

Advanced Stripper Gas Produced Water Remediation

Quarterly Technical Report for the Period Ending 12/31/2002

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Abstract

Natural gas and oil production from stripper wells also produces water contaminated with hydrocarbons, and in most locations, salts and trace elements. The hydrocarbons are not generally present in concentrations that allow the operator to economically recover these liquids. Produced liquids, (Stripper Gas Water) which are predominately water, present the operator with two options; purify the water to acceptable levels of contaminants, or pay for the disposal of the water.

The project scope involves testing SynCoal as a sorbent to reduce the levels of contamination in stripper gas well produced water to a level that the water can be put to a productive use. Produced water is to be filtered with SynCoal, a processed sub-bituminous coal. It is expected that the surface area of and in the SynCoal would sorb the hydrocarbons and other contaminants and the effluent would be usable for agricultural purposes.

Test plan anticipates using two well locations described as being disparate in the level and type of contaminants present. The loading capacity and the rate of loading for the sorbent should be quantified in field testing situations which include unregulated and widely varying liquid flow rates. This will require significant flexibility in the initial stages of the investigation.

The scope of work outlined below serves as the guidelines for the testing of SynCoal carbon product as a sorbent to remove hydrocarbons and other contaminants from the produced waters of natural gas wells. A maximum ratio of 1 lb carbon to 100 lbs water treated is the initial basis for economic design. While the levels of contaminants directly impact this ratio, the ultimate economics will be dictated by the filter servicing requirements. This experimental program is intended to identify those treatment parameters that yield the best technological practice for a given set of operating conditions. The goal of this research is to determine appropriate guidelines for field trials by accurately characterizing the performance of SynCoal over a full range of operating conditions.

Executive Summary

Due to a number of factors including the sale of Western SynCoal's parent company and continued transition efforts at NARCo related to its sale to Pan Canadian, no additional experimentation was completed during this period. The project team did meet, reviewed the data collected to date and developed a plan for the next phase of testing.

Work on the project has been hampered by the transition of both NARCo and Western SynCoal into their new organizations after their respective divestiture from Montana Power's group of companies. The overall program is achieving the desired removal of oils and greases from the produced waters but is suffering from inconsistent results leaving questions about the sampling and/or analyses.

Work completed in during the first quarter of 2001 included a second series of filtration tests and air sparging of the filtered produced water to test the ability to further reduce lighter weight hydrocarbon contents. The actual filtration tests were run on twelve (12) selected days between February 1 and 20. The filtering process was typically operated for 6-8 hour period during each day with samples taken near the end of the processing period. These tests used the same SynCoal column that was originally employed in the November testing. The column pressure remained constant throughout the test and the flowrate was held steady at a little over two gallons per minute. Analyses indicated that the hydrocarbon retention was reduced substantially from the initial test although the water clarity

continued to appear greatly improved at the column discharge. Major ions and trace metals were largely unaffected during the testing. Air sparging of the filtered water showed remarkable removal of BTEX constituents and appears to be promising.

Work completed from August 31, 2000 through December 31, 2000 included construction of test columns and support facilities to conduct the preliminary scoping trials to process produced water with SynCoal. The construction, installation, column packing and acquisition of water to be treated required three weeks of effort. The actual test runs ran two consecutive days of constant flow at a rate of approximately three gallons per minute. Three gallons per minute is similar to the rate of water production from the majority of wells in the Denver-Julesberg (D-J) basin. A single SynCoal column with 104 lbs of media was employed for this preliminary test. The column pressure remained constant throughout the test period. Analysis indicated good hydrocarbon retention on the SynCoal averaging 90% or better removal from the produced water. Water clarity was observed to be greatly improved at the column discharge. Major ions and trace metals were largely unaffected by exposure to SynCoal.

Experimental

08/31/2000 through 12/31/2000

After a prototype "filter canister" was designed and fabricated from steel pipe, the project team decided to construct the test units from six inch PVC pipe and fittings and mount them on portable

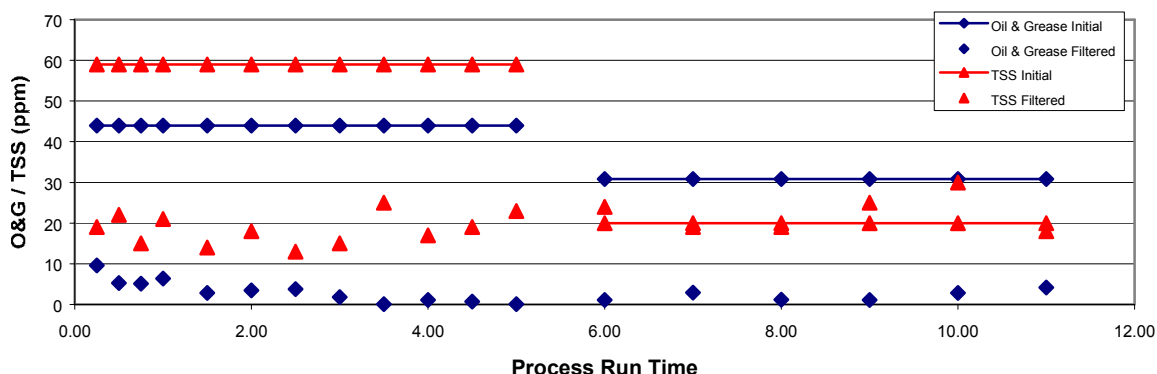
angle iron skid mounted stands to simplify construction, modification and installation. NARCo's Wattenberg shop and lab buildings were cleaned out to provide work space near the production field. In October 2000, fabrication of a two stage filter unit constructed of schedule 40 PVC pipe and related fittings was completed. Initially, some difficulty was encountered in preventing the migration of coal particles from the canister but after trying several screen / packing materials the filter retained the coal without losing particles with the water flow. The filter unit is mounted in an angle iron frame which can be easily be modified to provide an insulated shed around the filter unit with an access door in front. With the filter unit housed in an insulated structure coupled with a small catalytic heater, this unit should prove to be functional in winter as well as summer.

The filter unit was installed into a single state separator at the Ione #11 production battery in an effort to evaluate the functionality of the unit under field conditions. While the unit was determined to be of the correct design for generic connection to a separator unit, the surge discharge pressure of the single stage unit at the Ione battery was greater than anticipated (+ 200 psig). In order to use the PVC

filter unit on locations with pressures in excess of 50 psig, a surge tank is needed to control the impact force of the water discharged to the filter unit.

The filter unit was moved back to the Wattenberg location where it was installed with a surge feed tank, a pump to supply a steady flow to the filter and a "clean" water tank to characterize the filter. The initial testing was conducted on November 1 and 2, 2000 with water flowing through the filter for approximately 12 hours at a 3.0 to 3.5 gpm rate. Pre-treatment and post-treatment water samples were collected and analyzed. It appeared from visual observation of the pre-treatment and post-treatment water samples that the filter unit is removing insoluble and colloidal particulates in the produced water being treated. The analyses indicated that the oil and grease content was reduced from an average of over 36.8 mg/l to under 2.4 mg/l, approximately 93 percent reduction. Total suspended solids were reduced from 59 mg/l in the initial sample to under 20 mg/l on average. Copper and silver concentrations appeared to be reduced as well. Unfortunately, the sodium, calcium, magnesium and potassium concentrations did not appear to be affected.

Preliminary Column Test



01/01/2001 through 03/31/2001

In reviewing the data, it was postulated that the remaining oil and grease concentration was primarily light hydrocarbons that could be removed by simple air sparging. Additionally, concepts of using SynCoal with a high limestone content as the filter media may enhance the removal of the dissolved salt mineral content. Further testing to determine the filter material's saturation point, if the high limestone content SynCoal would remove more the the dissolved solids and if the remaining oil and grease levels can be removed by air sparging was determined to be an appropriate next step.

In late January, work was completed to winterize the test filter unit and a 110 volt $\frac{3}{4}$ h.p. electric centrifugal pump with a by-pass system and a 1 inch brass Neptune water meter with totalization capabilities were installed.

A second series of flow tests were conducted on the 1st stage of the prototype canister filter unit. The

primary objective of this test series was to determine the point of climax or saturation of the filter media.

From February 2 to February 20, the unit was tested in 6-8 hour sequences on 12 individual days. During that test period, 3 different batches of untreated water were received at the test facility and processed through the filter unit. A total of 10,520 gallons of produced water was processed through the stage one filter at an average inlet pressure of 10 psig and an average flow rate of 133.5 gallons per hour or 2.23 gpm. A sample of the treated and untreated water was collected initially and at the end of the first days' run; and at the end of each days run thereafter. The samples were placed in glass jars and labeled. As the attached data indicates, each new batch of untreated water presented a different set of organic concentrations. In an attempt to minimize the impact of this fluctuation in the chemical makeup of the water, the sample protocol for this phase of testing provided for the collection of a "in or untreated" sample at the beginning of the daily flow test

and an “out or treated” sample at the ending of the daily flow test. The water quality changed with each new batch of untreated water. The following is a brief summary of the organic concentration.

As part of the second stage testing protocol, the treated water from the filter unit was placed in a 100 barrel steel

water tank recently removed from an operating production facility. The tank was not cleaned prior to use in this test in an effort to maintain realistic field conditions. An air sparging unit was fabricated and air at 20 psig continuous flow was sparged through approximately 80 barrels of treated water.

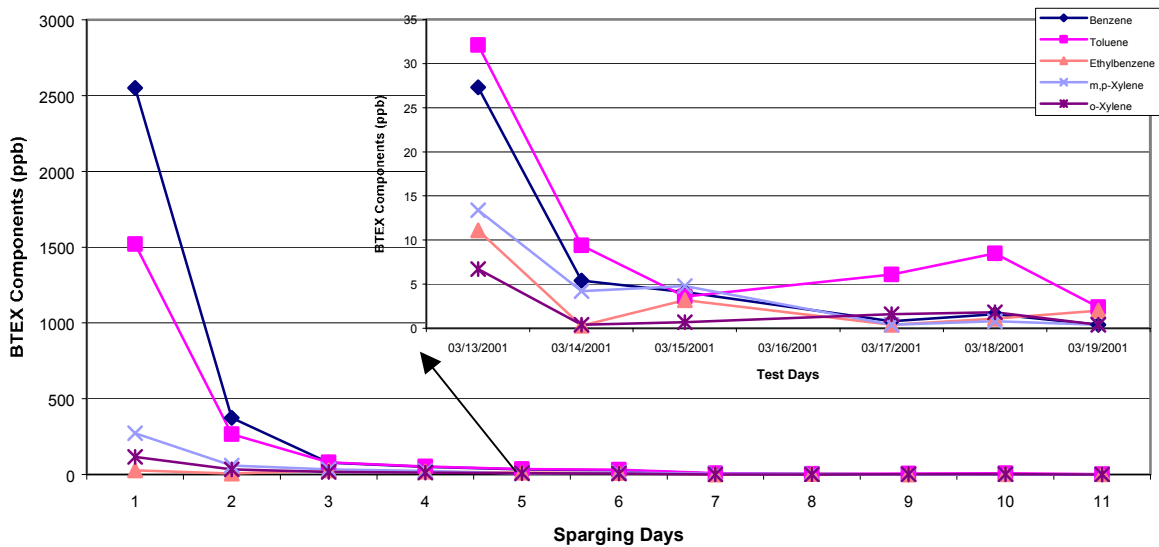
Second Test Series Summary Results

Sample ID	Untreated Oil & Grease ppm	Treated Oil & Grease ppm	Removal Efficiency	Untreated BTEX ppm	Treated BTEX ppm	Removal Efficiency
Batch 1 Ave	126.5	63.5	50%	1055	908	14%
Batch 2 Ave	228.6	234.8	-3%	2071	1685	19%
Batch 3 Ave	5.6	7.9	-41%	4462	4820	-8%

Sparging Test Results

Sample ID	Benzene (ppb)	Toluene (ppb)	Ethylbenzene (ppb)	m,p Xylene (ppb)	o-xylene (ppb)
Initial Sample	2550	1520	28.5	272	117
5 th Day Sample	33.2	35.5	12.8	6.0	8
11 th Day Sample	<0.4	2.4	2.0	0.4	<0.4

Sparging BTEX Removal



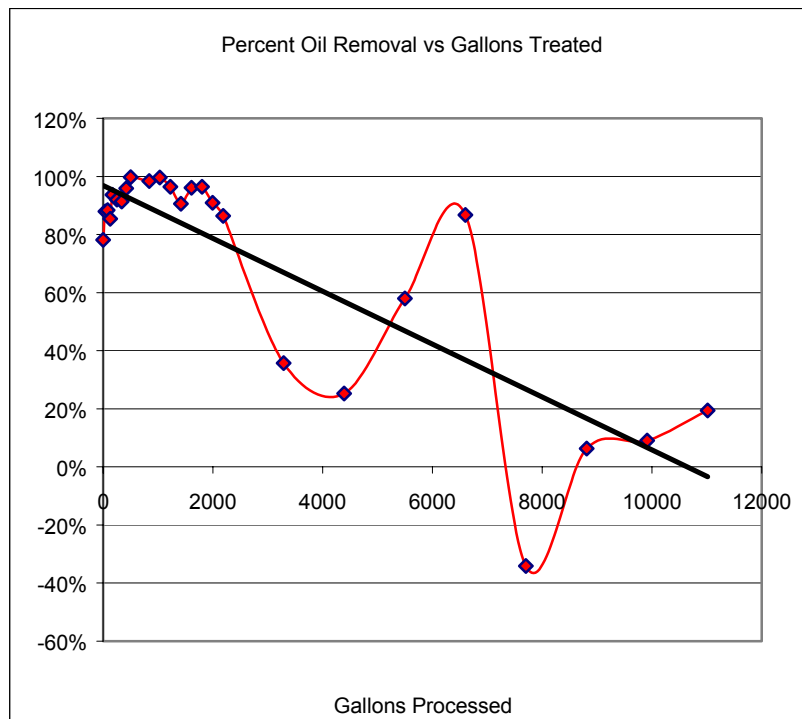
This data indicates a dramatic decrease in the BTEX concentration of the treated water in a very short period of time and under less than ideal conditions (cold temperatures and limited venting of the tank). Air sparging was conducted from the morning of February 27 through the evening of March 3 when the air sparging operation was suspended after the 5th day pending test results. After receiving the data, the air sparging tests were continued beginning March 13, 2001 and concluding March 19, 2001.

The second series of BTEX analysis on the 80 barrel batch of filtered and sparged produced water was completed and the results indicate that air sparging of the water filtered by a single stage of the test filter unit has removed all of the

enzyme and o-xylene with only a trace of the toluene, ethylbenzene, and m,p,-xylene all concentrations of which are well below any action levels.

The inorganic test data is confusing with regard to the effects the filtering process has on the inorganics in the water sampled. Of the salts, the sodium and potassium numbers are significant and appear to increase. Heavier metal seems to be decreased. The remaining inorganic numbers are insignificant. This raises the possibility that the treated water, once subject to air sparging to remove the VOC's could be recycled as makeup water for frac jobs or drilling fluids.

Lifetime Efficiency of the Initial Test Column



4/1/2001 through 6/30/2001

Due to a number of factors including the sale of Western SynCoal's parent company and continued transition efforts at NARCo related to its sale to Pan Canadian, no additional experimentation was completed during this period. The project team did meet and evaluate results to date. These efforts resulted in a plan for the next phase of testing which is anticipated to involve the following tasks:

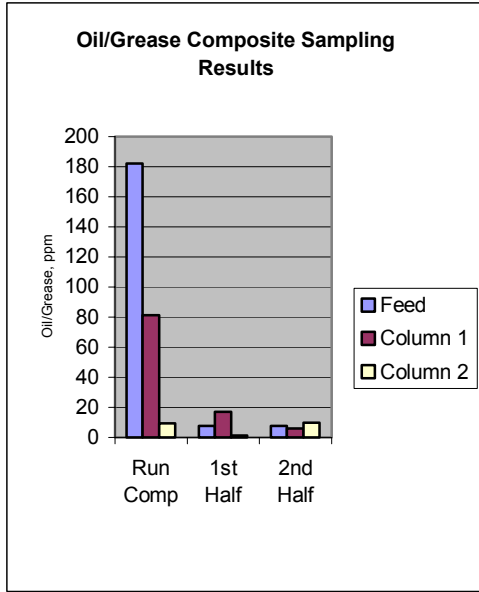
- 1) Reloading the twin filter canisters with SynCoal in the first stage canister and dolomitic SynCoal in the second stage canister.
- 2) The air sparge tank will be steam cleaned to remove any historical oil/grease residues.
- 3) The produced water from a single well unit will be used in this test protocol.
- 4) An initial water sample of the untreated produced water will be collected and analyzed for inorganics, organics, TRPH, and BTEX concentrations.
- 5) The filter/sparge system will be set up on a timer to operate 4 hours per 24 hour period. The flow rate will be established at 2 gallons/minute or 120 gallons/hr, or 480 gallons/day. Assuming an initial untreated volume of 80 bbls, the test will run for approximately 7 days.
- 6) At the end of the batch testing period, the same analysis profile as is listed in "4" above will be run to

provide the treated numbers for each batch.

- 7) The test protocol will be repeated with each 80 bbl. Batch of water taken from the designated well unit in an effort to determine the saturation point of the filter units. The marker to be used to determine the saturation point will be the TRPH analysis.

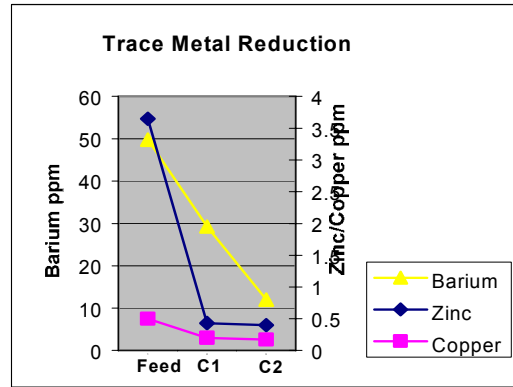
7/01/01 to 9/31/01

Test work this quarter focused on both sampling technique and preliminary evaluation of SynCoal tailings material to address inorganic contaminant reduction. A two column test apparatus was configured with 110 lbs of SynCoal Product in the first column and 120 lbs of SynCoal tails in the second column. SynCoal tailings contain partially calcined dolomites and stable massive pyrite, previously tested in weathering tests to exhibit a buffering capacity towards both pH and trace metals. Composite sampling was attempted to reduce the risk of previously encountered inconsistent results. The duration of the test was 6 hours at a feed rate of 3 gpm. The following graph shows the results from this sampling.



The first series of columns (Run Comp) is the analysis of composite sample increments taken hourly over the duration of the entire six hours. The second series of bars is a composite of hourly increments taken during the first half of the test run and the third data set shows a composite of hourly increments from the second half of the test run. Consistency of sampling and analysis remain problematic.

Major cation/anion reduction was not achieved by using the SynCoal tails in column 2, as the analyses indicates a slight increase in sodium and calcium. Trace metal analysis indicates a significant reduction in zinc, copper, barium, aluminium, cadmium, and chromium. Dissolved chloride appears unchanged.



Analytical results are shown in the appendix.

Review of the data sets to date still indicate a problem remains in sampling and/or analysis. The next investigation will configure four columns of SynCoal in series, the discharge of each to be sampled hourly with the goal of defining the hydrocarbon loading capacity of SynCoal tracing the “breakthrough” of hydrocarbons through a four column series.

Other activities during this quarter include the preparation of this projects presentation for a meeting of the Petroleum Technology Transfer Institute on November 27, 2001 in Denver, Colorado. Interest in the project has also been noted by several other groups in the industry requesting further information.

Results and Discussion

Further testing is necessary to verify the results obtained so far and to improve the experimental techniques. The anticipated testing for the next phase should accomplish these goals. The disruption in the activities caused by the changes in corporate ownership should be overcome (or significantly reduced)

during the next quarter allowing the project work to get back on track.

The testing during the first quarter of 2001 used the same filter and media as used in the initial testing. This filter media has been sitting unused for approximately 90 days since the last testing was concluded. The produced water treated came in three batches each with different organic concentrations. The first batch had substantially higher organic concentrations than the initial work in November and filters efficiency was much lower than the initial work indicated which could have been a result of the filter having a loading which started to effect its efficiency. Although looking at the data, it appears the final portion of this batch the filter was more efficient than the initial portion.

The second batch was even higher in organic concentration and it appeared that the filter had reached its saturation point as the untreated and treated water samples were effectively the same in oil and grease concentration for both the second and third batch of produced water.

The BTEX appeared to be generally unchanged throughout the range of samples from untreated to treated and generally increased in concentration with each batch processed. The inorganic analyses from samples taken on February 7 indicate that the sodium, potassium, calcium, magnesium, zinc, and manganese percentages concentrations were increased probably due to the leaching of the inorganic material within the filter media. The concentrations of iron, copper, cadmium, and lead were decreased at least theoretically due to the reaction with the

filter material. The theory that the organic material that was not being captured was lighter hydrocarbons, seems to be well worn out by the air sparging test work and the reduction in Betex material concentration over very short periods of time as a result of air sparging.

Conclusions

The SynCoal filter media can be effective in removing oil and grease until it becomes saturated at which point its effectiveness is reduced. It is possible that the filters effectiveness was impacted by the long layoff and some subsequent reaction of the previously loaded oil and grease on the carbon material or further oxidation of the carbon as it sat for the prolonged period of time in the filter.

The organic material that escapes the filter media appears to be lighter weight hydrocarbons that can be effectively removed by air sparging to achieve acceptable levels of hydrocarbon concentration.

Further the effective on dissolved inorganic concentrations is inconclusive although there is apparently leaching of sodium, potassium, calcium and magnesium materials from the filter media.

Further work is necessary to repeat these tests to eliminate the impact of the extended inactive period with the filter and to further define the capacity and limitations of the filter media.

The results to date are encouraging in that the SynCoal filter media removes substantial quantities of hydrocarbons.

If it is merged with a air sparging system the levels of total hydrocarbons can be reduced to very low levels. The inorganic concentrations are not effected significantly, although some results indicate that they are altered in some manner that is not yet understood. Future testing should attempt to repeat these observations and identify the

mechanisms impacting these concentrations. If they can be identified perhaps they can be controlled.

The inconsistent results between repetitive samples has been plaguing the test program causing the need to repeat test sequences and limiting the confidence in the conclusion.



Prototype "Filter Canister"
From Steel Pipe

Filter Unit Installed at
Field Site



Portable Skid
Mounted PVC Filters

Initial Column Test Notes & Results

Wattenburg Location

SynCoal Water Treatment Test Series 1

Nov. 1, 00

Sample Log

ID	Interval	Time	R Malmquist		
			pH	Cond, ms/cm	Notes
NA-001A	0.00	11:30	6.5	2.84	Flow adjusted to 3.3gpm, inlet = 11psi
NA-001B	0.00		6.5		Outlet = <3psi
NA-001C	0.00		6.5		
NA-002A	0.25	11:45	6.5	2.08	
NA-002B	0.25		6.5		
NA-002C	0.25		6.5		
NA-003A	0.50	12:00	6.5	2.30	Flow 3.3gpm
NA-003B	0.50		6.5		
NA-003C	0.50		6.5		
NA-004A	0.75	12:18	6.5	2.27	
NA-004B	0.75		6.5		
NA-004C	0.75		6.5		
NA-005A	1.00	12:32	6.5	2.26	Inlet = 12.5psi
NA-005B	1.00		6.5		Outlet = <3psi
NA-005C	1.00		6.5		
NA-006A	1.50	13:00	6.5	2.27	
NA-006B	1.50		6.5		
NA-006C	1.50		6.5		
NA-007A	2.00	13:35	6.5	2.25	
NA-007B	2.00		6.5		
NA-007C	2.00		6.5		
NA-008A	2.50	14:00	6.5	2.30	Inlet = 15 psi
NA-008B	2.50		6.5		Outlet = <3psi
NA-008C	2.50		6.5		
NA-009A	3.00	14:30	6.5	2.30	
NA-009B	3.00		6.5		inlet = 16psi
NA-009C	3.00		6.5		Flow = 2.1gpm
NA-010A	3.50	15:00	6.5	2.39	SAMPLES:
NA-010B	3.50		6.5		
NA-010C	3.50		6.5		
NA-011A	4.00	15:30	6.5	2.32	A: Preserved w/5ml HCl for Oil & Grease, 1 liter sample
NA-011B	4.00		6.5		B: Preserved w/1ml HNO3, filtered through 0.45 micron filter for trace metal
NA-011C	4.00		6.5		analysis. 250ml sample
NA-012A	4.50	16:00	6.5	2.39	C: Unpreserved sample for TSS, cation anion balance, pH, Cond. 500ml
NA-012B	4.50		6.5		
NA-012C	4.50		6.5		
NA-013A	5.00	16:30	6.5	2.39	
NA-013B	5.00		6.5		Estimated gallons treated =
NA-013C	5.00		6.5		
NA-014A	0.00	10:00	6.5	2.42	NA-014A,B,C IS INFEED SAMPLE
NA-014B	0.00				
NA-014C	0.00				flow 3.0gpm

NA-015A	1.00	11:00	6.5	2.35	inlet 17psi		
NA-015B	1.00						
NA-015C	1.00						
NA-016A	2.00	12:00	6.5	2.30	flow 3.3gpm		
NA-016B	2.00						
NA-016C	2.00						
NA-017A	3.00	13:02	6.5	2.33	inlet 17psi		
NA-017B	3.00						
NA-017C	3.00						
NA-018A	4.00	14:00	6.5	2.34			
NA-018B	4.00						
NA-018C	4.00				inlet 17psi		
NA-019A	5.00	15:00	6.5	2.36	flow 3.3gpm		
NA-019B	5.00						
NA-019C	5.00						
NA-020A	6.00	16:00			Gallons Treated	=	1152
NA-020B	6.00				Previous day		925
NA-020C	6.00				Total for single column to date =		2077

Preliminary Laboratory Results

Lab Results
Weld Laboratories, Inc.
Lab No: 4287

	Oil/Grease	Ca	Na	Mg	K	SO4	Cl
NA-001 FEED	44.0	231	5480	38.6	38.0	<1.0	4550
NA-002	9.6	241	5380	30.0	36.0	<1.0	4550
NA-003	5.3	239	5460	28.8	42.0	<1.0	4640
NA-004	5.1	242	5610	112.6	40.8	<1.0	4290
NA-005	6.4	223	5360	38.6	41.8	<1.0	4730
NA-006	2.8	239	5570	29.0	47.8	<1.0	4640
NA-007	3.5	238	5540	43.7	49.3	<1.0	4550
NA-008	3.8	241	5330	31.1	45.3	<1.0	4550
NA-009	1.8	246	5570	41.0	47.3	<1.0	4460
NA-010	<0.1	238	5310	32.3	51.3	<1.0	4640
NA-011	1.1	243	5230	35.1	48.5	<1.0	4640
NA-012	0.7	250	5230	27.7	48.3	<1.0	4550
NA-013	<0.1	245	5000	27.9	46.0	<1.0	4550
NA-014 FEED	30.8	245	5040	25.8	48.5	<1.0	4460
NA-015	1.1	245	5040	47.3	44.8	<1.0	4640
NA-016	2.9	246	5120	89.3	51.0	<1.0	4460
NA-017	1.2	242	5100	49.6	46.3	<1.0	4550
NA-018	1.1	240	5020	36.5	48.5	<1.0	4550
NA-019	2.8	245	5100	48.7	51.0	<1.0	4640
NA-020	4.2	237	5120	59.2	48.0	<1.0	4550

	Cr	Mo	Ni	Pb	Ag	Hg	Ba	Cd
NA-001 FEED	<0.01	<0.05	<0.01	<0.01	0.183	<0.0002	<1.0	0.080
NA-002	<0.01	<0.05	<0.01	<0.01	0.000	<0.0002	<1.0	0.065
NA-003	<0.01	<0.05	<0.01	<0.01	0.013	<0.0002	<1.0	0.065
NA-004	<0.01	<0.05	<0.01	<0.01	0.034	<0.0002	<1.0	0.063
NA-005	<0.01	<0.05	<0.01	<0.01	0.020	<0.0002	<1.0	0.065
NA-006	<0.01	<0.05	<0.01	<0.01	0.028	<0.0002	<1.0	0.065
NA-007	<0.01	<0.05	<0.01	<0.01	0.030	<0.0002	<1.0	0.065
NA-008	<0.01	<0.05	<0.01	<0.01	0.033	<0.0002	<1.0	0.070
NA-009	<0.01	<0.05	<0.01	<0.01	0.035	<0.0002	<1.0	0.065
NA-010	<0.01	<0.05	<0.01	<0.01	0.043	<0.0002	<1.0	0.068
NA-011	<0.01	<0.05	<0.01	<0.01	0.045	<0.0002	<1.0	0.068
NA-012	<0.01	<0.05	<0.01	<0.01	0.050	<0.0002	<1.0	0.068
NA-013	<0.01	<0.05	<0.01	<0.01	0.050	<0.0002	<1.0	0.068
NA-014 FEED	<0.01	<0.05	<0.01	<0.01	0.053	<0.0002	<1.0	0.068
NA-015	<0.01	<0.05	<0.01	<0.01	0.058	<0.0002	<1.0	0.075
NA-016	<0.01	<0.05	<0.01	<0.01	0.058	<0.0002	<1.0	0.068
NA-017	<0.01	<0.05	<0.01	<0.01	0.050	<0.0002	<1.0	0.070
NA-018	<0.01	<0.05	<0.01	<0.01	0.060	<0.0002	<1.0	0.068
NA-019	<0.01	<0.05	<0.01	<0.01	0.065	<0.0002	<1.0	0.068
NA-020	<0.01	<0.05	<0.01	<0.01	0.065	<0.0002	<1.0	0.068

	CO3	TSS	Zn	Fe	Mn	Cu	Al
NA-001 FEED	119	59	0.40	69.1	0.92	0.265	<0.1
NA-002	108	19	0.55	77.9	0.93	0.050	<0.1
NA-003	118	22	0.55	76.4	0.89	0.028	<0.1
NA-004	119	15	0.44	75.3	0.88	0.033	<0.1
NA-005	114	21	0.93	72.6	0.87	0.030	<0.1
NA-006	114	14	0.94	73.9	0.87	0.035	<0.1
NA-007	114	18	1.28	73.9	1.00	0.033	<0.1
NA-008	119	13	1.10	72.5	0.96	0.043	<0.1
NA-009	108	15	1.07	72.5	0.97	0.033	<0.1
NA-010	114	25	0.90	71.4	1.00	0.033	<0.1
NA-011	102	17	0.63	68.8	0.96	0.028	<0.1
NA-012	102	19	0.90	71.0	0.90	0.023	<0.1
NA-013	102	23	1.00	69.9	0.90	0.023	<0.1
NA-014 FEED	108	20	1.16	69.9	0.95	0.043	<0.1
NA-015	119	24	1.11	69.4	0.88	0.045	<0.1
NA-016	114	19	1.28	69.8	0.89	0.035	<0.1
NA-017	102	17	1.28	72.6	0.91	0.038	<0.1
NA-018	91	25	1.10	72.6	0.89	0.043	<0.1
NA-019	97	30	1.18	72.8	0.89	0.033	<0.1
NA-020	97	18	1.19	72.6	0.90	0.035	<0.1

Second Test Series Results Comparison

Sample ID	Untreated Oil & Grease ppm	Treated Oil & Grease ppm	Untreated BTEX ppb	Treated BTEX ppb
01-Feb	182	117	261	99
02-Feb	106	79.2	1165	1476
05-Feb	100	42	1255	1177
06-Feb	118	15.7	1539	881
07-Feb	258	403	1096	671
07-Feb	179	140	1214	883
08-Feb	255	239	3161	1907
09-Feb	276	251	1129	2522
12-Feb	175	141	3757	2441
13-Feb	12.6	17.1	1796	1976
13-Feb	4.1	8.3	2959	2773
14-Feb	4.2	3.8	2218	4293
19-Feb	4.3	8.2	7587	7853
20-Feb	2.8	2	7749	7203

Second Test Series Summary Results

Sample ID	Untreated Oil & Grease ppm	Treated Oil & Grease ppm	Removal Efficiency	Untreated BTEX ppm	Treated BTEX ppm	Removal Efficiency
Batch 1 Ave	126.5	63.5	50%	1055	908	14%
Batch 2 Ave	228.6	234.8	-3%	2071	1685	19%
Batch 3 Ave	5.6	7.9	-41%	4462	4820	-8%

Analyses for Samples Gathered July 18, 2001

Laboratory No. 4622

30-Aug-01

Sample ID	Feed Comp	1st Half Feed	2nd Half Feed	C1 Comp	C1 H 0/2	C1 H 4/6	C2 Comp	C2 H 0/2	C2 H 4/6
Oil & Grease	182	7.83	7.8	81.1	17.2	5.91	9.2	1.28	9.66
Calcium	1610	1690	1650	1830	1750	1790	1820	1840	1660
Magnesium	298	310	285	315	293	308	300	320	280
Sodium	13200	16400	16200	18300	17900	18800	19200	19700	18500
Potassium	5525	6050	5620	5870	5670	5770	5800	5900	5250
Zinc	3.65	4.08	0.38	0.43	0.53	0.68	0.4	0.33	2.18
Iron	400	528	383	385	373	368	358	398	340
Manganese	6	7.13	5.9	6.78	6	6.23	6.23	6.53	5.25
Copper	0.5	0.25	0.2	0.2	0.19	0.35	0.17	0.15	0.13
Aluminum	53.5	31.5	48.5	18.3	53.8	22.8	<1	7	41
Barium	49.8	21.8	14.8	29.3	23	9.8	12	19	8.8
Cadmium	0.15	0.1	0.07	0.08	0.09	0.07	0.11	0.04	0.09
Chromium	0.65	0.28	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01

Molybdenum	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Nickel	1.38	1.43	1.25	1.3	1.23	1.1	1.08	1.13	1.13
Lead	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Silver	0.063	0.065	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Mercury	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002
Chloride	13500	13200	12300	13200	12900	13800	12900	13200	14700