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# Development Novel Carbon Sorbents for Carbon Dioxide Capture

2010 NETL CO<sub>2</sub> Capture Technology Meeting  
September 13-17, 2010 in Pittsburgh, PA.

# Project Overview

- Participants:
  - SRI International, Menlo Park, CA
  - ATMI, Inc., Danbury, CT
  - DOE-National Energy Technology Center
- Period of Performance:
  - 10-1-2008 through 9-30-2011
- Funding:
  - U.S.: Department of Energy: \$1.35 million
  - Cost share: \$0.45 million
  - Total: \$1.8 million

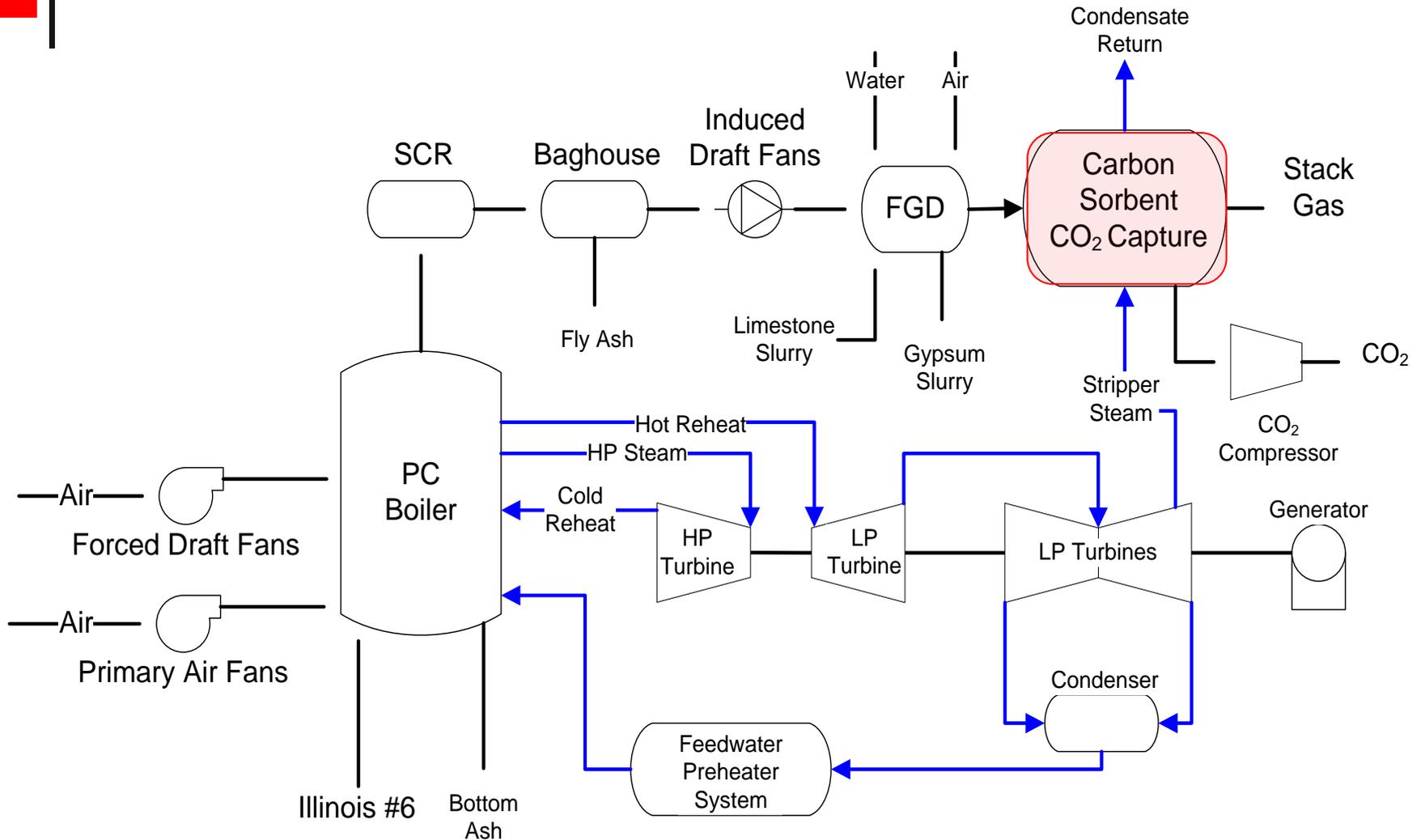
# Basic Principles

- Adsorption of CO<sub>2</sub> from flue gas on a selective and high capacity carbon sorbent.
- Ability to achieve rapid adsorption and desorption rates (no solid state diffusion limit).
- Minimize thermal energy requirements
  - Relatively low desorption temperature.
  - Low heat of desorption.
- Ability to desorb as pure CO<sub>2</sub>.

# Project Objectives

- Validate the performance of novel carbon sorbents for CO<sub>2</sub> capture on a bench-scale system for post-combustion applications.
- Perform parametric experiments to determine the optimum operating conditions.
- Evaluate the technical and economic viability of the technology.
- Pilot-scale testing in a future phase

# Block Flow Diagram



# Advantages of the Carbon Sorbents

- Low cost and stable sorbent: In the operating range of 20° to 100°C, the carbon sorbent is very stable.
- Relatively high CO<sub>2</sub> loading (0.1 to 0.2 kg of CO<sub>2</sub> per kg of sorbent in pure CO<sub>2</sub> at 1 atm).
- Low heat of absorption reaction
  - 25 to 28 kJ/mole CO<sub>2</sub>
- Low heat requirements for regeneration: Sorbent can be regenerated to release CO<sub>2</sub> at atmospheric pressure in a temperature range of 80° to 100°C.

# Technical Challenges

- Competitive adsorption of moisture and other flue gas components (SO<sub>x</sub>, and NO<sub>x</sub>).
- Adsorption temperature
  - Low temperature promotes adsorption
- Suitable reactor configuration for rapid cycling of the sorbent.
- Heat exchange between loaded (cold) and regenerated (hot) sorbents.

# Project Tasks

- Determination of the relevant properties of the sorbent.
  - Surface area, heat of adsorption and desorption, compressive strength and attrition resistance, size and shape of the sorbent particles.
- Improvements to the properties of the sorbent.
  - Structural modification of the pore structure
  - Functionalizing the surface
- Bench-scale parametric testing of the sorbent for adsorption and regeneration:
  - Screening tests
  - Parametric tests
  - Long-term tests
- Process technical and economic analysis.

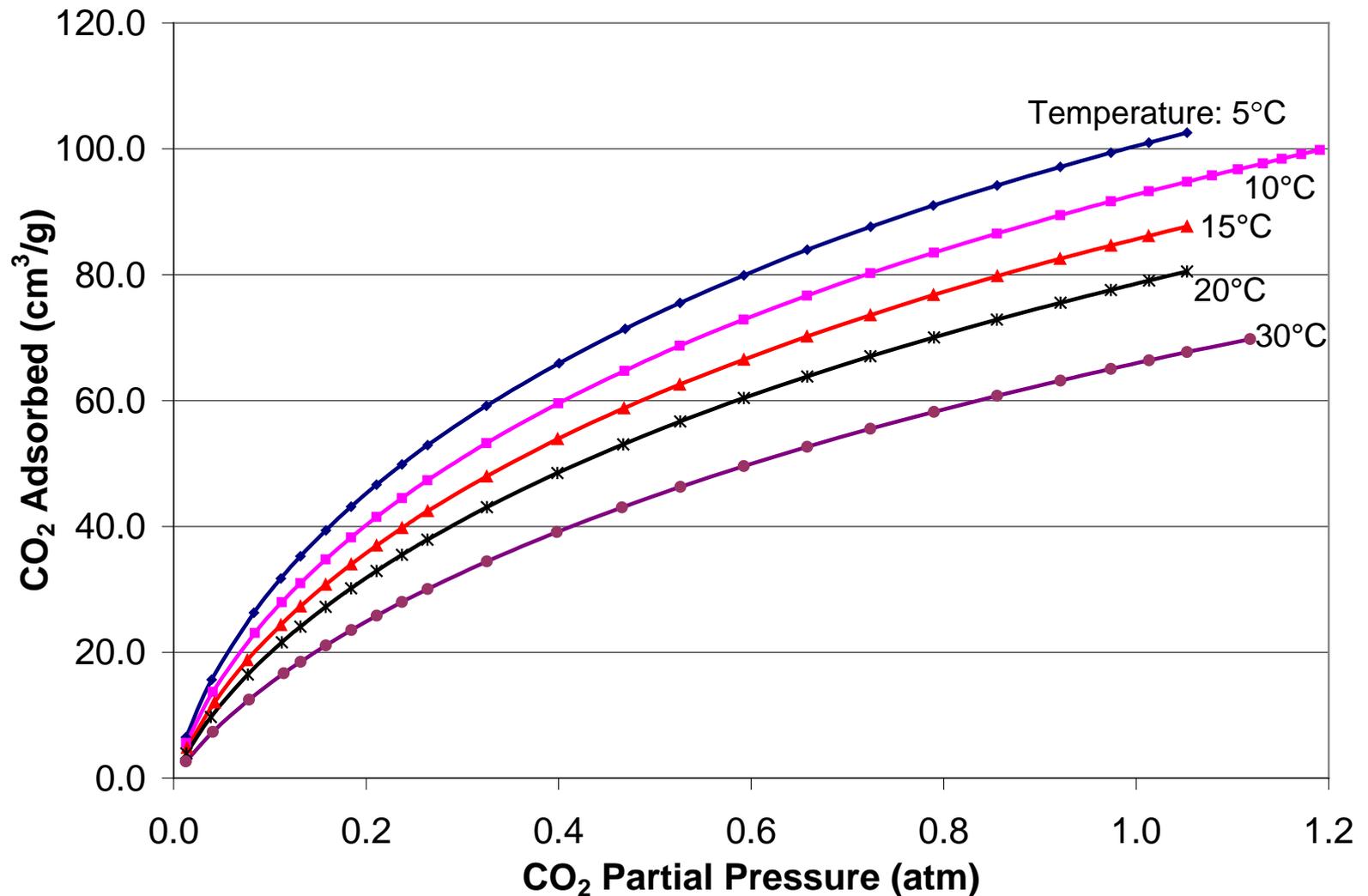
# Determination of the Relevant Properties of the Sorbent

- Adsorption/desorption isotherms
  - Gives capacity of sorbent as a function of temperature and CO<sub>2</sub> pressure
- Heat of adsorption and desorption
  - Gives cooling requirements during adsorption and heating requirements during regeneration
  - Provides a guide to improving sorbent for CO<sub>2</sub> selectivity
- Compression strength and attrition resistance, particle size and shape
  - Determines suitability for use in industrially-relevant reactors

# Physical Properties of the Sorbent

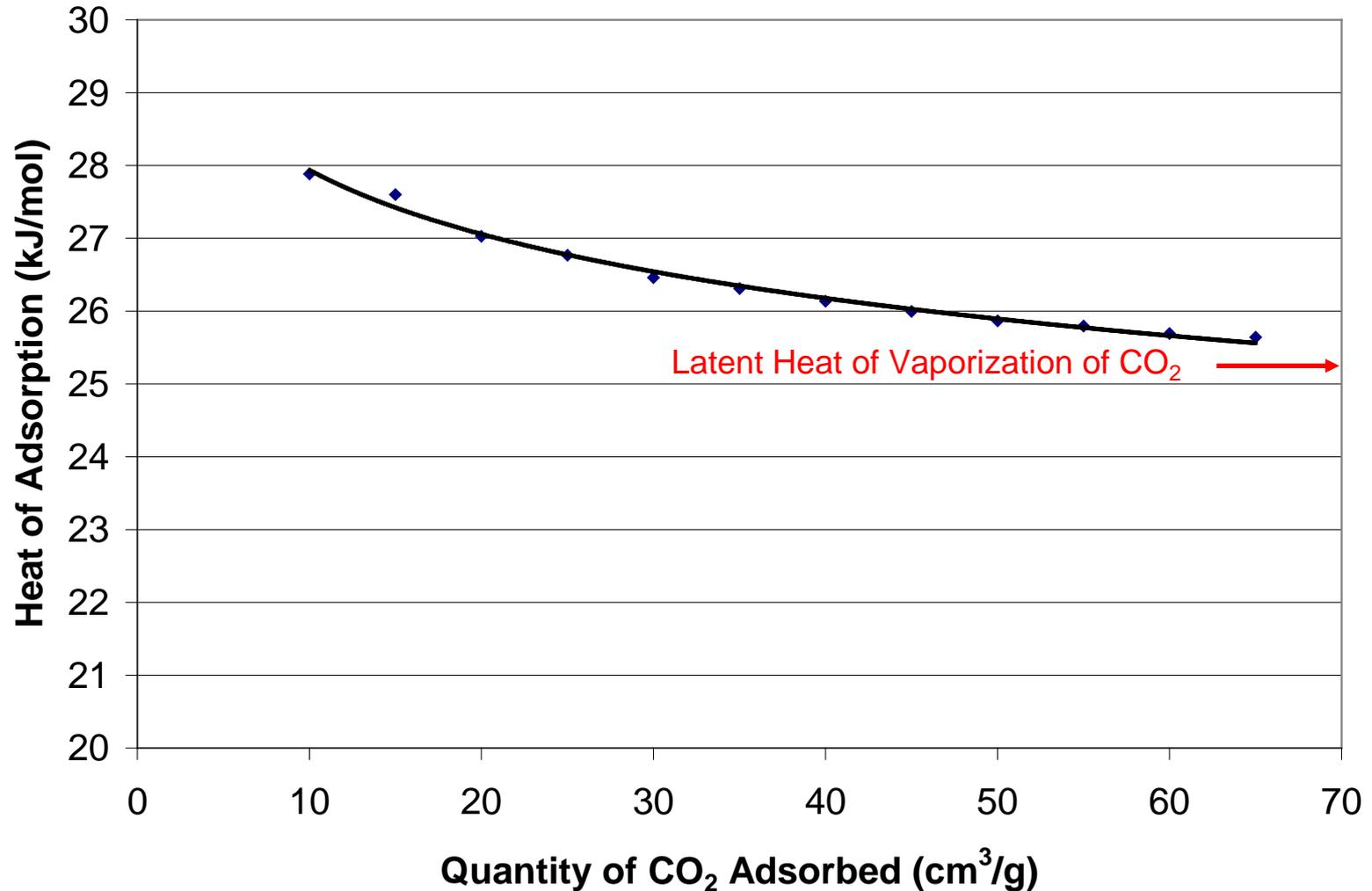
- A carbon sorbent currently manufactured by ATMI, Inc.
- High surface area (1300 m<sup>2</sup>/g).
- Low heat capacity (1 J/g-°K).
  - 25% of the heat capacity of water.
- High thermal conductivity (0.82 W/m-°K)
  - 5 times higher than typical porous catalysts.
- Low density (1.1 kg/liter).
- Unusually tough for a high surface area porous solid.

# CO<sub>2</sub> Adsorption Isotherms



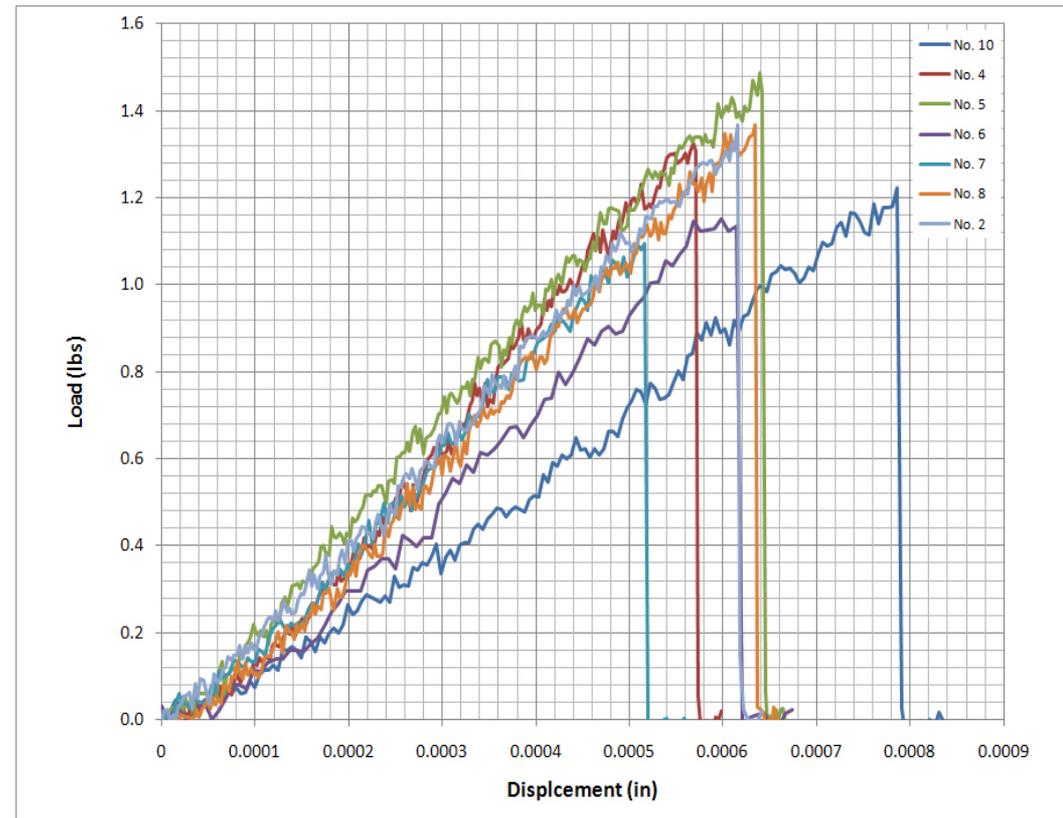
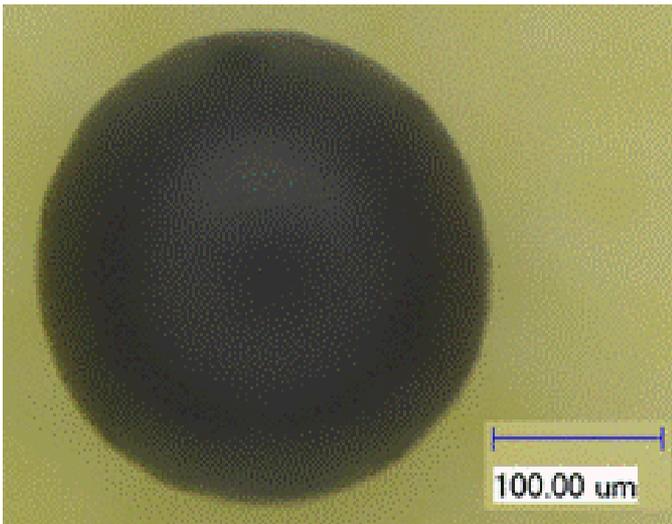
100 cm<sup>3</sup>/g = 20 wt% CO<sub>2</sub>

# Heat of Adsorption for CO<sub>2</sub>



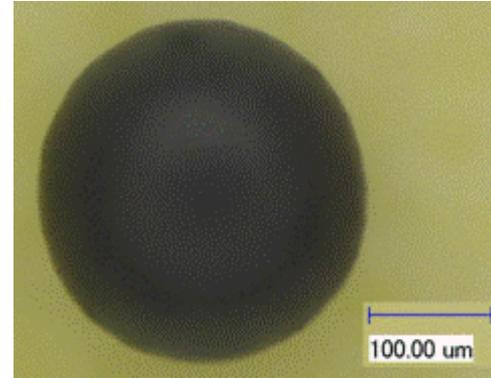
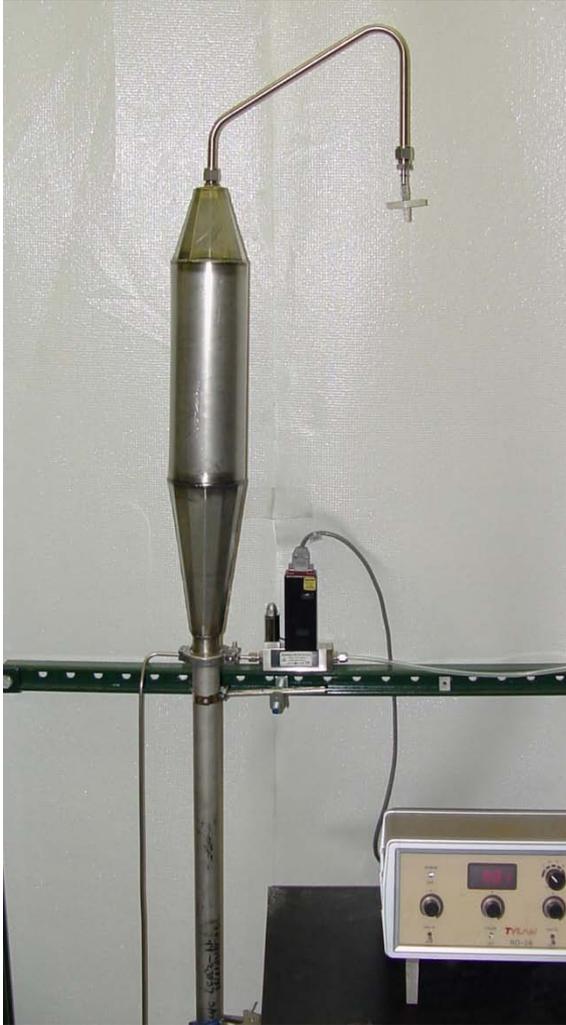
Note: Measured heat of desorption = 27 kJ/mole

# Morphology and Strength



12,000 to 18,000 psi compressive strength

# Attrition Resistance



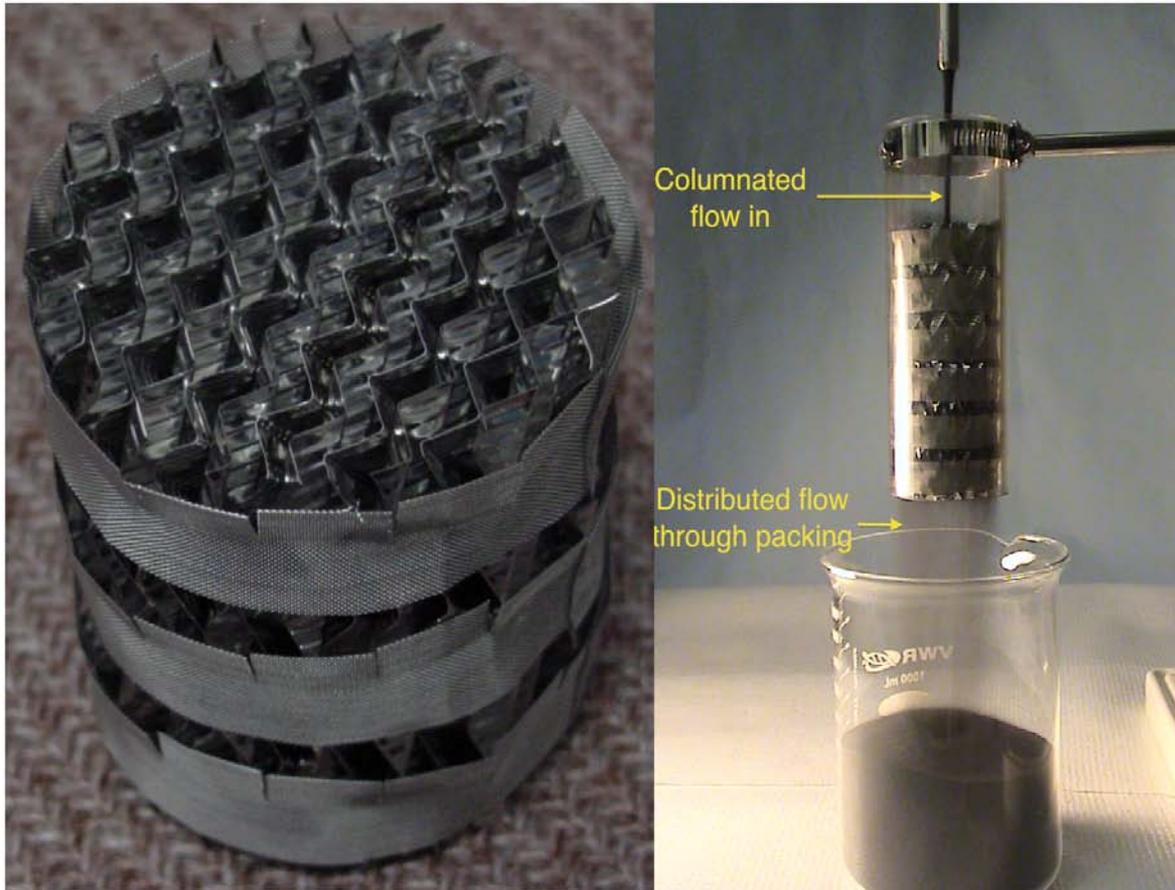
Photograph of a Sorbent Granule

ASTM Test D-5757: Accelerated attrition test using a high velocity gas jet.

Attrition Resistance: Very high;

Weight loss: <0.01%/Hour even at accelerated test conditions

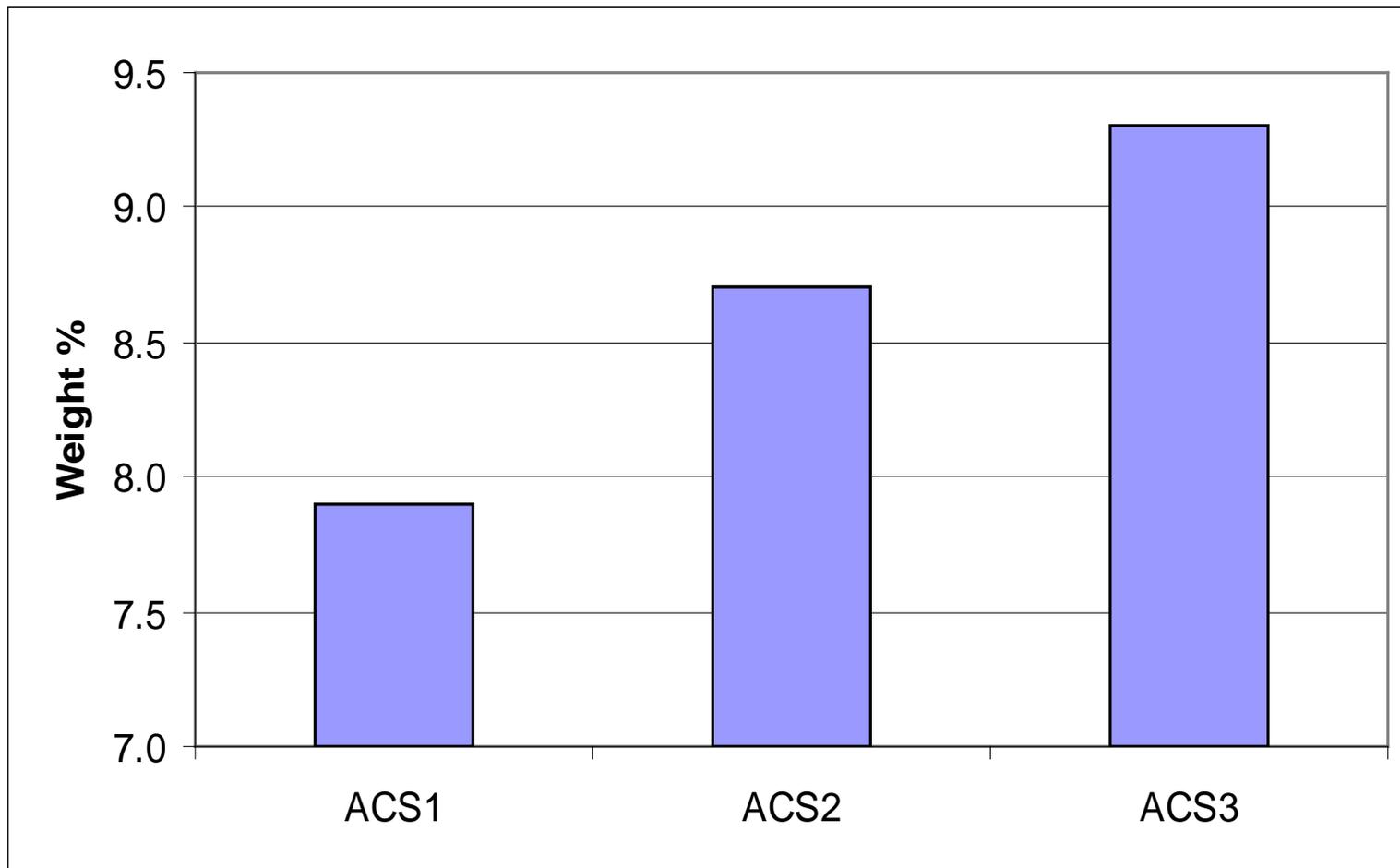
# Fluid-like Flow of Sorbents Through a Commercial Structural Packing



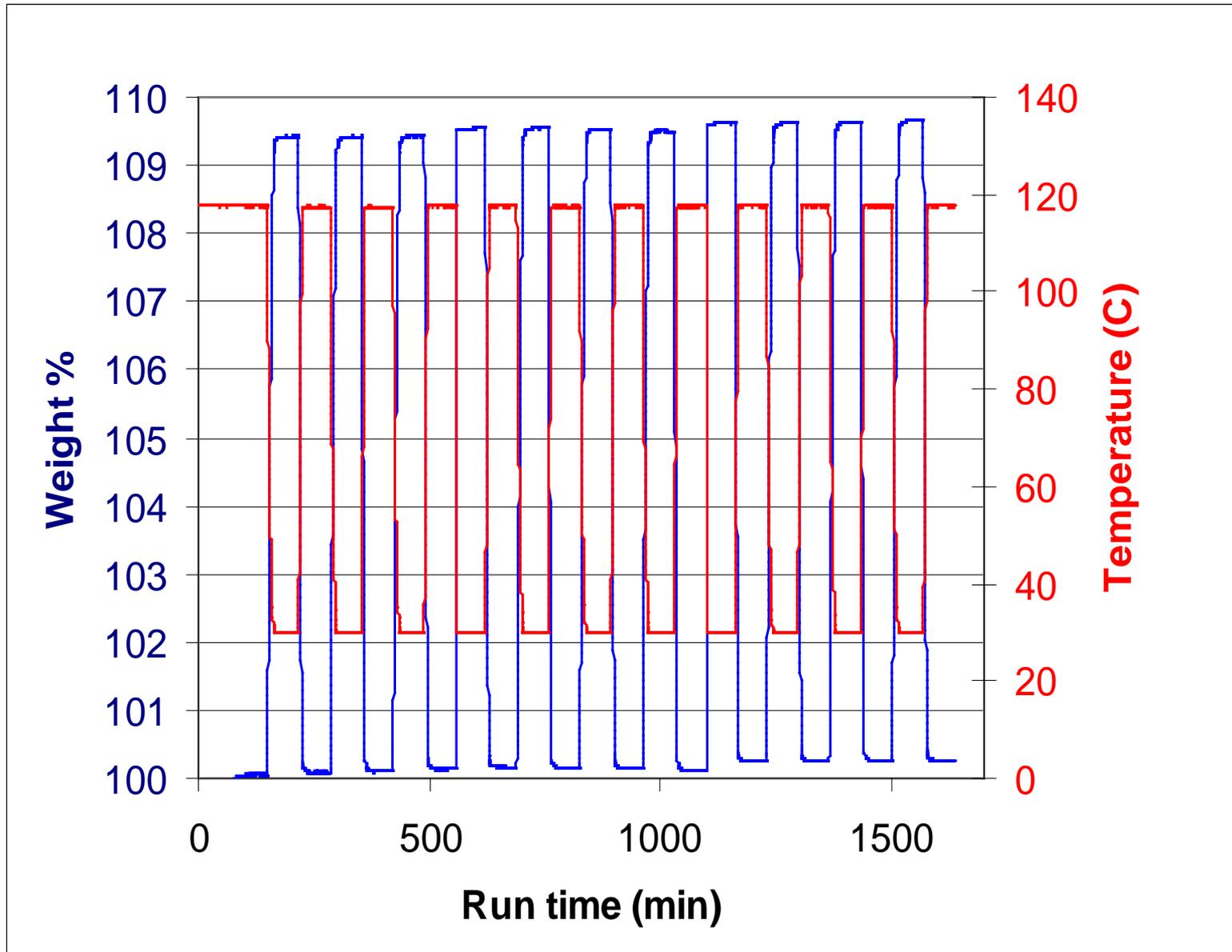
# Screening Tests

- Determine the CO<sub>2</sub> capture rate of the current and improved sorbents in a small bench-scale reactor
  - At 20 C using a simulated flue gas containing air and CO<sub>2</sub>.
  - Determine the adsorption kinetics and the CO<sub>2</sub> loading
- Heat to 110 C to desorb the CO<sub>2</sub>:
  - Determine the desorption kinetics and the CO<sub>2</sub> desorbed.
- Adsorption-regeneration cycles performed.
- No significant changes in the physical and mechanical properties of the aged sorbent.

# Differential CO<sub>2</sub> loading Between 30° and 110°C



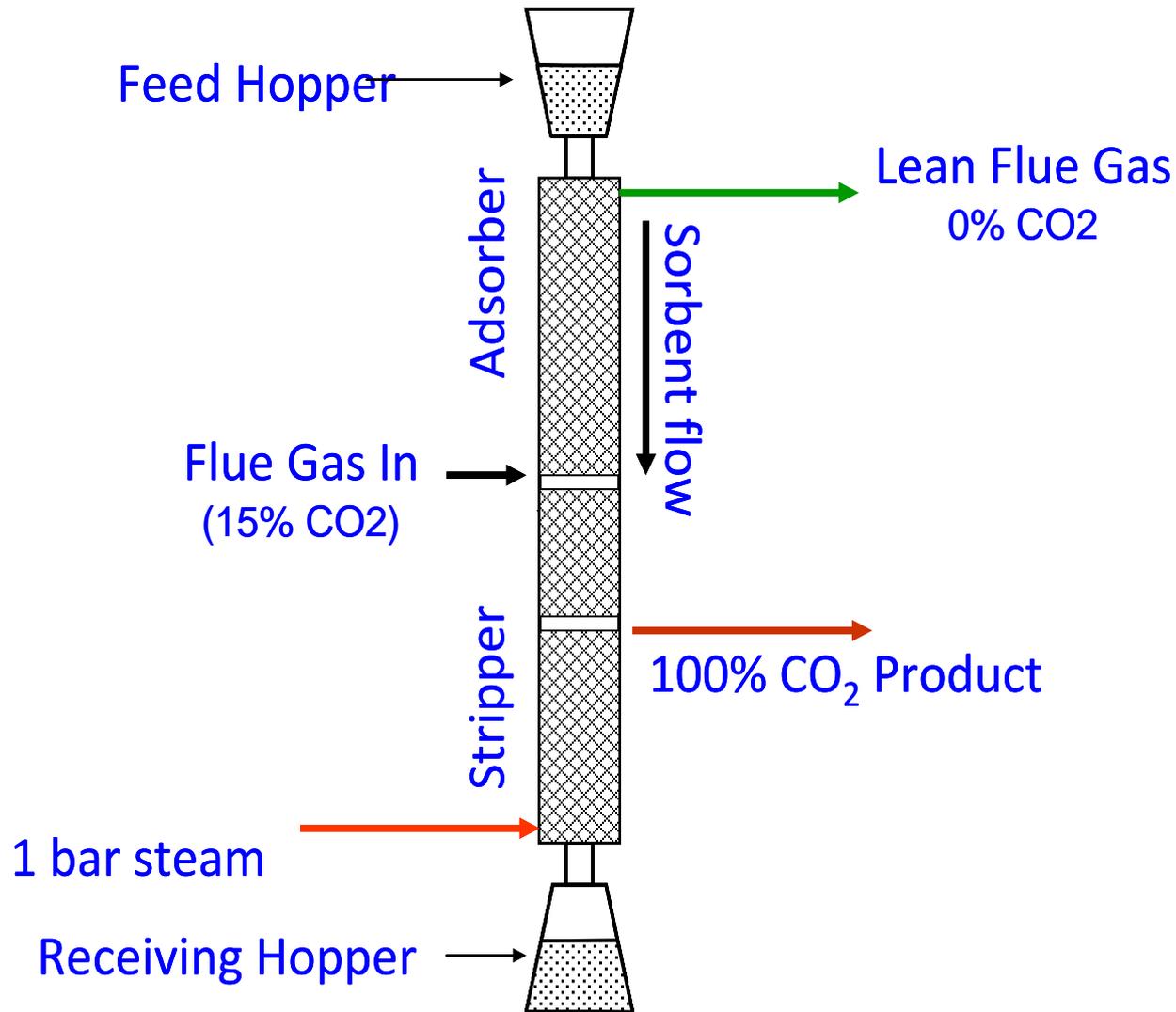
# Absorption-Desorption Cycle



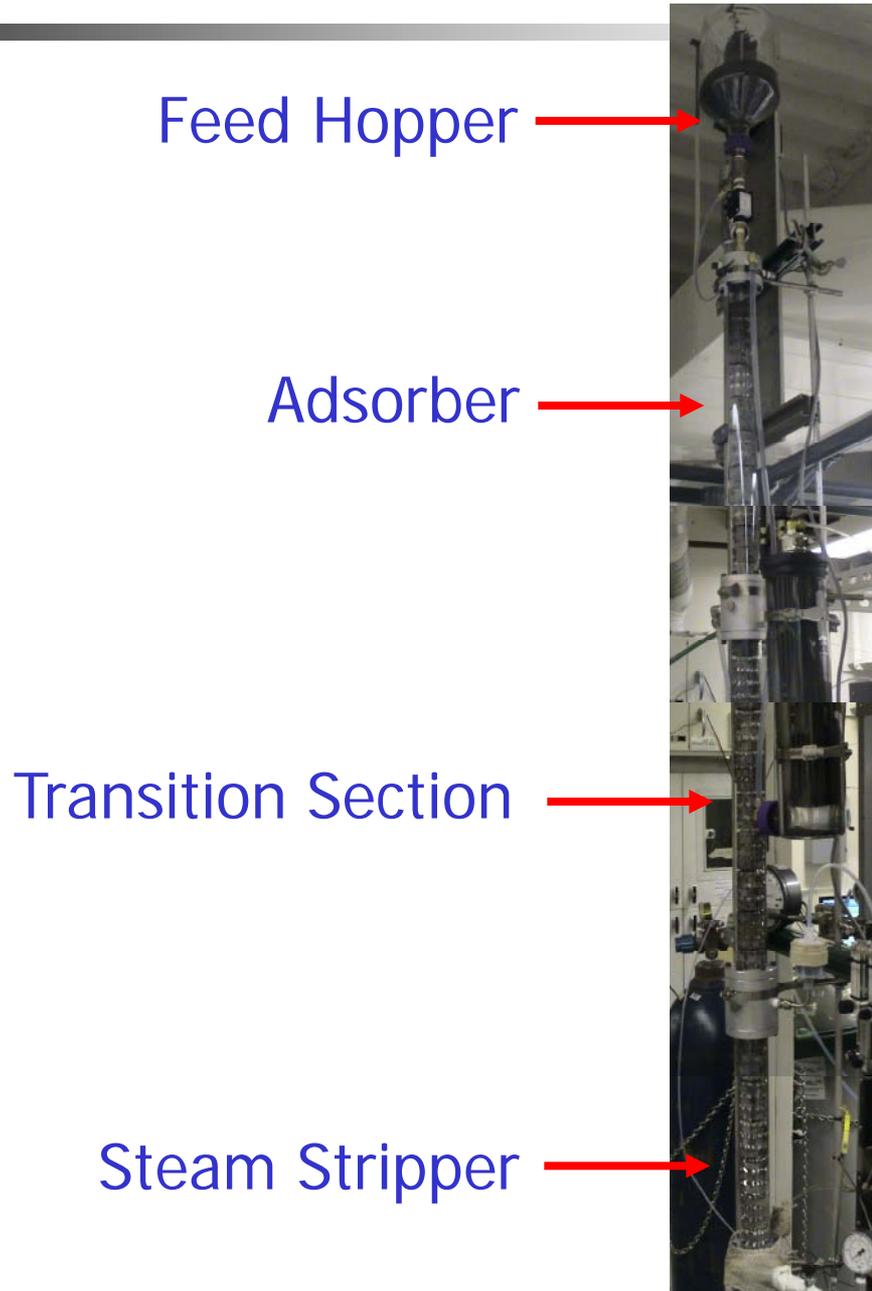
# Reactor Geometry

- Reactors for Gas-solid interactions are:
  - Fixed-bed reactors
  - Moving-bed reactors
  - Fluidized- or entrained-bed reactors
- We chose a moving-bed reactor type:
  - Uses a commercial structural packing to distribute the solid
  - Pressure drop for gas flow can be minimum
    - Low operating cost.
  - The sorbent can be cycled rapidly between CO<sub>2</sub> capture and release.
    - Low sorbent inventory
  - Reactor volume can be substantially small
    - Low capital cost.

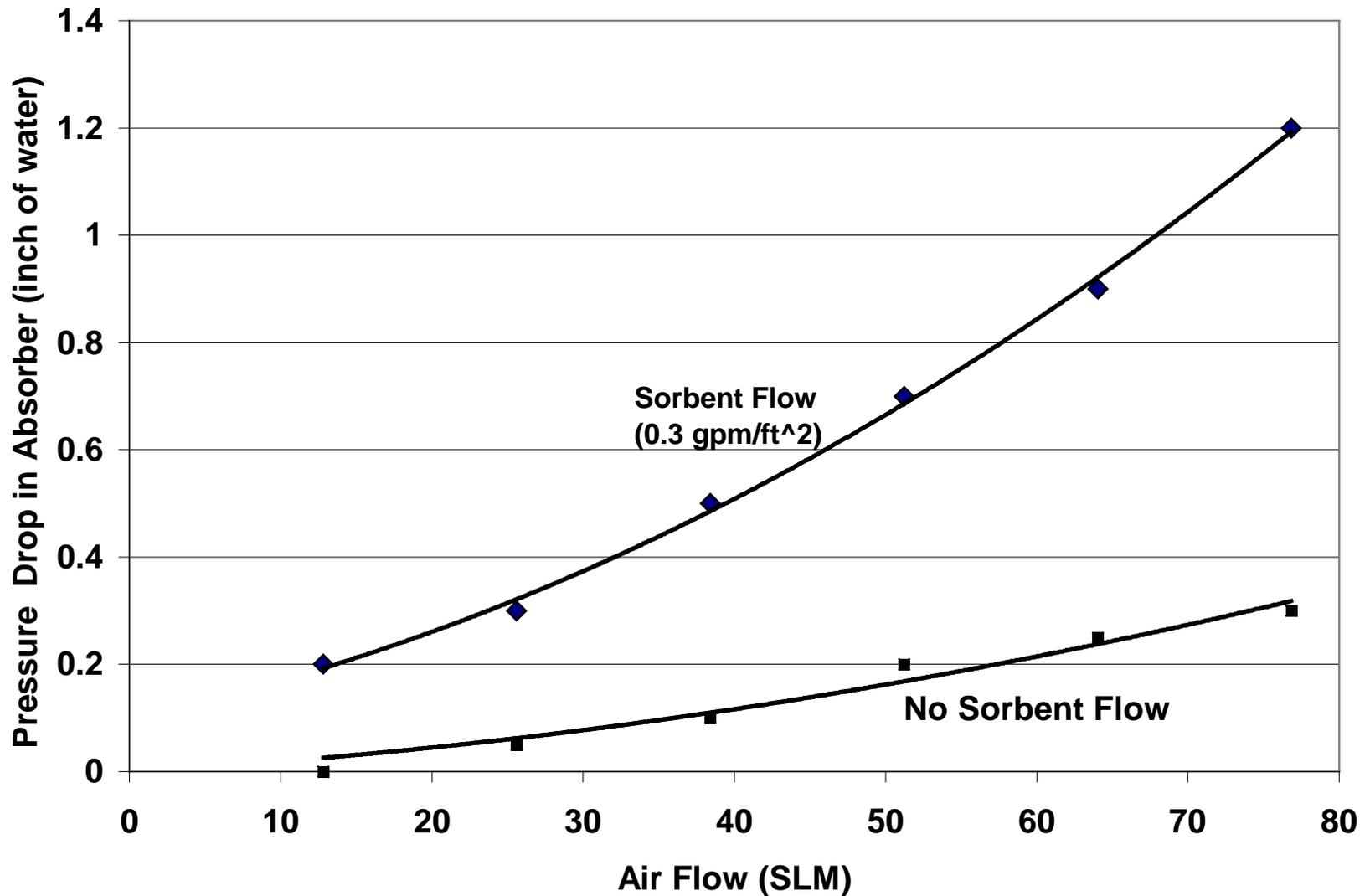
# Schematic Diagram of a Novel, Particle Cascading Reactor



# Integrated Absorber-Desorber (Small Bench-Scale Design)

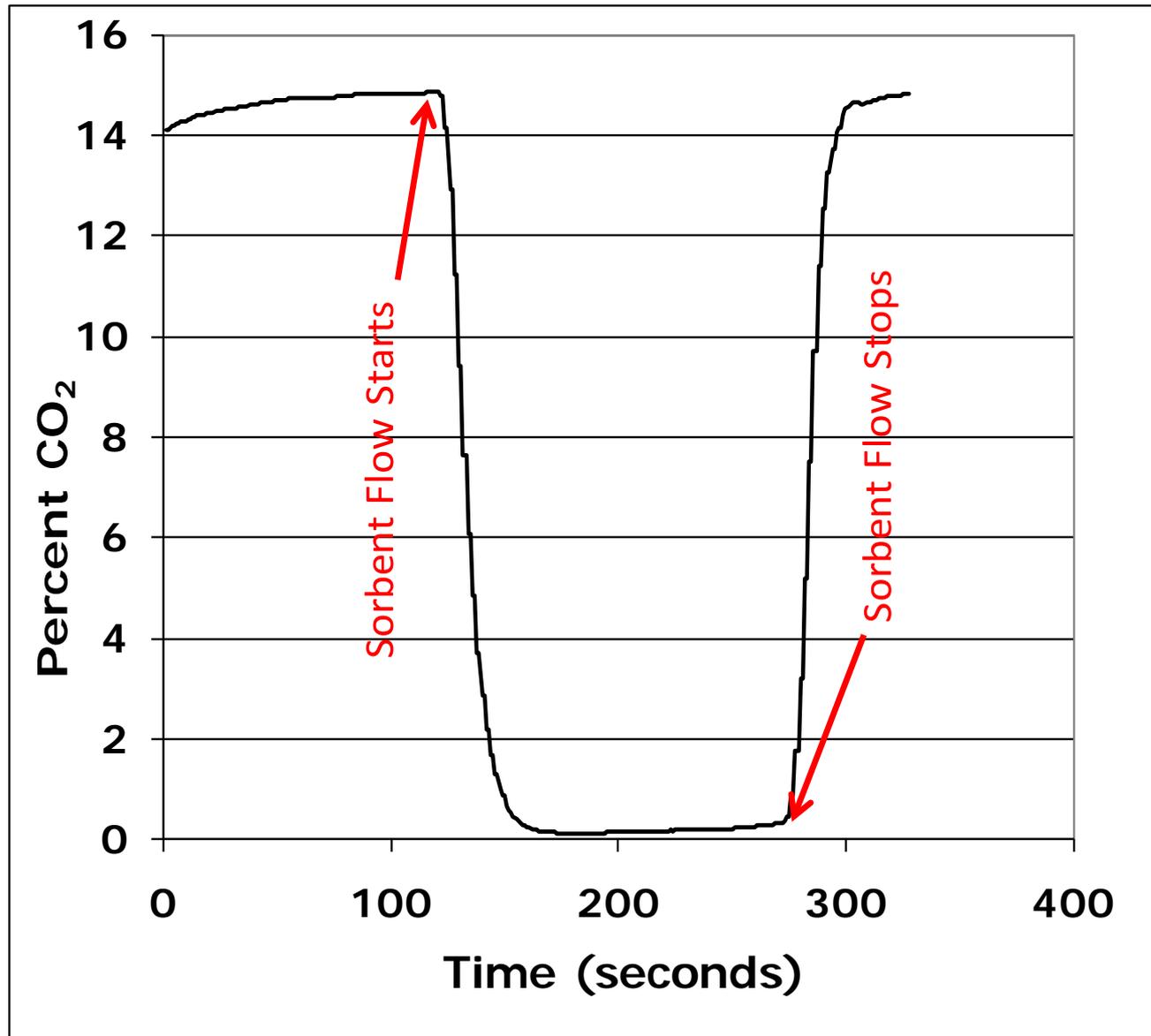


# Pressure Drop in the Absorber

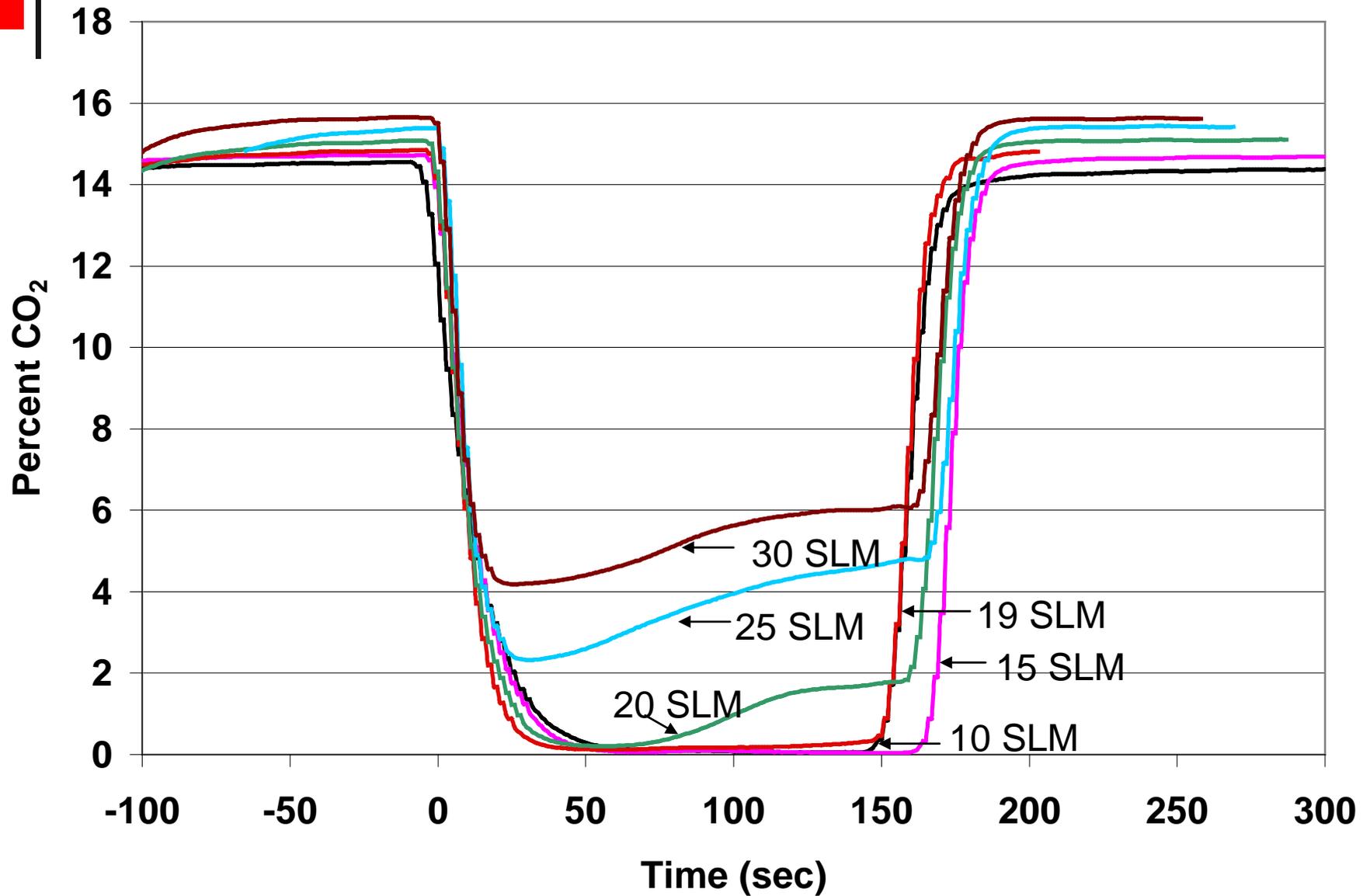


Flow rate of 100 Std.liters/min = 0.91 m/s

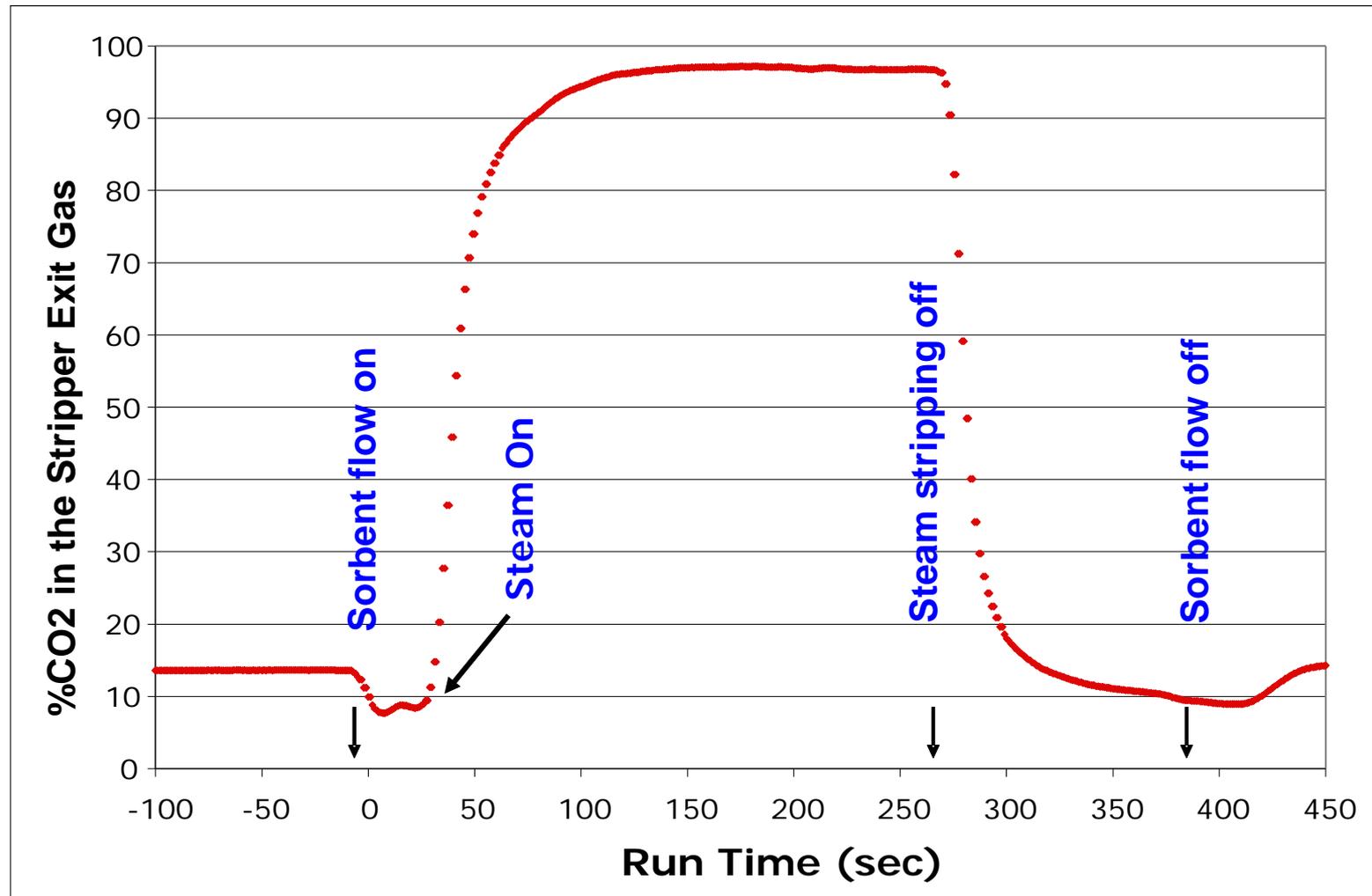
# Removal of CO<sub>2</sub> from Air-CO<sub>2</sub> Mixture



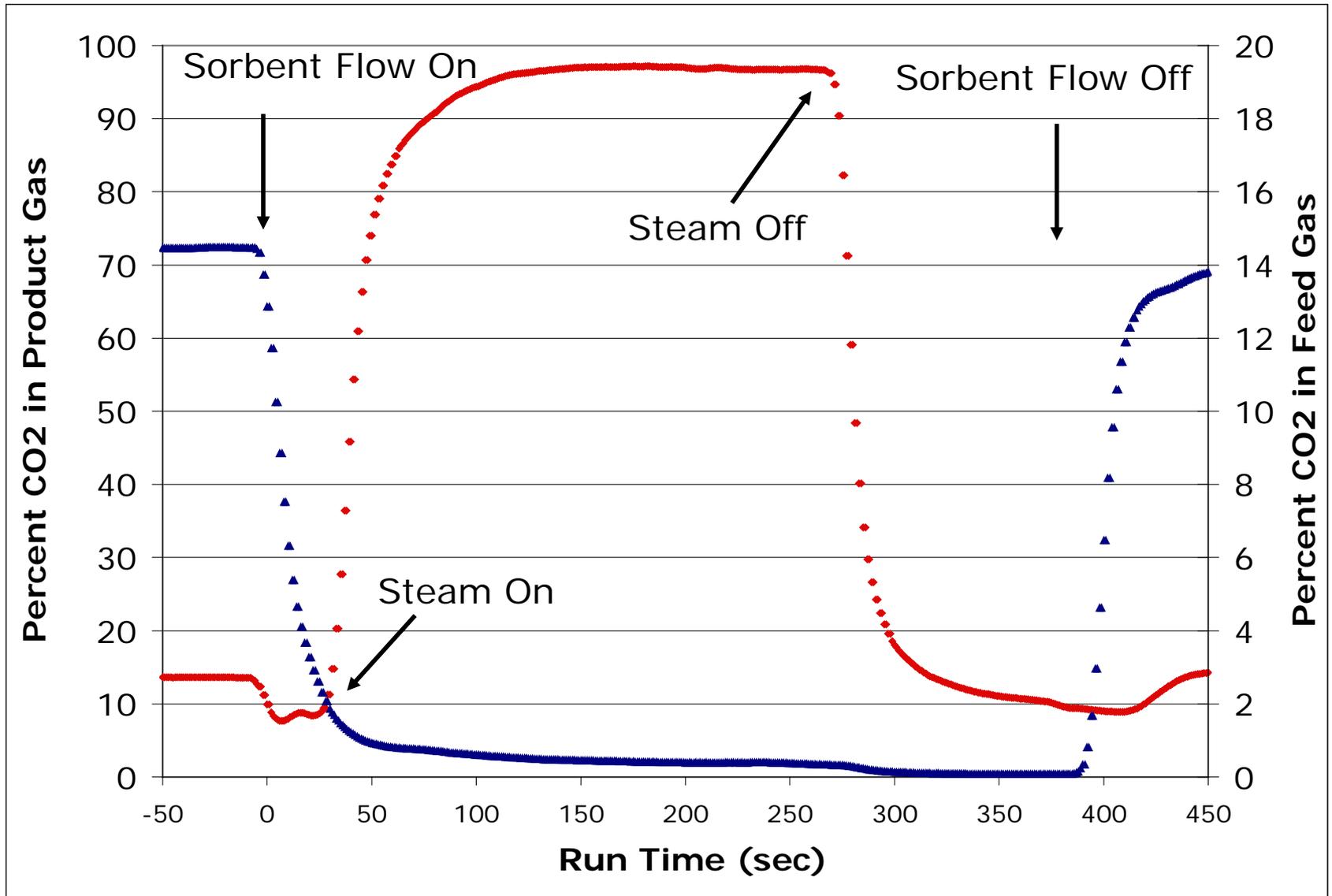
# Residual CO<sub>2</sub> concentration as a Function of Gas Flow Rate



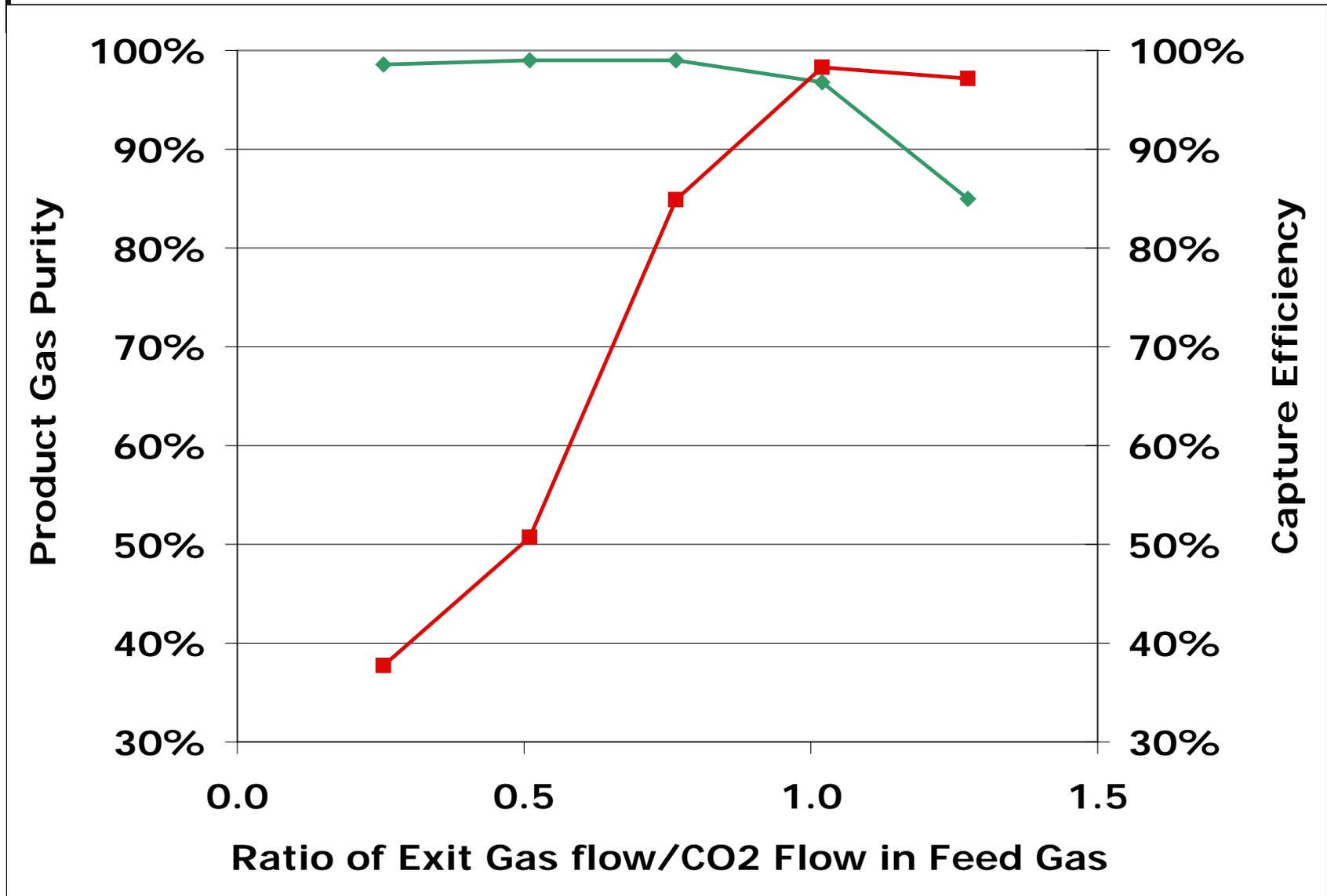
# Evolution of CO<sub>2</sub> in the Stripper



# Integrated Operation



# Capture Efficiency vs Product Gas Purity



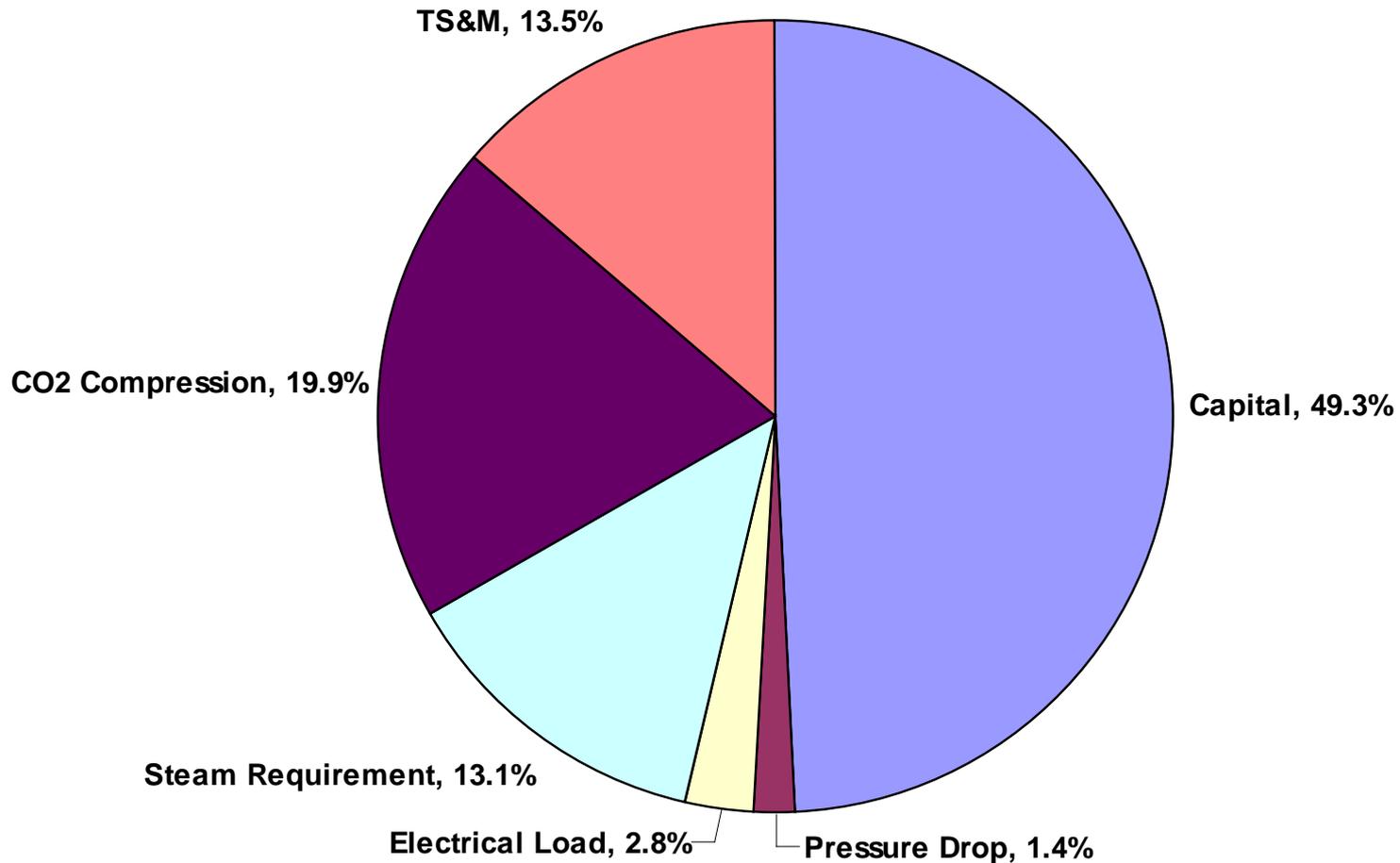
# Technical and Economic Analysis

- Steam-Pro model was used to generate the heat, material, and energy flows.
- Use DOE cost model was used to estimate the cost of electricity (COE).
- Base case is an air-fired greenfield supercritical PC plant (700 mW nominal) with no CO<sub>2</sub> capture.
- We compared a similar-size plant using CO<sub>2</sub> capture with carbon sorbent subsystem.

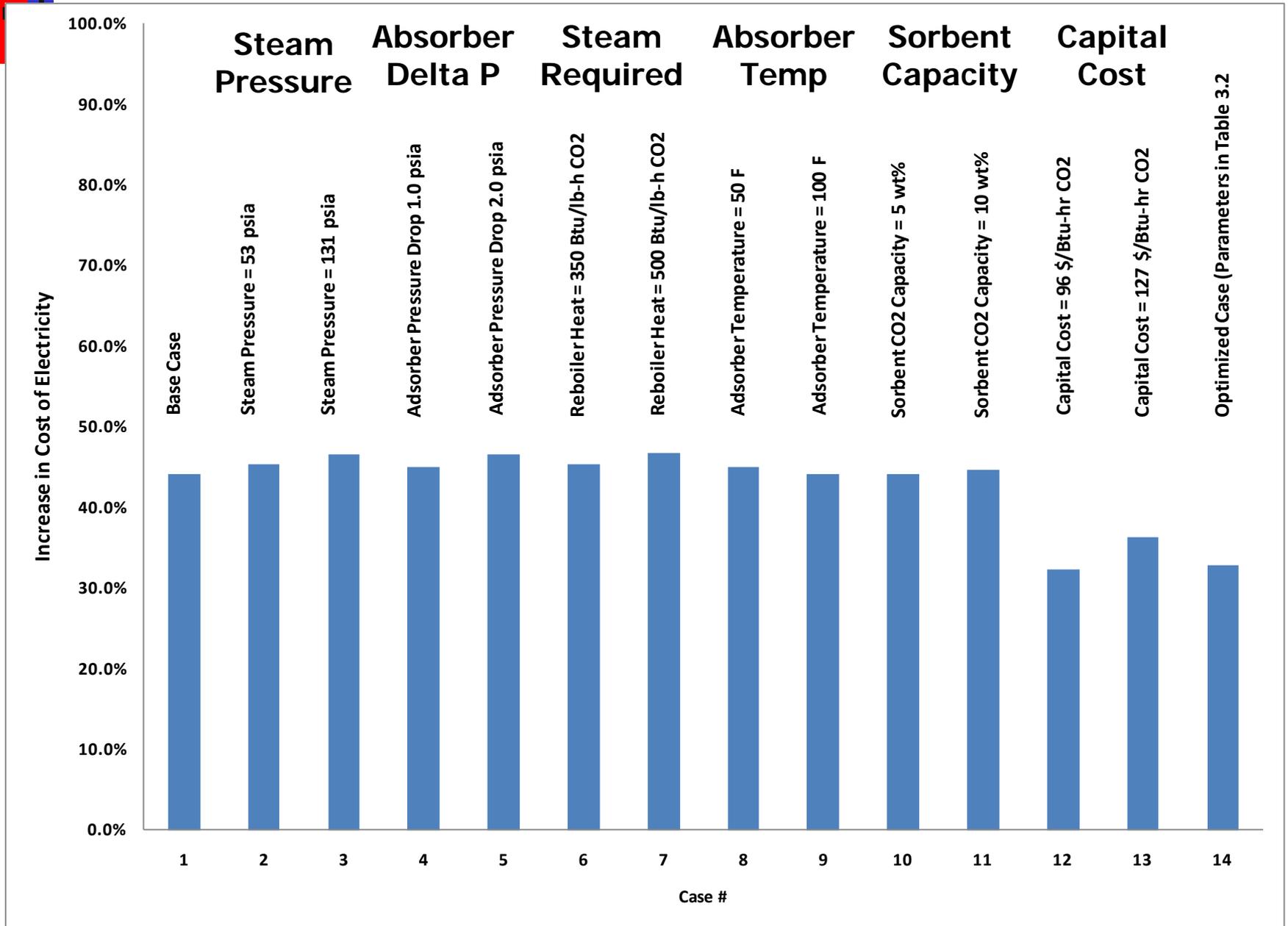
# Comparison of CO<sub>2</sub> Capture Costs

	<b>Base Case</b>	<b>Econamine FG+</b>	<b>Carbon Sorbent</b>
<b>CO<sub>2</sub> Capture</b>	<b>No</b>	<b>Yes</b>	<b>Yes</b>
Gross Power Output (kW)	581,034	679,911	640,421
Auxiliary Power Requirement (kW)	31,016	129,485	90,246
<b>Net Power Output (kW)</b>	<b>550,018</b>	<b>550,426</b>	<b>550,175</b>
<b>Net Plant HHV Efficiency (%)</b>	<b>38.9%</b>	<b>27.1%</b>	<b>35.1%</b>
Net Plant HHV Heat Rate (Btu/kW-hr)	8,859	12,590	9,717
Coal Flowrate (lb/hr)	414,000	594,000	458,280
CO <sub>2</sub> Emissions (lb/MWh)	1,790	252	194
Total Plant Cost (\$ x 1000)	872,118	1,586,765	1,224,213
Total Plant Cost (\$/kW)	1,586	2,883	2,225
<b>LCOE (¢/kWh)</b>	<b>6.40</b>	<b>11.58</b>	<b>9.23</b>
<b>Increase in COE (%)</b>	<b>0.0%</b>	<b>80.9%</b>	<b>44.2%</b>

# Break-Down of the CO<sub>2</sub> Capture Cost Using Advanced Carbon Sorbent (Initial)



# Increase in COE for Different Cases

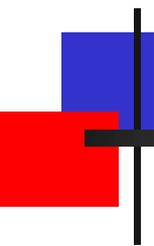


# Achievements

- Determined several physical and chemical properties of the advanced carbon sorbent in the context of flue gas CO<sub>2</sub> capture.
- Demonstrated an unique sorbent for CO<sub>2</sub> capture
  - Achieved ~99% CO<sub>2</sub> capture from air-CO<sub>2</sub> gas mixture
  - Achieved >98% pure CO<sub>2</sub> during regeneration
  - Capable of rapid adsorption and regeneration
  - Low heat requirements for regeneration
  - Fluid-like flow properties
  - High attrition resistance
- Developed an unique reactor system
  - Integrated absorber-desorber geometry
  - Minimize solids handling
  - Minimize heat exchanger requirements

# Future Plans

- Plans for next year:
  - 1000-cycle test under integrated adsorption-desorption conditions
  - Evaluation of the technical merits and increase in cost of electricity for CO<sub>2</sub> capture using the novel carbon sorbents
- Plans for the next phase:
  - Field testing of the process using a flue gas from an operating PC-fired boiler.



# Team

- SRI International
  - Dr. Gopala Krishnan – Associate Director (MRL) and PI
  - Dr. Angel Sanjurjo – Materials Research Laboratory Director and Project Supervisor
  - Dr. Marc Hornbostel, Senior Materials Scientist
- ATMI Inc.
  - Sorbent developer, Industry perspective
  - Dr. Joshua B. Sweeney, PhD; Director, Business Development
  - Dr. Lawrence H. Dubois Ph.D; Senior Vice President and Chief Technology Officer
  - Dr. Donald Carruthers; Senior Research Scientist
- DOE-NETL
  - Mr. Andrew O'Palko, Project Manager