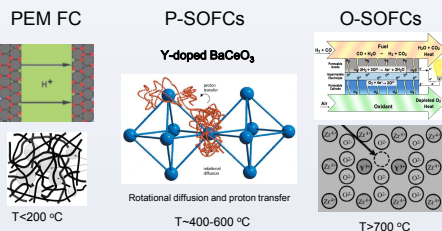


# Highly Active Hybrid Catalyst Impregnated Cathode for Proton Conducting Solid Oxide Fuel Cells

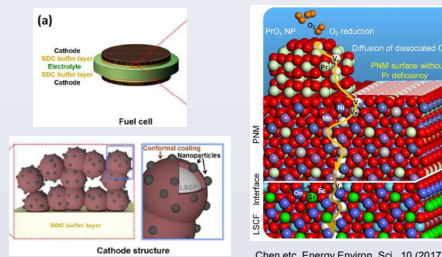
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## Introduction



ORR can be enhanced by adding a nanostructured catalyst on the electrode via infiltration/impregnation technique.

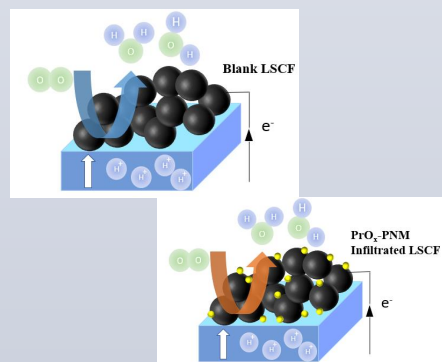


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## Research Objectives

The sluggish reaction kinetics in the cathode usually hinders the development of H-SOFCs operated at intermediate temperatures (400-650°C).

To improve the reaction kinetics of cathode, a novel hybrid catalyst consisting of  $\text{PrNi}_{0.5}\text{Mn}_{0.5}\text{O}_3$  and  $\text{PrO}_x$  is impregnated in the  $(\text{La}_{0.60}\text{Sr}_{0.40})_{0.95}\text{Co}_{0.20}\text{Fe}_{0.80}\text{O}_{3-\delta}$  (LSCF) cathode of H-SOFCs.



## Result and Discussion

### Part I Phase and Microstructure Characterization

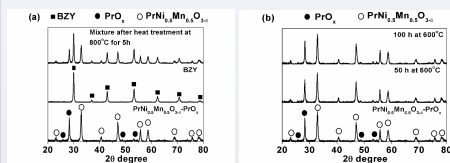


Figure 1 XRD patterns (a) chemical compatibility between  $\text{PrNi}_{0.5}\text{Mn}_{0.5}\text{O}_3$ ,  $\text{PrO}_x$ , and BZY; (b) Stability of  $\text{PrNi}_{0.5}\text{Mn}_{0.5}\text{O}_3$  and  $\text{PrO}_x$ .

Hybrid catalyst is chemically compatible with  $\text{BaZr}_{1-x}\text{Y}_x\text{O}_3$  (BZY) and shows good phase stability.

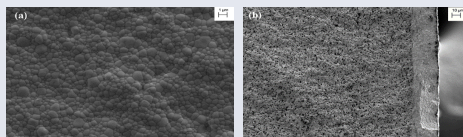


Figure 2 FESEM images of the single cells. (a) The surface of the BZY electrolyte; (b) The cross-section view of NiO-BZY anode and BZY electrolyte.

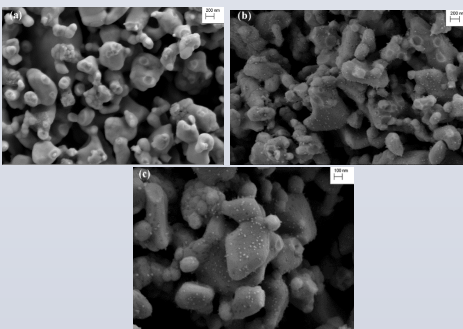


Figure 3 FESEM image of cathode (a) Microstructure of bare LSCF electrode; (b) Microstructure of LSCF electrode with  $\text{PrNi}_{0.5}\text{Mn}_{0.5}\text{O}_3$  and  $\text{PrO}_x$ ; (c) High magnification of image of LSCF electrode with  $\text{PrNi}_{0.5}\text{Mn}_{0.5}\text{O}_3$  and  $\text{PrO}_x$ .

H-SOFCs with hybrid catalyst impregnated LSCF cathode have been successfully fabricated.

### Part II Electrochemical Characterization

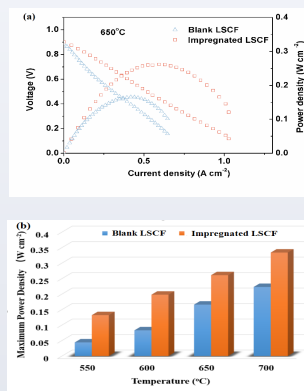


Figure 4 Electrochemical performance of single cells (a) I-V curves; (b) Maximum power densities of single cells at various temperature.

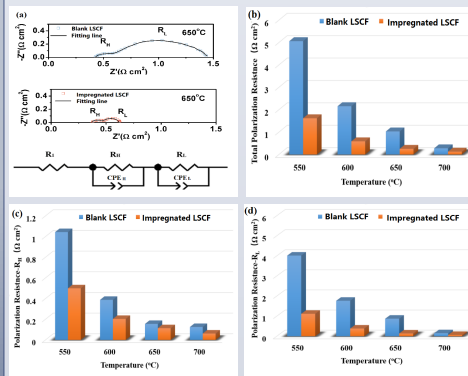


Figure 5 (a) Impedance spectra of single cells at 650 °C; (b) Polarization resistance of single cells at various temperatures; (c) Polarization resistance  $R_H$  at various temperatures; (d) Polarization resistance  $R_L$  at various temperatures.

Hybrid catalyst can greatly enhance the reaction kinetics of cathode, resulting in significant improvement of electrochemical performance of H-SOFCs, especially at lower temperatures.

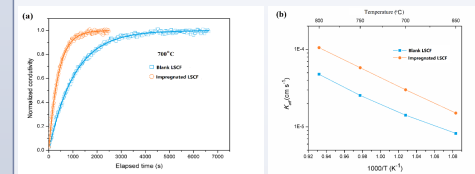


Figure 6 (a) Electrical conductivity relaxation as a function of time at 700°C; (b) Plots of the surface exchange coefficients ( $K_{eff}$ ) as function of temperatures.

Impregnation of hybrid catalyst can improve not only the kinetics of oxygen-ion transfer but also oxygen dissociation-absorption process (oxygen surface exchange) in the LSCF cathode, resulting in significant reduction of  $R_p$ ,  $R_H$  and  $R_L$ .

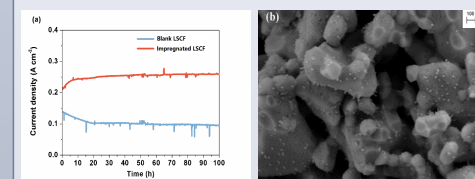


Figure 7 Stability of single cells at 600 °C (a) Operation of single cells at 0.6 V; (b) The microstructure of electrode after 100 h test.

## Conclusion

Impregnation of hybrid catalyst is a promising approach to improve the performance and durability of LSCF cathode for H-SOFCs.

## Acknowledgement

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