Investigation of Autoignition and Combustion Stability of High Pressure Supercritical Carbon Dioxide Oxycombustion

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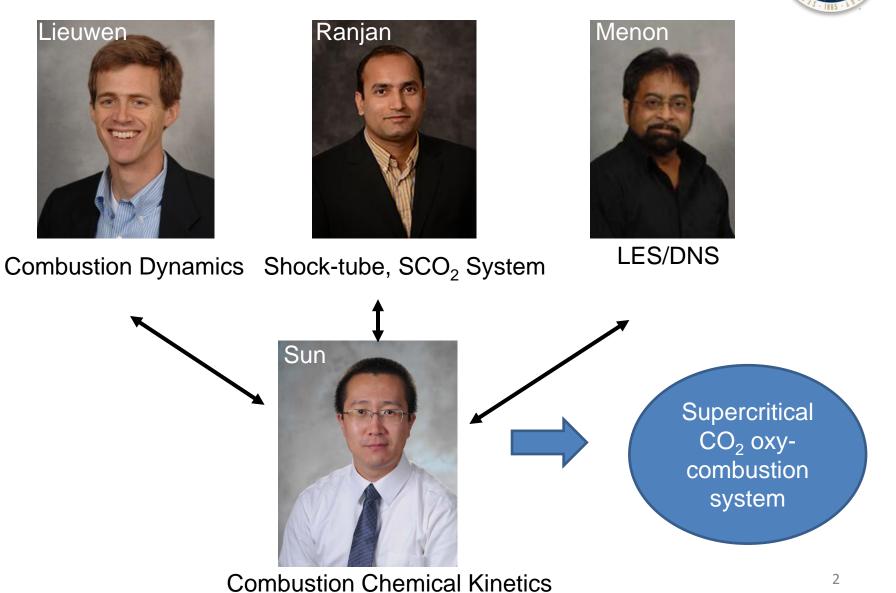


Performance period: Oct. 2015 – Sept. 2018

UTSR Project: DE-FE0025174 PM: Seth Lawson

Backstory

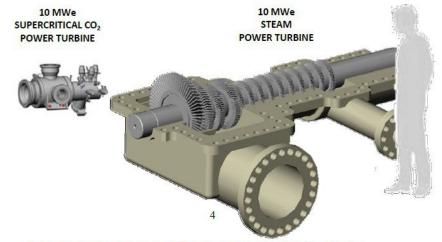




Background of Directly Fired Supercritical CO₂ cycle



- High plant conversion efficiencies exceeding 52% (LHV) with ~100% carbon capture
- Lower electricity cost (by ~15%)
- SCO₂ is a single-phase working fluid, and does not create the associated thermal fatigue or corrosion associated with two-phase flow (e.g., steam)
- SCO₂ undergoes drastic density change over small ranges of temperature and pressure → large amount of energy can be extracted → small equipment size

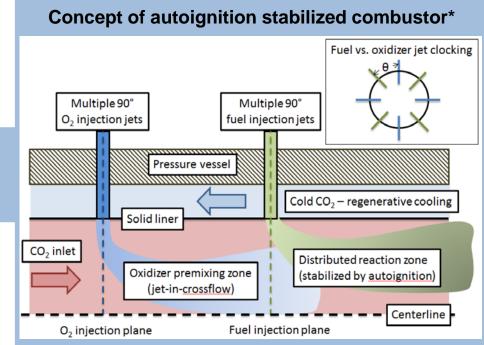


Echogen's 10 MWe sCO₂ power turbine compared to a 10 MWe steam turbine.

Overview of the Scientific Problem

- What fundamental combustion properties/knowledge we need in order to design combustor for SCO₂ oxy-combustion?
- High temperature (~1100 K) and high pressure (~200-300 atm) inlet condition
 - Conventional gas turbine combustor won't work owing to the failure of injector/flame holder at severe thermal environment

Autoignition delays and <u>combustion dynamics</u> of jet in crossflow



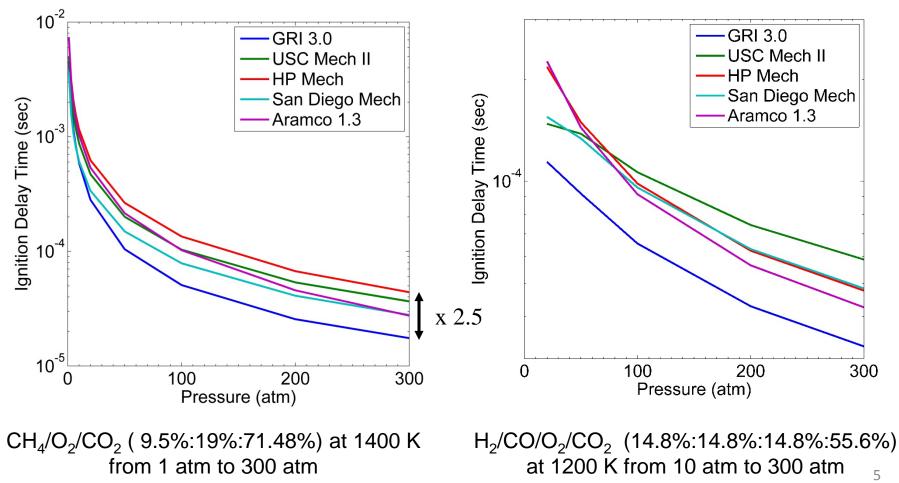


Motivation



Deviation increases with pressure: knowledge gap Kinetic models must be validated at regime of interest !!

Predicted autoignition delays from different kinetic models



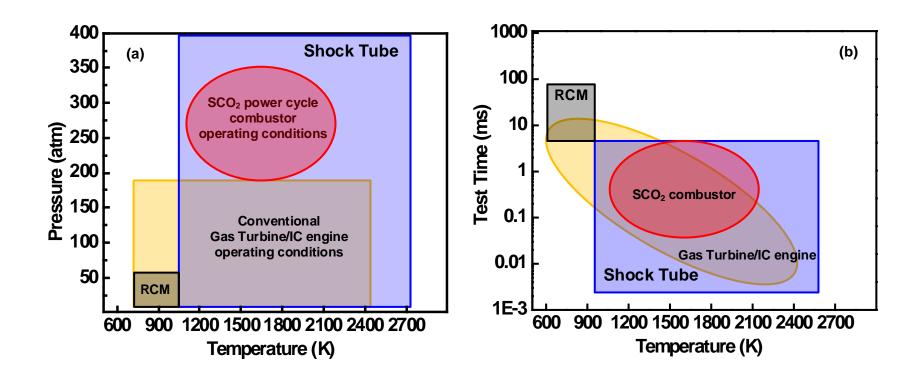
Overview of the Scientific Questions and Proposed Work



- What is the fundamental combustion properties?
 - Experimental investigation of chemical kinetic mechanisms for SCO₂ Oxy-combustion (Task 1&2: Ranjan & Sun)
- How can we use the mechanism to design combustors?
 - Development of a compact and optimized chemical kinetic mechanism for SCO₂ Oxy-combustion (Task 3: Sun)
- What is the combustor dynamics at this new condition?
 - theoretical and numerical investigation of combustion instability for SCO₂ Oxy-combustion (Task 4&5: Lieuwen, Menon & Sun)



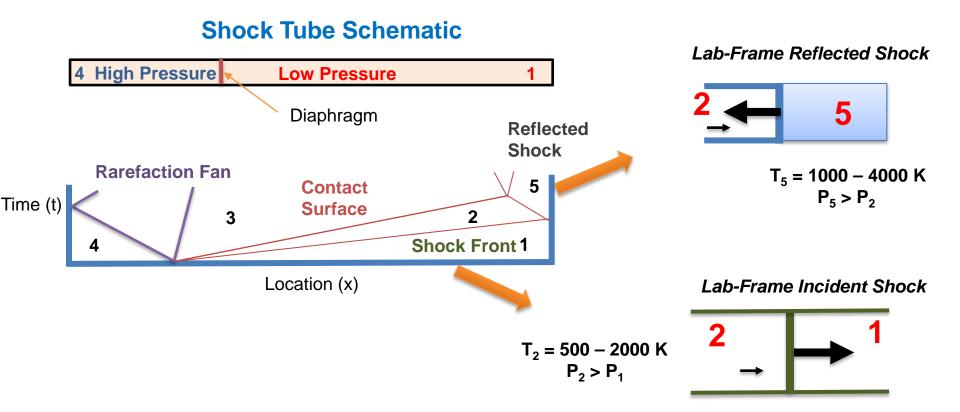
- How to study autoignition delays at SCO2 Oxycombustion condition?
 - Why Shock-Tube?



- Georgia Tech shock tube for fundamental autoignition study is under construction
- Wide pressure range (P up to 300 atm)
- Large ID (152.4 mm) to minimize non-ideal effect at very high pressure condition



Basics regarding the shock-tube:



Diagnostics: pressure and chemiluminescence Remind: currently no absorption spectroscopy can work at this condition (above 50 atm)

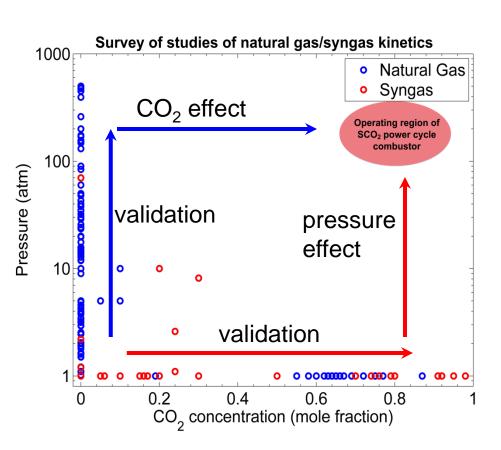


Key Capability of the GT Shock-tube

- Large internal bore (15.24 cm)—to minimize the boundary layer effect (very critical at high pressure conditions)
- It will be long (20 m total)
- Test time 50 ms (can achieve high value with modification of driver gas mixture)
- Diaphragm section replicate the current design in the operational shock-tube for turbulent mixing study
- Test pressure ~300 bar
- Preheating capability
- 0.2 μ m or better surface finish
- Optical access from end wall and side-wall
- Several locations for pressure transducers at the end wall and on side wall
- Diagnostic capability to understand the non-ideal effects in the shock-tube

Task 2: Investigation of Natural Gas and Syngas Autoignition in SCO₂ Environment

- Autoignition properties have never been investigated before in region of interest
- This task will investigate critical autoignition properties of natural gas and syngas diluted by CO₂ in region of interest
- Approach for high quality data:
 - Repeat existing experiments for validation
 - Ramp up pressure to study pressure effect
 - Ramp up CO₂ dilute concentration to study CO₂ dilution effect



A new regime to explore!

e.g.:

E.L. Petersen, et al, Symp. Combust., 1996(26), 799-806 11 S. Vasu, et al, Energy Fuels, 2011(25), 990-997



Task 3: Development of a Compact and Optimized Chemical Kinetic Mechanism for SCO₂ Oxy-combustion



- Develop an optimized, validated and compact chemical kinetic mechanism
- Employ the optimized mechanism in LES to study combustion stability
- Approach: optimize chemical kinetic mechanism based on experimental data obtained in task 2.
- Explore other methodology: Bayesian optimization for better optimization

Initial Generation of Mech Randomly generate several hundreds of mechanisms

Optimized Mech However, if this generation is good enough, we stop iteration and accept them as optimized mechanism Select the good mechanisms based on their performance of predicting auto-ignition delays, etc.

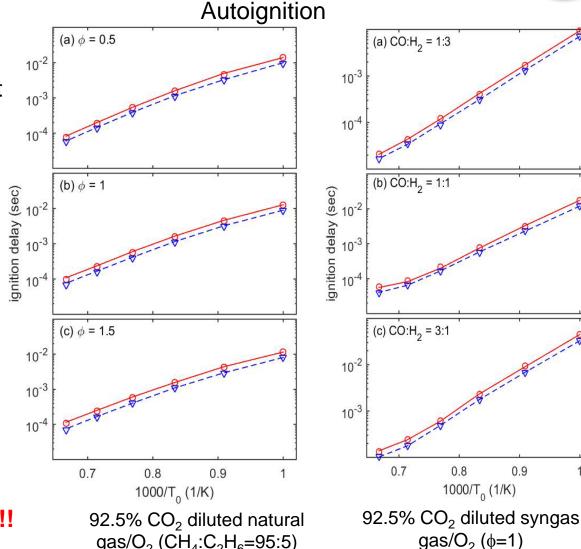
Give Birth to New Generation of Mech Good mechanims can "marry" with each other, and give birth to a new generation of mechanims

Flow chart of using Genetic Algorithm to optimize chemical kinetic mechanisms

Task 3: Development of a Compact and Optimized Chemical Kinetic Mechanism for SCO₂ Oxy-combustion

- Comparing to existing high pressure autoignition delay data, USC Mech II (111 species) has the best agreement¹. So it is used as a starting point for future optimized mechanism
- A 27 species reduced mechanism² for natural gas (CH_4/C_2H_6) and syngas (CO/H₂) is developed
- Comparison of the results from reduced (marker) and detailed mech (line). Solid lines (p = 200 a tm), dashed line (p = 300atm)

Warning: therm/trans data !! e.g., CO₂, different trend

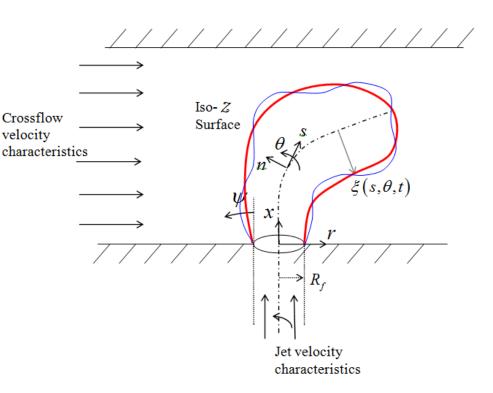




1. A. McClung, DE-FE0024041 Q1FY15 Research Performance Progress Report, SwRI 2. S. Coogan, X. Gao, W. Sun, Evaluation of Kinetic Mechanisms for Direct Fired Supercritical Oxy-Combustion of Natural Gas, TurboExpo 2016

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- The analytical work shall focus on physics based models of high pressure reacting jet in crossflow (JICF)
- A key goal of this work shall be to determine the relationship between flow disturbances and heat release oscillations

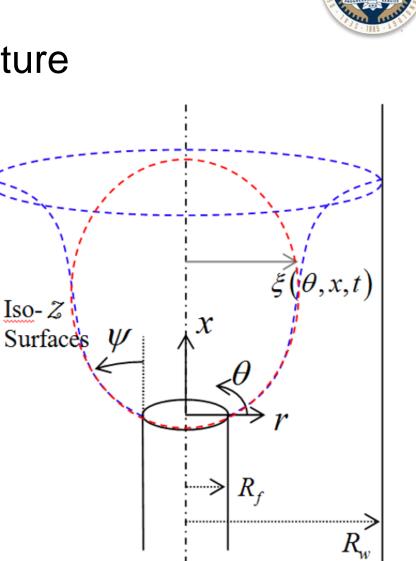


Analytic model of jet in crossflow



 Established model: Mixture fraction formulation

$$\frac{\partial Z}{\partial t} + \vec{u} \cdot \nabla Z = \nabla \cdot \left(\mathscr{D} \nabla Z \right)$$



Magina, N., Lieuwen, T. "Three-dimensional and swirl effects on harmonically forced, non-premixed flames". *9th US National Combustion Meeting* (2015).

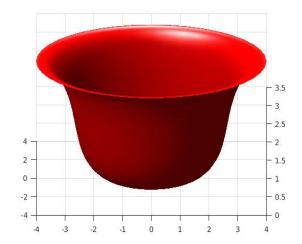


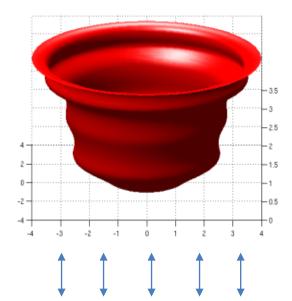


• Solution: Space-Time Dynamics of Z_{st} Surface

Bulk Axial Forcing $u_{x,1} = \varepsilon U_0 \exp[-i\omega t]$

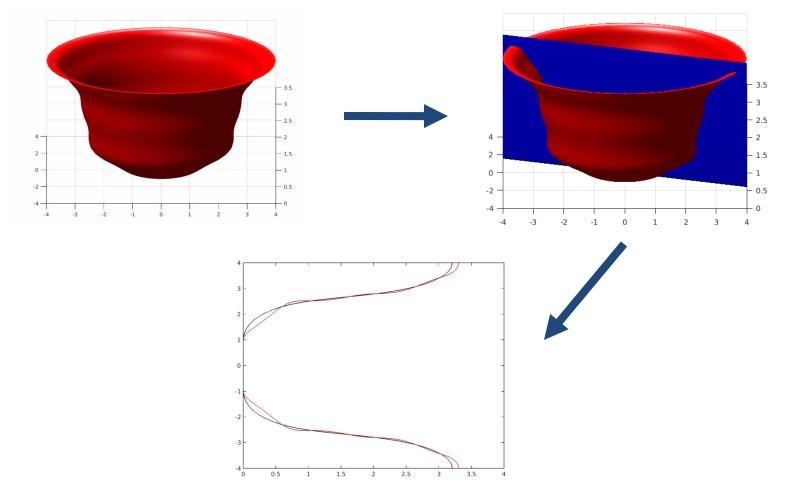
$$Pe >> 1 \qquad \frac{\xi_{1,n}(x,t)}{R_f} = \frac{i\varepsilon \exp[-i\omega t]}{2\pi St} \sin\psi_0(x) \left[1 - \exp\left[2\pi iSt\frac{x}{R_f}\right] \exp\left[-\frac{4\pi^2 St^2}{Pe}\frac{x}{R_f}\right]\right] + O\left(\frac{1}{Pe^2}\right)$$







• Solution: Space-Time Dynamics of Z_{st} Surface



Key Goals of Task 4

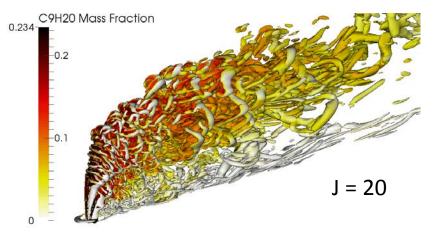


- Determine the gain-phase relationship between flow disturbances and heat release oscillations
- Compute time averaged flow and flame features
- Account for supercritical effects on diffusion coefficients, and radiation

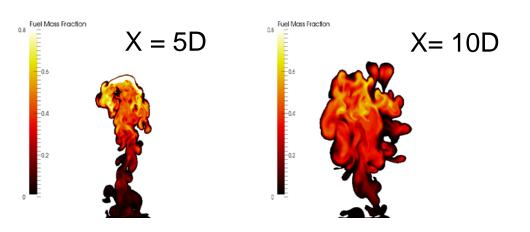
Task 5: LES Studies of Supercritical Mixing and Combustion



Supercritical Mixing in JICF (leveraged by our rocket engine work)



Vorticity Contours of supercritical Kerosene in air

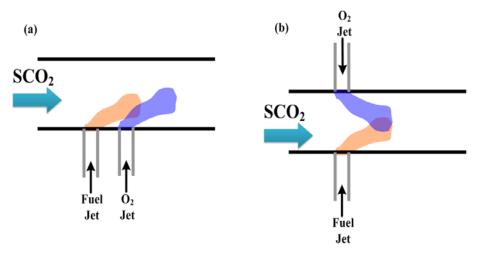


- LES capability exists to simulate supercritical mixing and reacting flows
- Uses Peng-Robinson EOS for real gas properties with finite-rate kinetics
- Simulations to be used to study mixing and combustion between SCO₂, fuel/oxidizer
- Effect of radiation

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Task 5: LES Studies of Supercritical Mixing and Combustion

- Task 5a: Simulate supercritical mixing/combustion in JICF
- Task 5b: Implement optimized kinetics from Task 3 for reacting studies
- Task 5c: Simulate and analyze conditions resulting in combustion stability in possible combustor geometries
 - Vary inflow and combustor operating conditions
 - Vary injection conditions
- Task 5d: Feedback sensitive reactions to Task 3 to further refine the mechanism



Possible circular combustor design for SCO2 power cycle (will be modeled)



Task 5: LES Studies of Supercritical Mixing and Combustion



Regime of interest: P = 200-300 atm

fluid	Critical temperature (K)	Critical pressure (atm)	Warning: Mixing rule !!
CO ₂	304	72.9	
H ₂ O	647	217.8	A mixture may have one, more
CH_4	190	45.4	than one, or no critical points
C_2H_6	305	48.1	$P_{1} = 72.9 atm$
H ₂	32.9	12.8	P _{c, CO2} = 72.9 atm P _{c, C16H34} = 25 atm P _{c,mixture} = 238 atm
CO	125.9	34.5	$P_{c,mixture} = 238 \text{ atm}$
O ₂	154.6	49.7	(CO ₂ :C ₁₆ H ₃₄ =0.94:0.06)

Transcritical regime exists and is very challenging to model

New physics and chemistry in gas turbine !!

Deliverables



- New fundamental combustion data base for SCO₂ power cycles
- Optimized predictive kinetic mechanism for natural gas and syngas
- Analytic and numerical models of jet in cross flow at SCO₂ power cycle operating conditions

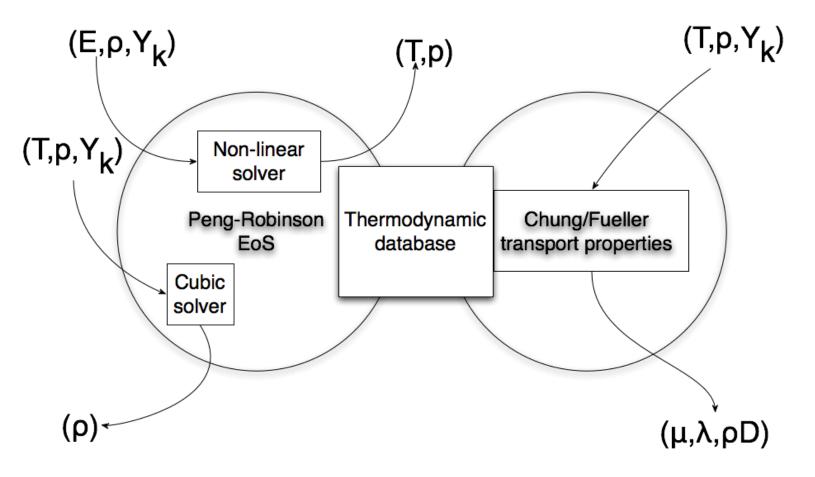


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Task 5: LES Studies of Supercritical Mixing and Combustion

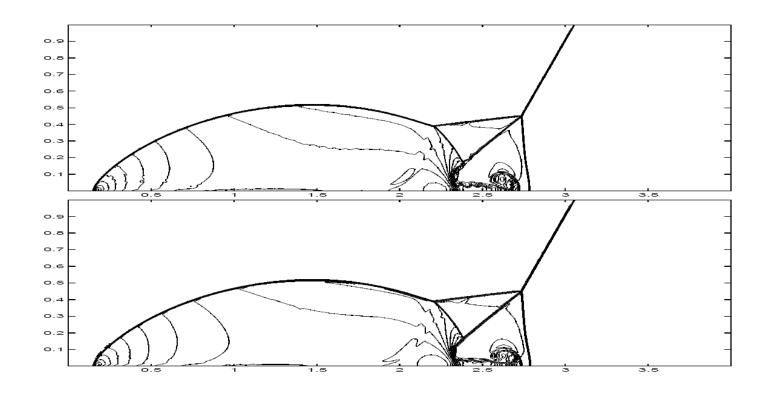


Real gas framework



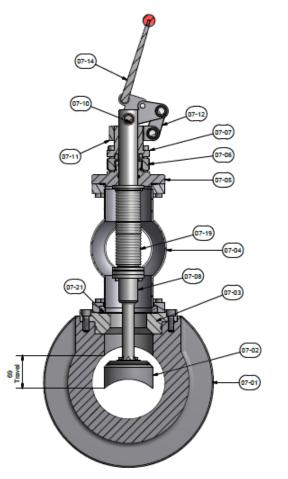
2D Effect in Shock Tube



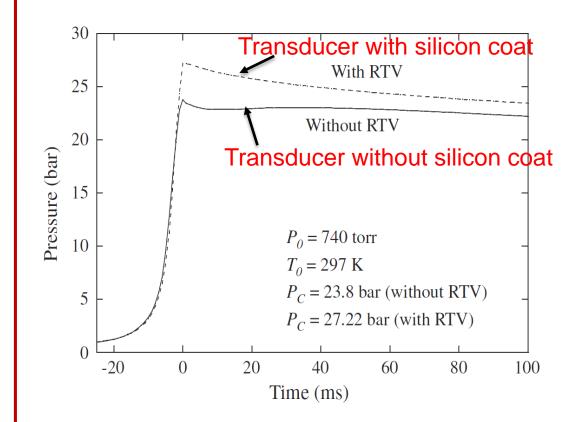


Density contour of Double Mach reflection

Plugs/valves have the same curvature with the inner surface of the tube to minimize the flow and shock obstruction



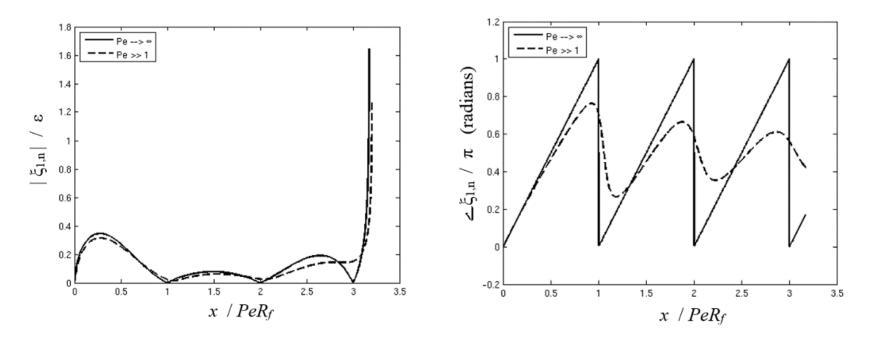
Pressure sensors for measuring the ignition delay



Mittal, Gaurav, and Anil Bhari. "A rapid compression machine with crevice 26 containment." *Combustion and Flame* 160.12 (2013): 2975-2981.



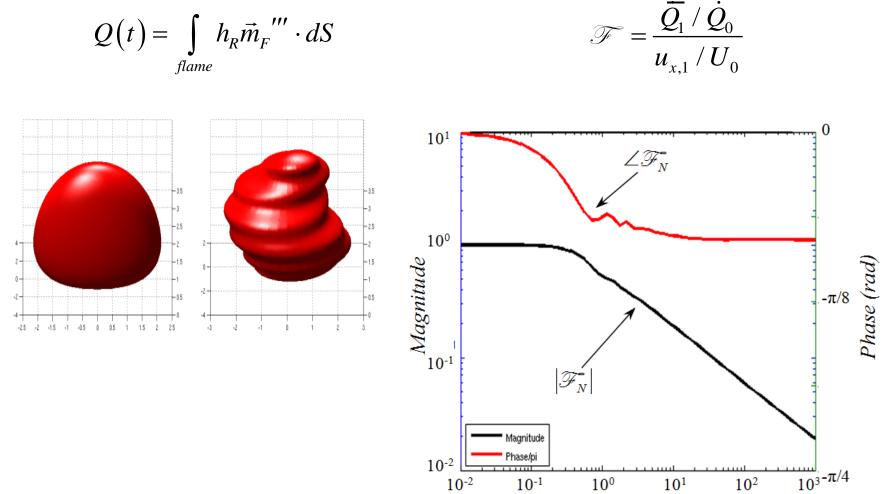
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• Solution: Space-Time Dynamics of Z_{st} Surface



10-2

10-1

100

101

 St_L

10²

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Objectives and Approaches



Objectives:

- Perform fundamental R&D on combustion kinetics and dynamics at supercritical CO₂ power cycle operating conditions for natural gas and syngas
- Focus on key knowledge gaps associated with supercritical CO₂ oxycombustion including autoignition properties, development of predictive chemical kinetic mechanism, and analyses of flow, mixing, and flame dynamics

Approaches:

- New experiments to generate fundamental data base
- Development of new predictive kinetic mechanism assisted by experiments
- Study of flame/flow stability based on newly developed kinetic mechanism
- Integration of experimental, numerical, and theoretical efforts