

Tuning Surface Stoichiometry of SOFC Electrodes at the Molecular and Nano-Scale for Enhanced Performance and Durability

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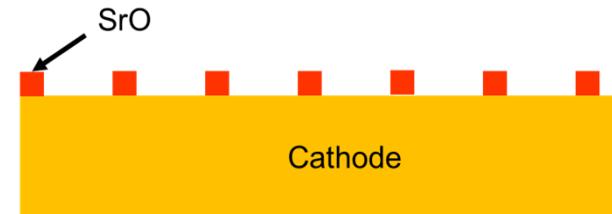
May 1, 2019



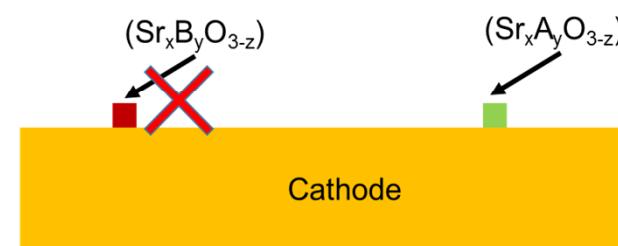
Project Objectives

- Control cathode and anode surface stoichiometry
- Correlate catalytic activity through surface defect chemistry
- Quantify degradation rates/performance and reveal the underlying mechanisms on tuned stoichiometric electrodes
- Utilize thermodynamics to minimize phase segregation (e.g., SrO/SrCO₃) in cathodes and advanced ceramic anodes
- Develop cost-effective and scalable techniques to modify the electrode surface stoichiometry for enhanced performance and long-term stability

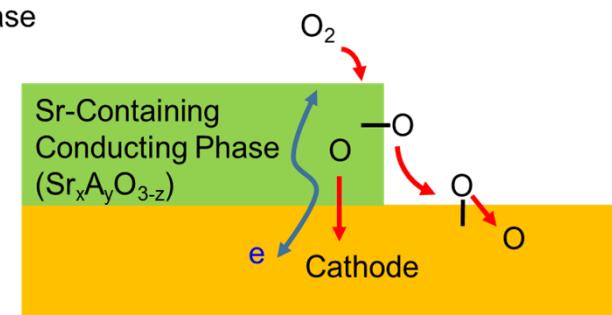
(a) Self-assembled or controlled SrO segregation



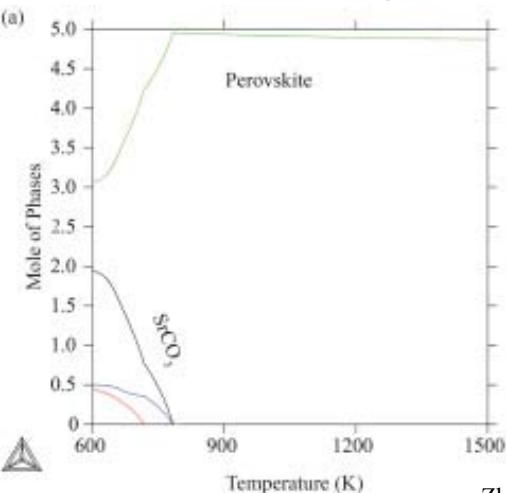
(b) Introduce transition metal dopants to achieve either insulating or conducting phases



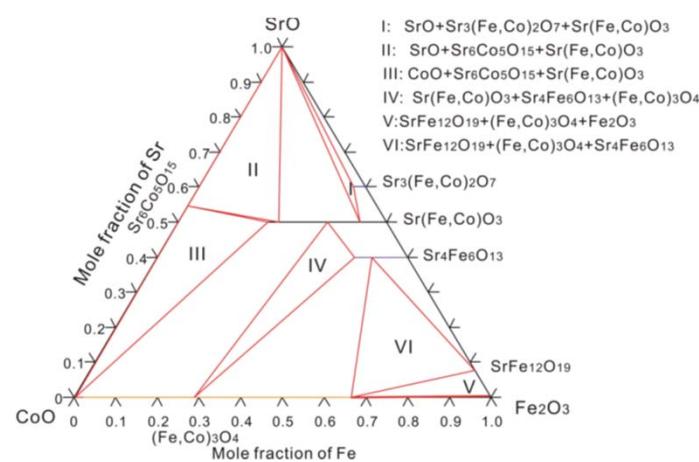
(c) The ORR is enhanced through Sr-containing phase



SrO-LSCF Phase Diagram



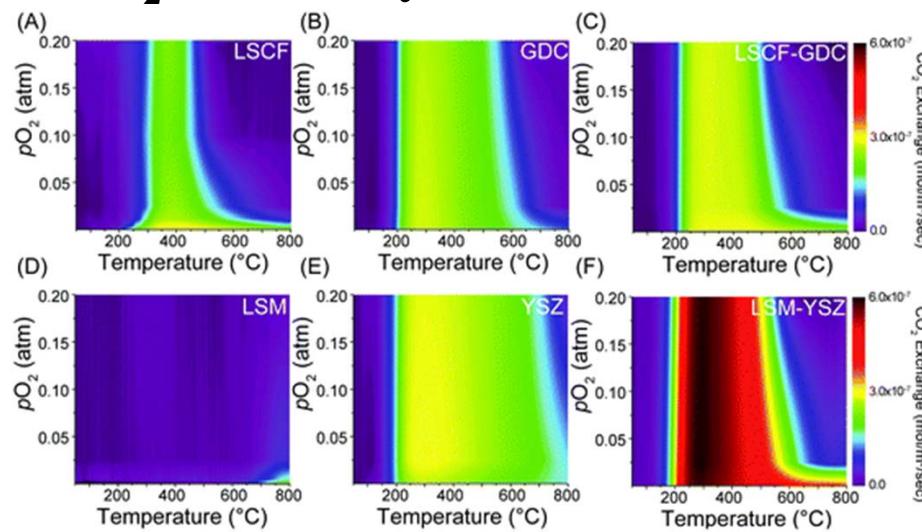
Sr-Co-Fe-O Phase Diagram



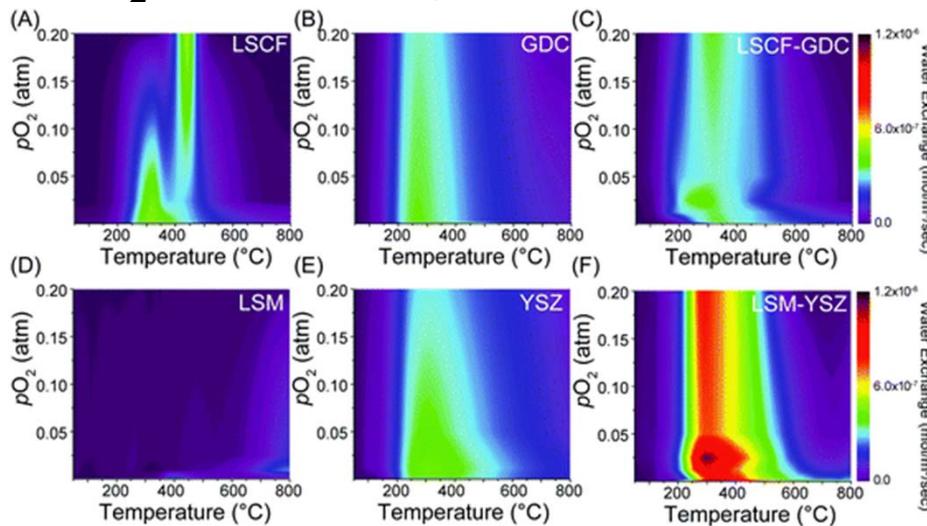
Zhang, W.-W.; Chen, M.; Povoden-Karadeniz, E.; Hendriksen, P. V. *Solid State Ionics* 2016, 292, 88-97.
Darvish, S.; Gopalan, S.; Zhong, Y. *Journal of Power Sources* 2016, 336, 351-359.

Surface Reactivity and Surface Cation Segregation

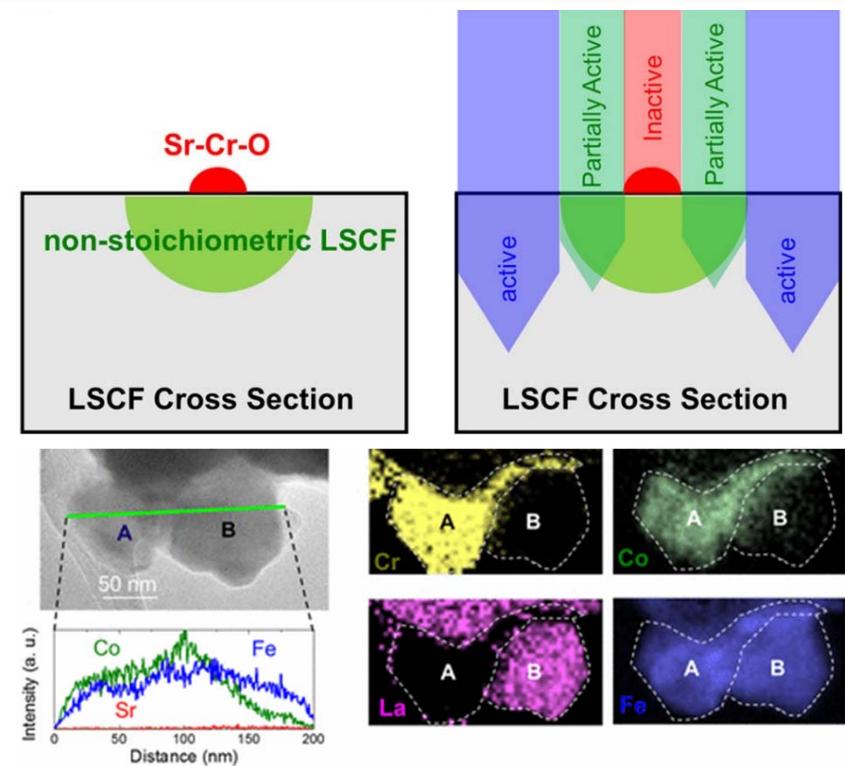
CO₂ Reactivity



H₂O Reactivity



Y.-L. Huang, C Pellegrinelli, A Geller, SC Liou, A Jarry, L Wang, Y Yu, H. Bluhm, E. J. Crumlin, K. J Gaskell, B. W. Eichhorn, E. D Wachsman, *Energy & Environmental Science*, **10**, 919-923 (2017)

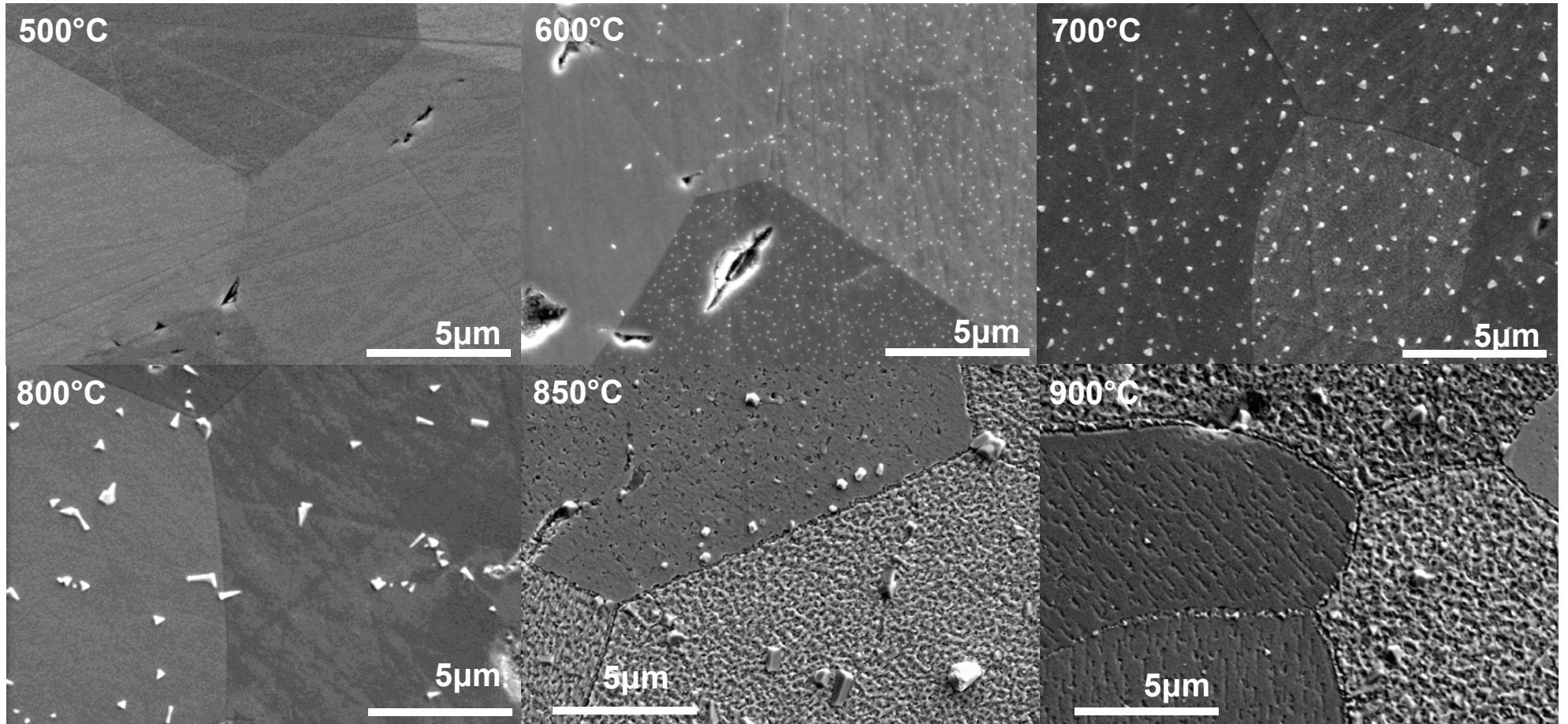


Y.-L. Huang, A. M. Hussain, C. Pellegrinelli, C. Xiong and E. D. Wachsman,
ACS Applied Materials & Interfaces, **9**, 16660-16668 (2017).

- CO₂ and H₂O reactivity on cathodes indicate surface carbonate & hydroxide formation impacts O₂ exchange
- Oxygen transport can be divided into three surface pathways:
 1. Electrochemically inactive pathway (Sr–Cr–O secondary phase)
 2. Partially active pathway (effective region)
 3. Normal active pathway.

Temperature Effect on LSCF Surface Segregation

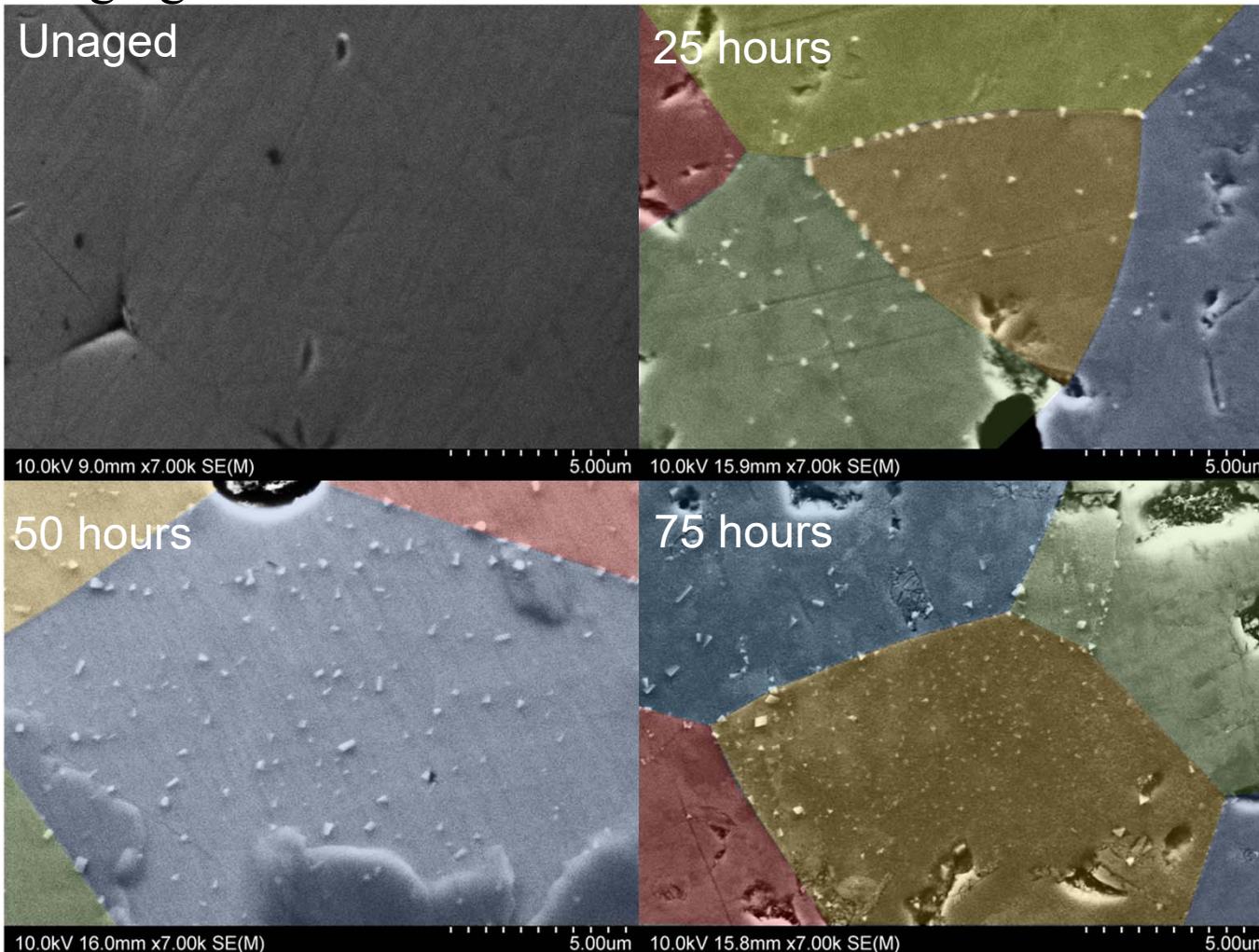
Aging of Dense LSCF Surface in Synthetic Air for 25 hrs



- Increase in temperature promotes surface SrO precipitation (size, numbers)
- Different mechanism observed at $>850\text{ }^{\circ}\text{C}$, porous surface and grain orientation dependence.

Time Effect on LSCF Surface Segregation

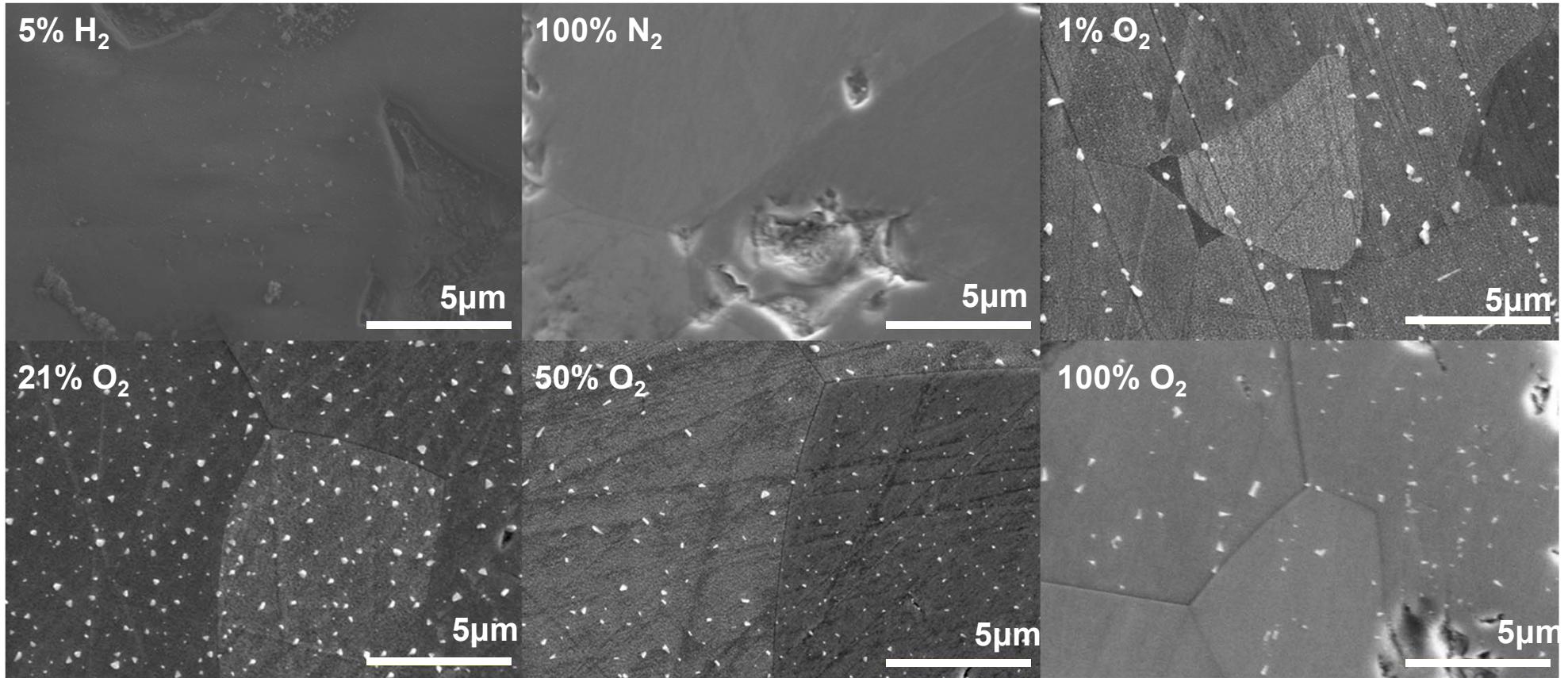
Aging of Dense LSCF Surface at 800 °C in Static Air



- Different grains color highlighted
 - Segregation Process: SrO nucleates at grain boundaries and then migrates to grain center.

$p\text{O}_2$ Effect on LSCF Surface Segregation

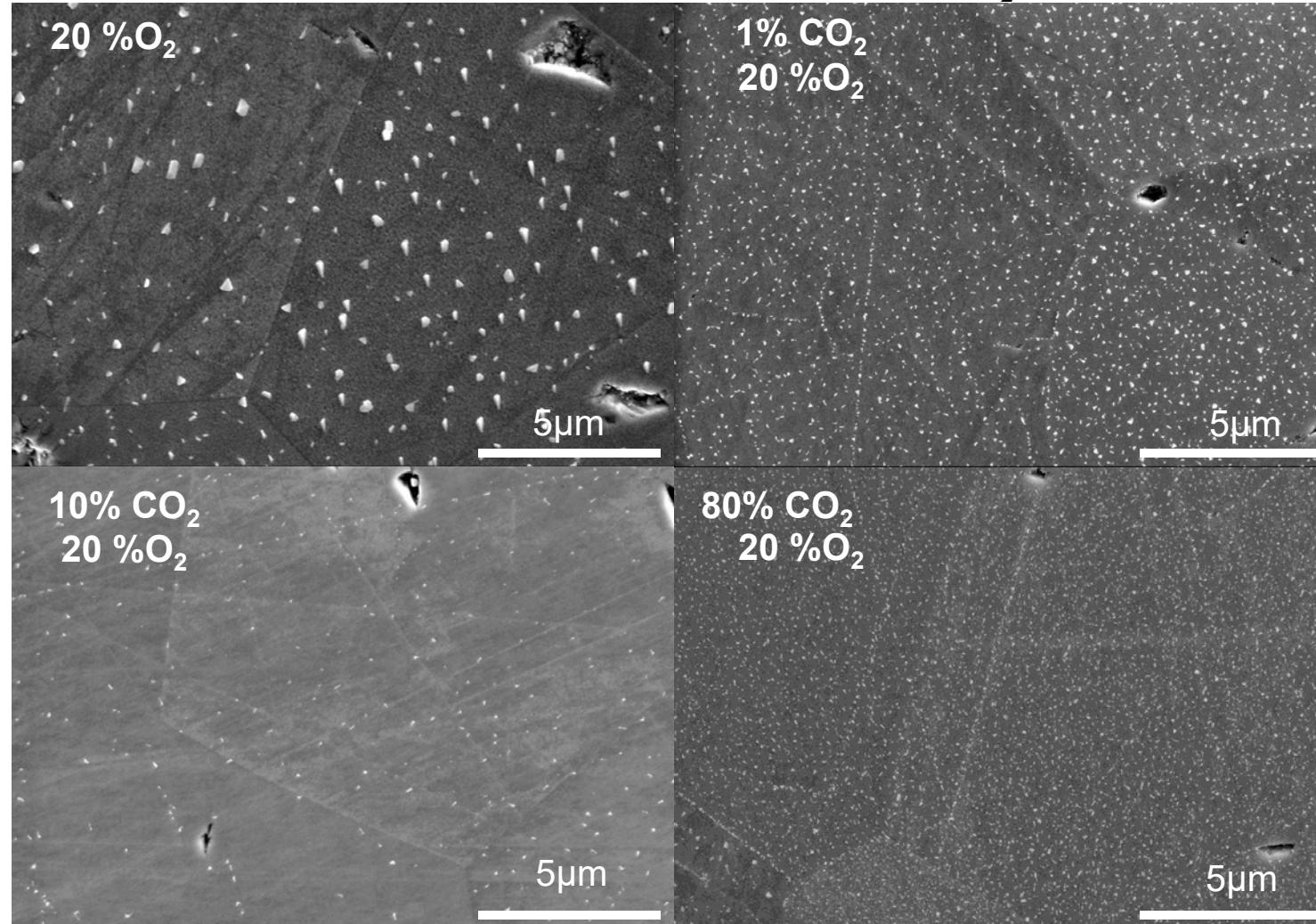
Aging of Dense LSCF Surface at 700 °C for 25 hrs



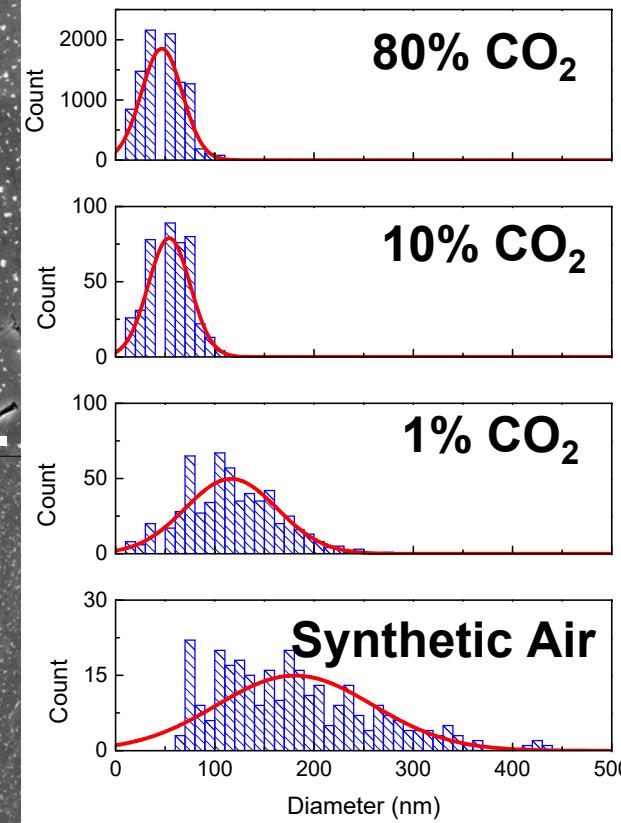
- Increase in $p\text{O}_2$ first promotes SrO segregation up to $\sim 21\%$
- However, further increase in $p\text{O}_2$ suppresses SrO segregation
- Likely correlate to defect chemistry of LSCF

*p*CO₂ Effect on LSCF Surface Segregation

700 ° C for 25 hrs, balanced with N₂



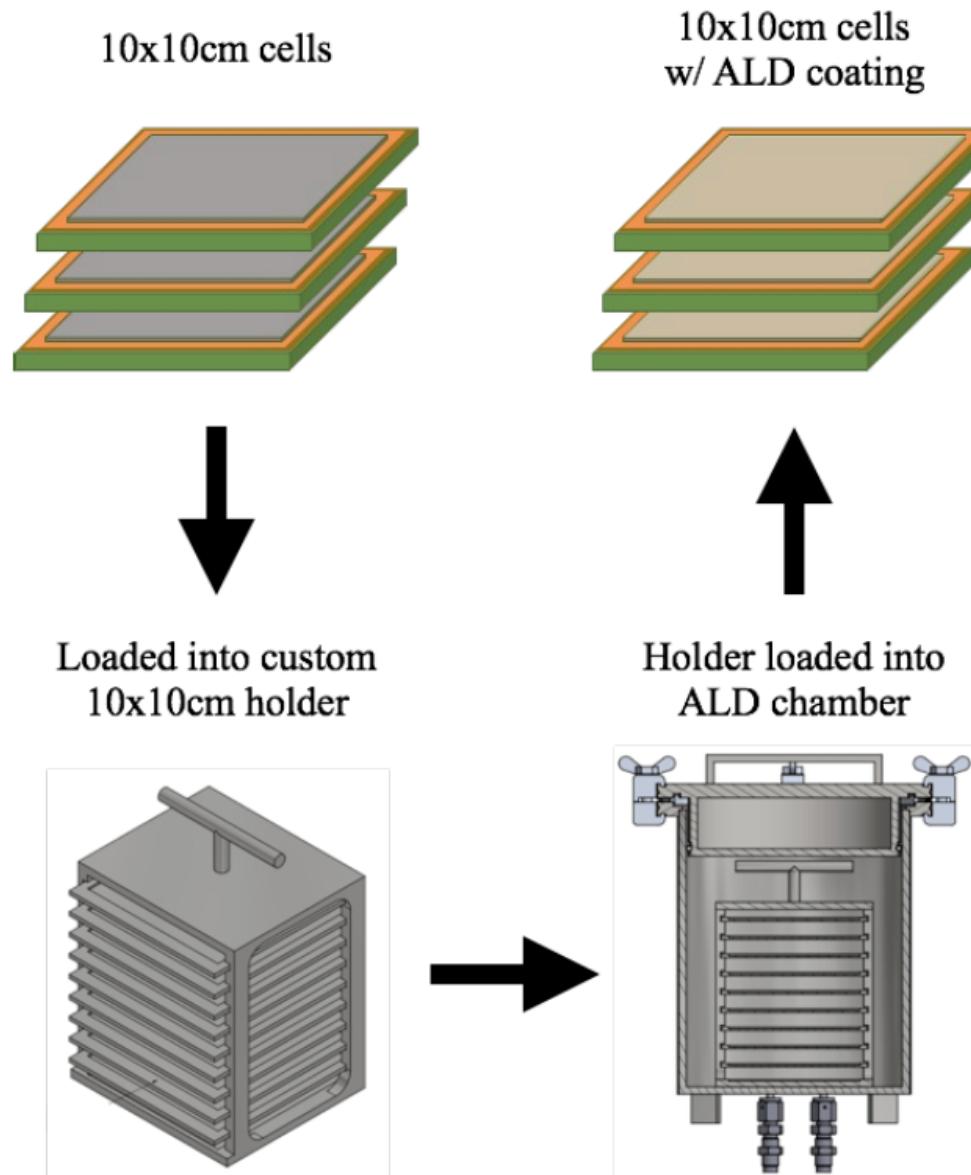
Particle Size Distribution



- Increase in *p*CO₂ decreases precipitate particle size and increases particle number.
- Promotes nucleation and suppress particle migration/growth.



ALD of MO_x Electrocatalysts

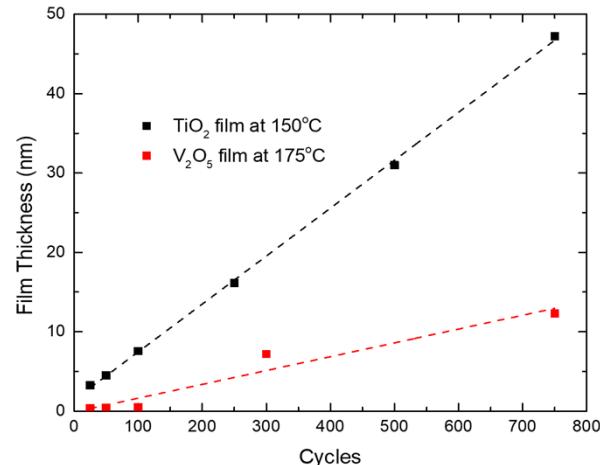


Custom ALD system capable of multiple 10 cm x 10 cm cells



TiO_x and VO_x ALD Surface Modification of LSCF

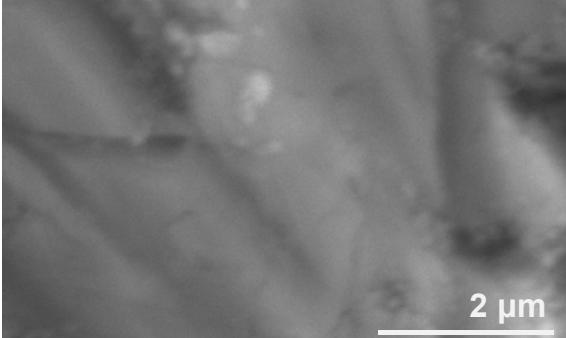
Growth Rate Calibration



- ALD deposited TiO_x and VO_x to form SrTiO_x and SrVO_x
- Growth Rate per ALD cycle shows clear linear trend for Ti but not V
- ALD deposition on sintered LSCF surface shows uniform deposition of Ti but not V

Surface Morphology as a function of ALD cycles and temperatures

Unmodified LSCF



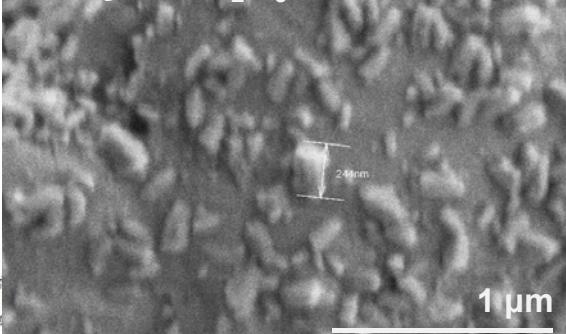
750 cycles TiO₂ – 150°C



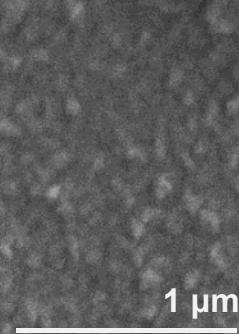
1500 cycles TiO₂ – 150°C



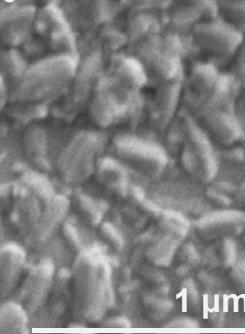
750 cycles V₂O₅ – 150°C



750 cycles V₂O₅ – 175°C

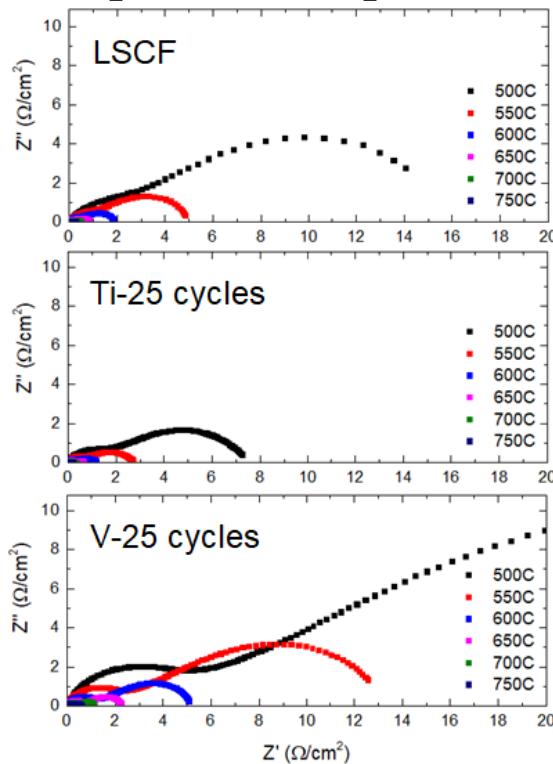


1500 cycles V₂O₅ – 150°C

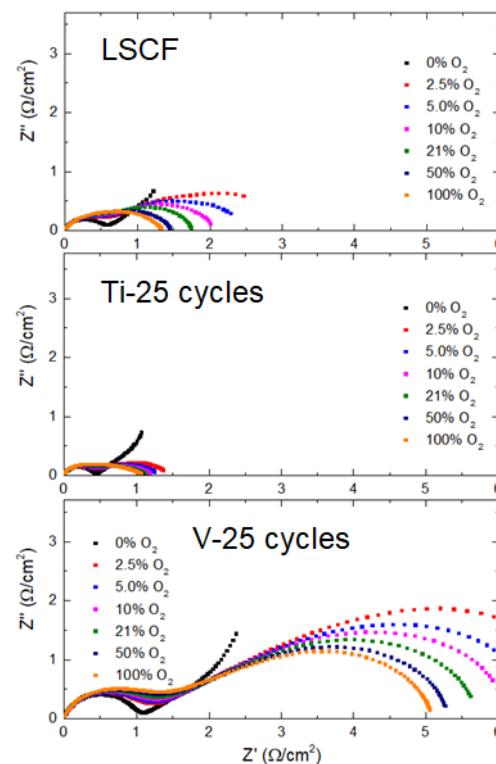


Electrochemical Performance of ALD Modified LSCF-GDC

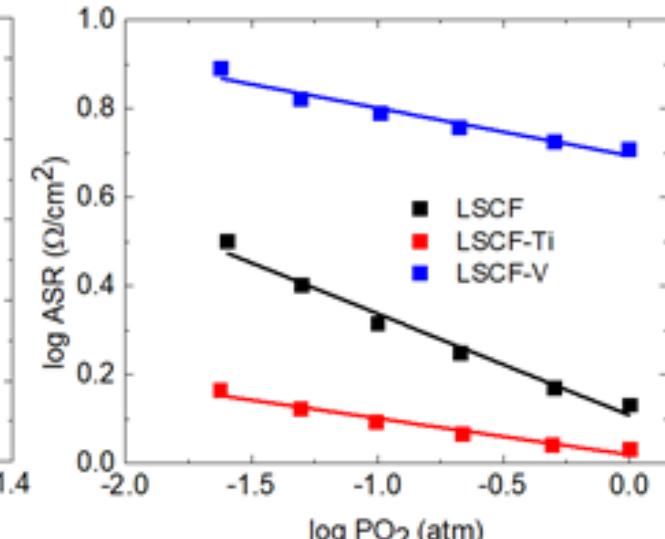
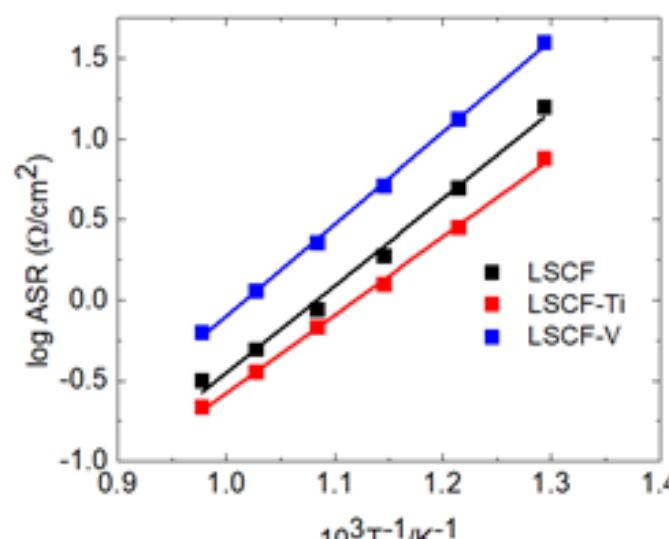
Temperature Dependence



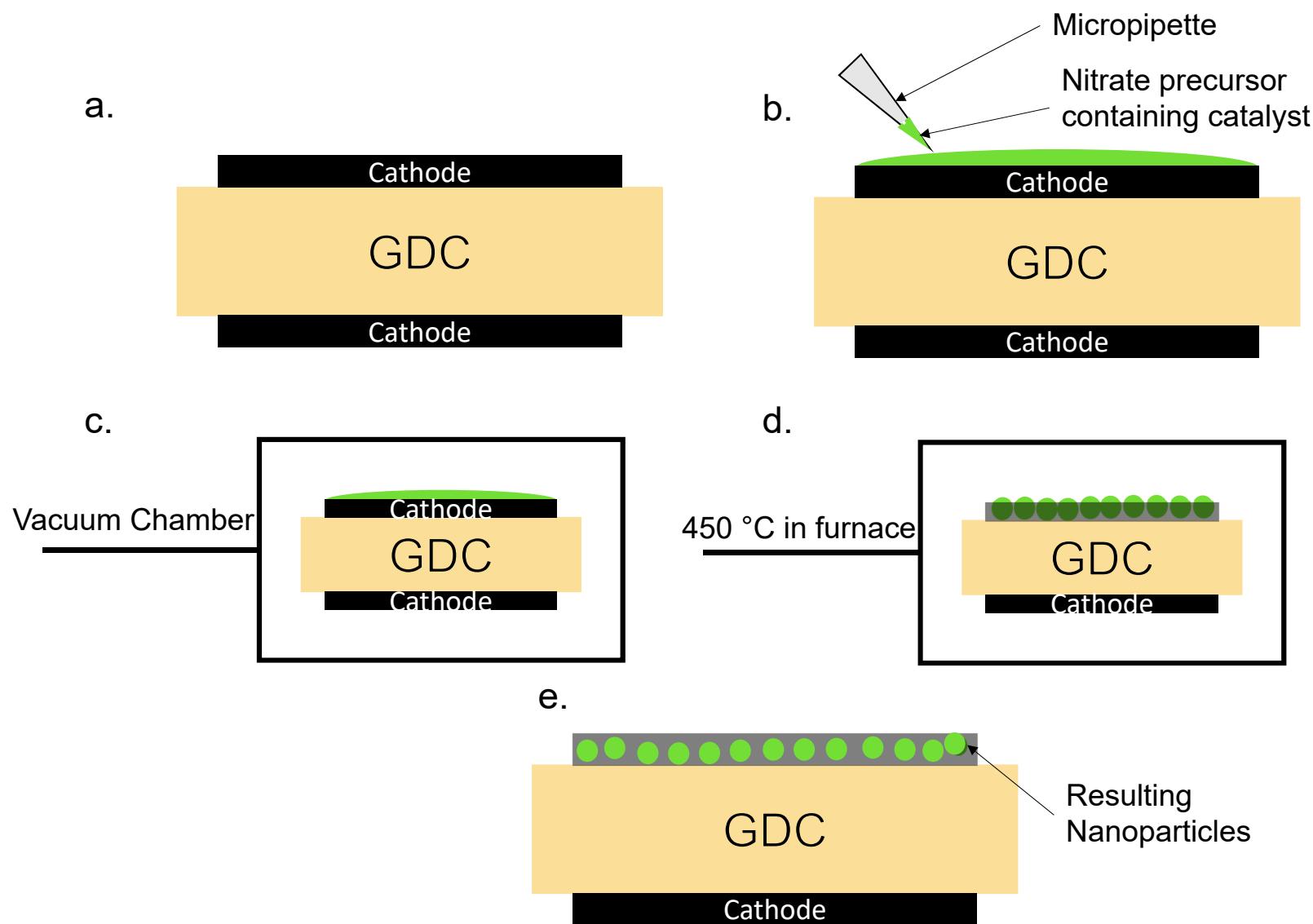
$p\text{O}_2$ Dependence



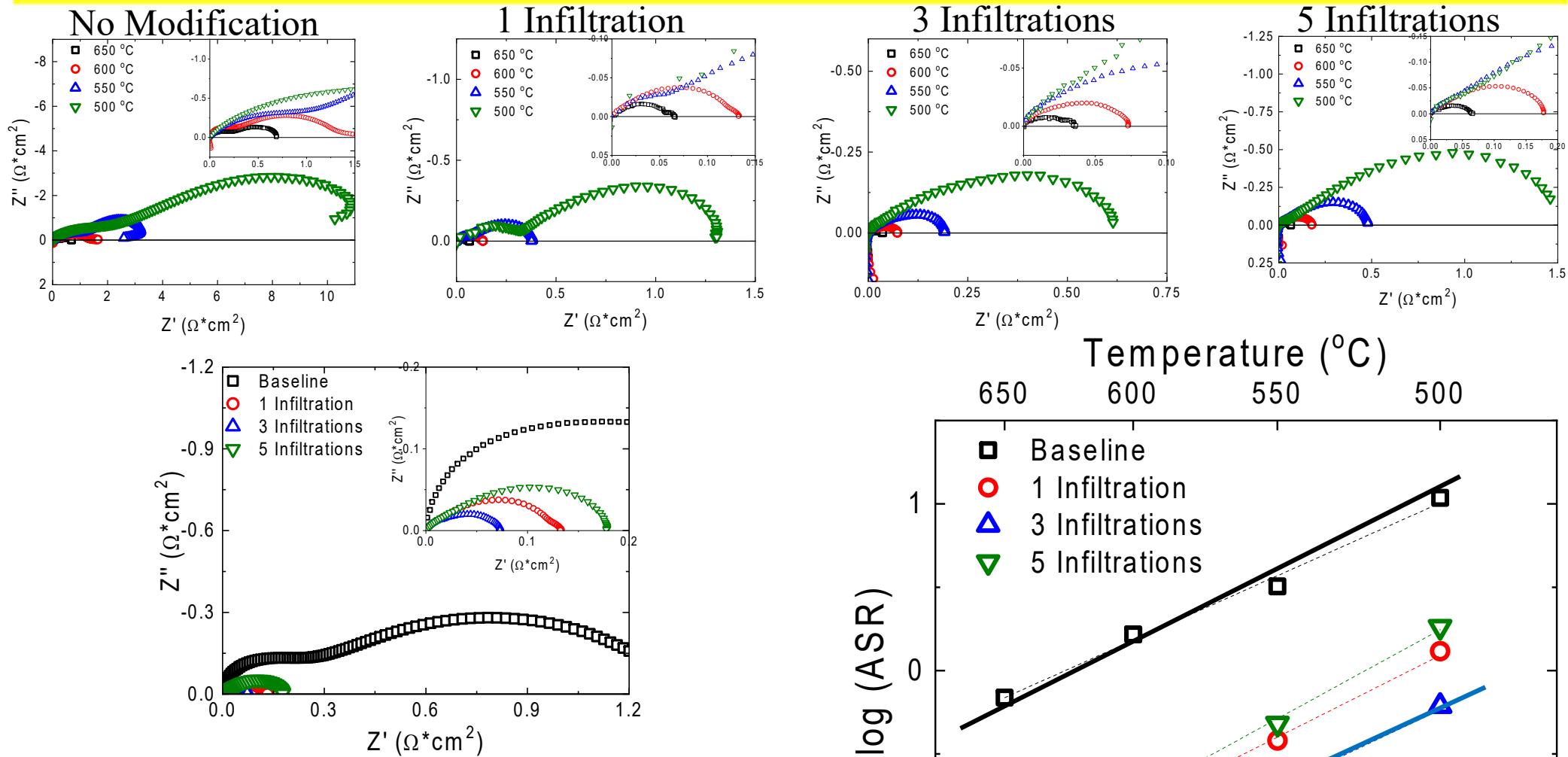
- ALD of Ti significantly lowered LSCF cathode impedance for all temperatures and $p\text{O}_2$ investigated while V increased it



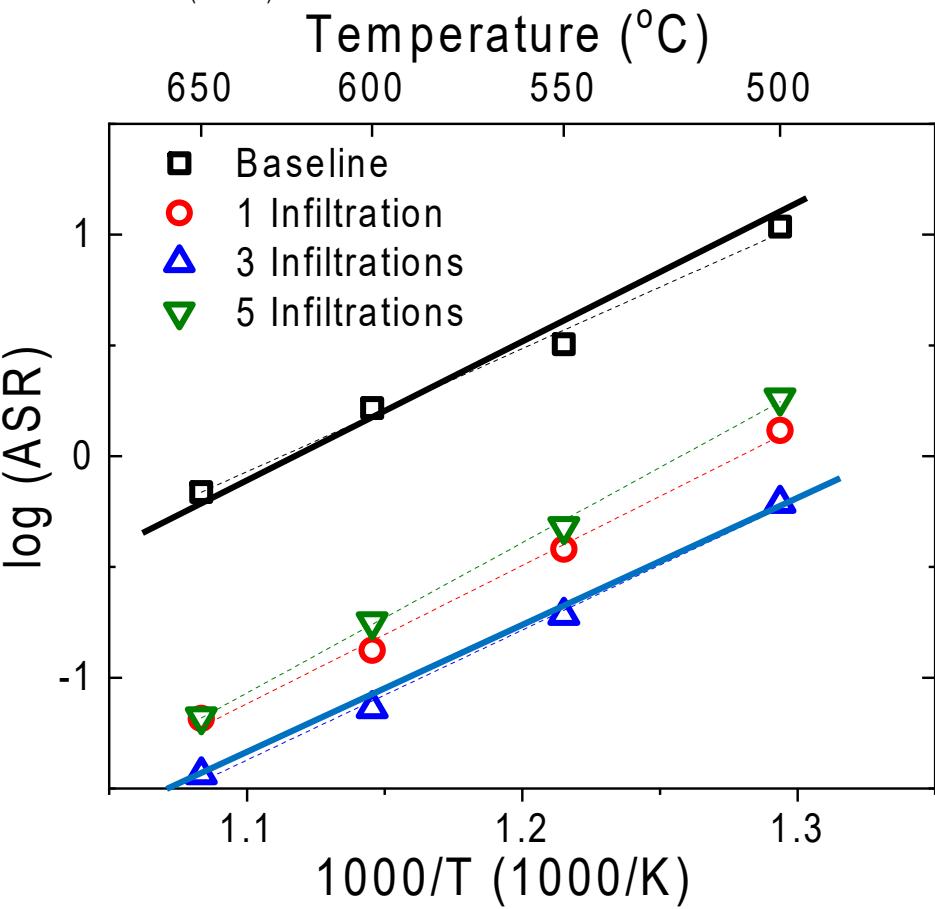
Solution Infiltration of MO_x Electrocatalysts



Solution Infiltration of MO_x Electrocatalysts

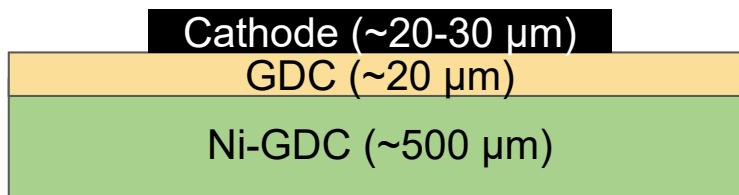
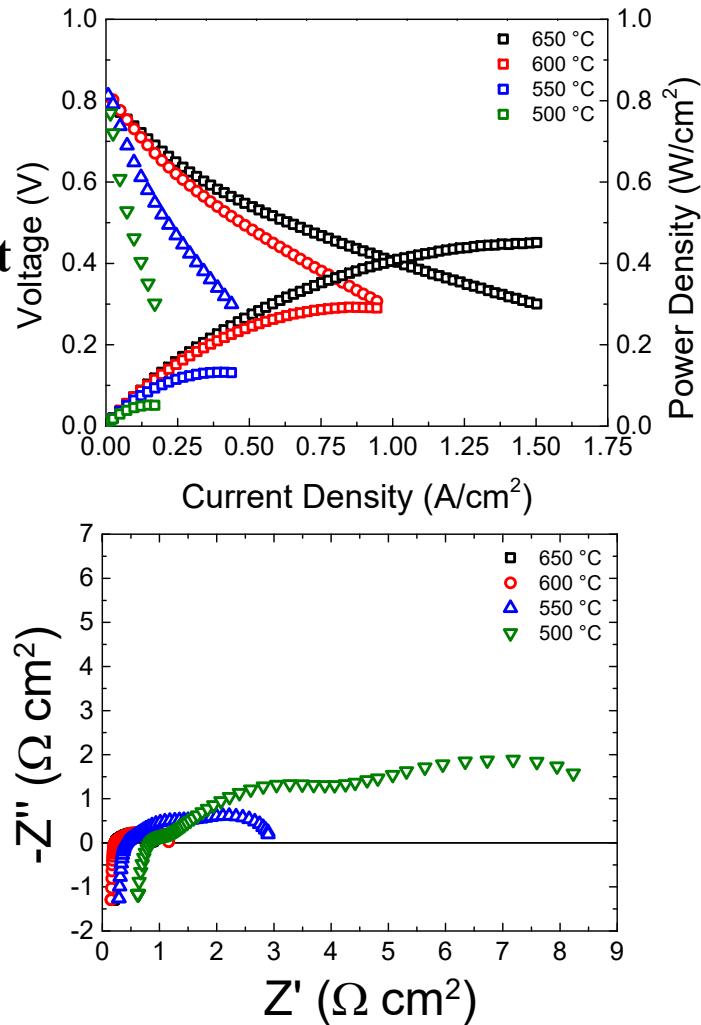


- Modification reduced ASR by order of magnitude
- 3 Infiltrations showed best improvement ASR @ 600 °C of $0.073 \Omega \text{ cm}^2$ (unmodified cell $1.64 \Omega \text{ cm}^2$)
- Higher loading eventually blocks active sites.

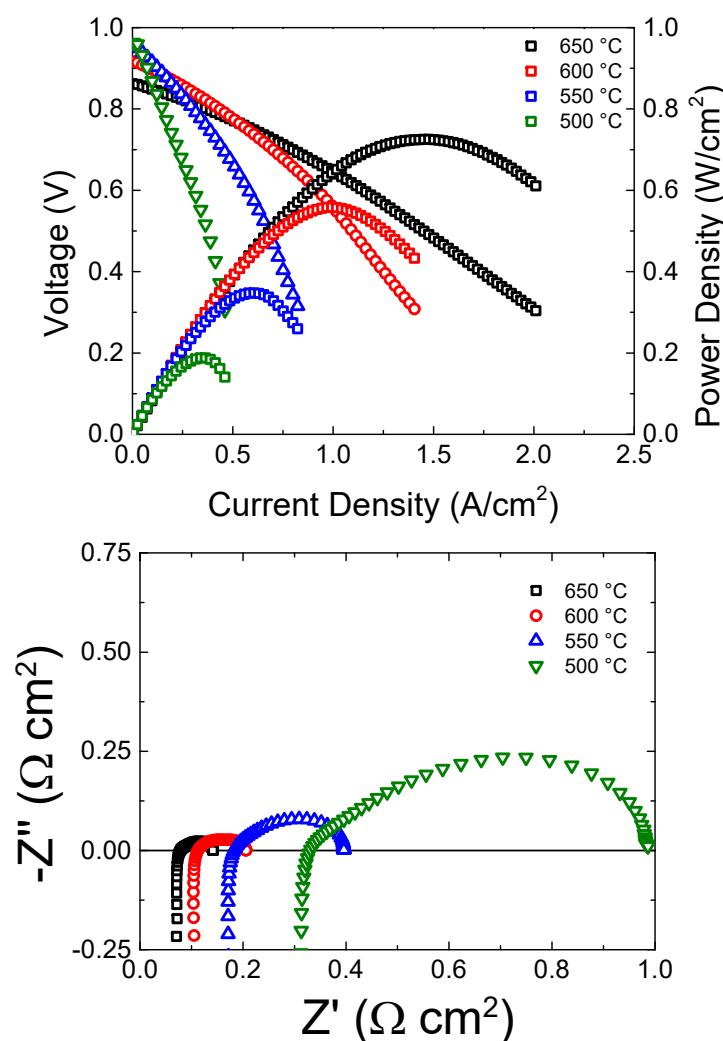


MO_x Surface Modified SOFC Performance

Baseline Cell: LSCF-GDC/GDC/Ni-GDC



Cathode Modified Cell

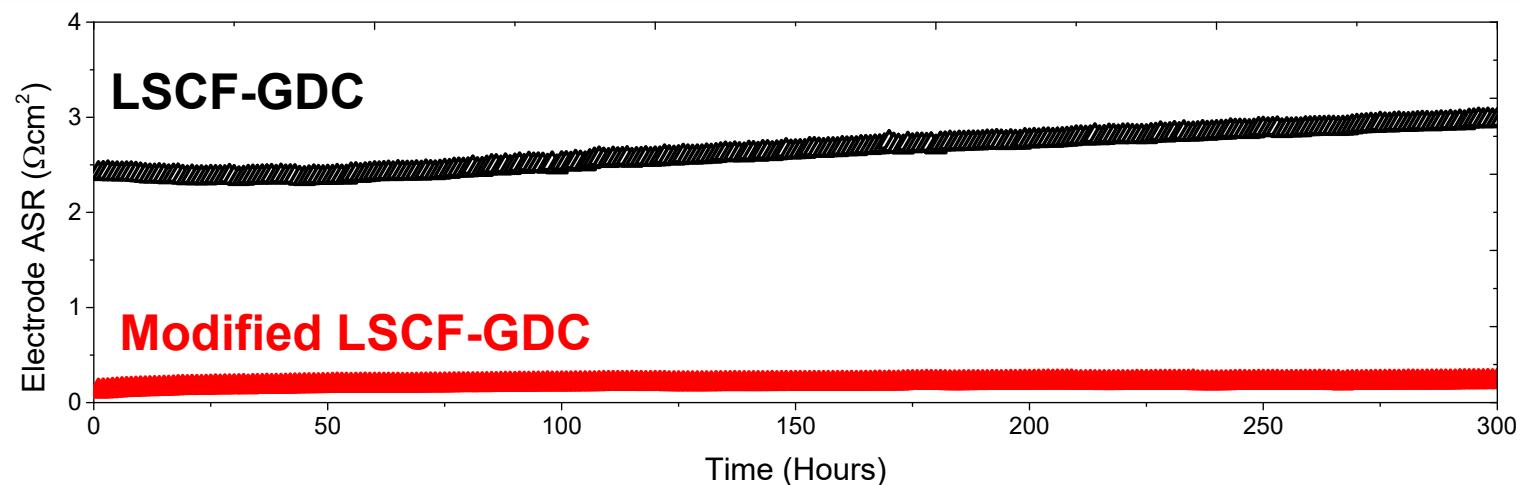


- Higher OCP
- Lower Activation Polarization
- Higher Energy Density

Stability of MO_x Surface Modified Cathode



EIS 600°C air



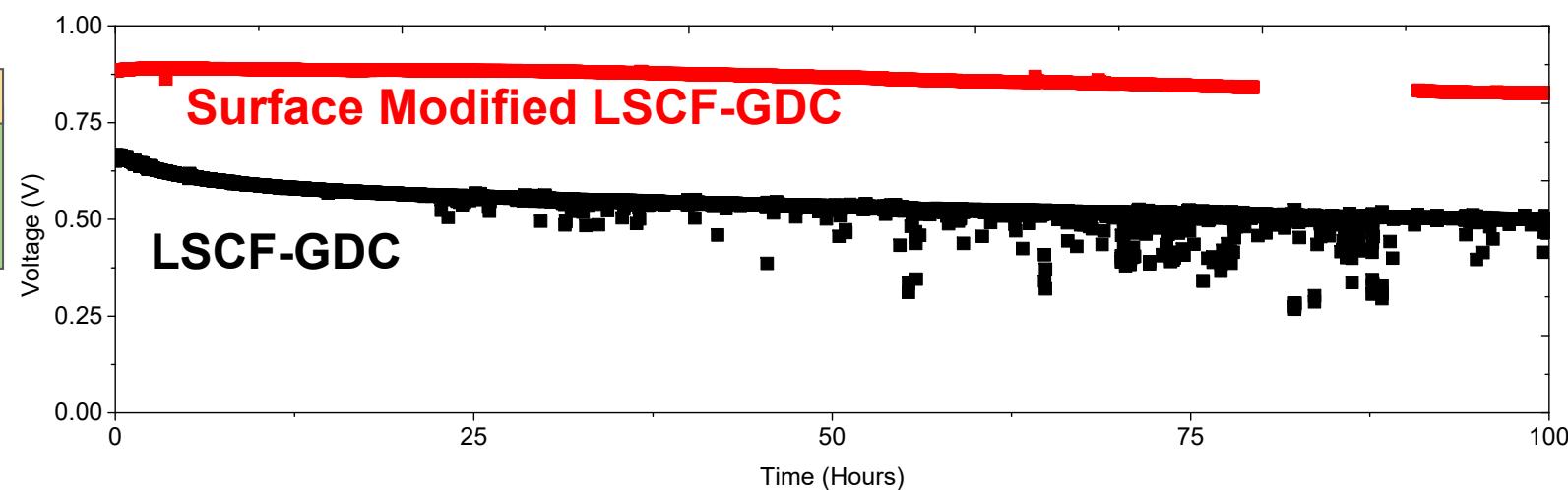
Cathode (~20-30 μm)

GDC (~20 μm)

Ni-GDC (~500 μm)

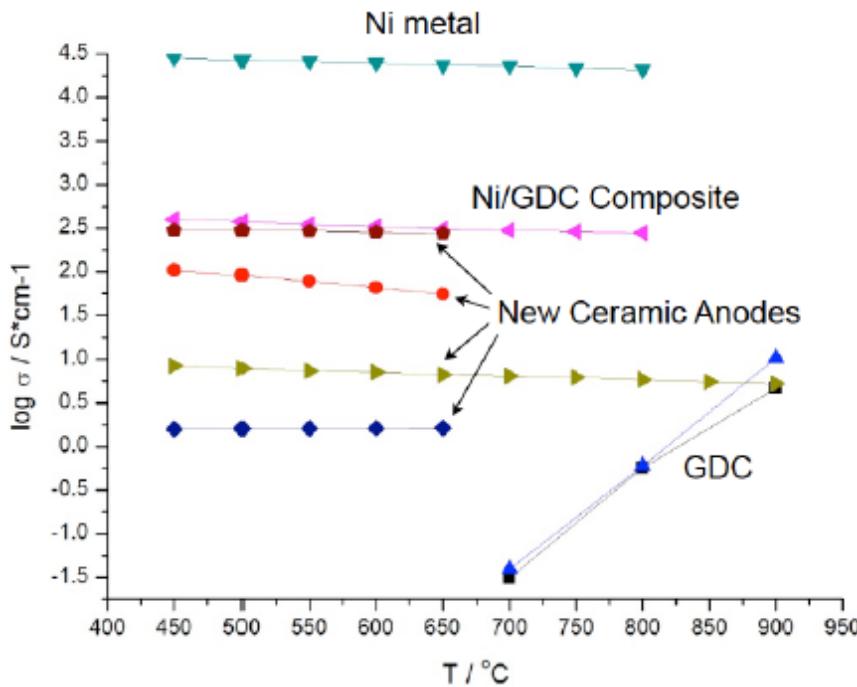
200 mA/cm² 600°C

H₂-3%H₂O/air



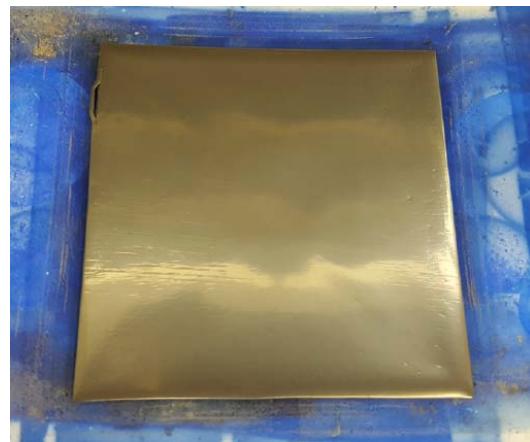
- Higher and more stable voltage at constant current
- Still running

Extend to Redox Tolerant Ceramic Anodes

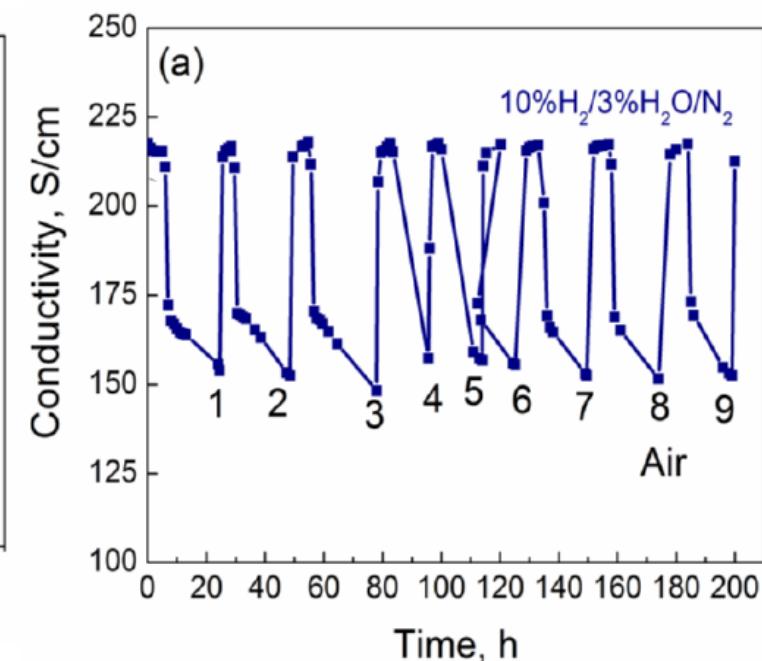
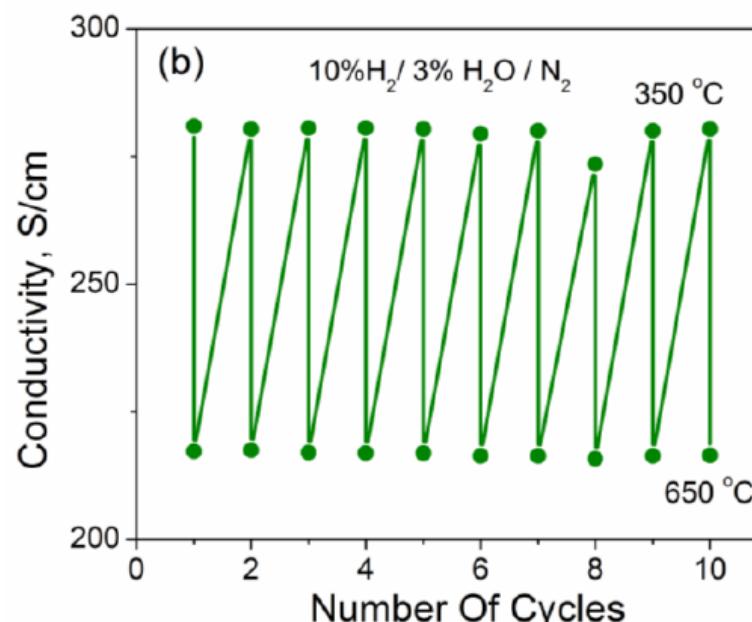


New ceramic anodes developed in our lab:

- Comparable conductivity to Ni/GDC
- Enable thermal and fuel to air cycling
- Will be using above approaches to mitigate Sr segregation issues



5 cm x 5 cm Cell



Acknowledgments

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