

Plasma Arc Based Rare Earth Element Recovery from Coal Fly Ash



Presenter: Ken Jeffers, PE

kjeffers@southernresearch.org

Principal Investigator: Jay Renew, PE

jrenew@southernresearch.org

DOE-NETL 2017 REE Portfolio Review

22-Mar-2017

DOE Award No. DE-FE0027102

1-Mar-2016 – 31-Aug-2017



Project Goals and Objectives (Phase 1)

- Overall Objective – pilot-scale thermal process to concentrate REEs from coal fly ash
- DOE Key Goal – final product(s) with mixed total REE 2 wt%
- Task 1 – Project Management Planning
- Task 2 – Sampling and Characterization of Proposed Feedstocks
- Task 3 – Feasibility Study
 - 3.1 Bench-scale Experimentation
 - 3.2 Computational Modeling
- Go/No-Go Decision Point
- Task 4 – Design Package for Phase 2 Pilot-Scale Unit
- Task 5 – Phase 1 Summary Report and Phase 2 Renewal Application

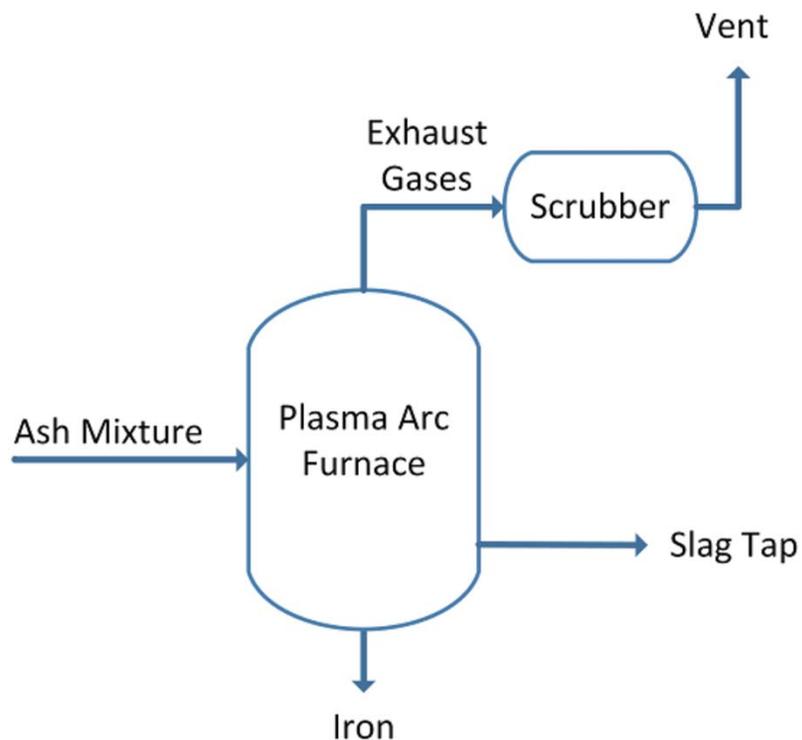
Presentation Outline

- Plasma-arc based recovery process technology
- Coal ash feedstock characteristics
- Bench-scale experiments
- Computational modeling study
- Simple Economic Analysis
- Preliminary Phase 2 pilot-scale design features
- Summary and conclusions

Proposed REE Recovery Technology

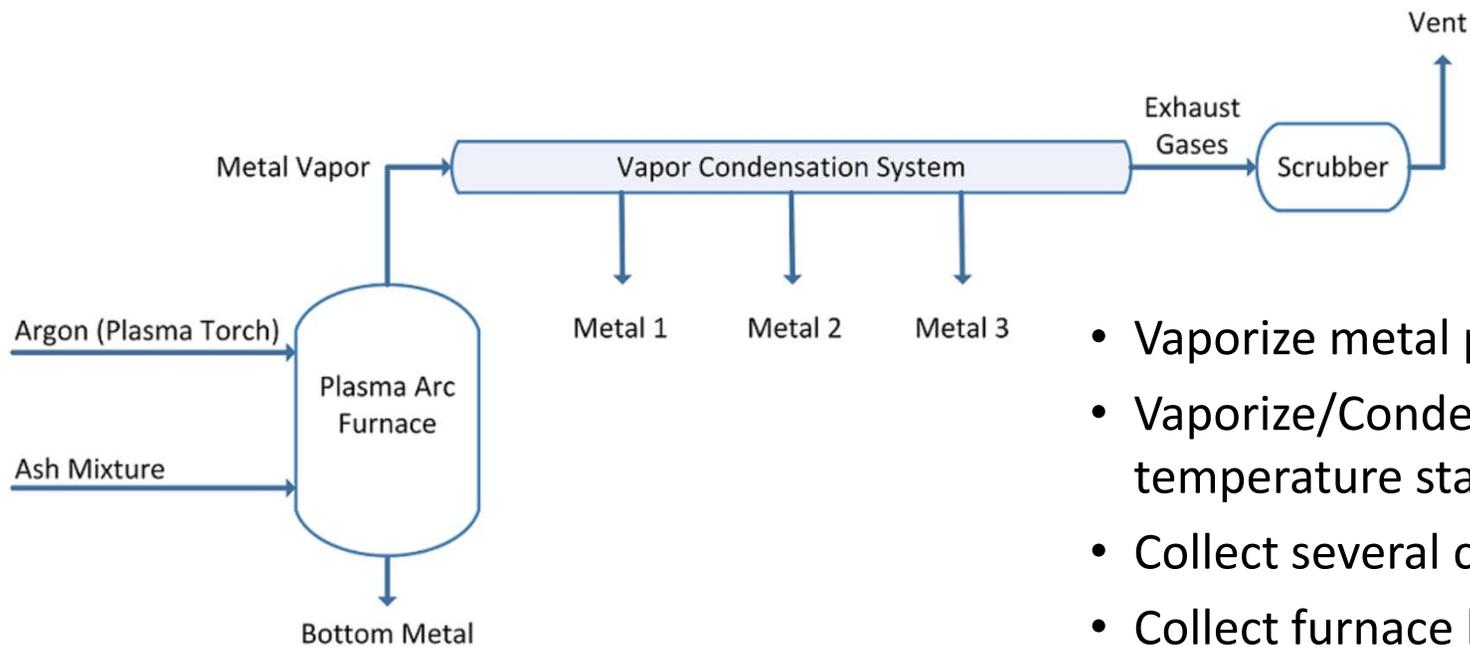
- Thermal process using plasma arc technology
- Two process options:
 - 1 – Smelting ash into slag and metal fractions, REEs collect in metal.
 - 2 – Smelting + Vaporization/condensation of metal fraction to produce further REE enrichments.

Ash Smelting Process



- Heat ash mixture under reducing conditions to 1500 – 1900 C
- Reducing conditions
 - Addition of carbon reductant
 - Conducted in absence of air(O₂)
- Reduce Fe₂O₃ to iron metal
- Separate ash into slag and metal
- Reduce REOs, collect REEs in iron
- Goal: iron product with ~2 wt% REEs
- Bench-scale experiments for proof-of-concept

Smelting + Vaporization/Condensation

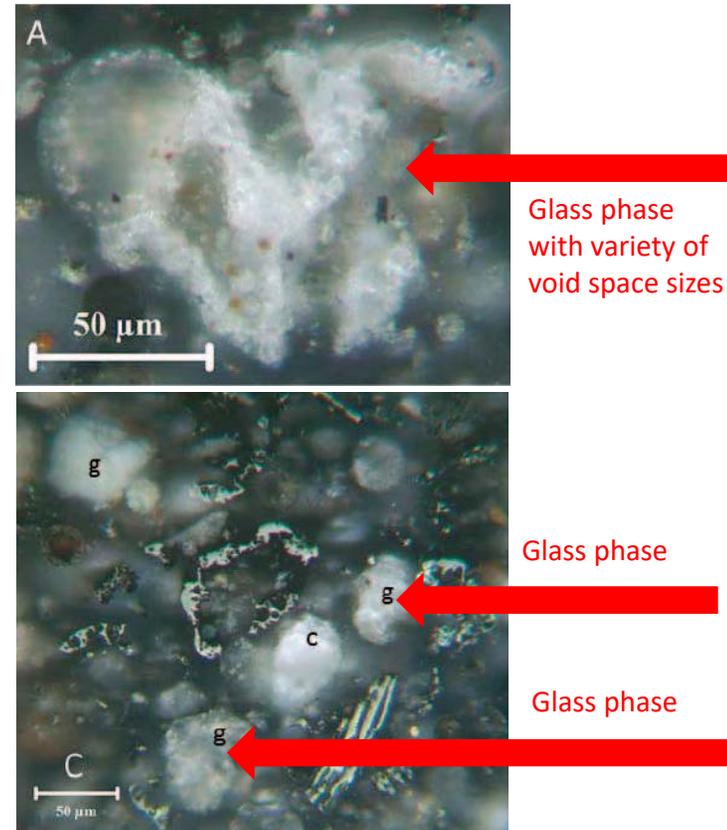


- Vaporize metal produced during smelting
- Vaporize/Condense at discrete temperature stages (2000 – 3000 C)
- Collect several condensed metal fractions
- Collect furnace bottom fraction
- Equilibrium and CFD modeling for Proof-of-concept

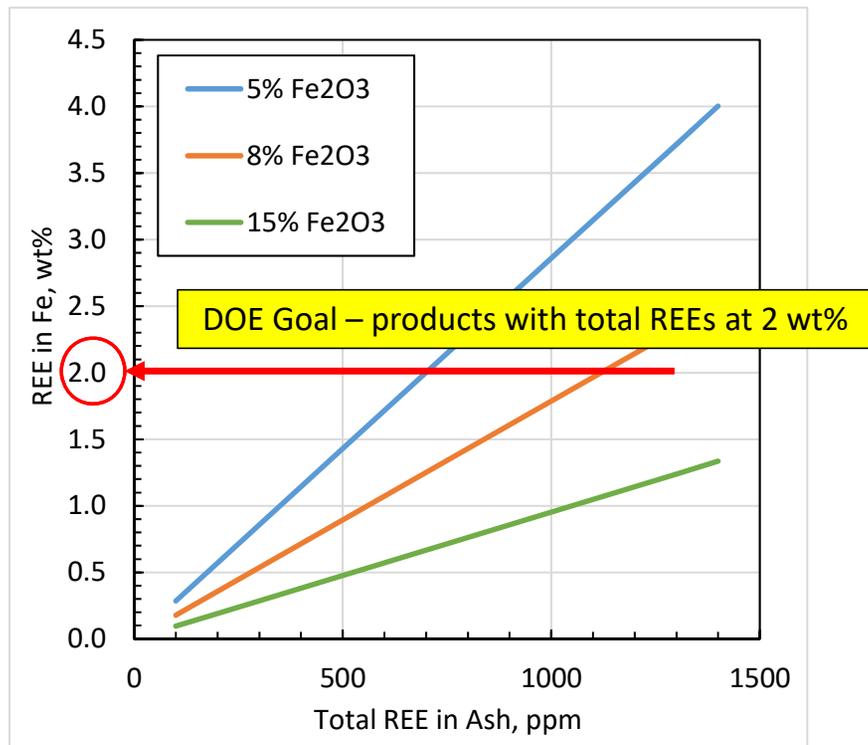
REE Occurrence in Coal Ash

- REEs are expected to behave similarly in combustion process.
- REEs are non-volatile and not found enriched on ash particle surface (Hower et al., 2013).
- REEs have been found evenly distributed in the amorphous glassy alumina-silicate phase of ash (Hower et al., 2013).
- Challenge - entire glassy phase would have to be leached to recover 100% REEs from CFA (Hower et al., 2013).

CFA SEM Graphic Source - Hower et al., 2013 Study

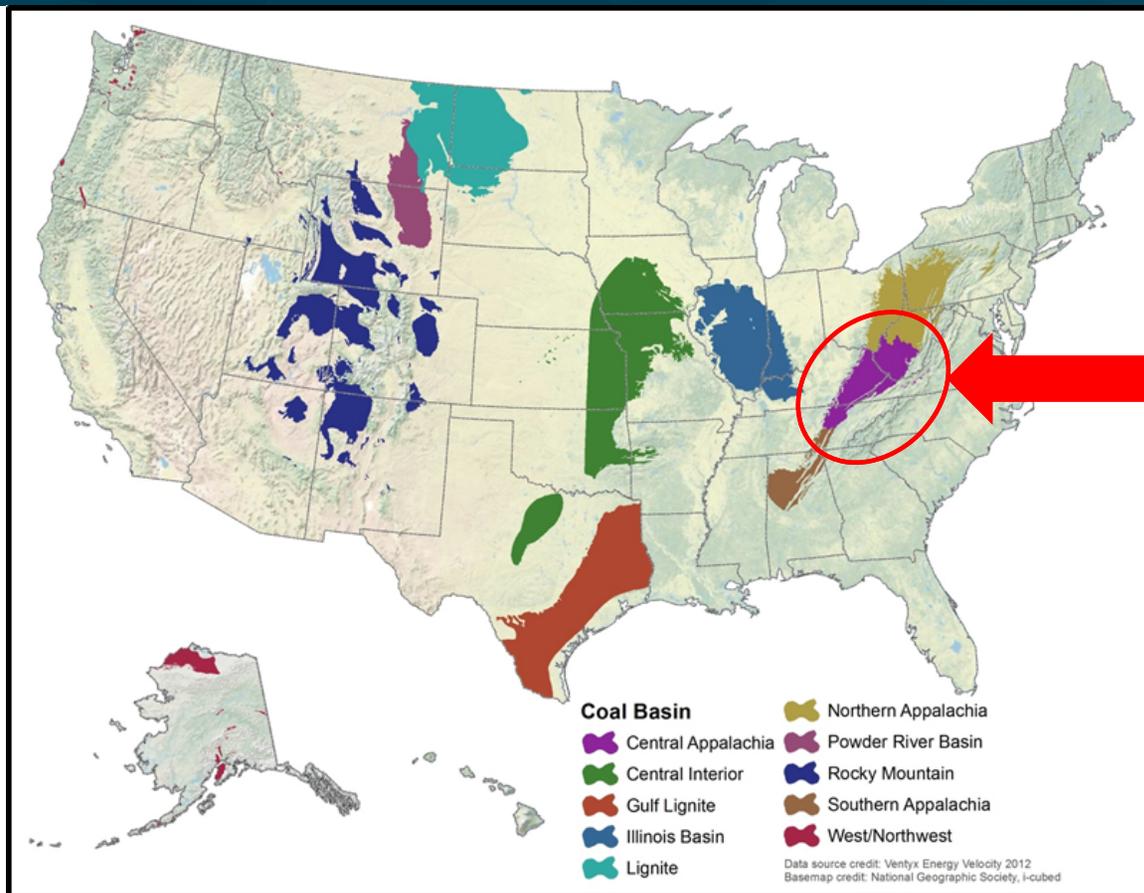


Coal Ash Feedstock Selection



- REE content in metal product depends on ash composition
- Total REE Content > 700 ppm (average ~500 ppm)
- Iron oxide content 5 – 8% (range is 4 – 25%)
- Best opportunity: ash from Central Appalachian coals

Coal Seams In United States



Central Appalachia Coal – total REE + Y concentrations as high as 1,668 ppm in coal fly ash from plants using this coal (Hower et al)

Graphic Source - ("Rare Earth Elements from Coal and Coal By-Products," 2016)

Smelting Experiments

Sample Mixtures

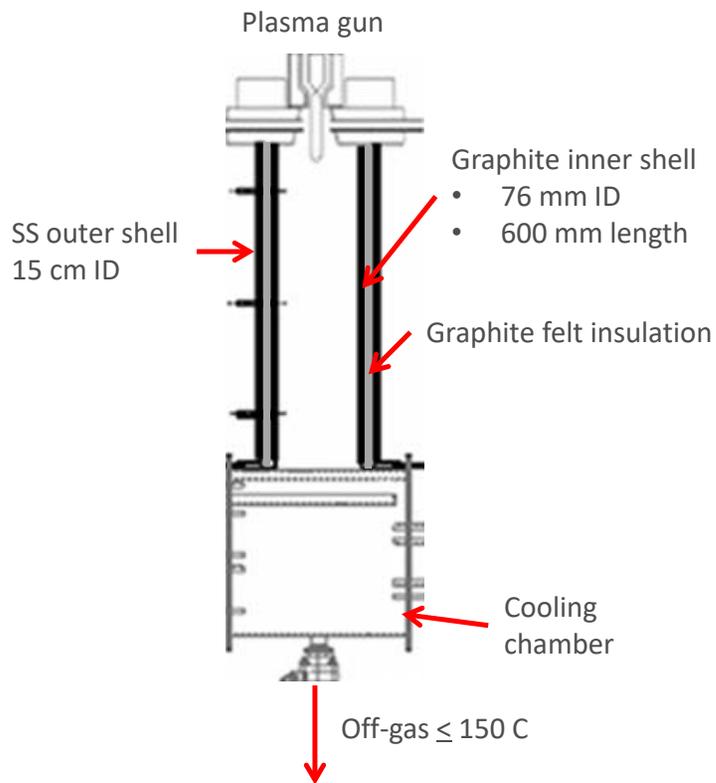
- Raw ash:
 - 400 – 1200 ppm Total REE
 - ~5% Fe_2O_3
- Fluxing agent (CaO/Lime)
- Reductant (activated charcoal)
- Supplemental metal

Key Processing Variables

- Smelting temperature
- Smelting time



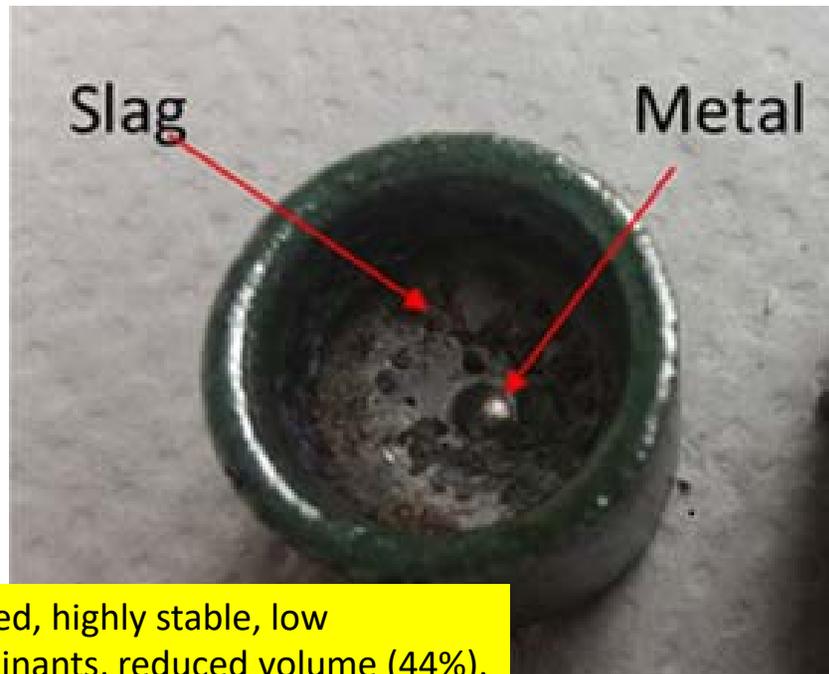
Bench-scale Plasma Reactor – U of UT



Water cooled reactor vessel



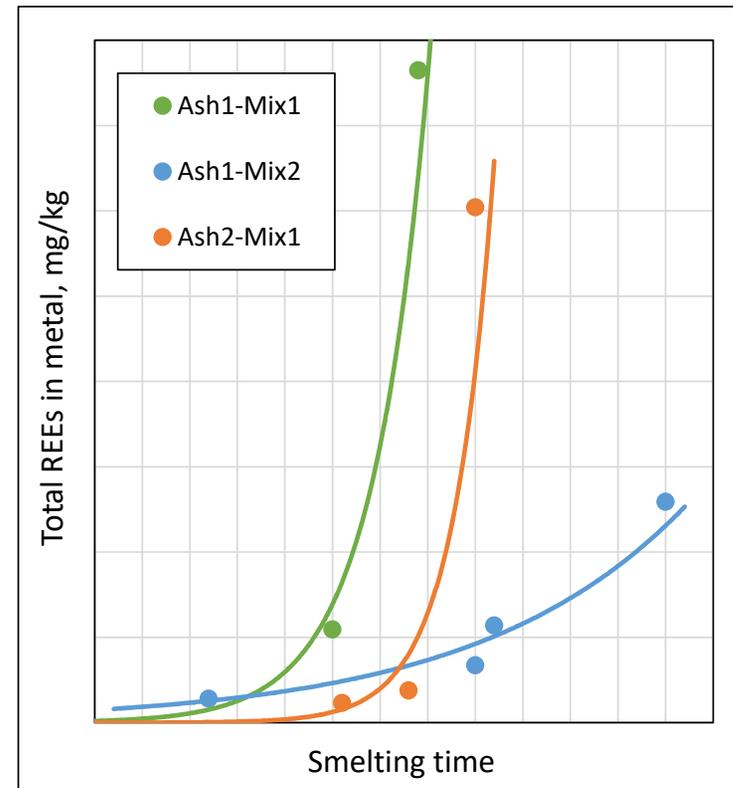
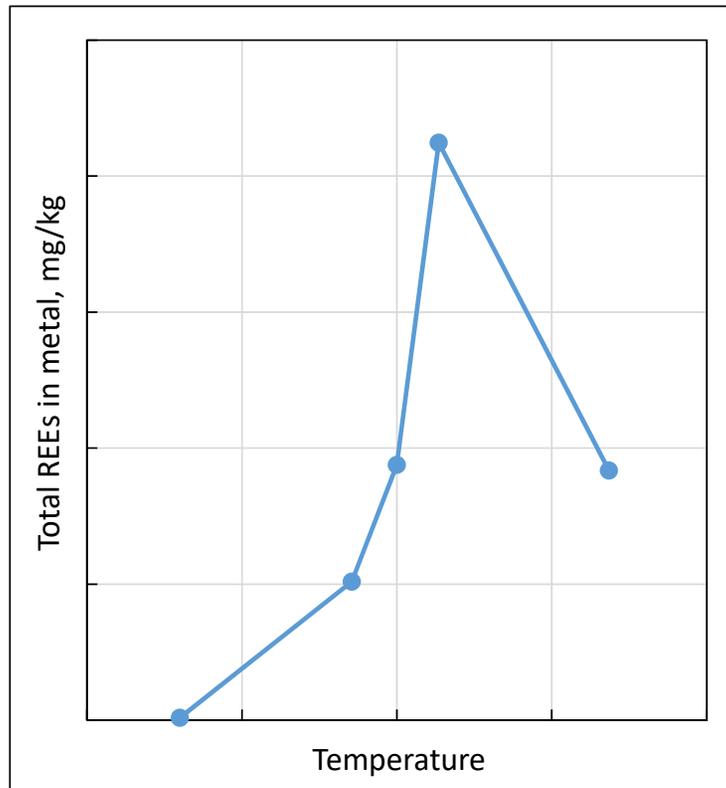
Slag-Metal Separation



Slag compared to raw ash: vitrified, highly stable, low susceptibility to leaching contaminants, reduced volume (44%).

Metal and Slag portions analyzed for REE content using ICP-MS

Smelting Results



Vaporization/Condensation Modeling

Equilibrium Calculations

- Predict species in gas phase
- Determine potential separation of individual REE species

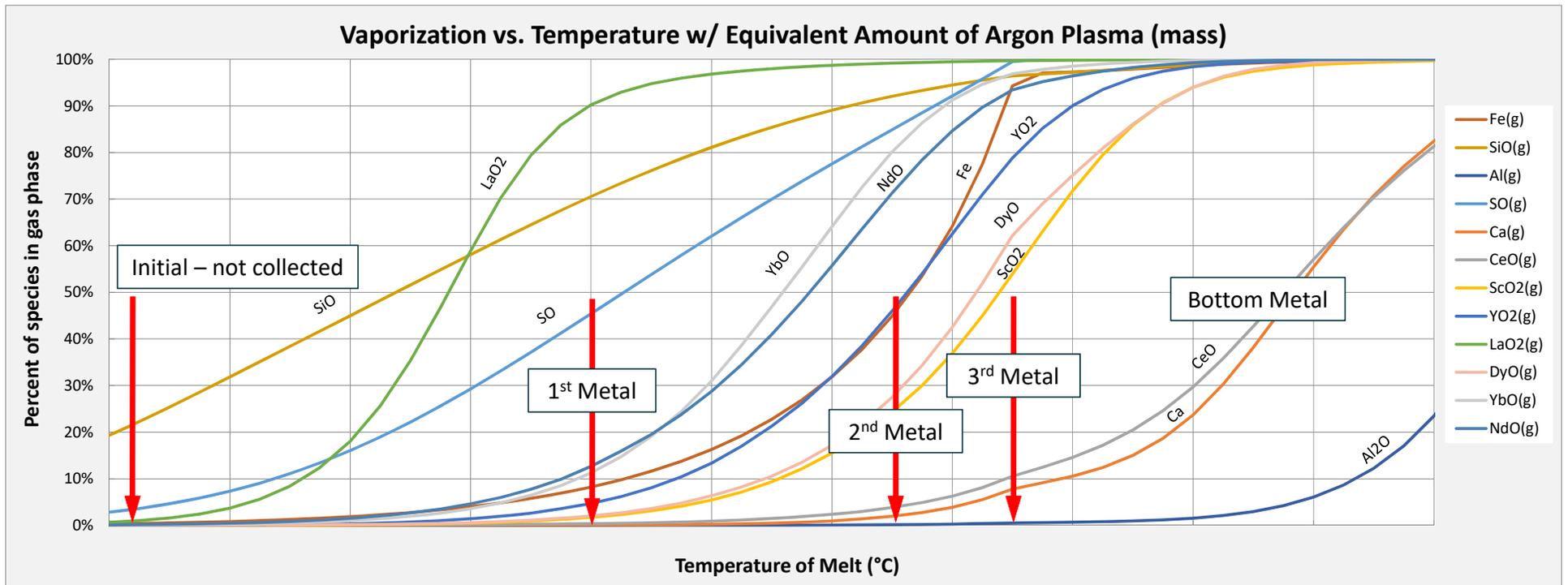
CFD Modeling

- Nucleation of metal fume particles
- Particle growth by coagulation
- Deposition on condenser surfaces

Output/Results

- Equilibrium gas phase composition vs temperature
- Select condensation temperature stages
- Composition of new condensed metal fractions
- Flow profile through the condensation/collection system

Equilibrium Calculations



Produced using Outotec HSC Chemistry 9.0

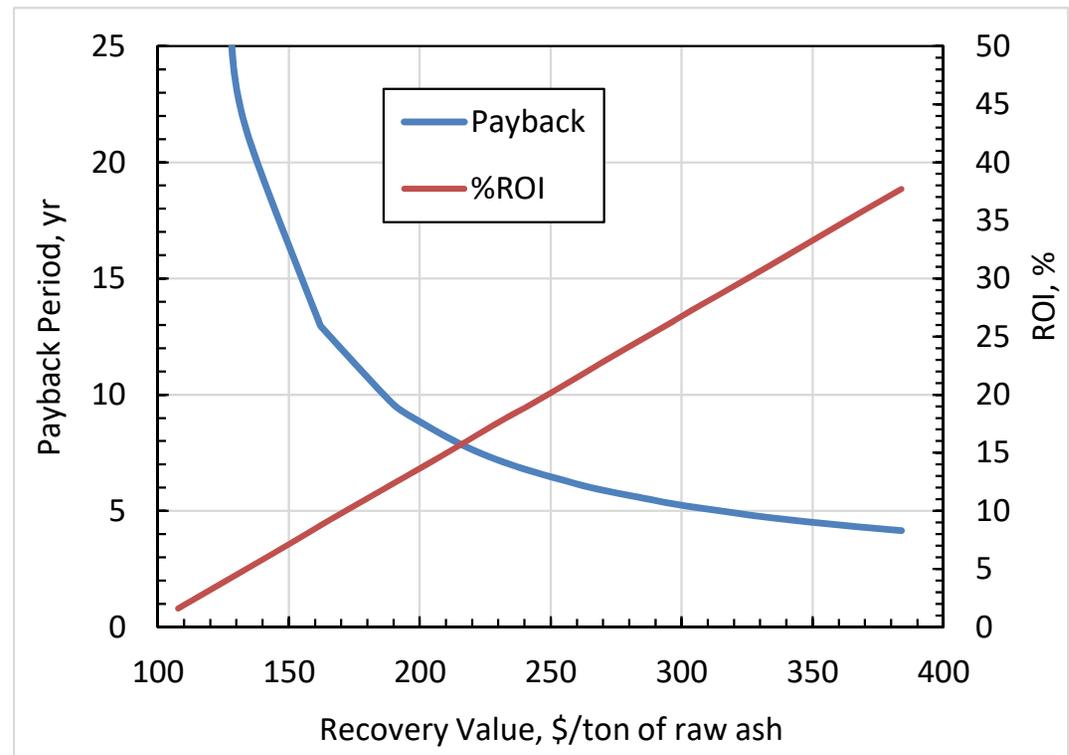
Modeling Case Study

| Element | Enrichment: Concentration in metal product/concentration in raw ash | | | |
|----------------------------|---|------------|------------|------------|
| | 1st Metal | 2nd Metal | 3rd Metal | Bottom |
| Ce | 0.5 | 1.4 | 2.1 | 95.1 |
| Sc | 2.4 | 9.2 | 9.3 | 48.9 |
| Y | 6.6 | 16.9 | 10.3 | 22.5 |
| La | 124.4 | 3.7 | 0.2 | 0.3 |
| Nd | 17.4 | 24.0 | 7.0 | 7.0 |
| Dy | 2.8 | 10.4 | 10.9 | 40.2 |
| Yb | 15.6 | 28.2 | 5.3 | 3.3 |
| Total REEs, wt% | 2.8 | 1.0 | 0.5 | 4.7 |

Base assumptions – starting ash with 1200 ppm total REEs, 50% REE recovery during smelting.

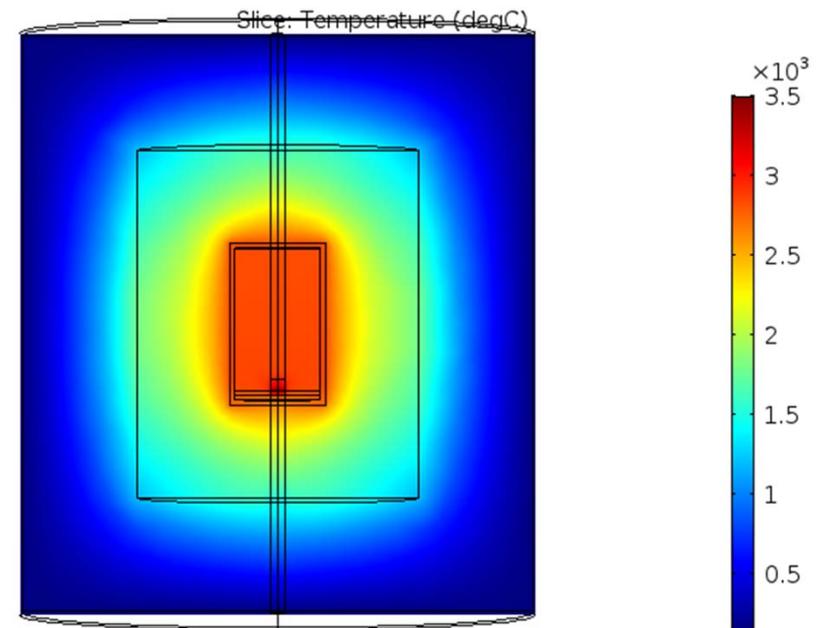
Simple Economic Analysis

- Full-scale unit
 - 350 tpd raw ash
 - Co-located at ~1000 MW power plant
- Value Streams
 - REEs
 - Metal (iron)
 - Slag for construction materials



Preliminary Phase 2 Pilot-Scale Design

- DC plasma arc furnace, graphite electrodes
- Thermal profile modeling – conduction and radiation
 - Insulation/refractory requirements
 - Overall furnace dimensions/weight
- Core crucible sizing
 - Feed/processing rates
 - Power requirements



Summary and Conclusions

- Smelting process is effective to separate ash into slag and metal
- Observed positive correlations between REE recovery and smelting time/temperature
- Vitriified slag environmentally benign compared to raw ash
- Vaporization/Condensation process modeling demonstrated
 - Individual REE separations achievable
 - Metal fractions with higher REE enrichments produced
- Economic potential
- Iterative process for pilot-scale plasma furnace design

Disclaimer

This presentation was prepared as an account of work sponsored by an agency of the United States Government (DOE-NETL, Office of Fossil Energy). Neither the United States Government nor any agency thereof, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.

Thanks for Listening! Questions?



Solving the world's
hardest problems.