

## Recent technology development of KS-1 CO<sub>2</sub> recovery process

Takashi Kamijo, Nobuo Imai, Masaki Iijima,  
Toru Takashina, Hiroshi Tanaka,

Mitsubishi Heavy Industries, Ltd.

May, 2004



### Outline

- Background of KS-1 CO<sub>2</sub> recovery process
- Highlight operating results from the commercial plant
- Recent technology development
- Low energy regeneration system
- Comparison of improved KS-1 process with conventional technologies
- Summary



### ➤ Background of KS-1 CO<sub>2</sub> recovery process

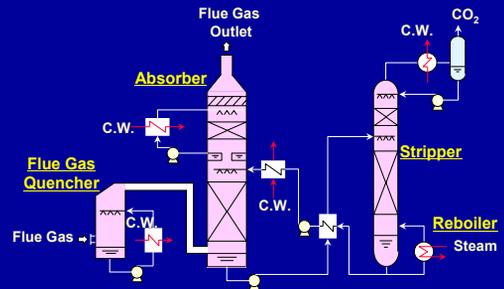
#### ■ Features - KS-1 Technology

- Mitsubishi Heavy Industries, Ltd (MHI) & Kansai Electric Power Company (KEPCO) 's proprietary Flue gas CO<sub>2</sub> capturing technology
- Hindered Amine Solvent "KS-1" with the special proprietary equipment
- Advantages
  - Low Energy Consumption
  - High CO<sub>2</sub> Loading
  - Low Corrosion
  - Low Degradation
  - Low Amine Loss



### ➤ Background of KS-1 CO<sub>2</sub> recovery process

#### ■ Typical Process Flow Diagram - KS-1 Technology



### ➤ Highlight operating results from the commercial plant (1/2)



•Client : Petronas Fertilizer (Keda) Sdn. Bhd  
•Location : Kedah Darul Aman, Malaysia  
•Flue Gas Source : Steam Reformer Flue Gas

Comp.	vol.%
N <sub>2</sub>	67.79
O <sub>2</sub>	0.85
H <sub>2</sub> O	22.28
CO <sub>2</sub>	8.08
Ar	1.00
SO <sub>x</sub>	0.85 ppmv
NO <sub>x</sub>	160 ppmv

•Capacity : CO<sub>2</sub> Recovery 210 Tons/day Max  
•CO<sub>2</sub> Recovery % : 90%  
•Solvent : KS-1 Solvent  
•Use of CO<sub>2</sub> : Urea Production  
•Start of Operation : October 1999  
•Project Scope : Turnkey Lumpsum



### ➤ Highlight operating results from the commercial plant (2/2)

#### ■ Operating Result

CO <sub>2</sub> Capturing	: Max. 210 Metric ton/day
CO <sub>2</sub> purity	: 99.95 % vol.dry
Utility requirement :	
Steam	1.5 ton/ton-CO <sub>2</sub>
Electricity	18 kWh/ton-CO <sub>2</sub>
Cooling water	159 m3/ton-CO <sub>2</sub>
Chemical requirement :	
Solvent	0.4-0.5 kg/ton-CO <sub>2</sub>



### Recent technology development

- CO<sub>2</sub> absorption & regeneration model
- Gas & Liquid flow simulation
- Design consideration for a large single train CO<sub>2</sub> recovery plant
- Less utility consumption by improving heat recovery
- Less solvent consumption by refining absorber
- Tower packing optimization for compact absorber
- CO<sub>2</sub> recovery from high sulfur coal firing flue gas



### Rigorous Model for CO<sub>2</sub> absorber & Stripper

- VLE : Bench Scale Test result of KS-1 solvent
- Mass & Heat transfer
- Assumption : Steady state  
Bulk Temperature = Interface Temperature  
No axial dispersion

#### Simulator

Input : L<sub>IN</sub>, G<sub>IN</sub>, T<sub>L,IN</sub>, T<sub>G,IN</sub>,  
G&L Inlet composition

Output : L<sub>OUT</sub>, G<sub>OUT</sub>, T<sub>L,OUT</sub>, T<sub>G,OUT</sub>,  
G&L Outlet composition



### Differential Equation

#### Material Balance

All components :  $-\frac{d}{dz}(\rho_L u_L) = mcMc + mwMw$        $\frac{d}{dz}(\rho_G u_G) = -mc - mw$

CO<sub>2</sub> :  $-\frac{d}{dz}(\rho_L y_C u_L) = mcMc$        $\frac{d}{dz}(\rho_G x_C u_G) = -mc$

H<sub>2</sub>O :  $-\frac{d}{dz}(\rho_L y_W u_L) = mwMw$        $\frac{d}{dz}(\rho_G x_W u_G) = -mw$

#### Heat Balance

- Gas Phase :  $\frac{d}{dz}(\rho_G C_G T_G u_G) = Q_R + Q_W + Q_{GL}$

- Liquid Phase :  $-\frac{d}{dz}(\rho_L C_L T_L u_L) = Q_{GL}$

#### Mass Transfer

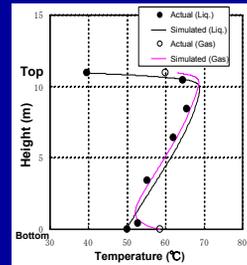
$$m_C = K_{GC} a (p_C - p_C^*)$$

K<sub>GC</sub>: Mass Transfer Coefficient

p<sub>C</sub><sup>\*</sup>: Vapor – Liquid Equilibrium

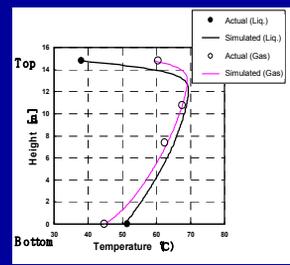
### Simulator results vs. actual operating data

Actual Simulation  
CO<sub>2</sub> inlet (vol.% dry) : 10.2 (Input)  
CO<sub>2</sub> outlet (vol.% dry) : 1.10 1.10



Nanko Pilot Test

Actual Simulation  
CO<sub>2</sub> inlet (vol.% dry) : 9.1 (Input)  
CO<sub>2</sub> outlet (vol.% dry) : 0.62 0.62



Malaysia Commercial Plant



### Design consideration for a large single train CO<sub>2</sub> recovery plant

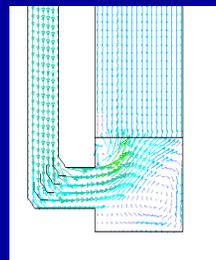
- CFD Technology
- Distribution Test by 1mx1m Rectangular Tower
- Determine operating condition with effective Gas/Liquid contact

#### Result

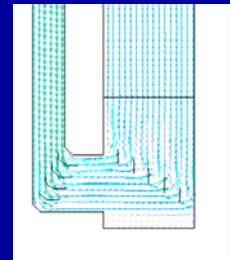
Large scale CO<sub>2</sub> Absorber for 5,000-10,000 Metric ton/day capacity can be realized.



### CFD results for Large scale CO<sub>2</sub> Rectangular Absorber



Without Absorber inlet guide vane



With Absorber inlet guide vane

Side View of Absorber



### Rectangular Tower Test in MHI R&D center

**Test facility**

**<Test condition>**

Fluid	water
Temperature	30°C
Gas velocity	1.4~5m/s
Liquid velocity	11~17m <sup>3</sup> /m <sup>2</sup> ·h

**Fig. Outside view**

**Fig. Appearance of liquid flow below distributor (Liquid flow rate 17m<sup>3</sup>/m<sup>2</sup>·h)**

MITSUBISHI HEAVY INDUSTRIES, LTD.

### Determine operating condition with effective Gas/Liquid contact

**CO<sub>2</sub> absorption rate  $\eta = \int \eta(A) dA$**

**Gas distribution**

**Liquid distribution**

MITSUBISHI HEAVY INDUSTRIES, LTD.

### Less Utility Consumption

- Less steam and cooling water consumption (15% steam reduction)
- Lower Flue Gas Blower Power by Structured Packing
- Define Optimum Operating Condition by rigorous simulator
  - Lean & Rich loading
  - Packing selection & height
  - Solvent circulation rate
- Integrated heat recovery system with the process of flue gas source or CO<sub>2</sub> compression system, etc.

MITSUBISHI HEAVY INDUSTRIES, LTD.

### Less solvent consumption by refining absorber

- Installing proprietary device in top of CO<sub>2</sub> Absorber to wash and remove amine mist

	Before	After
Solvent loss	0.45 kg/T-CO <sub>2</sub>	→ 0.35 kg/T-CO <sub>2</sub>
Condition	Flue gas CO <sub>2</sub> 8 % dry SO <sub>x</sub> 1.5 ppmv NO <sub>x</sub> 140 ppmv	

MITSUBISHI HEAVY INDUSTRIES, LTD.

### Tower Packing Optimization

CMR packing vs. Structured packing

**Pressure drop**

**Gas-Liquid effective contact area**

MITSUBISHI HEAVY INDUSTRIES, LTD.

### Relationship between Gas/Liquid Effective contact Area and Flow Parameter

Result of Pilot Plant Operating data  
Packing: Structured Packing (200 m<sup>2</sup>/m<sup>3</sup>)

Condition:  
Temp. 45 degC  
Press. Atm.  
Nanko Pilot Plant Absorber

Flow Parameter,  
Flooding Approach =  $L/G \cdot (\text{Density}_G / \text{Density}_L)^{0.5}$

MITSUBISHI HEAVY INDUSTRIES, LTD.

## ■ Low energy regeneration system

- Improve heat recovery around CO<sub>2</sub> stripper
  - Utilize heat of Lean solvent & Steam Condensate from Stripper Reboiler
  - Hot Lean Solvent and Steam Condensate heat up intermediary stage of CO<sub>2</sub> stripper
  - Addition several Heat exchanger & Pumps compared with base flow scheme.



## ■ Comparison of improved KS-1 process with conventional technologies

### ➤ Case study result for 3,000 Metric ton/day CO<sub>2</sub> recovery plant

	Conventional	Advanced
Solvent	KS-1	→
Process Flow	Base	Improved heat recovery around CO <sub>2</sub> stripper
Utility consumption		
Steam	Base	Base x 0.86
Electricity	Base	Base x 1.10
Cooling Water	Base	Base x 0.91
Chemical consumption		
Solvent loss	0.45 kg/T-CO <sub>2</sub>	0.35 kg/T-CO <sub>2</sub>
Condition	Flue gas, CO <sub>2</sub> 8 % dry	→
	SOx 1.5 ppmv	
	NOx 140 ppmv	



## ➤ Summary

- KS-1 technology revealing best performance in Malaysia commercial plant among any other amine based flue gas CO<sub>2</sub> recovery processes
- MHI's recent technology development results in better performance than conventional KS-1 process
- Large Single train CO<sub>2</sub> recovery Plant capacity of 5,000-10,000 Metric ton/day can be realized
- Solvent loss is reduced further by installing proprietary device in top of CO<sub>2</sub> Absorber
- Structured Packing enable compact Tower and utility reduction
- Improved heat recovery around CO<sub>2</sub> stripper results in 14 % reduction of steam requirements

