An Intensified Electro-Catalytic Process for Production of Formic Acid from Power Plant CO₂ Emissions (FE0031720)

Center for Applied Energy Research

FE (%)



UK CAER aims to decrease the cost of CO₂ capture from coal-derived sources through novel electrochemical methods to convert CO₂ into formic acid.

Formic acid (HCO₂H) has been selected as the target due to: 1) Lowest Gibbs energy input 1) Lowest atomic input (protons + electrons) 2) Potential for expanding the commercial market for formic acid





Project Overview

Develop and test a novel electro-catalytic method to produce high-value formic acid from coal-derived CO₂ as a strategy to offset the cost of CO₂ capture.

The project involves the development and testing of an engineered catalyst to selectively reduce CO₂ directly and exclusively to formic acid, along with process intensification and numerous aspects of a novel reactor design.

Project Period: 1/1/2019 - 6/30/2021 (30 months)

Funding: Federal - \$800K; Cost share - \$201K; Total - \$1M

Project Team: UK CAER and UNIST

NETL Project Manager: Naomi O'Neil **Center for Applied Energy Research**





Charge carriers, such as ethyl viologen (EV), shuttle charge between the cathode and catalyst, enabling formic acid production from only protons and CO₂





- MV 0.25 Fig. 3. Immobilization of the catalyst in the production cell using a mesh membrane enables high flow rates and minimal pressure. Fig. 4. Stability of charge carrier compounds after accepting electrons in the reduction cell Each step in this project has resulted in a

significant improvement to both the formic acid (formate) production and charge efficiency in UK CAER Andora dual-cell

flow-through process.

200

Time (hrs)

216 hrs

Remaining Work

- Increase formic acid production to final target of 100 mM

- Perform Life Cycle Analysis

- Conduct high-level TEA and evaluate the application of conventional formate separation techniques to this process