



Pilot Testing of a Modular Oxygen Production System Using Oxygen Binding Adsorbents

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Objective: The design, fabrication, and testing of a 10 to 20 kg/day modular oxygen (O₂) production system

- Be cost competitive with current state-of-art process
- Modular process for small scale oxygen production
- Sorbent bed-factor less than 600 lb-sorbent/TPD O₂ (tons/day O₂)
- O₂ purity greater than 95%

Specific Challenges

- Rapid PSA cycle development
- Structured sorbent module development
- Rapid cycle modeling tool development and cycle optimization
- Material and module scale up and manufacturing
- Design and fabrication of pilot O₂ production system
- Parametric and long-term testing
- Techno-economic analysis

Timeframe: 12/1/2017 to 11/30/2020

Technical approach

- Modular O₂ production
- O₂ binding materials
- Structured adsorbent beds
- Rapid PSA cycles,
- Tools to optimize the structured bed and rapid PSA cycles
- Pilot system testing
- Techno-economic analysis

Adsorbent Properties and Projected Operation Conditions		
	Units	Projected Performance
Material Properties		
Adsorbent materials		Complexed cobalt solids with permanent mesoporosity, Co-PEI complexes on silica support, and Li-exchanged zeolite
Adsorbent O₂ capacity	Wt%	>0.5
Operating Conditions		
Feed gas	–	Air
Temperature	°C	20–40
Pressure (adsorption)	kPa	Absorber: ~120 to 1,000 kPa; Regeneration: under vacuum (0.1–0.5 kPa) or atmospheric pressure
Process (adsorption)		Vacuum-swing adsorption or PSA

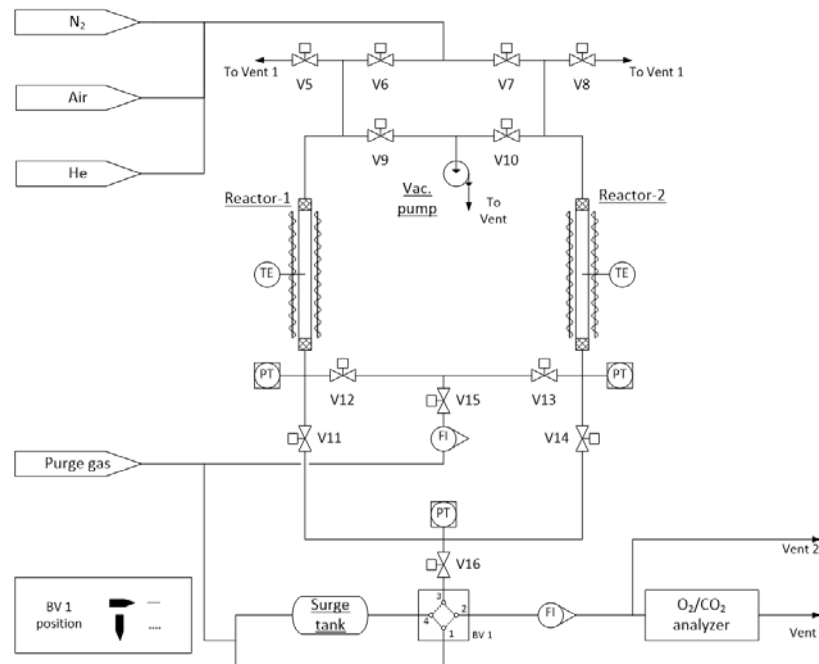
Success Criteria

1. Structured sorbent stable >1,000 rapid sorption/desorption cycles
2. BSF < 600
3. Target O₂ production cost ≤ \$46.74/ton O₂ (1996 Constant dollar) , \$61.49 in 2016\$
(cost benchmark for state-of-the-art cryogenic air distillation)

Process design package & final techno-economic analysis for full-scale, modular, rapid-cycle PSA O₂ production system using novel structured sorbent to produce 10-50 TPD of high-purity (≥95%) O₂ from air as the oxygen feed for DOE's 1- to 5-MW oxygen-blown REMS gasifier skid for power generation



L&C-300



Wide range of VPSA (Vacuum Pressure Swing Adsorption) testing capabilities

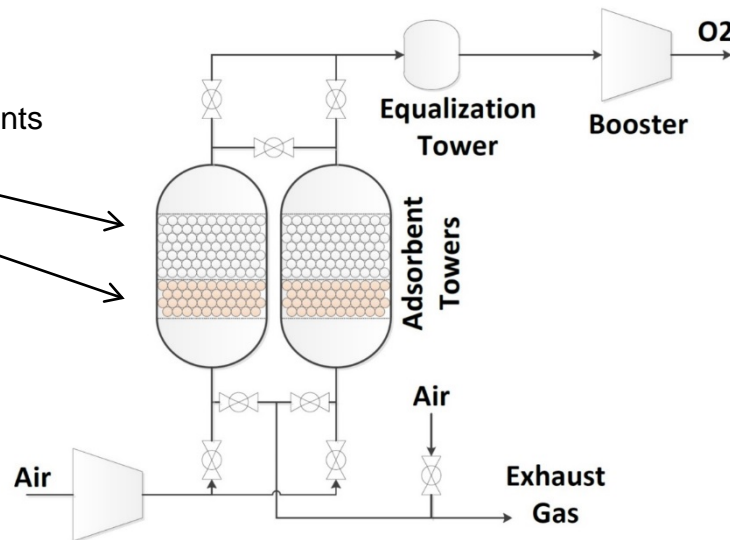
- While O_2 in air is adsorbed at sorbent in Reactor-1, product N_2 purity is monitored via effluent O_2 analyzer; meanwhile Reactor-2 is evacuated through vacuum for regeneration

Air Liquide Objectives:

- Develop and characterize structured adsorbents formulations using commercial adsorbents
- Develop and characterize structured adsorbents beds
- Manufacture and ship 2 to 4 structured adsorbents for pilot testing
- Support activities (e.g. Pilot design, operating parameters for traditional adsorbents)

BP1 Approach

- Conventional O₂ VSA → uses 2 adsorbents
- Top adsorbent used to capture N₂
- Bottom adsorbent used for air drying





Air Dehumidification:

- 2 options: activated alumina (AA) or silica gel (SG)

Multi-Step Approach

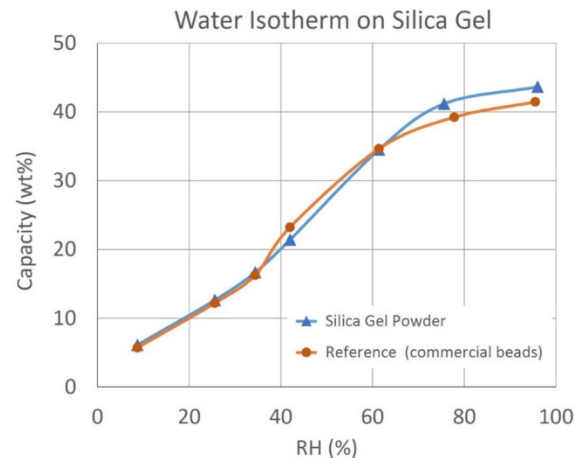
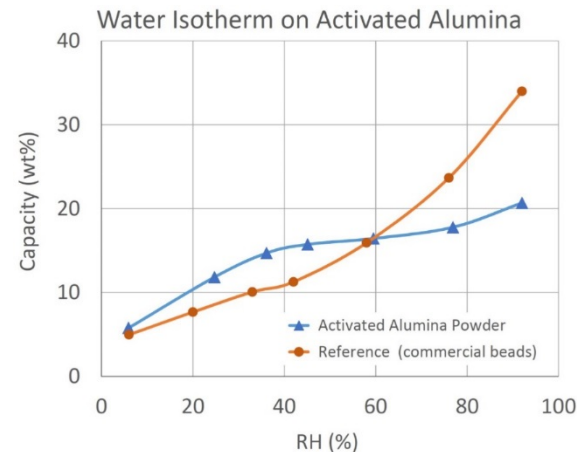
- Selection of powdery sources of AA and SG
- Characterization (Water isotherm)
- Sources with adequate capacity can be formed with a binder

Results

- H₂O capacity of AA powder is below expectation
- H₂O capacity of SG powder meets expectation

Next steps

- Form SG powder with a binder
- Make / characterize SG structured adsorbent bed



N₂-Binding Adsorbent

- Various zeolites typically used as N₂ binding adsorbents

Multi-Step Approach

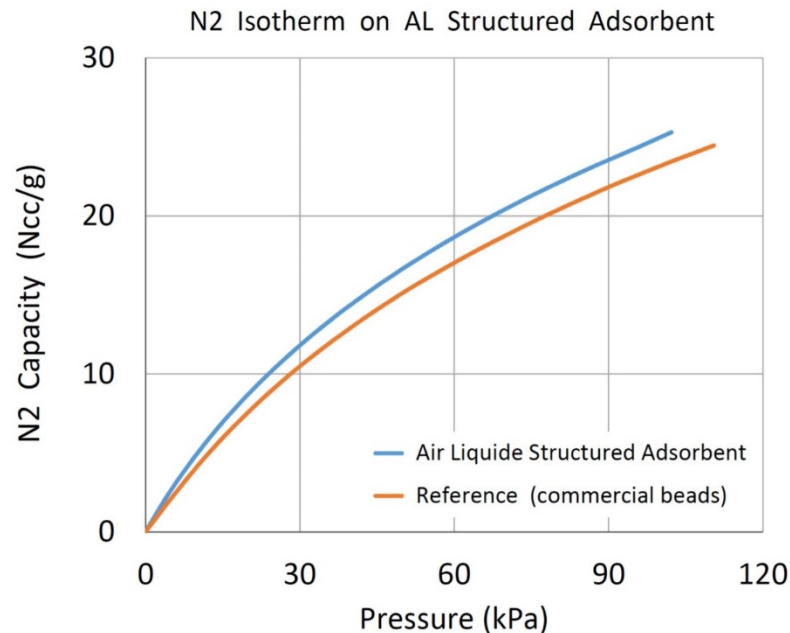
- Selection of zeolite powder
- Forming with binder
- N₂ isotherm carried after activation

Result

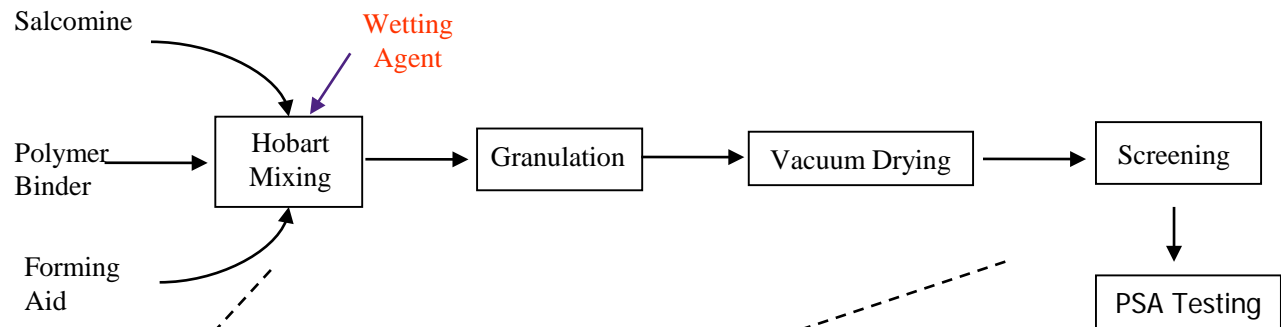
- Similar N₂ capacity compared to commercial adsorbents

Next steps

- Generate O₂ isotherm to derive N₂/O₂ selectivity
- Make and characterize Air Liquide structured adsorbent bed



Bead Forming of O₂-Sorbent for PSA Evaluation



Hobart Planetary Blender

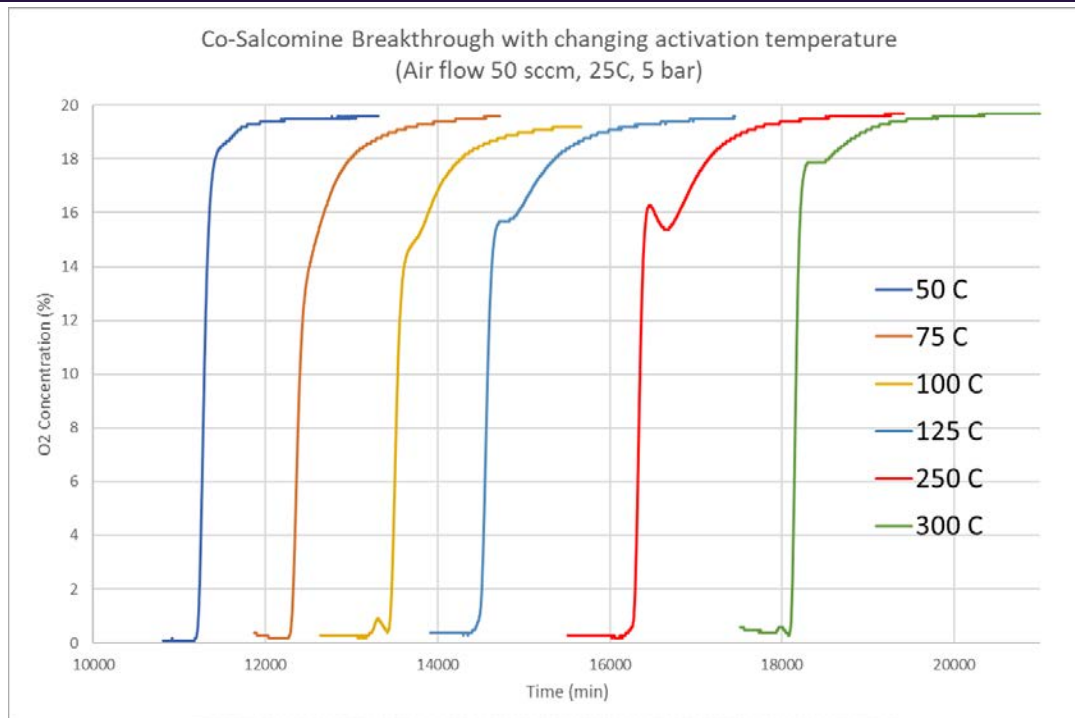
- 40-50 mesh, 0.33mm
- 30-40 mesh, 0.45mm
- 18-30 mesh, 0.75 mm
- **16-18 mesh, 1.06mm**
- **10-16 mesh, 1.51 mm**
- **8-10 mesh, 2.15 mm**



N₂-sorbent

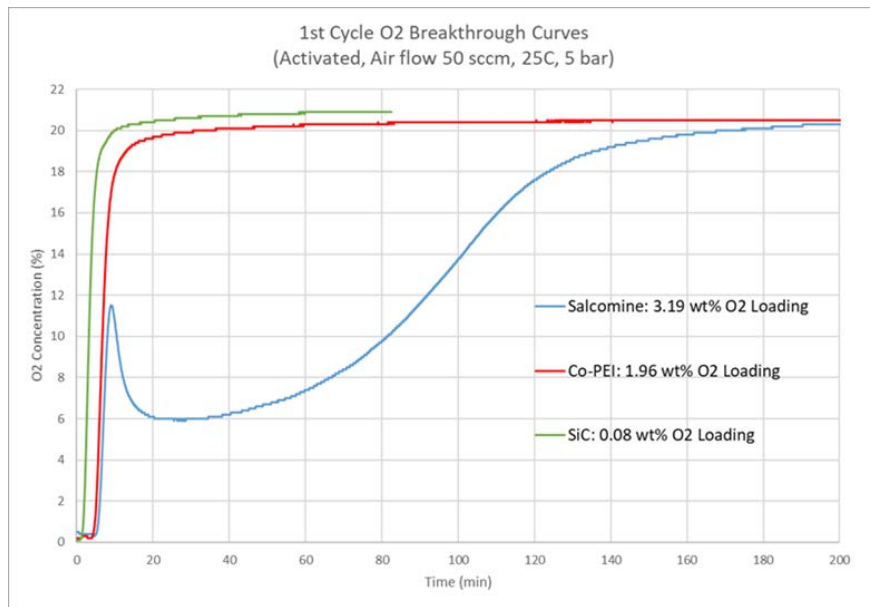
O₂-sorbent

Optimize Activation Conditions for O₂-Sorbent



- Elevated temperature needed to pre-activate Co-complex to promote active sites readily available for O₂ adsorption, while maintaining molecular structure intact

Breakthrough Testing – Manual Mode

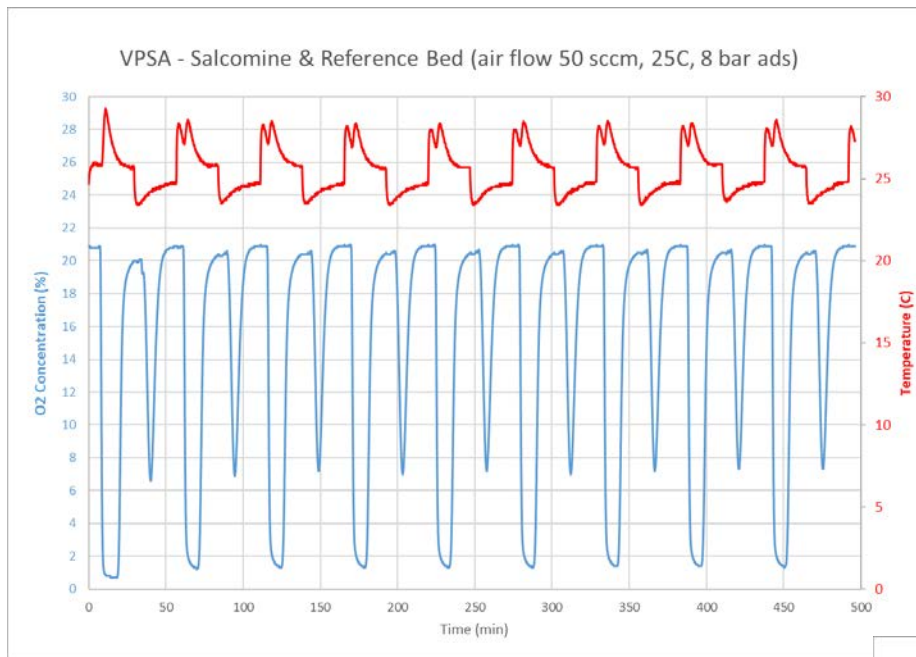


2nd Cycle O₂ Loading:

- Salcomine: 1.09 wt%
- Co-PEI: 0.76 wt%
- SiC (reference): 0.11 wt%

Findings: (1) Commercial Co(II)-salen showed both regenerable O₂ sorption and irreversible sorption; (2) the slopes of reversible portion of breakthrough curves are fast; (3) Confirmed > 0.5 wt% total O₂ loading on both Co-Complex sorbents

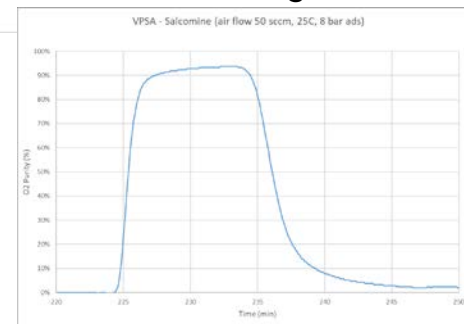
VPSA Cyclic Testing



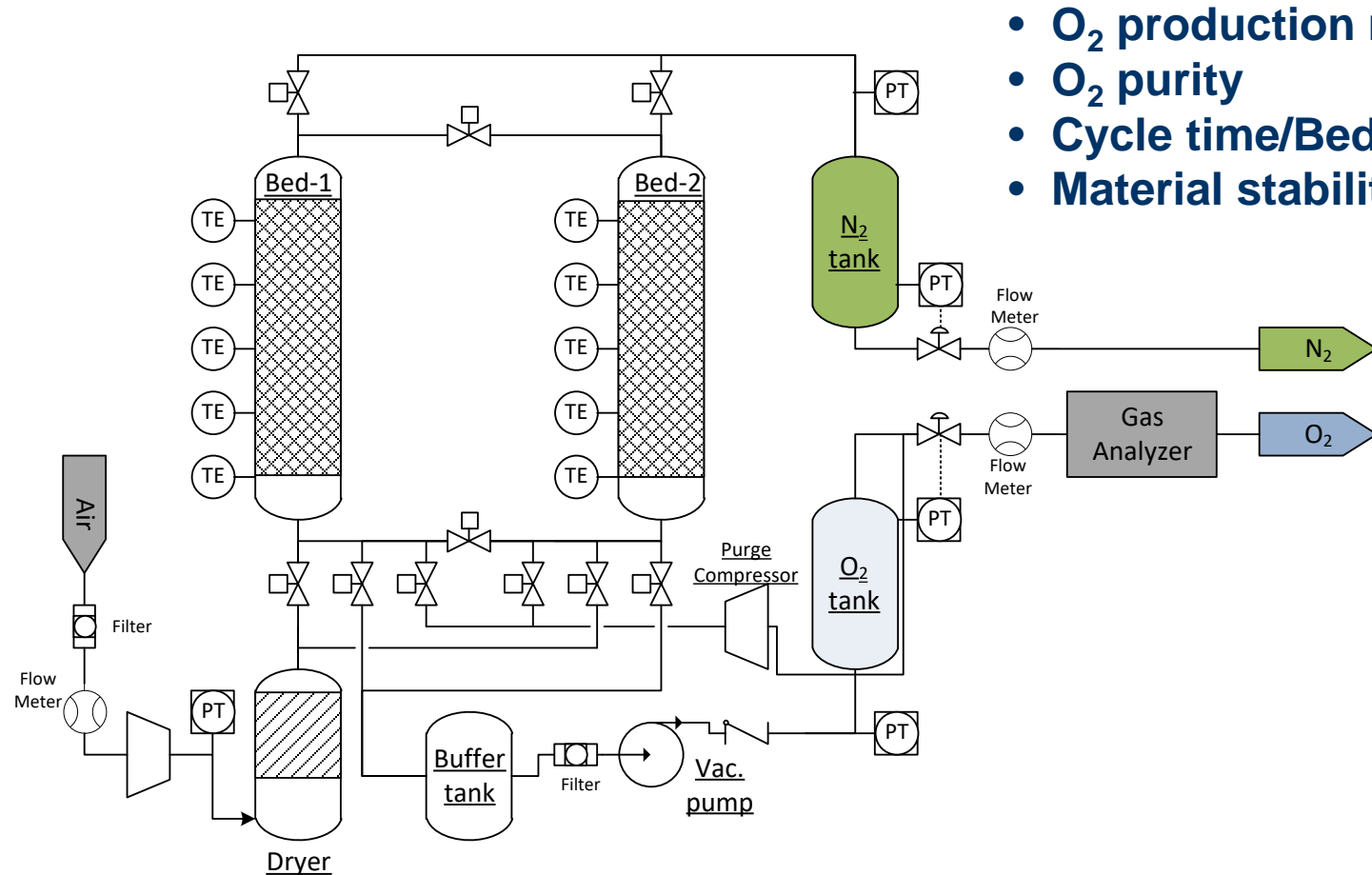
- Reactor temperature profile reflecting the heat of adsorption
- T rose at adsorption step and dropped at regeneration step
- Effluent O₂ concentration showing quick drop when air (21% O₂) passing through sorbent
- O₂ concentration reached 1.3%, indicating 95% product purity in effluent gas



Achieved 95% O₂ purity of cyclic sorption efficiency →



O₂ Pilot Plant Process Flow Diagram



- O₂ production rate
- O₂ purity
- Cycle time/Bed size factor
- Material stability

Technoeconomic Analysis - Results

Parameter	Cost
Purchased Equipment Cost (PEC)	\$ 313,300
Installation Cost (Matls + Labor)	\$ 432,200
Total Installed Costs	\$ 745,500
EPC (8% of TIC)	\$ 59,640
Contingency (15% of TIC+BEC)	\$ 120,770
Total Plant Cost (TPC)	\$ 925,910
Taxes & Insurance (2% of TPC)	\$ 18518
Spare parts (0.5% of TPC)	\$ 4630
Financing Costs (2.7% of TPC)	\$ 25000
Initial sorbent fill	\$ 42,972
Total Owner's Cost	\$1,017,030
OPEX (Electricity costs) (\$8.00/MWh)	\$ 91,250

Levelized Cost of O₂

$$\frac{\text{TOC} * \text{CCF} + \text{VOPEX} * \text{CF}}{\text{O}_2 \text{ Production} * \text{CF}}$$

where

TOC = Total Owner's Cost

CCF = Capital Charge Factor

VOPEX = Variable Operating costs

CF = Capacity factor

Levelized Cost of O₂ = \$59.4/ton of O₂
 SOTA O₂ = \$61.49/ton O₂ in 2016\$

- CAPEX estimated using Aspen Process Economics Analyzer
- Cost of electricity takes from DOE's Annual Energy Outlook, 2019 for electricity costs for industrial sector

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Thank you!

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