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# Novel Functional Graded Thermal Barrier Coatings in Coal-fired Power Plant Turbines

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# Acknowledgement

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- Subcontract: James Knapp (Praxair Surface Technologies)
- Collaborators: Li Li, Don Lemen (Praxair Surface Technologies)
- Yeon-Gil Jung (Changwon National University)
- Yang Ren, Jiangang Sun (Argonne National Laboratory)
- Changdong Wei (OSU), Bin Hu (Dartmouth)
- Ph.D. graduate students: Xingye Guo, Yi Zhang

# Outline

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- Introduction
  - Coating fabrications
  - Single ceramic layer (SCL) architecture
  - Double ceramic layer (DCL) architecture
  - Characterization of physical and mechanical properties
  - Microstructure and composition
  - Porosity and hardness
  - Bond strength test
  - Erosion test
  - Characterization of thermal properties
  - Thermal conductivity and specific heat measurements
  - Jet engine thermal shock tests
  - Thermal gradient mechanical fatigue tests
  - Summary and future work
-

# Limitation of yttria stabilized zirconia

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- Zirconia partially stabilized with 7 wt% yttria (7YSZ) is the current state-of-the-art thermal barrier coating material.
- However, at temperatures higher than 1200 °C, YSZ layers are prone to **sintering**, which increases thermal conductivity and makes them less effective.
- The sintered and densified coatings can also **reduce thermal stress and strain tolerance**, which can reduce the coating's durability significantly.

# Motivation and objective

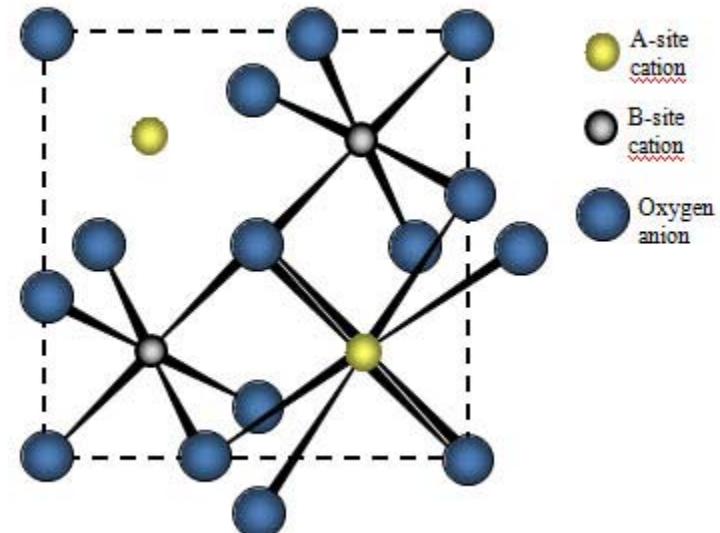
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- To further increase the operating temperature of turbine engines, alternative TBC materials with lower thermal conductivity, higher operating temperatures and better sintering resistance are required.
- The objective of the project is to develop a novel lanthanum zirconate based multi-layer thermal barrier coating system.
- The ultimate goal is to develop a manufacturing process to produce pyrochlore oxide based coating with improved high-temperature properties.

# Pyrochlore - $A_2B_2O_7$

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Pyrochlore-type rare earth zirconium oxides ( $(Re_2Zr_2O_7$ , Re = rare earth) are promising candidates for thermal barrier coatings, high-permittivity dielectrics, potential solid electrolytes in high-temperature fuel cells, and immobilization hosts of actinides in nuclear waste.

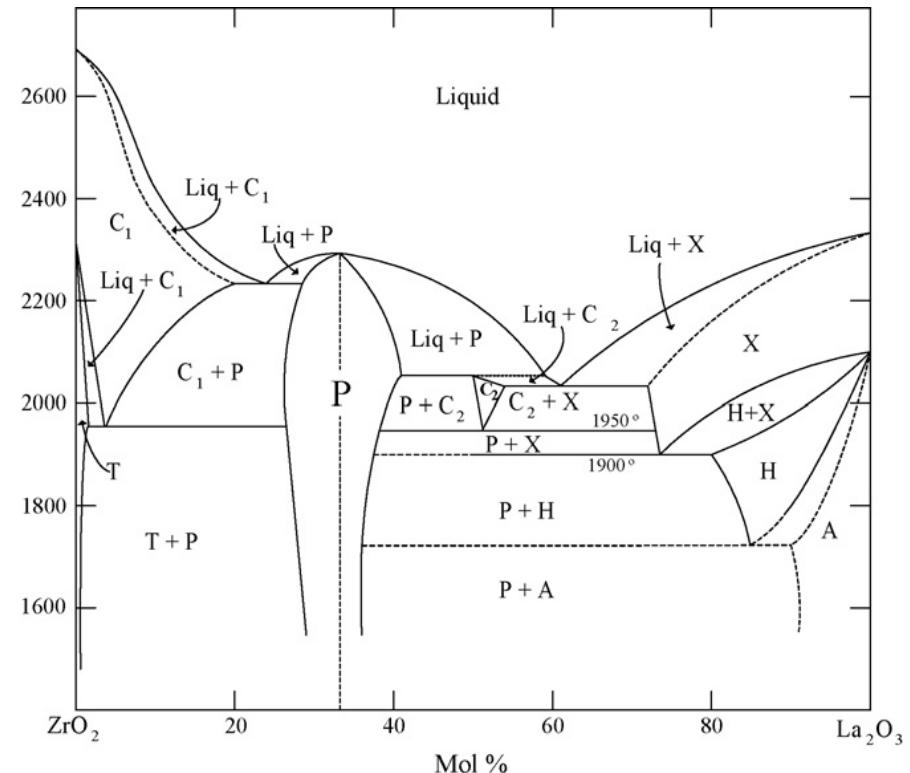


Pyrochlore crystal structure:  
 $A_2B_2O_7$ . A and B are metals incorporated into the structure in various combinations. (credit: NETL)

# Why $\text{La}_2\text{Zr}_2\text{O}_7$ ?

Compared with YSZ,  $\text{La}_2\text{Zr}_2\text{O}_7$  has

- Higher temperature phase stability. No phase transformation
- Lower sintering rate at elevated temperature
- Lower thermal conductivity
- Lower CTE



Phase diagram of  $\text{La}_2\text{O}_3-\text{ZrO}_2$

# **La<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> vs. YSZ**

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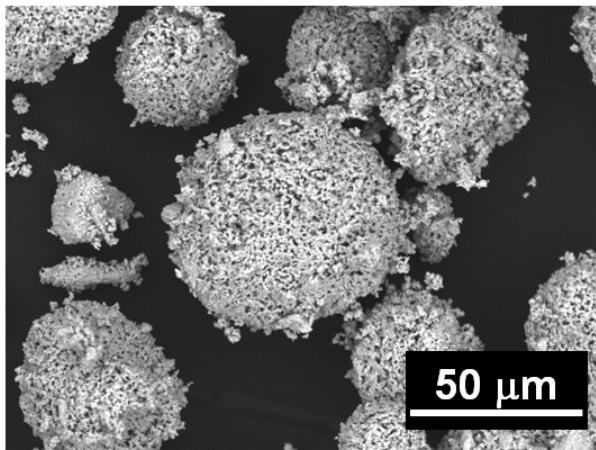
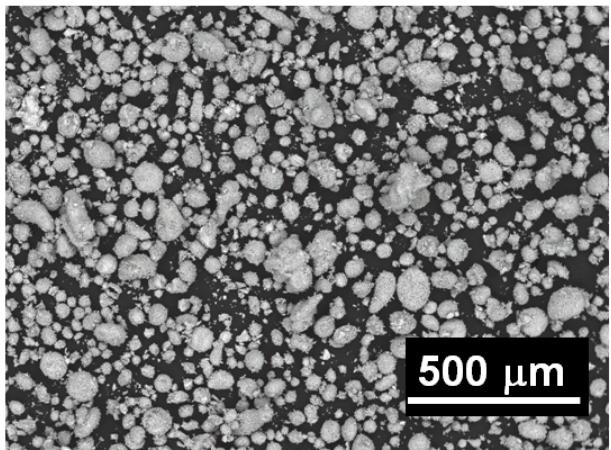
<b>Materials property</b>	<b>8YSZ</b>	<b>La<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub></b>
Melting Point (°C)	2680	2300
Maximum Operating Temperature (°C)	1200	>1300
Thermal Conductivity (W/m-K) (@ 800°C )	2.12	1.6
Coefficient of Thermal Expansion (x10 <sup>-6</sup> /K) (@1000 °C)	11.0	8.9-9.1
Density (g/cm <sup>3</sup> )	6.07	6.00
Specific heat (J/g-K) (@1000 °C)	0.64	0.54

# Layered coating architecture

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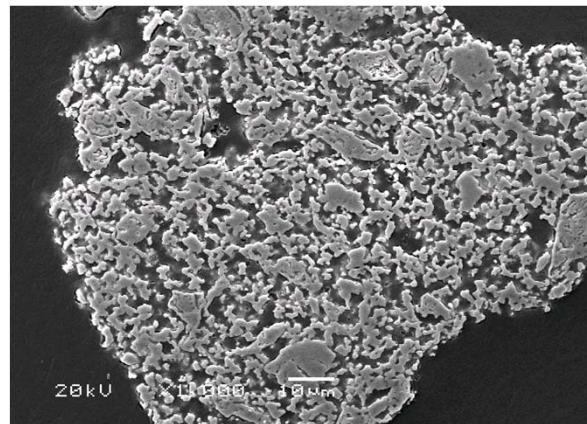
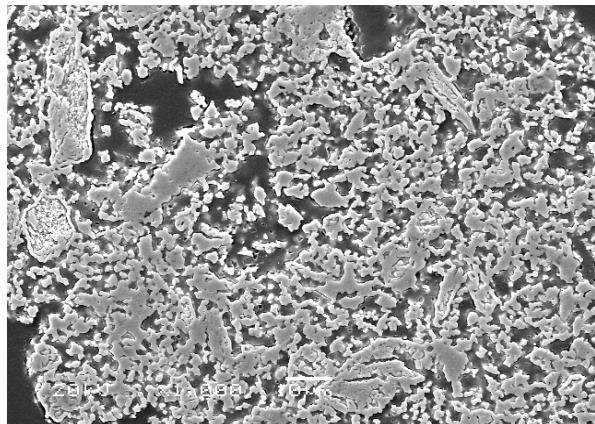
- The coefficient of thermal expansion of  $\text{La}_2\text{Zr}_2\text{O}_7$  ( $10 \times 10^{-6} /^\circ\text{C}$ ) is lower than those of both substrate and bondcoat (about  $15 \times 10^{-6} /^\circ\text{C}$  @  $1000^\circ\text{C}$ ). As a result, the thermal cycling properties may be a concern
- The layered topcoat architecture is believed to be a feasible solution to improve thermal strain tolerance
- In this work, we develop a multi-layer, functionally graded, pyrochlore oxide based TBC system

# $\text{La}_2\text{Zr}_2\text{O}_7$ spray powder morphology



## Powder surface morphology

- Spherical shape with a rough surface
- Good flowability and high density
- Particle size between 30 ~ 100  $\mu\text{m}$

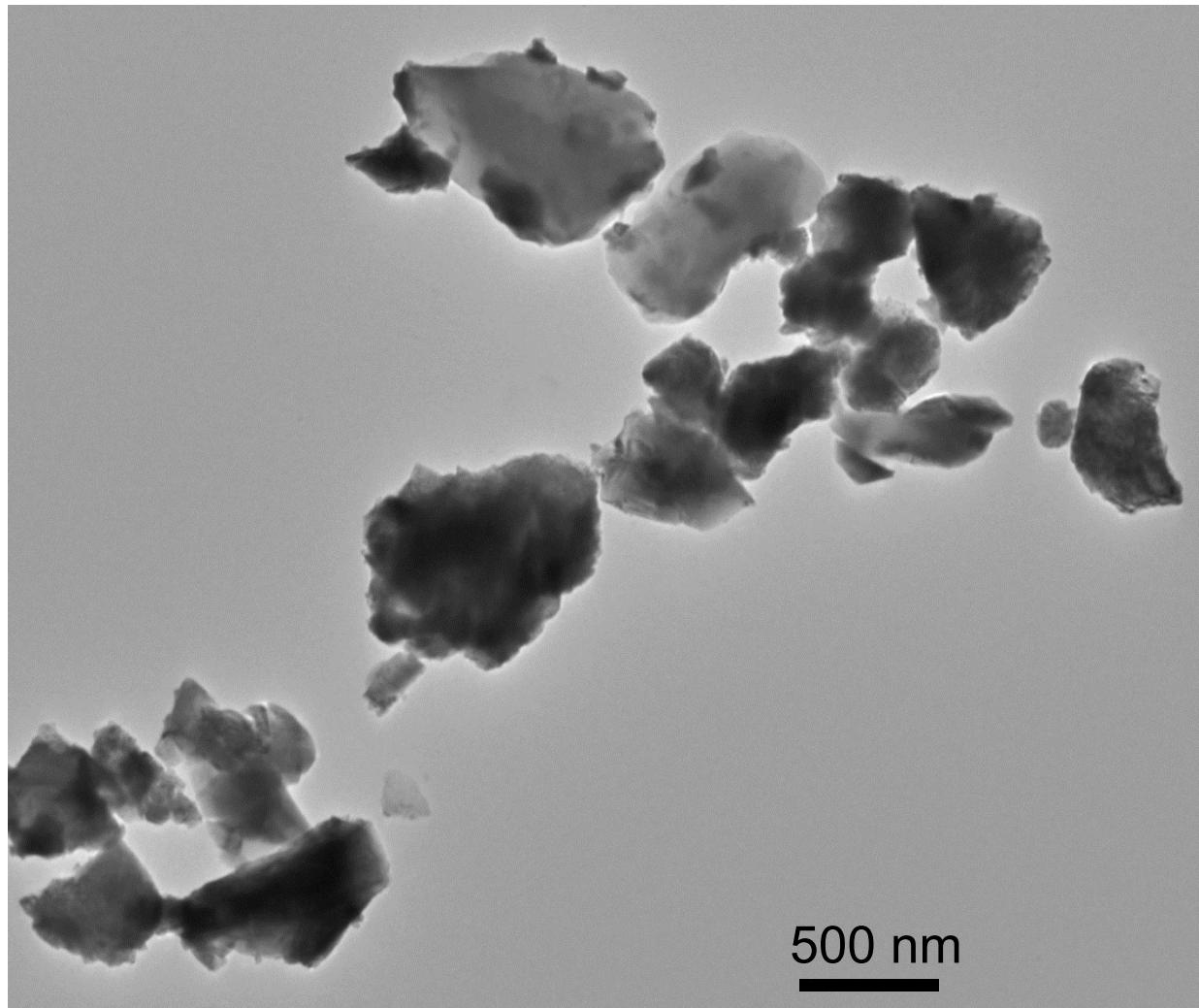


## Powder cross-section

- Porous interior

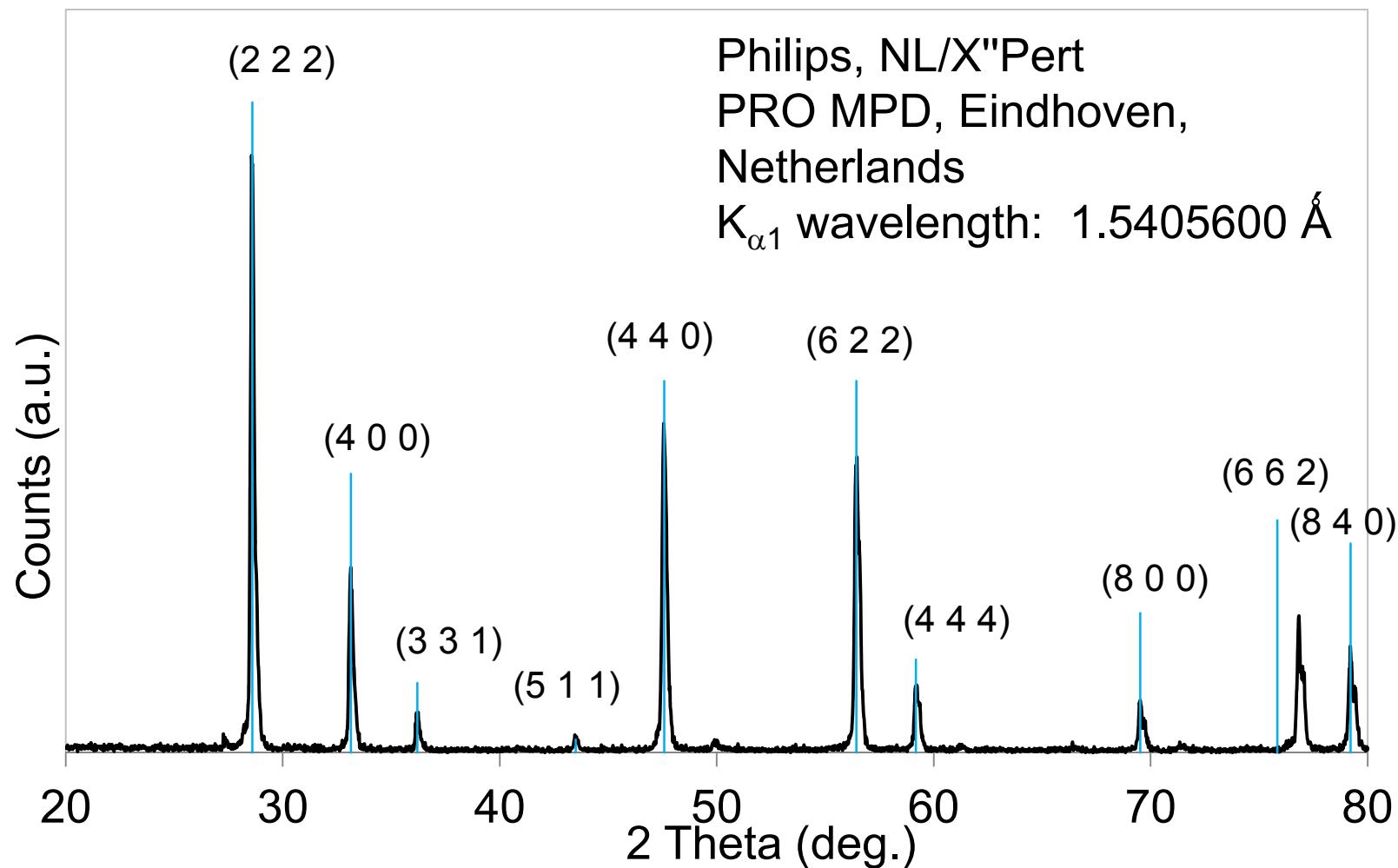
# TEM image of $\text{La}_2\text{Zr}_2\text{O}_7$

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credit: Bin Hu @ Dartmouth

# $\text{La}_2\text{Zr}_2\text{O}_7$ powder XRD analysis

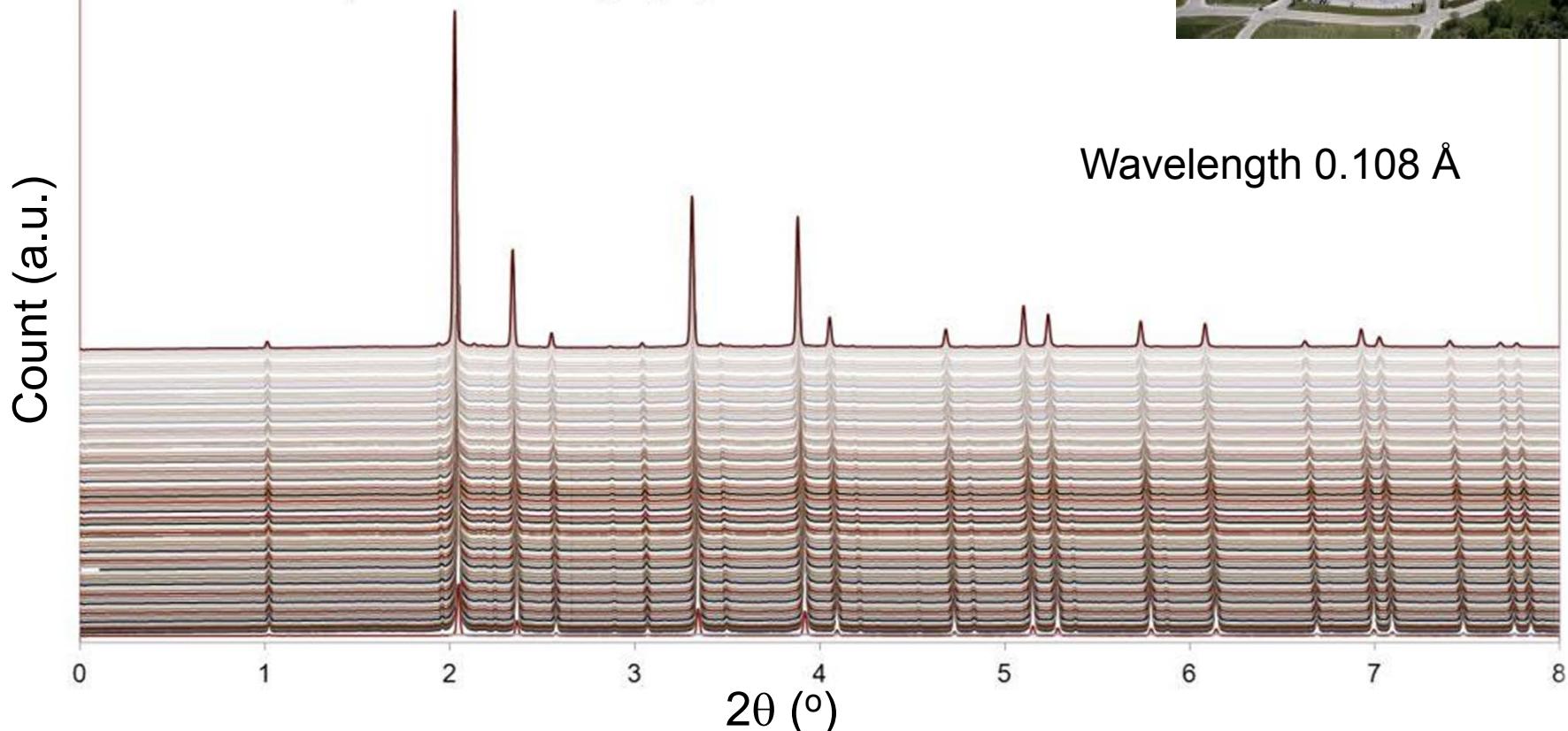


XRD data show that the powder composition is  $\text{La}_2\text{Zr}_2\text{O}_7$

# Synchrotron XRD



In situ HEXRD profiles of  $\text{La}_2\text{Zr}_2\text{O}_7$  from 30°C to 1400°C



*In situ* Synchrotron XRD shows no compositional change at high temperatures

credit: Yang Ren @ ANL

# Coating fabrication using APS

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- $\text{La}_2\text{Zr}_2\text{O}_7$  coatings were deposited using air plasma spray (APS) technique by a Praxair patented plasma spray torch.
- Haynes 188 superalloy was used as the substrate.

Haynes 188	Co	Ni	Cr	W	Si	C	La	Fe	Mn
(w%)	39	22	22	14	0.35	0.10	0.03	3	1.25

- The bond coat is Ni-based intermetallic LN-65 using APS, with a thickness of 228  $\mu\text{m}$

LN-65	Ni	Cr	Al	Y	O
(w%)	67.3	21.12	9.94	1.02	0.19

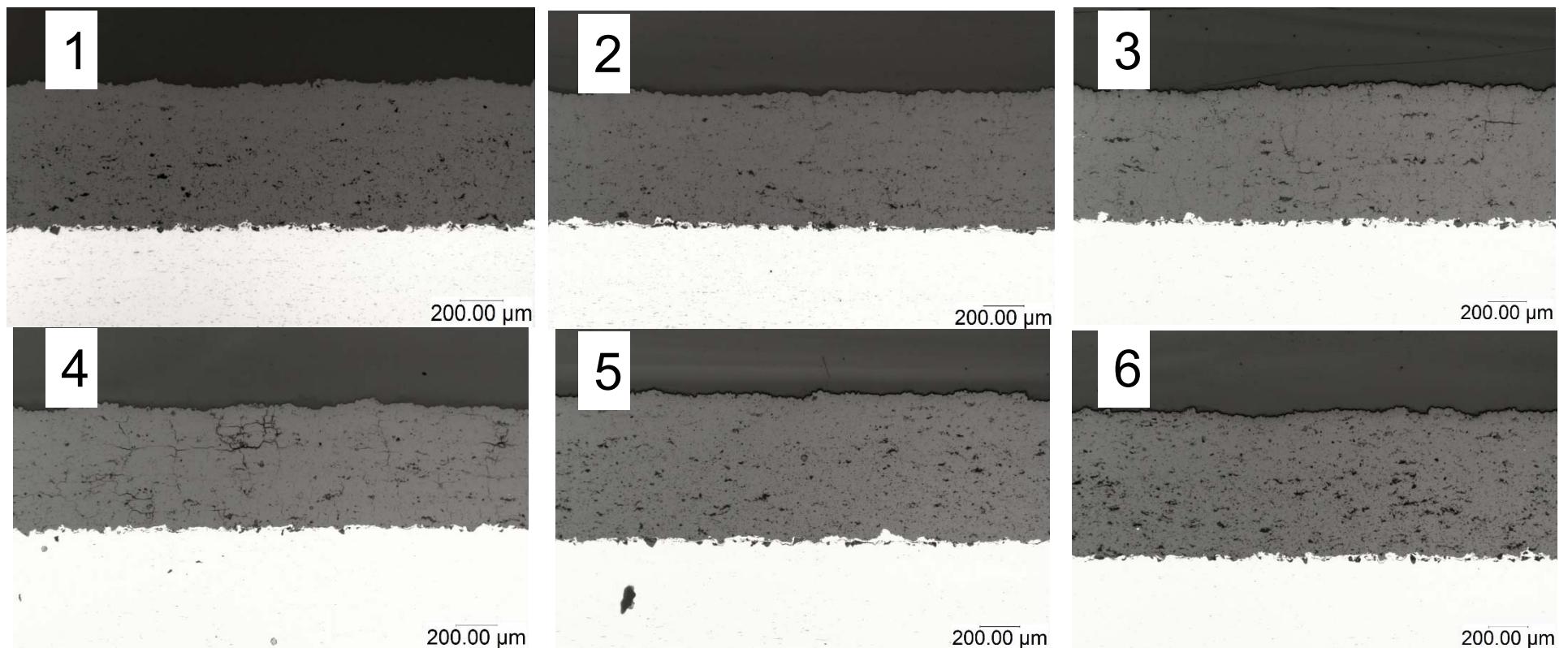
- Controlled spray parameters:
- Powder feed ratio
- Torch current
- Torch gas (Argon), Carrier gas (Argon), Shield gas (Argon), Secondary gas (Hydrogen)
- Standoff distance
- Sample rig surface rotation speed (RPM and surface speed)

# Outline

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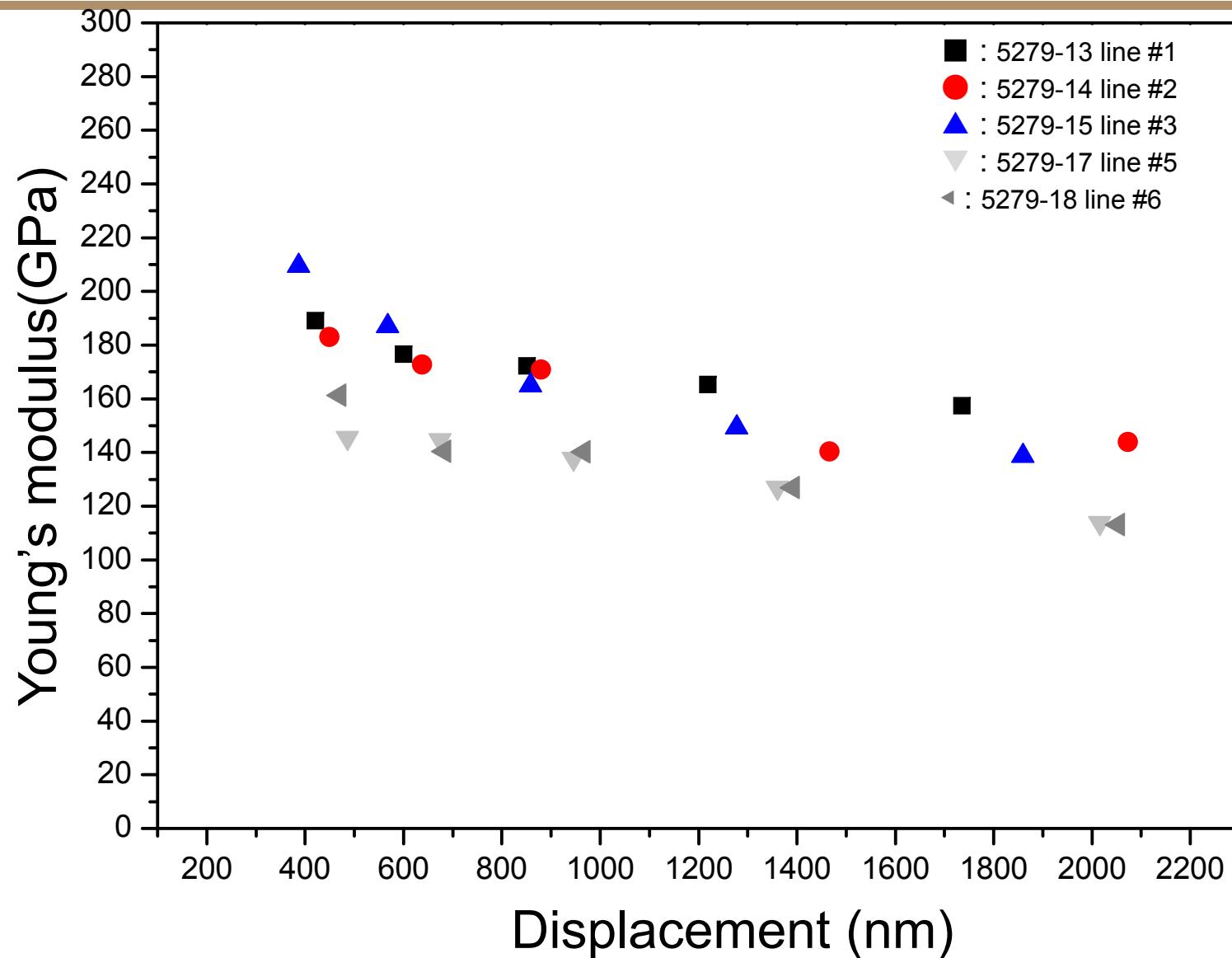
- Introduction
  - Coating fabrications
  - Single ceramic layer (SCL) architecture – dense coating
  - Double ceramic layer (DCL) architecture
  - Characterization of physical and mechanical properties
  - Microstructure
  - Hardness and Young's modulus
  - Bond strength test
  - Erosion test
  - Characterization of thermal properties
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# Cross sectional view of dense coating

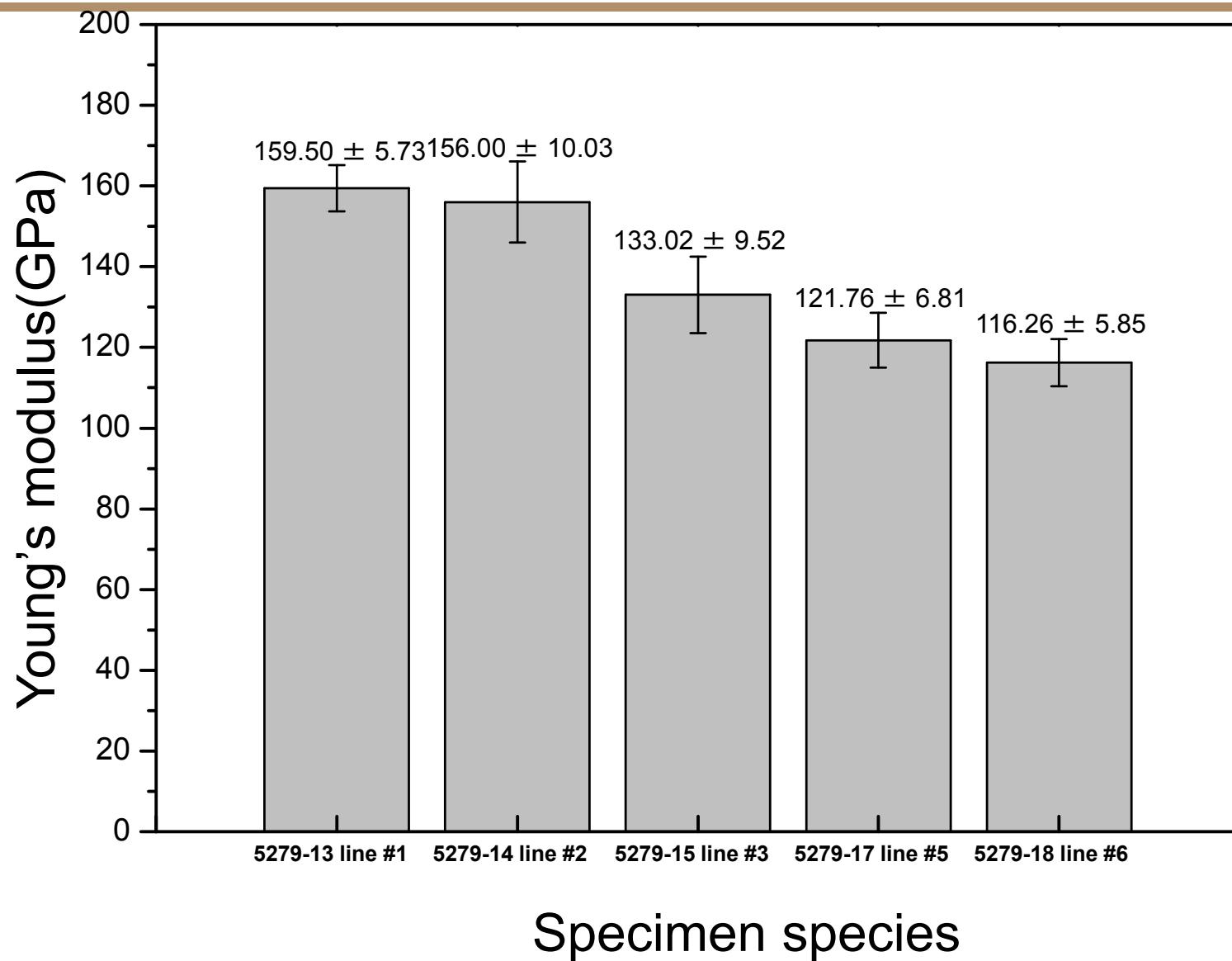


Processing parameters (powder feed rate, surface speed, current, stand off ) were varied to control the porosity.

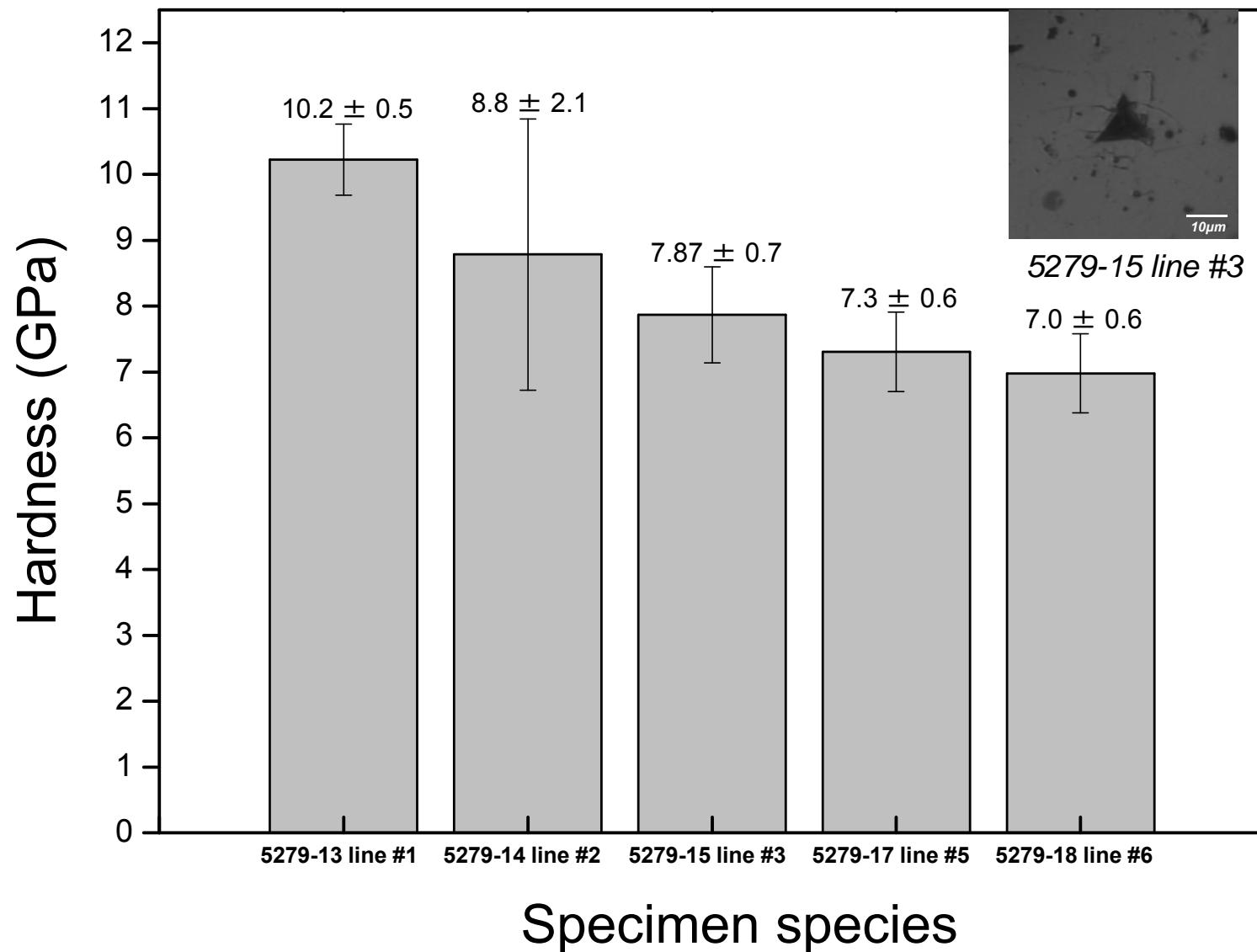
# Nanoindentation Young's modulus vs. displacement



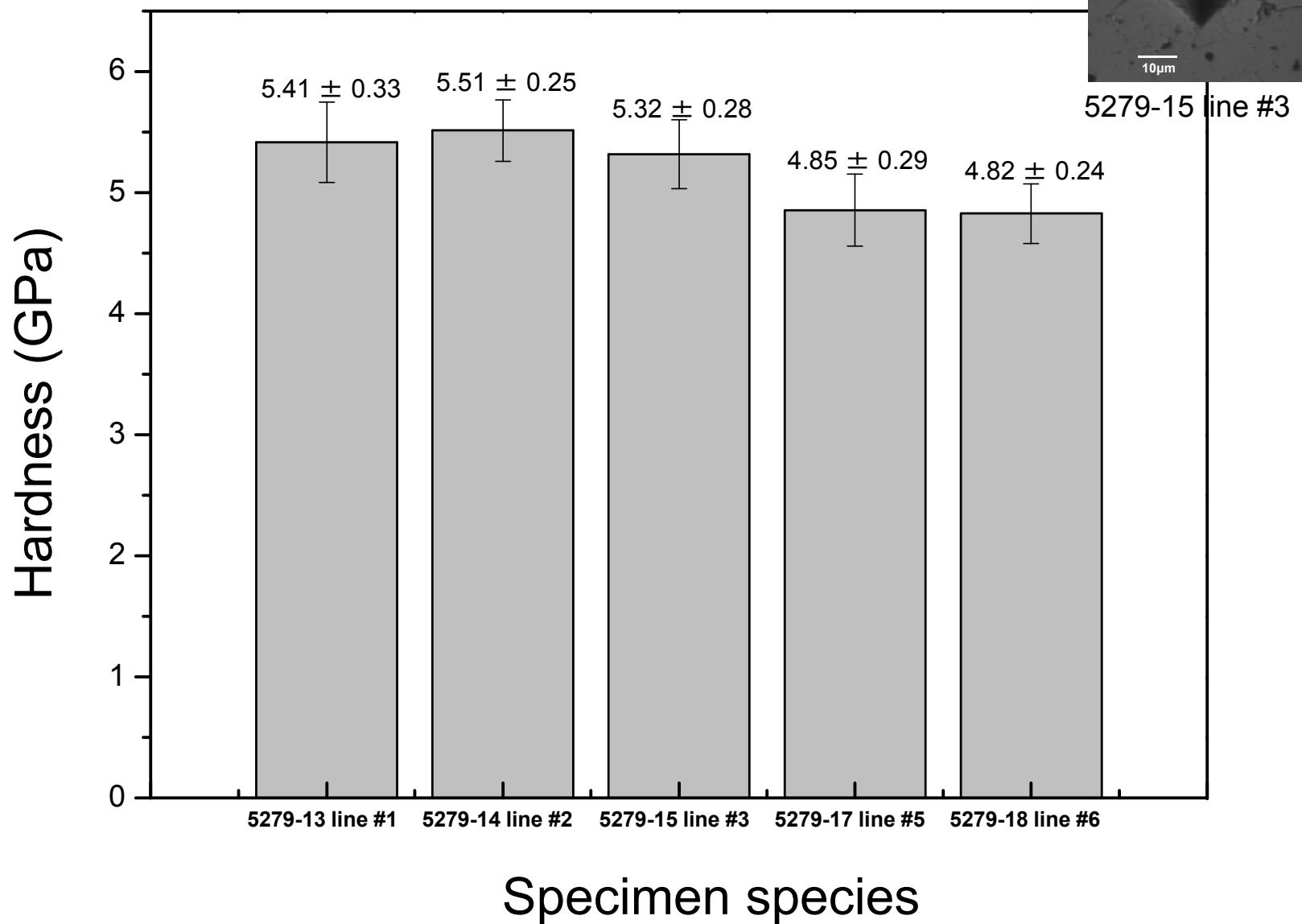
# Nanoindentation Young's modulus



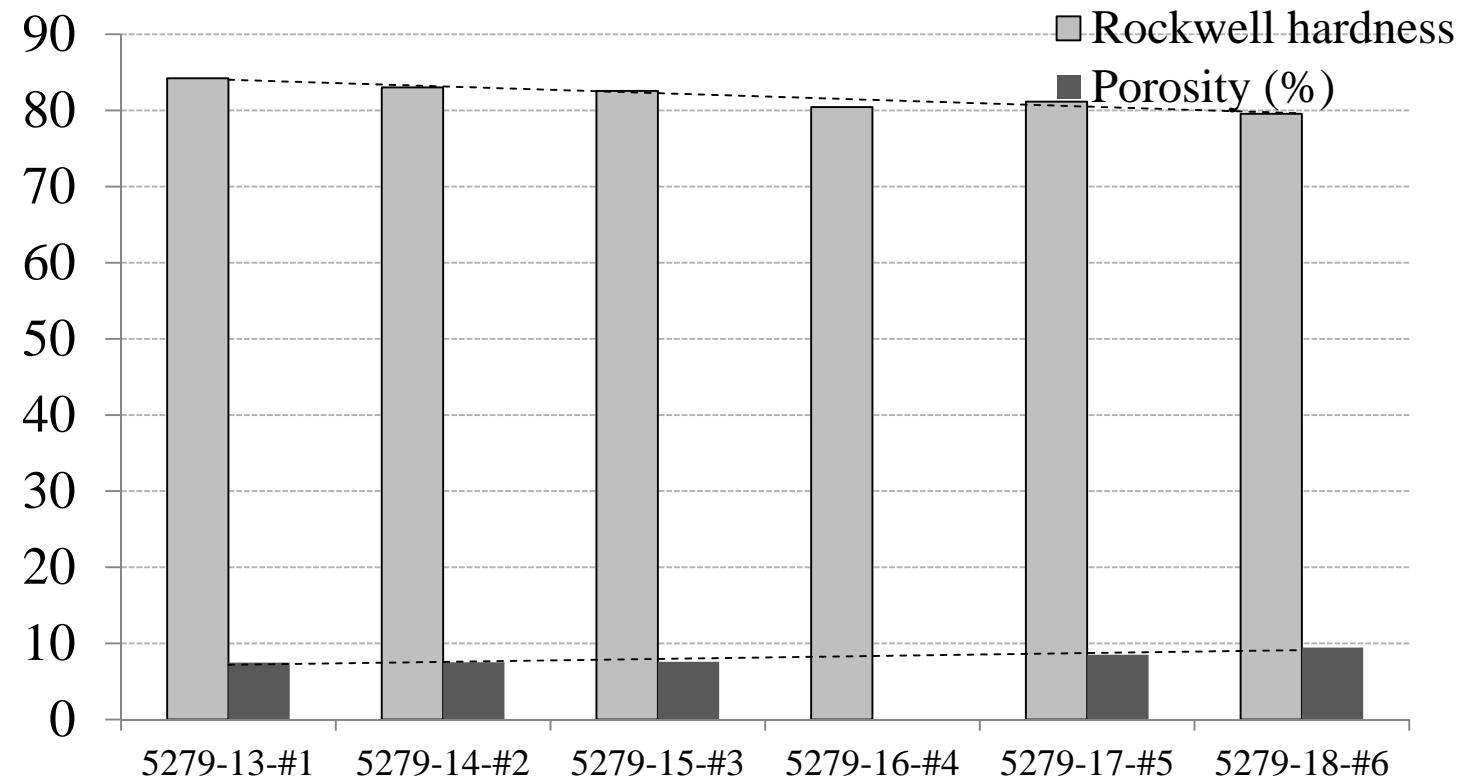
# Nanoindentation hardness



# Vicker's indentation hardness



# Rockwell's indentation hardness



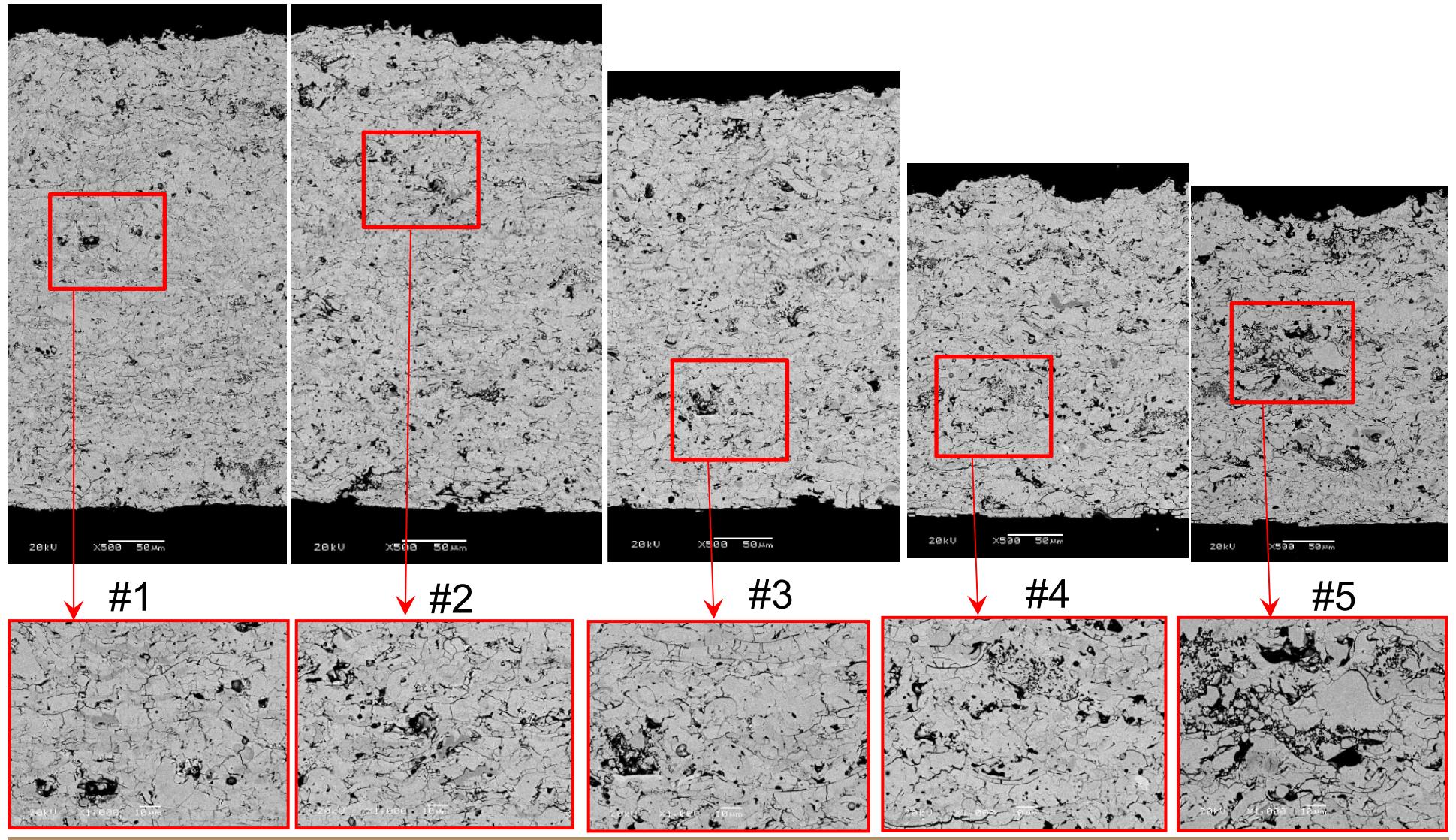
- Low density coatings with porosity between 7~10 % were achieved.
- Porosity and hardness can be tuned via changing processing conditions
- Powder feed rate↑ or current↓ → porosity↑ → hardness↓  
[Hardness =  $1.99 \times (100 - \text{porosity}) - 100$ ]

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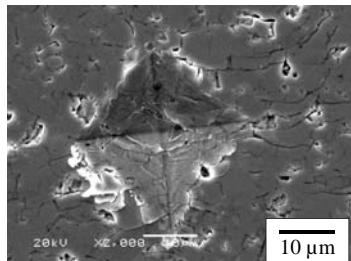
# Cross sections of SCL $\text{La}_2\text{Zr}_2\text{O}_7$ coatings



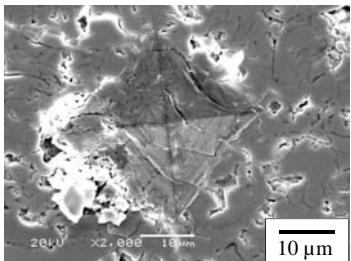
# Vickers hardness indentation

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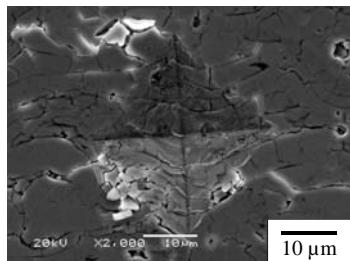
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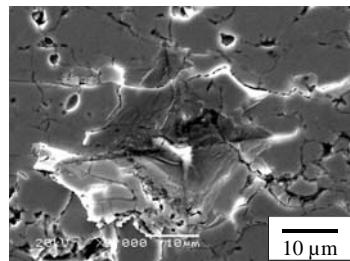
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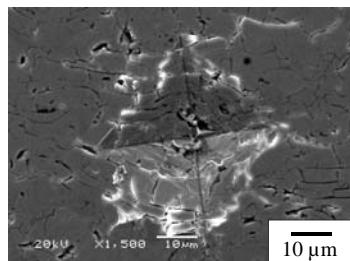
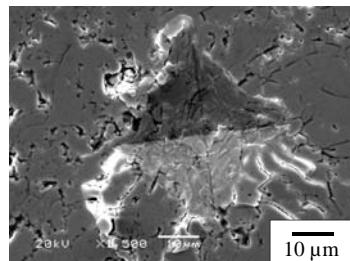
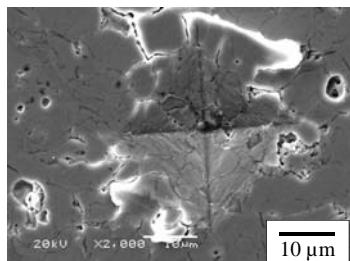
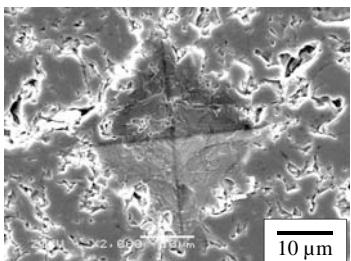
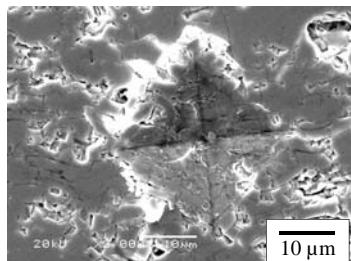
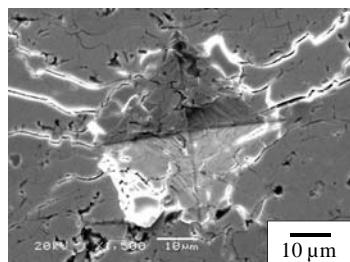
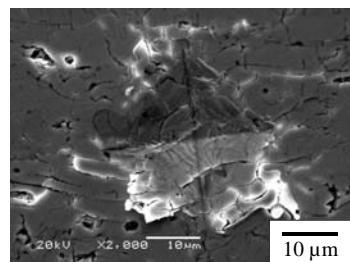
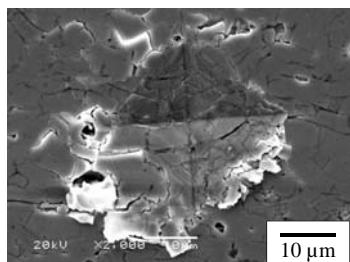
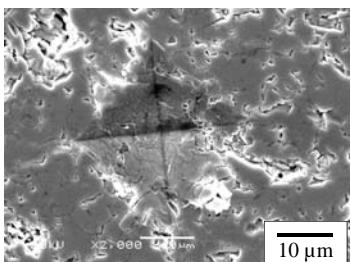
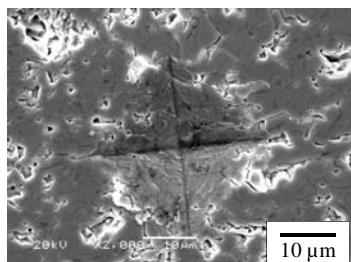
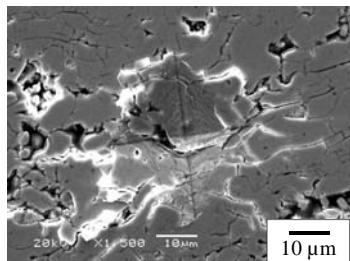
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#4

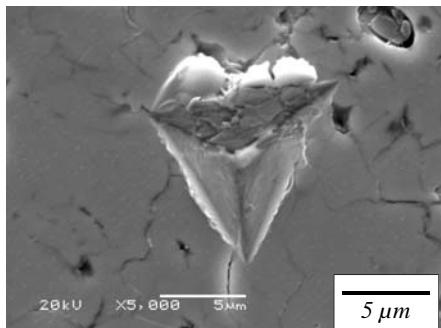


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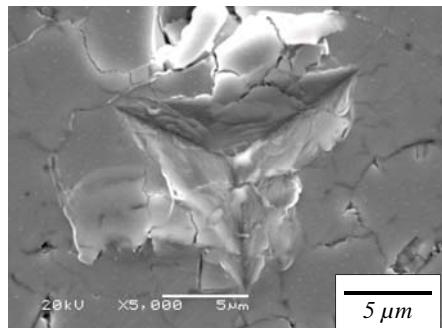


# Nanoindentation

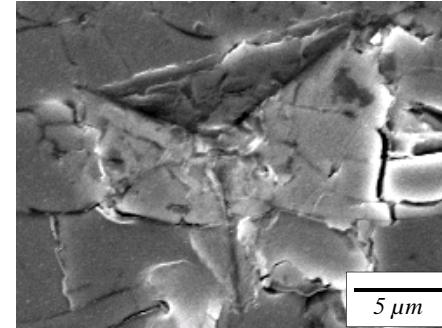
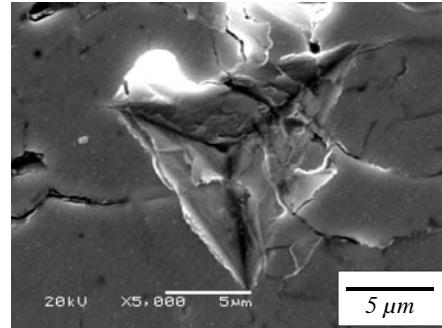
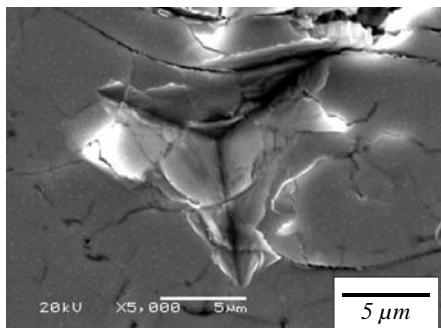
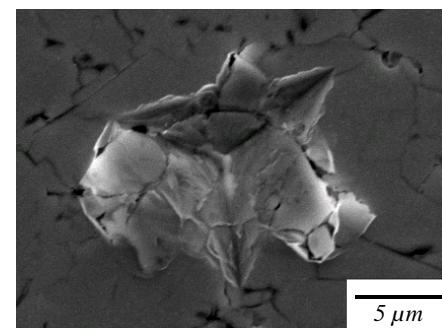
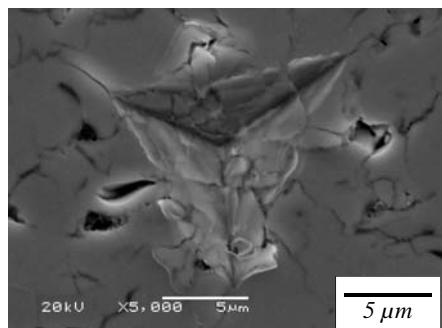
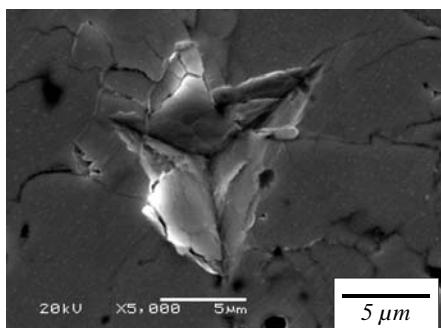
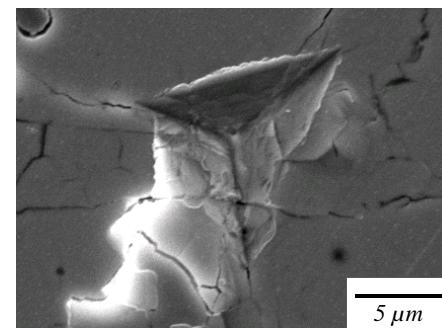
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#4

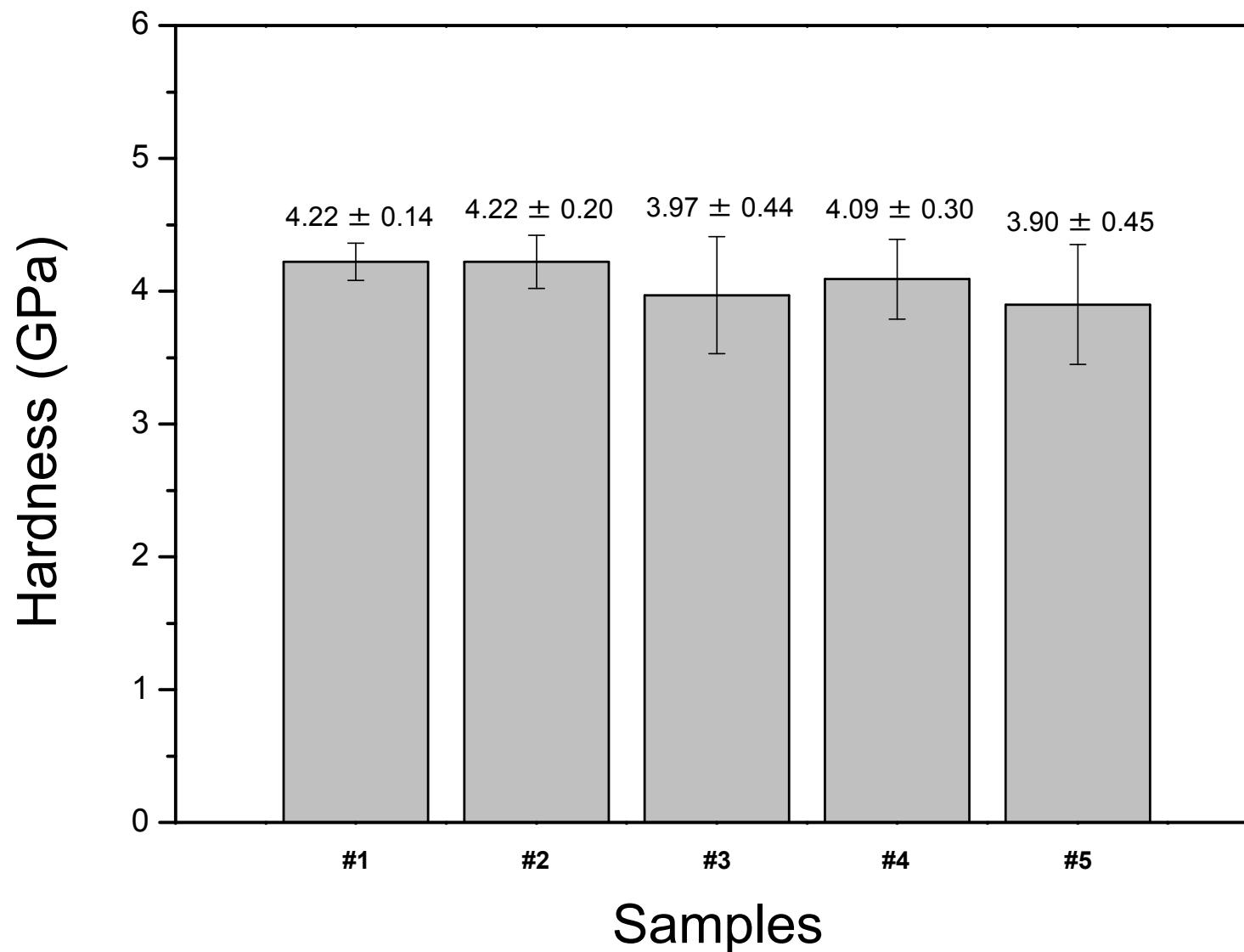


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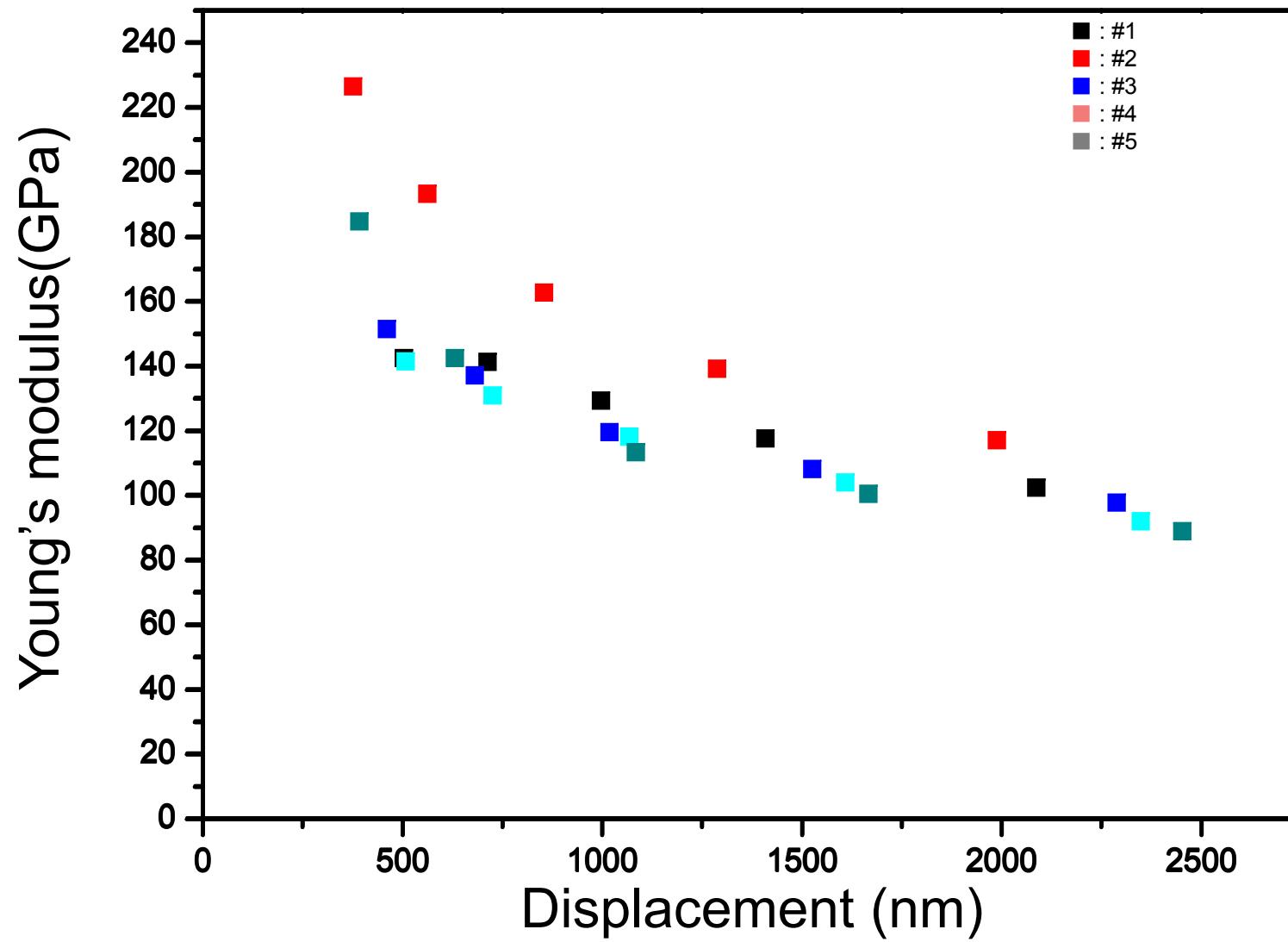
# Vickers indentation hardness

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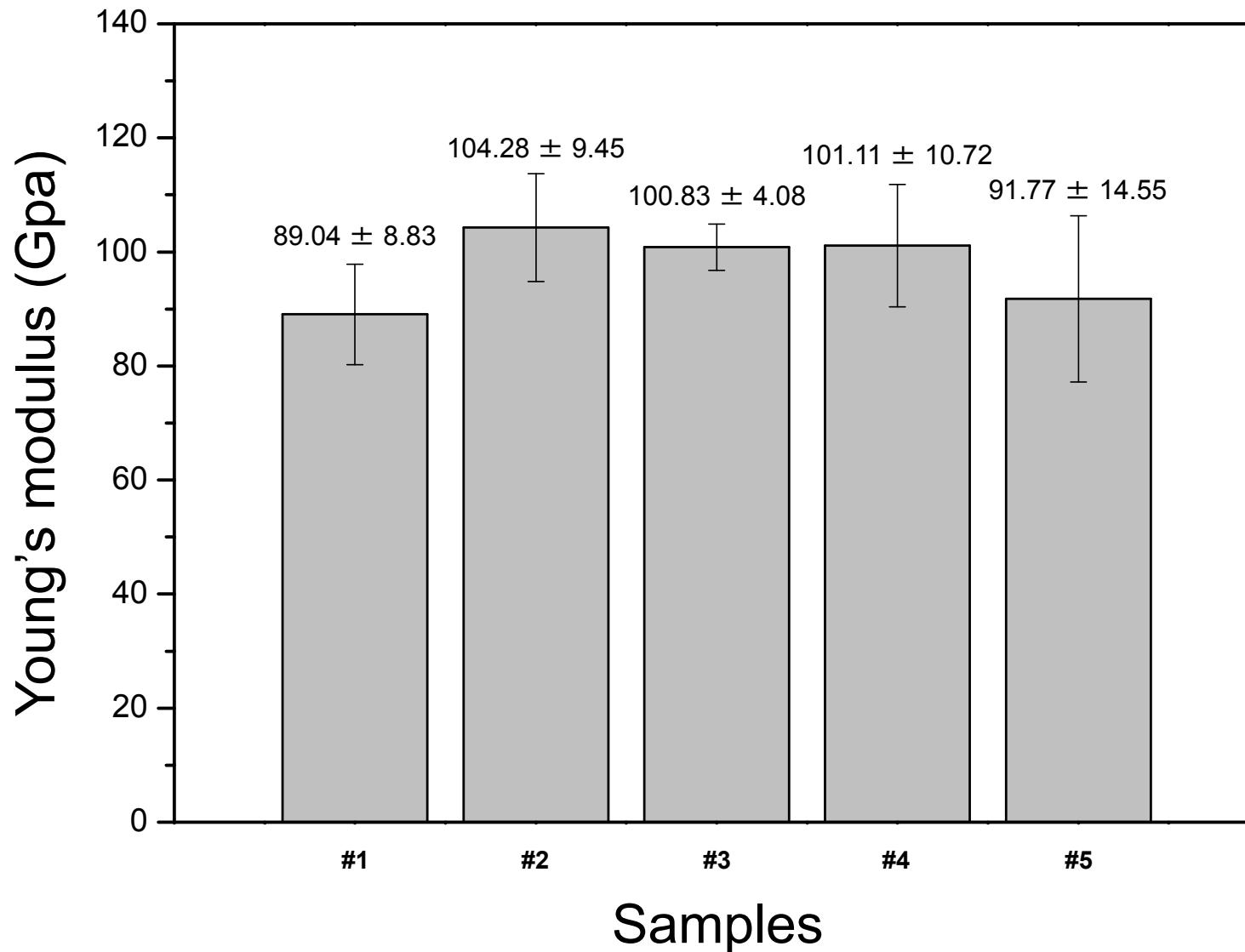
# Nano indentation Young's modulus vs. displacement

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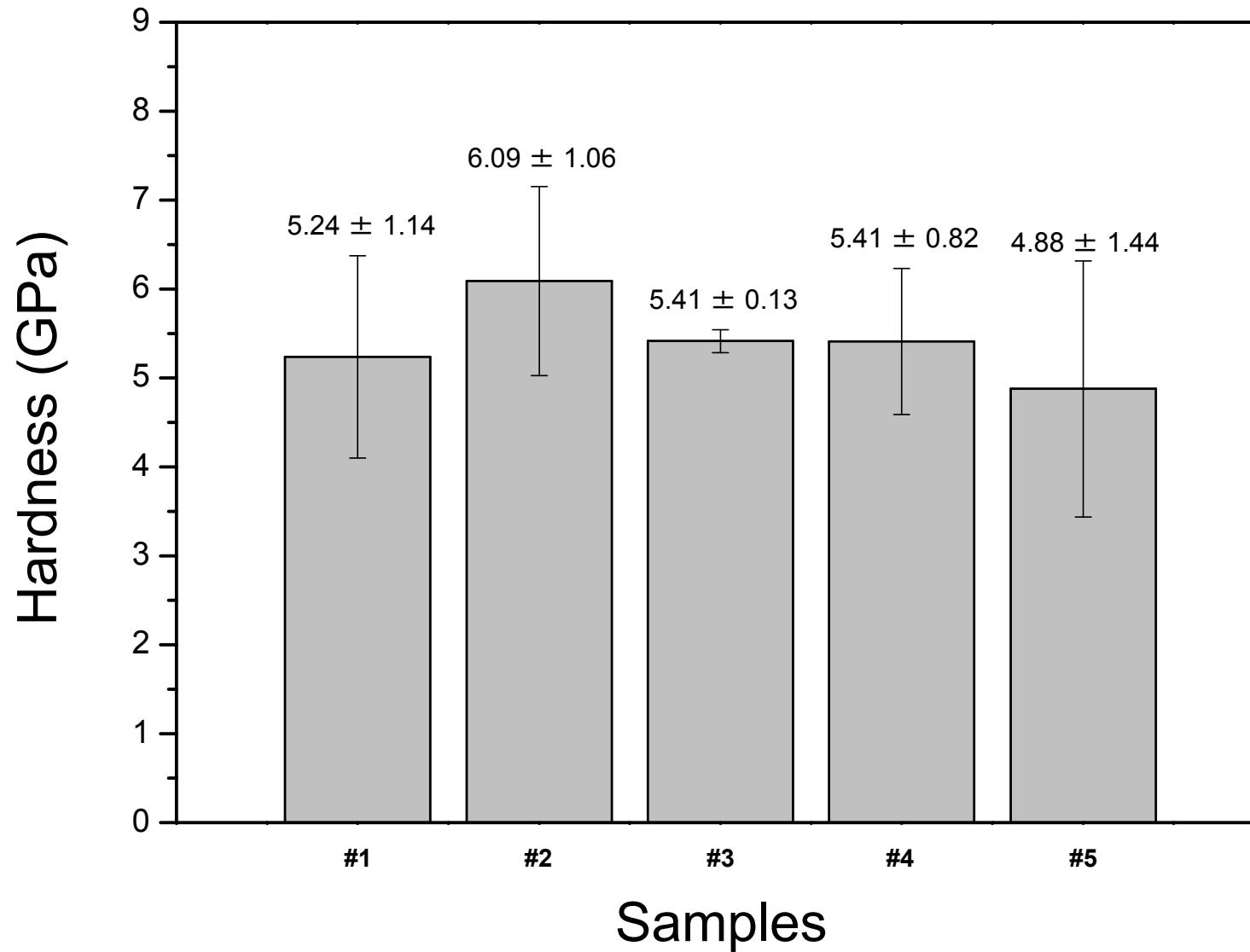
# Nanoindentation Young's modulus

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# Nanoindentation hardness

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# Porosity of low density SCL coating

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Line #	Density (g/cm <sup>3</sup> )	Porosity (%)
7	5.3182	11.36
8	5.2587	12.36
9	5.2584	12.36
10	5.2917	11.81
11	5.2614	12.31
12	5.0089	16.52

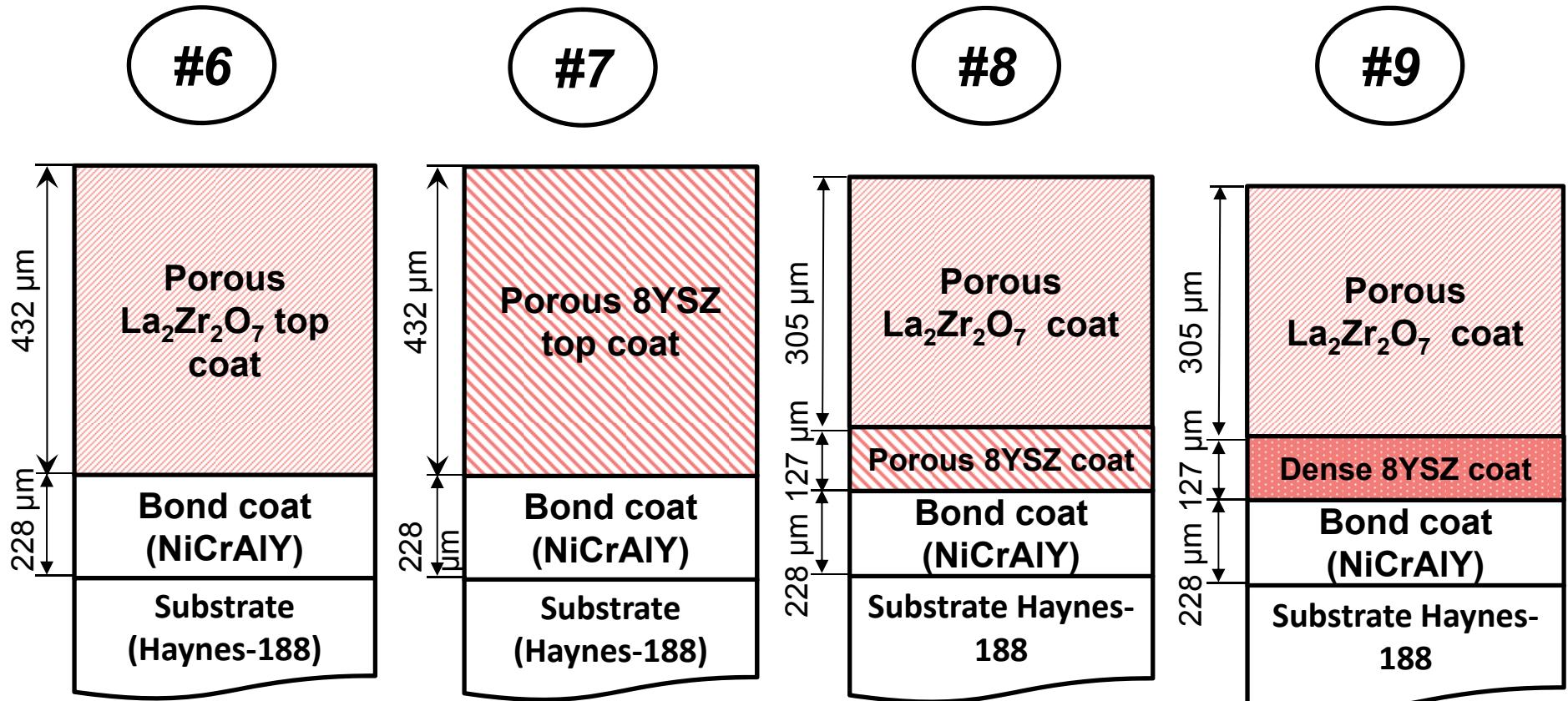
Low density coatings with porosity between 11~17% were achieved.

# Outline

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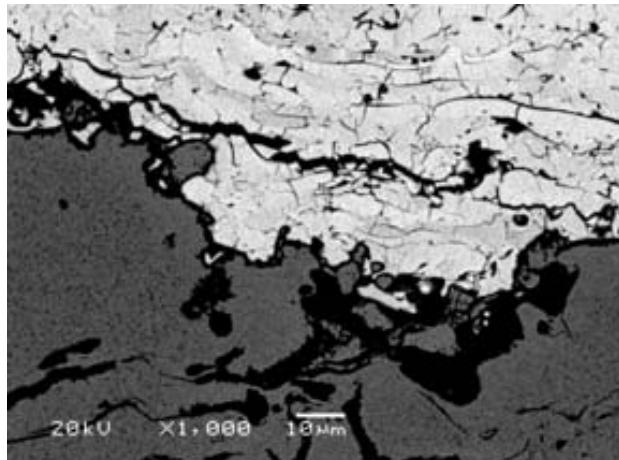
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# Double ceramic layer (DCL) architectures

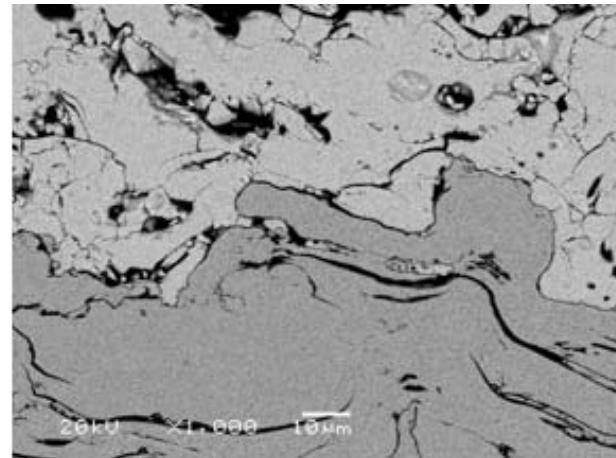


# Interfaces of DCL coatings

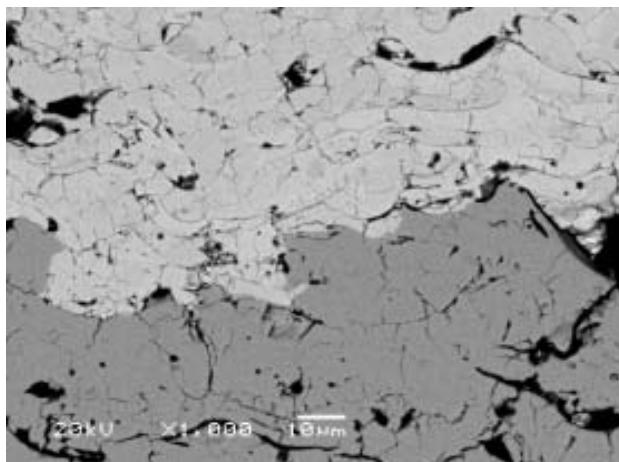
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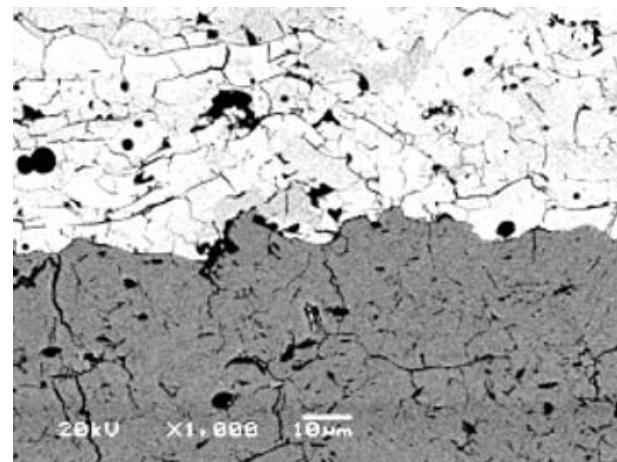
#6 La<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> and bond coat interface



#7 porous 8YSZ and bond coat interface



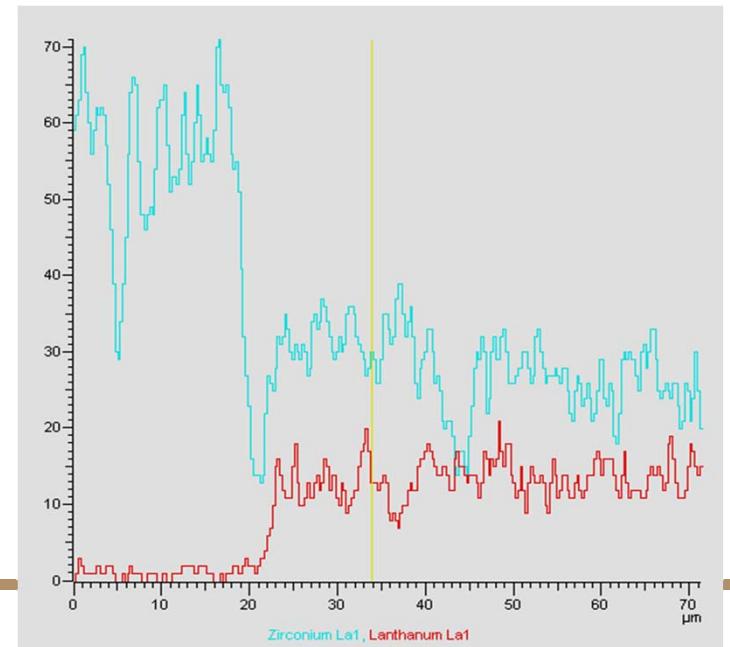
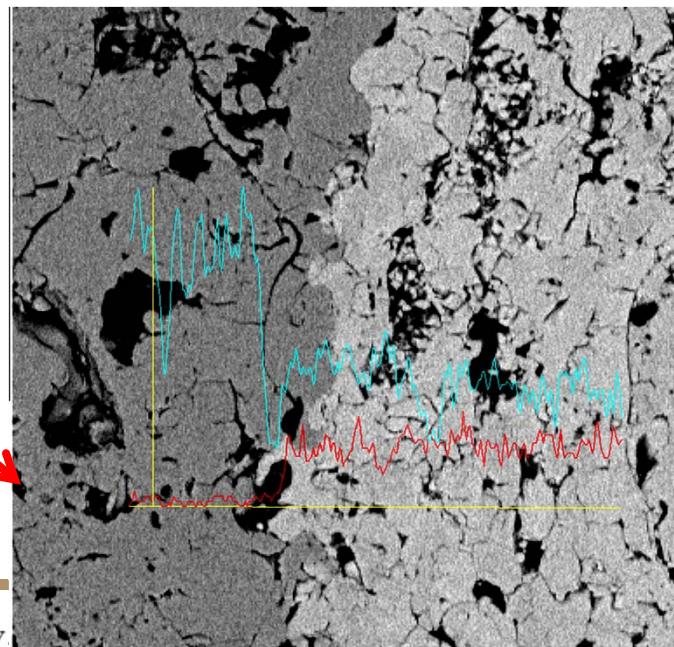
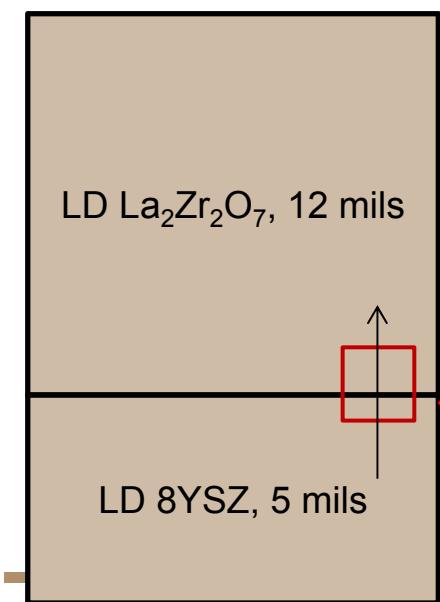
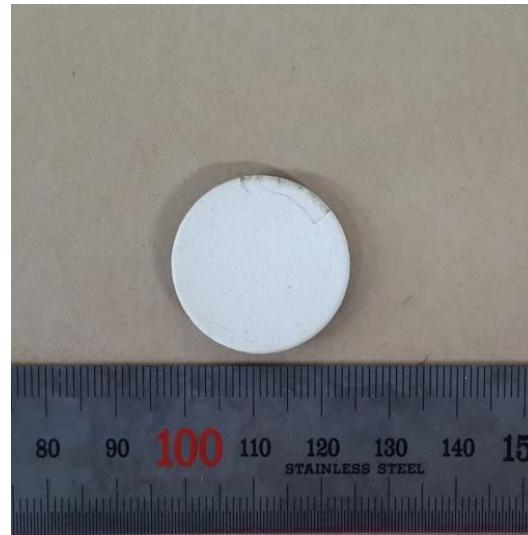
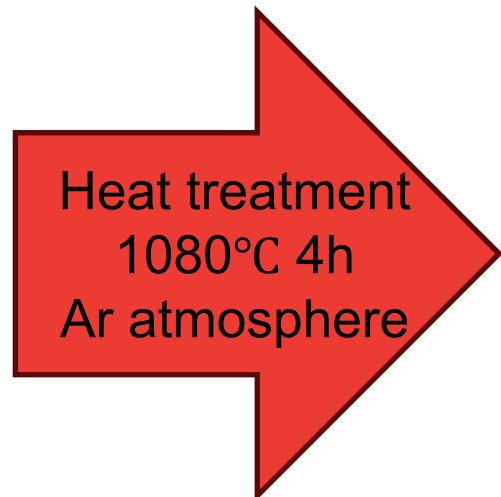
#8 La<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> and porous 8YSZ interface



#9 La<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> and dense 8YSZ interface

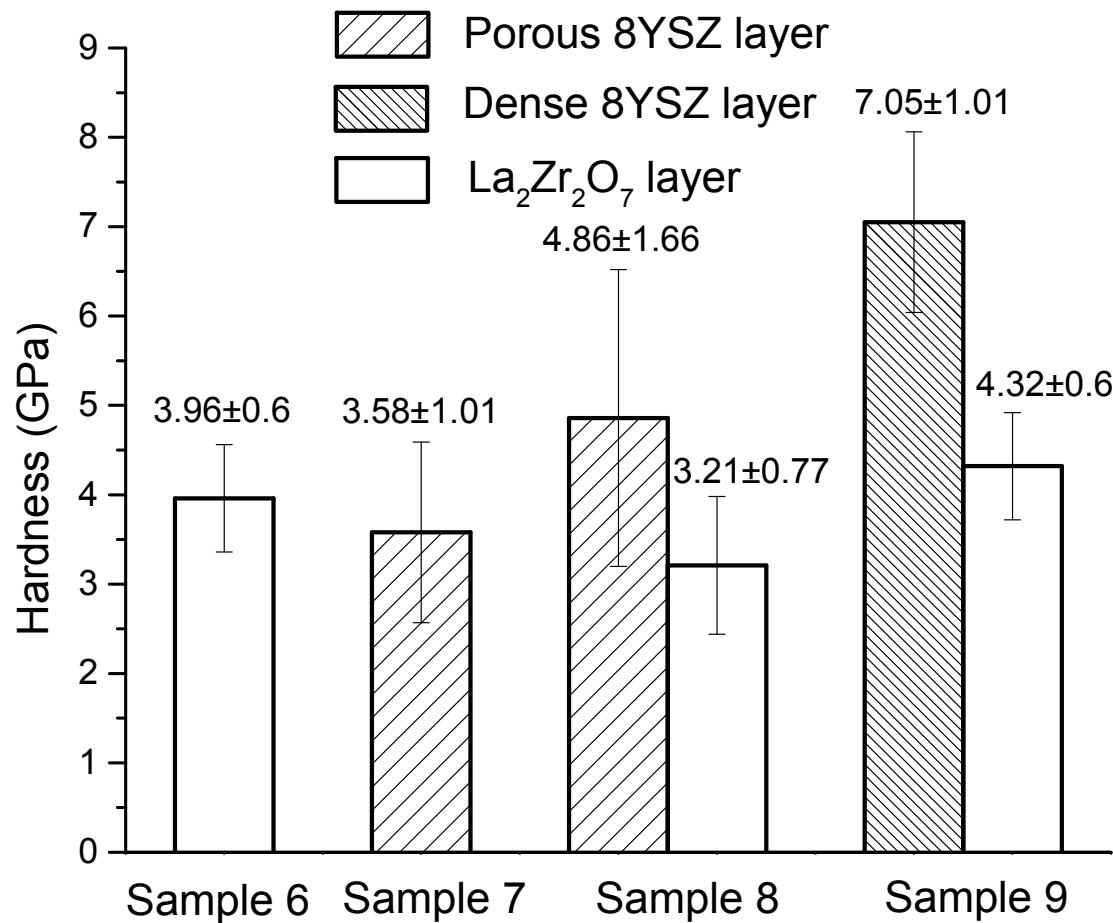
# Energy-dispersive X-ray spectroscopy

Applied heat treatments on sample #8



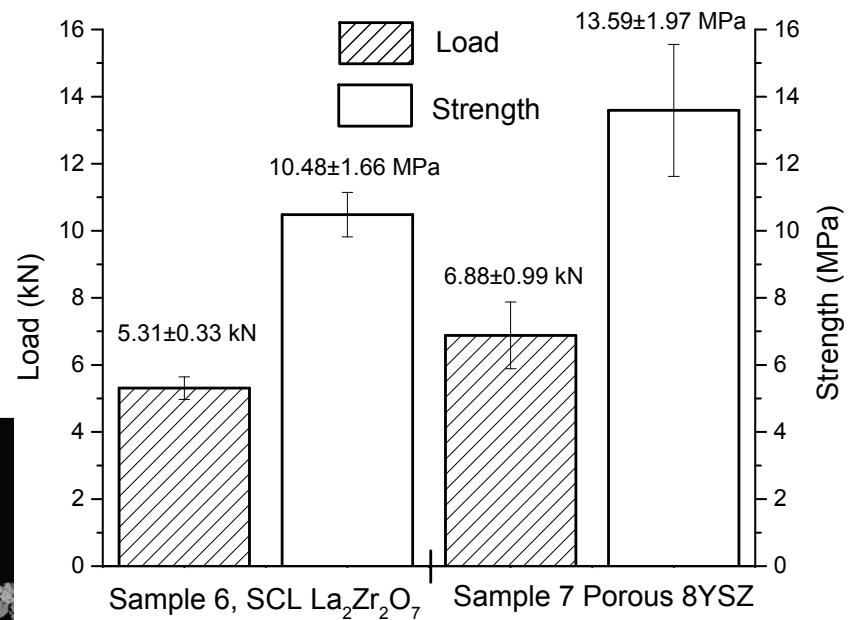
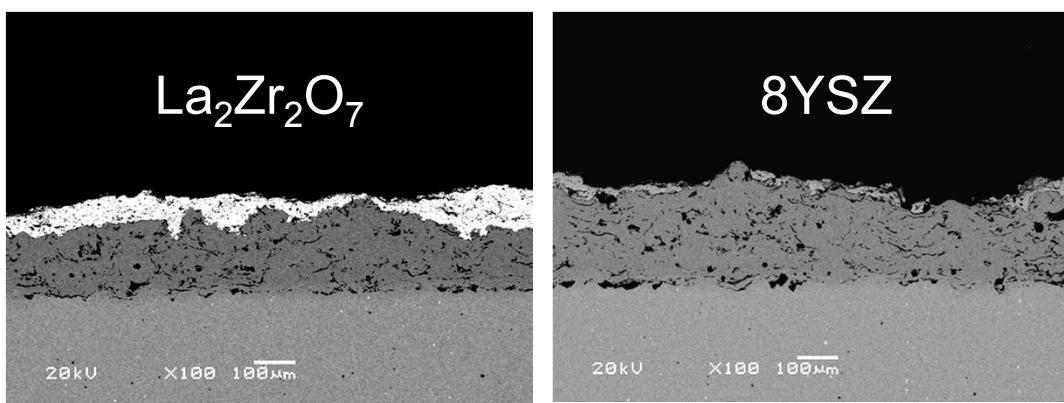
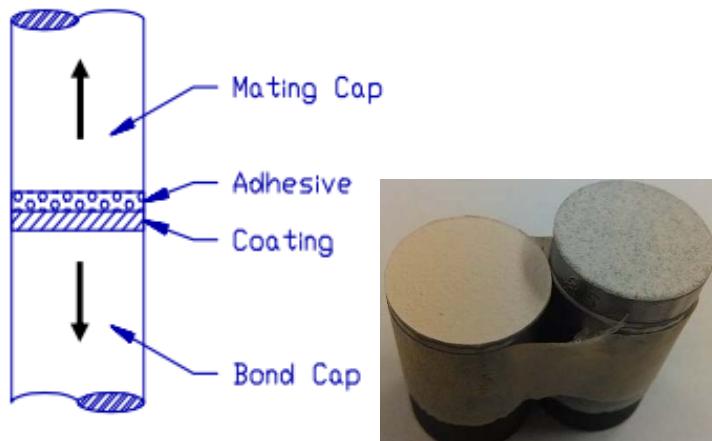
# Vickers hardness of DCL

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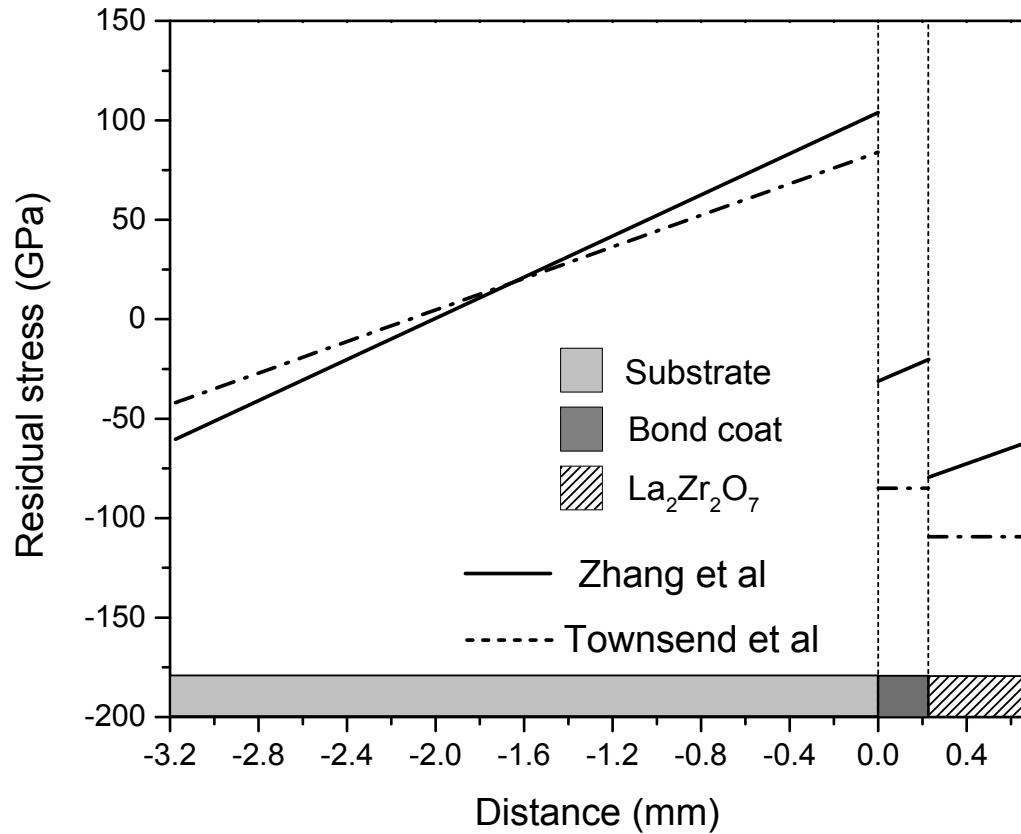


# Bond strength test

Epoxy (FM 1000 adhesive film) to glue coating buttons to a mating cap.  
Tensile test according to ASTM-C-633.



# Residual stress distribution in coating



$$\sigma_s = E_s \left[ \varepsilon_s + K(z + \delta) \right]$$

$$\sigma_i = E_i \left[ \varepsilon_i + K(z + \delta) \right]$$

$$\text{where } \varepsilon_i = \Delta\alpha\Delta T + \sum_{k=1}^n \frac{E_k t_k}{E_s t_s} (\alpha_k - \alpha_i) \Delta T$$

$$\varepsilon_s = -\sum_{i=1}^n \frac{E_i t_i}{E_s t_s} \Delta\alpha\Delta T$$

$$\delta = \frac{t_s}{2} - \sum_{i=1}^n \frac{E_i t_i}{E_s t_s} (2h_{i-1} + t_i)$$

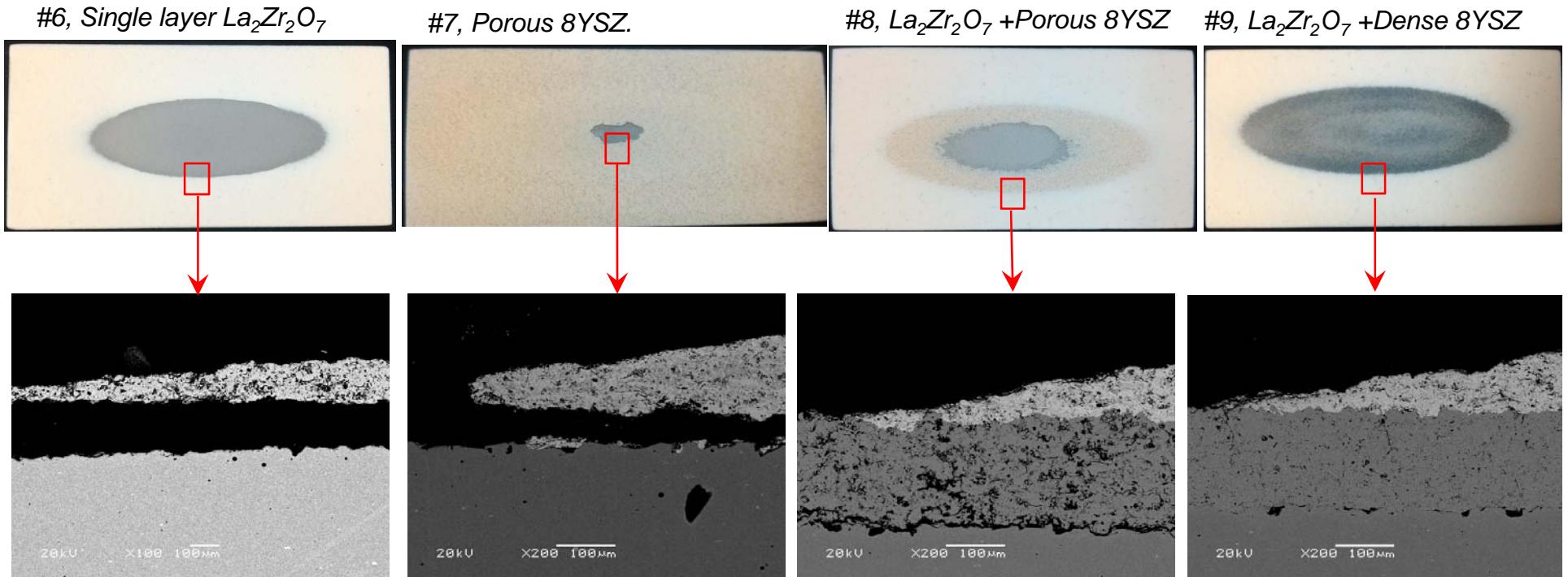
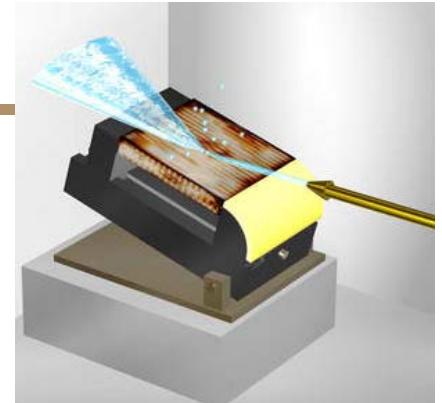
$$K = -\sum_{i=1}^n \frac{6E_i t_i \Delta\alpha\Delta T}{E_s t_s^2}$$

where  $\alpha$  is the coefficient of thermal expansion (CTE),  $k$  is the ceramic coating layers range from 1 to  $n$ ,  $t_i$  is the thickness of  $i^{\text{th}}$  layer.

X.C. Zhang, *Thin Solid Films*, 488 (2005) 274-282.

# Erosion test

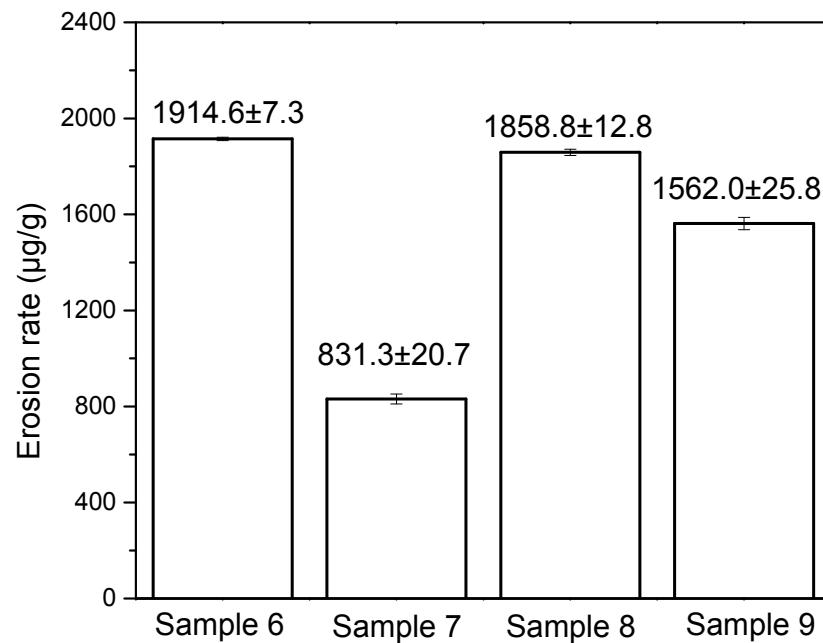
- $600 \pm 0.2\text{g}$  alumina sands with a diameter of  $50\text{ }\mu\text{m}$
- Spray rate  $6\text{ g/s}$ ; duration  $100\text{ s}$ ; spray angle  $20^\circ$



# Erosion rate & critical erosion velocity

Erosion rate describes the erosion resistance of TBC sample [1]:

$$R_{\text{erosion}} = \frac{W_{\text{removed material}}}{W_{\text{impacting particles}}}$$

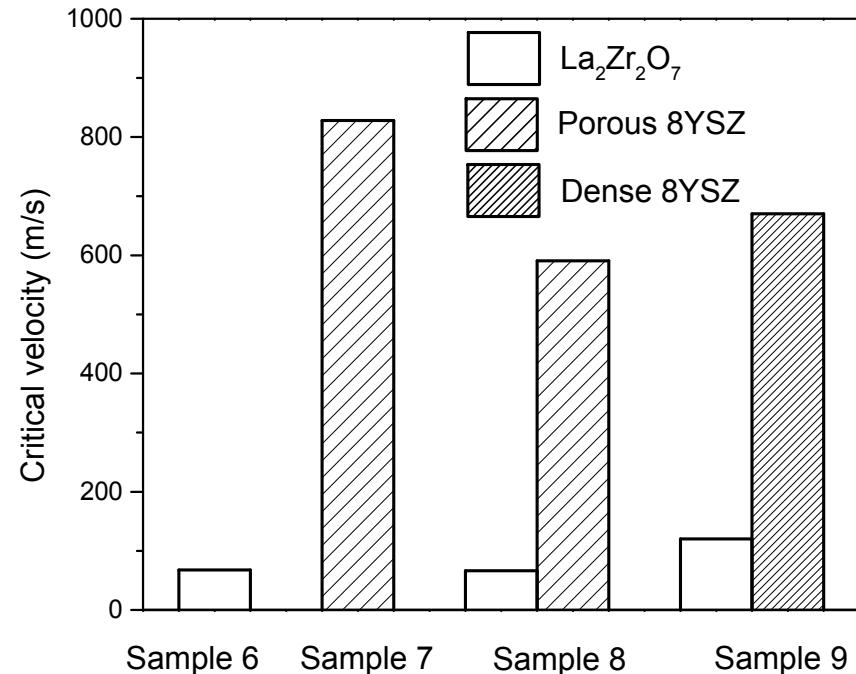


[1] D. Park, Int J Adv Manuf Technol, 23 (2004) 444-450.

Critical erosion velocity is used to express the critical condition to initiate cracks [2]:

$$V_{\text{crit}} = 105 \frac{E^{3/4} K_{\text{IC}}^3}{H^{13/4} \rho^{1/2} R^{3/2}}$$

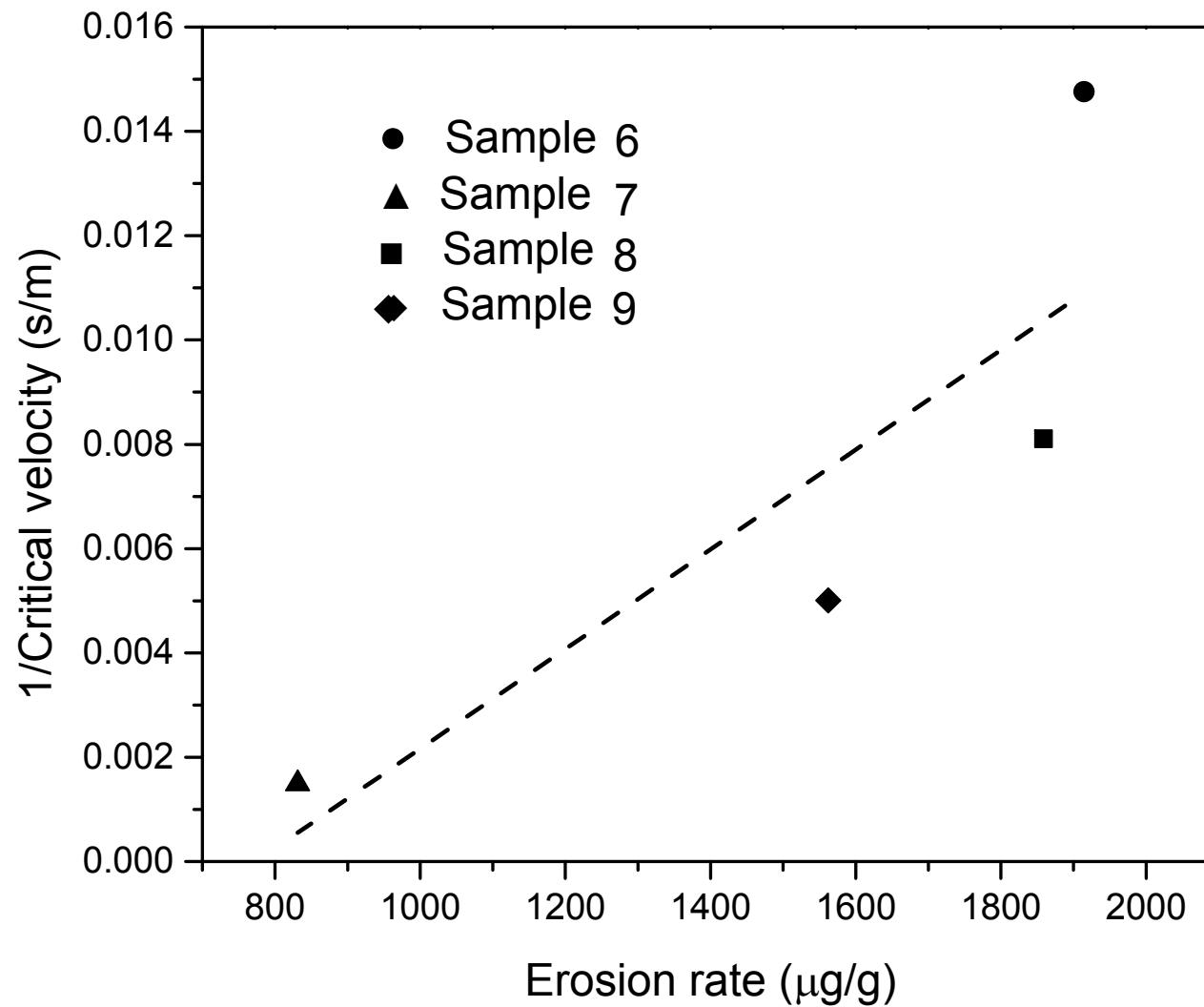
E: Young's modulus  
H: hardness  
 $K_{\text{IC}}$ : fracture toughness  
 $\rho$ : density of erodent particle  
R: particle radius



[2] R.G. Wellman, Wear, 256 (2004) 889-899.

# Relationship between $V_{crit}$ and erosion rate

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# Outline

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- Introduction
  - Coating fabrications
  - Single ceramic layer (SCL) architecture
  - Double ceramic layer (DCL) architecture
  - Characterization of physical and mechanical properties
  - Microstructure and composition
  - Porosity and hardness
  - Bond strength test
  - Erosion test
  - **Characterization of thermal properties**
  - Thermal properties
  - Jet engine thermal shock tests
  - Thermal gradient mechanical fatigue tests
  - Summary and future work
-

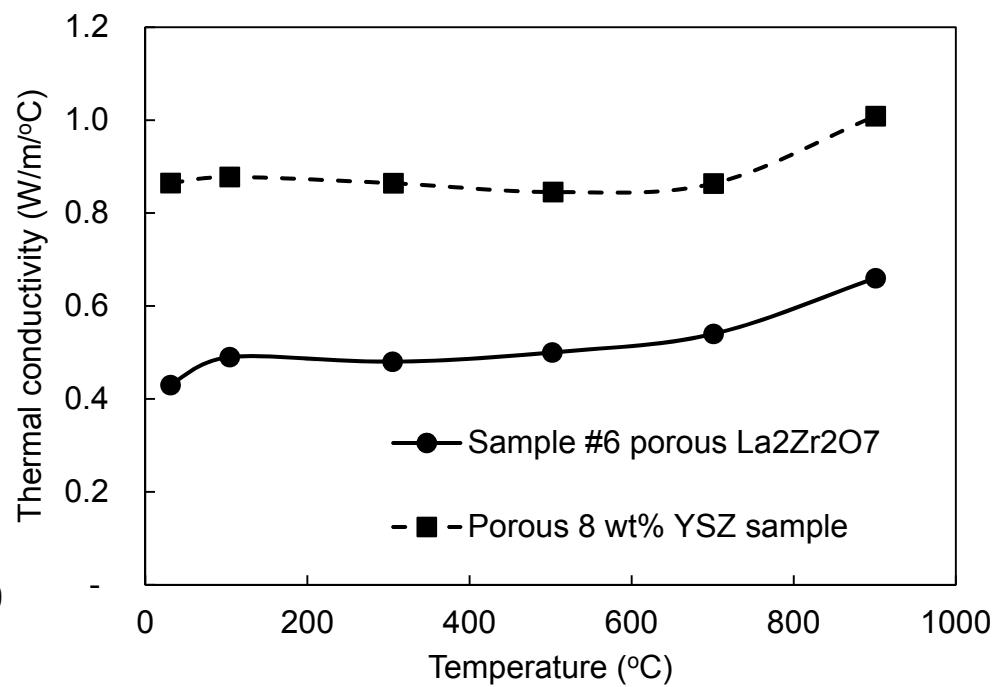
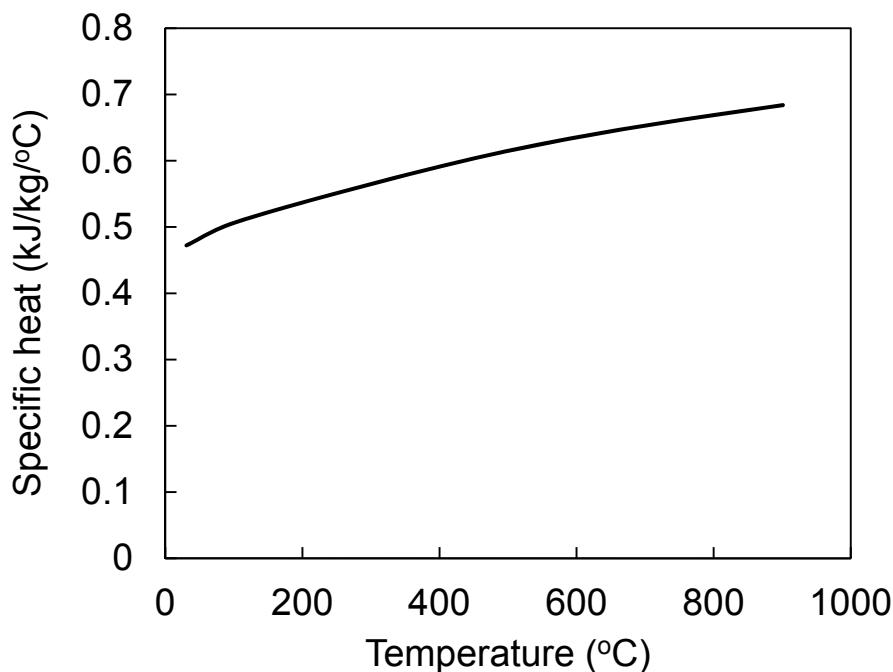
# Thermal conductivity

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Thermal conductivity is determined from thermal diffusivity  $D_{th}$ , specific heat capacity  $C_p$ , and measured density  $\rho$ :

$$k = D_{th} \cdot C_p \cdot \rho$$

Thermal diffusivity is measured using laser flash diffusivity system (TA instrument DLF1200). Specific heat is measured by analytical method (TA instrument DLF1200)



# Thermal conductivity and heat capacity map

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credit:  
Jiangan Sun  
@ ANL

TBC is 90.55% dense ( $\rho=5.478\text{g/cc}$ ), with a nominal thickness of  $600\mu\text{m}$   
Indentation marks are from previous study

# Thermal conductivity and heat capacity map

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## Sample information

TBC:

Material:  $\text{La}_2\text{Zr}_2\text{O}_7$

Thickness:  $\sim 600 \mu\text{m}$  (this is used in calculation)

Density: 90.55% dense, dense density=6 g/cc, so density  $\rho = 5.478 \text{ g/cc}$

Specific heat:  $c = 0.54 \text{ J/g-K}$  @1000C

Substrate (following are room temperature properties obtained from matweb):

Material: Haynes 188

Density:  $\rho = 8.98 \text{ g/cc}$

Thermal conductivity:  $k = 10.4 \text{ W/m-K}$ ,

Specific heat:  $c = 0.403 \text{ J/g-K}$ , (therefore,  $\rho c = 3.62 \text{ J/cm}^3\text{-K}$ )

Thickness used in calculation:  $L = 4 \text{ mm}$  (may have a small effect to results)

## Test condition

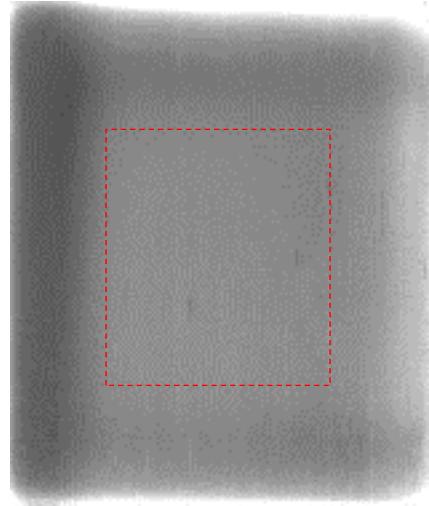
Flash thermal imaging test with one flash lamp

Imaging speed: 994 Hz; imaging duration: 3 seconds

# Measured TBC thermal properties

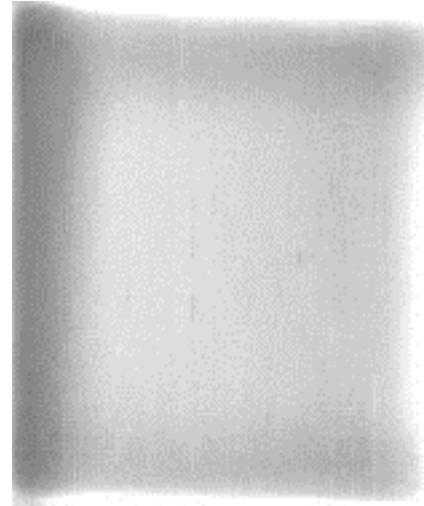
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Thermal conductivity  $k$  image



0 1 W/m-K

Heat capacity  $\rho c$  image



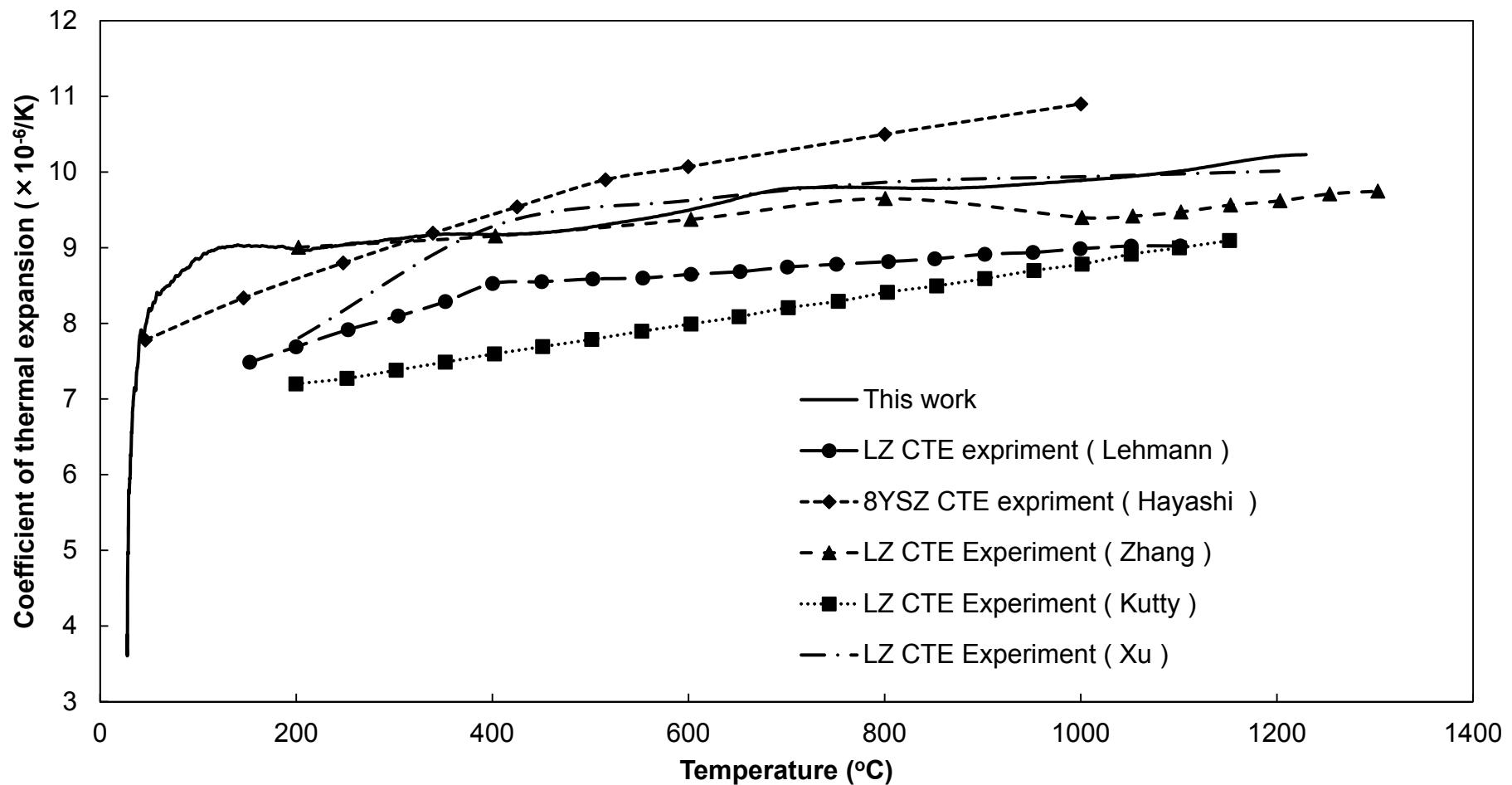
0 2.5  $\text{J}/\text{cm}^3\text{-K}$

credit:  
Jiangan Sun  
@ ANL

Predicted average TBC properties (within red rectangular area):  
 $k = 0.55 \text{ W}/\text{m}\text{-K}$ ,  $\rho c = 2.16 \text{ J}/\text{cm}^3\text{-K}$

- These results were based on a TBC thickness of  $600 \mu\text{m}$
- TBC specific heat @RT:  $c = 0.393 \text{ J}/\text{g}\text{-K}$ ; predicted TBC density is:  $\rho = \rho c / c = 2.16 / 0.393 = 5.5 \text{ g}/\text{cc}$

# Coefficient of thermal expansion (CTE)



CTE is measured using a BAEHR dilatometer from 25 to 1400  $^{\circ}\text{C}$ .

H. Lehmann, D. Pitzer, G. Pracht, R. Vassen, D. Stöver, Journal of the American Ceramic Society, 86 (2003) 1338-1344.

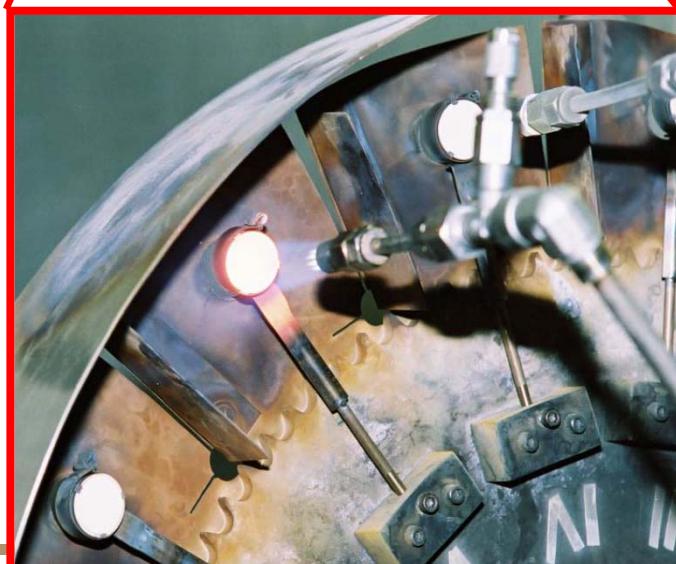
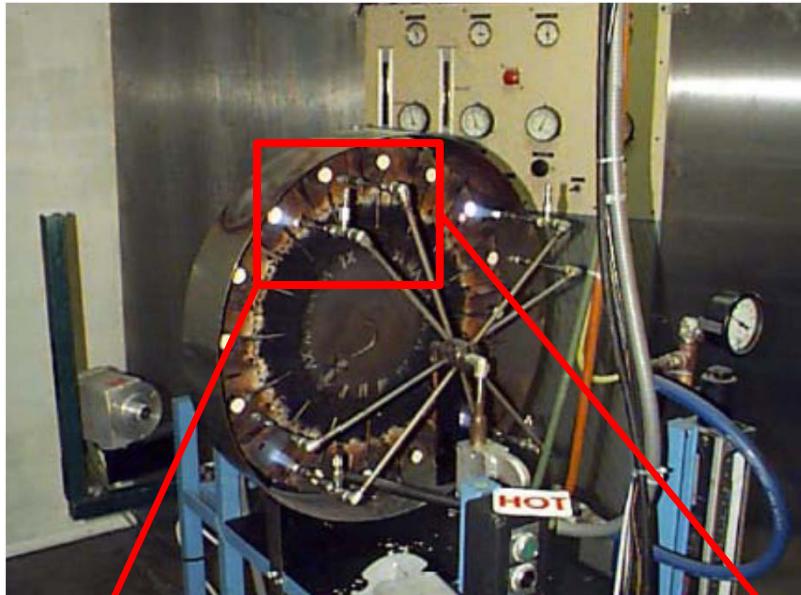
H. Hayashi, T. Saitou, N. Maruyama, H. Inaba, K. Kawamura, M. Mori, Solid State Ionics, 176 (2005) 613-619.

J. Zhang, J. Yu, X. Cheng, S. Hou, Journal of Alloys and Compounds, 525 (2012) 78-81.

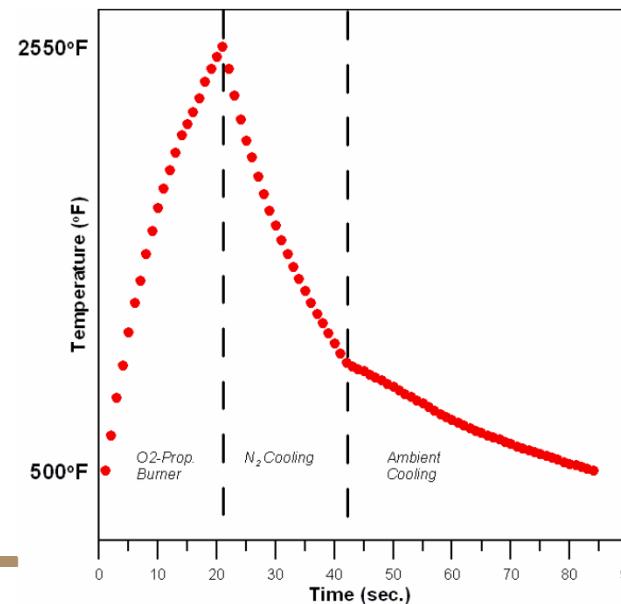
K.V.G. Kutty, S. Rajagopalan, C.K. Mathews, U.V. Varadaraju, Materials Research Bulletin, 29 (1994) 759-766

C. Xu, C. Wang, C. Chan, K. Ho, Physical Review B, 43 (1991) 5024-5027.

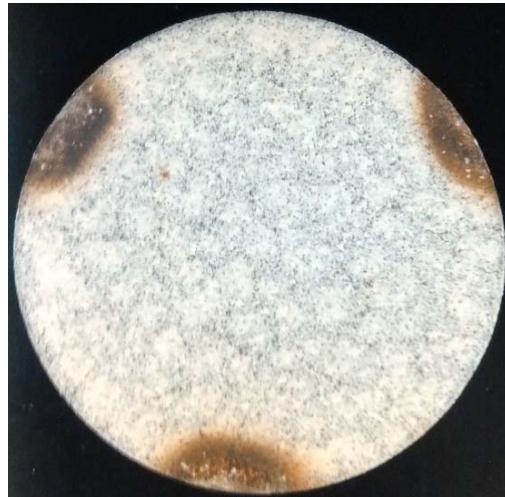
# Jet engine thermal shock tests (JETS)



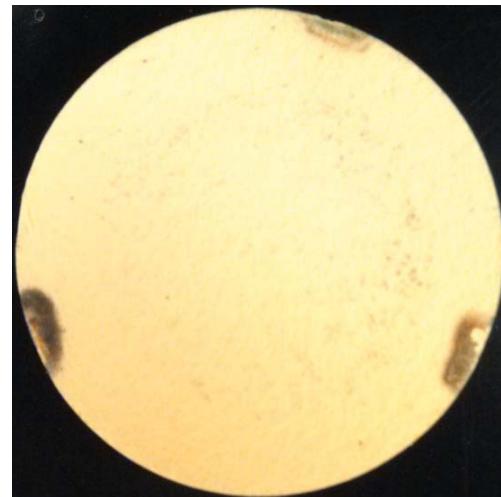
- Jet engine thermal shock (JETS) tests are conducted to investigate the thermal cycling performance.
- TBC samples are heated to 2250 °F (1232.2 °C) at the center for 20 s, and then cooled by compressed N<sub>2</sub> cooling for 20 s, and then ambient cooling for 40 s.
- Temperatures are measured by thermal couple and pyrometer.



# Jet engine thermal shock test (JETS) results



#6, Single layer  $\text{La}_2\text{Zr}_2\text{O}_7$



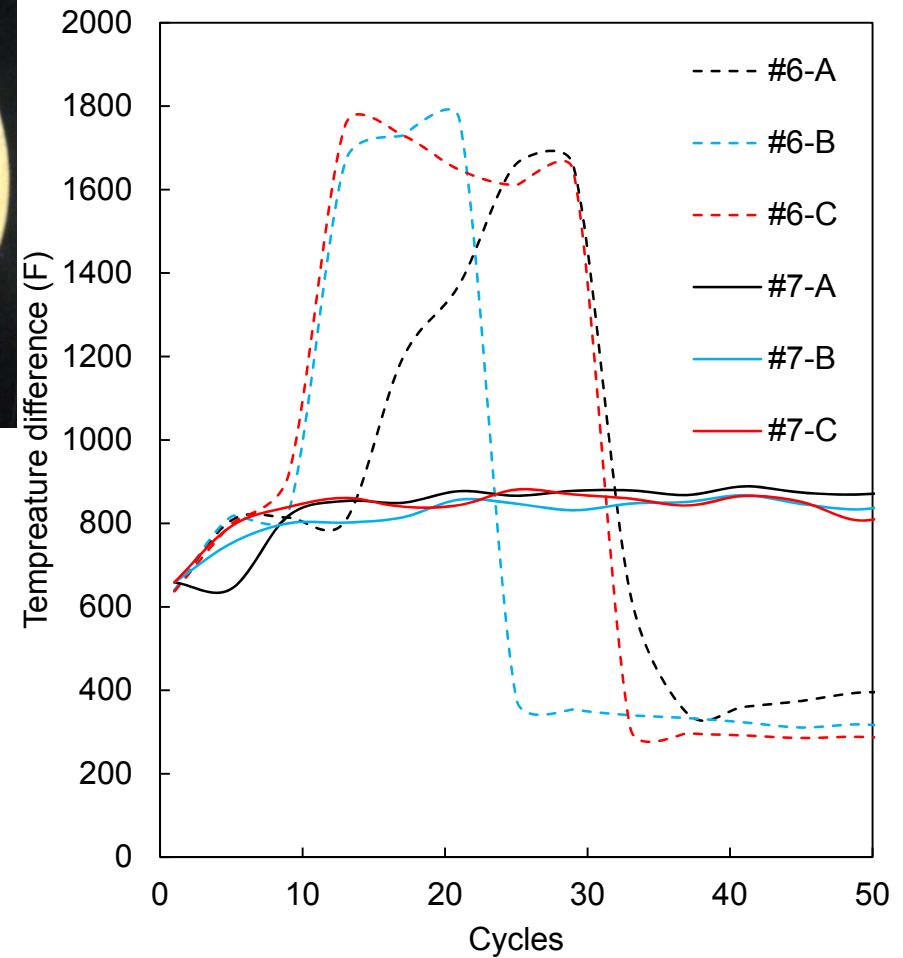
#7, Porous 8YSZ



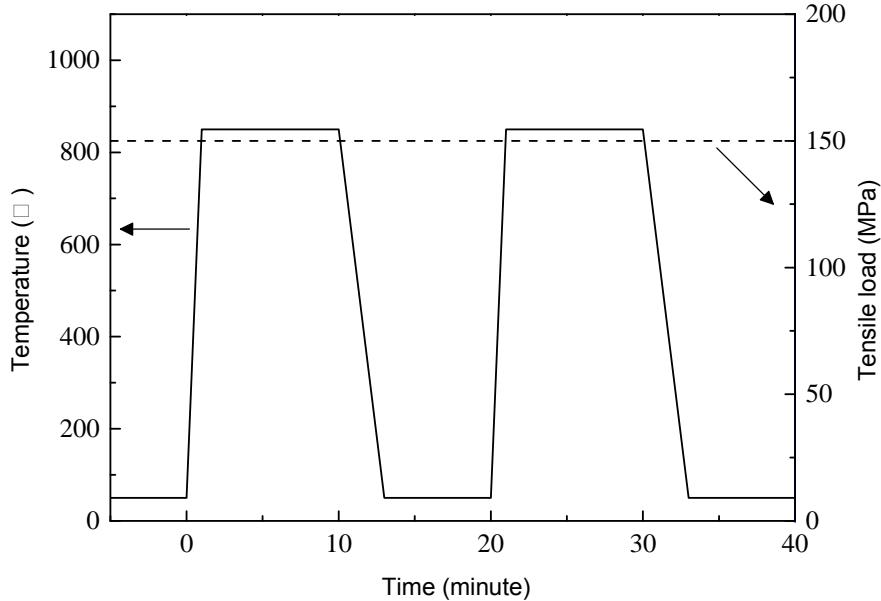
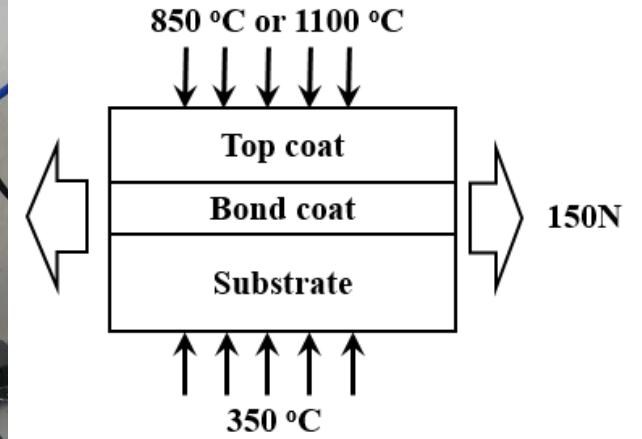
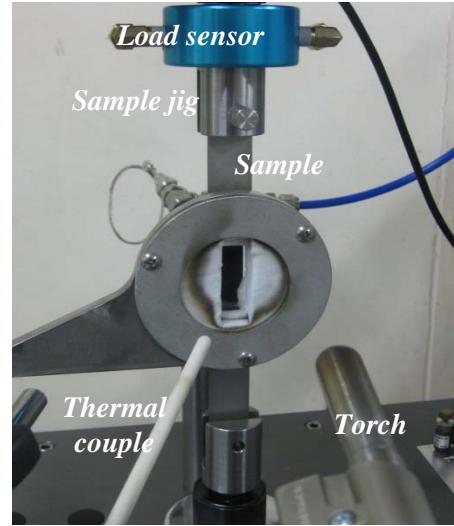
#8, Porous 8YSZ+  $\text{La}_2\text{Zr}_2\text{O}_7$



#9, Dense 8YSZ+  $\text{La}_2\text{Zr}_2\text{O}_7$



# Thermal gradient mechanical fatigue (TGMF)

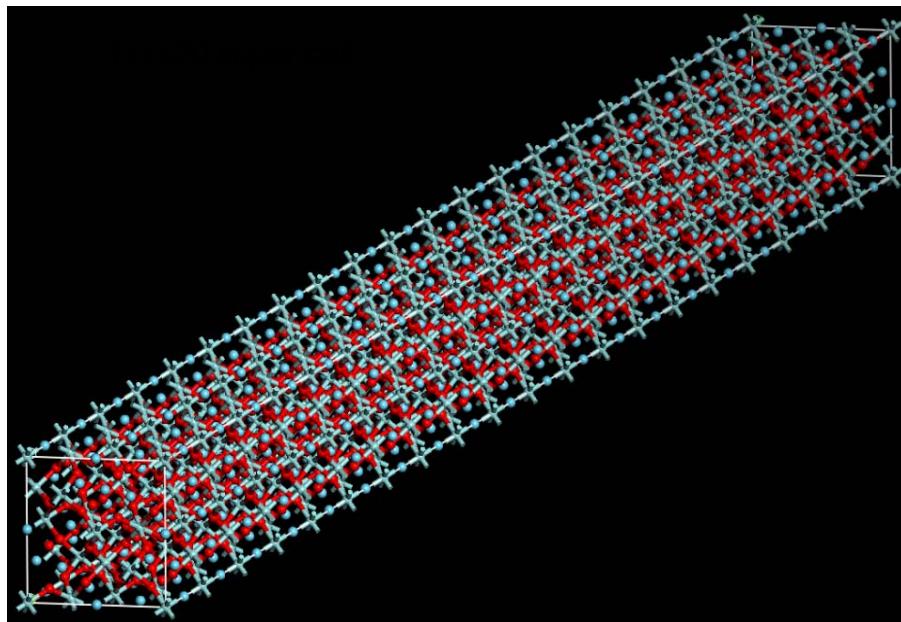


Sample	Test cycle
SCL porous 8YSZ	1200
DCL porous 8YSZ + La <sub>2</sub> Zr <sub>2</sub> O <sub>7</sub>	220
DCL dense 8YSZ + La <sub>2</sub> Zr <sub>2</sub> O <sub>7</sub>	50

At 1100 °C

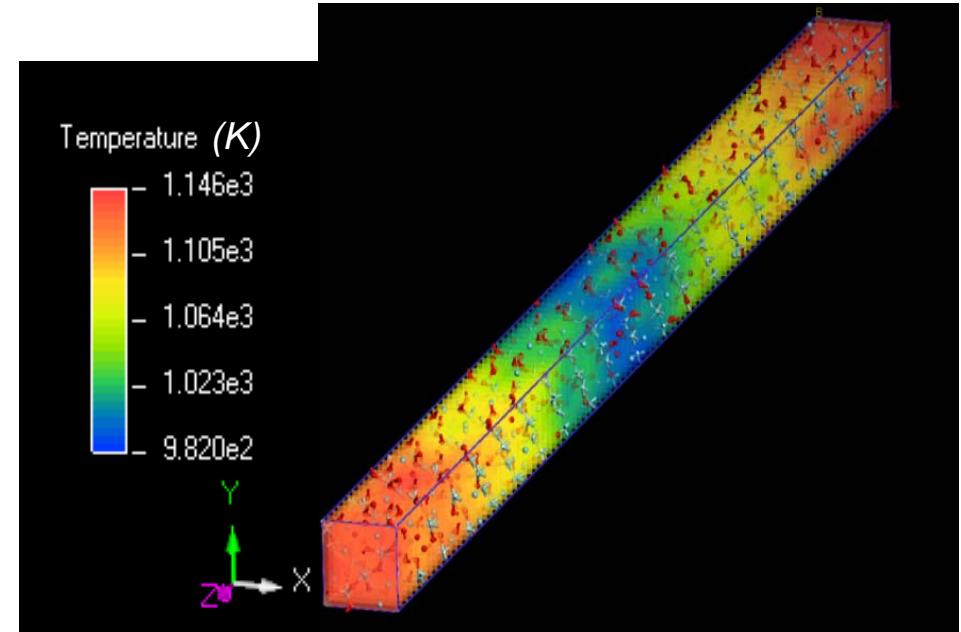
Sample	Test cycle
DCL porous 8YSZ + La <sub>2</sub> Zr <sub>2</sub> O <sub>7</sub>	38
DCL dense 8YSZ + La <sub>2</sub> Zr <sub>2</sub> O <sub>7</sub>	49

# $\text{La}_2\text{Zr}_2\text{O}_7$ thermal conductivity calculation



Replicate 20 conventional cells along the heat flow direction to form a super cell

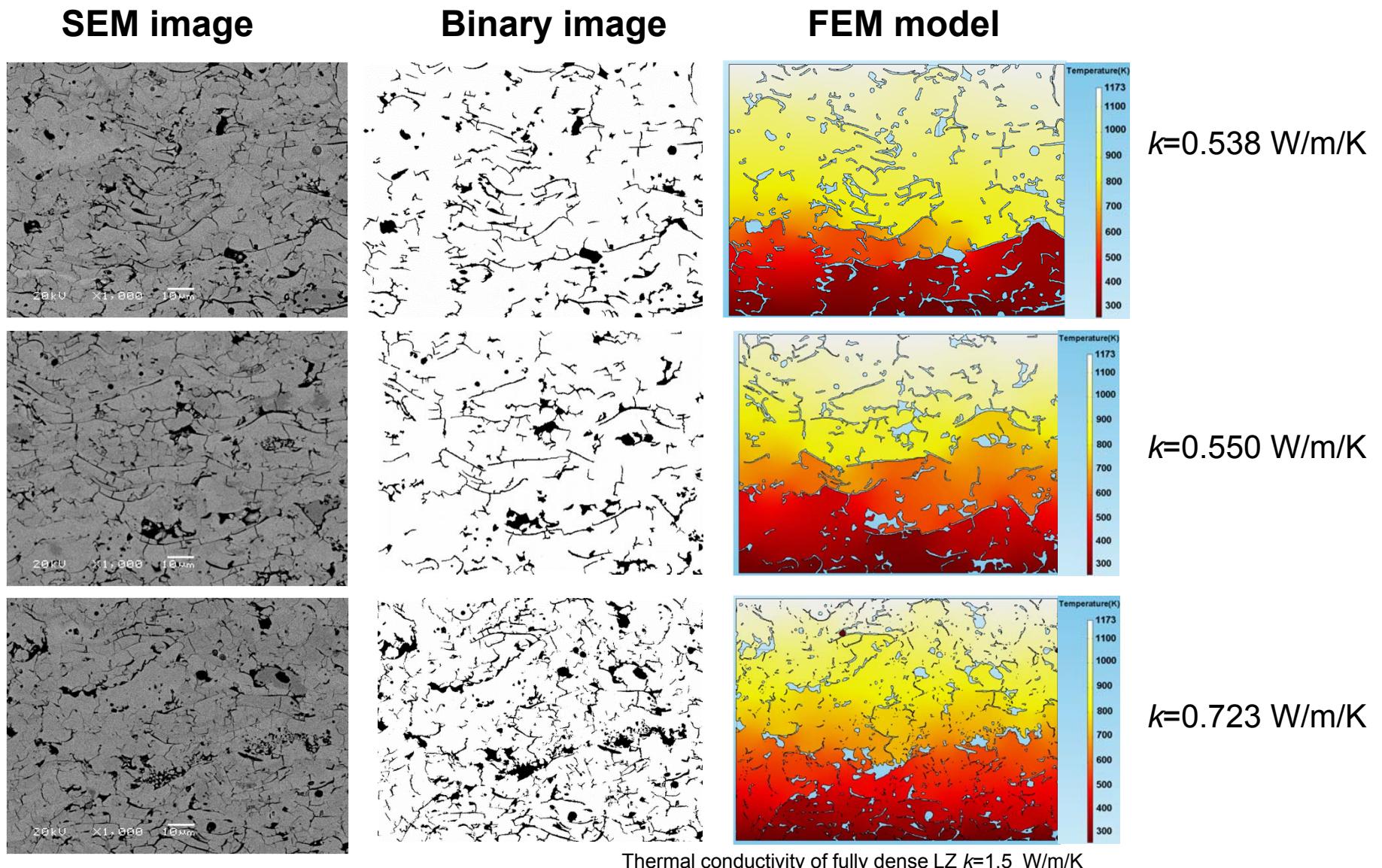
The calculated thermal conductivity is 1.2 W/m/K at the temperature of 1000 °C, which is reasonably in agreement with the experimentally measured thermal conductivity ~1.5 W/m/K [1].



Calculated temperature contour based on Fourier's law  $k = -\vec{q}''/\nabla T$

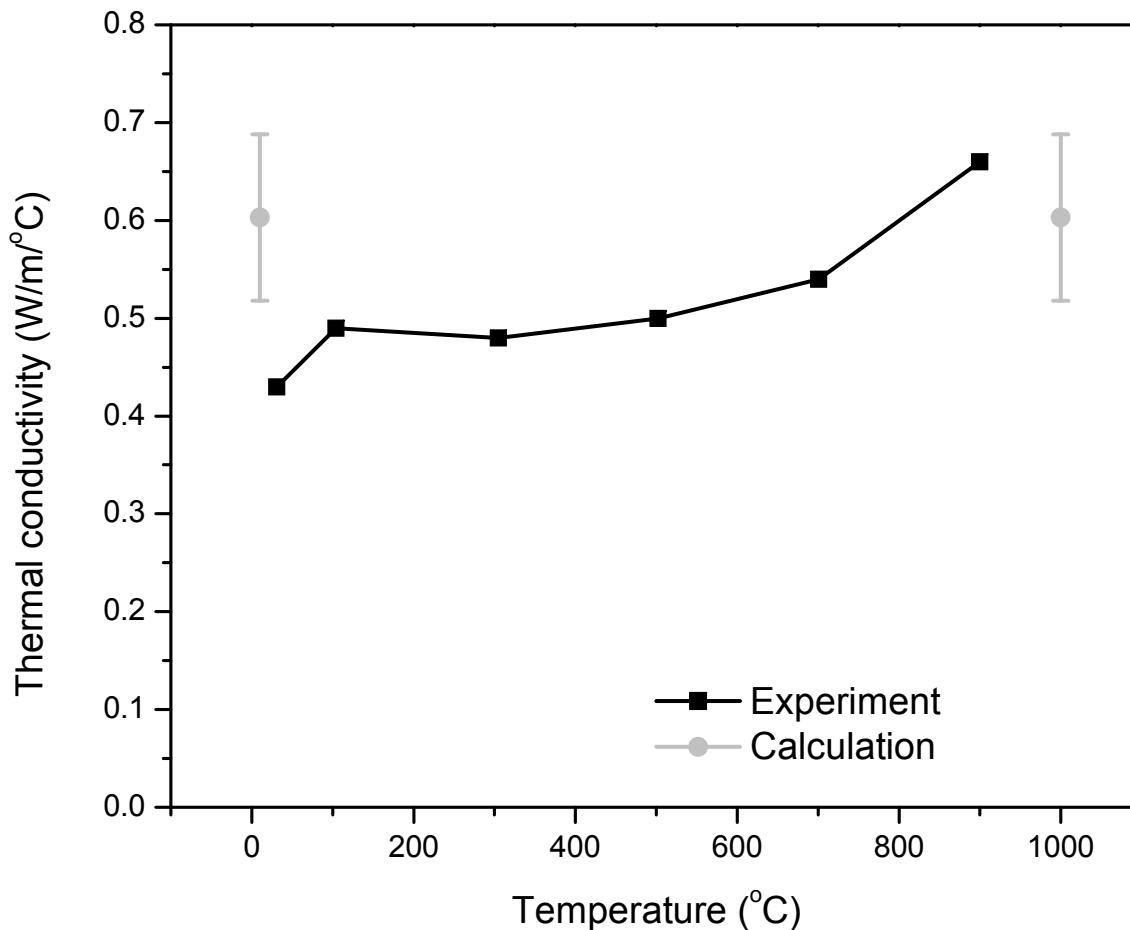
[1] R. Vassen, X. Cao, F. Tietz, J. Am. Ceram. Soc., 83 (2000) 2023–2028.

# Imaged based FEM calculation of thermal conductivity of $\text{La}_2\text{Zr}_2\text{O}_7$ TBC



# Imaged based FEM calculation of thermal conductivity of $\text{La}_2\text{Zr}_2\text{O}_7$ coating

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Calculated thermal conductivity  $0.60 \pm 0.08$  W/m-K,  
in good agreement with experimental data.

# Summary

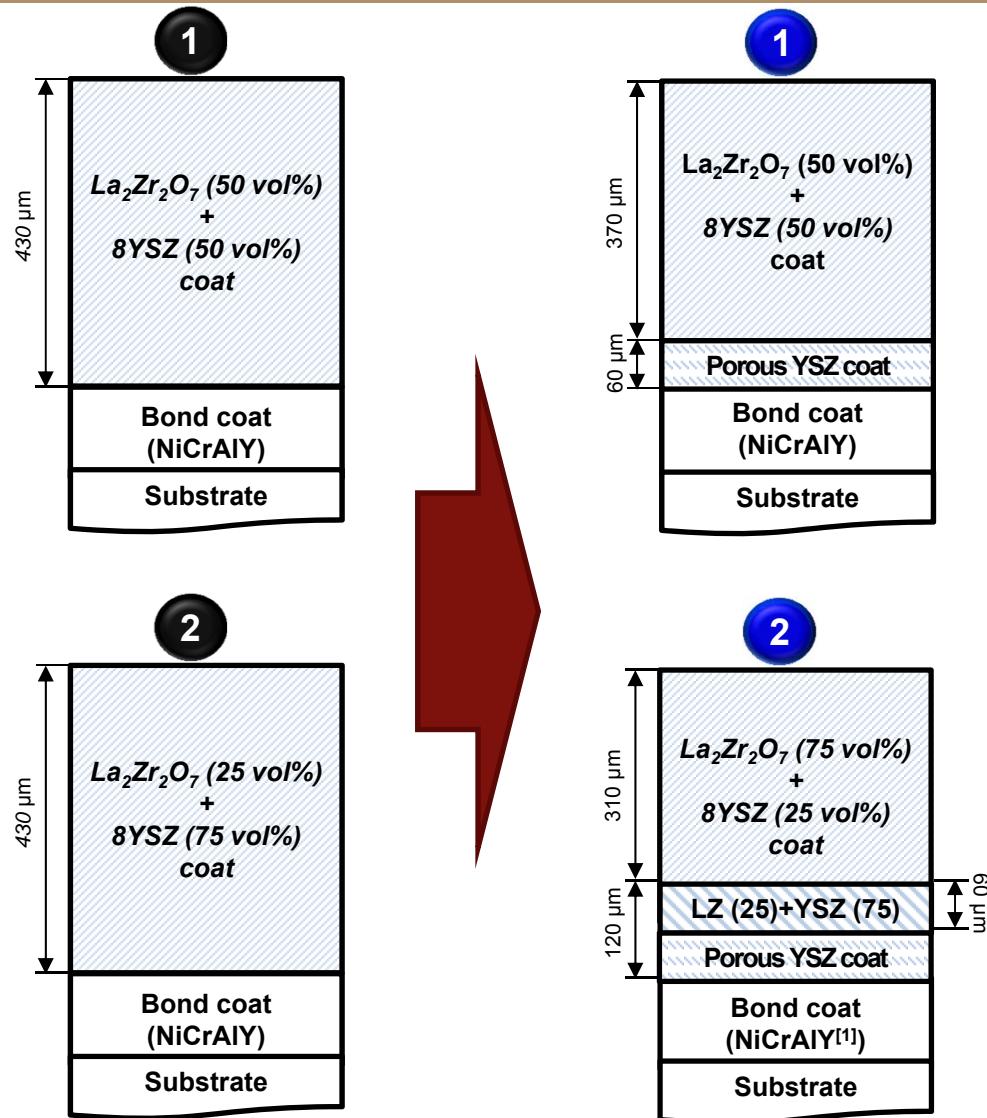
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- $\text{La}_2\text{Zr}_2\text{O}_7$  powder, coating microstructure and chemistry characterizations show that  $\text{La}_2\text{Zr}_2\text{O}_7$  is stable at high temperatures, which makes it suitable for TBC applications.
- Mechanical properties (hardness, bond strength) are similar to 8YSZ.
- Thermal conductivity of  $\text{La}_2\text{Zr}_2\text{O}_7$  is lower than 8YSZ of similar porosity.
- Thermal properties using *ab initio* and image-based finite element model calculations are in good agreement with experiments.

# Future work

- Thermal cycling behavior of  $\text{La}_2\text{Zr}_2\text{O}_7$  needs to be improved.

# Composite coatings with buffer layers



**Composite top coats:**  
thermal conductivity + matching  
CTEs

**Introducing buffer layer:**  
Increasing strain compliance  
+ Decreasing CTEs mismatch

**2<sup>nd</sup> buffer layer:**  
Further decrease CTEs  
mismatch

# Publications and presentations

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1. Jing Zhang, Yeon-Gil Jung, Li Li, co-organize "Advanced Coating Materials for Energy and Environmental Applications" symposium in Materials Science & Technology 2015 (MS&T15), October 4-8, 2015, Columbus, OH
  2. Jing Zhang, Yeon-Gil Jung (eds.), 1st International Joint Mini-Symposium on Advanced Coatings, Materials Today: Proceedings, 2014
  3. Yeon-Gil Jung, Zhe Lu, Ungyu Paik, and Jing Zhang, Lifetime Performance of Thermal Barrier Coatings in Thermally Graded Mechanical Fatigue Environments, The 11th International Conference of Pacific Rim Ceramic Societies(PacRim-11), Jeju, Korea, August 30 - September 4, 2015
  4. Yeon-Gil Jung, Zhe Lu, Qi-Zheng Cui, Sang-Won Myoung, and Jing Zhang, Thermal Durability and Fracture Behavior of Thermal Barrier Coatings in Thermally Graded Mechanical Fatigue Environments, the International Symposium on Green Manufacturing and Applications 2015 (ISGMA 2015), Qingdao, China, June 23 - June 27, 2015
  5. Xingye Guo, Jing Zhang, Zhe Lu, Yeon-Gil Jung, Theoretical prediction of thermal and mechanical properties of lanthanum zirconate nanocrystal, the 1st International Conference & Exhibition for Nanopia, Changwon Exhibition Convention Center, Gyeongsangnam-do Province, Miryang City, Korea, November 13-14, 2014
  6. Sang-Won Myoung, Zhe Lu, Qizheng Cui, Je-Hyun Lee, Yeon-Gil Jung, Jing Zhang, Thermomechanical properties of thermal barrier coatings with microstructure design in cyclic thermal exposure, the 1st International Conference & Exhibition for Nanopia, Changwon Exhibition Convention Center, Gyeongsangnam-do Province, Miryang City, Korea, November 13-14, 2014
  7. Zhang, J., X. Guo, Y.-G. Jung, L. Li, and J. Knapp, Microstructural Non-uniformity and Mechanical Property of Air Plasma-sprayed Dense Lanthanum Zirconate Thermal Barrier Coating. Materials Today: Proceedings, 2014. 1(1): p. 11-16.
  8. Guo, X. and J. Zhang, First Principles Study of Thermodynamic Properties of Lanthanum Zirconate. Materials Today: Proceedings, 2014. 1(1): p. 25-34.
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