#### **SRI International**

# CO<sub>2</sub> Capture from IGCC Gas Streams Using the AC-ABC Process

2011 NETL CO<sub>2</sub> Capture Technology Meeting August 22-26, 2010 Pittsburgh, PA.

#### **Project Overview**

- Partners:
  - DOE-National Energy Technology Center, Awarding Agency
  - SRI International, Menlo Park, CA
  - EIG, Sunnyvale, CA
  - New Partnerships, 2011
- Period of Performance:
  - 10-1-2009 through 3-30-2012
- Funding:
  - U.S.: Department of Energy: \$3.42 million
  - Cost share: \$1.08 million
  - Total: \$4.5 million

#### **Project Objectives**

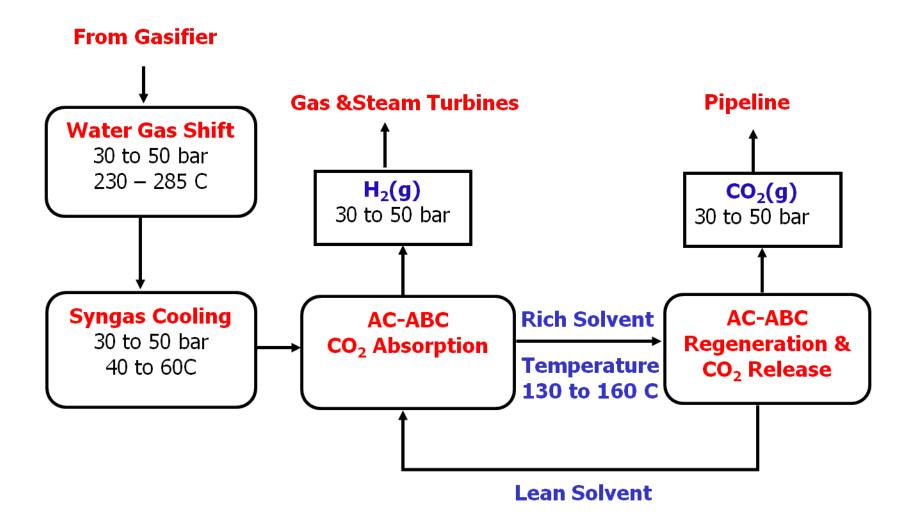
#### Overall objective:

 To develop an innovative, low-cost CO<sub>2</sub> capture technology based on absorption on a high-capacity and low-cost aqueous ammoniated solution.

#### Specific objectives:

- Test the technology on a bench scale batch reactor to validate the concept.
- Determine the optimum operating conditions for a small pilot-scale reactor.
- Design and build a small pilot-scale reactor capable of continuous integrated operation.
- Perform tests to evaluate the process in a coal gasifier environment.
- Perform a technical and economic evaluation on the technology.

### **Process Flow Diagram**



## NH<sub>3</sub>, CO<sub>2</sub>(NH<sub>3</sub>)\* NH<sub>3</sub>, CO<sub>2 (g)</sub> NH<sub>4</sub>(NH<sub>2</sub>CO<sub>2</sub>)

#### **Sound Underlying Concepts**

Uses well-known reaction between CO₂ and aqueous ammonia

```
NH_4OH+CO_2 \longleftrightarrow NH_4HCO_3

(NH_4)_2CO_3+CO_2 + H_2O \longleftrightarrow 2NH_4HCO_3

NH_4(NH_2CO_2)+CO_2+2H_2O \longleftrightarrow 2NH_4HCO_3
```

- Reactions are reversible
  - Absorption reactions at lower temperature
  - Desorption reactions at higher temperature
- Relatively low heat of reaction (300-600 Btu/lb of  $CO_2$  depending on the  $NH_{3}/CO_2$  ratio in the solution ).
- High pressure operation enhances absorption of CO<sub>2</sub>.
- A similar set of reactions occur between H<sub>2</sub>S and ammoniated solution.

#### **Process Highlights**

- Concentrated ammoniated solution is used to capture CO<sub>2</sub> and H<sub>2</sub>S from syngas at high pressure.
- Absorber operation at 40°-60° C temperature; No refrigeration is needed.
- CO<sub>2</sub> is released at high pressures:
  - The size of CO<sub>2</sub> stripper, the number of stages of CO<sub>2</sub> compression and inter-cooling and the electric power consumption for compression of CO<sub>2</sub> compression to the pipeline pressure is reduced.
- High net CO<sub>2</sub> loading, up to 20% by weight.
- H<sub>2</sub>S is released at conditions suitable for sulfur recovery.

#### **Process Advantages**

- Low cost and readily available reagent.
- Reagent is chemically stable under the operating conditions.
- Low heat consumption for CO<sub>2</sub> stripping (<600 Btu/lb CO<sub>2</sub>).
- Extremely low solubility of H<sub>2</sub>, CO, and CH<sub>4</sub> in absorber solution.
  - Minimizes losses of fuel species.
- Absorber and regenerator can operate at similar pressure.
  - No need to pump solution cross pressure boundaries. Low energy consumption for pumping.

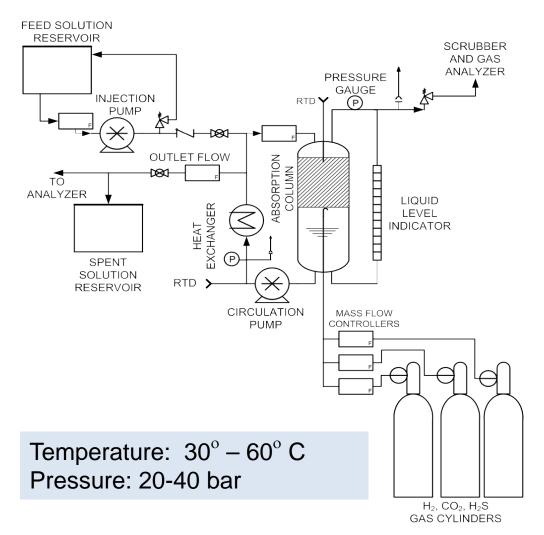
### **Project Tasks**

- 1. Bench-scale Batch Tests
- 2. Pilot-Scale Integrated, Continuous Tests
- 3. Project Management

#### **Bench-Scale Absorber Testing**

- Determination of solubility:
  - Shifted-gas components (H<sub>2</sub>, CO, N<sub>2</sub>, Ar)
- Determination of reactivity of CO<sub>2</sub> and H<sub>2</sub>S:
  - Function of composition, pressure, and temperature.
- Mixed-gas testing to determine the relative reaction kinetics.

### **Bench-Scale Absorber System**





Reactor ID: 4-in

Low Pressure Drop

Specific Area: 425 m<sup>2</sup>/m<sup>3</sup>

Packing Height: 2-ft

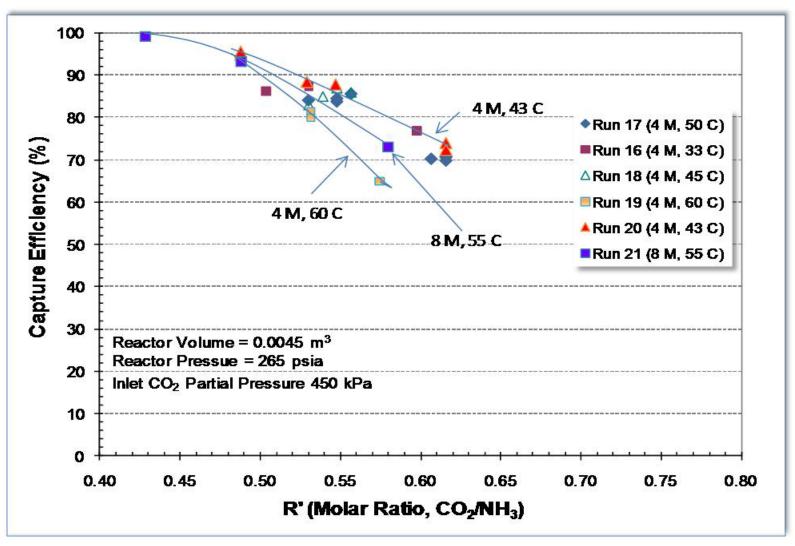
Liquid Loading: 5 – 10%

### **Determination of Fuel Gas Component Solubility**

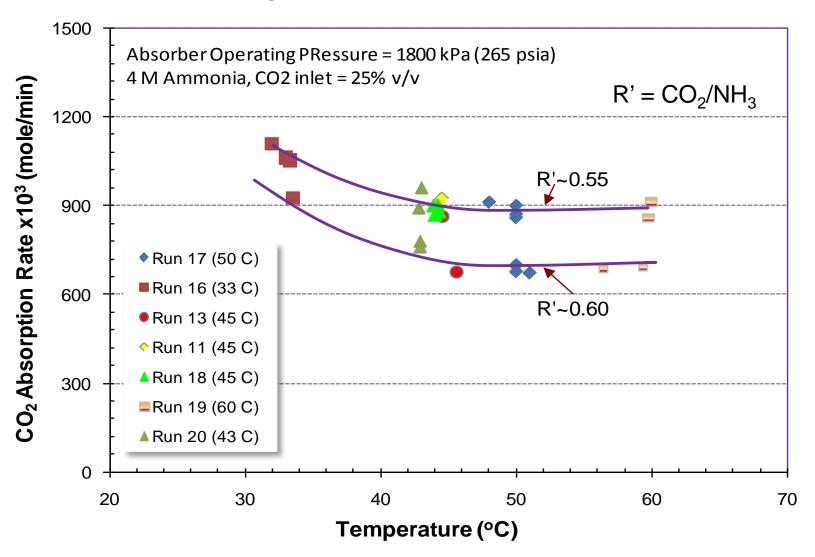
Gas Type	Gas Component Concentration (%v/v)	Dissolved Gas (g/kg solution at 40 atm)
H <sub>2</sub>	50.0	6.53x10 <sup>-3</sup>
СО	2.0	3.62x10 <sup>-4</sup>
CH <sub>4</sub>	2.0	4.67x10 <sup>-4</sup>
$N_2$	1.0	1.11×10 <sup>-4</sup>

### CO<sub>2</sub> Capture Efficiency vs Solution Composition

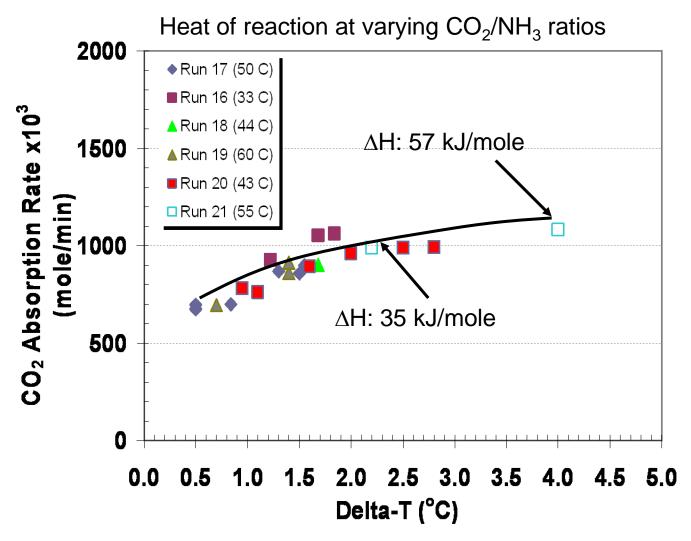
CO<sub>2</sub> Capture Efficiency Exceeds 90%



## No Significant Decrease in the Rate of Absorption at Elevated Temperatures

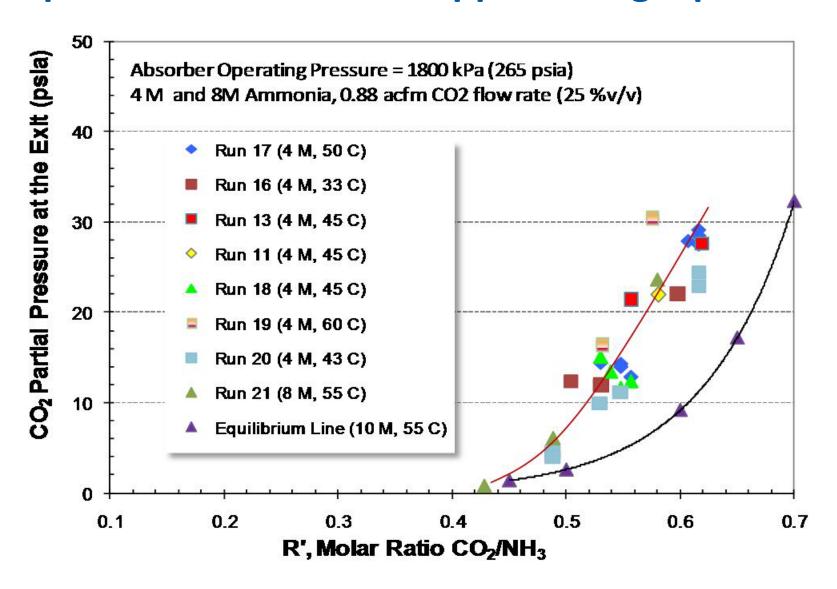


## Moderate Heat of Absorption to Reduce Cooling Requirements



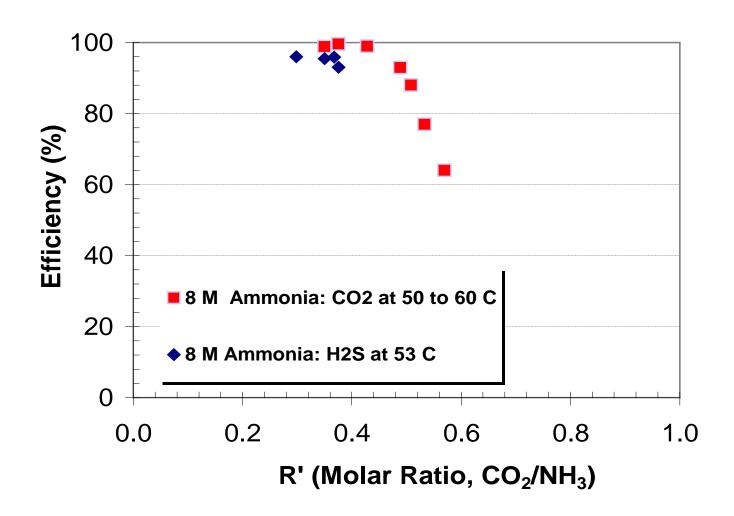
Target: <60 kJ/mole

#### Rapid Rate of Reactions Approaching Equilibrium

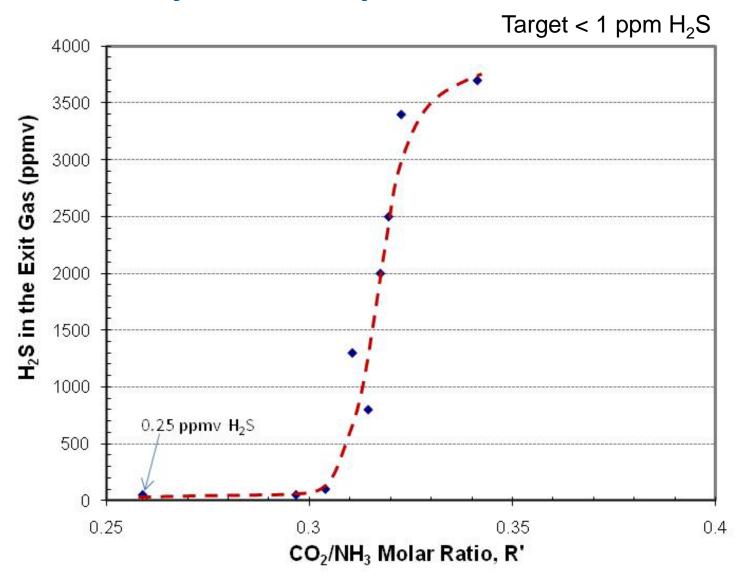


#### High Efficiency of H<sub>2</sub>S Capture

(2-ft, Single Stage Absorber Column)

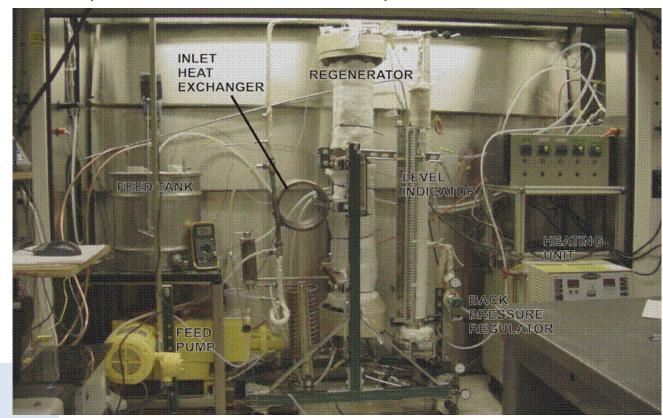


### High Efficiency of H<sub>2</sub>S Capture



### **Bench-Scale Regenerator Testing**

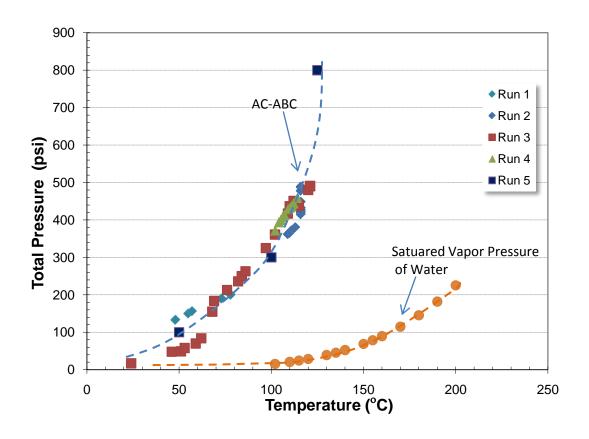
- Determination of CO<sub>2</sub> and H<sub>2</sub>S release characteristics
  - Function of temperature, pressure, and solution composition



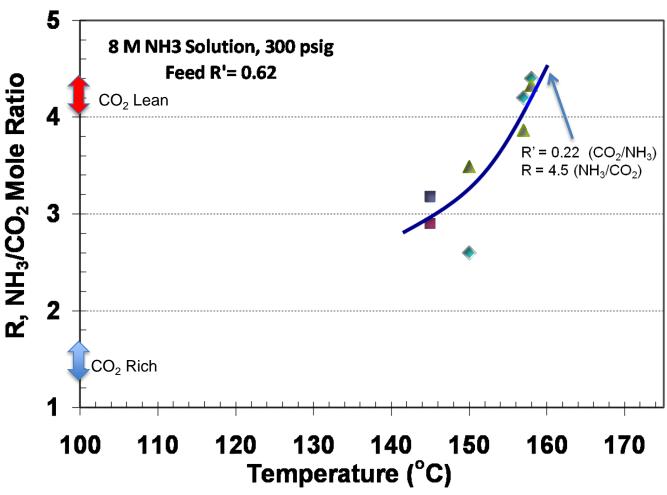
T: 100 - 170 C P: 10-40 bar

Feed CO<sub>2</sub> Loading: 10-20 wt%

## Measured CO<sub>2</sub> Attainable Pressure Function of Temperature



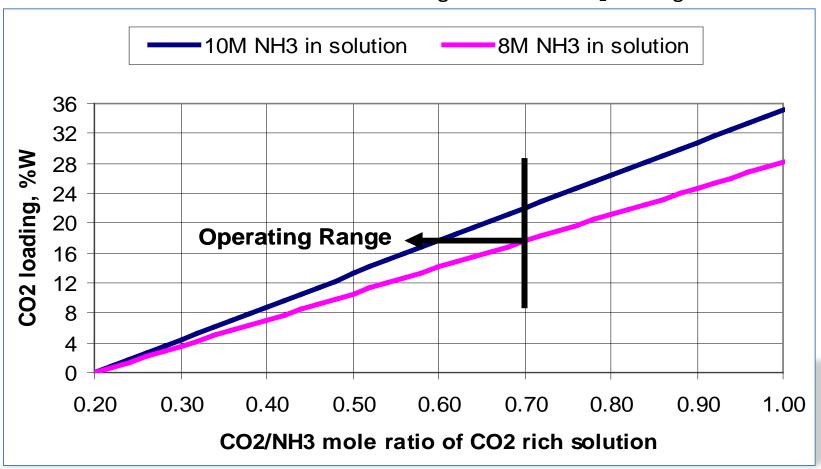
## High Levels of Solvent Regeneration at Moderate Temperatures



Target R Value: >4

## CO<sub>2</sub> Capacity: Function of Solution Composition

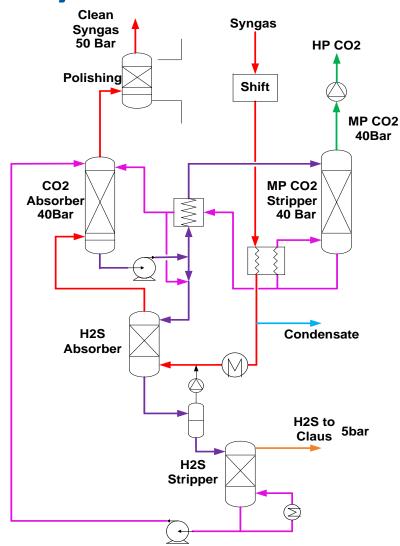
Target: >15 wt% CO<sub>2</sub> loading



Solubility of NH<sub>4</sub>HCO<sub>3</sub> at 50°C: 70 wt%

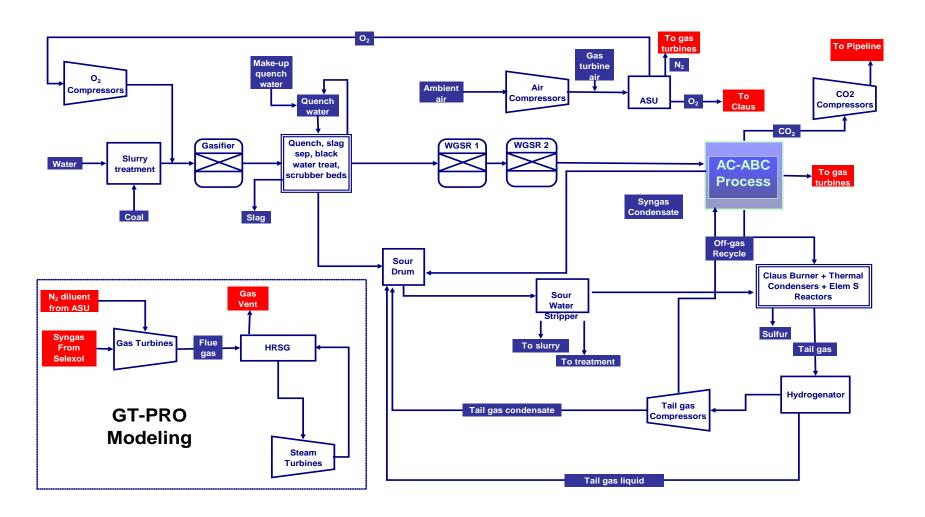
#### **Technical and Economic Analysis**

- Compare the AC-ABC process with a similar-size plant using CO<sub>2</sub> capture with Selexol subsystem.
- Base case is an IGCC plant (750 MW nominal) with no CO<sub>2</sub> capture.
- Generate the equipment sizing, heat, and material flows using Aspen and GT-Pro modeling.
- Use DOE spread sheet to generate cost.



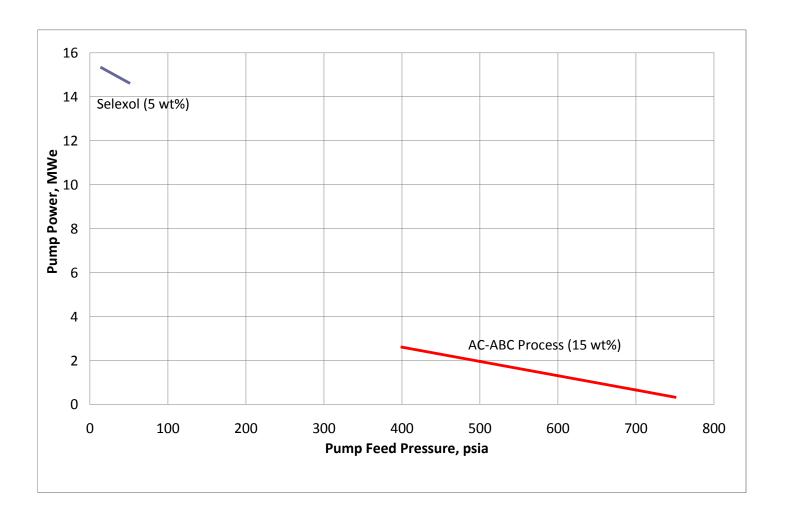
Schematic of the CO<sub>2</sub> and H<sub>2</sub>S Capture System

### **Block Flow Diagram**

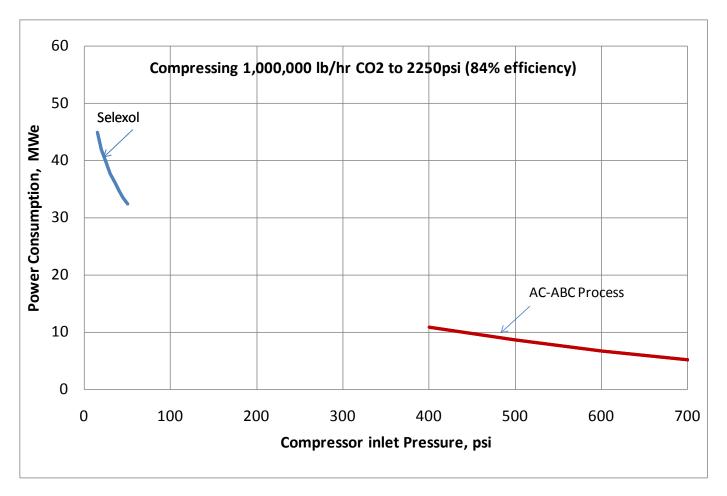


Process Energy Requirements: CO<sub>2</sub> stripping, solution pumping, and CO<sub>2</sub> compression

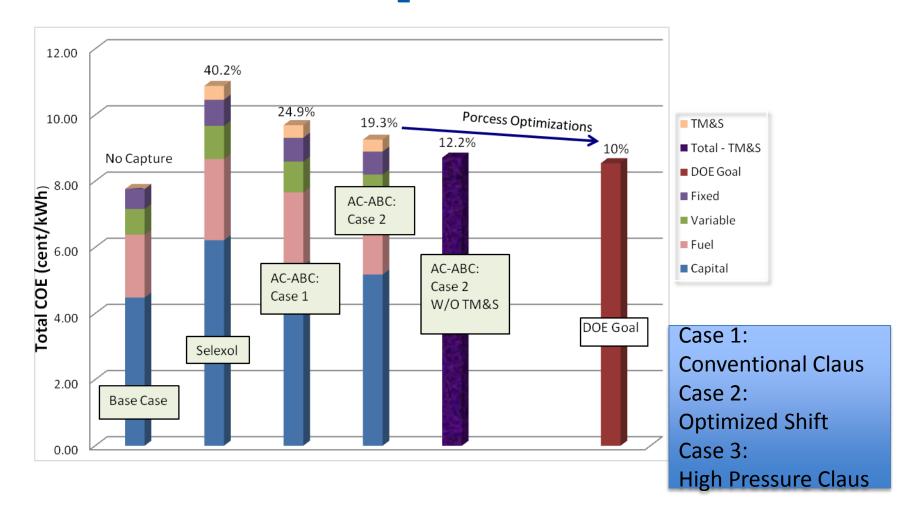
### **Energy Saving in Solution Recirculation**



## **CO<sub>2</sub> Compression Power Saving**

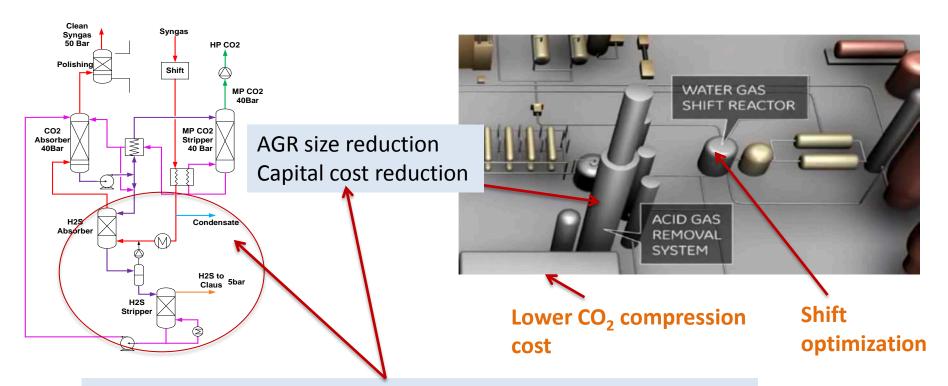


## Performance and Economic Factors: Preliminary Cost of CO<sub>2</sub> Capture



CO2 capture: 3.3 million tons/year; Plant operting life: 30 years; Capacity factor: 80%

## Performance Improvements to Reach The DOE Goal of 10% COE



- Sulfur recovery from the high pressure CO<sub>2</sub> stream
  - No need for selective CO<sub>2</sub> and H<sub>2</sub>S absorption
  - Single stage absorber for both H<sub>2</sub>S and CO<sub>2</sub> Capture
  - H<sub>2</sub>S and CO<sub>2</sub> regenerated together

AC-ABC: Case 3

#### Bench-Sale Testing

#### COMPLETED

- Smooth operation of bench-scale system without fouling (20 bar, 170 C, 20 wt% loading)
- Very high levels of CO<sub>2</sub> and H<sub>2</sub>S capture efficiencies (>90%)
- Rapid CO<sub>2</sub> absorption and regeneration
- Preliminary estimation of the cost of CO<sub>2</sub> capture in IGCC systems using the AC-ABC process
- Selection of the gasifier location & operator for conducting pilot-scale tests

#### Pilot –Scale Testing

#### **FUTURE WORK**

- Design of a pilot-scale continuous, integrated test system
- Construction of the pilot-scale system
- Development of pilot-scale test plans
- Performance of pilot-scale tests
- Process modeling based on pilotscale tests
- Continued process economic analysis
- Technology transfer to commercial sector

### **Anticipated Benefits, if Successful**

- Demonstration of a technology under a real-world condition in the early stages of development at a minimum cost and time.
- Potential for capturing other trace acidic compounds such as HCl vapor.
- Rapid development to a commercial scale CO<sub>2</sub> capture technology that has the potential to achieve the goal at an affordable cost.

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#### **Thank You**

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