# Optimum Microstructure and Phase Composition of Solid Oxide Fuel Cell Electrodes with Infiltrated Nanocatalysts

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# Introduction

# Background & Motivation

- Infiltration with nanocatalysts has been widely adopted as a post-processing of electrode fabrication due to its simplicity and efficiency in improving electrochemical activity of composite electrode backbones.
- However, the conventional electrode scaffold or backbone was not designed for the additional process. A conventional electrode backbone is required to be modified for the additional infiltration process for optimization in performance.

## $\succ$ Purpose of the study

To understand whether the conventional SOFC cathode microstructure, typically optimized without considering the performance change associated with infiltration, needs to be adjusted to benefit fully from the infiltrated nanocatalysts.

# Methodology

## Calibrated Multistep ORR modeling





- (a) 3D sketch,
- (b) pathways with detailed reaction steps.

activity. Research performed by Leidos Research

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(b)

 $+ V_{0}^{"}_{VSZ} + e^{-} \leftrightarrow O_{0}^{X}_{VSZ}$ ce adsorption & dissocia  $1/2O_2 + S + Oe^- \leftrightarrow O_{ad}^$ bulk charge transfer (bct):  $+ V_{\alpha,YSZ}^{\circ} \leftrightarrow O_{\alpha,YSZ}^{X} + V_{\alpha,YSZ}^{\circ}$ 

irface charge transfer (sct)

- The developed multiphysics model considers all the basic physical/electrochemical processes occurring within the SOFCs.
- Mathematically, these highly coupled processes are formulated by sets of partial differential equations with specific expressions for different corresponding physics.

**Charge conservation** (electron-conducting phase)

$$a_{int}C_{DL}\frac{\partial}{\partial t}(\varphi_e - \varphi_i) + \nabla \cdot (-\sigma_e \nabla \varphi_e) = i_F$$

Species transport

$$\varepsilon \frac{\partial \phi}{\partial t} = \nabla \cdot \left( D_{\phi}^{eff} \nabla \phi \right) - S_{\phi}$$

 In our study, the complicated nonlinear coupled system of partial differential equations is solved, numerically, by in-house developed FORTRAN codes..

#### References

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- 2017. This work was performed in support of the US **U.S. DEPARTMENT OF** Department of Energy National Energy Technology Laboratory's ongoing Solid Oxide Fuel Cell research



agency thereof.



# > Baseline LSM-YSZ backbones (Volume fractions & Grain size)

#### Parameters:

#### 1) Volume fraction

 $V_{LSM}$ :  $V_{YSZ} = 40\%$ : 60%, 50%: 50%, or 60%: 40%

#### 2) Grain size $\beta = 0$ : baseline (P-0) $\beta = 1$ : finer grain (P-1)

 $\beta = 2$ : finest grain (P-2)

 $r_{LSM} = 0.67^{\beta_{LSM}} \cdot r_{LSM,ref}$  $r_{YSZ} = 0.5^{\beta_{YSZ}} \cdot r_{YSZ,ref}$  $r_{LSM,ref} = 0.57 \times 10^{-6} m$  $r_{YSZ,ref} = 0.35 \times 10^{-6} m$ 

### Sample notation

ex) LY\_46-P\_21 volume fraction:  $V_{LSM}$ :  $V_{YSZ} = 40\%$ : 60% grain size:  $\beta_{LSM} = 2$ ,  $\beta_{YSZ} = 1$ 



Resistance components associated with reaction steps and species transport for baseline LSM/YSZ composite cathode and nano-LSC infiltrated backbones



Total 11 backbones with different LSM/YSZ volume fractions and grain sizes. The name of backbones, LY\_*ab*-P\_*cd*, indicates the volume ratio (*ab*) and the size scale (*cd*) of LSM/YSZ.

#### Resistance components for (a) baseline cell and (b) LSC infiltrated cell.

The infiltrated nanoparticles with higher ORR activity (e.g., LSF, LSC, 50%LSM-50%LSC) significantly affected the surface adsorption/dissociation step and the transport of oxygen ion and oxygen vacancy. The gas diffusion

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ohmic resistance reduction is necessary for further improvements of infiltrated cathodes.

The thorough analysis of performance change with different infiltrates helped filter out the optimum backbone for infiltration technology

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# Results

# in oxygen reduction reactions:



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