



Progress in High Temperature Oxidation Modeling: Internal and External Oxidation

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#### **IPT Task 3.4: Computational Aspects in Alloy Design & Life Prediction**



(Source: Internet)

**Metal Oxidation Modeling** 



#### **State-of-the-art of Computational Modeling**





In this intermediate length regime, some common simplifications are not applicable such as local charge neutrality or constant electric field.



#### **Length Scale Gaps in Oxidation Theories**





#### Moderate film thickness regime: The coupling of charge interaction, ionic diffusion, and chemical reaction have to be addressed.



#### **Oxidation Modeling in Pure Metals**



Chemical reaction

Coupling physics with no viable simplification

- Mass transport
- Charge interaction
- Evolving structure
- Evolving electric field



## **Phase-field Method**

#### **Governing Equations for Metal Oxidation**

ReactionDiffusion + Electromigration[X<sup>-</sup>]:
$$\frac{\partial c_1}{\partial t} = K_I \Lambda_{\zeta} (Q\tilde{c}_2 - \tilde{c}_1) - K_{II} \Lambda_{\eta} \tilde{c}_1 + \nabla \cdot (\tilde{D}_1 \nabla \tilde{c}_1) - \frac{e}{k_B T} \nabla \cdot (D_1 c_1 z_1 \mathbf{E})$$
[e<sup>-</sup>]: $\frac{\partial c_2}{\partial t} = -K_I \Lambda_{\zeta} (Q\tilde{c}_2 - \tilde{c}_1) + K_{II} \Lambda_{\eta} \tilde{c}_1 + \nabla \cdot (\tilde{D}_2 \nabla \tilde{c}_2) - \frac{e}{k_B T} \nabla \cdot (D_2 c_2 z_2 \mathbf{E})$ [c<sup>+</sup>]: $\frac{\partial c_3}{\partial t} = \nabla \cdot (D_3 \nabla c_3) - \frac{e}{k_B T} \nabla \cdot (D_3 c_3 z_3 \mathbf{E})$ [M]: $\frac{\partial \eta}{\partial t} = -K_V K_{II} \Lambda_{\eta} \tilde{c}_1 + M_{\eta} \nabla^2 (\partial f / \partial \eta - \beta \nabla^2 \eta)$ 

The electric field, satisfying Poisson's equation, is solved by **an efficient numerical scheme** for arbitrary **dielectric heterogeneity** 

$$\nabla \cdot [\mathcal{E}(\mathbf{r})\nabla \varphi(\mathbf{r})] + \rho_f(\mathbf{r}) = 0$$



#### Simulated Results on Oxidation Kinetics Modeling Linear → Parabolic kinetics Transition



T Cheng, Y Wen and J Hawk, J. Phys. Chem. C 118(2014), 1269-1284



## **Oxide Growth Rate vs. Film Thickness**



Space charge effect can not be ignored



# Physics-based Modeling Capability on **External Oxidation** so far ...

## **Modeling Internal Oxidation**

#### Why we care?

- Oxidation usually starts with internal selective oxidation of certain solute elements
- Transition from internal to external oxidation is the basis for alloy design regarding oxidation resistance.





Oxidation Map: Compositional effects on<br/>the oxidation of Ni-Al alloys<br/>(N. Birks, G. Meier, F. Pettit, 2006)Transition from internal to external<br/>oxidation in Co-8.99% Ti: 900°C for 528h<br/>(J. Megusar; G. Meier, 1976)

## **Modeling Internal Oxidation**

**State-of-the-art about theoretical understanding** 

- Analyzed by C. Wagner in 1959 for a binary system with 1D assumption Oversimplification!
- Remains an open problem especially in consideration of the complex microstructure in 3D situations







Transition from internal to external oxidation in Co-8.99% Ti at 900°C for 528h, (J. Megusar; G. Meier, 1976)

## Wagner's theory on the transition from internal to external oxidation



Schematic: transition from internal to external oxidation of solute B to the formation of an external layer of  $BO_v$ . (a) internal oxidation,

(b) external oxidation with higher CB (after N. Birks, G. Meier, F. Pettit, 2006)

Transition criterion (C. Wagner 1959):  

$$N_B^{(O)} > \left[\frac{\pi f^*}{2\nu} \frac{V_m}{V_{ox}} \frac{D_O N_O^{(S)}}{D_B}\right]^{1/2}$$



Schematic: Concentration profiles for internal oxidation of A-B (Birks, Meier, Pettit, 2006)

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 $f^* = 0.3$  (for Ag-In system) (R. A. Rapp, 1961)  $f^* \sim 0.5$  (for Fe-Si system) (W. Zhao, Y. Kang, J. Orozco, B. Gleeson, 2015) There is no universal/general  $f^*$  even just for binary systems!

#### **No Predication Capability!**

#### **Phase-Field Modeling of Internal Oxidation**





Co-8.99%Ti oxidized at 900°C for 528h, (J. Megusar; G. Meier, 1976) Ni-8.67Al oxidized at 1000°C (after deep etching) (A. M.-Villafane, F. Stott, J. C.-Nava, G. Wood, 2002) TiO<sub>2</sub> needles in Co-3.7Ti alloys after reaction front passed. (J. Megusar and G. Meier, 1976)

Microstructure evolution in internal oxidation can be very different depending on specific material systems and it is no surprise that the critical volume fraction for the transition from internal to external oxidation is not a constant

#### **Technical Challenges #I**

Table 5.1 Oxide-metal-volume ratios of some

common metals

| Oxide                          | Oxide-metal-volume ratio |
|--------------------------------|--------------------------|
| K <sub>2</sub> O               | 0.45                     |
| MgO                            | 0.81                     |
| Na <sub>2</sub> O              | 0.97                     |
| $Al_2O_3$                      | 1.28                     |
| ThO <sub>2</sub>               | 1.30                     |
| $ZrO_2$                        | 1.56                     |
| Cu <sub>2</sub> O              | 1.64                     |
| NiO                            | 1.65                     |
| FeO (on $\alpha$ -Fe)          | 1.68                     |
| TiO <sub>2</sub>               | 1.70-1.78                |
| CoO                            | 1.86                     |
| Cr <sub>2</sub> O <sub>3</sub> | 2.07                     |
| $Fe_3O_4$ (on $\alpha$ -Fe)    | 2.10                     |
| $Fe_2O_3$ (on $\alpha$ -Fe)    | 2.14                     |
| $Ta_2O_5$                      | 2.50                     |
| $Nb_2O_5$                      | 2.68                     |
| $V_2O_5$                       | 3.19                     |
| WO <sub>3</sub>                | 3.30                     |

Plastic deformation with oxide growth quite common

- relieves elastic strain energy (oxide morphology evolution)
- may be time dependent (viscoplasticity)

(Hancock and Hurst, 1974)

Need to Develop a Plastic and Viscoplastic Deformation Modeling Capability!

#### **Technical Challenges #II**



Oxide-matrix interface:

• typically coherent for small

precipitates

Loses
 coherency
 upon growth



(after D. Porter and K. Easterling, 1996)

Interface Coherence with Oxide Growth

 $\Delta G = 4\mu\delta^2 \cdot \frac{4\pi r^3}{3} + 4\pi r^2 \cdot \gamma$ 

#### **Technical Challenges #II**



#### **Technical Challenges #II**

#### An incoherent interface typically:

- Higher interfacial energy but lower elastic energy
- No accommodation of shear stress
- Short-circuit diffusion path

Coherency state should have a dominant effect on oxide morphology and growth kinetics!

**Coupling between stress and diffusion: affect oxide precipitate morphology & its evolution** 

Interface Coherence with Oxide Growth

OXIDE

Need to develop a comprehensive coherency transition modeling capability!

#### Outstanding Challenges for Phase-Field Modeling of Internal Oxidation

- I. Modeling plasticity due to volume expansion with oxidation
- II. Modeling coherency loss involving transition between coherent, semicoherent, and incoherent interfaces

#### **State of the Art**

#### **Phase-field models for plasticity:**

- Proposed by X. Guo, S. Shi and X. Ma (2005)
  - 1) Assumed elastic-perfect plasticity no hardening behavior
  - 2) Solves plastic strain by calculating the variation of deviatoric strain energy

#### **State of the Art**

#### **Phase-field models for plasticity:**

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Can the deviatoric strain energy be expressed as  $E^{dis} = E^{dis}(e_{ij}^0, \mathcal{E}_{ij}^p)$ ? ( $e_{ij}^0 = \mathcal{E}_{ij}^0 - \mathcal{E}_{kk}^0 \delta_{ij} / 3$  is the deviatoric part of eigenstrain)

Considering the solution of classical elastic inclusion problem by Eshelby, for a spherical inclusion with dilatational eigenstrain

$$\varepsilon_r = \varepsilon_t = \frac{1}{3} \frac{1+\nu}{1-\nu} \varepsilon^{00} \text{ (inside)} \quad \varepsilon_r = -\frac{2}{3} \frac{1+\nu}{1-\nu} \frac{a^3}{r^3} \varepsilon^{00}, \quad \varepsilon_t = \frac{1}{3} \frac{1+\nu}{1-\nu} \frac{a^3}{r^3} \varepsilon^{00} \text{ (outside)}$$

Deviatoric stress is large near the interface outside the inclusion!

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#### A New Phase-Field Plasticity Model: Preliminary Results: Spherical Dilatational Inclusion in an Infinite Elasto-Perfect Plastic Matrix



Analytical solution by J. Lee, Y. Earmme, H. Aaronson, K. Russell 1980

#### **Preliminary Results: Prediction of coherent to incoherent transition based on energetics**



#### **Preliminary Results: Coherent precipitate with both dilatational and deviatoric eigenstrain**



Von Mises stress and hydrostatic stress distribution with a coherent interface

#### Preliminary Results: Incoherent precipitate with both dilatational and deviatoric eigenstrain

$$\boldsymbol{\varepsilon}_{ij}^{0} = \begin{bmatrix} 0.03 & & \\ & 0 & \\ & & 0 \end{bmatrix} = \begin{bmatrix} 0.01 & & & \\ & 0.01 & & \\ & & 0.01 \end{bmatrix} + \begin{bmatrix} 0.02 & & & \\ & -0.01 & & \\ & & -0.01 \end{bmatrix}$$

Von Mises stress and hydrostatic stress distribution with an incoherent interface: hydrostatic stress could not be completely removed in the precipitate!



Porter and Eastering's model needs to be reconsidered as they assumed elastic strain energy can be completely removed after loss of coherency.

## Summary and Outlook

- Developed a <u>simulation capability</u> based on Phase-Field Method to simulate external oxidation in simple systems
- Further development is on-going to develop an internal oxidation modeling capability that will eventually be able to simulate the transition from internal to external oxidation

