

# Intermediate Temperature Proton Conducting Fuel Cells for Transportation Applications

## ARPA-E Project (2012 Open)

Award No. DE-AR0000314

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Completed Q10

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# Project Team

Team Member	Project Role
<p>Ceramatec, Inc.</p> <p>Location: Salt Lake City, UT</p> <p>Focus: Ion conducting ceramics Electrochemistry Advanced Materials</p>	<p>Prime</p> <p>Materials Scale up Stack Testing</p>
<p>Los Alamos National Lab.</p> <p>Location: Los Alamos, NM</p>	<p>National Lab Partner</p> <p>Materials Development, Synthesis, &amp; Characterization Cell Testing</p>
<p>Nissan Technical Center North America</p> <p>Location: Farmington Hills, MI</p>	<p>Commercialization Partner</p> <p>Cell validation System Modeling Requirement Definition</p>

# Project Objectives

- Develop a proton conducting fuel cell based on Tin Pyrophosphate (TPP) that operates at 200 – 250 ° C
  - ❖ **Mid-Temp and Low RH will simplify the Balance of Plant in the system.**
  - ❖ **This simplification will reduce significant portion of the Balance of Plant cost.**

# Project Target

- Fuel Cell Testing using thin, composite membrane
  - Demonstration of 25 to 50 cm<sup>2</sup> fuel cell
  - 500 mW/cm<sup>2</sup> at 200° - 250° C, relative humidity < 5%

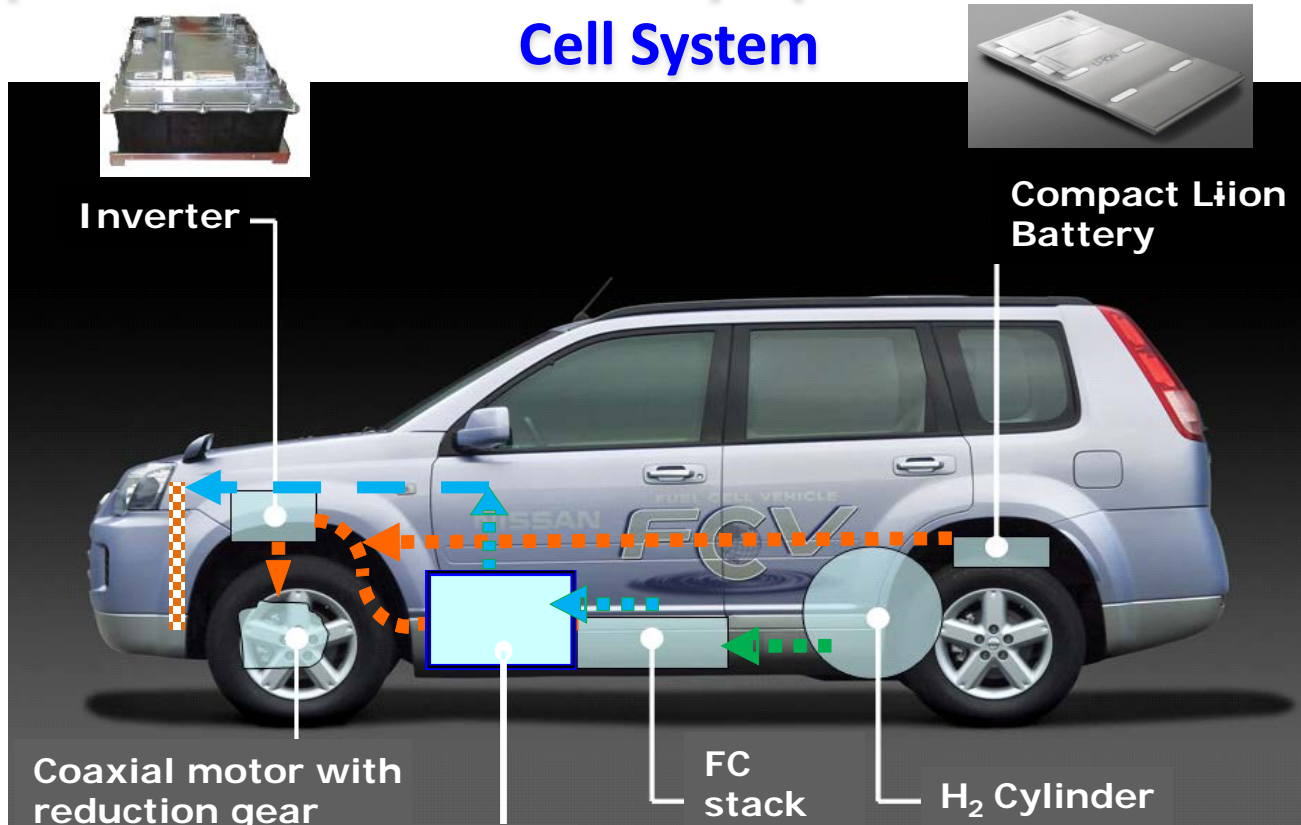
# Conclusions

- Reproducible, high conductivity in scaled up powder batches
  - Proton Conductivity of 0.1 S/cm
- High loading of TPP in polymer composite
- Single 5 cm<sup>2</sup> membrane performance of ~ 300 mW/cm<sup>2</sup> demonstrated (High Pt loading) – porous membrane
- Dense composite membrane fabricated
- Low Pt loading (0.2 mg/cm<sup>2</sup>), 5 cm<sup>2</sup> cell demonstrated >400 mW/cm<sup>2</sup>
- Early versions of cells demonstrated in 50 cm<sup>2</sup> size

# MOTIVATION



# Mid-Temperature and Low Humidity Operation Benefits for the Fuel Cell System



## BOP system

Possible Cost Saving

- Radiator
- Humidification
- Compressor-Expander Unit
- H<sub>2</sub> Purification



- Hydrogen
- Electricity
- Coolant

Mid temperature operation can lower the FC system cost by simplifying the Balance of Plant (BOP)

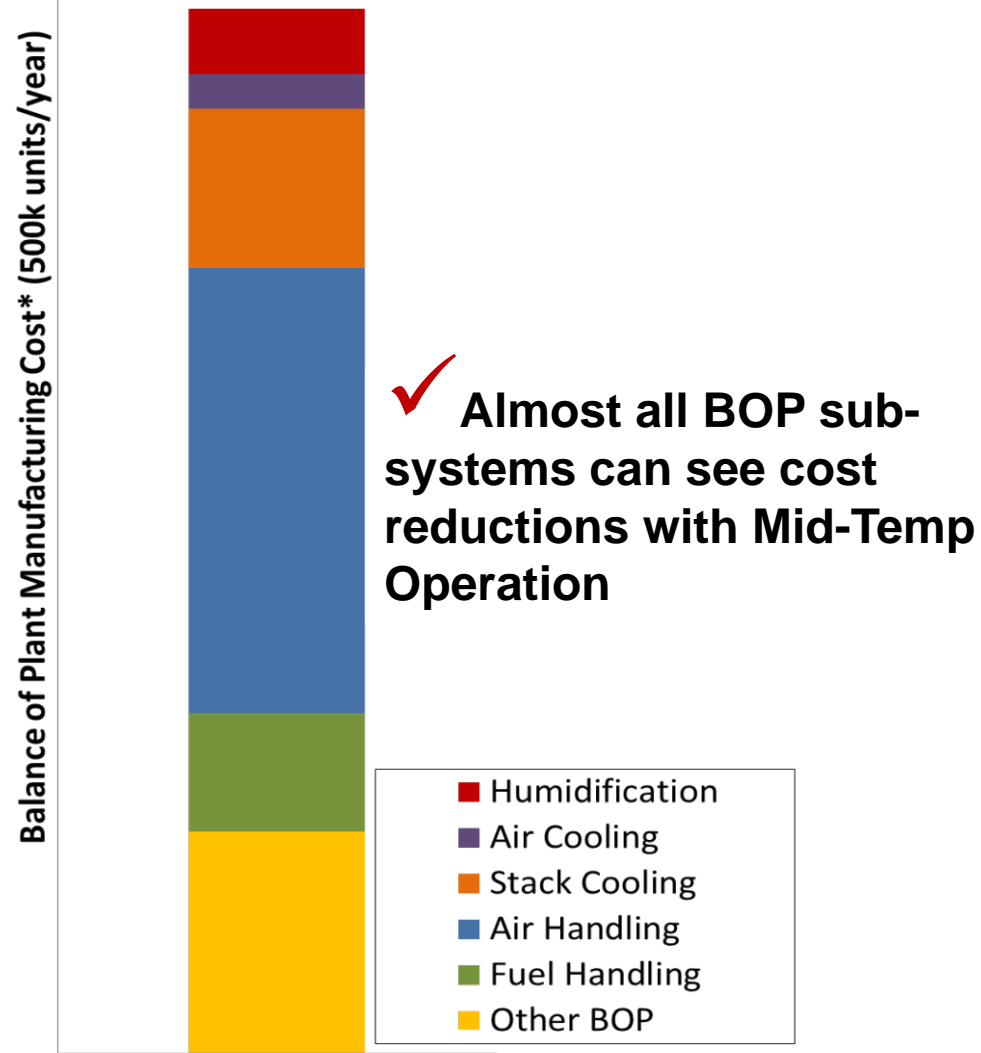
# FCEV System-Level Modeling

## Mid-Temp FCEV System Cost Estimation

### Major Cost Saving component/system \*

1. Air Handling
  - ✓ Compressor
  - ✓ Expander
  
2. Water/ Heat Recovery
  - ✓ Humidifier
  - ✓ Radiator
  - ✓ Coolant Loop

2013 FC System\*\*



\* Compared to conventional FC system

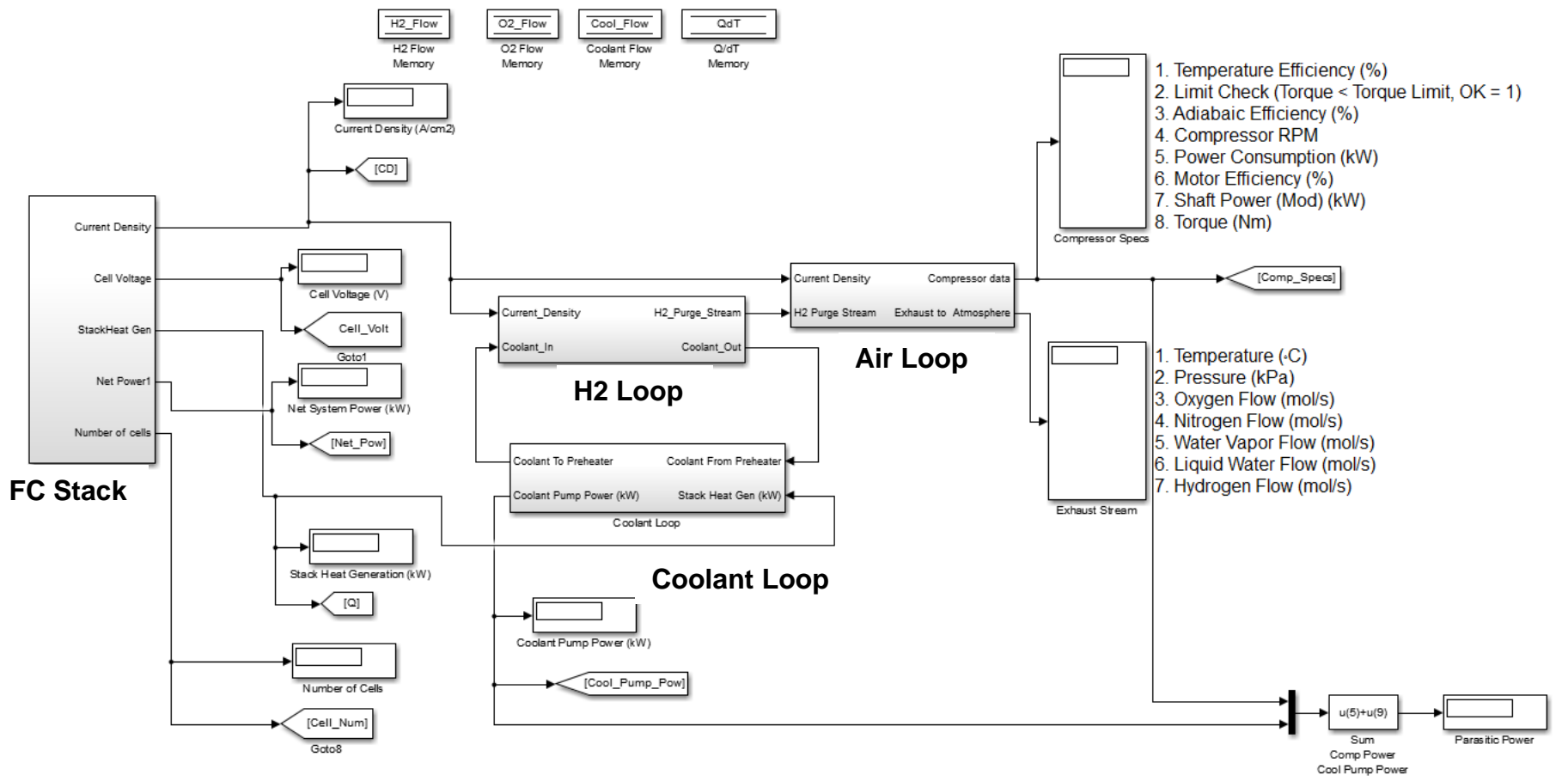


\*\* James Brian D, " Fuel Cell Transportation Cost analysis Prelim Results" DOE Annual Review, May 2013



# FCEV System-Level Modeling

## Overall Model in Matlab+Simulink



Lot more sub-layer and sub-systems also built-in

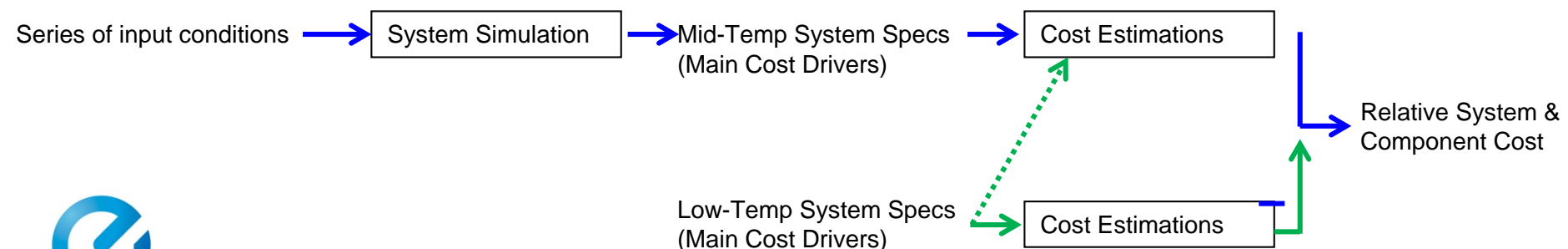


# FCEV System-Level Modeling

## Mid-Temp FCEV System Cost Estimation



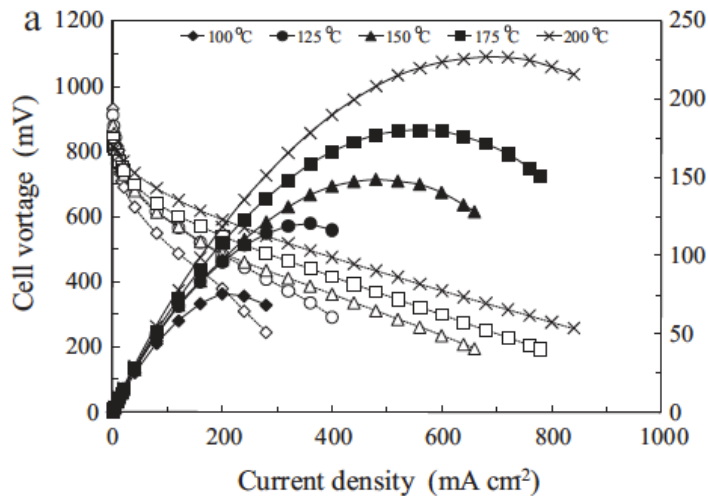
- **System specifications are calculated from the system simulation.**
  - The necessary specification ranges for FCEV operation will be determined
  
- **The determined specs will be used to estimate the relative cost of the system components with respect to a Low-Temp FCEV System**
  - Main cost drivers to be determined and will be the focus of the simulation
  
- **For Example**
  - The air compressor cost is a major cost driver for the system
  - Compressor cost is primarily determined by the required pressure ratio and the torque
    - These specs are calculated over a range of operating conditions



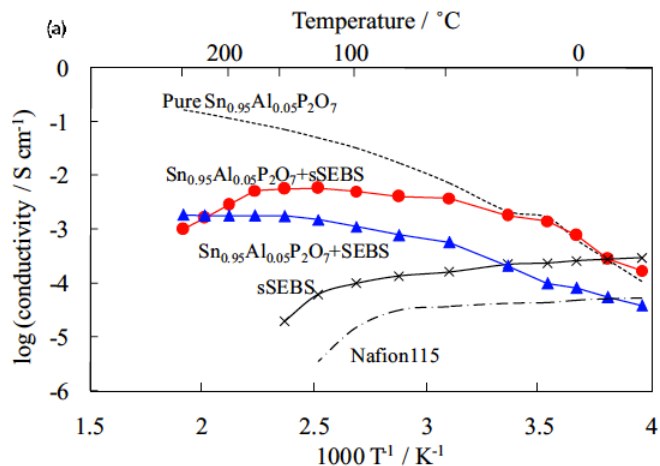
# MEMBRANE MATERIAL



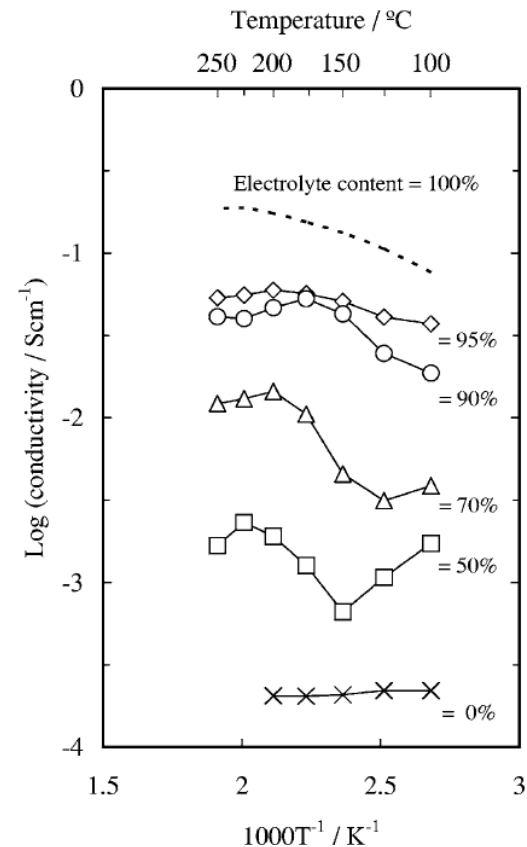
# State of the Art – Indium Tin Pyrophosphate (ITPP Fuel Cells and Composite Membranes)



Y.C. Jin et al. / *Journal of Power Sources* 196 (2011) 4905–4910



*Electrochemical and Solid-State Letters*, 13 (2) B8-B10 (2010)



*Journal of The Electrochemical Society*, 154 (1) B63-B67 (2007)

## Project Goals

- ✓ Double State of the art power density
- ✓ Improve Conductivity 5 times

# Conductivity of $\text{In}_{0.1}\text{Sn}_{0.9}\text{P}_2\text{O}_7$ with varying P:M

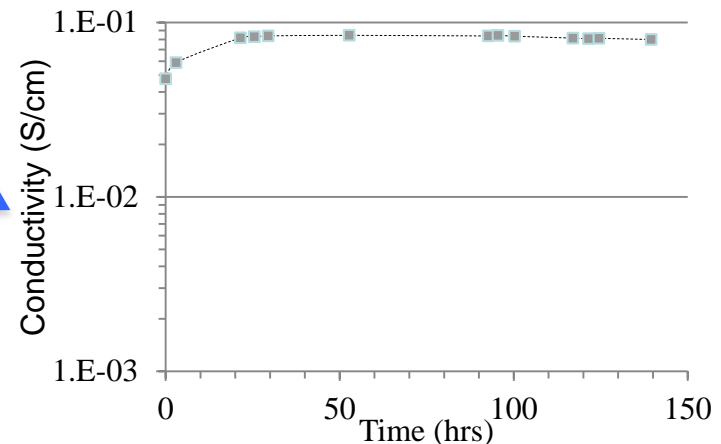
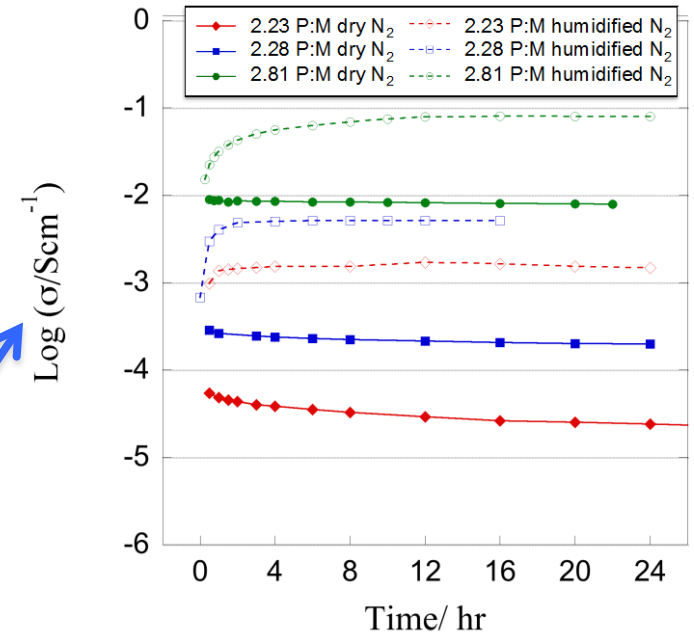
- High Proton Conductivity at Intermediate temp. in anhydrous reported for In-doped Sn pyrophosphates
- Inconsistent reproducibility in conductivity reported

Composition optimization for reproducible, high conductivity - LANL

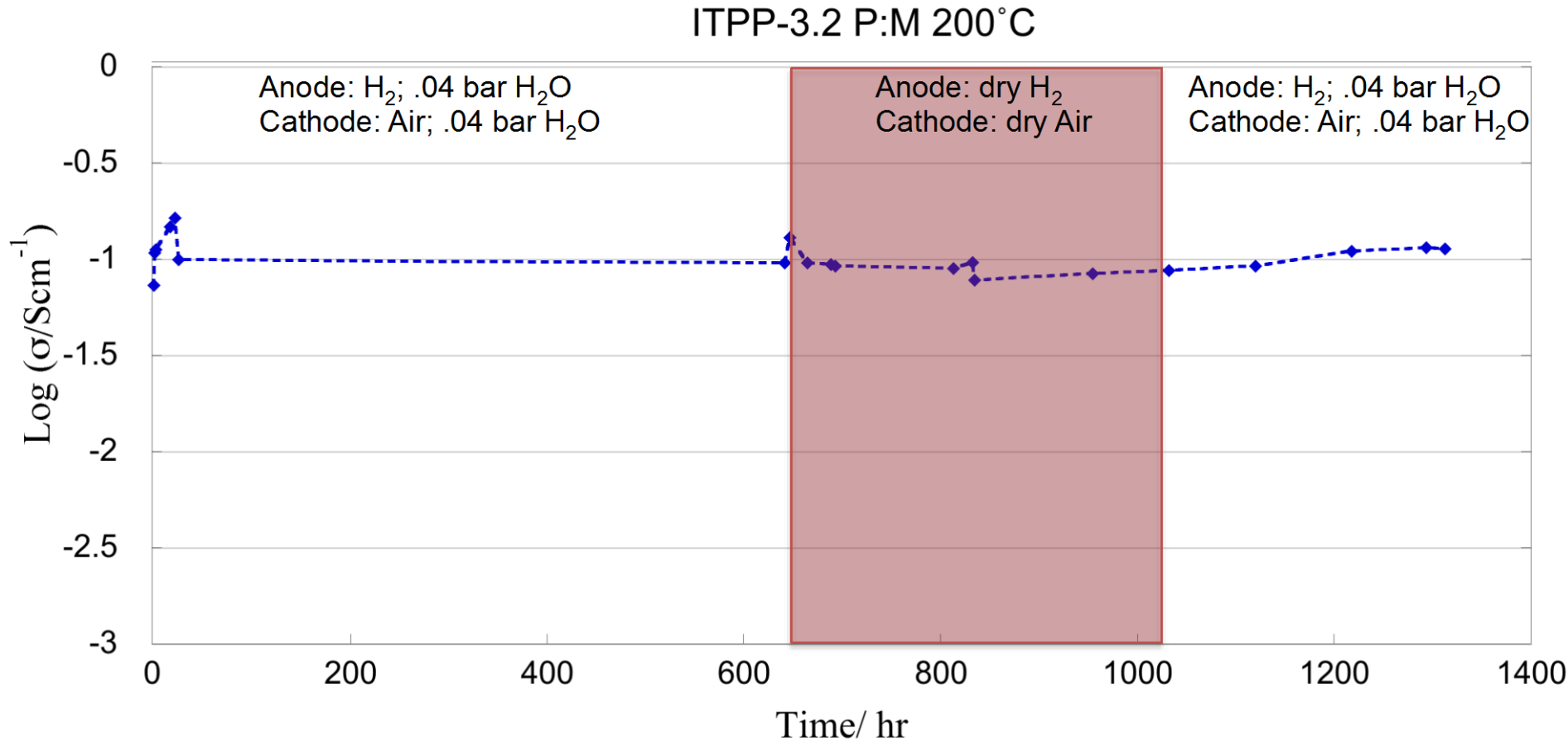
Batch scale up and high conductivity - Ceramatec

- ✓ Conductivity of nominal material (2.02 P:M) is negligible at 250°C.
- ✓ P/M > 3;  $\sigma \approx 10^{-1} \text{ Scm}^{-1}$

Kreller, C.R.; Wilson, M.S.; Mukundan, R.; Brosha, E.L.; Garzon, F.H. *ECS Electrochemistry Letters* 2013; 2(9): F61-F63.



# Stability of Conductivity

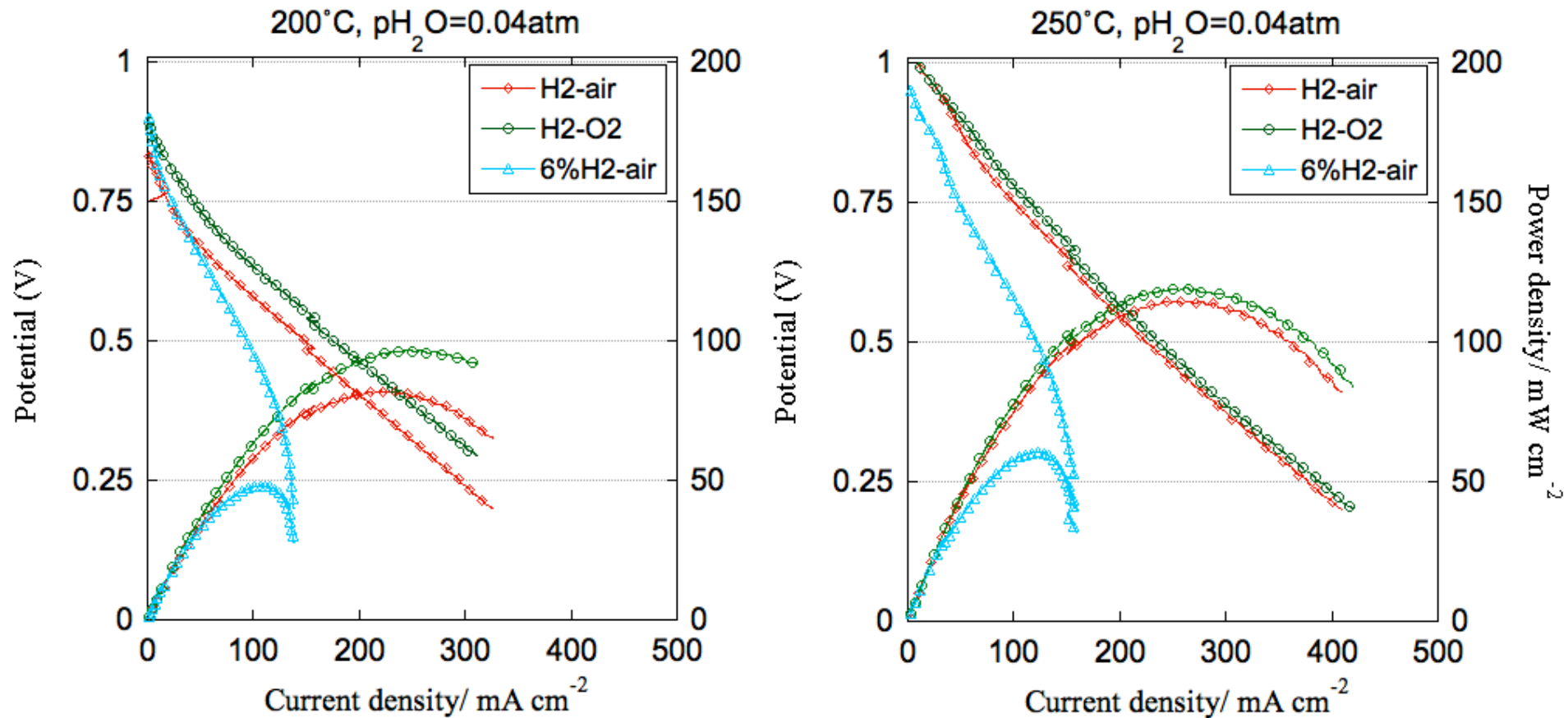


# SINGLE CELL TESTING



# Electrolyte 1mm thick ITPP

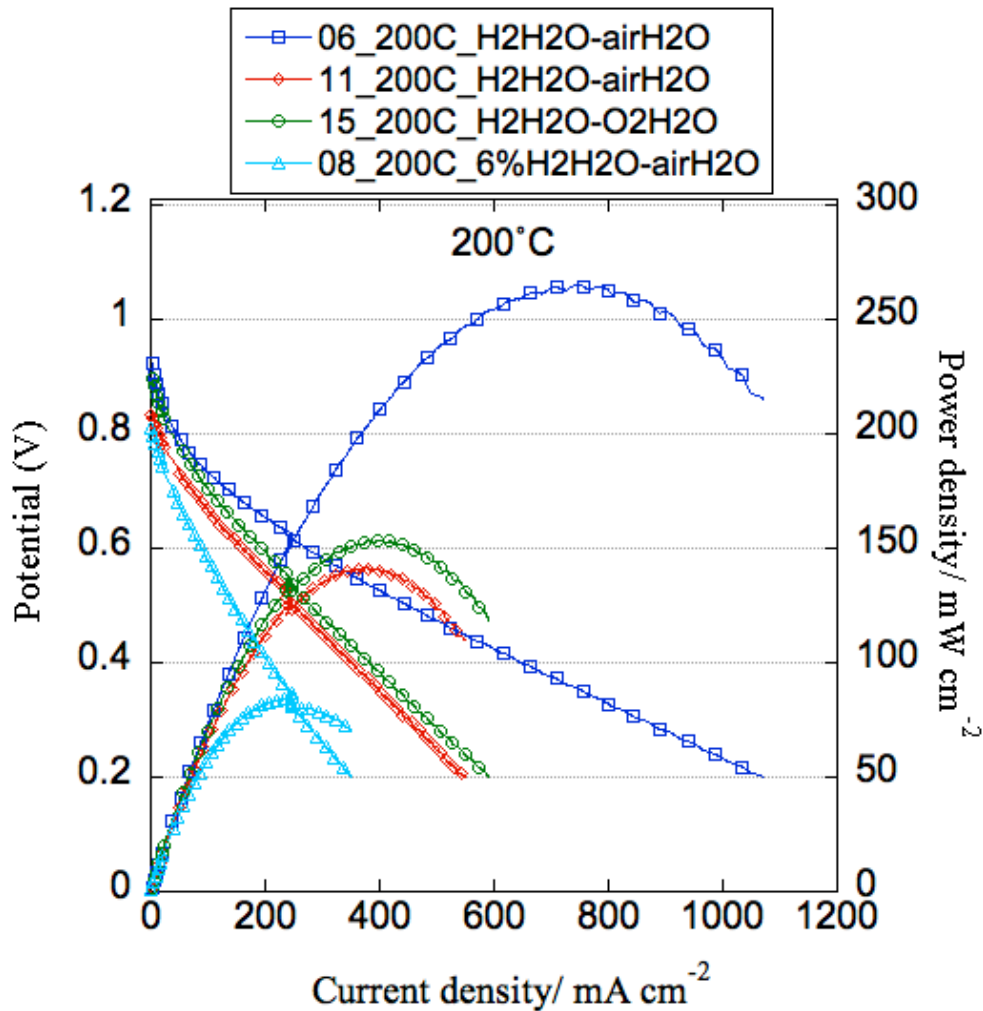
## Electrode: 10mg/cm<sup>2</sup> GDE + Phos Acid



- ✓ Anode performance is limiting in 6% H<sub>2</sub>
- ✓ Electrolyte conductivity is critical

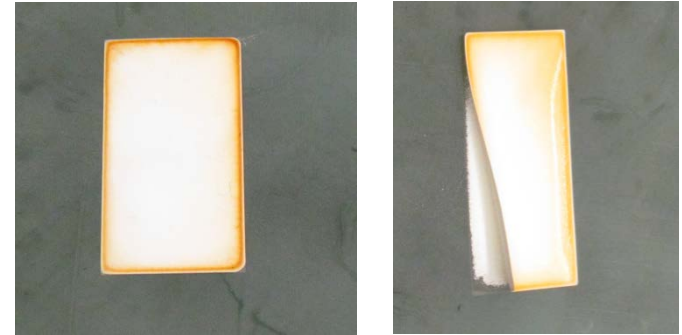
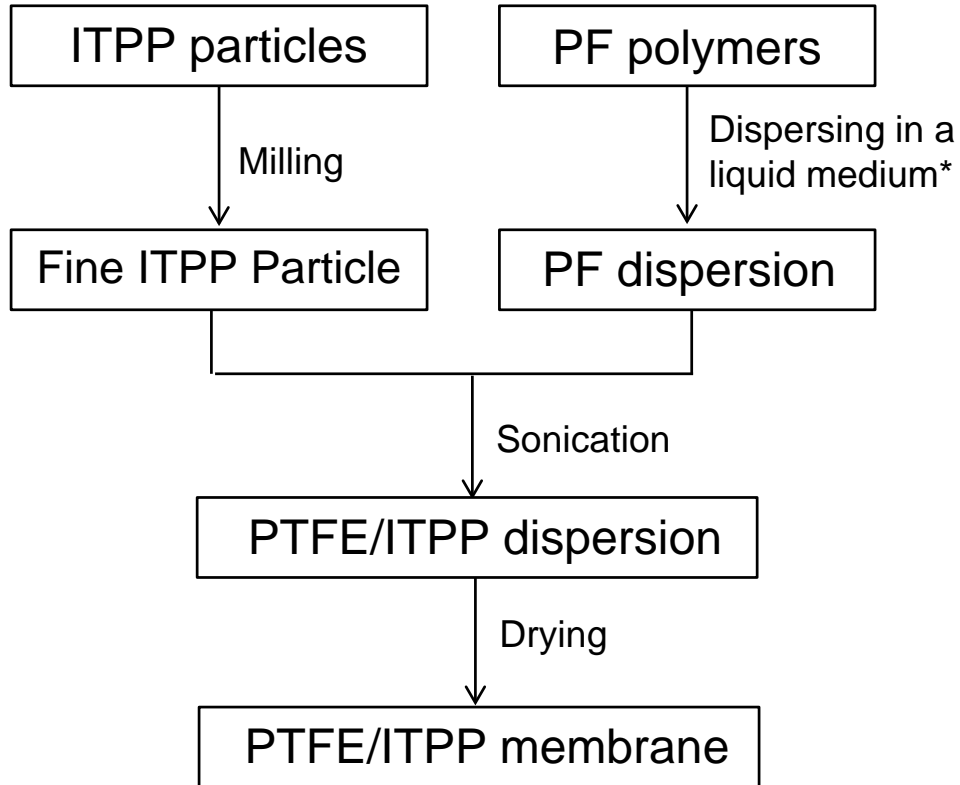


Electrolyte 0.025mm thick TPP 15wt% SiC whiskers  
 Electrode: Pt 10mg/cm<sup>2</sup> GDE + Phos Acid: 4μl cathode/2μl anode



✓ Significant performance improvement with thin electrolytes

# Fabrication process of composite membranes



## Cast composite membranes

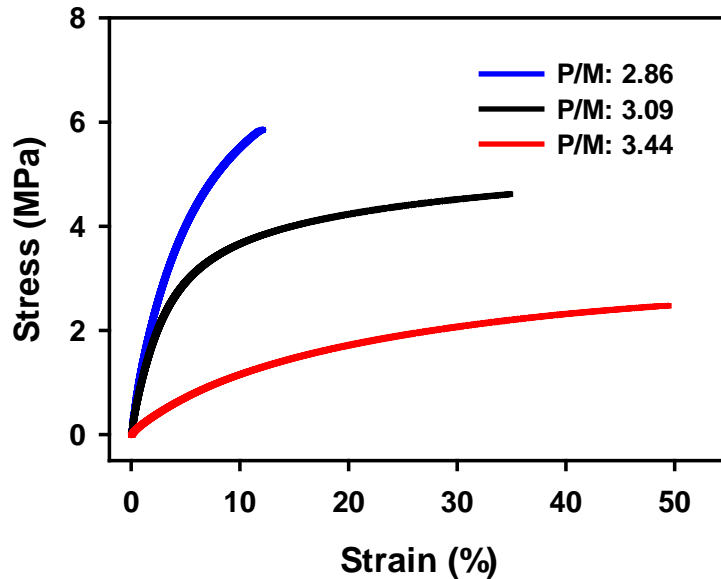
ITPP content: up to 90%

Thickness: 10 - 350  $\mu\text{m}$

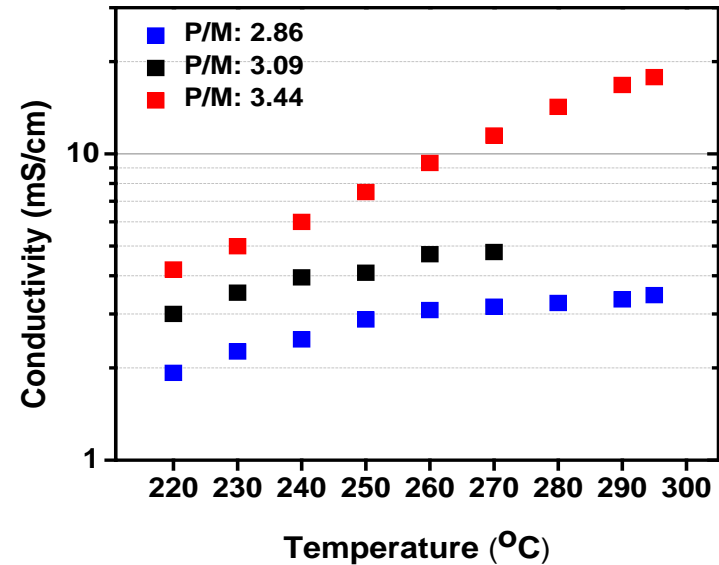
Conventional sintering results in loss of proton conductivity

# Effect of P:M ratio on mechanical and electrochemical properties

**Stress-strain curves**  
(ITPP content: 75wt.%)



**Proton conductivity**  
(ITPP content: 75wt.%)

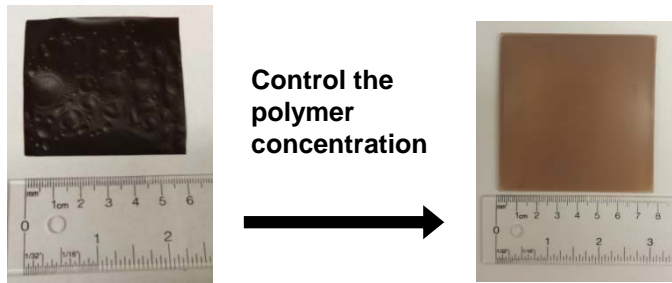


- ✓ Elongation Strength ↑; Stress and Modulus ↓ with increasing P:M ratio
- ✓ Conductivity increased with P:M ratio
- ✓ Further optimizations are needed in terms of P:M ratio, ITPP content and casting solvent

# Large Area Membrane Fabrication

□ Optimization of membrane formation : Issue of surface uniformity

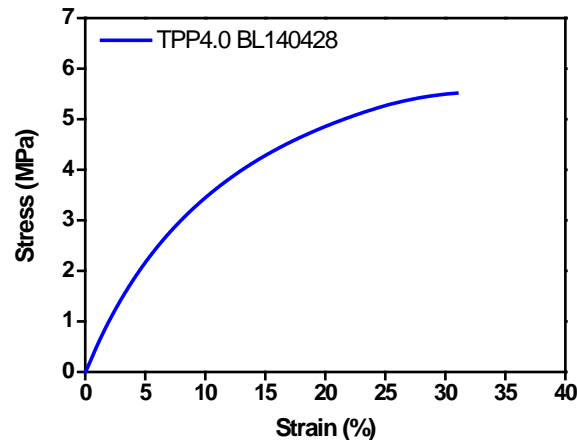
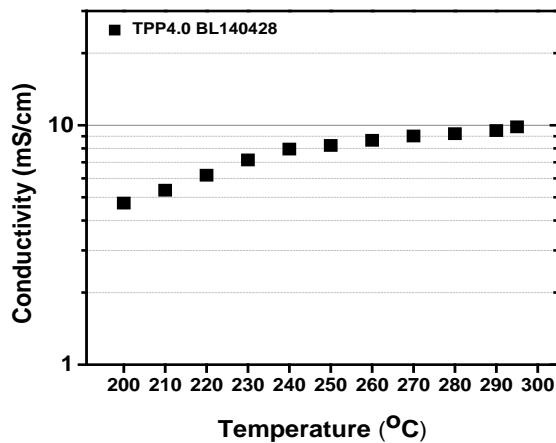
→ PF polymer solution concentration: 5~8 wt%



3 X 3 in<sup>2</sup> membranes  
ITPP content: 75 %  
Thickness: 50 μm ± 2



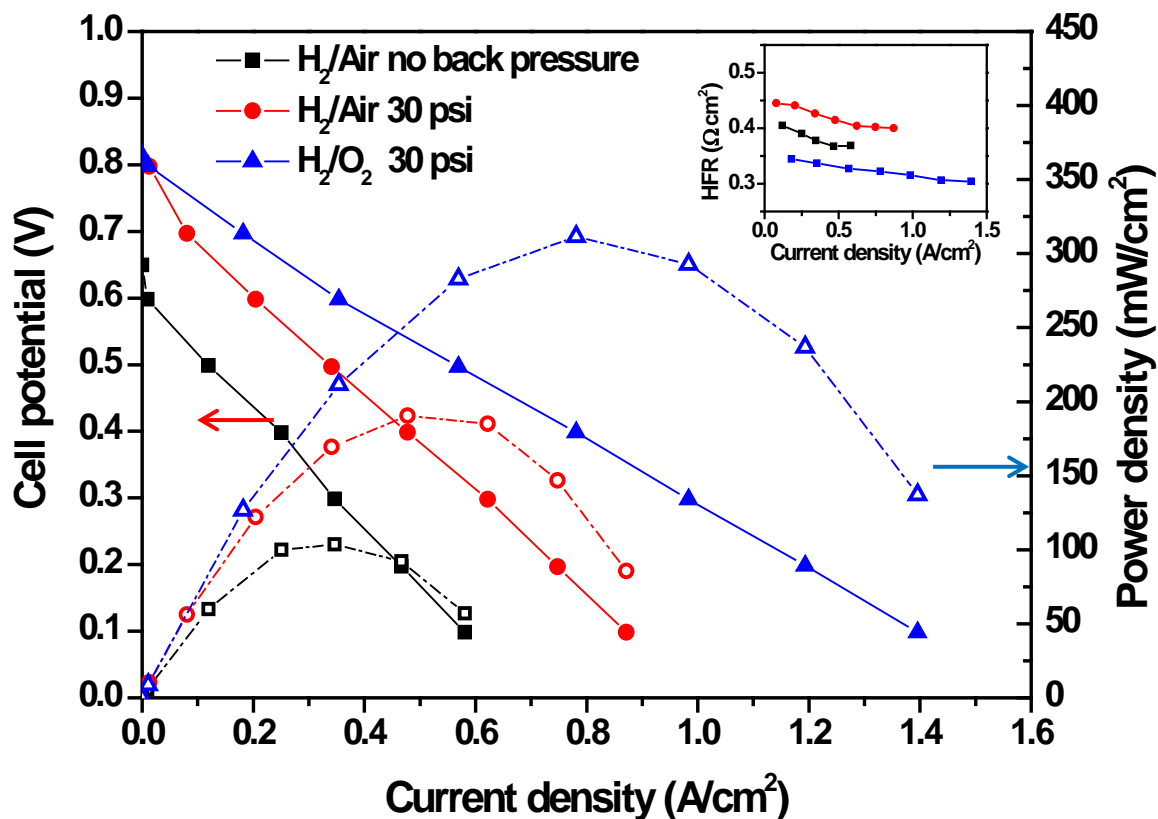
## Proton conductivity & Mechanical property (Stress-strain curves)



ITPP sample	Ceramatec sample # TPP4.0 BL140428 (P:M ratio : 3.28)
ITPP content	75 wt%
Casting temperature	140 °C
Casting time	4 h

# Performance\_TPP90wt%/PA

## TPP90wt%/PA (electrode: Pt Black + TPP)



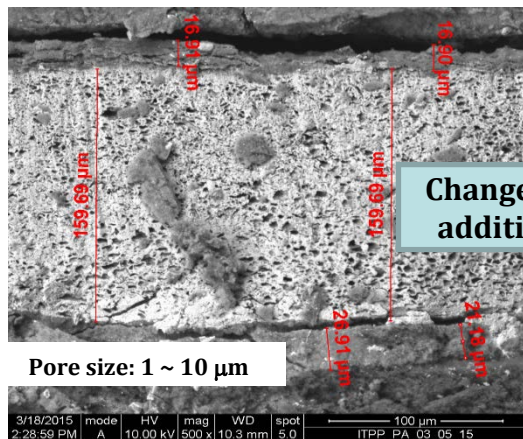
Sample	HFR (Ω cm <sup>2</sup> )	Conductivity (mS/cm)
H <sub>2</sub> /Air no back pressure	0.37 ~ 0.4	30 ~ 33
H <sub>2</sub> /Air 30 psi	0.40 ~ 0.44	27 ~ 30
H <sub>2</sub> /O <sub>2</sub> 30 psi	0.30 ~ 0.34	35 ~ 40

### Condition

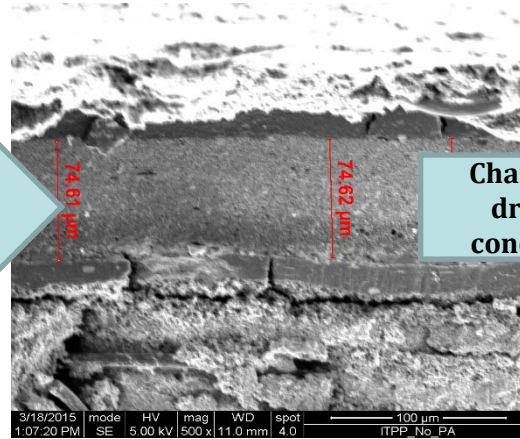
- Membrane thickness: 120 μm
- Anode/Cathode/Cell Temp: 80/80/220 °C
- H<sub>2</sub>/Air (H<sub>2</sub>/O<sub>2</sub>): 200/200 sccm
- Back pressure: varied (0-30 psi)
- Pt loading: 3.5 mg/cm<sup>2</sup>

✓ Best performance in H<sub>2</sub>/O<sub>2</sub> condition with 30psi back pressure

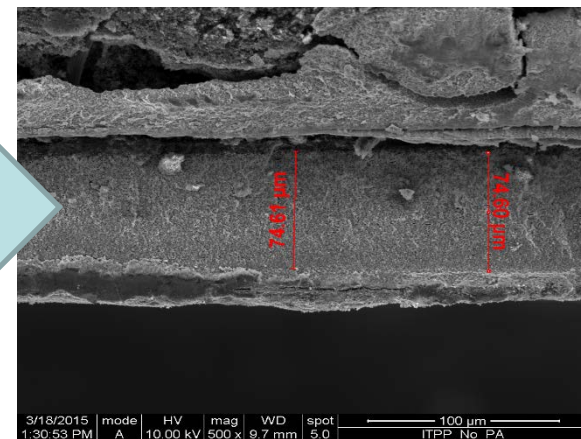
# SEM images of TPP/PF composite membrane



Change in additive



Change in drying condition



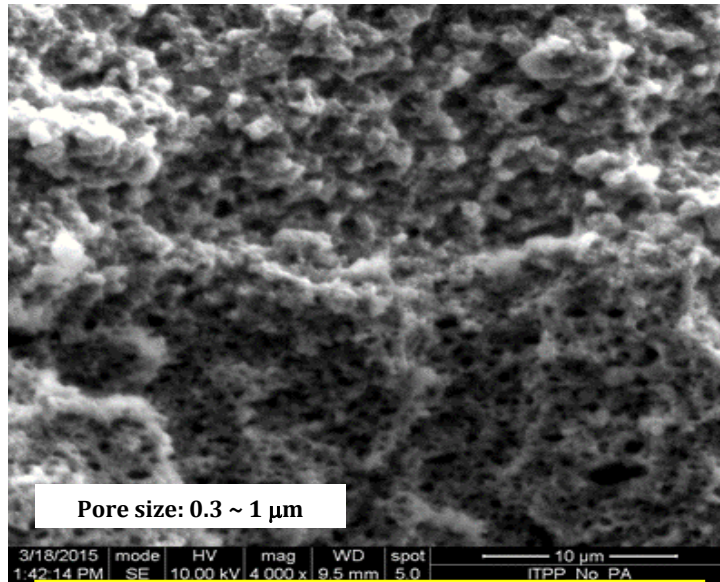
- Membrane thickness: 180  $\mu\text{m}$
- **No back pressure**
- **No humidification**
- $\text{H}_2/\text{O}_2$ : 200/200 sccm
- **Pt loading: 0.2 mg/cm<sup>2</sup>**
- Cell temperature: 200  $^\circ\text{C}$
- OCV = 820 mV.

- Membrane thickness: 120  $\mu\text{m}$
- **No back pressure**
- **No humidification**
- $\text{H}_2/\text{O}_2$ : 200/200 sccm
- **Pt loading: 0.2 mg/cm<sup>2</sup>**
- Cell temperature: 220  $^\circ\text{C}$
- OCV = 840 mV.

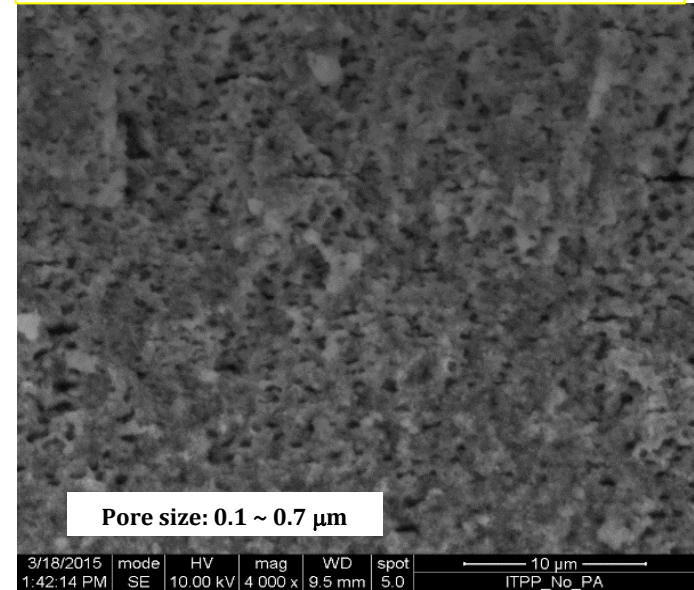
- Membrane thickness: 100  $\mu\text{m}$
- Back pressure: 30 psi
- No humidification
- $\text{H}_2/\text{O}_2$ : 200/200 sccm
- **Pt loading: 0.2 mg/cm<sup>2</sup>**
- Cell temperature: 200  $^\circ\text{C}$
- OCV = 810 mV.

# Membrane Porosity

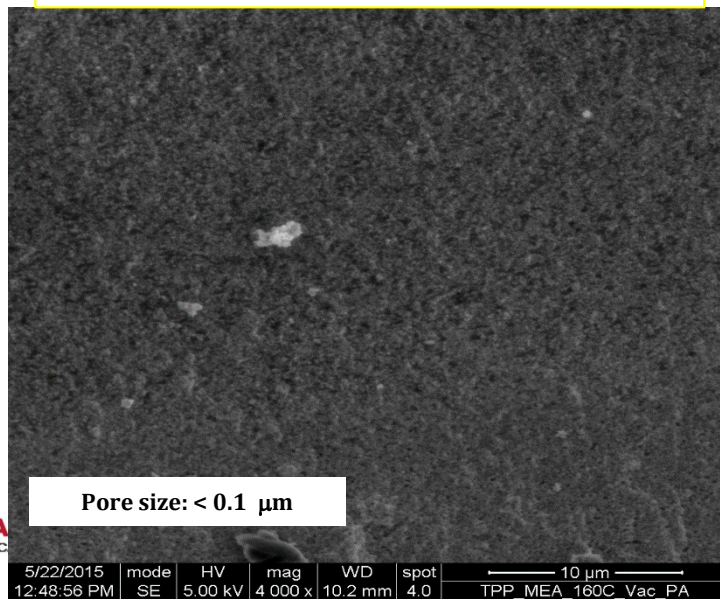
Initial Process



Change in drying condition

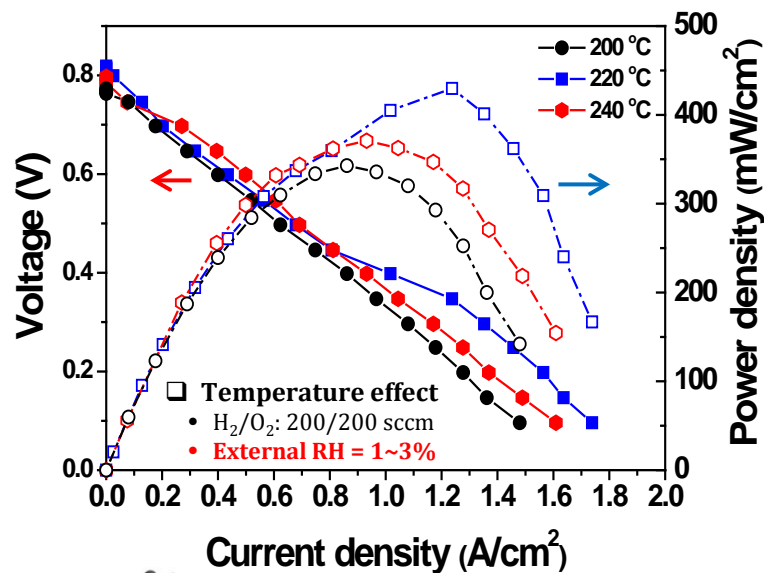
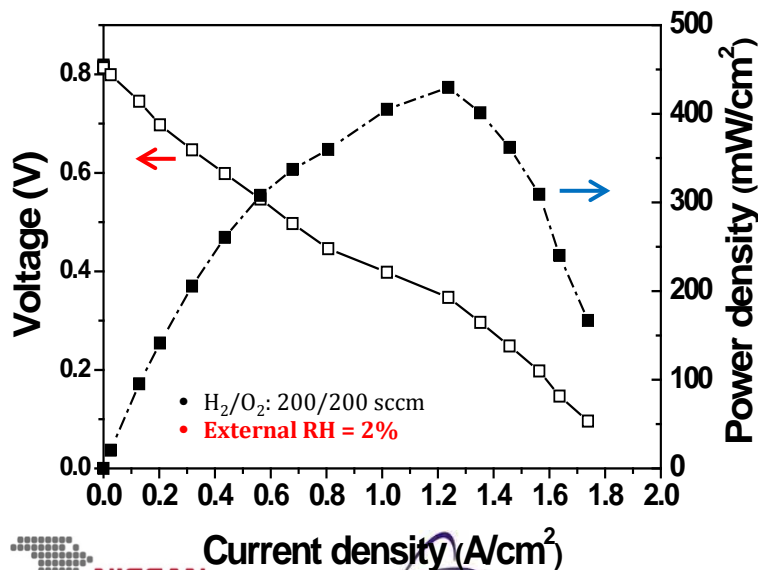
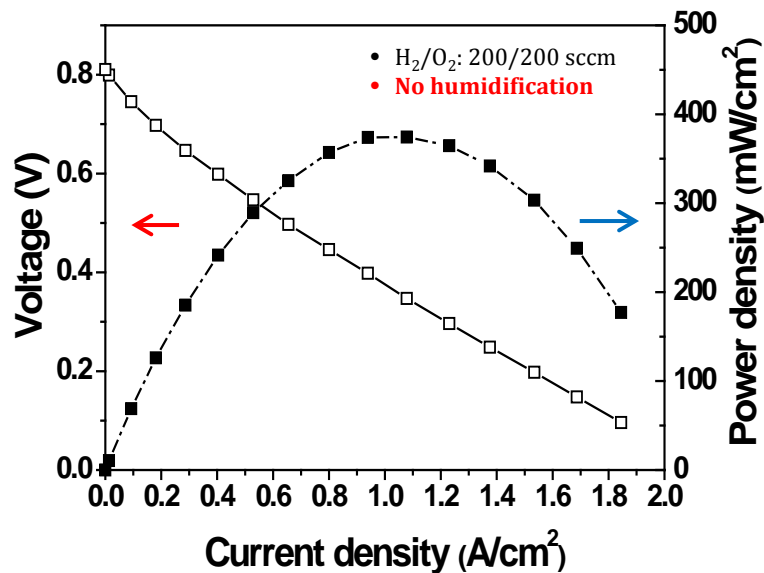
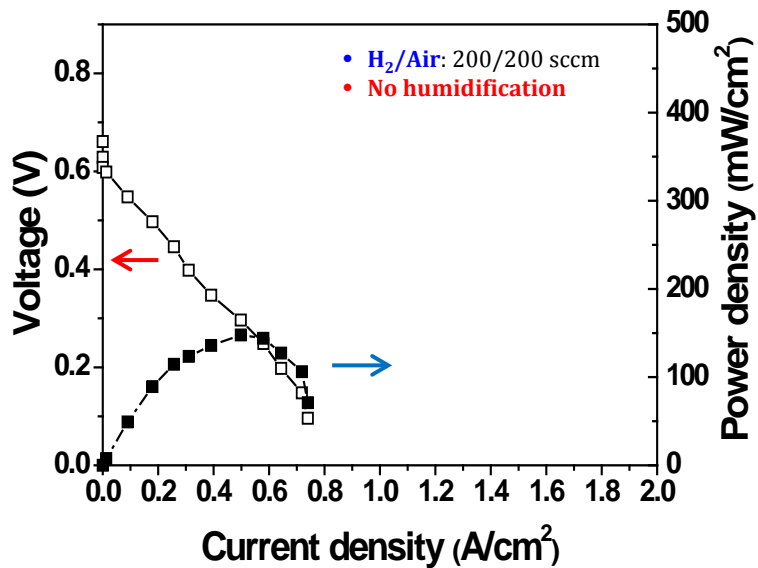


Change in drying condition and Temperature



✓ Successfully prepared denser membranes

- Modified slip preparation and drying conditions



□ Membrane  
 - TPP content: 90 wt%  
 - Thickness: 80  $\mu m$   
 - Drying condition changed

□ Catalyst  
 - 20wt% Pt/C (BASF)  
 - 0.2  $mg_{PGM}/cm^2$  for each electrode

□ Electrode  
 - No TPP particles  
 - Added  $H_3PO_4$   
 - LANL developed ionomer  
 - Ionomer is stable at high temperature operation.

□ Operating cond.  
 - Cell temp.: 220  $^{\circ}C$   
 - Backpressure: 30psig  
 - Cell break-in: 24 h

□ MEA preparation  
 - Active area: 4.4  $cm^2$   
 - PTFE gaskets  
 - High temperature GDL  
 - GDL painting without hotpressing

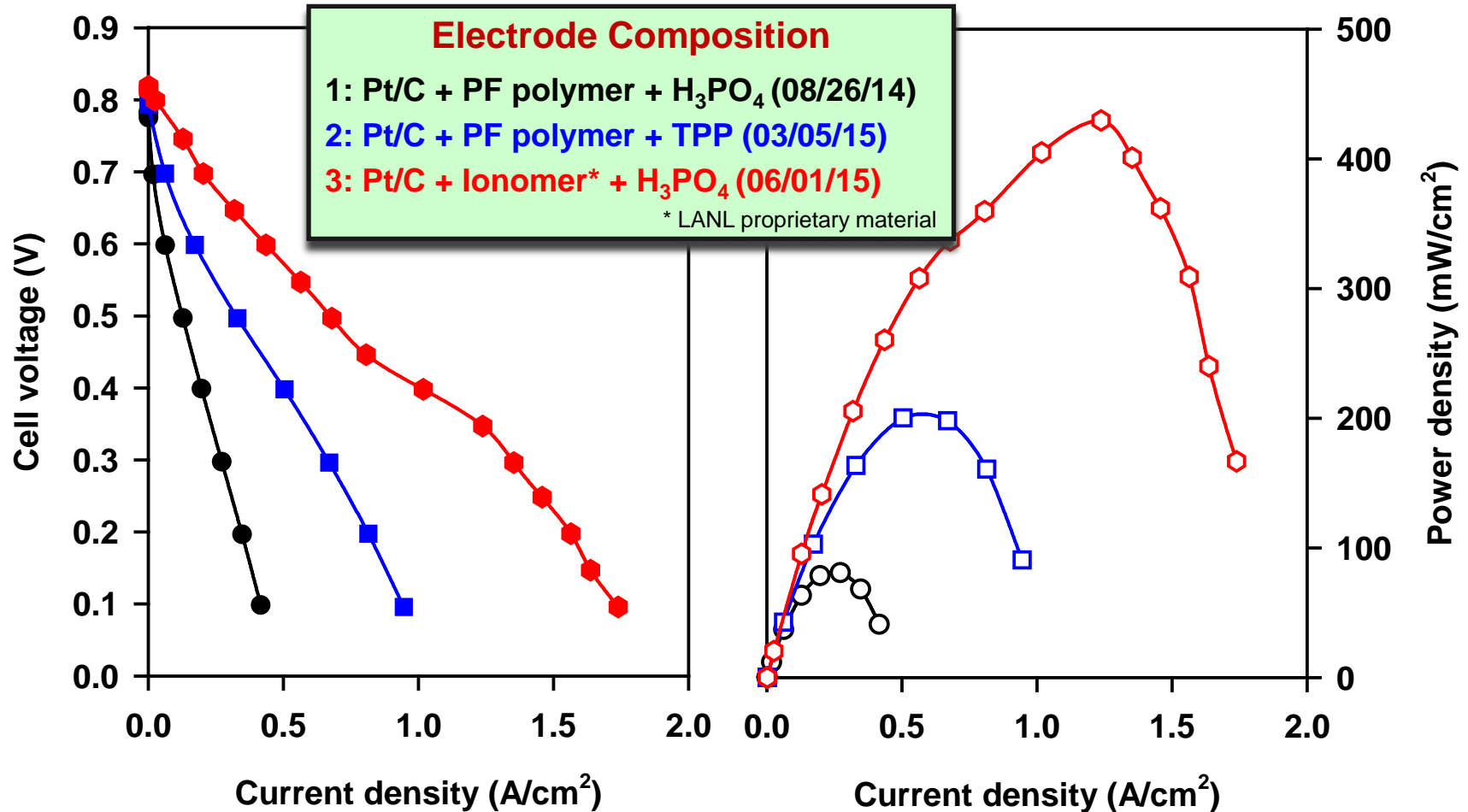




# Electrode Optimization towards Better Fuel Cell Performance

Test conditions: H<sub>2</sub> (200 sccm)/O<sub>2</sub> (200 sccm); external RH = 0-2%; 5 cm<sup>2</sup> hardware  
 Operating temperature: 1: 240° C, 2: 230° C, 3: 220° C  
 Membrane TPP content: 90 wt.%; Membrane thickness: 80-180 μm  
 Catalyst: 20 wt.% Pt/C (BASF); catalyst loading: 0.2 mg/cm<sup>2</sup> for each electrode

● 1: 08/26/14  
 ■ 2: 03/05/15  
 ◆ 3: 06/01/15



# Conclusions

- Reproducible, high conductivity in scaled up powder batches
  - Proton Conductivity of 0.1 S/cm
- High loading of TPP in polymer composite
- Single 5 cm<sup>2</sup> membrane performance of ~ 300 mW/cm<sup>2</sup> demonstrated (High Pt loading) – porous membrane
- Dense composite membrane fabricated
- Low Pt loading (0.2 mg/cm<sup>2</sup>), ionomer in GDE, 5 cm<sup>2</sup> cell demonstrated >400 mW/cm<sup>2</sup>
- Early versions of cells demonstrated in 50 cm<sup>2</sup> size

# Remaining Challenges

- Increase in OCV
- High performance cells in 25 cm<sup>2</sup> and 50 cm<sup>2</sup> size
- Design/build/test multi-cell stack
- Long term performance stability evaluation
- CO tolerance evaluation
- Complete cost model

# Acknowledgement

- ARPA-E Team
  - Program Director: Dr. John Lemmon
  - Technical Support: Dr. Scott Litzelman
  - Tech to Market: Mr. Sven Mumme

