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Development and Demonstration of Waste Heat Integration with Solvent Process for More Efficient CO<sub>2</sub> Removal from Coal-Fired Flue Gas

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## **Project Participants**



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### Project Objectives and Overview

### **Team-Member Roles**



Southern Company

### Project Management Funding Host Site

## AECOM

Reporting Detailed Design Flue Gas Measurement



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### Technology Provider Reporting



Project Management Funding

# In this project, advanced heat integration was demonstrated on a coal EGU

- The heat integration was chosen for its ability to provide:
- Increased plant efficiency,
- Mitigation of parasitic losses from a CO<sub>2</sub> capture system (CCS),

- Reduced water consumption and cooling water use, and
- Improvement in air quality system performance
- The heat integration included heat recovery for use in the coal EGU Rankine cycle. The heat was sourced from:
- A pilot CO<sub>2</sub> capture facility and
- The coal EGU flue gas.

## Project objectives were chosen to quantify effects of heat integration on the coal EGU



## Heat integration system transfers heat into boiler using two heat exchangers

Together, the two heat exchangers and associated balance of plant are known as the High Efficiency System (HES):

- <u>CO<sub>2</sub> Cooler</u>: Recovers heat from the outlet of the stripper in the CO<sub>2</sub> capture facility.
- <u>Flue Gas Cooler</u>: Recovers heat from the coal EGU flue gas downstream of the plant air heater.

A standard heat exchanger can be used for the  $CO_2$  Cooler but the Flue Gas Cooler is based on a heat exchanger used to recover heat from flue gas in Japan.

# The Flue Gas Cooler is based on a similar process used in Japan

#### Hirono P/S Japan - 600MW





→ Water Loop Flue Gas **Plume Abatement** 

## Here, plume abatement may not be desirable; the heat can instead be used to improve heat rate



(a) Application for Europe and Japan



# Flue Gas Cooler condenses SO<sub>3</sub> onto the fly ash

- Operates downstream of the APH
- Mechanism for removal of SO<sub>3</sub> from flue gas
  - $-SO_{3}(g) + H_{2}O(g) --> H_{2}SO_{4}(g)$
  - $H_2SO_4 (g) --> H_2SO_4 (I)$
  - $H_2SO_4$  (I) condenses on fly ash in flue gas and a protective layer of ash on tube bundles

- Flue Gas Cooler tube skin temperature < SO<sub>3</sub> dewpoint
  - Alkaline species in fly ash (Ca, Na) neutralize H<sub>2</sub>SO<sub>4</sub>
  - Silicates, etc. physically adsorb H<sub>2</sub>SO<sub>4</sub>

# Corrosion in the Flue Gas Cooler can be mitigated by fly ash in the flue gas

Carbon steel tubes in good condition after 2 years of operation in Japan.





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Ratio of fly ash or dust in the flue gas can be used to determine corrosion rate.

# The Flue Gas Cooler can also provide environmental benefits including:

• Reduced water consumption in the FGD and cooling water use in the CCS facility due to reduction in gas temperature;

- Better  $SO_3$  capture through condensation of the  $SO_3$  on the fly ash;
- Better particulate control device performance through both reduced gas volume and lower ash resistivity due to reduced temperature and moisture adsorption to fly ash; and
- Increased capture of Hazardous Air Pollutants (mercury, other toxic metals, etc.) due to reduced flue gas temperatures and SO<sub>3</sub> concentrations as well as improved particulate capture performance.

# In the HES, heat is also recovered from the $CO_2$ capture system



## In the HES, boiler condensate is first heated in the CO<sub>2</sub> Cooler and then the Flue Gas Cooler



## Heat integration can eliminate the need for low pressure heaters



- Pilot included a CO<sub>2</sub> capture system to provide heat for the CO<sub>2</sub> Cooler
- Pilot used a slipstream of plant flue gas to provide heat for the Flue Gas Cooler

- Various measurements were taken in the flue gas and boiler condensate around each heat exchanger
- Pilot integrated balance of plant and control equipment with the host site

### Pilot Unit at Plant Barry

## To demonstrate the HES, a 25-MW pilot was built at Plant Barry



### Plant Barry was chosen for the 25-MW CCS plant already in place

Funded by industry consortium

- Fully integrated CO<sub>2</sub> capture/compression
- Storage in Citronelle Dome
- Capacity: 500 metric tons CO<sub>2</sub>/day

### CCS plant at Barry uses Kansai Mitsubishi Carbon Dioxide Recovery Process (KM CDR)®

- KM CDR uses a proprietary solvent, KS-1, as the absorbent media for  $\mathrm{CO}_{\mathrm{2}}$
- Dominant reaction of KS-1 requires a lower molar ratio than that of MEA
- KS-1 has been shown to degrade (via formation of heat stable salts) more slowly than MEA
- KS-1 is more efficient at adsorbing and desorbing CO<sub>2</sub>



## The CO<sub>2</sub> and Flue Gas Coolers were integrated as shown; a mini-ESP was also included



## The CO<sub>2</sub> Cooler was located near the CCS plant; The FGC was located downstream of the air heater



## Both the Flue Gas Cooler and CO<sub>2</sub> Cooler were sized for a 25-MW coal EGU





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#### Flue Gas Cooler

### CO<sub>2</sub> Cooler

## A blower was used to pull the slipstream of flue gas through the Flue Gas Cooler and ESP





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Flue Gas Blower

Pilot ESP (0.25 MW)

## Several deviations from the intended design and operation occurred but are believed to be minor

- Initially the team proposed heating CCS process condensate as well as boiler condensate; a techno-economic study found this to not be advantageous
- In the CO<sub>2</sub> Cooler, steam, rather than product CO<sub>2</sub> was often used to heat the boiler condensate due to scheduling conflicts with operating the CCS plant
- A condition to test the effect of CaBr<sub>2</sub> injection on the Flue Gas Cooler was initially planned but not carried out.
- Only 900 hours of operation were achieved due to scheduling conflicts with the host unit and issues with the flue gas blower

# Erosion of the flue gas blower caused significant delays and limited runtime

- Fly ash caused erosion of critical components of the flue gas blower.
- This equipment was only necessary for the pilot to pull the slipstream of flue gas through the Flue Gas Cooler; it would not be used in a fullscale HES.
- Erosion was stopped by applying a thick coating; however, the coating began to chip off due to thermal expansion/contraction of the coated elements.





### Test Program and Results

## The test program was organized into five tests to satisfy the project objectives

**Performance Test** - Evaluate the CO<sub>2</sub> Cooler and Flue Gas Cooler performances and verify controllability of the temperature control valves.

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**Turndown Load Operation Test** - Evaluate the Flue Gas Cooler performance under reduced flue gas flow conditions.

**Impurities Removal Test** - Evaluate the effect of cooling the flue gas via the Flue Gas Cooler on the pilot ESP performance for particulate matter, sulfur oxides, and trace metals.

**Long-Term Durability Test** - Evaluate the system data and physical condition of the Flue Gas Cooler, such as vibration and mechanical damage

Material Evaluation Test – Evaluate any corrosion, erosion or boiler condensate leakage in the Flue Gas Cooler.

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**Purpose:** Evaluate the  $CO_2$  Cooler and Flue Gas Cooler performances and verify controllability of the temperature control valves.

### Items evaluated:

- Heat recovery performance of CO<sub>2</sub> Cooler and Flue Gas Cooler and effect on plant generation (via modeling)
- Flue gas pressure drop across the Flue Gas Cooler
- Water consumption reduction for the existing FGD system and cooling water use reduction in the CO<sub>2</sub> capture system via calculations

## Several heat recovery modes were investigated in the Performance Test

Recovered heat was calculated by measuring the boiler condensate temperatures at the inlet and outlet of the CO<sub>2</sub> Cooler and Flue Gas Cooler.

| Test Condition   | R1-1   | R1-2      | R1-3      | R1-4      | R1-5    |
|--|--------|-----------|-----------|-----------|---------|
| CO2 Cooler heat recovery mode                          | Normal | Normal    | Increased | Increased | Reduced |
| Flue Gas Cooler heat recovery mode                     | Normal | Increased | Normal    | Increased | Normal  |
| Flue gas flowrate scfm                                 | 60,000 | 60,000    | 60,000    | 60,000    | 60,000  |
| Flue Gas Cooler flue gas outlet temperature set point  | 203F   | 185F      | 203F      | 185F      | 203F    |
| Flue Gas Cooler heat transfer coefficient (Btu/ft²hrF) | 2.7    | 4.3       | 2.4       | 3.9       | 2.1     |
| Total Heat Recovered (MMBtu/hr)                        | 11.5   | 13.6      | 11.1      | 13.5      | 10.1    |
| Percentage recovered by the CO <sub>2</sub> Cooler     | 42%    | 32%       | 54%       | 44%       | 32%     |
| Total heat recovered for a 550-MW coal EGU (MMBtu/hr)  | 253    | 300       | 244       | 297       | 222     |

## The DOE Case 10 plant (subcritical PC EGU with CCS) was used as the basis for the model.

|  | Original Case 10<br>Value | Gain or Loss (-) Due<br>to HES |
|--|---------------------------|--------------------------------|
| Total LP feedwater heater and deaerator steam extraction | 421,000 lb/hr             | -366,000 lb/hr                 |
| Turbine generation                                       | 673 MW                    | 18.7 MW                        |
| Cooling fan and water pumps power consumption increase   | -                         | 1.6 MW                         |
| Induced draft fan power consumption                      | 12.1 MW                   | -1.3 MW                        |
| Total Power Gain   | -                         | 18.3 MW                        |
| Plant Thermal Efficiency                                 | 26.2%                     | 0.9% points                    |

## Flue gas pressure drop across the Flue Gas Cooler was monitored throughout testing

Pressure drop across Flue Gas Cooler ranged from 2-4" H<sub>2</sub>0



## The Flue Gas Cooler can reduce evaporative water consumption in the FGD

- By cooling the flue gas, FGD makeup water can be reduced.
- Percentage of water saved was calculated, not measured.
- For coal EGUs with an air heater outlet temperature of 300°F, up to 60% of the FGD makeup water can be saved.

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• For a 550-MW plant, 400 gpm of FGD makeup water would be saved.



# The High Efficiency System can reduce

- For coal EGUs with an air heater outlet temperature of 300°F, up to 36% of the CCS cooling water use can be reduced.
  - For a 550-MW plant, 27,000 gpm of CCS cooling water use would be reduced by the Flue Gas Cooler.
- An additional 20% reduction in cooling water use can be realized by using the CO<sub>2</sub> Cooler to cool the product CO<sub>2</sub>
  - For a 550-MW plant, 18,000 gpm of CCS cooling water use would be reduced by the CO<sub>2</sub> Cooler.

**Purpose:** Evaluate the CO<sub>2</sub> Cooler and Flue Gas Cooler performances at reduced flue gas flowrate.

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### Items evaluated:

- Heat recovery performance of CO<sub>2</sub> Cooler and Flue Gas Cooler
- Flue gas pressure drop across the Flue Gas Cooler
- Water quality of BC at the outlet of the Flue Gas Cooler

## Heat recovery with a reduced flue gas flow was investigated in the Turndown Load Operation Test

Flue gas flowrate was reduced to 70-75% of the design value. Only issue encountered was vibrations from the Flue Gas Blower at low flowrates.

| Test Condition   | R2-1   | R2-2      | R2-3    | R2-4      |
|--|--------|-----------|---------|-----------|
| CO2 Cooler heat recovery mode                          | Normal | Normal    | Reduced | Reduced   |
| Flue Gas Cooler heat recovery mode                     | Normal | Increased | Normal  | Increased |
| Flue gas flowrate scfm                                 | 42,000 | 42,000    | 45,000  | 45,000    |
| Flue Gas Cooler flue gas outlet temperature set point  | 203F   | 185F      | 203F    | 185F      |
| Flue Gas Cooler heat transfer coefficient (Btu/ft²hrF) | 1.8    | 1.3       | 2.9     | 3.2       |
| Total Heat Recovered (MMBtu/hr)                        | 8.1    | 9.1       | 6.2     | 9.0       |
| Total heat recovered for a 550-MW coal EGU (MMBtu/hr)  | 178    | 201       | 136     | 198       |

**Purpose:** Evaluate the effect of cooling the flue gas via the FGC on the pilot ESP performance for particulate matter, sulfur oxides, and trace metals.

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### Items evaluated:

- ESP SO<sub>3</sub>, particulate matter, and trace metals removal performance
- Characteristics of ash collected at the ESP

## Flue gas was sampled for particulate, metals, and sulfur oxides in the Impurities Removal Test



## Test conditions were chosen to evaluate the effect of the Flue Gas Cooler and SO<sub>3</sub> concentration

### Test Conditions included:

- <u>No FGC 300F</u>: No boiler condensate flowed through the FGC, the flue gas was not cooled by the Flue Gas Cooler.
- <u>FGC 203F</u>: The flue gas at the FGC outlet was cooled to 203°F by the Flue Gas Cooler.
- <u>FGC 185F</u>: The flue gas was further cooled down to 185°F by the Flue Gas Cooler.
- <u>FGC 203F + SO<sub>3</sub></u>: The flue gas was cooled to 203°F by the Flue Gas Cooler and SO<sub>3</sub> was injected. Although, no significant increase in flue gas SO<sub>3</sub> was measured due to injection.

## Despite injection of SO<sub>3</sub> into the flue gas, very low concentrations were measured

- For condition FGC 203F + SO3, an 8ppm equivalent of SO<sub>3</sub> was injected into the flue gas via Plant Barry's ESP SO<sub>3</sub> conditioning system.
- Very little SO<sub>3</sub> was measured at either the FGC inlet or ESP outlet.
- The injected SO<sub>3</sub> removed by the alkaline fly ash.
- However, an appreciable effect was measured on mercury removal due to SO<sub>3</sub> injection.

| Condition        | FGC Inlet                                    | ESP outlet |  |  |
|------------------|--|------------|--|--|
|                  | (ppmd SO <sub>3</sub> at 3% O <sub>2</sub> ) |            |  |  |
| NO FGC 300F      | 0.11   | 0.03       |  |  |
| FGC 203F+<br>SO3 | 0.18   | 0.04       |  |  |
| FGC 203F         | 0.17   | 0.04       |  |  |
| FGC 185F         | 0.11   | 0.02       |  |  |

## Impurities removal was enhanced by Flue Gas Cooler operation

- Native mercury removal by fly ash increased significantly from 28 to >86% due to the Flue Gas Cooler
- Selenium removal increased from 96 to 98%
- No discernable effect due to temperature decrease from 203 to 185°F on either metal or particulate matter
- SO<sub>3</sub> removal not calculated due to low concentrations



## SO<sub>3</sub> injection inhibited mercury capture, no effect on selenium or particulate matter

- Mercury removal decreased from >92 to 40%
- Mercury removal still higher during SO<sub>3</sub> injection than without FGC operation
- Selenium removal unchanged
- Particulate matter removal unchanged
- SO<sub>3</sub> removal not calculated due to low concentrations



**Purpose:** Evaluate the system data and physical conditions, including vibration, mechanical damage, etc.

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### Items evaluated:

- Flue Gas Cooler internal surfaces via visual inspection
- Flue Gas Cooler internal equipment such as soot blowers via visual inspection

# The HES was operated for 913 hours for the Long-Term Durability Test

• Flue Gas Cooler internal surfaces were visually inspected before, during and after operation.

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- No mechanical damage to tubes found via visual inspection (see pictures below)
- No damage to soot blowers found via visual inspection
- No ash deposition or accumulation on tube walls



\*The remaining fly ash can be easily removed by soot-blowers.

**Purpose:** Evaluate any corrosion, erosion or plugging in the Flue Gas Cooler.

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Items evaluated:

- Flue Gas Cooler tubes wall loss via corrosion
- Boiler condensate leakage via flowmeters at the inlet and outlet of the Flue Gas Cooler
- Water quality of boiler condensate at the outlet of the Flue Gas Cooler

## Heat transfer tubes from the Flue Gas Cooler were analyzed upon project completion

- Tubes were cut from the FGC and sent to Det Norske Veritas (DNV GL) for analysis. Control samples of tubing not exposed to flue gas used for comparison.
- The fins were removed and scale and deposits scrubbed off.
- Wall loss measurements were taken via a three-dimensional optical microscope.





### General corrosion was found on all tubes, likely due to presence of moisture

- The highest localized corrosion rate was estimated to be 174 mils per year.
- This sample was located **near** a duct wall.
- The sample with the most uniform corrosion provided a rate of 40 mils/year
- Flue gas was not purged from the duct after operation like would be done in a fullscale plant.

| Tube<br>Bundle | Inlet or<br>Outlet | Row, 1 is lowest<br>out of 32 | Scale<br>Present | Corrosion<br>Present | Calculated<br>Corrosion Rate<br>(mils/year) |
|----------------|--------------------|-------------------------------|------------------|----------------------|---|
| 1              | Inlet              | 1                             | Yes              | Yes, localized       | 20  |
| 1              | Inlet              | 8                             | Yes              | Yes, localized       | 18  |
| 1              | Outlet             | 4                             | Yes              | Yes, localized       | 144   |
| 1              | Outlet             | 15                            | Yes              | Yes, localized       | 18  |
| 2              | Inlet              | 15                            | Yes              | Yes, localized       | 36  |
| 2              | Outlet             | 2                             | Yes              | Yes, localized       | 27  |
| 3              | Inlet              | 1                             | Yes              | Yes, localized       | 34  |
| 3              | Inlet              | 4                             | Yes              | Yes, localized       | 31  |
| 3              | Outlet             | 1                             | Yes              | Yes, localized       | 173   |
| 4              | Inlet              | 13                            | Yes              | Yes, localized       | 134   |
| 4              | Outlet             | 1                             | Yes              | Yes, prevalent       | 40  |
| Not used       | Not used           | Not used                      | -                | Flash rust only      | -   |
| Not used       | Not used           | Not used                      | -                | Flash rust only      | -   |
| Not used       | Not used           | Not used                      | -                | Flash rust only      | -   |
| Not used       | Not used           | Not used                      | -                | Flash rust only      | -   |
| Not used       | Not used           | Not used                      | -                | Flash rust only      | -   |

## Boiler condensate differential flow varied throughout the demonstration.

Differential flowrate was measured via flow meters in the boiler condensate at the inlet and outlet of the Flue Gas Cooler



## No significant impact or trend with boiler water quality detected

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Water quality was measured via conductivity meter in the boiler condensate at the outlet of the Flue Gas Cooler



### Techno-Economic Assessment

### Several cases were compared in the Techno-Economic Assessment

 Case 9 – DOE/NETL case for a 550-MW subcritical coal EGU without CCS, burning bituminous coal;

- Case 10 DOE/NETL case for a 550-MW subcritical bituminous coal EGU using the monoethanolamine (MEA) solvent, Econamine, CCS system
- Case 10b 550-MW subcritical bituminous coal EGU using the KM CDR Process for the CCS system, also has SO<sub>3</sub> control
- Case 10c 550-MW subcritical bituminous coal EGU using the KM CDR Process for the CCS system, also has SO<sub>3</sub> control and High Efficiency System

### Techno-Economic Assessment

| Case                                       |          | 9                              | 10                             | 10b  | 10c   |
|--|----------|--------------------------------|--------------------------------|--|---|
| Plant Configuration                        |          | Subcritical<br>PC w/out<br>CCS | Subcritical<br>PC w MEA<br>CCS | Subcritical<br>PC w KM<br>CDR <sup>®</sup> CCS | Subcritical PC w<br>KM CDR <sup>®</sup> CCS w<br>heat integration |
| Avoided Cost                               | \$/ton   |                                | 70.6                           | 58.5   | 51.4  |
| Total Overnight Cost                       | MM\$     | 1,098                          | 1,985                          | 1,800  | 1,741   |
| Cost of Electricity                        | mils/kWh | 59.4                           | 109.6                          | 101.5  | 96.5  |
| Percent Increase in<br>COE<br>from Case 9  |          | -                              | 98%                            | 71%  | 62%   |
| Percent Decrease in<br>COE<br>from Case 10 |          | -                              | -                              | 13.7%  | 18.0%   |



### Environment, Health, and Safety Assessment

# Six streams in a 550-MW plant with CCS were analyzed in the EH&S Assessment

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### Streams affected by the HES included:

- 1. Fly ash capture via particulate control device
- 2. FGD and polishing-scrubber wastewater
- 3. FGD gypsum or other solids
- 4. CO<sub>2</sub> capture system reclaimed waste
- 5. Product CO<sub>2</sub>
- 6. Treated flue gas exiting via the stack

## Cooling the flue gas caused an increase in uptake of metals on the fly ash

 Analysis of the fly ash captured by the pilot ESP showed increases of mercury and selenium.

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• For fly ash to be reused in concrete manufacturing, mercury limits should be examined on a site-specific basis.

|          | Concent | Concentration in Ash (µg/g <sub>ash</sub> ) |                 |  |  |
|----------|---------|---|-----------------|--|--|
| Analyte  | No FGC  | FGC FGC 20                                  |                 |  |  |
|          | 300F    | Operation                                   | SO <sub>3</sub> |  |  |
| Mercury  | 0.69    | 1.67  | 0.87            |  |  |
| Selenium | 67.5    | 134   | 103             |  |  |

## Increased concentrations of metals were also measured in leachate from the fly ash

- Ash captured by the pilot ESP was subjected to the Toxicity Characteristics Leaching Procedure (TCLP).
- The concentrations of selenium, mercury, and arsenic increased due to FGC operation.

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• All concentrations were far below the RCRA levels.

|          | Concent        | Concentration in Ash Leachate<br>(µg/l) |                 |       | <b>tory Limits</b><br>µg/l)      |
|----------|----------------|---|-----------------|-------|----------------------------------|
| Analyte  | No FGC<br>300F | FGC<br>Operation                        | FGC 203F<br>SO3 | RCRA  | Maximum<br>Contaminant<br>Limits |
| Mercury  | 0.00           | 0.18                                    | 0.04            | 200   | 2.0                              |
| Selenium | 72.8           | 166                                     | 138             | 1,000 | 50                               |
| Arsenic  | 7.76           | 15.2                                    | 10.7            | 5,000 | 10                               |

## Metals and other contaminants are expected to be reduced in other streams due to the HES

• FGD and polishing scrubber wastewater and other byproducts would have reduced metals due to increased uptake by the fly ash.

- Less CO<sub>2</sub> capture solvent reclaimed waste would be created due to the reduction of SO<sub>3</sub> entering the CCS system that creates sulfur-based heat-stable salts.
- Fugitive amine emissions would also be reduced as this is also a byproduct of SO<sub>3</sub> entering the CCS system.
- Treated flue gas exiting the plant's stack would have reduced metals and other contaminants due to increased uptake by the fly ash.

### Conclusions

### **Summary of Project Objectives**

| Quantify energy<br>efficiency<br>improvements | Identify and/or<br>resolve integration<br>problems | Quantify ancillary benefits                   |
|---|--|---|
| Unit heat rate improvement                    | Effect on water quality                            | Better ESP<br>performance                     |
| Flue gas pressure<br>drop                     | Corrosion, erosion,<br>or plugging                 | Increased SO <sub>3</sub> ,<br>Hg, Se capture |
|   | Issues with high-<br>sulfur flue gas               | Reduced water<br>consumption and<br>use       |

### Energy improvements were quantified

Quantify energy efficiency improvements

| Unit heat rate |
|----------------|
| improvement    |

Flue gas pressure drop Use of the HES can increase the generation of a 550-MW plant with CCS by 18.3 MW.

- Thermal efficiency can be increased by 0.9 percentage points (i.e. from 26.2 to 27.1%), alternately heat rate could decrease from 13,050 to 12,630 Btu/kWh.
- Use of the HES, can reduce the cost of electricity 4-5% from that of the DOE Case 10 plant with MEA CCS.
- Pressure drop across the Flue Gas Cooler was measured to be 2-4 inWc.

# Potential integration challenges were measured but high sulfur flue gas was not tested

Identify and/or resolve integration problems

| Effect on water<br>quality         |
|------------------------------------|
| Corrosion, erosion,<br>or plugging |

lssues with highsulfur flue gas

- Boiler condensate water quality was found to be unaffected by the HES.
- Corrosion was found on the Flue Gas Cooler tubes. Corrosion may have been increased due to the lack of a flue gas purge.
- No plugging was found in the Flue Gas Cooler.
- Little to no SO<sub>3</sub> was measured in the flue gas, even during injection of SO<sub>3</sub>.

## Ancillary benefits of the HES were shown to be significant



Via the reduced flue gas temperature:

- ESP outlet flue gas particulate matter concentration decreased by 36%,
- ESP outlet flue gas mercury concentration decreased by 80%,

- ESP outlet flue gas selenium concentration decreased by 33-56%,
  - Up to 60% of FGD makeup water can be saved, and
- Up to 50% of CCS cooling water can be saved.

