

Materials Issues for Advanced Supercritical CO₂ Cycles and High Efficiency Gas Turbines

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Office of Fossil Energy (B. White, project monitor)

Acknowledgments

TBC Task leaders: J. A. Haynes - coating procurement
K. A. Unocic - characterization (TEM)
S. Sampath, Stonybrook U. - processing

M. Lance - PSLS, 3D LM

H. Aldridge, U. Florida PhD student - lifetime modeling

G. Garner, M. Stephens - oxidation experiments

sCO₂: Jim Keiser - autoclave design

Mike Howell - construction and operations

T. Lowe - characterization

D. W. Coffey - TEM specimen preparation, FIB

T. Jordan - metallography

D. Leonard - EPMA

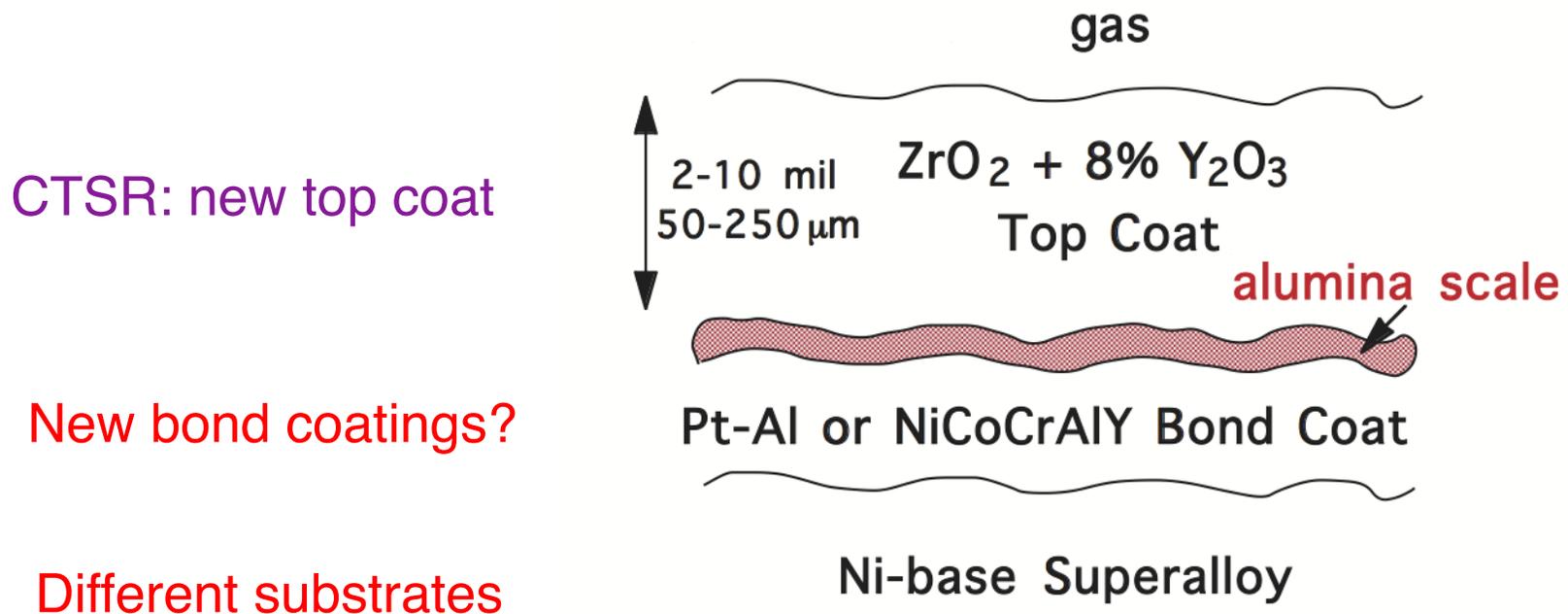
Superalloys: Howmet, Siemens, Capstone Turbines

Alloys: Haynes, Special Metals, ATI, Sumitomo, Sandvik...

Research sponsored by: U. S. Department of Energy, Office of Coal and Power R&D, Office of Fossil Energy

Looking for coating solutions

New environments (higher H_2O , CO_2 , SO_2)



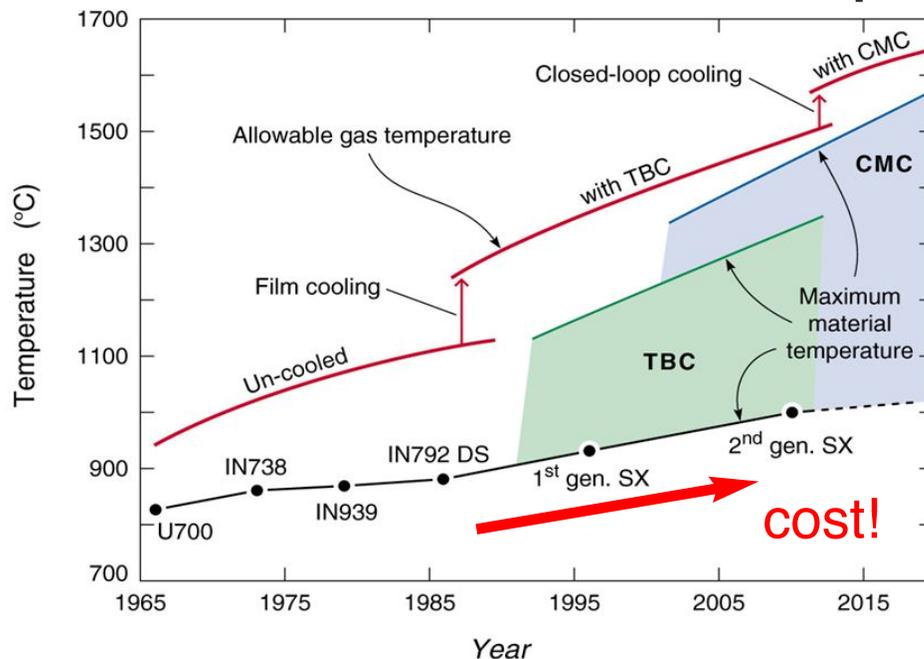
#1 More durable coatings will benefit
IGCC and NGCC

#2 Focus on alumina scale as “weak link”

Coatings for Syngas-H₂ Turbines completed in 2015

Used furnace cycle testing to evaluate:

- Effect of H₂O, CO₂, etc. on TBC lifetime
- 1-h vs. 100-h cycles
- NiCoCrAlYHfSi bond coatings vs. Ti-B additions
- effect of substrate composition (de-evolution):



MarM247 <- PWA1483 <- CMSX4
8Cr+1Hf 12Cr+4Ti 6Cr+3Re

Standardized coating procedures

16mm disks: superalloy substrates (all at.%):

X4: 13.0Al 1.2Ti 6.4Cr **0.9Re** 0.03Hf 17ppmaS

1483: **7.3Al** **4.9Ti** **13.6Cr** 0Re <0.001Hf <3ppmaS

247: 12.6Al 1.3Ti 9.7Cr 0Re **0.47Hf** <3ppmaS

High Velocity Oxygen Fuel (HVOF) bond coating:

Ni-18Co-16Cr-23Al-0.4Y-0.07Hf-0.65Si

Roughness: final coarse powder spray

APS top coating: ZrO_2 - Y_2O_3 (1 side)

Oxidation: 1-h and 100-h cycles

900° and 1100°C: air + 10% H₂O

Characterization: Metallographic cross-sections

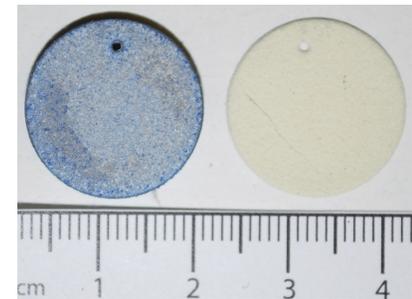
SEM/EDS/EBSD

EPMA (WDS)

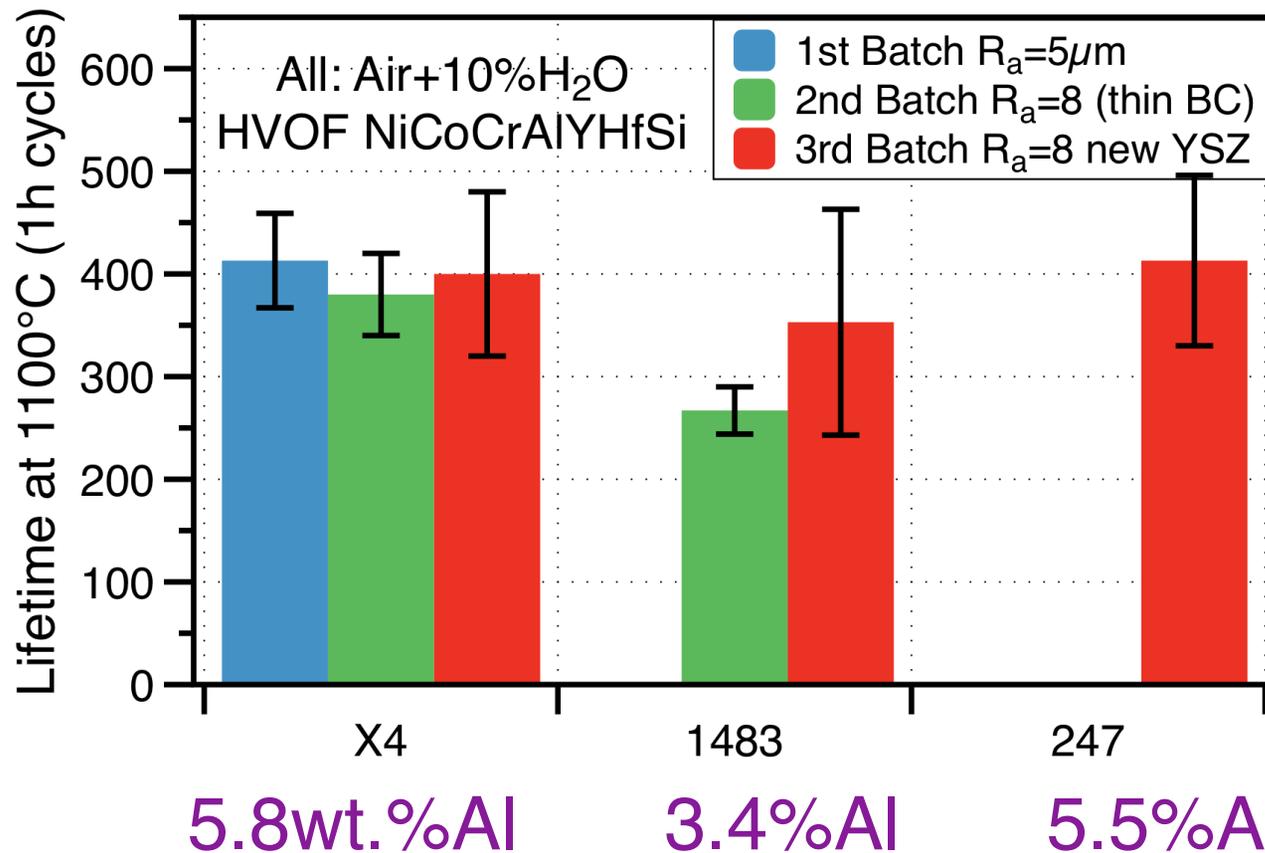
PSLS, 3D LM

FIB/TEM

16mm diameter coupon



1-h cycles: DS 247 similar to X4 HVOF NiCoCrAlYHfSi/APS YSZ coatings

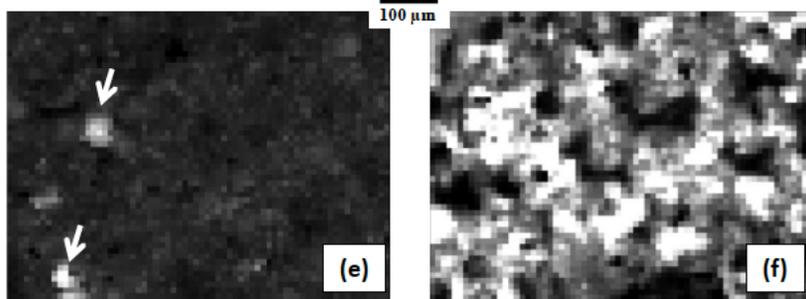
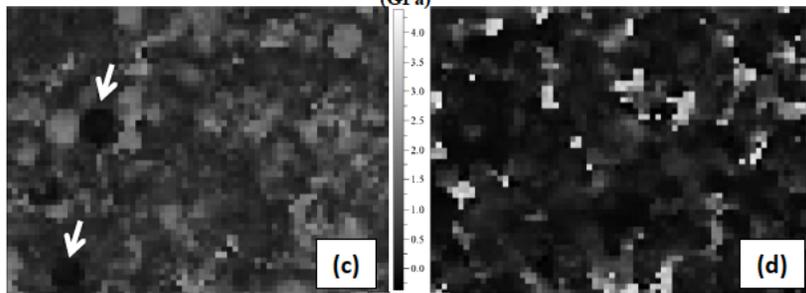
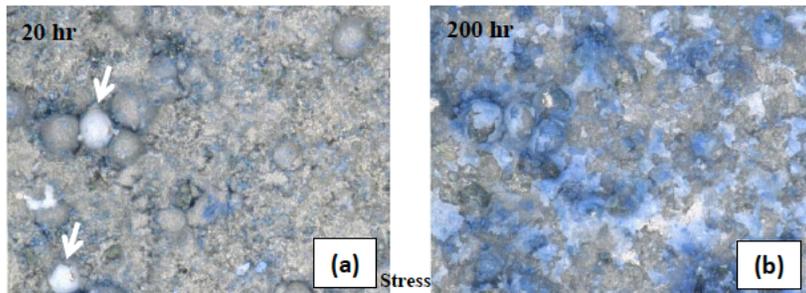


1100°C average lifetime of 3 similarly coated specimens

No parameter effect #1

3D image + PSLS: maps & histograms

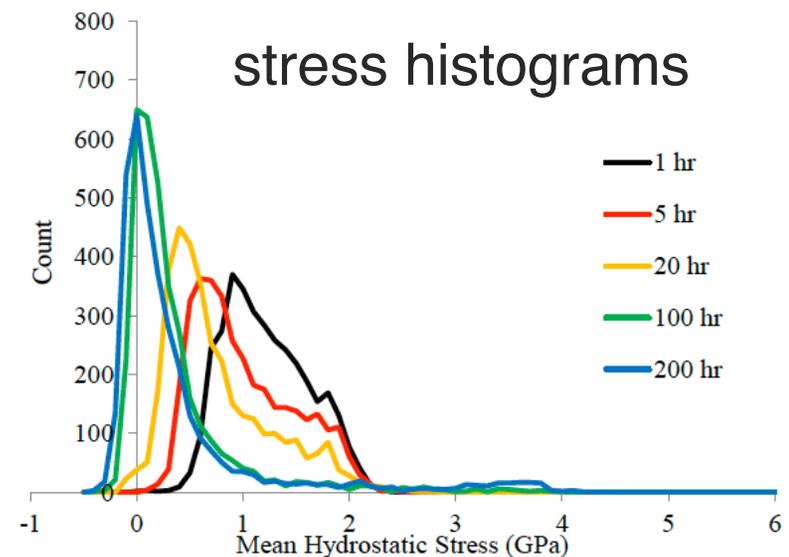
1483: 1100°C, dry air, 1h cycles



20 cycles

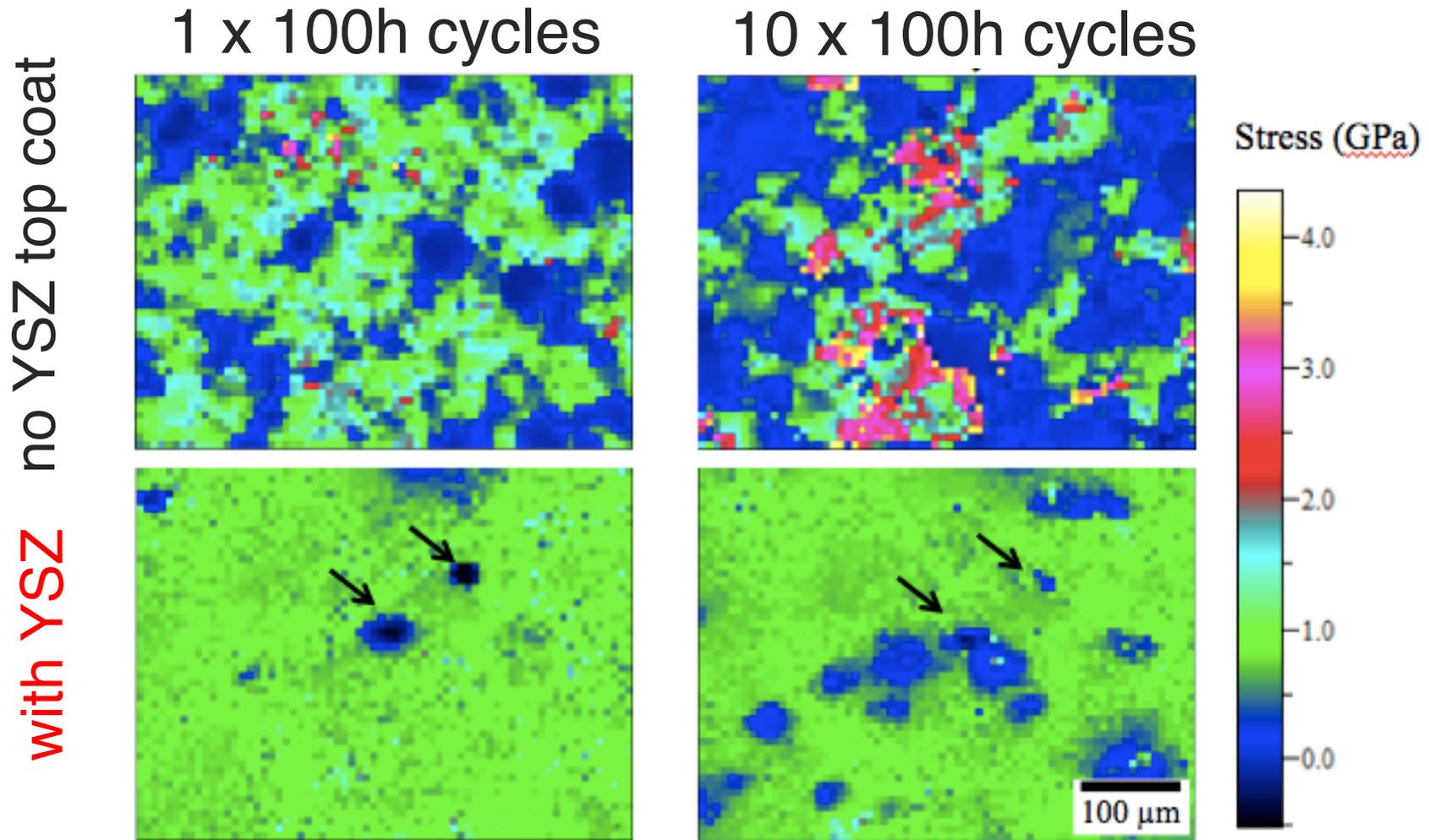
200 cycles

No YSZ top coat:
nothing to constrain spallation



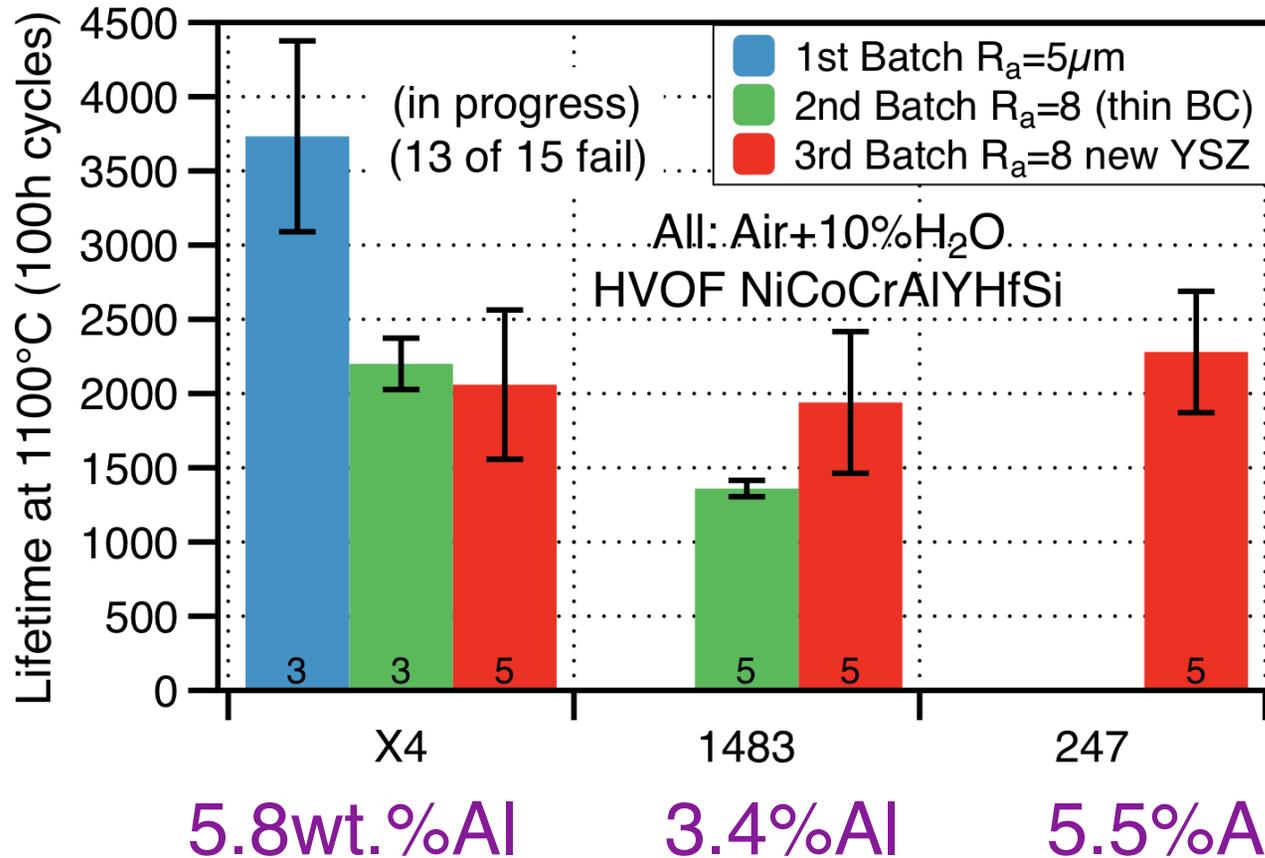
Able to measure stress under YSZ

HVOF MCrAlYHfSi/APS YSZ on 1483 in 50% H_2O



Stress in alumina scale measured by PSLS or PLPS

Also similar lifetime with 100h cycle HVOF NiCoCrAlYHfSi/APS YSZ coatings



1100°C
2012°F

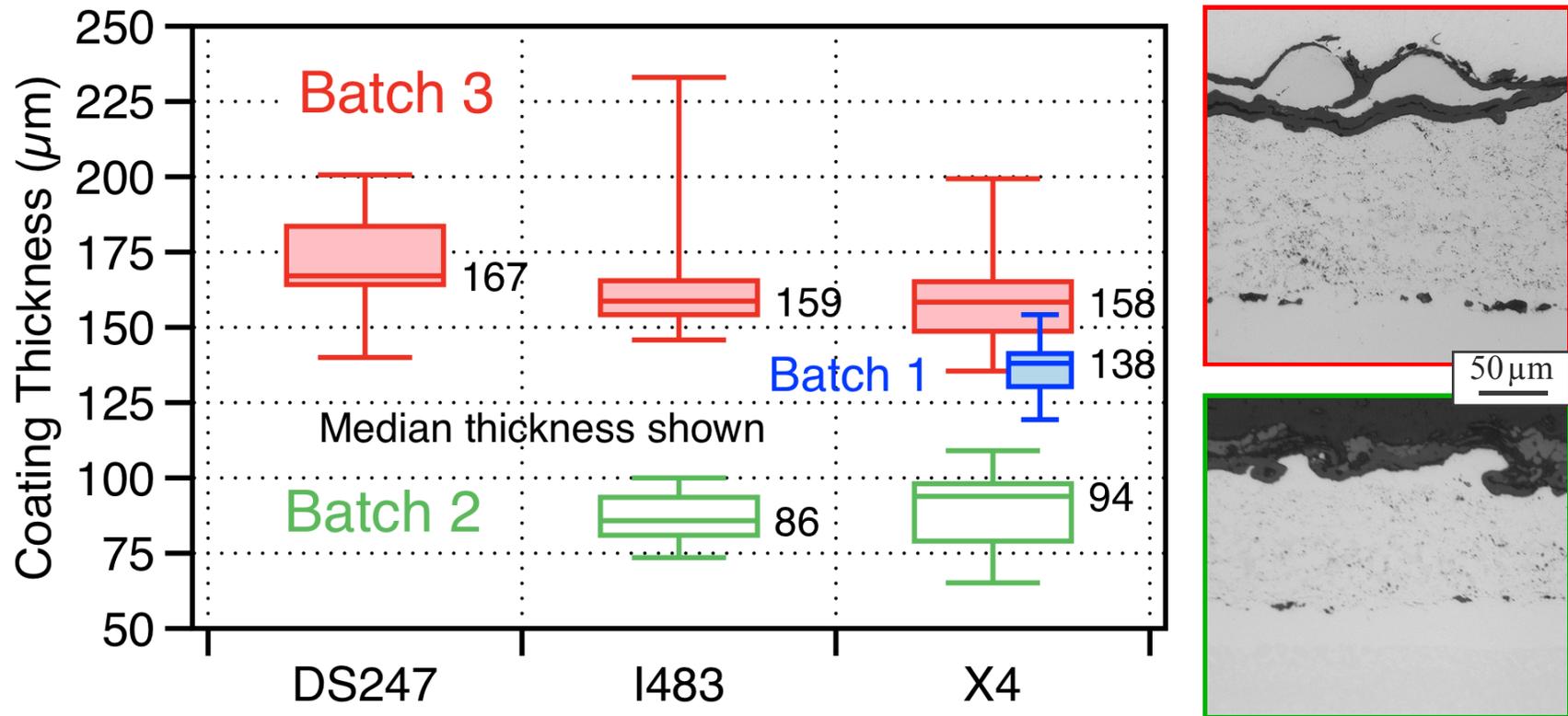
10% H_2O
simulated
exhaust

5 similarly coated specimens with each substrate
100h cycles more relevant to baseload generation

No parameter effect #2

Thicker bond coating appeared to eliminate substrate effect

3 batches of HVOF NiCoCrAlYHfSi coatings

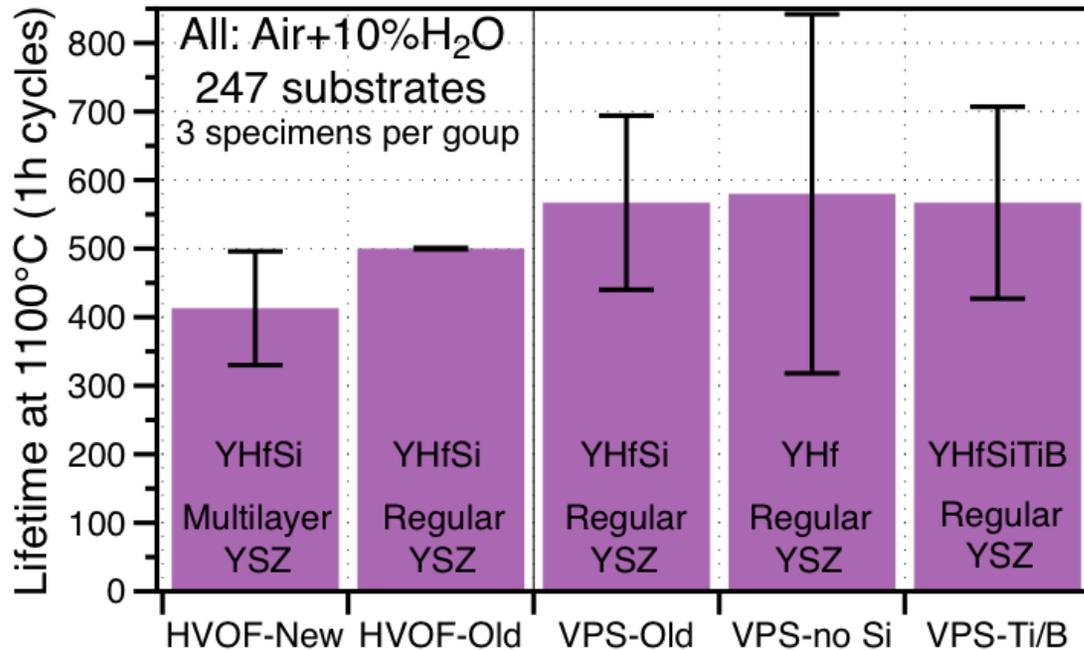


Batch 2 - saw lower lifetime for 1483 substrates

Batch 3 - no substrate effect observed

Last set of coupons: similar lifetimes

1-h cycles, 1100°C, air + 10%H₂O



1100°C, 2012°F
247 substrates
3 specimen average

Bars 1-2: single vs. CTSR double layer YSZ

Bars 2-3: HVOF vs. VPS NiCoCrAlYHfSi

Bars 3-4: VPS YHfSi vs. YHf (effect of Si)

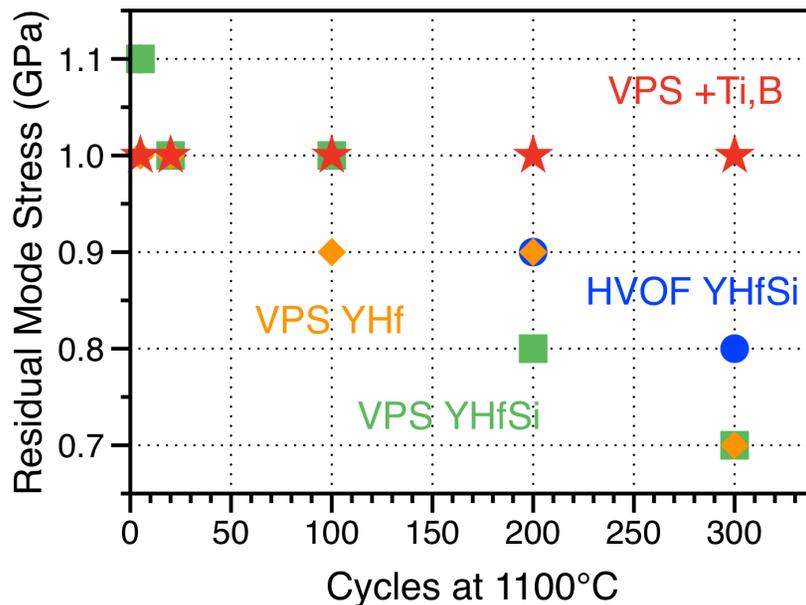
Bars 3-5: VPS YHfSi vs. YHfSiTiB (effect of Ti,B)

No parameter effects #3, #4, #5, #6

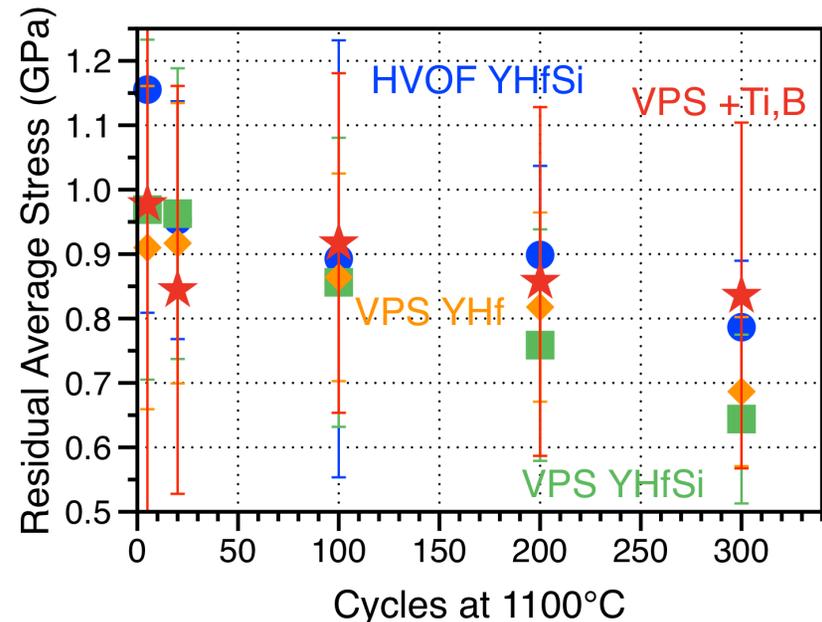
Different stress with Ti-B addition

PSLS measurements during 1-h cycles at 1100°C

mode stress



average stress



Mode stress - peak of stress histogram

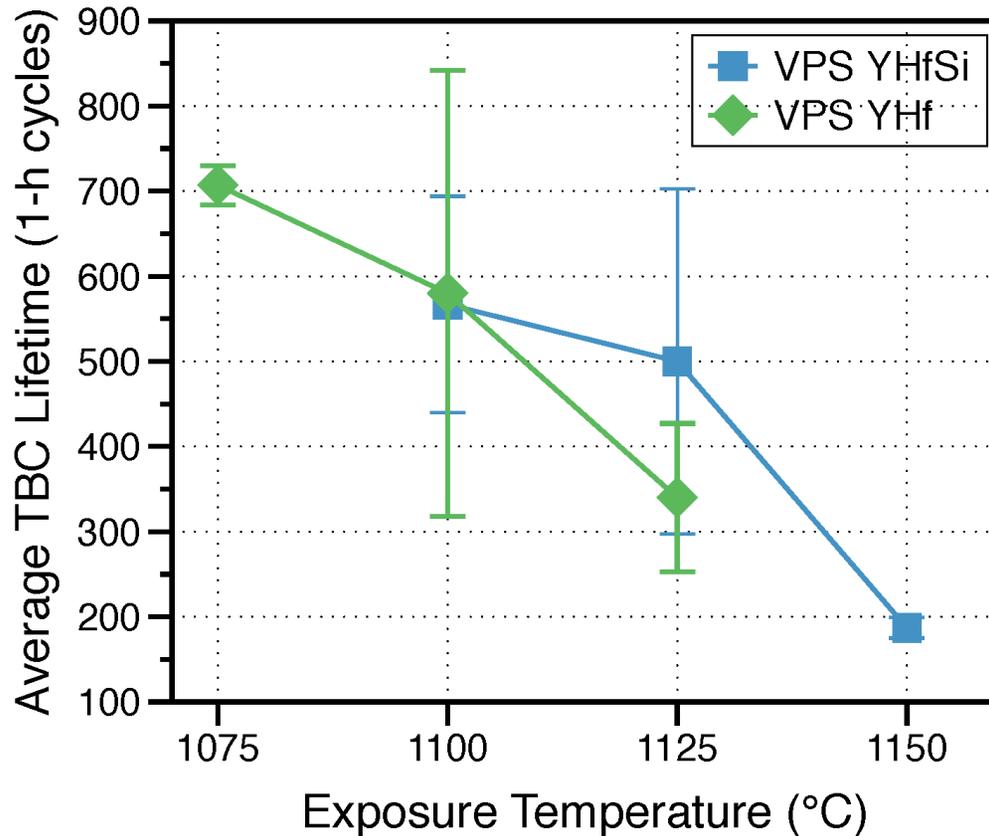
Higher residual stress with VPS NiCoCrAlYHfSiTiB

- average stress: not significant difference

- did not result in increased lifetime

Temperature showed lifetime effect

VPS NiCoCrAlYHf±Si/APS YSZ coatings



air+10%H₂O
simulated exhaust
3 samples/group

No change in lifetime w/o Si addition

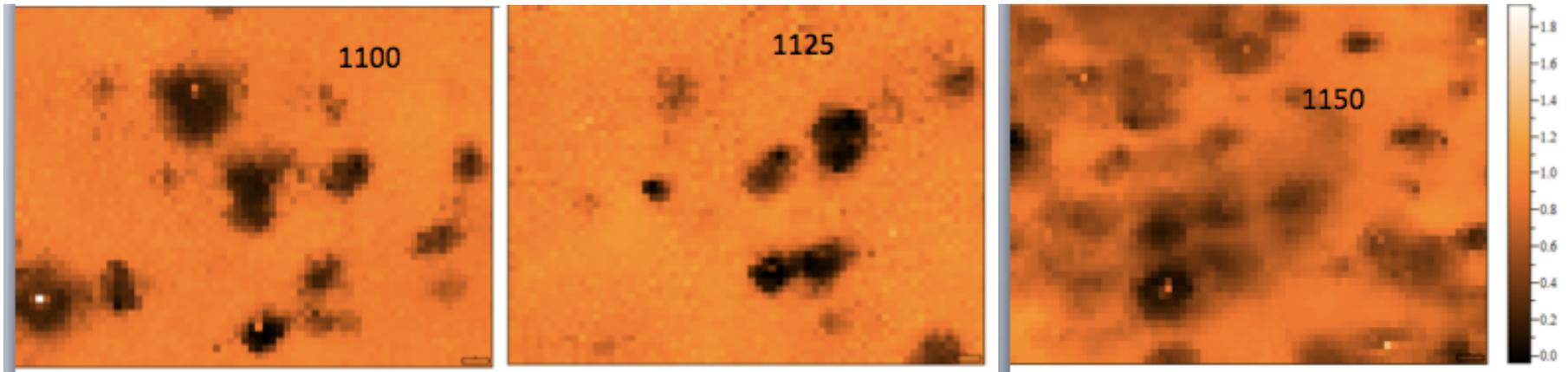
PSLS: lower stress at 1150°C

All 100 1-h cycles in air+10%H₂O

1100°C

1125°C

1150°C

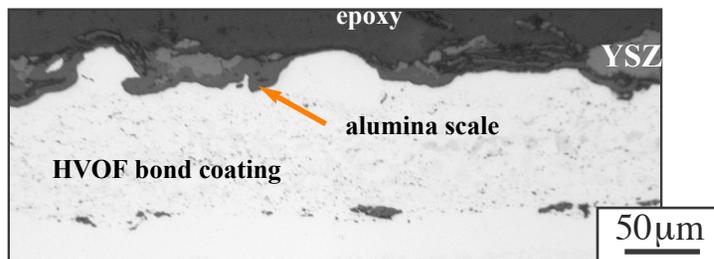


Average lifetime:

567 cycles

500 cycles

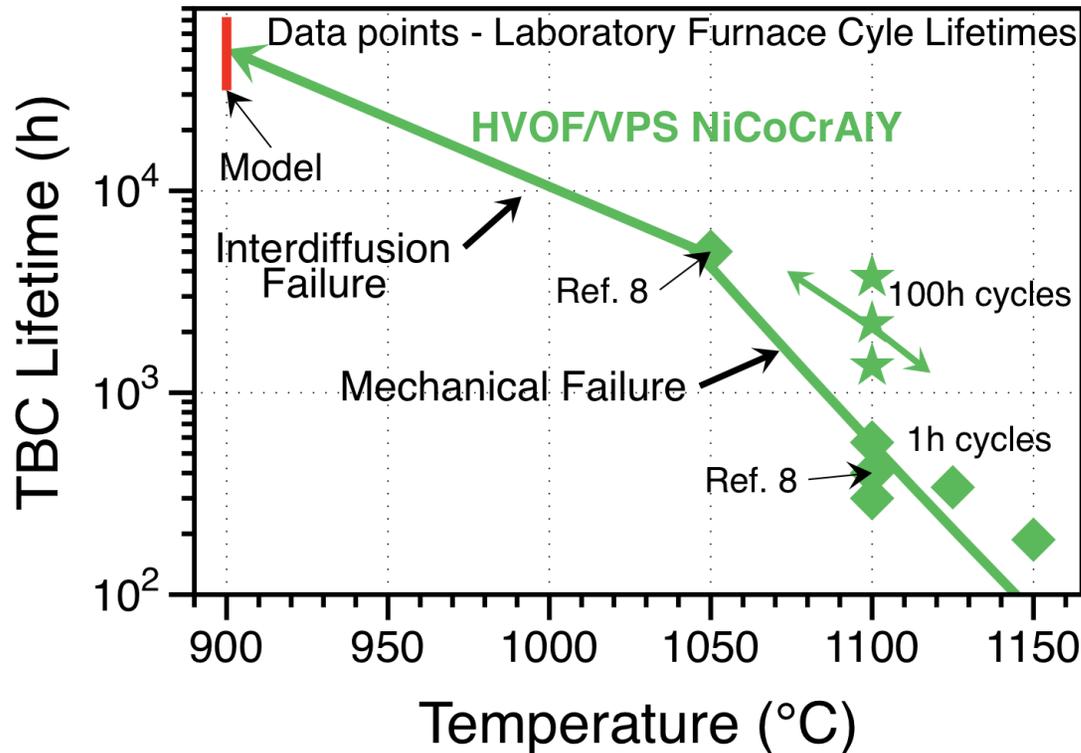
187 cycles



Low stress areas due to asperities in bond coating
- areas growing after 100 cycles at 1150°C

Initial model work focused on 900°C

Model inputs based on 900°C 5-20 kh samples



Failure map from Schütze et al.

Lifetime modeling:

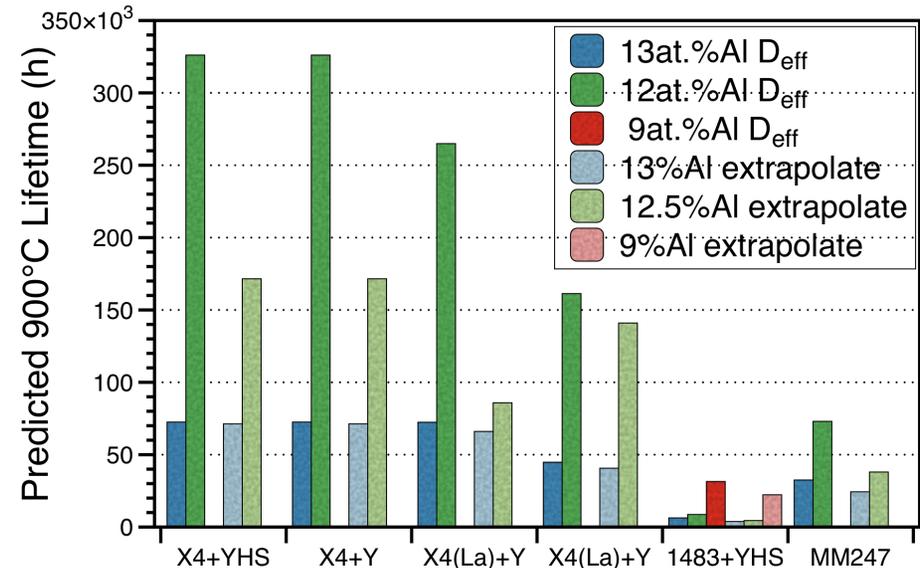
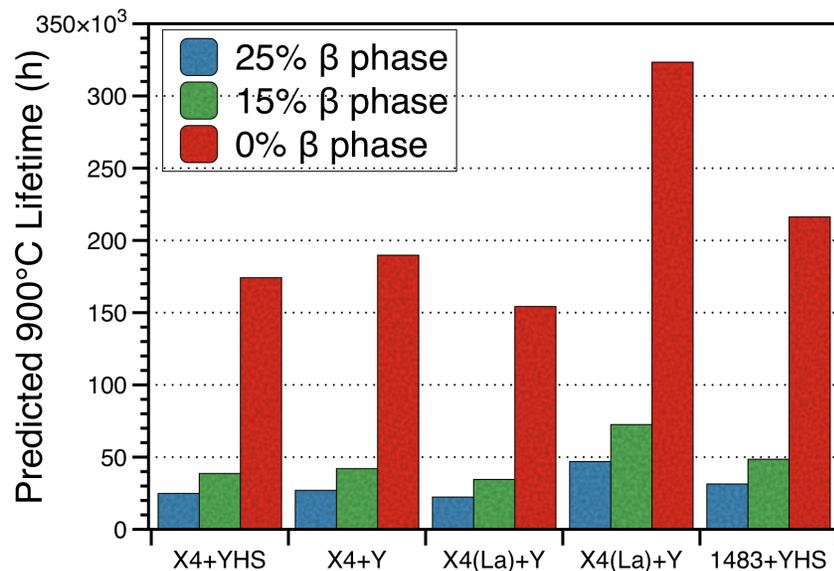
1. Beta depletion
- 2a. NASA COSIM model with D_{eff} from EPMA
- 2b. COSIM model with extrapolated D

Wide range of lifetimes predicted

Highly dependent on parameter inputs at 900°C

β phase depletion

COSIM



HVOF bond coating

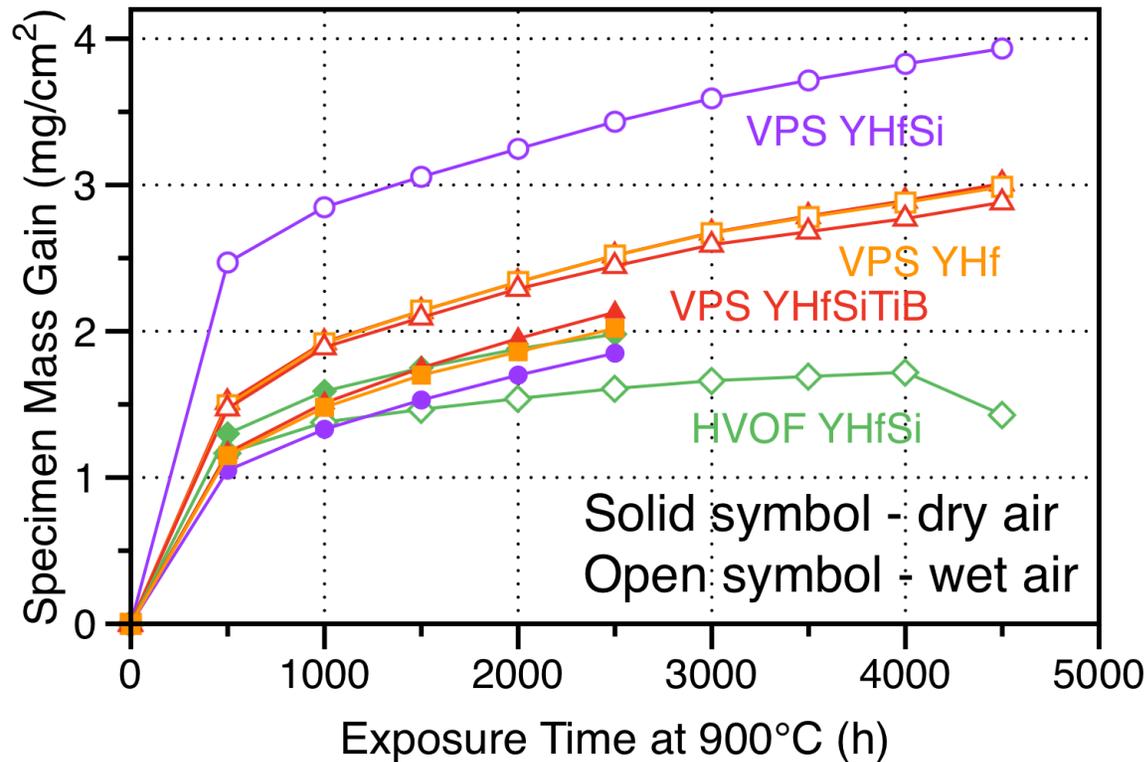
Coating on 1483 (low Al): low life by COSIM

Little effect of YHfSi compared to Y alone

Next phase: collect data between 900°-1100°C

What is effect of H₂O at 900°C?

500-h cycles in wet air and laboratory air



Slight change in rate constants with H₂O
Stop specimens at 5,000h for metallography
- compare change in oxide thickness

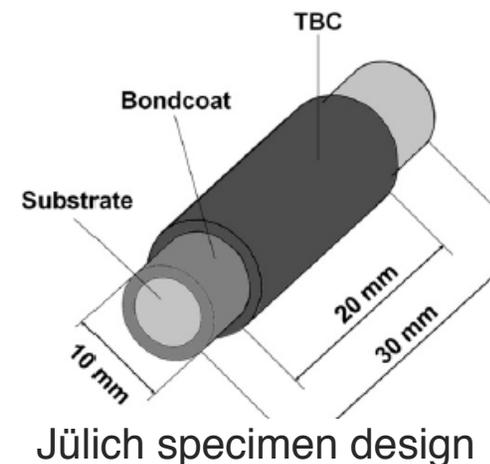
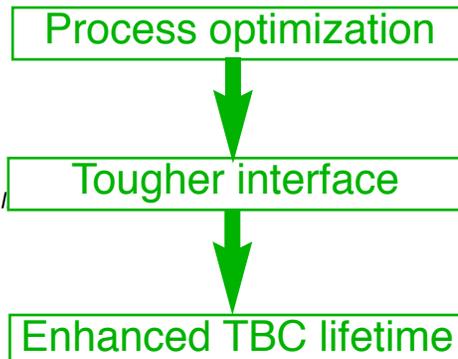
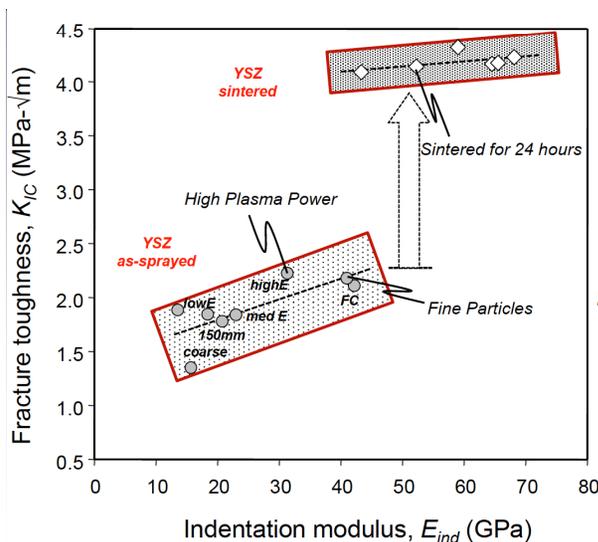
2015 focus areas

1. Use CTSR/ORNL experience for best TBC
2. Get away from testing flat coupons
Cranfield/Jülich coat more complex shapes



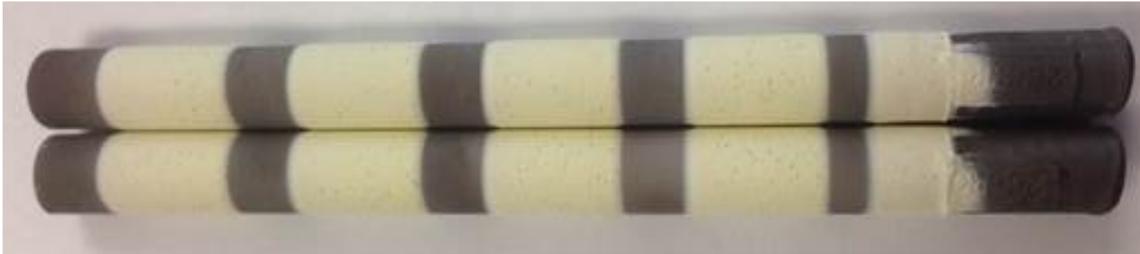
~4 specimens from each rod/tube: 100h cycles
- measure lifetime, alumina stress...

Optimize coating process for rod (CTSR)



Rod specimens begun testing

Wet air at 900°, 1100° and 1150°C



APS YSZ top coating; NiCoCrAlYHfSi APS + HVOF

Exposures:

500-h cycle at 900°C (in progress)

100-h exposure at 1100°C

100-h exposure at 1150°C

run to 5 kh

run to fail

run to fail

APS MCrAlYHfSi bond coating

After 100h at 1100°C



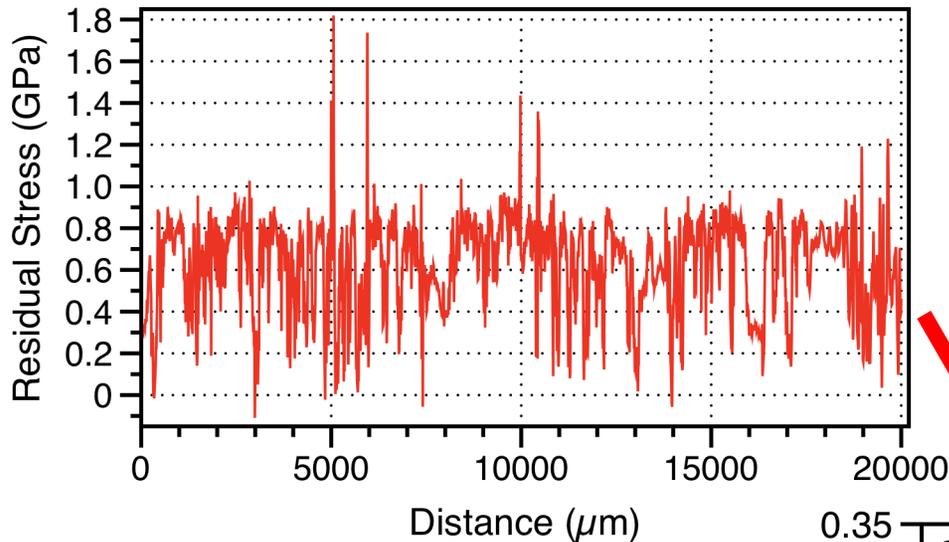
APS MCrAlYHfSi bond coating

Failure after 100h at 1150°C

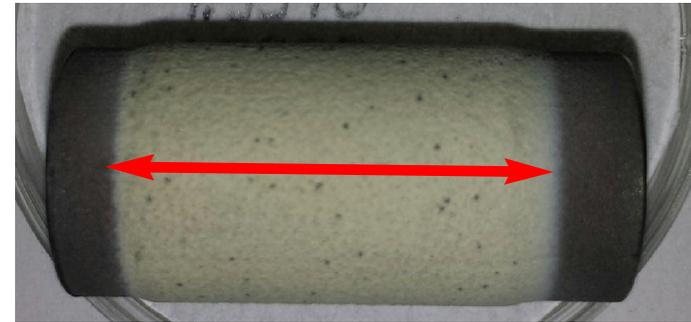


Initial stress measurements

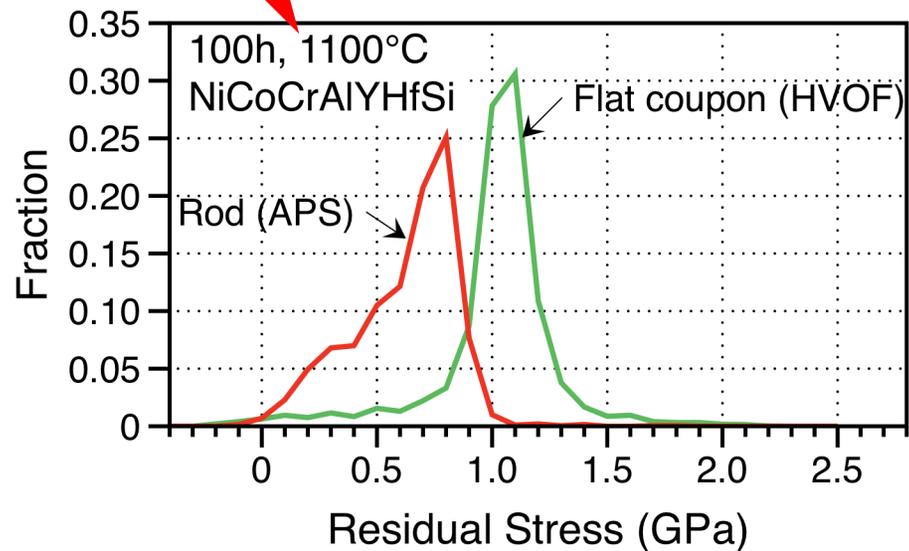
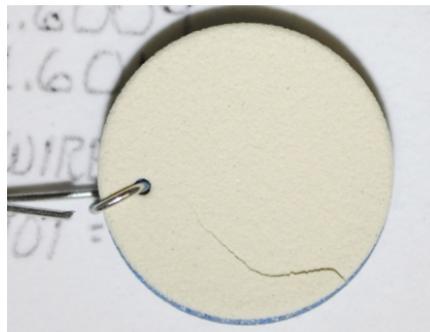
One 100-h cycle at 1100°C in air+10%H₂O



PSLS measurement along length of rod



Next measurement at 500 h



Lower mode stress in APS rod specimen after 1 cycle

TBC Summary

2014-15 work concluded multi-year investigation:

Effect of environment (H_2O , CO_2 ...)

Effect of bond coating chemistry

Effect of superalloy chemistry

Effect of cycle frequency

Lifetime at 900°C (NASA COSIM model)

Furnace cycle testing on 16mm disk specimens

1. 100h cycles at 1100°C failed by interdiffusion
2. Suspect early edge failures on coupons

2015 project

coating 12.5mm diameter rod specimens (247)

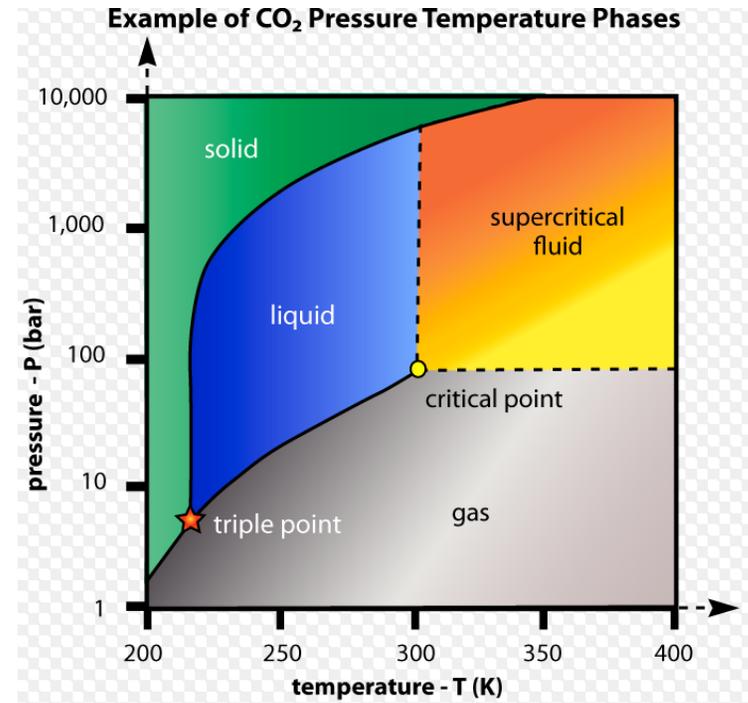
initial exposures started at 900° , 1100° , 1150°C

Focus on lifetime modeling at $<1100^\circ\text{C}$

Why use supercritical CO₂?

Potential supercritical CO₂ (sCO₂) advantages:

- no phase changes
- high efficiency
- more compact turbine
- short heat up
- less complex
- lower cost (?)



Direct- and indirect-fired sCO₂ Brayton cycles for:

- fossil energy (coal or natural gas) **FY13- ?**
- concentrated solar power **FY16-FY18**
- nuclear (paired with sodium for safety)
- waste heat recovery/bottoming cycle

Many possible applications



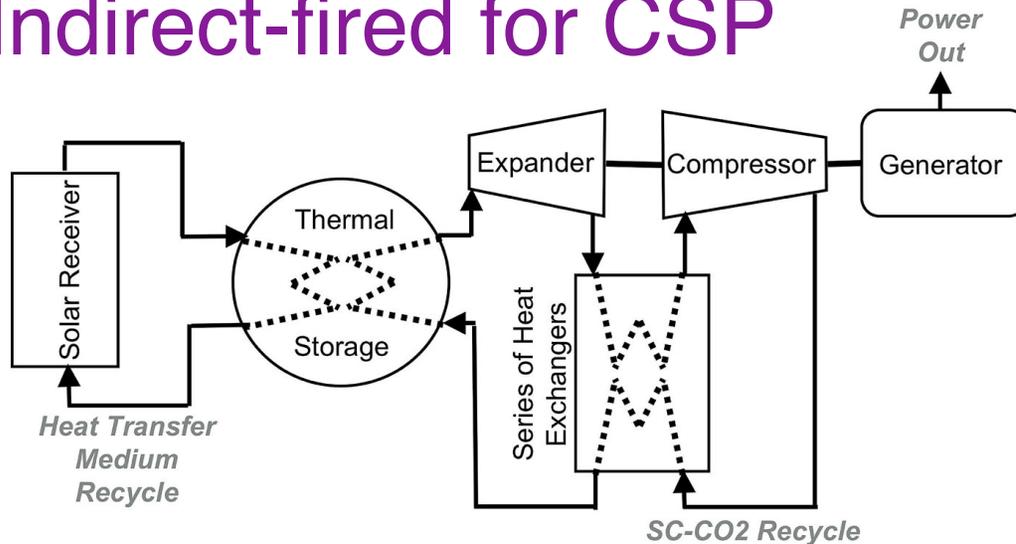
Smaller fossil fuel turbines



7MW Echogen (WHR)

Indirect & direct-fired systems of interest

Indirect-fired for CSP



Closed loop of relatively pure CO₂
 - primary HX (>700°C)
 - recuperators (<600°C)

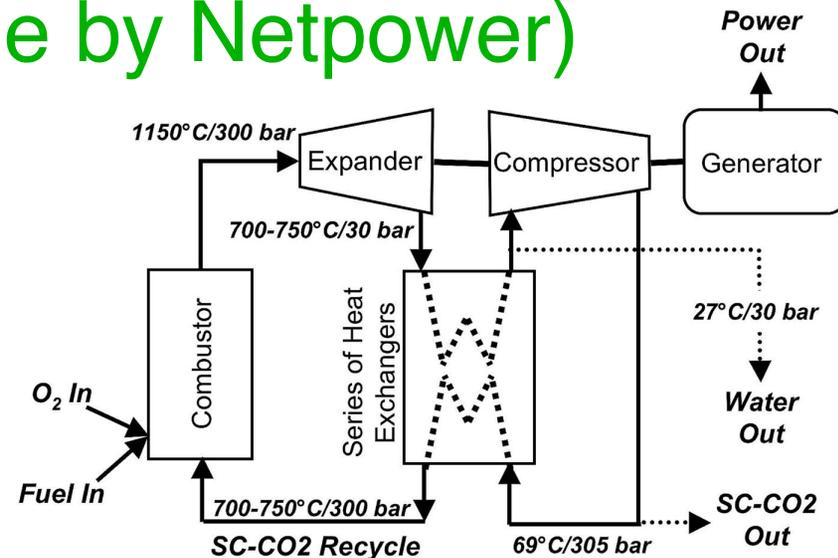
Also, waste heat recovery, bottoming cycle

Direct-fired (Allam cycle by Netpower)

In: natural gas + O₂

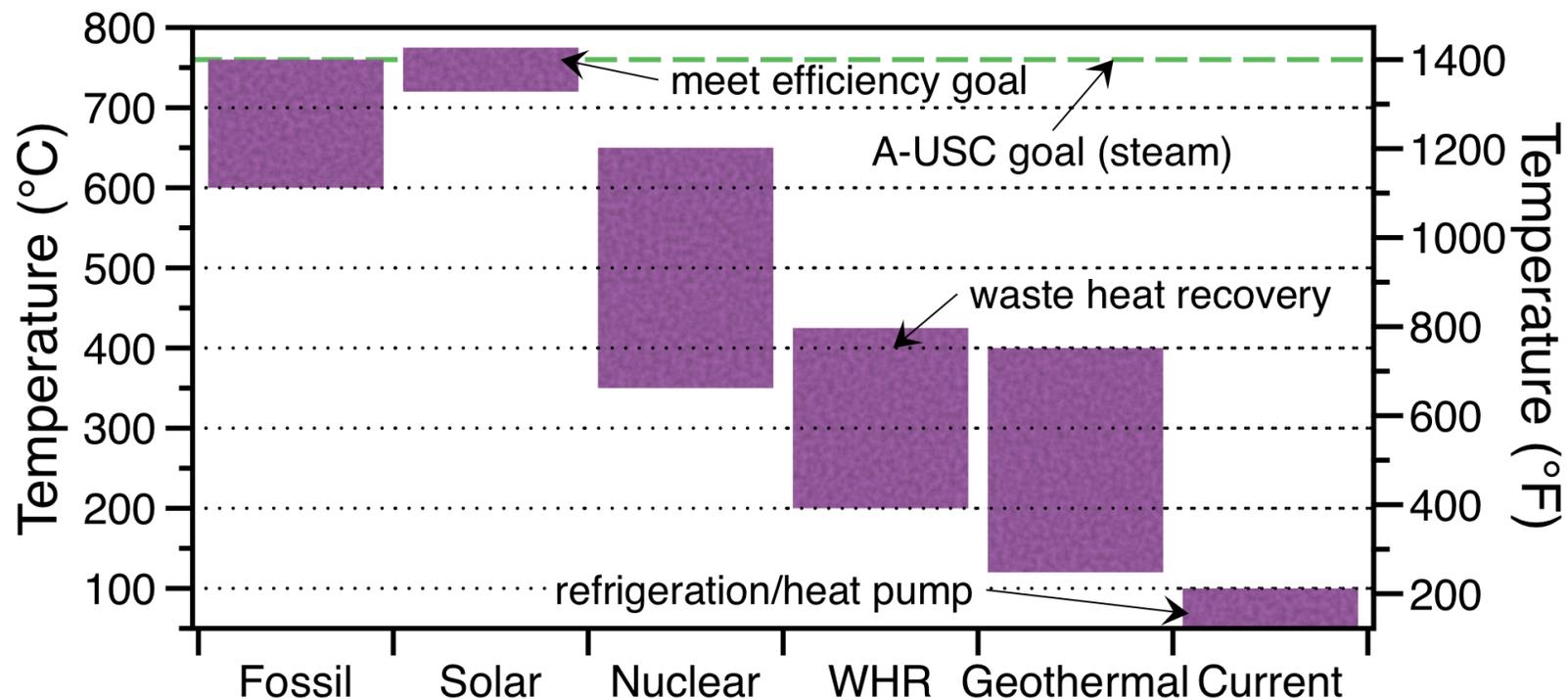
Impurities: ~10% H₂O
 ~1% O₂, CH₄?

Out: CO₂ for EOR
 (enhanced oil recovery)



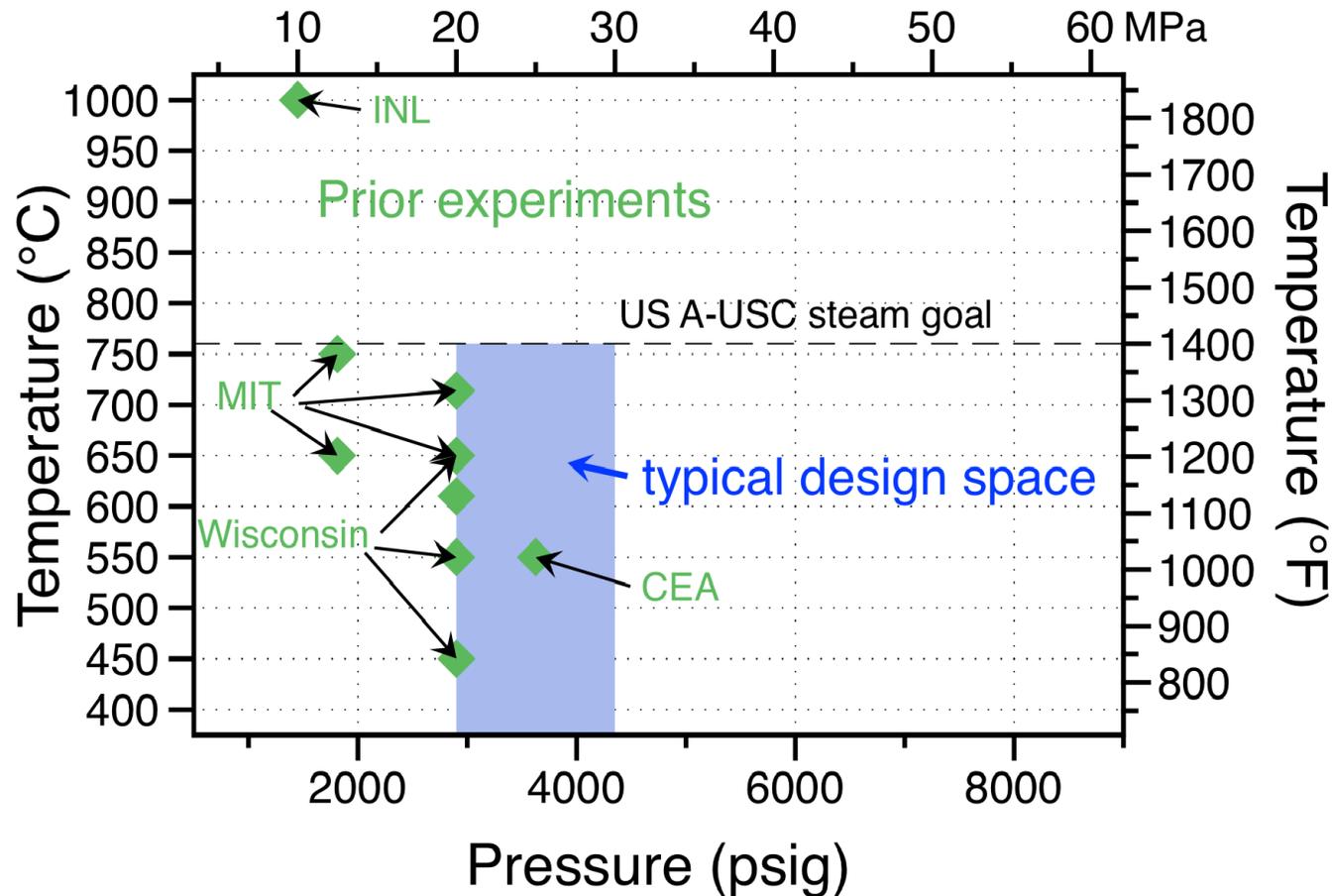
Different temperature targets

- Uncertainty about ranges for sCO₂ applications
- Fossil energy interest for power generation
coal/natural gas: replace steam with closed cycle



Relatively little prior sCO₂ work

Especially at >650°C and 300 bar



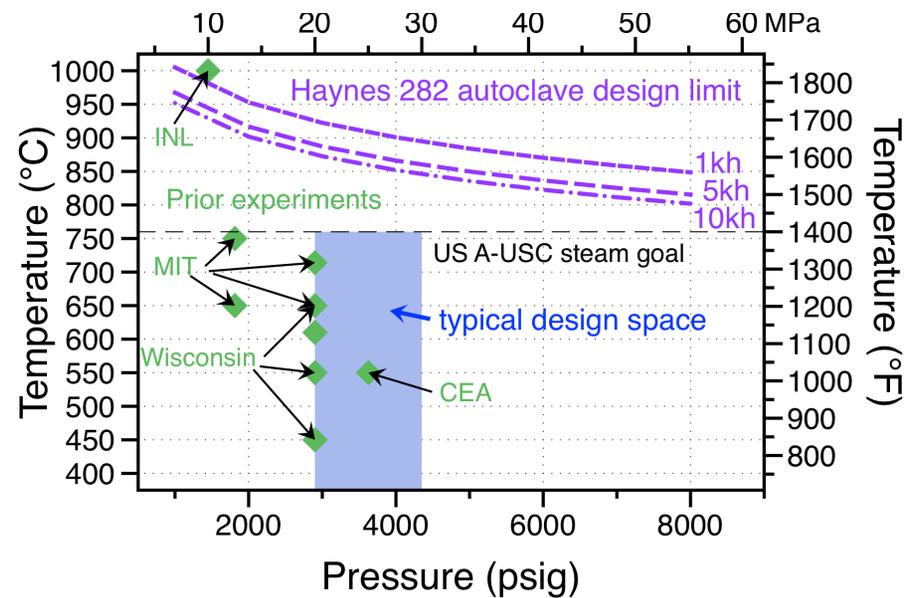
Several groups active in the past 10 years

U. Wisconsin group has published the most results

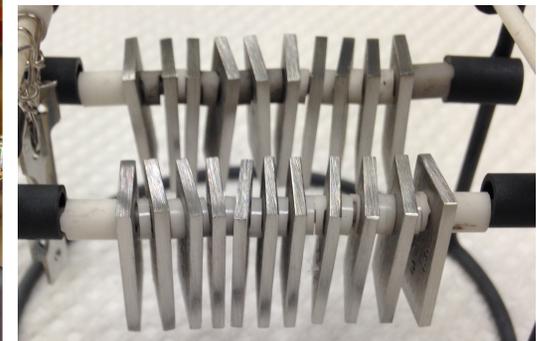
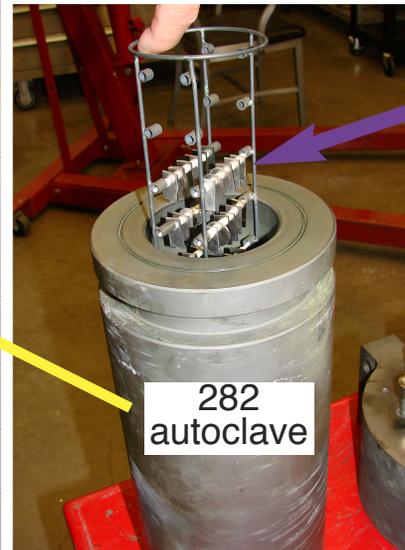
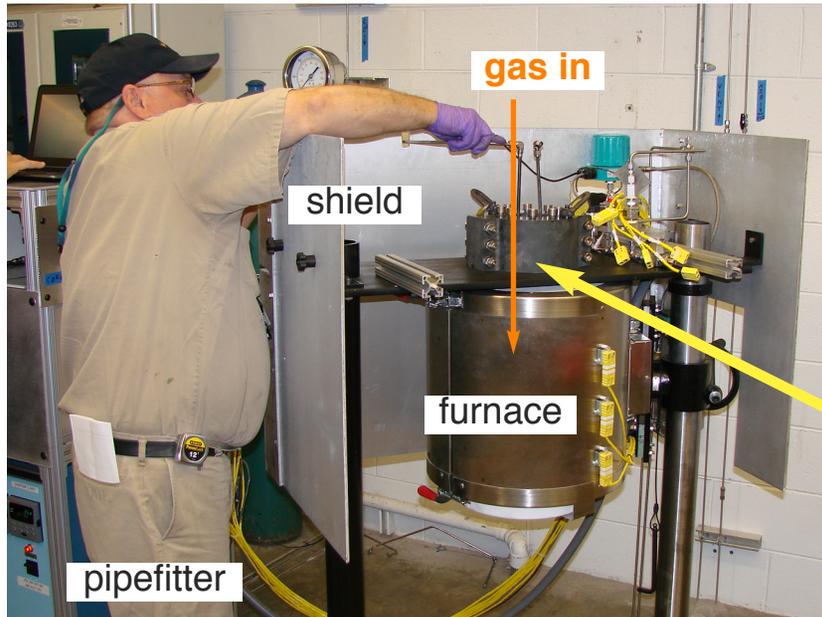
Temperature/pressure limited by autoclave design

ORNL sCO₂ rig finished in 2014

- ORNL design team: 100+ years of experience
- Haynes 282 autoclave
152mm (6") dia.
1 ml/min flow

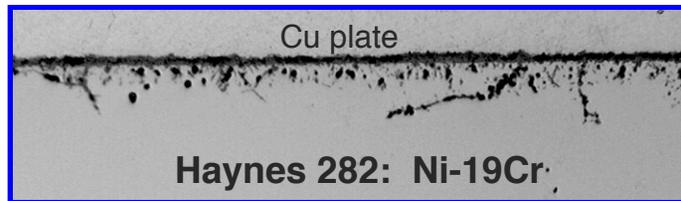


ORNL sCO₂ rig:



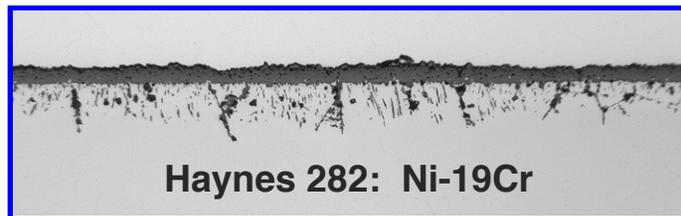
Why worry about 740/282?

5-10kh at 800°C still form thin reaction product in air



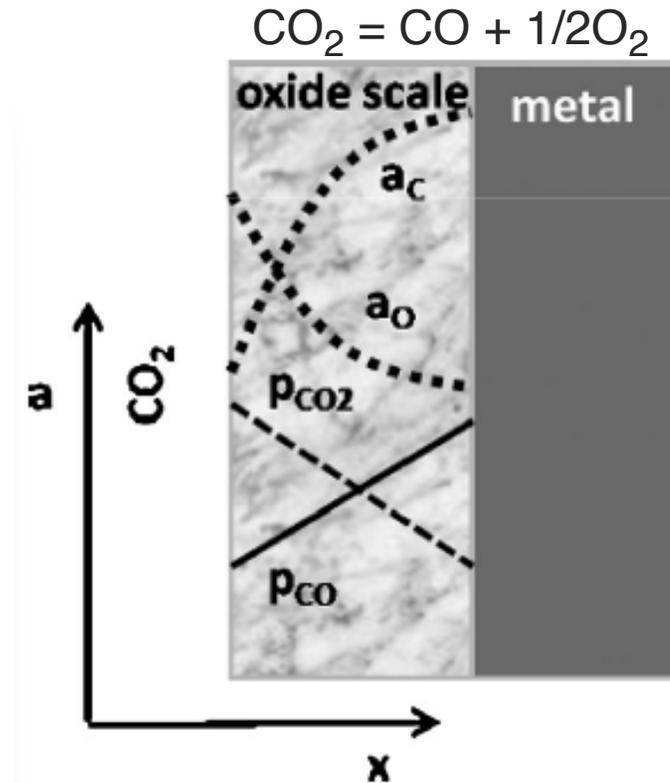
800°C, 5,000h in wet air

10 μ m



800°C, 10,000h in dry air
both exposures: 500h cycles

Al+Ti internally oxidize beneath
 Cr_2O_3 oxide scale



from Young, et al. 2011

General: C activity (a_c) relatively low, favors oxidation

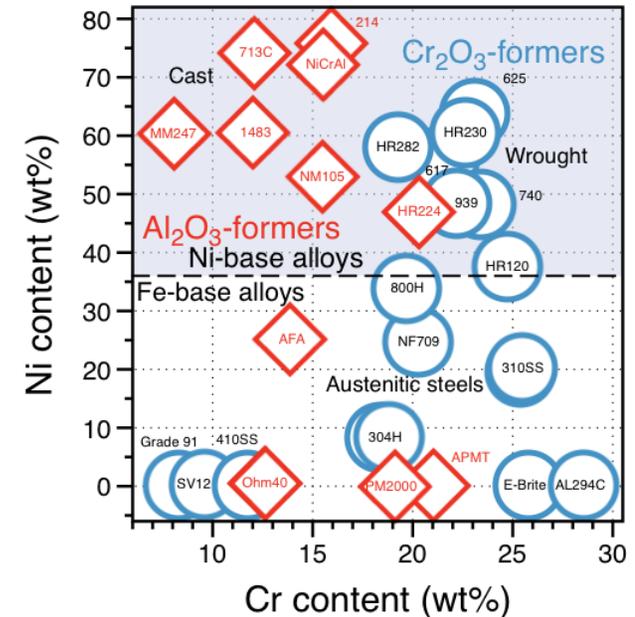
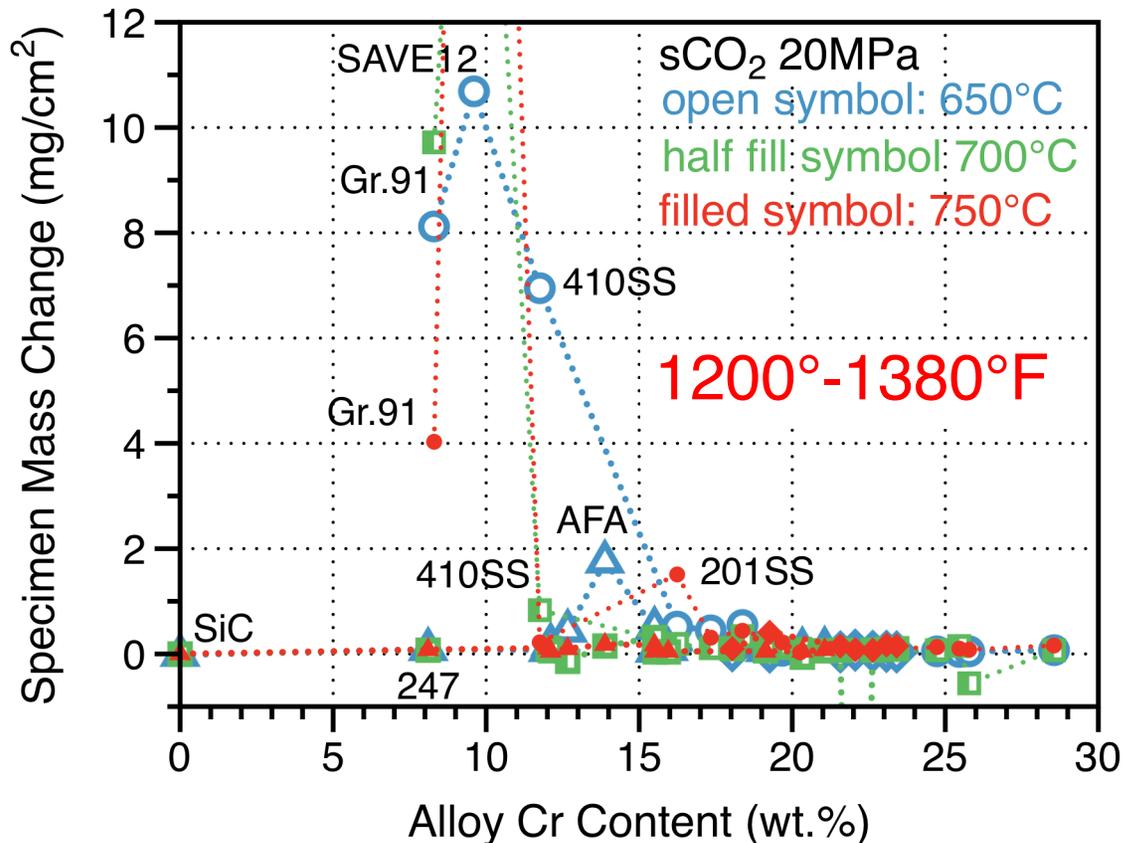
McCoy 1965: 18Cr-8Ni steel internally carburized in 1bar CO_2

High a_c predicted - what about NiCr in $s\text{CO}_2$ after 10 years?

Many alloys exposed 650°-750°C

~30 alloy coupons exposed 500h at 200 bar (20 MPa)

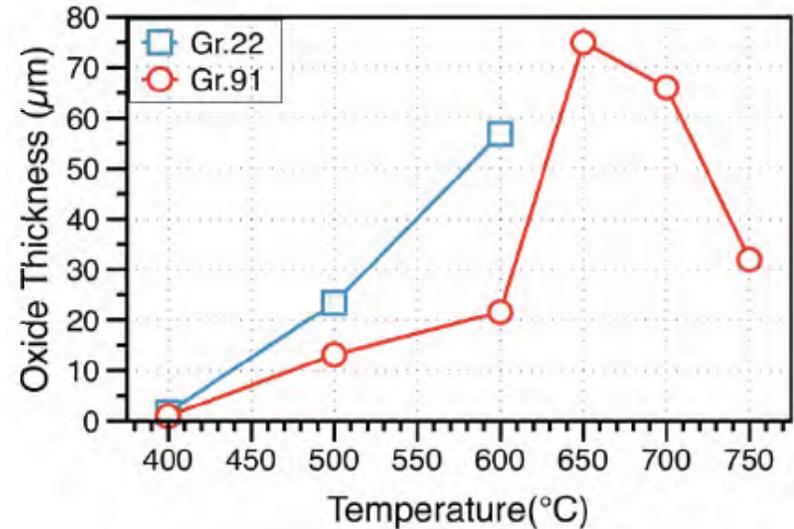
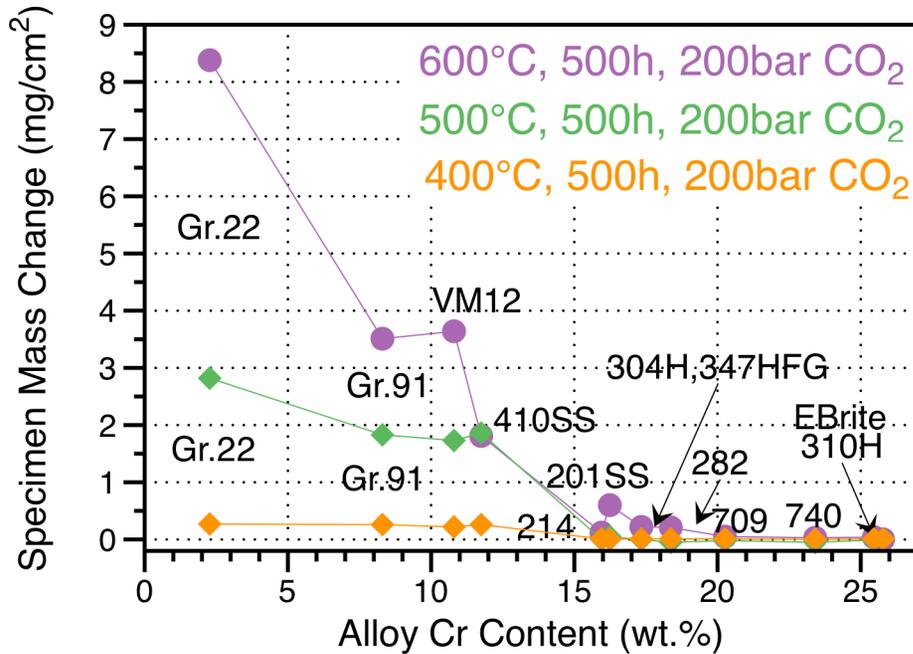
Most Ni-base alloys did fine



Fe-9Cr-1Mo (Gr91):
- similar to other studies

BUT, only 500h

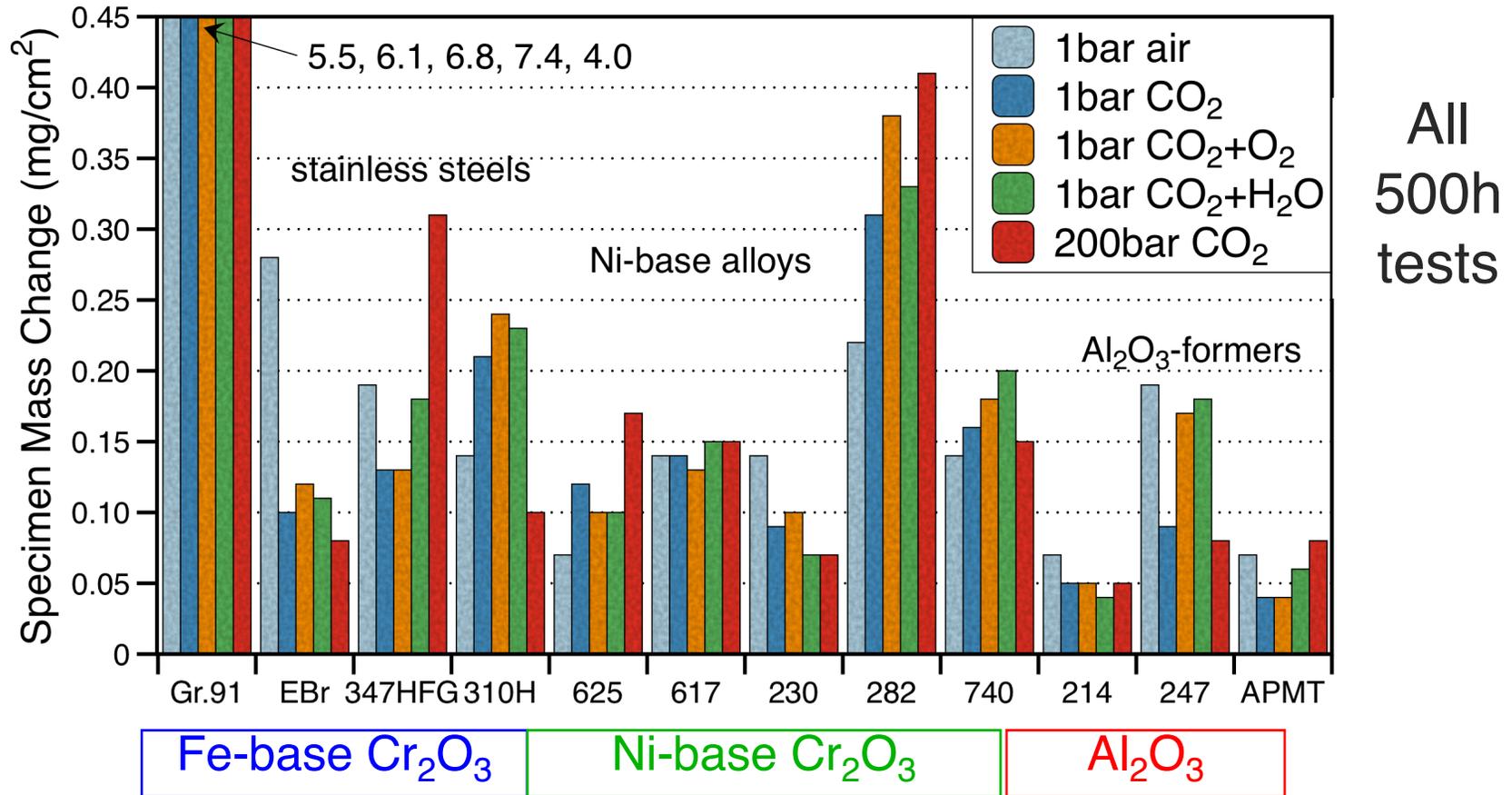
Steels exposed at 400°-600°C 500h exposures in 20 MPa CO₂



Industry interested in where
low-cost alloys can be used

Baseline created at 750°C 1bar

Core 12 alloys, 1bar: dry air, CO₂, CO₂+10%H₂O

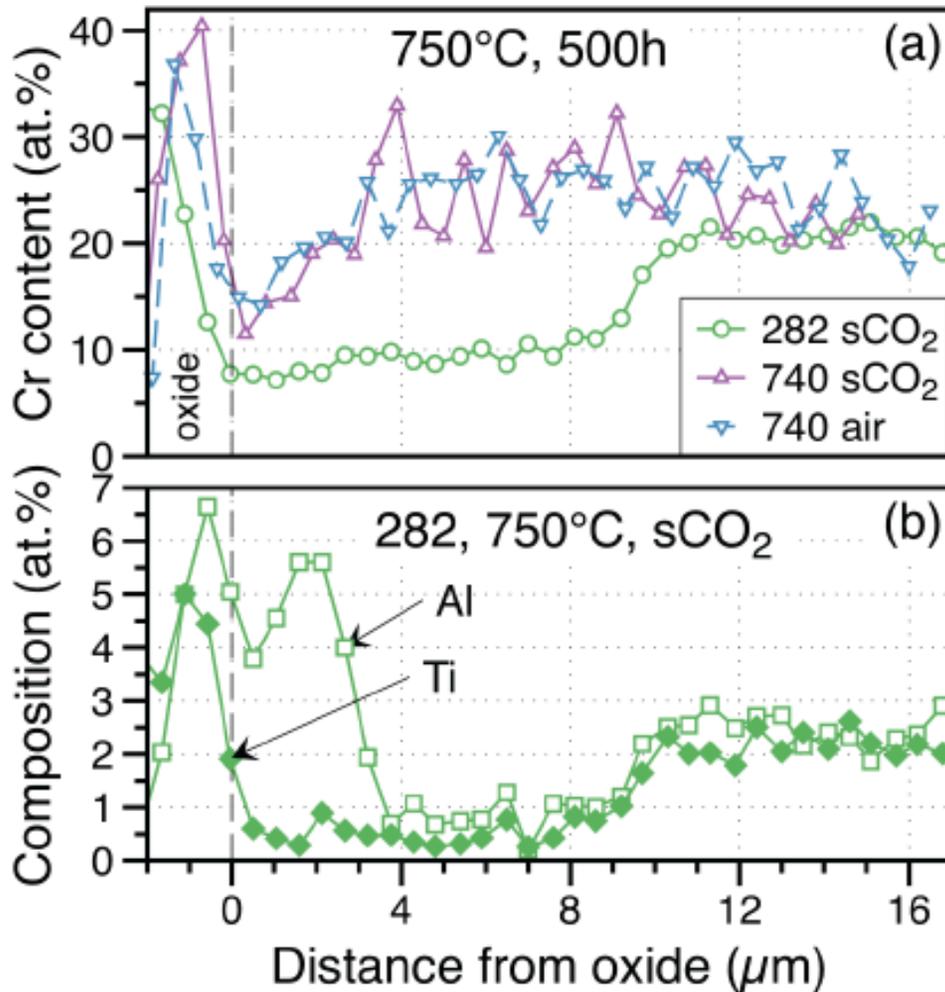


0.1 mg/cm² ~ 0.5 μm surface oxide
 10 mg/cm² ~ 50 μm (2 mils)

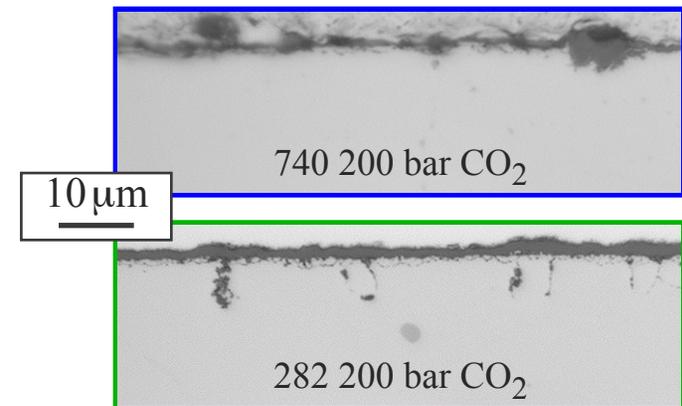
Consistently higher mass gains for alloy 282

282 deeper Cr depletion than 740

EPMA depth profiles beneath scale at 750°C



1 bar air vs. 200bar CO₂



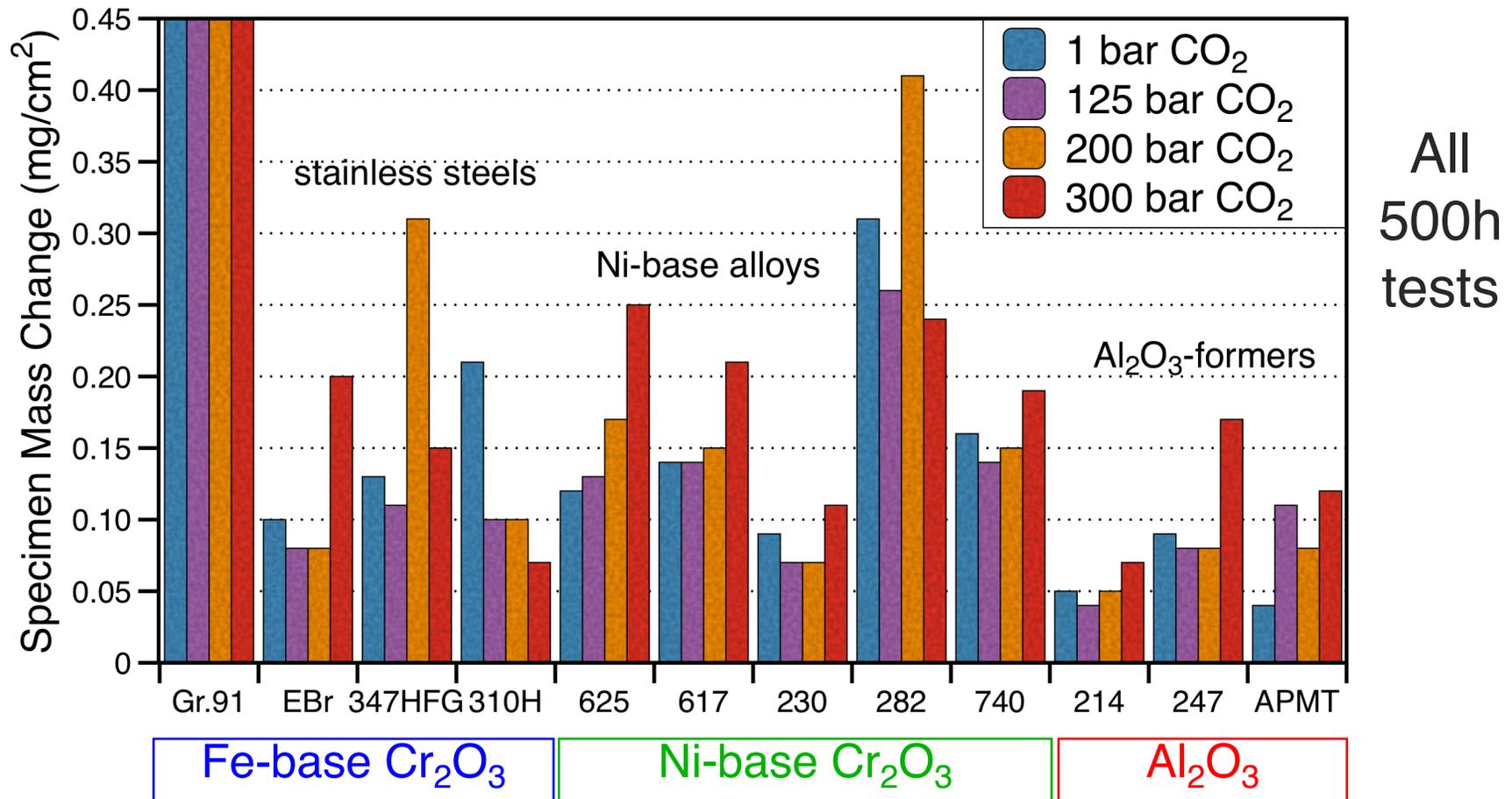
282: 58Ni-19Cr-10Co-8Mo-1.5Al-2.2Ti

740: 49Ni-24.6Cr-20Co-0.5Mo-1.3Al-1.5Ti

Little effect of pressure observed

500h exposures at 750°C

Core group of 12 alloys evaluated



0.1 mg/cm² ~ 0.5 μm surface oxide
10 mg/cm² ~ 50 μm (2 mils)

Typical Fe-rich oxide on Gr.91

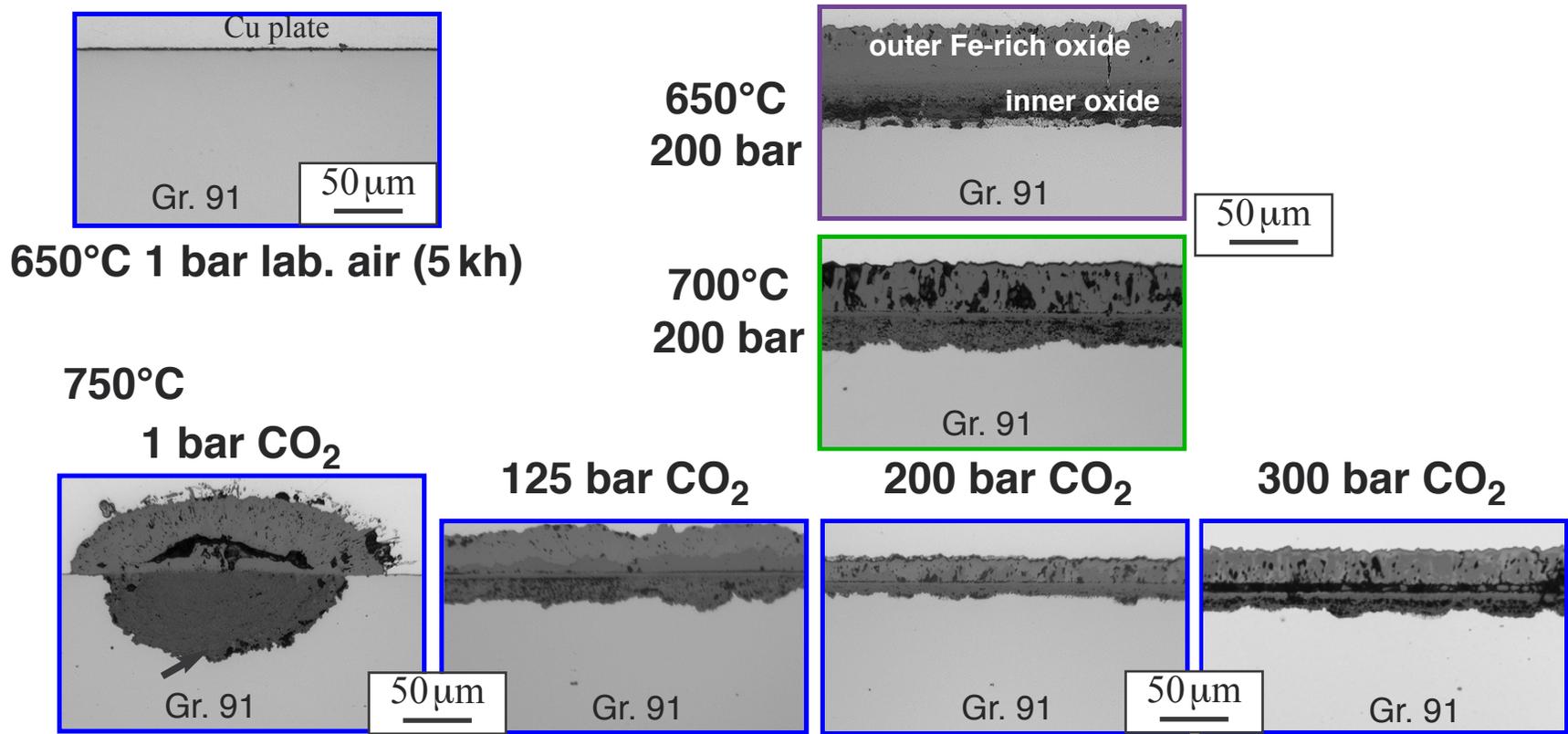
However, inner/outer ratio appears to change with P

Outer $\text{Fe}_2\text{O}_3/\text{Fe}_3\text{O}_4$ layer

Inner $(\text{Fe,Cr})_3\text{O}_4$ layer

Grade 91: Fe-9Cr-1Mo

Some thin-protective Cr-rich scale at 1 bar

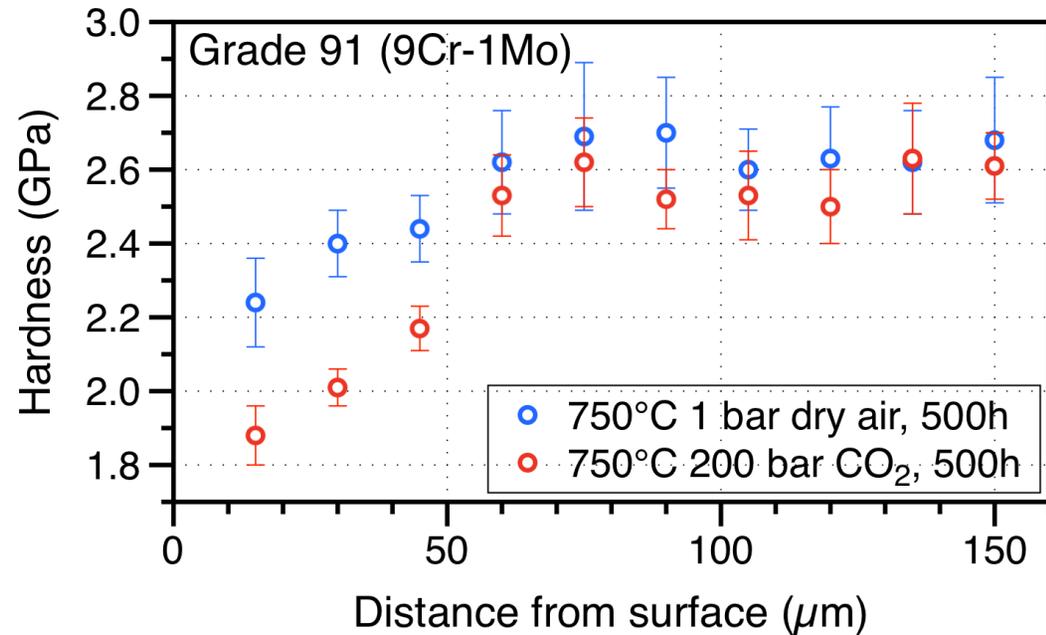


light microscopy of polished cross-sections

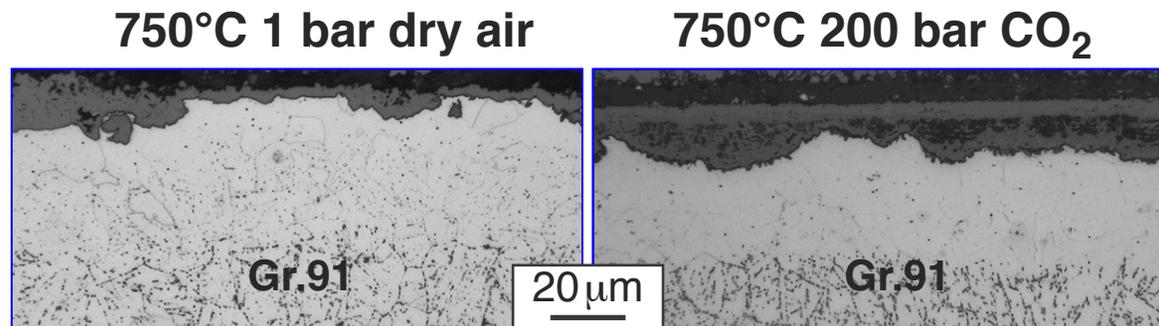
Looking for internal carburization

High a_c predicted, McCoy (1965) observed in 1 bar CO_2

microhardness of Gr.91



oxalic acid etch

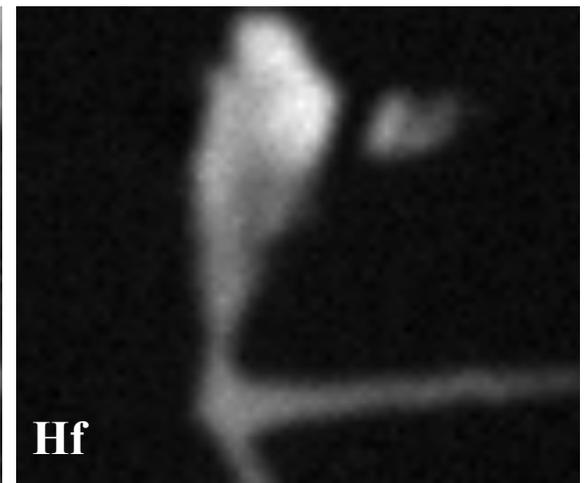
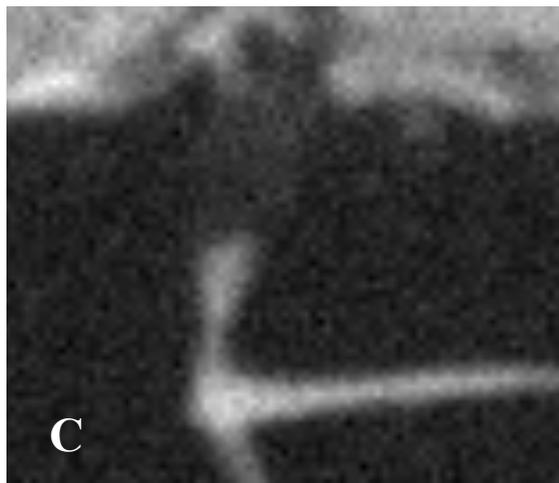
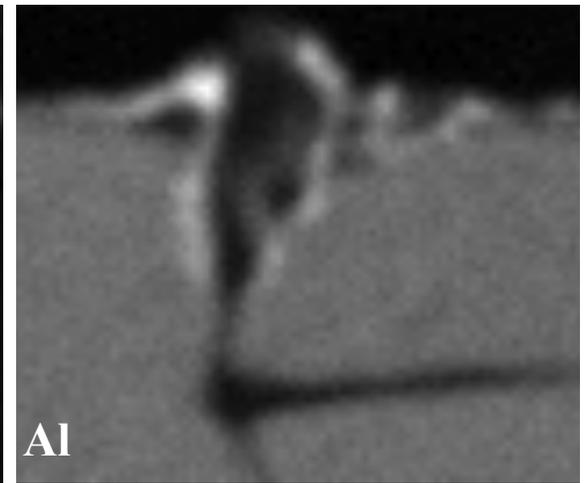
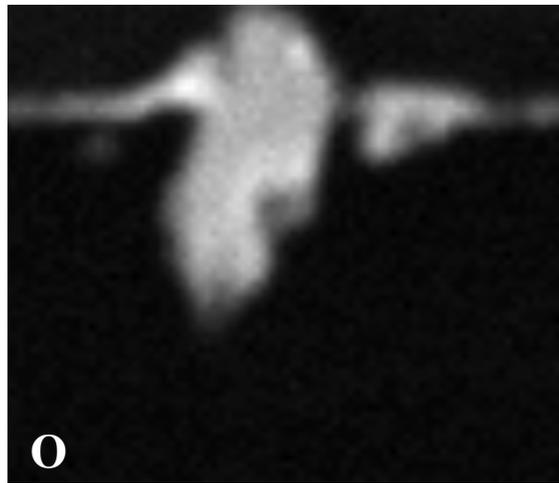
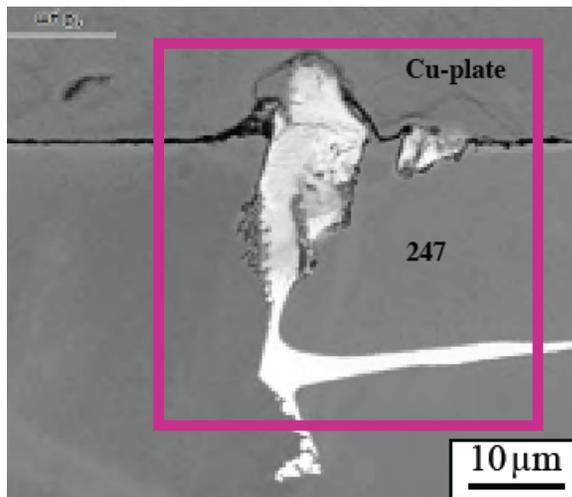


Did Cr depletion dissolve the carbides at 750°C?

Hf-rich carbide oxidized in sCO₂

EPMA: carbide transformed to oxide at 750°C 20MPa

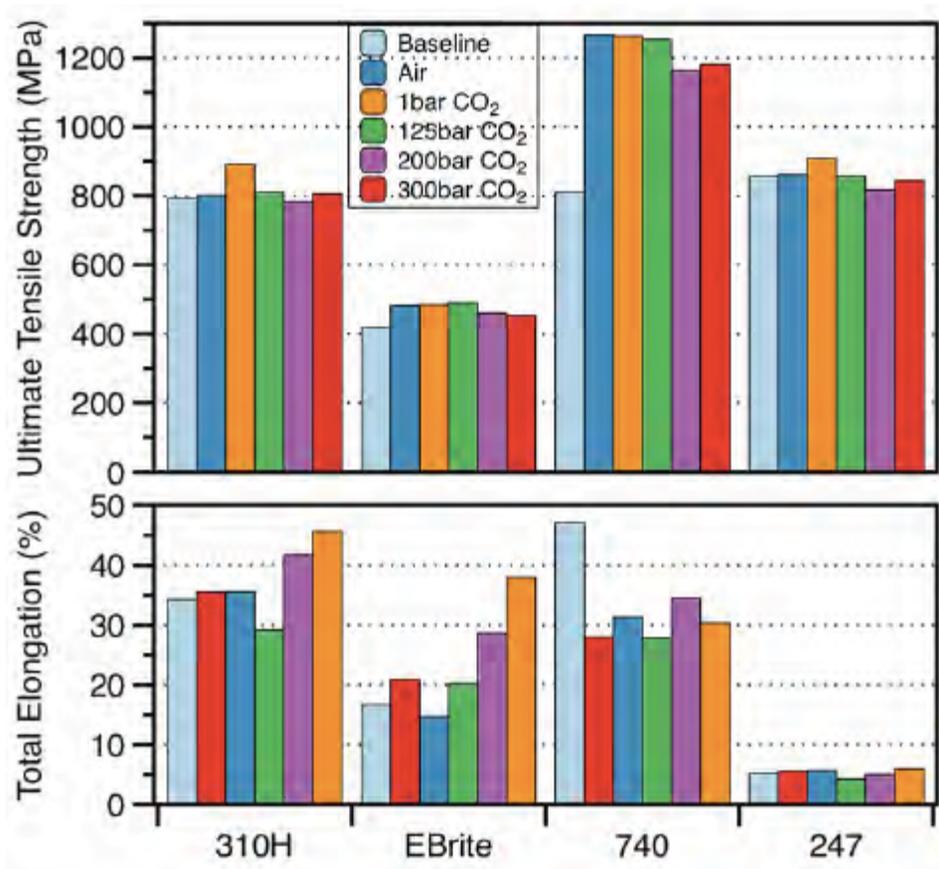
247: Ni-8Cr-6Al-10Co-10W-3Ta-1.4Hf



750°C: initial tensile experiments showed little effect of sCO₂

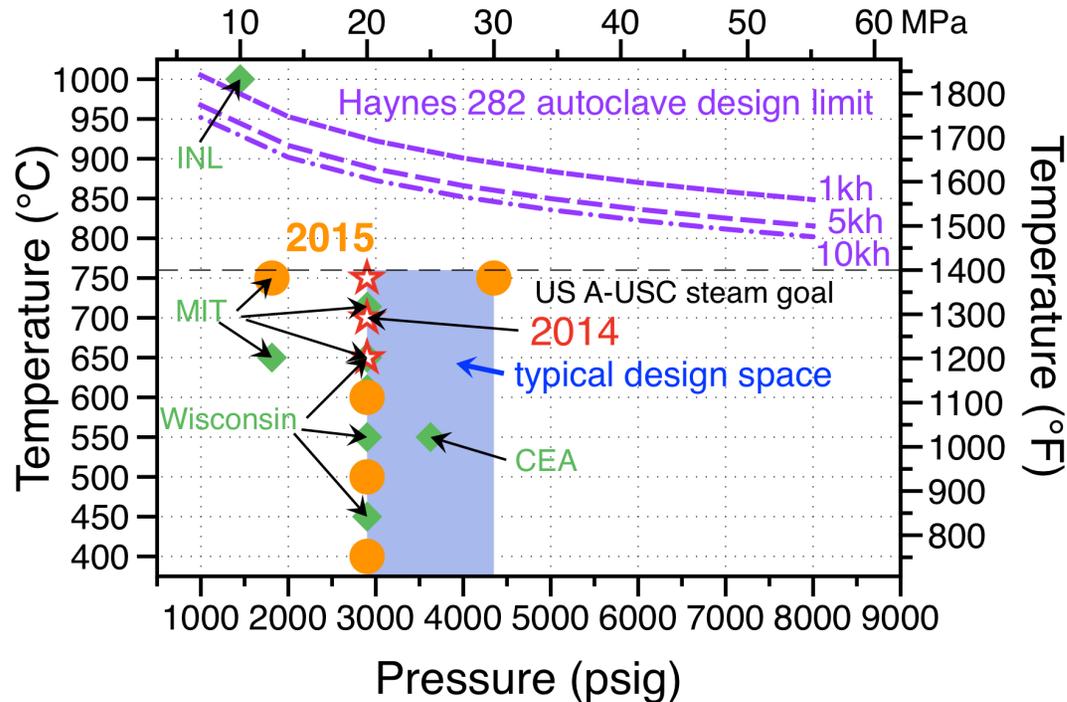
25mm tensile bars exposed at each condition

Tensile test at room temperature: 10⁻³/s strain rate



sCO₂ Summary

Completed five 500h sCO₂ tests in 2015:



- Wide range of alloys exposed (12)
- Surprisingly little effect of pressure
- Little effect on 25°C tensile properties
for alloys 740, 310HCbN, E-Brite & 247
- Fe-9Cr formed thick scales, similar to other studies

Future work

TBC project:

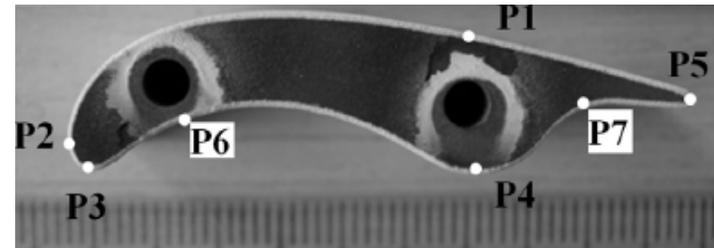
Long-term testing of rod specimens

Develop a path to deployment

- solicit OEM input
- next step: burner rig testing? provide samples
- address issues identified

Specimens with more complex geometry

Cranfield design:



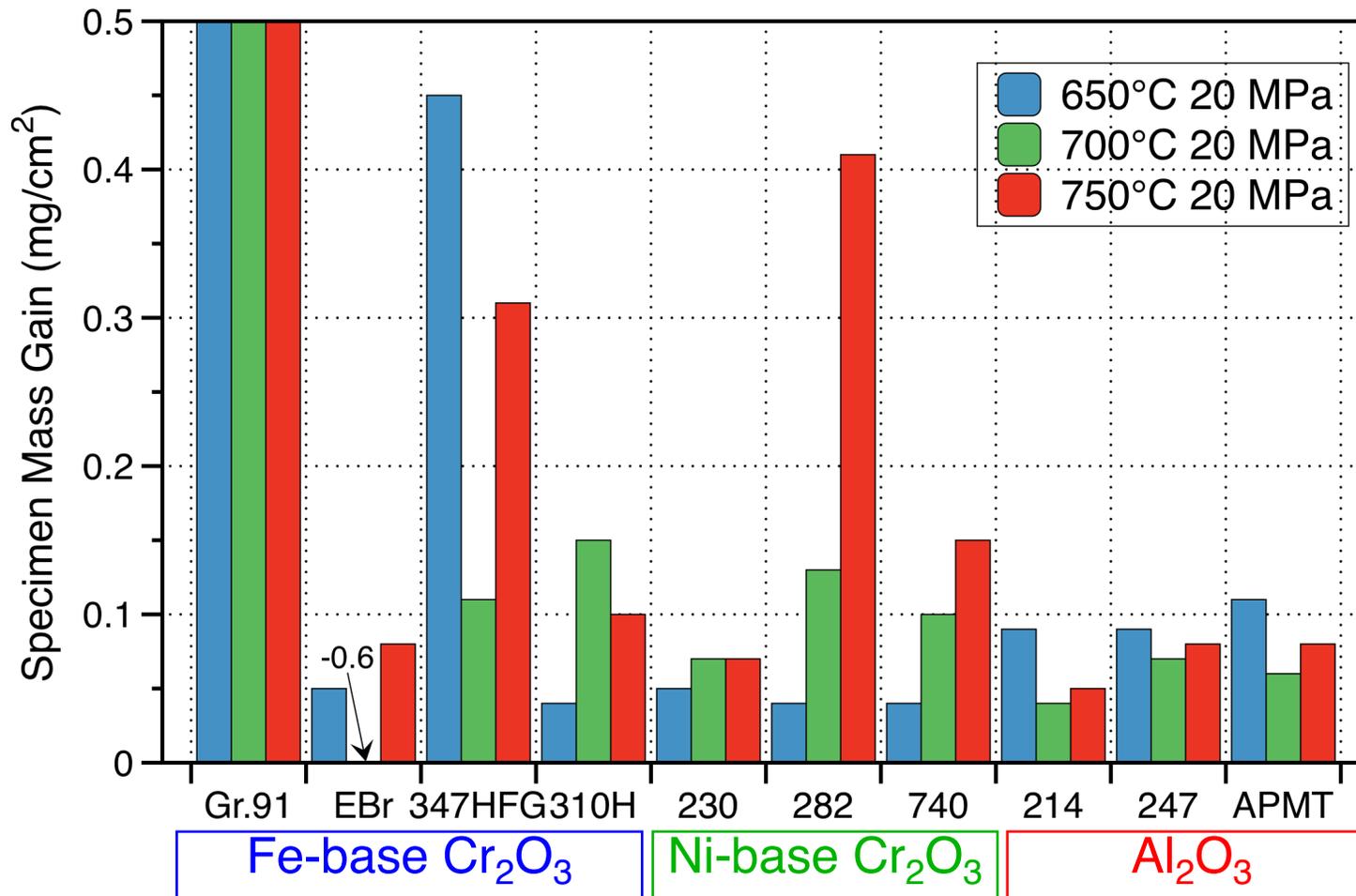
Supercritical CO₂:

- 2nd autoclave for impurity effects (measure in + out)
- developing on-line O₂, H₂O detector at pressure
- long-term exposures at 750°C/300 bar (30 MPa)

backup slides

1st: 20MPa runs, most low gains

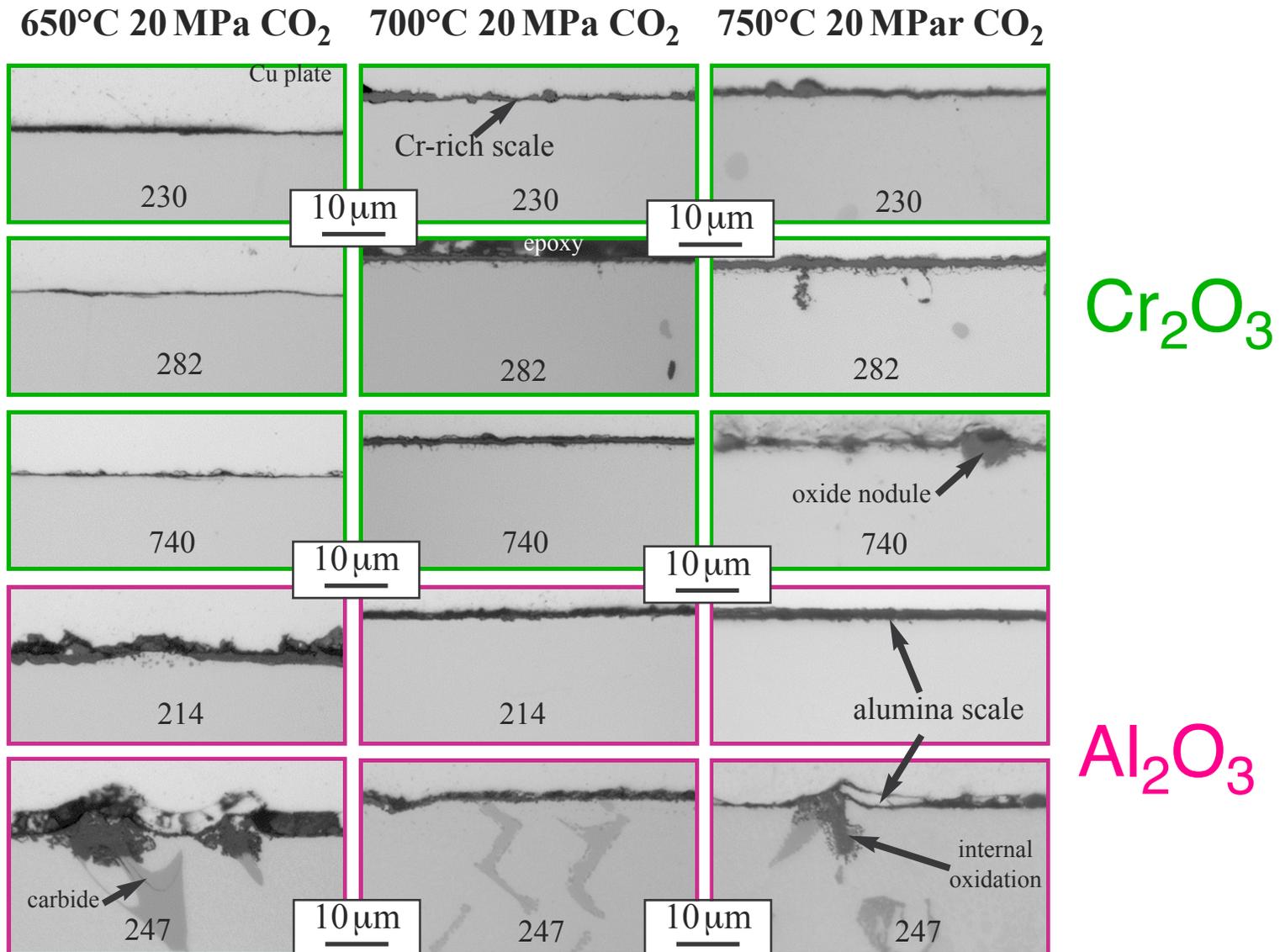
10 representative alloys were focus of metallography



0.1 mg/cm² ~ 0.5 μm surface oxide
10 mg/cm² ~ 50 μm (2 mils)

Ni-base alloys: thin scales

All thin Cr-rich or Al-rich scales in 20 MPa sCO₂



Thoughts

More characterization needed of current results
Better understand some unusual results

Concern:

Degradation by C penetration through Cr_2O_3

H. McCoy 1965 at 1bar

D. J. Young et al. (2011-2014) at 1bar

Need to evaluate longer times + ex-situ ductility

Al_2O_3 thought to be better barrier to C ingress

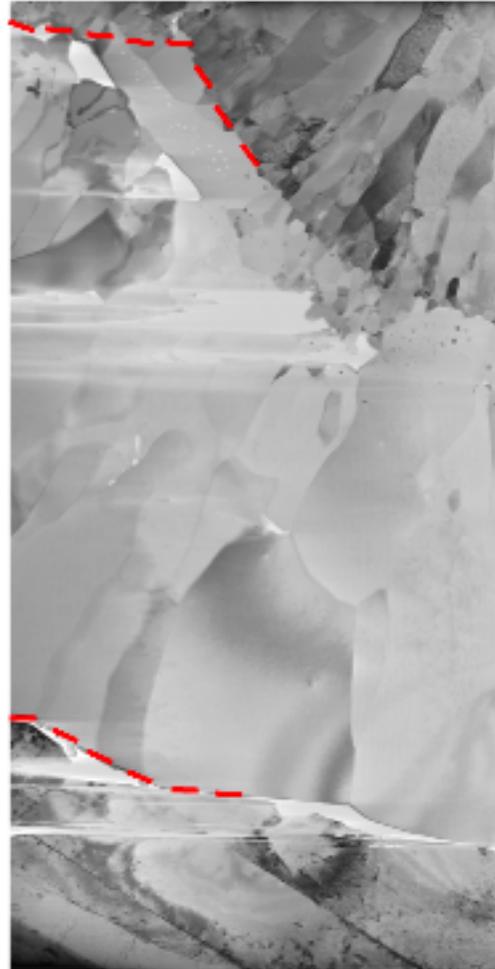
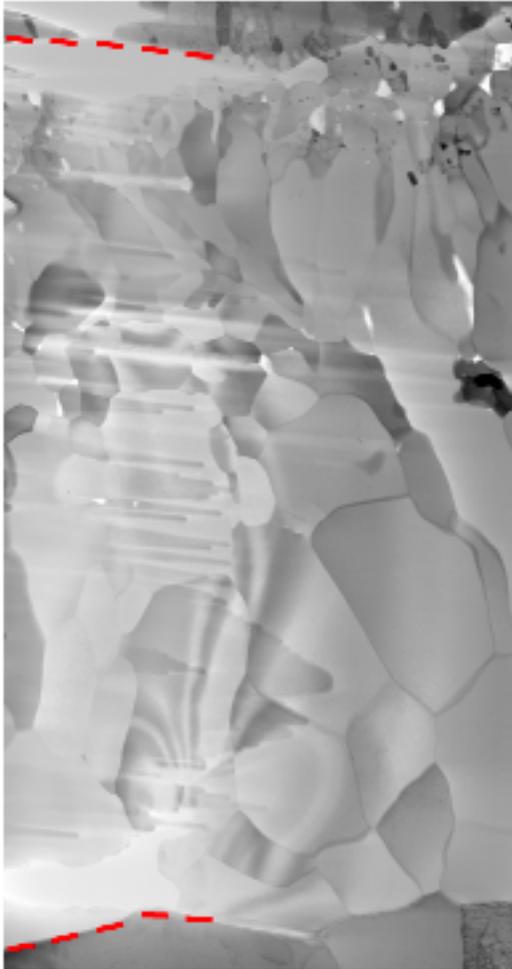
Pre-oxidation may assist in Al_2O_3 formation

Al-containing alloys can be difficult to fabricate

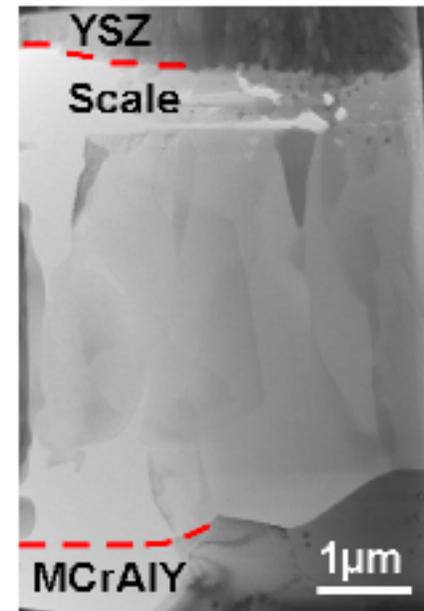
Possible to make FIB/TEM specimens but not as “pretty” as model systems

X4

1483

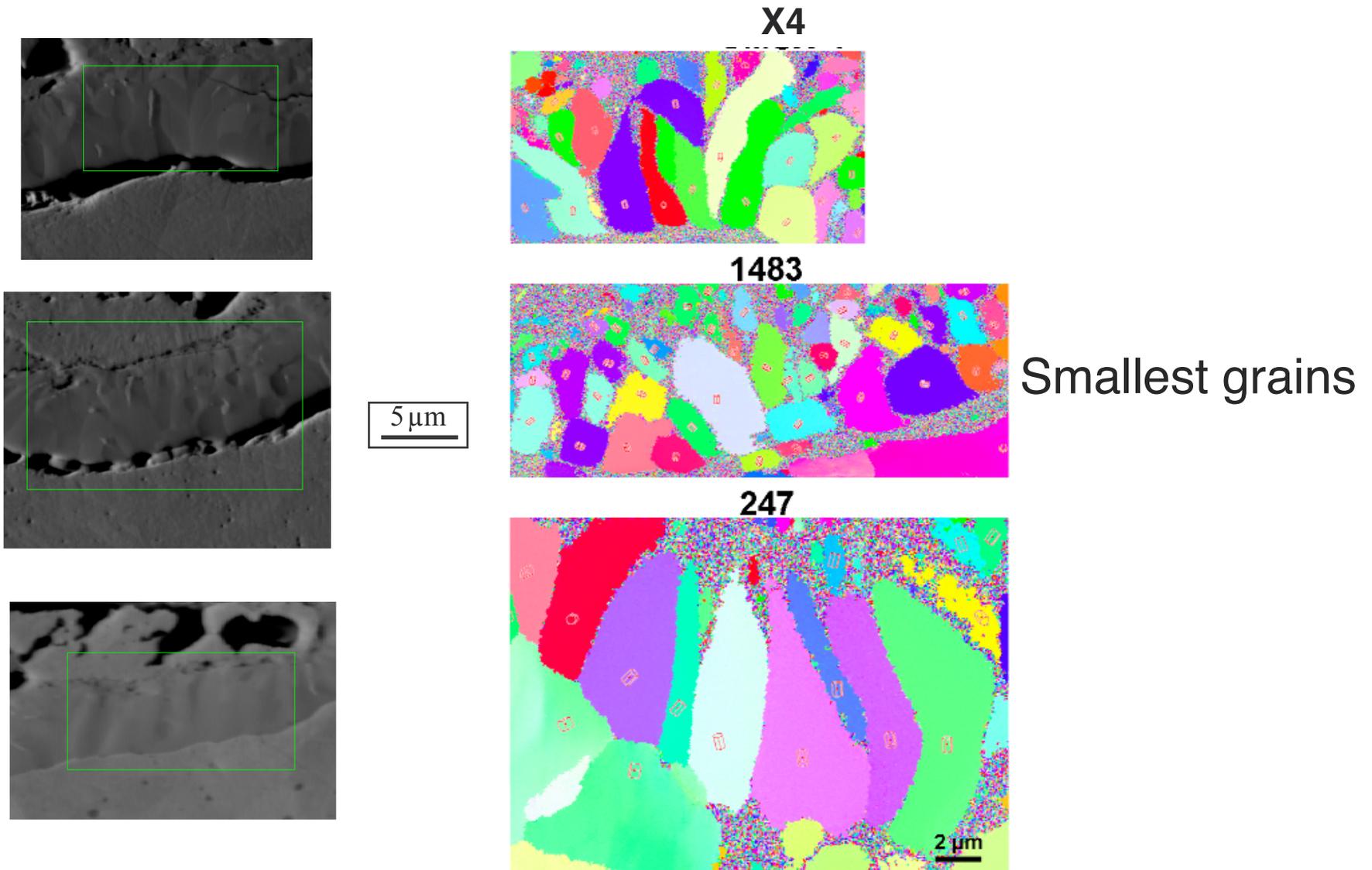


247



Each specimen failed after 480 1-h cycles at 1100°C

EBSD showed more fine grains for scale formed on 1483 with highest Ti content



Each specimen failed after 480 1-h cycles at 1100°C