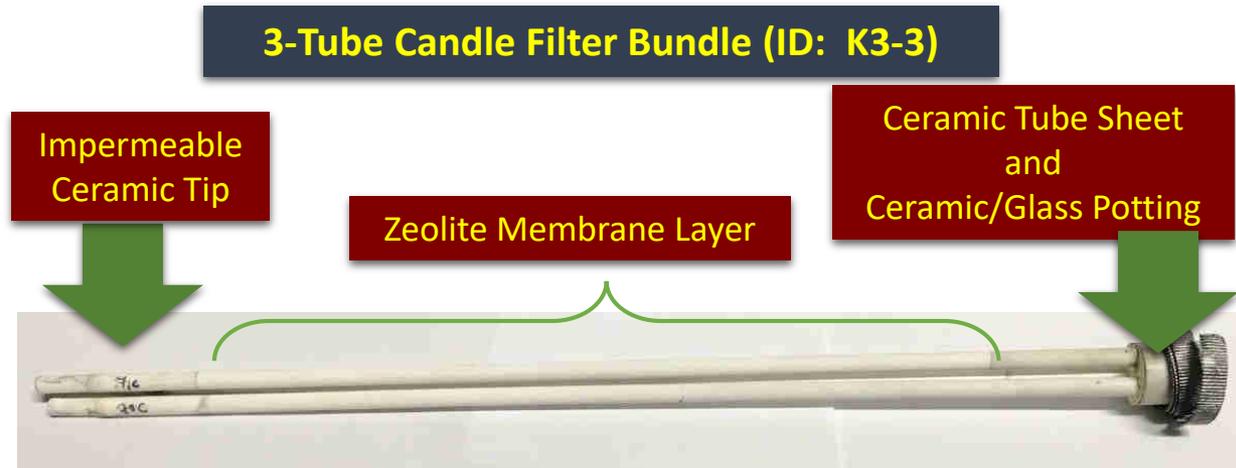
Bundle Leak Rate during Challenge Testing



Task 14: Building Bench-Scale Zeolite Membrane Reactors

Assembling and Testing Bench Scale Zeolite Membrane Reactor

Fabrication and Performance Testing of a 3-tube Zeolite Membrane Bundle



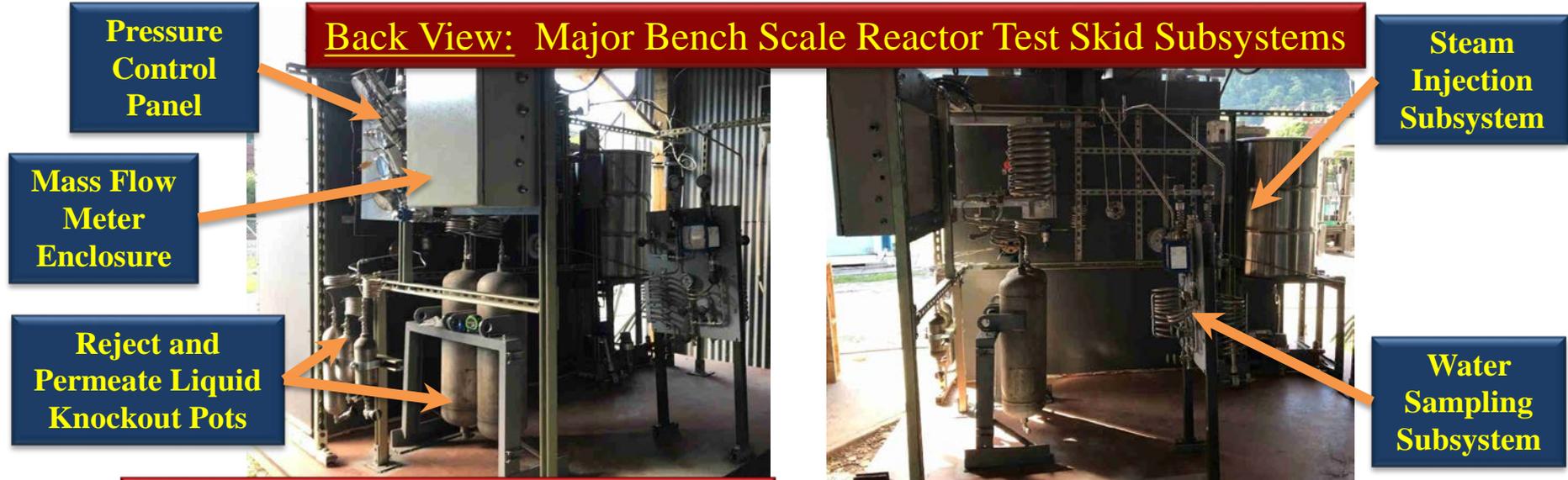
Single-gas permeation at 200°C

	Gas permeance (m ³ /m ² /hr/bar)	
	He	N ₂
Tube 70B	1.449	0.811
Tube 70C	1.336	0.711
Tube 71C	1.422	0.742
Tube average	1.400	0.755
3-tube bundle	1.310	0.707

Task 14: Building Bench-Scale Zeolite Membrane Reactors

Modification and Installation of Bench-Scale Reactor Test Skid

Back View: Major Bench Scale Reactor Test Skid Subsystems



Front View: Oven Chamber with Membrane Test Module and a 3" x 85-tube Membrane Bundle for Perspective



Power Distribution Skid



Conclusions

- Mathematical models of WGS-ZMR and Reaction kinetics of WGS catalysts established.
- 25-cm long CCD-modified zeolite membranes scaled up on alumina substrates.
- Multiple-tube ZMRs assembled and evaluated, and test skid for bench-scale test modified.

Acknowledgement

- Financial support: U.S. Department of Energy and National Energy Technology Laboratory (Federal Grant: DE-FE0026435).



Ashim D'Silva / Unsplash.com
Lower Antelope Canyon, Arizona

Thank You!