

# Time-dependent stability of SOFC activated by nano-sized cathode electrocatalyst

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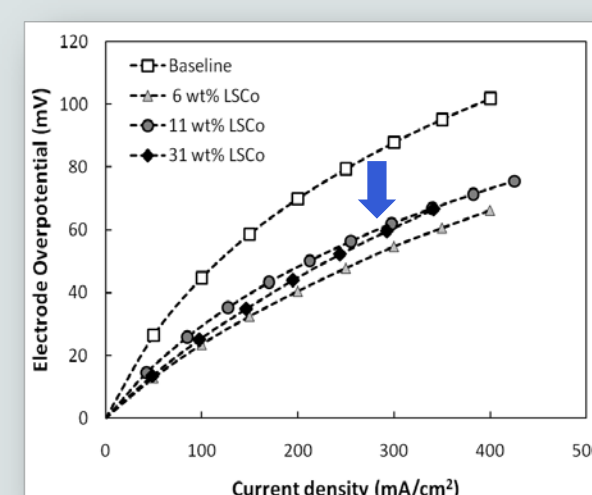
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## Introduction

### Cathode Infiltration with electrocatalysts

⇒ **Dramatic decrease in electrode overpotential due to enhanced oxygen reduction reaction.**

\* Shiwoo Lee et al., J. Electrochem. Soc., 158 (2011) B735.



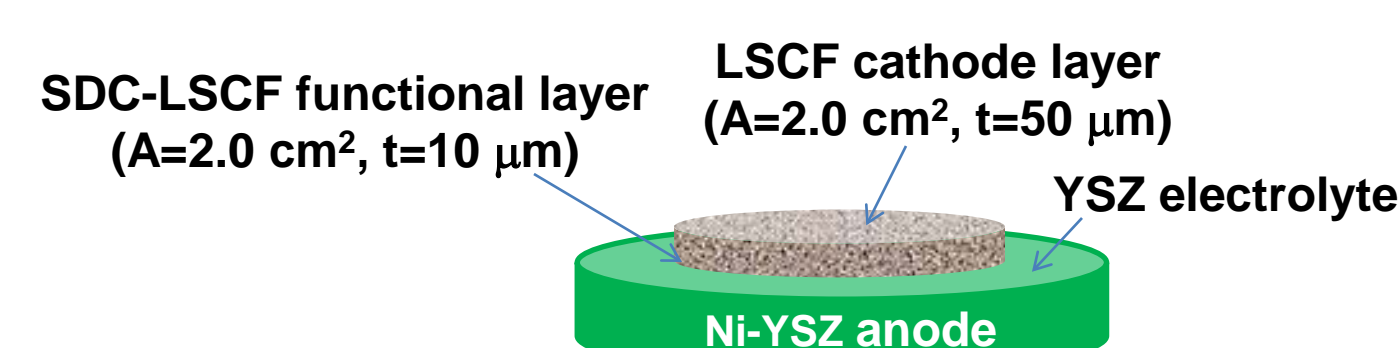
**But, long-term stability is questionable... because of electrocatalysts'**

- 1) **Nanometer size** (= small curvature radius)
- 2) **Instability** (decomposition or interaction)

### Theme of this study

⇒ **Time-dependent stability of infiltrated cathode**

## Features of this study



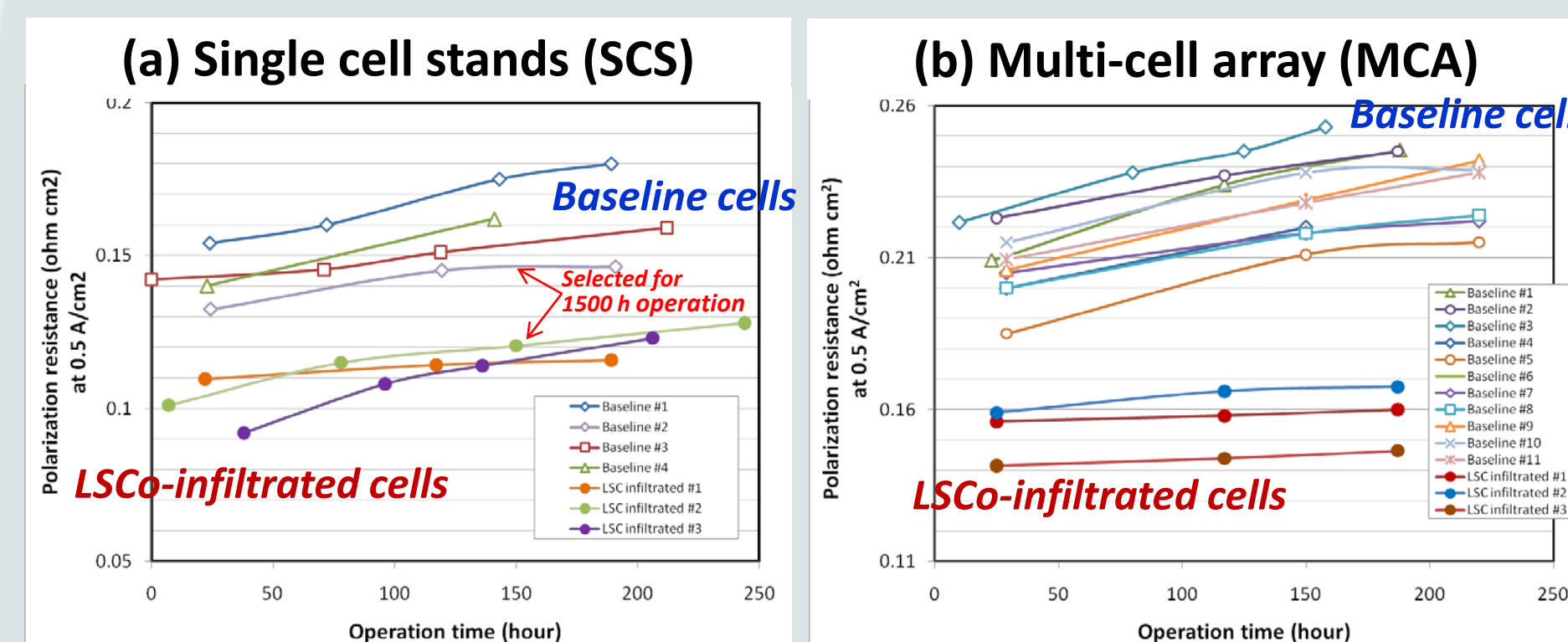
- 1) **Comparative study**  
commercial baseline cells (MSRI Co.) vs. LSCo-infiltrated cells
- 2) **Cathode backbone**: composite of SDC-LSCF
- 3) **Multiple cell tests for statistical reliability**  
utilizing single cell test stands (SCS) or multi cell array (MCA)
- 4) **Impedance analysis** for degradation mechanism investigation

	Single cell stands (SCS)	Multi-cell array (MCA)
<b>Fuel composition</b>	97% H <sub>2</sub> + 3% H <sub>2</sub> O	Pure H <sub>2</sub>
<b>Gas flow rate</b>	400 sccm for cathode 400 sccm for anode	2000 sccm for cathode 2000 sccm for anode (12 cells)
<b>Sealing material</b>	Mica	Glass
<b>Current collectors</b>	Paste	Mesh
<b>Heating schedule</b>	Direct heating to 750C	850C (30 min) aging before 750C operation



## Multiple cell tests for 200 h

Fig. Polarization resistance (Rp) variation with time (DC 0.5 A/cm²)



(a) Rp obtained at SCS (average value at the operation of 24 h)

Baseline cell = 0.14 ohmcm²

Infiltrated cell = 0.10 ohmcm²

29% decrease in Rp by infiltration  
(= 35% increase in power density under 0.7V, assuming Ro=0.2 ohmcm²)

(b) Rp obtained at MCA, (average value at the operation of 24 h)

Baseline cell = 0.20 ohmcm²

Infiltrated cell = 0.15 ohmcm²

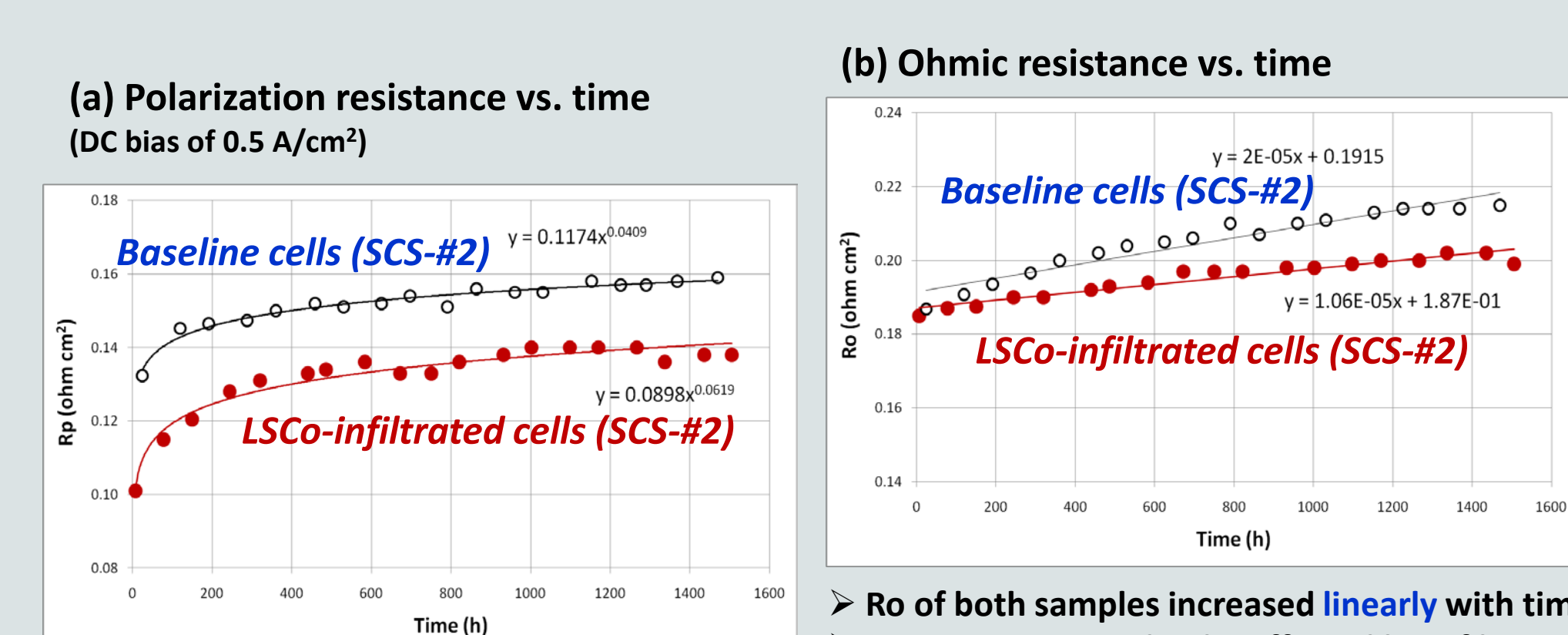
25% decrease in Rp by infiltration

Time-dependent stability for 200 h:

- (1) Regardless of test stands, cathode was activated by LSCo infiltration.
- (2) All the tested cells showed Rp increase with operation time.
- (3) **No accelerated degradation was shown for infiltrated cells.**

## 1500 h test for selected cells

Fig. (a) Rp and (b) Ro variation of the selected cells tested 1500 h



➢ Rp vs. time (1500 h): power law relationship

$y = 0.1174x^{0.0409}$  for Baseline cell (SCS-#2)

$y = 0.0898x^{0.0619}$  for LSCo infiltrated cell (SCS-#2)

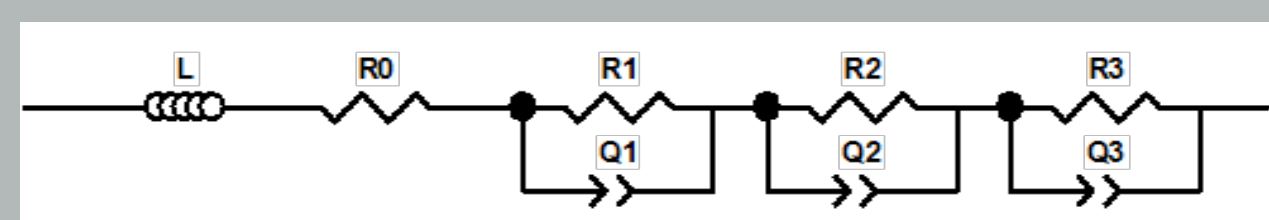
Time-dependent stability for 1500 h:

- (1) **No accelerated degradation by infiltration was confirmed.**
- (2) **70% of Rp increase over 1500 h occurs within initial 200 h operation.**

➢ Ro of both samples increased **linearly** with time.  
➢ Ro is not supposed to be affected by infiltration.

## Impedance Study

### Equivalent circuit used for fitting

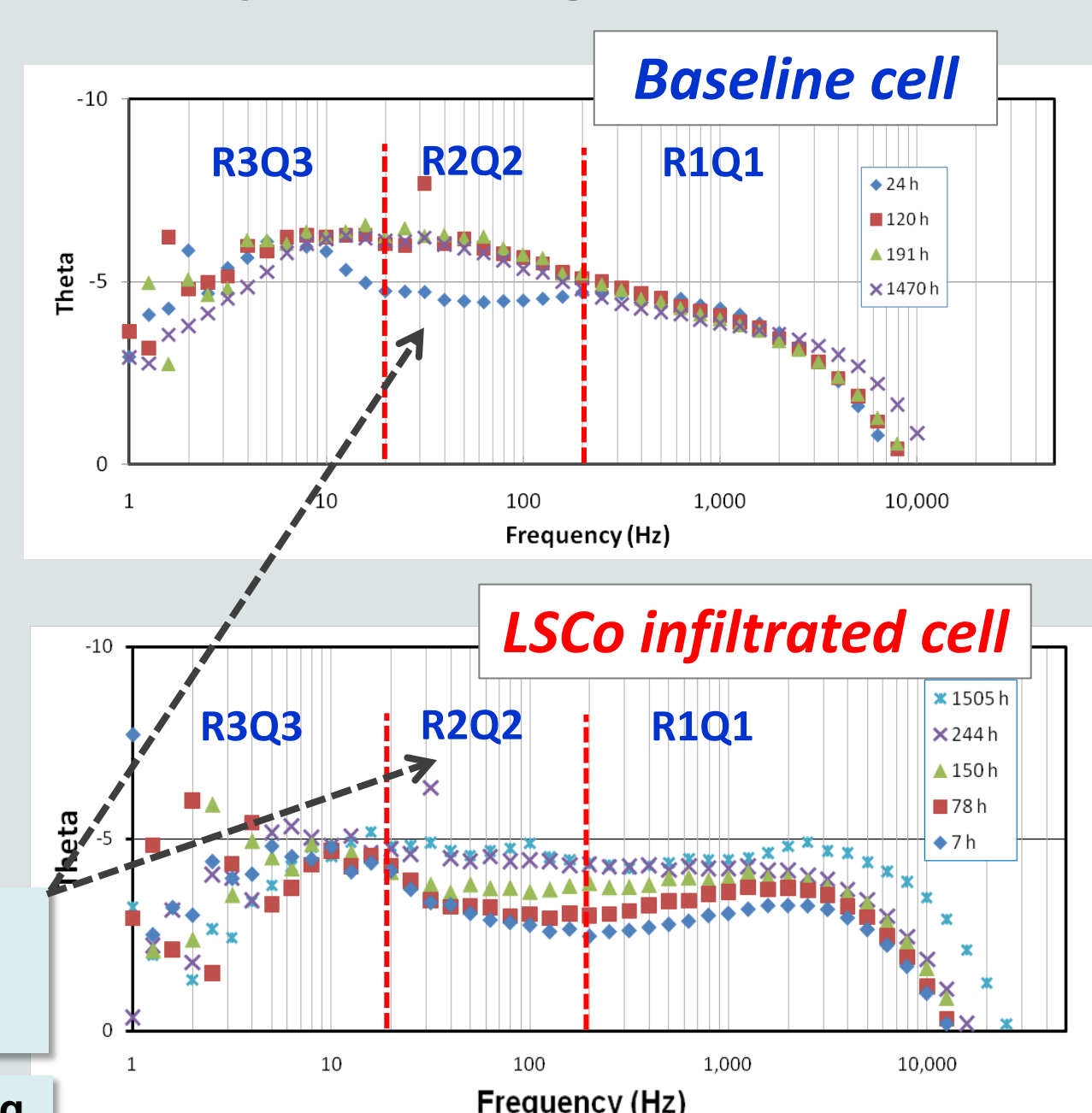


- (1) **R0**: ohmic resistance
- (2) **R1Q1**: High Frequency (HF,  $> 2 \times 10^2$  Hz): charge transfer / bulk diffusion
- (3) **R2Q2**: Intermediate Frequency (MF,  $2 \times 10^1 - 2 \times 10^2$  Hz): incorporation of oxygen molecule into cathode
- (4) **R3Q3**: Low Frequency (LF,  $10^0 - 2 \times 10^1$  Hz): seems to be related to anode reactions
- (5)  $< 10^0$  Hz: gas diffusion polarization

Characteristic appearance of a new feature in the **MF range** with a summit frequency of **20-200 Hz**, while R1 and R3 were nearly constant.

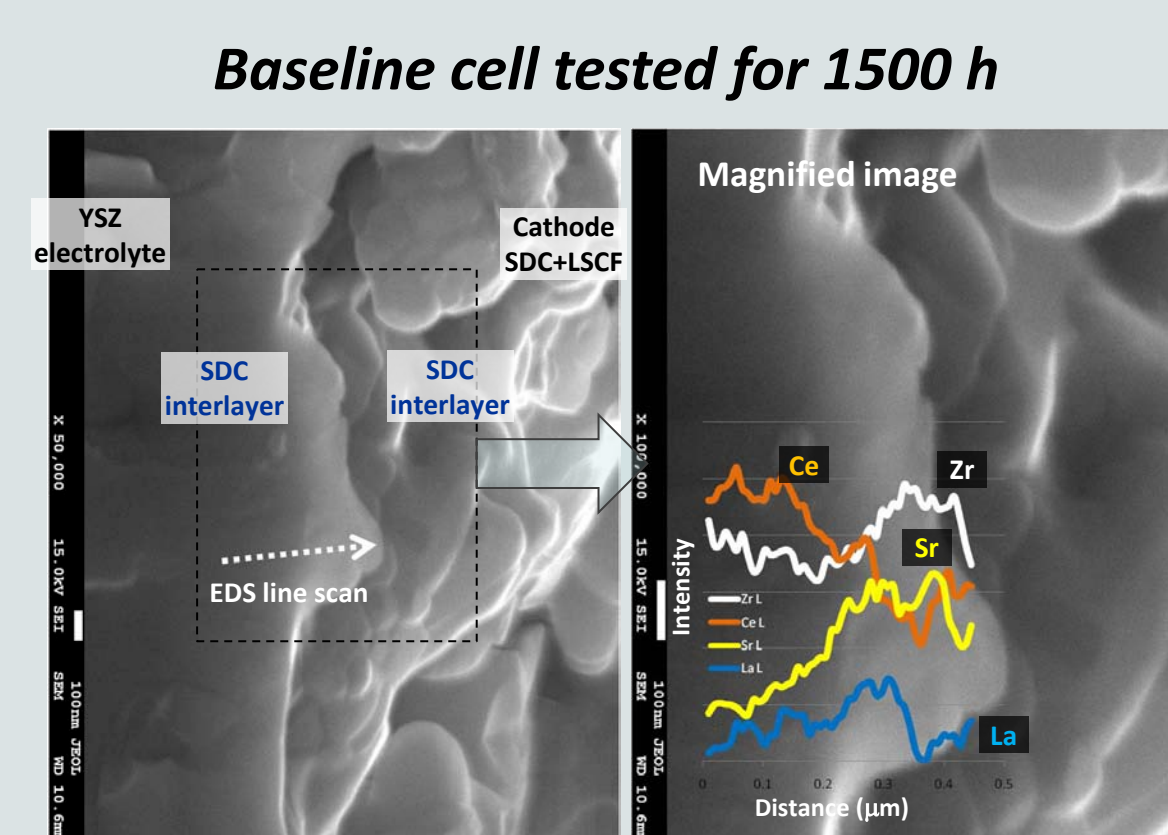
Bode plots confirm that Rp change occurs mostly during initial 200 h.

### Bode plot change over 1500 h



## Microstructure analysis

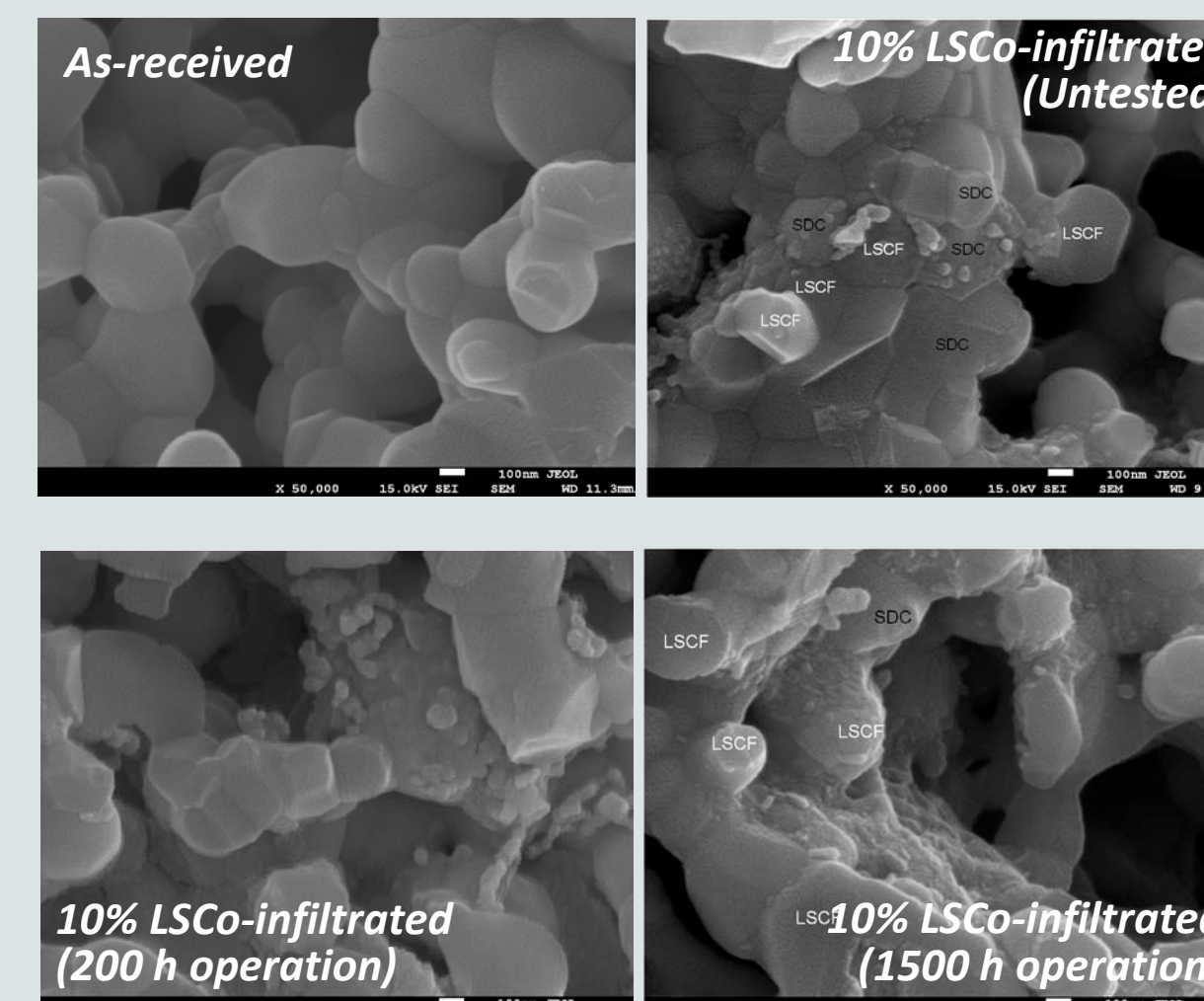
### (1) SrZrO<sub>3</sub> formation



Strontium (Sr) and zirconium (Zr) were detected as major elements at the interface of YSZ electrolyte and SDC interlayer.

**SrZrO<sub>3</sub> formation** may be one of the reasons of cathode degradation with time for both baseline cell and infiltrated cell.

### (2) Coarsening



- Overall microstructural changes with operation time seem to be minimal.
- There is **no clear evidence of particle coarsening** (loss of surface area).
- Particle **wetting behavior** on backbone cathode needs to be considered seriously because of its high contribution to cathode microstructure.

## Conclusion

- ✓ Long-term stability of infiltrated cells has been investigated systematically utilizing (1) commercial cells with composite cathode as baseline and (2) impedance spectrometry.
- ✓ There was virtually no difference in Rp degradation behavior between the two cells, baseline cell and LSCo-infiltrated cell, tested for 1500 h.
- ✓ Most of Rp degradation happened during initial a few hundred hours of operation.
- ✓ Intermediate frequency (MF) arc has been mostly affected by time-dependent degradation for both cells tested for 1500 h.
- ✓ SrZrO<sub>3</sub> formation in spite of SDC interlayer seems to be one of the reasons of cathode degradation.
- ✓ **Conclusively, there was no accelerated degradation by nano-sized LSCo electrocatalyst.**