

# **Advanced Metallic Interconnect Development**

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# Interconnect Development

## Objectives:

- Develop cost-effective, optimized materials and coatings for intermediate temperature SOFC interconnect and interconnect/electrode interface applications.
- Identify and understand degradation processes in interconnects and at their interfaces.

## Approaches:

- Evaluation of conventional and newly developed alloys (chemical, electrical, mechanical properties, cost).
- Investigation and understanding of degradations in bulk alloy interconnects and at their interfaces under SOFC operating conditions.
- Materials development
  - ❖ Surface modification
  - ❖ Bulk modification or alloy development
  - ❖ Cathode/interconnect interfaces

# Focus Areas & Progress

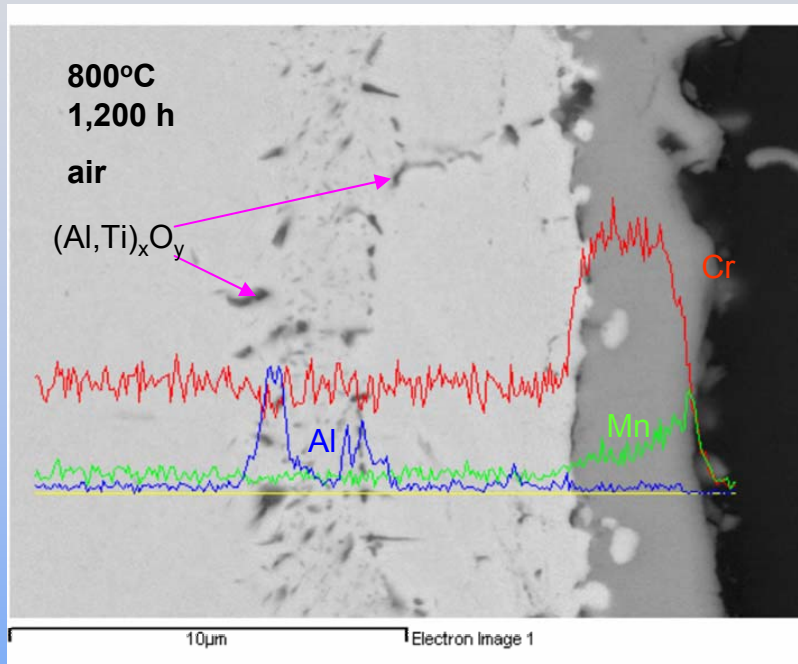
- Study of Ni-based alloys.
- Investigation of oxidation behavior of candidate alloys under SOFC operating conditions
- Development of cathode-side functional interfaces

# Focus Areas & Progress

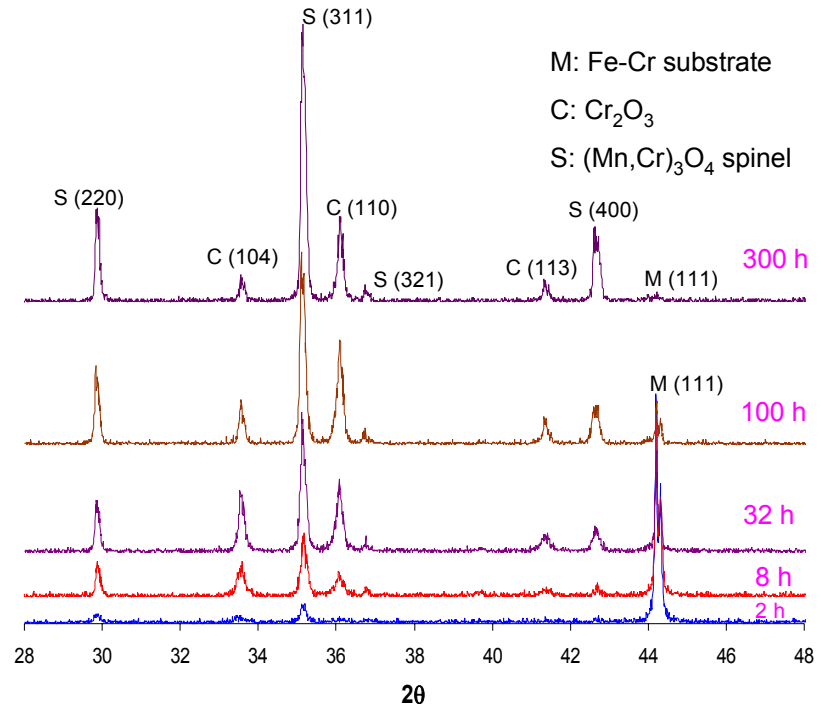
- Study of Ni-based alloys
- Investigation of oxidation behavior of candidate alloys under SOFC operating conditions
- Development of cathode-side functional interfaces

# Ferritic Stainless Steels: Status and Issues

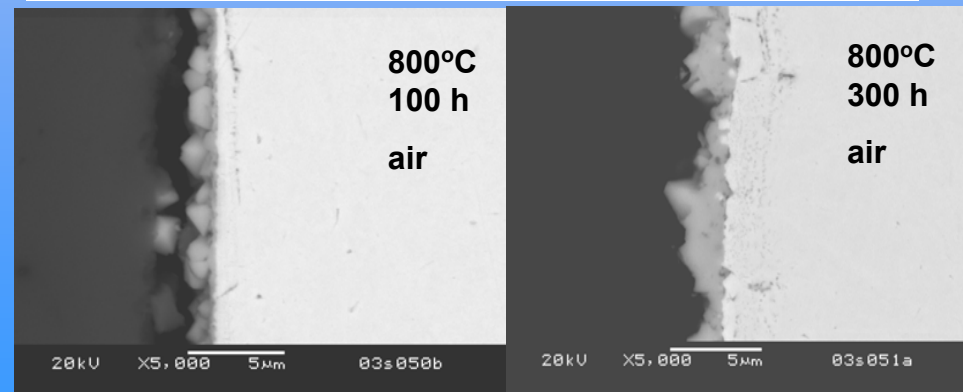
Crofer22 APU



## In-situ X-Ray Diffraction Analysis



- Scale volatility;
- Long term oxidation resistance under SOFC operating conditions;
- Life time scale electrical properties;
- Mechanical/thermomechanical stability.



# Study and Evaluation of Ni-Based Alloys

Alloys <sup>a</sup>	Nominal composition, wt%														
	Ni	Cr	Fe	Co	C	Mn	Si	Mo	W	C <sub>b</sub>	Ti	Al	B	V	Others
Haynes242	Bal	8.0	2.0 <sup>b</sup>	2.5 <sup>b</sup>	0.03 <sup>b</sup>	0.8 <sup>b</sup>	0.8 <sup>b</sup>	25.0	--	--	--	0.5 <sup>b</sup>	0.006 <sup>b</sup>	--	0.5 <sup>b</sup> Cu
LTES700	Bal	12.0						18.0			1.1	0.9			
Hastelloy C-4	Bal	16.0	3.0 <sup>b</sup>	2.0 <sup>b</sup>	0.01 <sup>b</sup>	1.0 <sup>b</sup>	0.08 <sup>b</sup>	16.0	--	--	0.70	--	--	--	
Hastelloy S	Bal	16.0	3.0 <sup>b</sup>	2.0 <sup>b</sup>	0.02 <sup>b</sup>	0.5	0.4	15.0	1.0 <sup>b</sup>	--	--	0.25	0.015 <sup>b</sup>	--	0.02La
Haynes230	Bal	22.0	3.0 <sup>b</sup>	5.0 <sup>b</sup>	0.10	0.5	0.4	2.0	14.0	--	--	0.3	0.015	--	--

Haynes242, C, S, and 230 were developed by Haynes International; LTES700 by Mitsubishi Heavy Industries.

## Why Ni-based Alloys?

- Excellent oxidation resistance, super high temperature strength, and good manufacturability.
- Formation of NiO top scale as potential Cr stopping layer.
- CTE can be modified through alloying.
- Scale can be potentially engineered for improved electrical conductivity.

## Questions

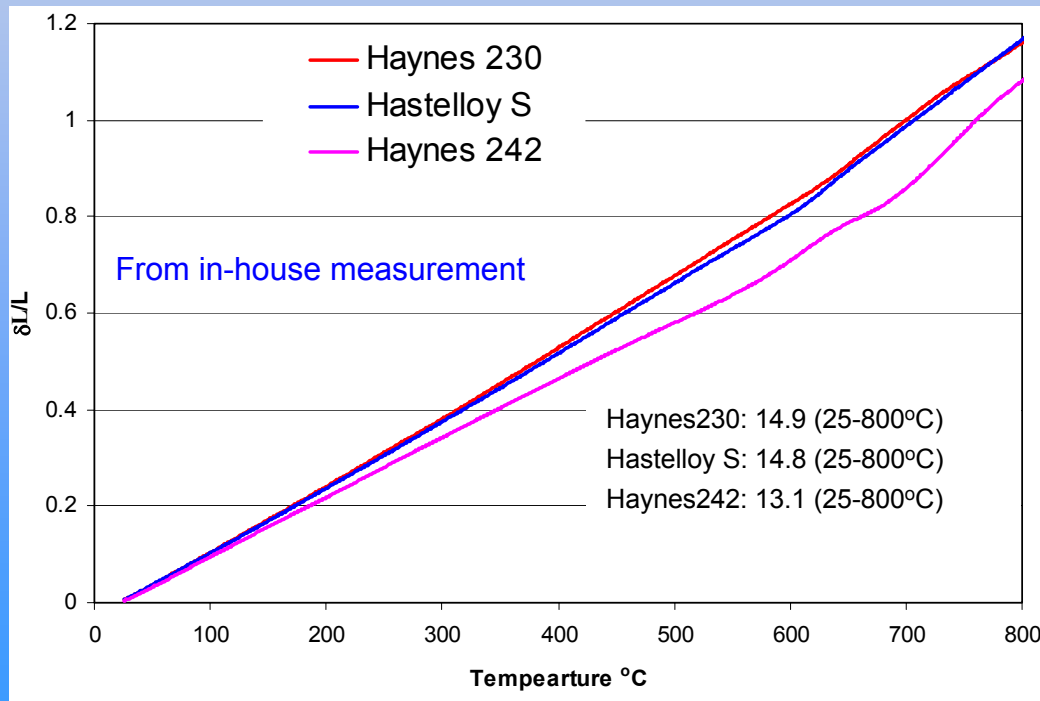
- Can the required combination of properties be found in a single alloy composition?
- Cost?

# Low CTE Ni-Based Alloys

- ❖ Traditional Ni-based alloys have a CTE of 15.0~19.0  $\mu\text{m/m.K}^{-1}$  (RT~800°C). A relatively low CTE of 13.0~14.5  $\mu\text{m/m.K}^{-1}$  (RT~800°C) can be achieved via alloying.
- ❖ Mo, W, Ti and Al reduce CTE of Ni-based alloys; while Cr, Ta+Nb and Co increase it;
- ❖ Cr concentration has to be relatively low in these alloys.

$$\alpha = 13.87 + 7.28 \times 10^{-2} Cr + 3.75 \times 10^{-2} (Ta + 1.95 Nb) + 1.98 \times 10^{-2} Co - 1.84 \times 10^{-2} Al - 7.95 \times 10^{-2} W - 8.24 \times 10^{-2} Mo - 1.63 \times 10^{-1} Ti$$

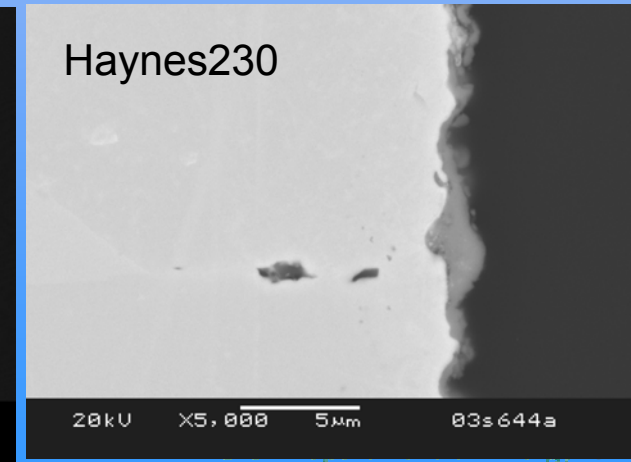
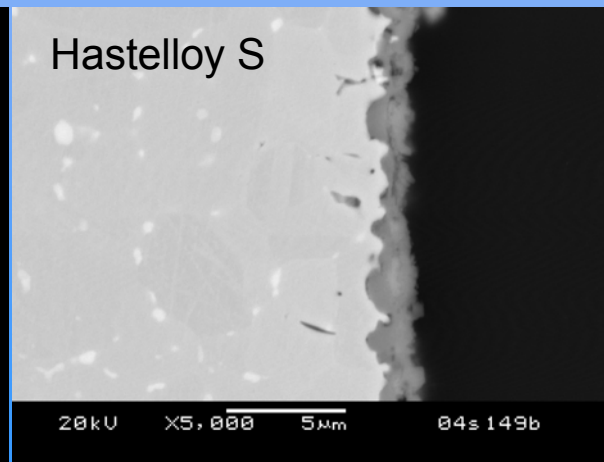
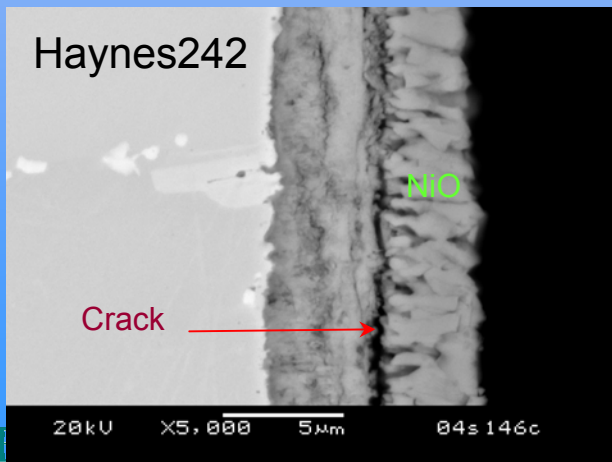
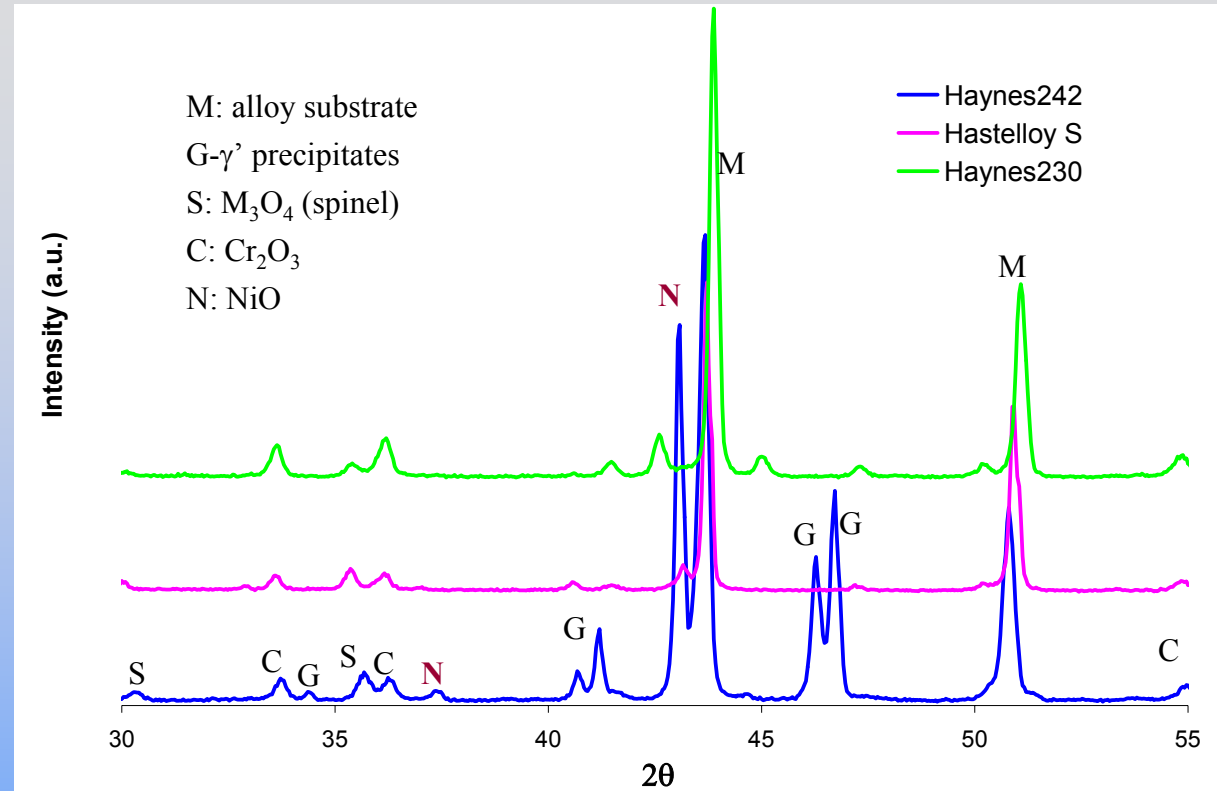
Yamamoto, et al.



Alloys	TEC $\times 10^{-6} \cdot \text{K}^{-1}$ (from manufacturers)
Crofer22 APU	<b>12.2</b> RT-760°C
Haynes242	<b>12.2-13.9</b> 20-540-760°C
LTES700	<b>13.6</b> RT-760°C
Hastelloy C-4	<b>13.3-14.4</b> 20-540-760°C
Hastelloy S	<b>13.3-14.4</b> 20-540-760°C
Haynes 230	<b>15.2</b> 25-800°C

# Scale Structure and Composition

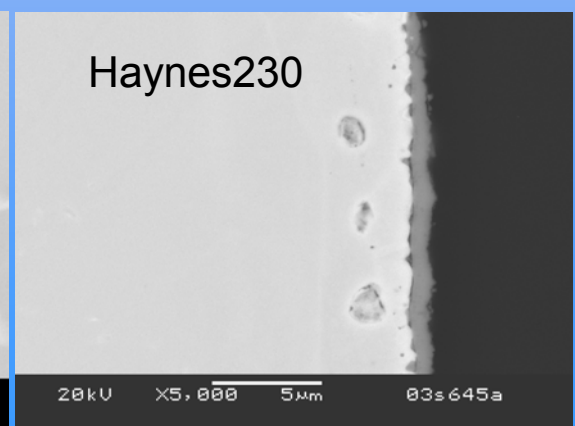
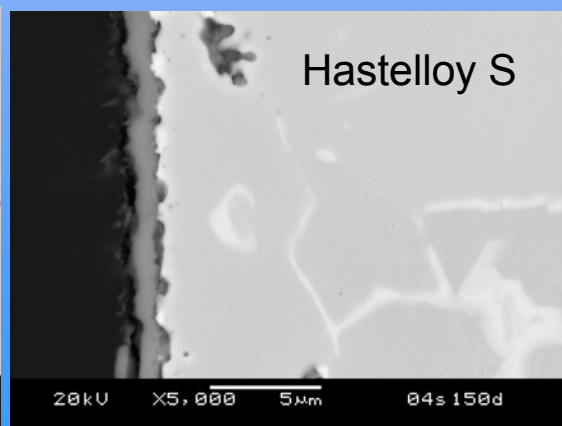
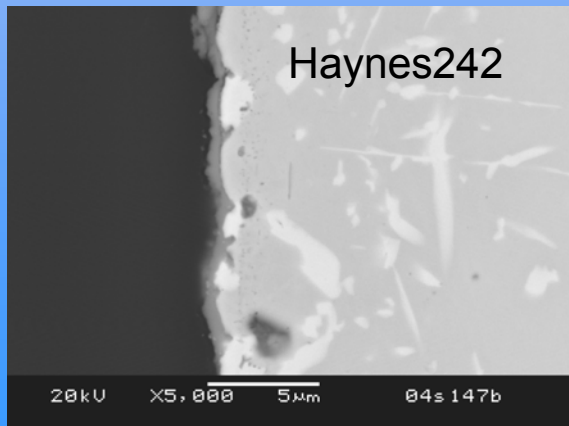
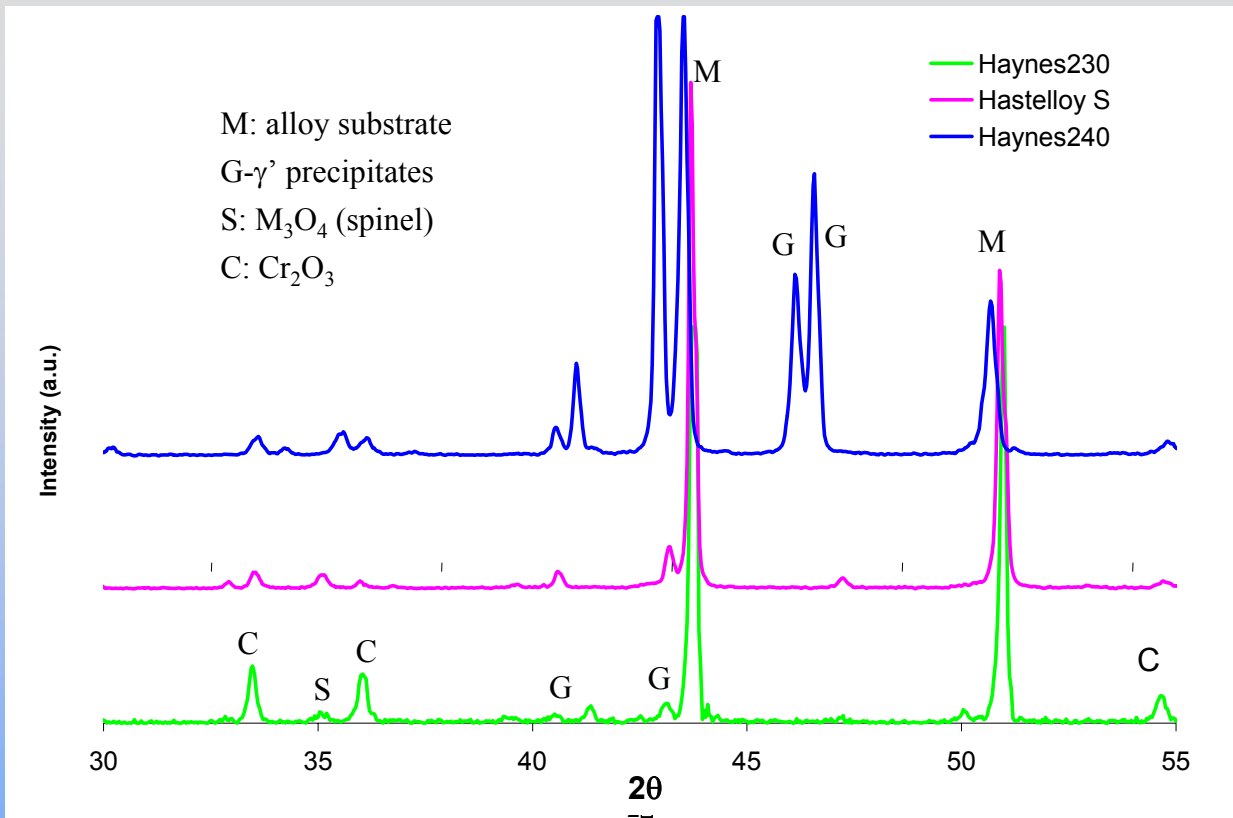
After  
oxidation at  
800°C for  
300 hours in  
**moist AIR**.





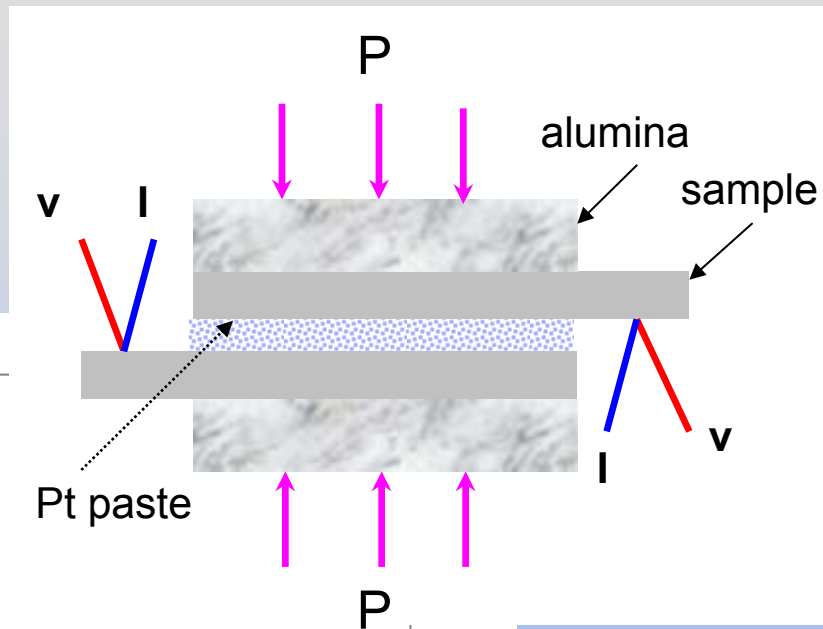
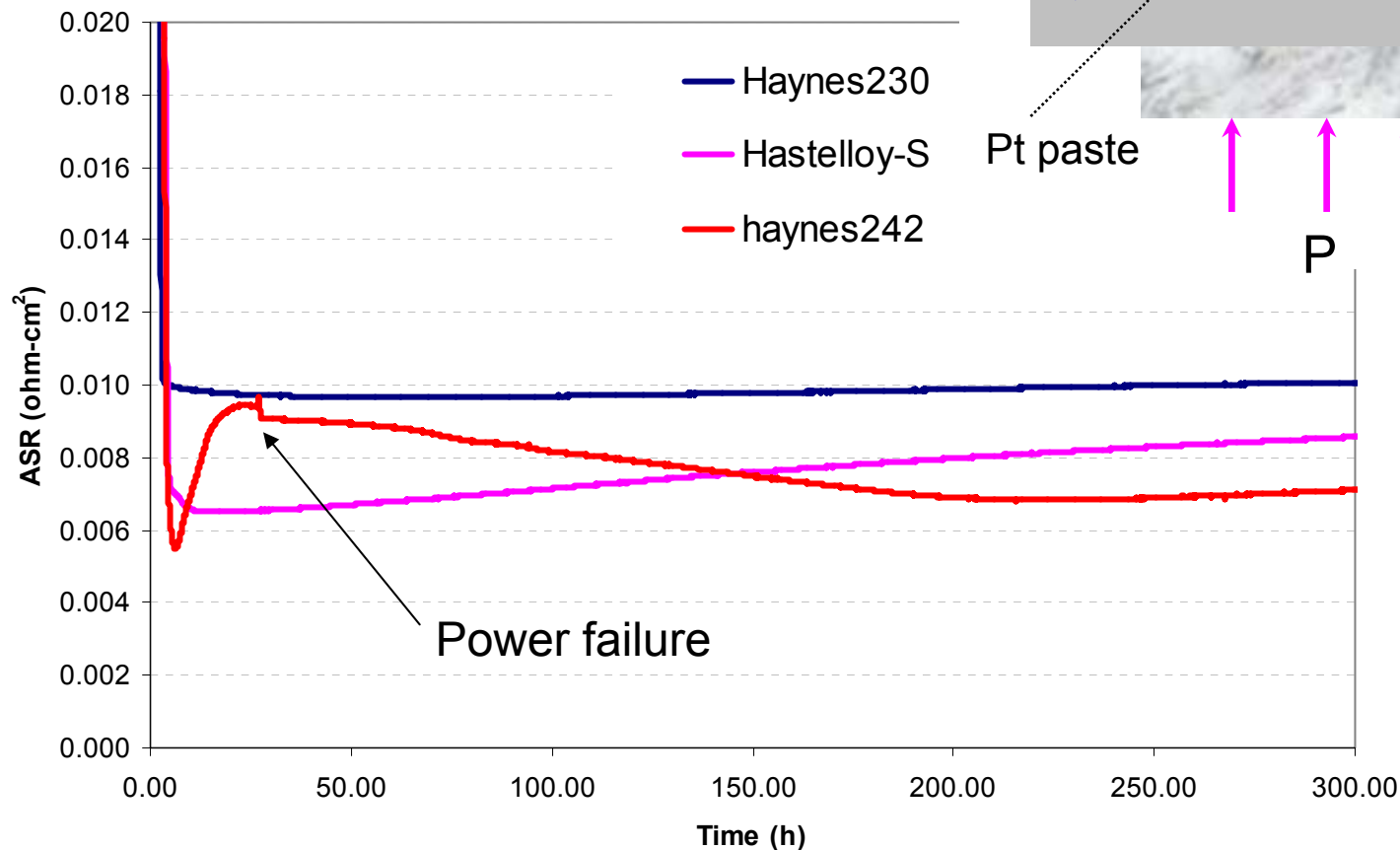
# Scale Structure and Composition

After oxidation  
at 800°C for  
300 hours in  
**moist**  
**HYDROGEN**.



# Scale ASR

The measurement was carried out at 800°C in moist air.



# Summary

## Ferritic Stainless Steels

- The newly developed FSS demonstrates reduced scale volatility, good CTE matching, reduced scaled resistance, and improved surface compatibility with sealing glasses.
- There is however a need for further improvement in long term scale chemical, electrical, and mechanical stability (for temperatures  $>700^{\circ}\text{C}$ ).

## Ni-Based Alloys

- CTE of Ni-based alloys can be adjusted to a relatively low value via lowering Cr% and adding metal elements such as W, Mo, etc.
- The decreased Cr% may however raises concerns over the oxidation resistance of an alloy in cathode environment; The heavy alloying also creates nonlinearity in the CTE curve.
- A scale with a NiO outer-layer can be formed on low Cr% Ni-alloys in cathode-side environment, but its suitability as an electrically conductive protective layer is questionable.

# Focus Areas & Progress

- Study of Ni-based alloys
- Investigation of oxidation behavior of candidate alloys under SOFC operating conditions
- Development of cathode-side functional interfaces

# Oxidation Behavior of Alloys under Interconnect Dual Exposures

## Motivation:

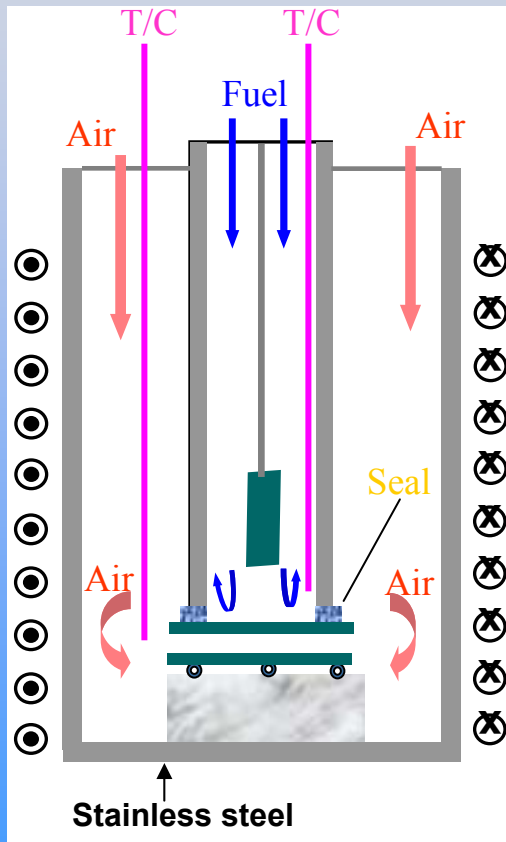
- Oxidation study has been a common area of interest, but typically under single atmosphere exposure.
- Dual exposures are commonly found in SOFC stacks and BOP, as well as other systems.
- Understanding helps develop robust materials.

## Materials studied:

NiBS	{	Haynes 230-22%Cr	FeSS	{	E-brite-27%Cr
		Hastelloy S-17%Cr			Crofer22-22%Cr
		Haynes 242-8%Cr			AISI430-17%Cr

## Variables:

- Alloy composition
- Isothermal vs. cycling
- Moisture



# Anomalous Oxidation of FSS under Interconnect Dual Exposures: A Summary

The DUAL exposures lead to an anomalous oxidation behavior of ferritic stainless steels under the SOFC interconnect dual exposure conditions;

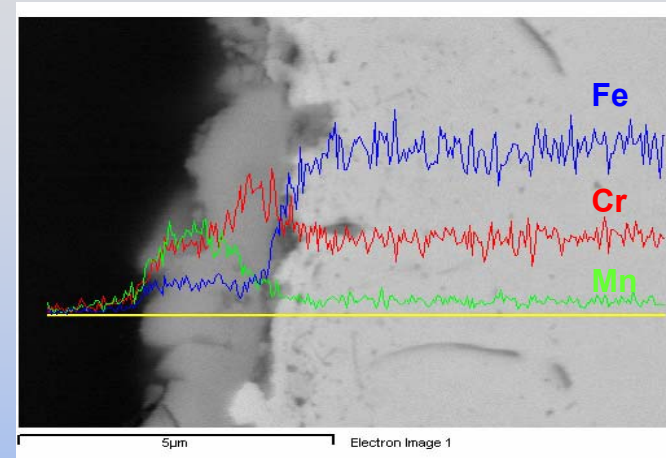
The anomalous oxidation behavior appears to be caused by hydrogen diffusion from the fuel side to the airside of alloy interconnects.

- For 430 with 17% Cr, dual exposures enhanced the iron transport in the scale on the airside, leading to hematite formation and localized attack;

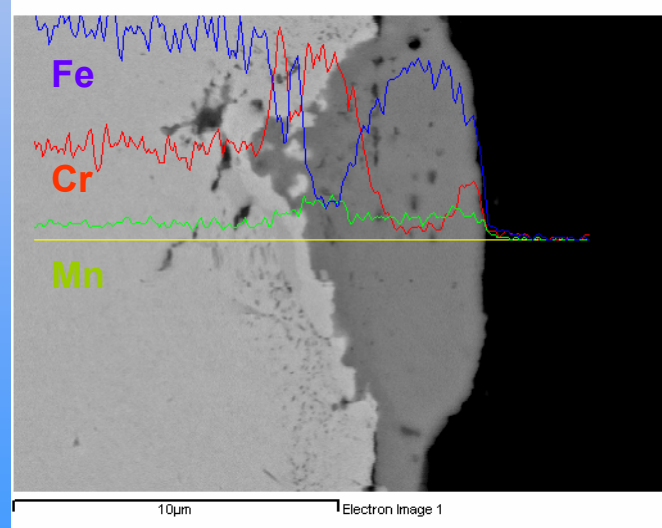
- For Crofer22 (22% Cr), Fe enrichment was found in the spinel layer after isothermal oxidation; thermal cycling resulted in the hematite nodule formation and localized attack;

- For ferritic stainless steels with enough chromium, e.g. E-brite (27% Cr), the accelerated iron transport and iron oxide formation are inhibited, though differences in scale microstructure and morphology are observed

## Airside of Crofer22 APU



Isothermal: 800°C, 300h

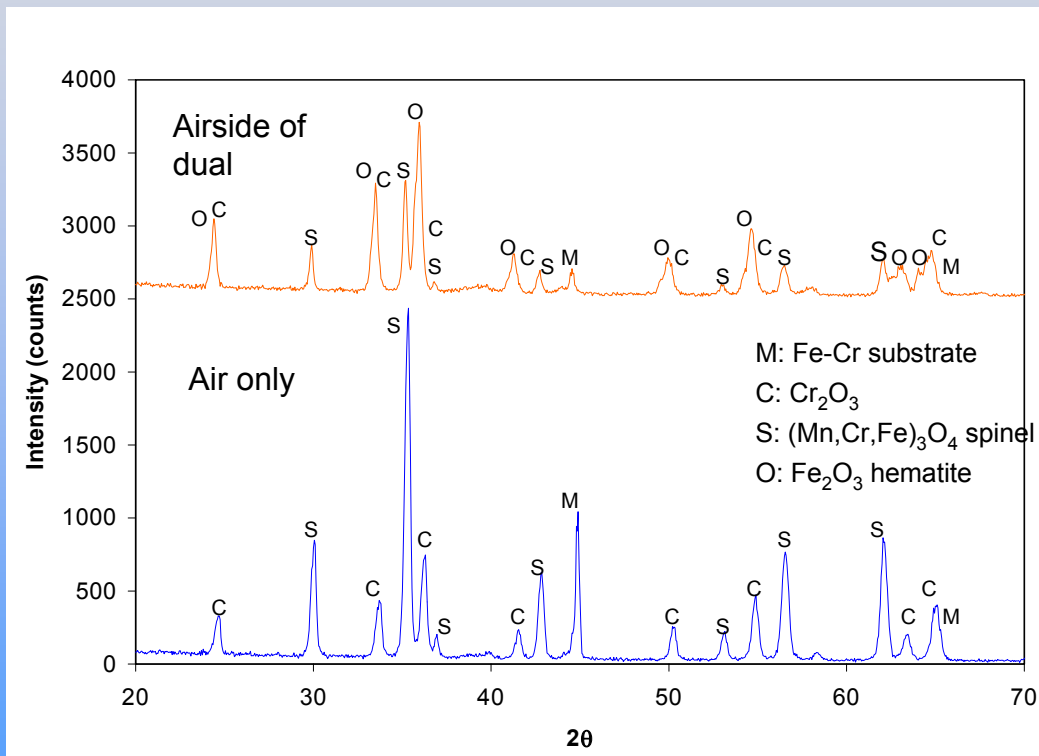


Thermal cycling: 800°C, 3x100h

Pacific Northwest National Laboratory  
U.S. Department of Energy 14

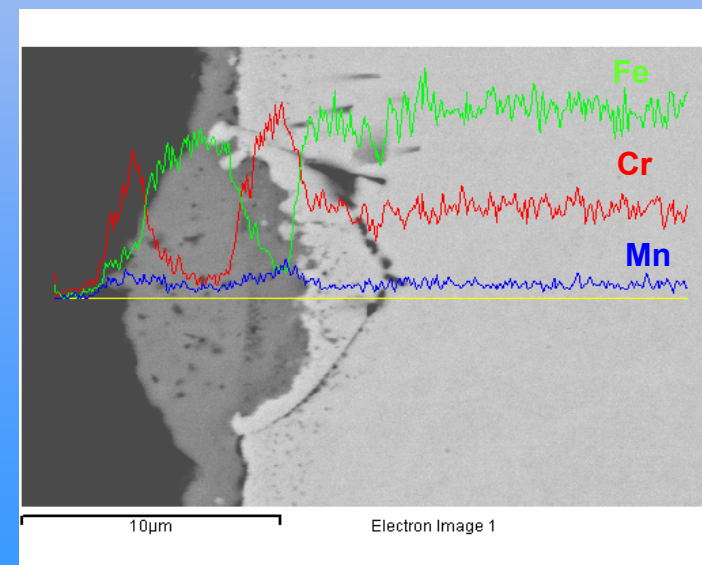
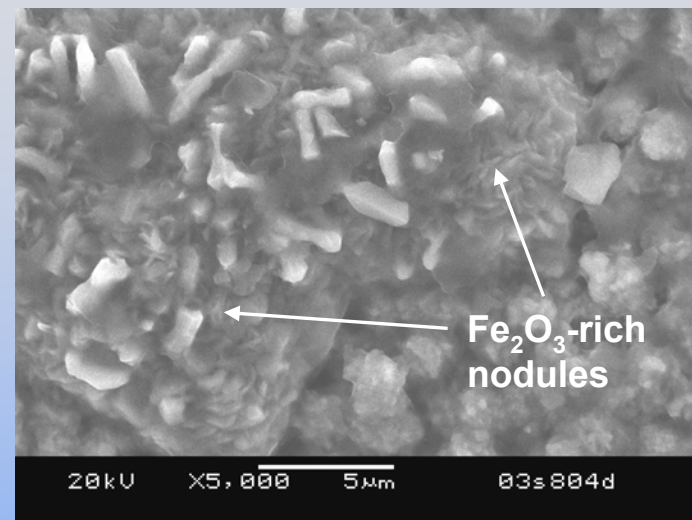
# Crofer22 APU: Effects of Moisture

Grown on the coupon in **moist (3% $\text{H}_2\text{O}$ )** air only and on the airside of the coupon that was **ISOTHERMALLY** heat-treated at 800°C, 300 hours.



**Presence of moisture accelerated the anomalous oxidation.**

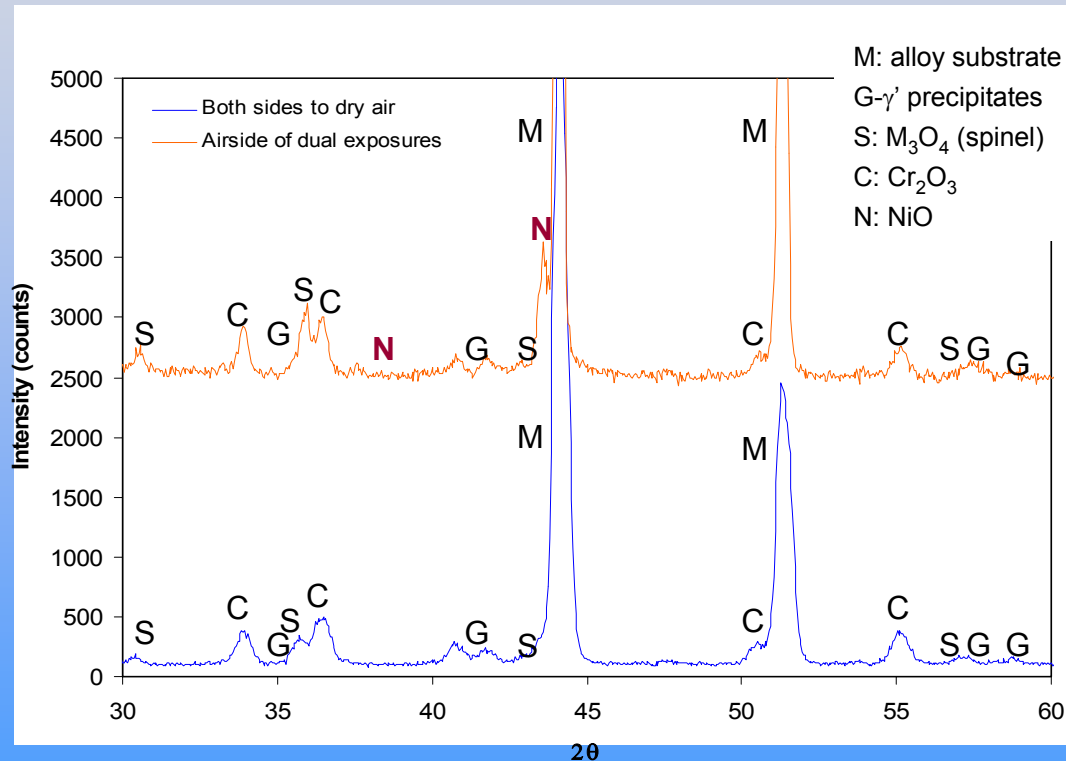
## Airside of dual exposures



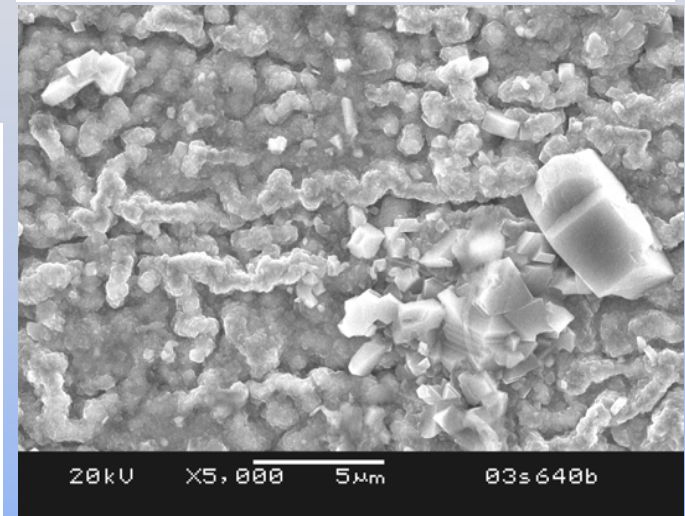


# Haynes230: Oxidation Behavior

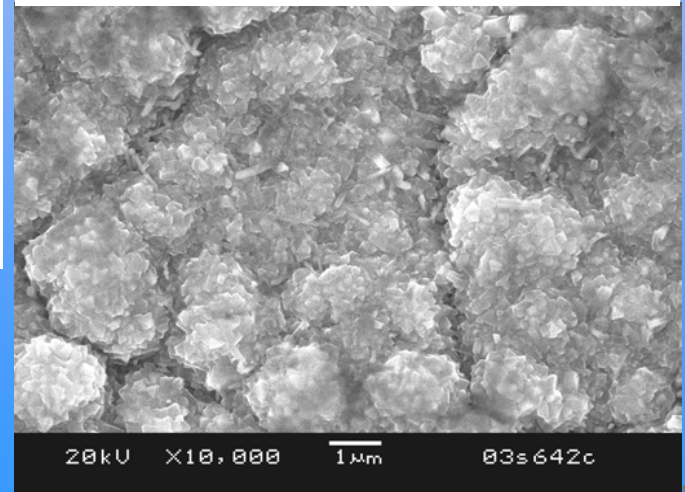
Grown on the coupon in air only (ambient air) and on the airside of the coupon that was isothermally heat-treated at 800°C, 300 hours.



Air exposure at both sides



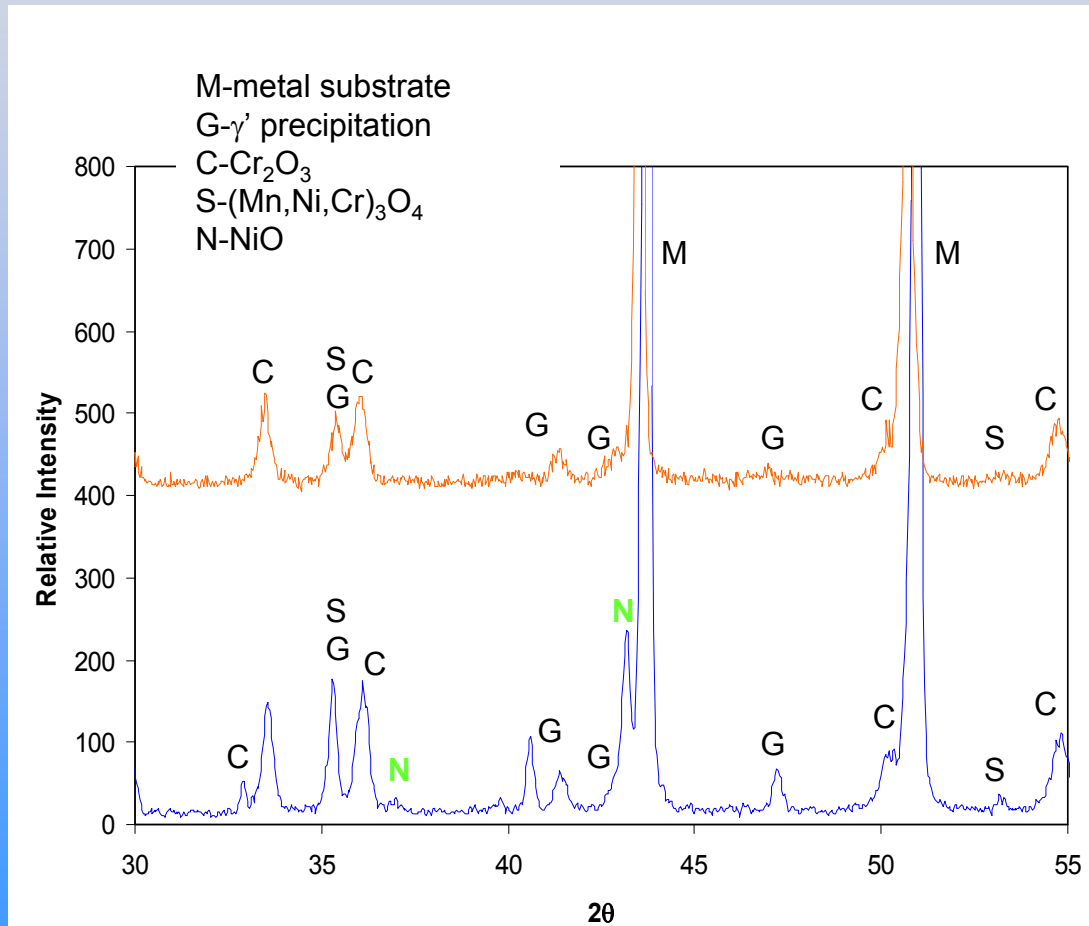
Airside of dual exposures



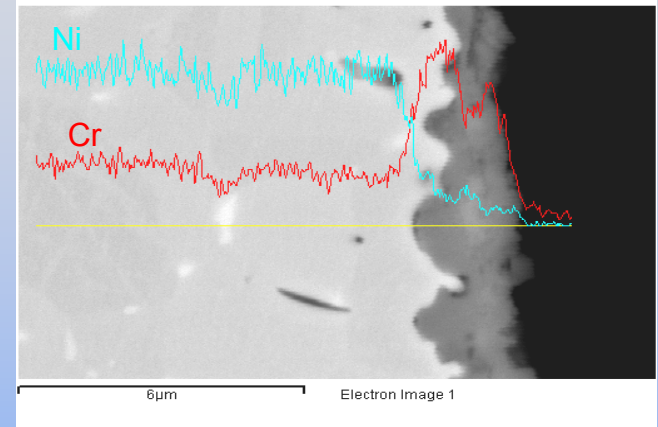


# Hastelloy S: Oxidation Behavior

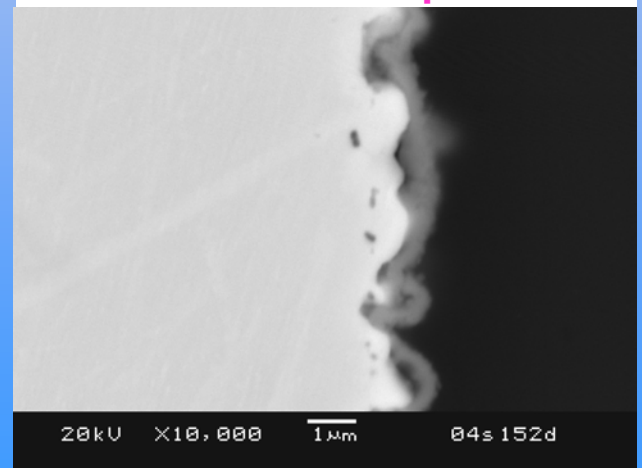
Grown on the coupon in air only and on the airside of the coupon that was isothermally heat-treated at 800°C, 300 hours.



Air exposure at both sides

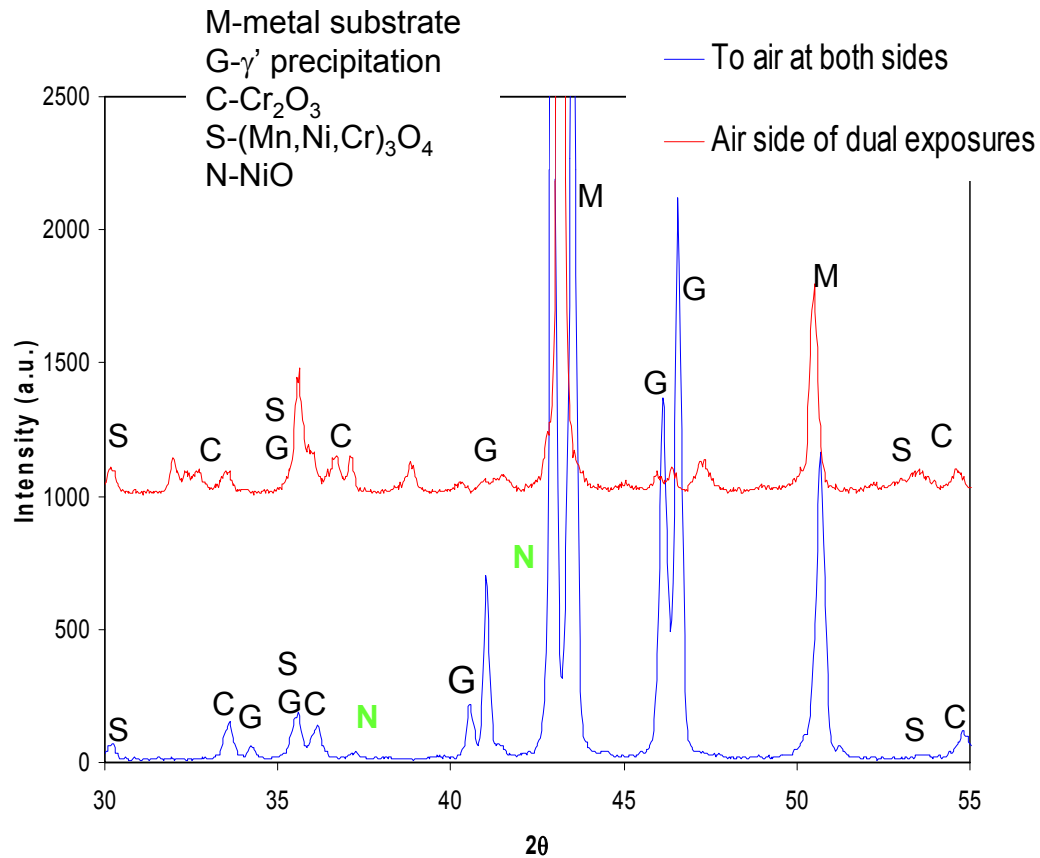


Airside of dual exposures

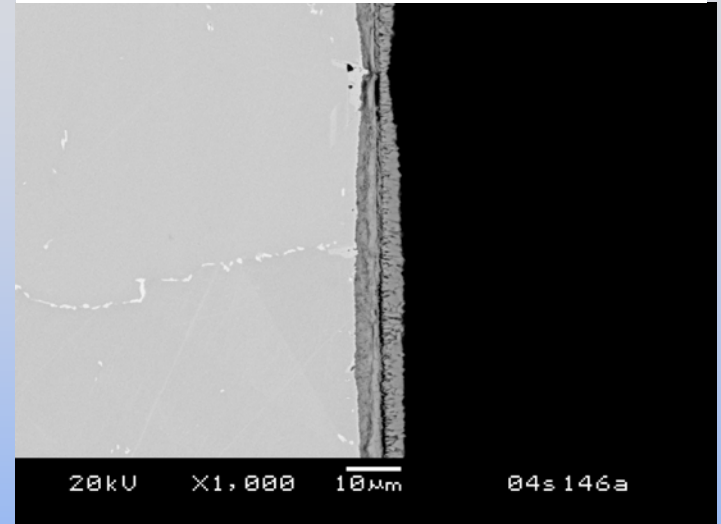


# Haynes242: Oxidation Behavior

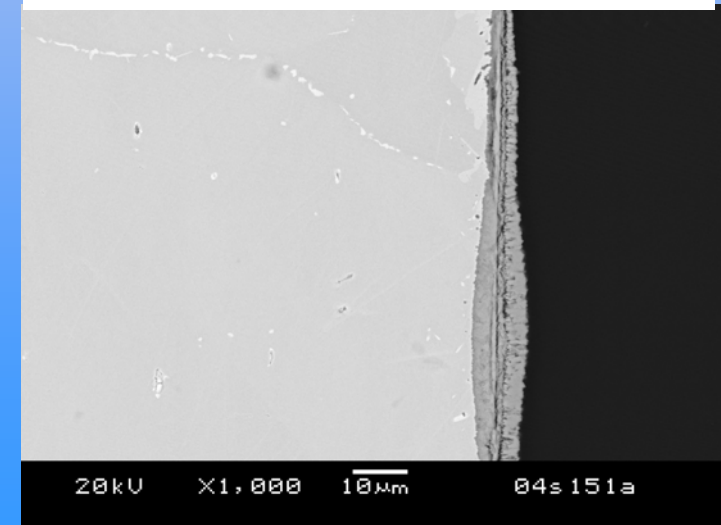
Grown on the coupon in air only and on the airside of the coupon that was isothermally heat-treated at 800°C, 300 hours.



Air exposure at both sides



Airside of dual exposures



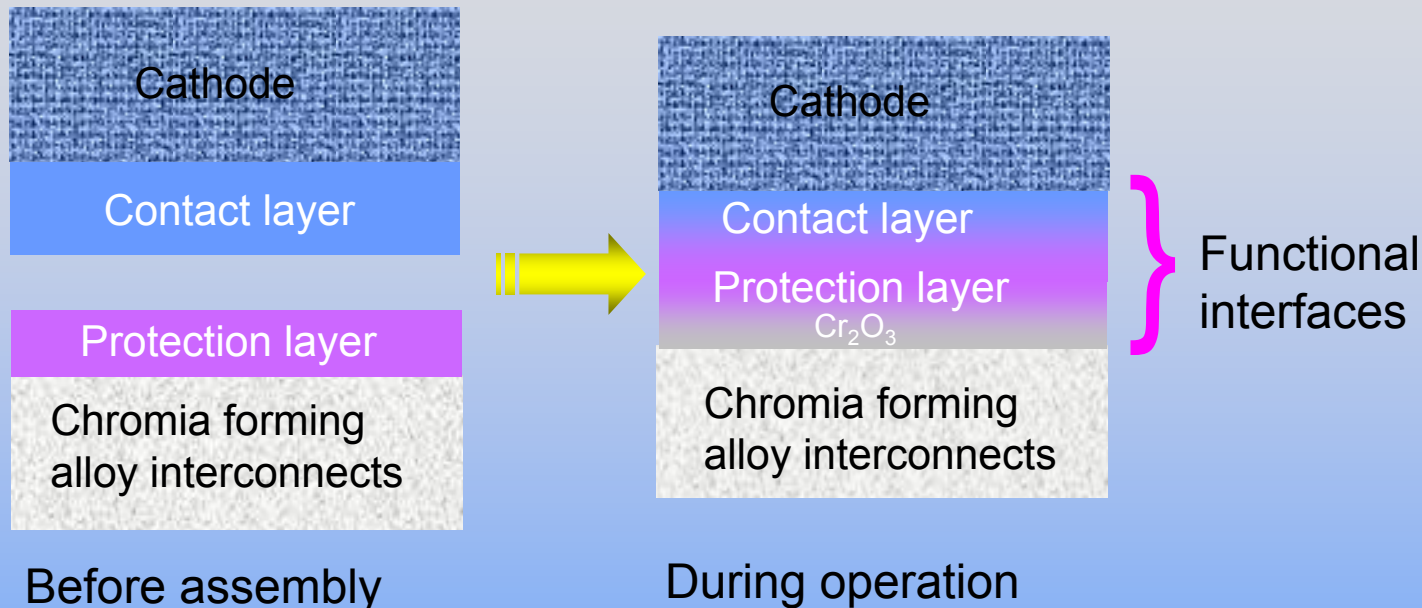
# Summary

- For ferritic stainless steels with relatively low chromium levels (22% or less), dual exposure enhances the iron transport in scale on the airside, leading to hematite formation and localized attack.
- The presence of moisture enhances the anomalous oxidation, leading to localized attack.
- For Ni-based alloys, dual atmosphere exposure tends to reduce NiO formation, and to facilitate the formation of a uniform chromia/spinel dominated scale.

# Focus Areas & Progress

- Study of Ni-based alloys
- Investigation of oxidation behavior of candidate alloys under SOFC operating conditions
- Development of cathode-side functional interfaces

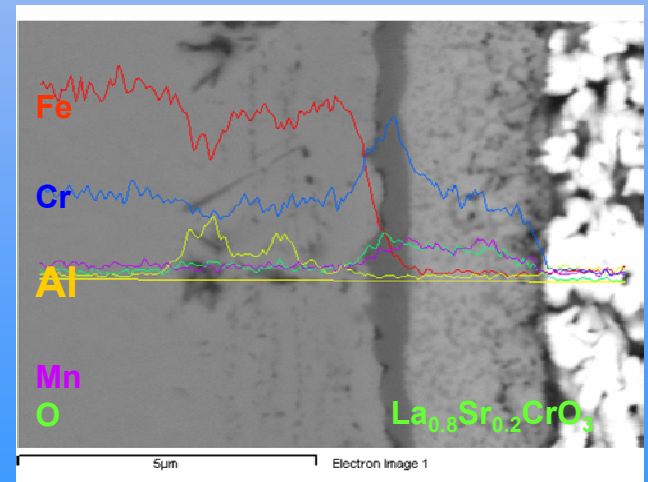
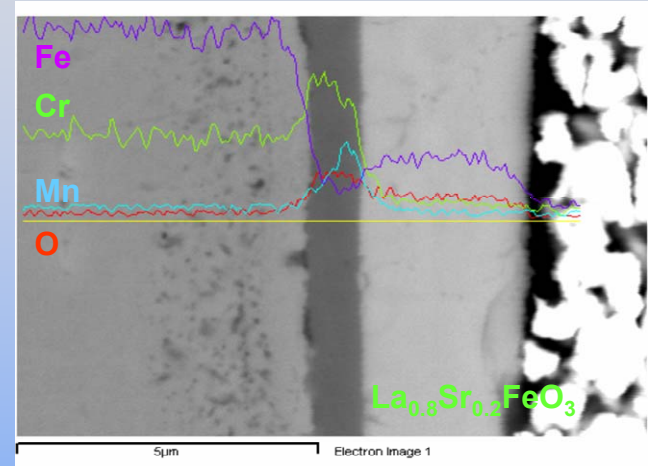
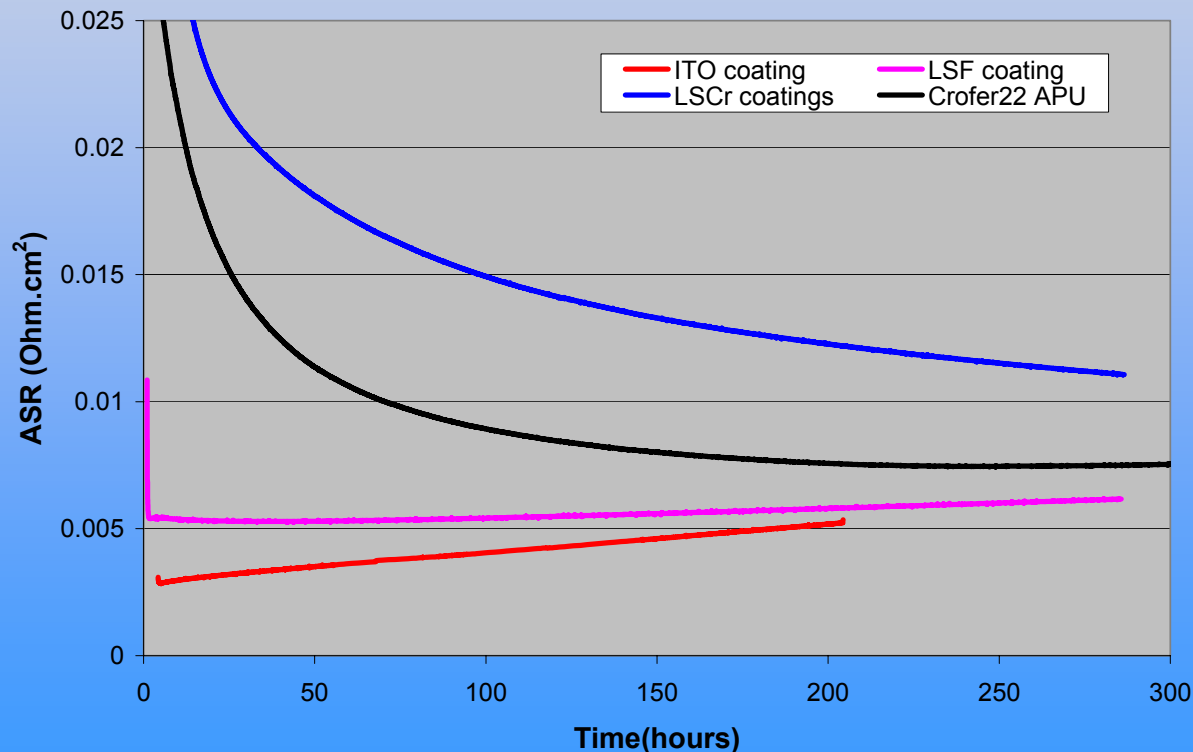
# Cathode-Side Functional Interfaces



- Protection layer acts as a mass barrier to mitigate or prevent Cr migration via both gas transport and solid state reactions, as well as to decrease electrical contact resistance. The subsequently grown chromia sub-scale serves as cation and anion transport barrier, protecting the alloy interconnect.
- Contact layer promotes contact between cathodes and interconnects, and helps minimize interfacial resistance and power loss.

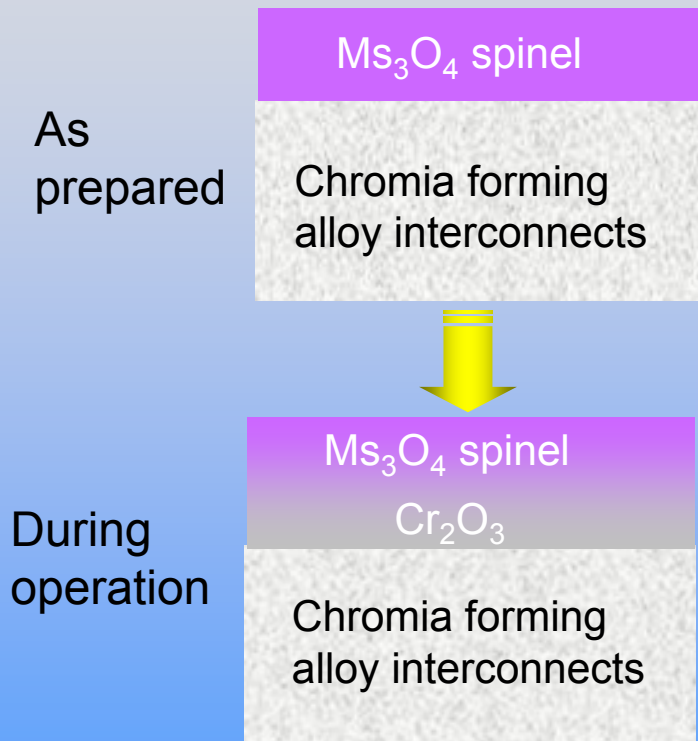
# Provskite Coatings as Protection Layers

- The provskite coatings decrease electrical resistance and mitigate or prevent Cr migration;
- The growth rate of the chromia beneath the coatings and the eventual scale depends on the ionic conductivity of coatings.
- Long term stability needs to be further studied.



# Thermal Grown Spinel Protection Layer

## Concept



● The protection layer is intended to be thermally grown.

## Approach

Solution coating, PVD, CVD or EC plating of spinel formation metals.

Growth of a thin spinel layer via reactions during a heat treatment in an optimized environment

Formation of a spinel-chromia functional scale on interconnects during subsequent oxidation or SOFC operations.

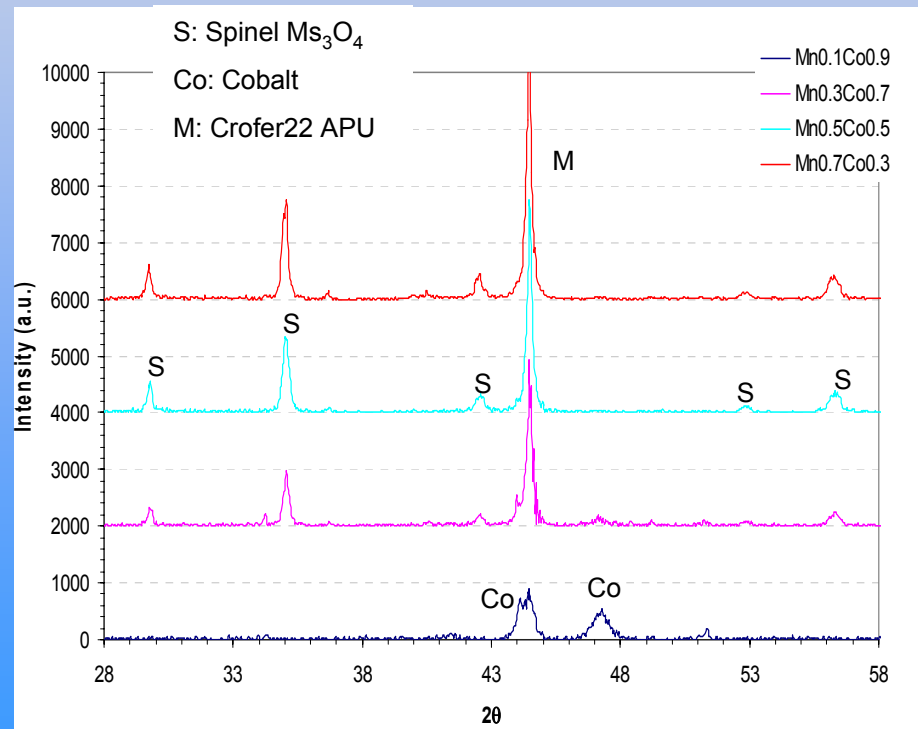
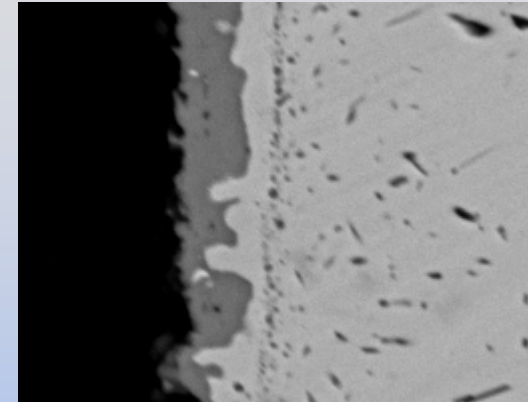
# Growth of $(\text{Mn,Co})_3\text{O}_4$ on Crofer22 APU

Current focus is on thermally grown spinels which contain no Cr and/or are more stable than  $(\text{Cr,Mn})_3\text{O}_4$ .

$\text{MnCO}_3 + \text{Co}_3\text{O}_4$   
Slurry coating

Heat treating in  
 $2.75\text{H}_2 + \text{Ar}$  at  
 $950^\circ\text{C}$  for 24  
hours.

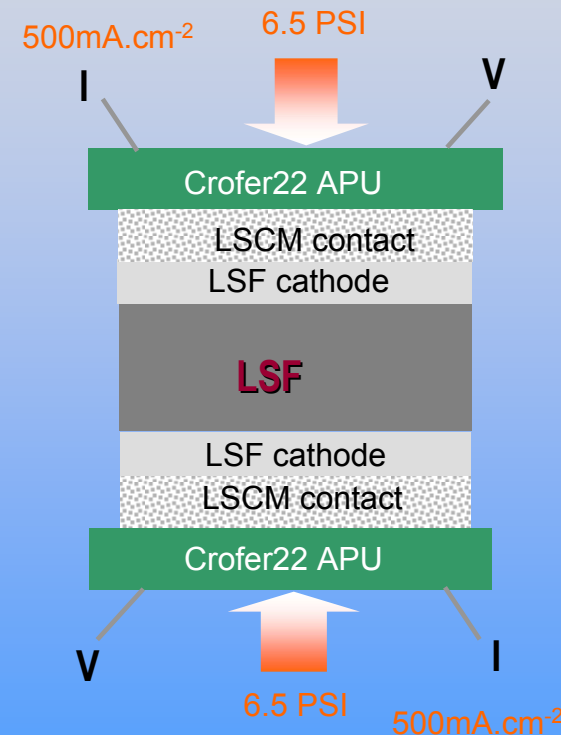
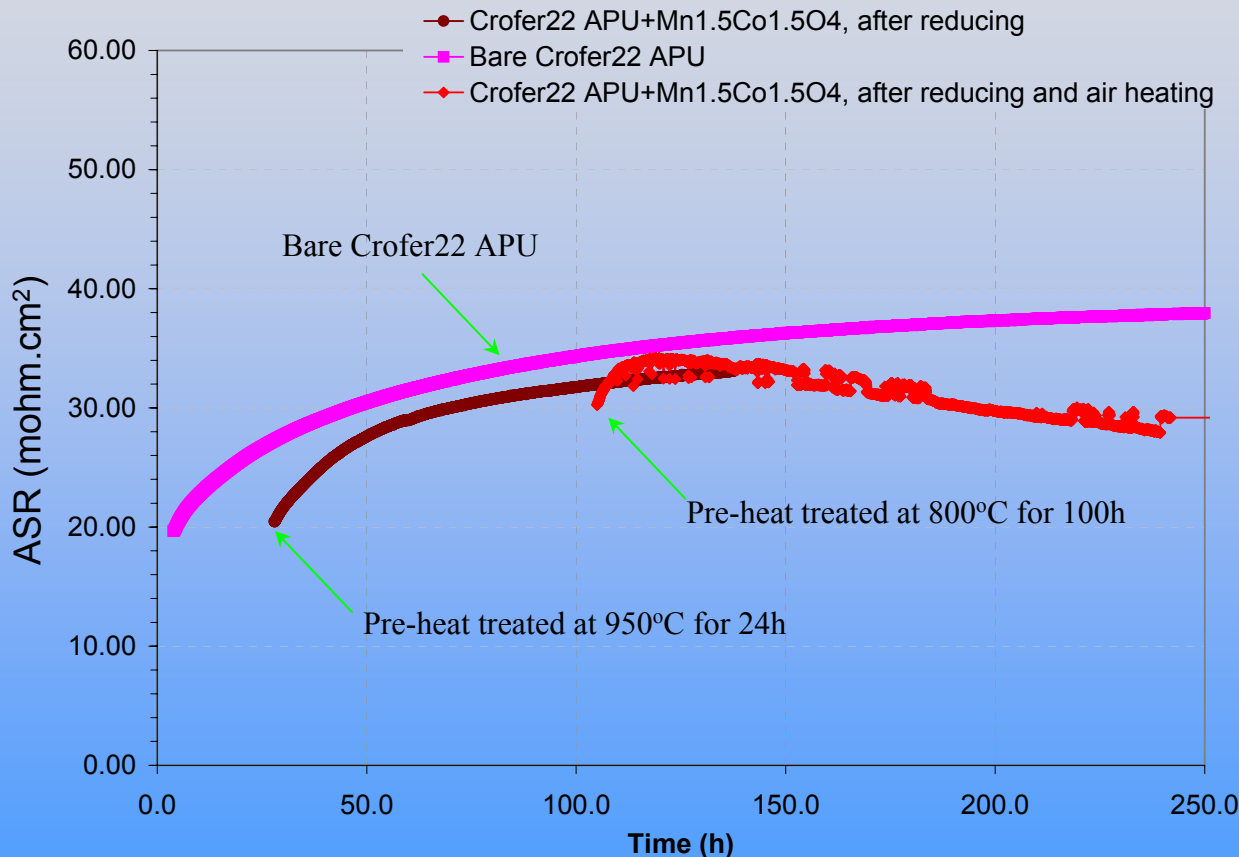
Oxidation in  
oxidizing  
environment





# Interfacial ASR of Crofer22 APU Grown with Spinel Protection Layers

The  $(\text{Mn},\text{Co})_3\text{O}_4$  spinel protection layer on Crofer22 APU minimizes the interfacial resistance when  $(\text{La}_{0.8}\text{Sr}_{0.2})\text{Co}_{0.5}\text{Mn}_{0.5}\text{O}_3$  used as an electrical contact.



$$ASR_{\text{cathode/interconnect}} = \Phi(\text{scale}, \text{contacts}, \text{reactions})$$

# **Summary**

- **Continuous, thin spinel protection layer can be thermally grown on chromia forming alloys during optimized pre-heat treating; the spinel protection layer is intended to help minimize volatilization of Cr vapor species and the interfacial electrical resistance.**
- **Preliminary work on Co/Mn spinel layers indicates low interfacial electrical resistance.**
- **Mitigation of Cr volatility to be verified experimentally.**

# Future Work:

## Study oxidation behavior under dual exposures

- Mechanistic understanding: Interaction and transport of H/H<sup>+</sup> at the metal/oxide interface and in the oxide scale; their effects on defect structure, transport properties, scale growth.
- Study effects of dual exposure on scale electrical conductivity.
- Oxidation behavior of alloys under the reforming gas/air dual exposures.

## Investigate and develop cathode-side functional interfaces

- Spinel protection layers: Continue to screen and search for spinels that compatible to candidate alloys and more thermochemically stable than (Mn,Cr)<sub>3</sub>O<sub>4</sub>; optimize processing and materials composition.
- Electrical contact layers: Continue to study the interactions between conductive oxides and candidate alloys; investigate the interfacial ASR and optimize the composition for a minimized interfacial resistance.

## Develop and investigate cladded composite-structure interconnects

- Continue to the proof of concept investigation.
- Study interdiffusion and predict life via modeling.
- Optimize structure and compositions.

## Acknowledgements

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