



# High-Efficiency Post Combustion Carbon Capture System DOE SBIR Phase 2 Contract # DE-SC0017221 June 26, 2018-May 20, 2020

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#### Outline

- Introduction
- Background
- Phase II Technical Objectives
- Phase II work plan
- Results to date



### Innovation





Compact, modular Post Combustion Carbon Capture System utilizing high internal volume nanosorbents, for carbon capture, supported on a tailorable mesh substrate (Microlith<sup>®</sup>).

Our system enables

- low pressure drop,
- high volumetric utilization,
- high mass transfer,
- lower energy of regeneration.

Capital and operating costs are reduced based on lowered energy demand in capture and sorbent regeneration.



### Microlith Substrate



Microlith mesh coated with sorbent, cell opening 600  $\mu$ m



CFD analysis of fluid flow behavior in 2D cross-sections through a stack of 21 Microlith mesh elements (left), a single channel in a honeycomb monolith (right).

#### **Benefits of Microlith**<sup>®</sup> substrate:

- High surface area per unit volume
- High heat & mass transfer rates
- Low sorbent usage & small size
- Lower ΔP vs. pellet sorbent bed
- Comparable ΔP vs. monoliths
- Design flexibility (e.g., planar, radial)
- Modular design, easily scalable
- Resistive heating



#### **PCCC Unit**





#### **Proposed PCCC module assembly:**

- Two modules working in parallel one adsorbing and one desorbing
- Some water condensed at inlet heat could be recovered for regeneration
- Regeneration heat can be supplied resistively (heating up the mesh) or by heating purge gas with waste heat.



# Phase II Technical Objectives



- Optimize the sorbent to higher capacities with good selectivity towards CO<sub>2</sub>. Sorbents are produced at the large scale (can be easily scaled to tons) with acceptable cost.
- Optimize mesh geometry to achieve higher loadings without affecting sorbent structure at increased production capacity.
- Continue plant modeling from Phase I with steady and dynamic process and CFD modeling of our system to simulate a scale-up design.
- Integrate with DOE Carbon Capture Initiative Software.
- Full scale economic modelling of CO<sub>2</sub> capture with sensitivity analysis and balance of plant components to demonstrate achievement of the DOE target of under 35 \$/ton of CO<sub>2</sub> captured by 2035.

#### PCI Precision Combustion, Inc Sorbent/Microlith Test Results – Phase I



Test assembly for sorbent coated on Microlith mesh for rapid adsorption-desorption capability at low pressure drop; left- micrograph of sorbent coated on support (opening is  $800 \ \mu m$ ).



- Stable performance over multiple cycles of adsorption desorption of CO<sub>2</sub> from simulated flue gas (15 % vol. CO<sub>2</sub>).
- Fast breakthrough and fast regeneration.
- Means for increasing capacity currently being developed at PCI.



### **Phase II Work Plan**



Task I: Sorbent optimization (ongoing): Optimization of a durable, robust sorbent material that can meet the CO<sub>2</sub> working capacity demand.

Task 2: Substrate and coating optimization (ongoing): Demonstrate an adhesive and cohesive robust sorbent-based coating on the mesh substrate that meets target loadings without compromising CO<sub>2</sub> retention capacity.

Task 3: Lab-scale durability and operating envelope (ongoing): Determine a lab-scale operational envelope for the optimized system that meets target capacity is regenerable at low temperatures and demonstrates durability over at least 200 cycles with simulated flue gas.

Task 4: Modelling (ongoing): Develop a CFD model that accurately describes the Microlith-based full scale PCCCS unit with relevant energy and mass fluxes.



### Phase II Work Plan – ctd.



<u>Task 5: Small-scale</u> module assembly and testing (not started): field trial with actual flue gas to demonstrate the efficacy of the system and further refine operating conditions and parameters.

**Task 6: Economic modelling (not started):** Fully develop an economic model that demonstrates we can meet projected cost targets for CO<sub>2</sub> capture 35 \$/ton of CO<sub>2</sub> avoided by 2035.

**Task 7**: Quarterly reports and a final report as required will be prepared. The final report will include recommendations on the feasibility of proceeding with Phase III pilot scale testing and potential for full-scale commercialization.



#### **Results to Date**



**Expand sorbent selection to additional sorbent materials and composites:** 

**Rapid testing of material:** 

- Fixed bed flow-through with simulated flue gas.
- Multi-cycle TGA analysis.

Investigate pathways for rapid large-scale production of sorbent-substrate system with minimal material loss

Set up CFD model of full scale PCCC unit

## **Sorbent Evaluation – Rapid Selection**

# Fixed bed rapid breakthrough cycling with simulated flue gas (capacity) – thermal regeneration at 80 °C



Thermal gravimetric analysis (cyclic stability) Adsorption in  $CO_2$  at 30 °C Desorption in N<sub>2</sub> at 80 °C





Detail of TGA cycle showing fast adsorption/ desorption kinetics Allows for operation at high space velocity (low residence time)

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Axisymmetric model of the adsorber design is being developed in OpenFOAM with the Microlith substrate being modeled as porous media. Phase I experimental data is currently used to validate the CFD approach.

#### Model will yield:

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- > Simulated breakthrough behavior (Exit  $CO_2$  as function of time for provided flow/temperature regimes).
- > Simulated desorption behavior (Exit  $CO_2$  as function of time for provided flow/temperature regimes).
- > Flow, temperature, and species distribution for adsorption and desorption processes modelled.
- Pressure drop as a function of reactor design.
- $\succ$  Simulation campaign to investigate effects of design parameters on CO<sub>2</sub> capture.

CFD model will be integrated with DOE's Carbon Capture Initiative Software.

An economic model of the PCCCS will be developed using the model data to determine the overall cost per ton of CO<sub>2</sub> captured.



## **Phase II Targets**



CASE	No capture	MEA	PCI system
Increase in cost of electricity, %	0.0	63.3	39.2

#### **Targets achieved through:**

- Increased sorbent capacity
- Sorbent stability over multiple cycles
- Low regeneration temperature
- Fast kinetics of adsorption/desorption







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