



# Gasification Technology Options for SOFC Applications

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# Presentation Outline

- SOFC Syngas Purity Requirements
- Gasification Technologies and Characteristics
- Gas clean up options
  - Currently available
  - In Development
  - R&D Status

# Contaminants in Coal

- Your average lump of coal contains a lot more than just Carbon, Hydrogen, Oxygen, Nitrogen and Sulfur (with a touch of Chlorine)

## Typical Metal Contaminants in Coal

Coal Type	Hg (ppm)	As (ppm)	Se (ppm)	Cd (ppm)
Pittsburg	0.11	4.1	0.6	0.06
Elkhorn/Hazard	0.13	4.0	3.1	0.31
Illinois No.6	0.22	2.7	2.2	0.15
Wyodak	0.19	1.3	1.6	0.30

Bool et al., 1997

- In addition to these metal contaminants, coal ash contains the following:
  - Si, Al, Fe, Ti, P, Ca, Mg, Na, K, S, and more

# Coal Gasification Chemistry & Reactions

## Important reactions in coal gasification:

<b>Coal Devolatilization = CH<sub>4</sub> + CO + CO<sub>2</sub> + Oils + Tars + C (Char)</b>	
<b>C + O<sub>2</sub> = CO<sub>2</sub></b>	(exothermic – rapid)
<b>C + 1/2O<sub>2</sub> = CO</b>	(exothermic – rapid)
<b>C + H<sub>2</sub>O = CO + H<sub>2</sub></b>	(endothermic – slower than oxidation)
<b>C + CO<sub>2</sub> = 2CO</b>	(endothermic – slower than oxidation)
<b>CO + H<sub>2</sub>O = CO<sub>2</sub> + H<sub>2</sub></b>	<b>Shift Reaction (slightly exothermic)</b>
<b>CO + 3H<sub>2</sub> = CH<sub>4</sub> + H<sub>2</sub>O</b>	Methanation (exothermic)
<b>C + 2H<sub>2</sub> = CH<sub>4</sub></b>	Direct Methanation (exothermic)

# Typical O<sub>2</sub>-Blown Gasifier Produced Syngas

- H<sub>2</sub> 30 - 50%
- CO 40 - 60%
- CO<sub>2</sub> 4 - 20%
- H<sub>2</sub>S 0.5 - 2%
- COS ~500<sup>+</sup> ppmv
- Ar 0.5 - 1%
- N<sub>2</sub> 0.7 - 6%
- NH<sub>3</sub> ~50-100 ppmv
- Ni & Fe\* Carbonyls 1 to 4 ppmv
- HCN ~50-100 ppmv
- HCOOH\* ~50 ppmv

## Acid Gases

*Trace Components Include: As, P, Hg, Cd, Zn, Bi, Sb, Pb, Na, K, Fe, Ni*

*+ H<sub>2</sub>S: COS is typically about 95%:5% of total sulfur*

*\* Carbonyls and Formic Acid (HCOOH) formed downstream of Gasifier*

# Estimated Thermodynamic Equilibrium State of Trace Components

Element	>1000°C	400° to 800°C	100° to 400°C	<100°C
As	AsO, As <sub>2</sub>	AsO, As <sub>4</sub>	As <sub>2</sub>	AsH <sub>3</sub>
Be	Be(OH) <sub>2</sub>	Condensed Species	Condensed Species	Condensed Species
Hg	Hg	Hg	Hg, HgCl <sub>2</sub>	Hg, HgCl <sub>2</sub>
B	HBO	HBO	HBO	-
V	VO <sub>2</sub>	Condensed Species	Condensed Species	Condensed Species
Se	H <sub>2</sub> Se, Se, SeO	H <sub>2</sub> Se	H <sub>2</sub> Se	H <sub>2</sub> Se
Ni	NiCl, NiCl <sub>2</sub>	Condensed Species	Ni(CO) <sub>4</sub>	Ni(CO) <sub>4</sub>
Co	CoCl <sub>2</sub> , CoCl	Condensed Species	Condensed Species	Condensed Species
Sb	SbO, Sb <sub>2</sub>	SbO, Sb <sub>2</sub>	Sb <sub>4</sub>	Condensed Species
Cd	Cd	Cd	CdCl <sub>2</sub>	Condensed Species
Pb	Pb, PbCl <sub>2</sub>	PbS, Pb, PbCl <sub>2</sub>	Condensed Species	Condensed Species
Zn	Zn	Zn, ZnCl <sub>2</sub>	Condensed Species	Condensed Species

Gasifier operating range

Source: SRI International presentation from 2006 SECA Review Meeting

# Known Issue: Nickel Carbonyls

- Nickel in feedstock ash can react with CO in the syngas if at high partial pressure and form Nickel carbonyl
  - $\text{Ni} + 4\text{CO} \rightleftharpoons \text{Ni}(\text{CO})_4$
  - $\text{NiS} + 4\text{CO} + \text{H}_2 \rightleftharpoons \text{Ni}(\text{CO})_4 + \text{H}_2\text{S}$
- Both ISAB and Puertollano have reported that Ni carbonyl has passed through all the gas clean-up steps and reached the CT where it has plated out on the hot section parts
- Direct water quench should remove some carbonyls
- Some AGR processes will remove carbonyls
  - Rectisol, yes
  - Selexol, expected to remove carbonyls but no verified experience
  - MDEA solvents do not remove carbonyls
- Activated carbon beds should remove iron and nickel carbonyls

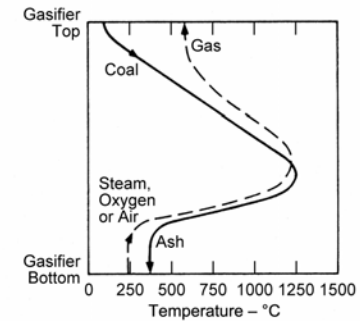
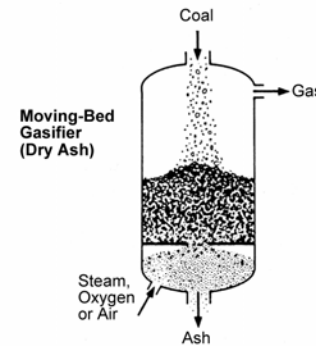
# SOFC Gas Purity Requirements

- No chlorides (reduced H<sub>2</sub>/CO adsorption on Ni)
- <5 ppm H<sub>2</sub>S (reduced steam reforming activity)
- Very low (Zero?) Aromatic Hydrocarbons ( Benzene, Naphthalene etc)
- No Cl, P, As (<1 ppm), Hg
- CO, H<sub>2</sub>, CH<sub>4</sub> are the main fuel components but the SOFC must be designed to handle the appropriate heat balance for the actual composition
- CO<sub>2</sub> and H<sub>2</sub>O can be tolerated but take up space and are preferably minimized
  
- Contaminant impacts require further study

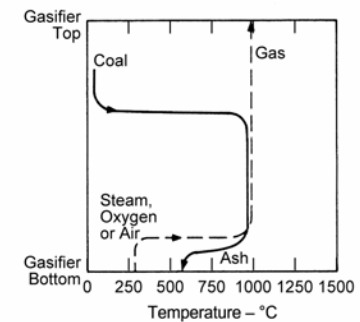
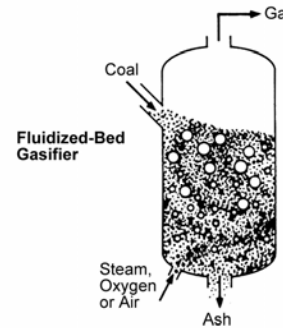


# The 3 Major Types of Gasification Processes

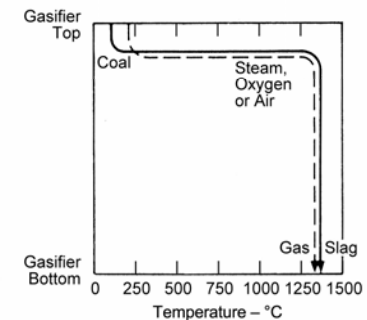
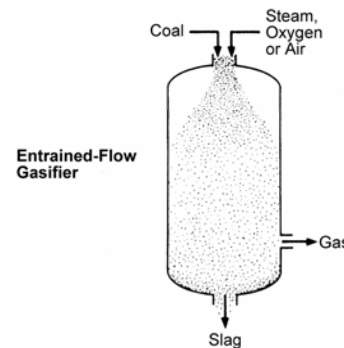
## 1. Moving-Bed Gasifier (Dry Ash)



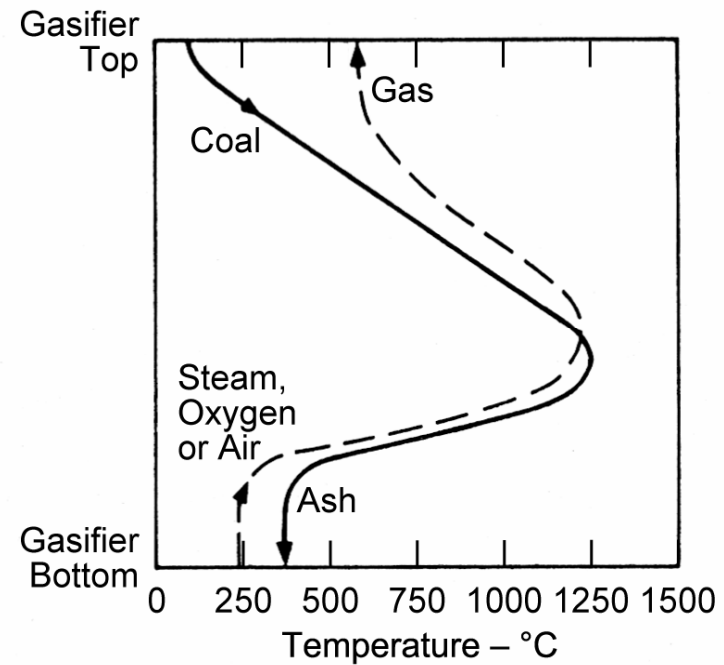
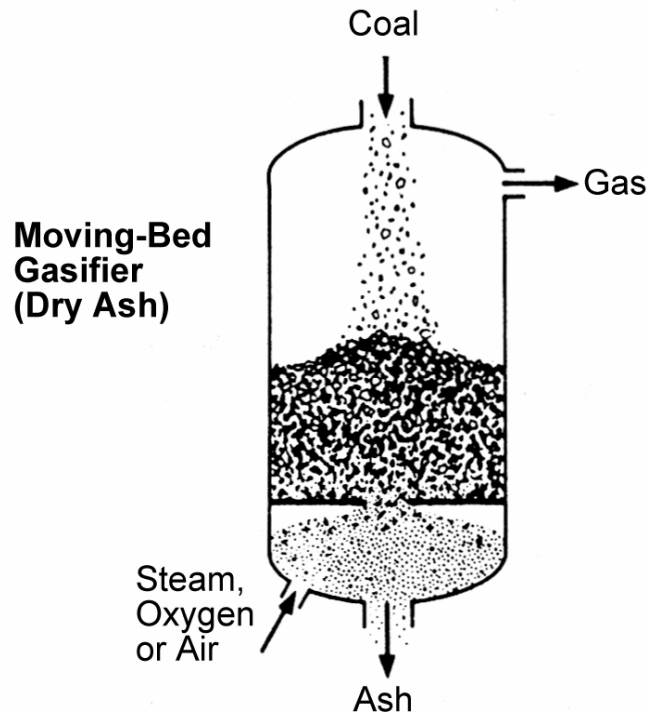
## 2. Fluidized-Bed Gasifier



## 3. Entrained-Flow Gasifier



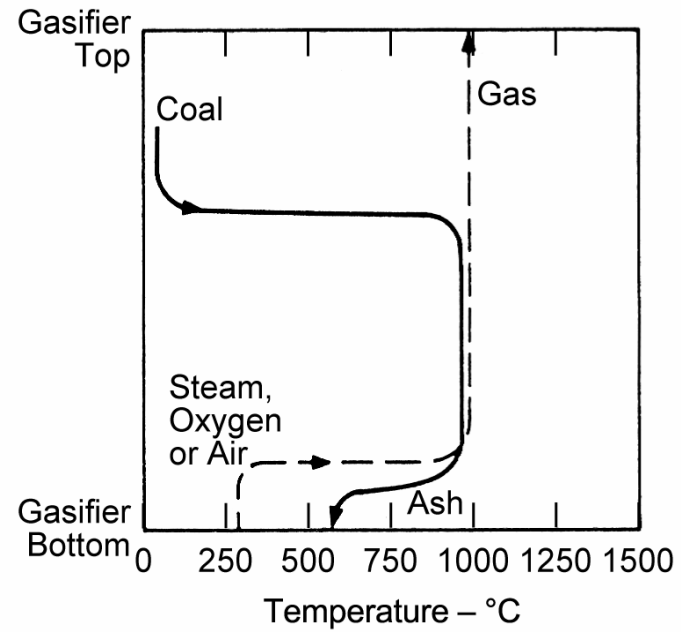
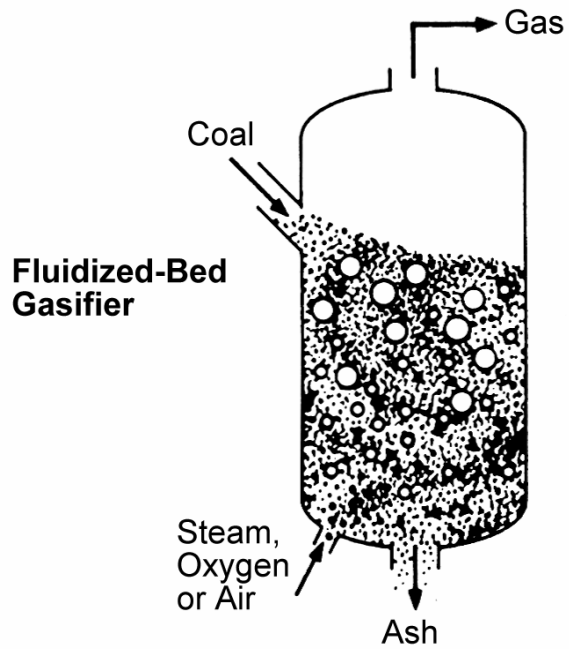
# 1. Moving-Bed Gasifier (Dry Ash)



# Moving Bed Gasifiers – Main Features (Lurgi dry ash, BGL etc)

- Lock Hopper top feed. Sized coal 2" x ¼" required
- Countercurrent operation
- Low outlet temperature 600-1000 F
- Tars, oils and phenolic byproducts
- Syngas 9-10% CH<sub>4</sub> or ~ 15-18% of Carbon in coal. Syngas not well suited for synthesis of Hydrogen, Ammonia, Methanol, DME, but OK for SNG.
- Lurgi dry ash units in operation worldwide (Sasol F-T, BEPC SNG)
- Steam added to keep coal below ash softening point in dry ash version. Results in high H<sub>2</sub> to CO ratio
- Most experience with lignites and lower rank coals. Bituminous coals need mechanical stirrer
- Atmospheric pressure units (Wellman etc) once widely used are not suitable for most current gasification applications
- BGL slagging version has some improvements over dry ash but limited commercial experience (one unit at Schwarze Pumpe)

## 2. Fluidized-Bed Gasifier

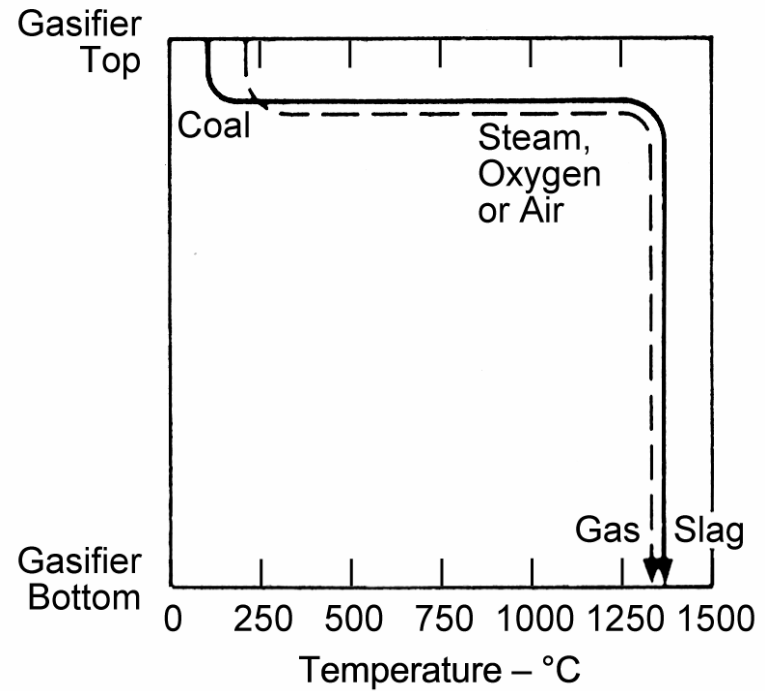
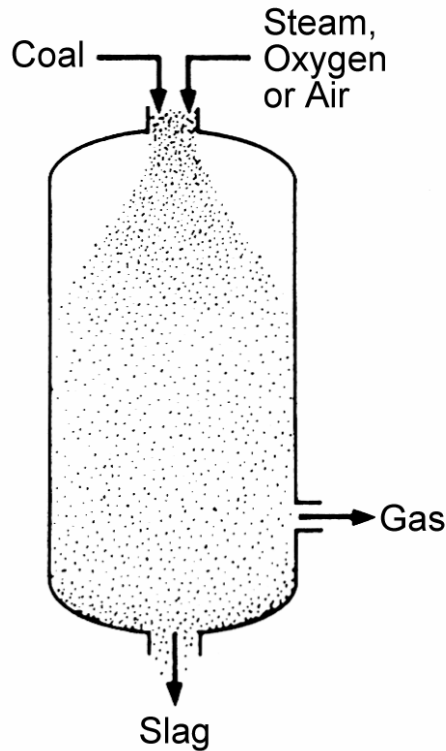


# Fluid Bed Gasifiers – Main Features (KBR, HT Winkler, U Gas etc)

- Dry coal feed 1/8" minus
- Steam added to control temperature below ash softening , however GTI U Gas and KRW piloted ash agglomerating mode
- Should operate at temperature sufficient to destroy tars yet not slag.
- Lower carbon conversion than entrained dependent on coal reactivity
- Main experience low rank coals. Poor carbon conversion with bituminous coal and pet coke
- Syngas contains CH<sub>4</sub> so not well suited for synthesis of Hydrogen, Ammonia, Methanol or DME but OK for SNG.
- Can be either air blown or Oxygen blown
- Need scale up in pressure and to commercial size
- Three velocity modes : Bubbling bed <3 fps (GTI U Gas, KRW), Circulating 8-16 fps (HT Winkler, GRI U Gas) and Fast or Transport (KBR) 20-45 fps.

# 3. Entrained-Flow Gasifier (GE, COP, Shell, Siemens, MHI etc)

Entrained-Flow Gasifier



# Entrained Flow Gasification

## - Main Features

- Operates in the slagging region 2400-3000F. Inert slag produced.
- No troublesome tars and very low CH<sub>4</sub> in single stage gasifiers
- Single stage gasifiers (GE, Shell) very suitable for Hydrogen, Ammonia, Methanol and F-T production
- In two stage gasifiers outlet temperature decreases and CH<sub>4</sub> content increases as more coal is fed to the second stage (COP, MHI).
- Slurry fed gasifiers can be run at up to 1000 psig (Eastman)
- Slurry fed gasifier (GE, COP) efficiency deteriorates and oxygen usage increases with high moisture and high ash coals
- Dry coal fed gasifiers (Shell, Siemens/FutureEnergy, Eagle, MHI) need pre drying of high moisture coals for reliable feed control but can handle a wide range of coals.
- GE, COP and Shell all proven at commercial size in IGCC plants.

# Gasification Process Selection

- Selection depends upon:
- Product/Application – Hydrogen, Synthesis (Ammonia, Methanol, Fischer-Tropsch liquids), SNG, Power only, Co-production or Polygeneration
- Coal types or range
- Overall Plant/Project Objectives
  - Lowest Cost-of-Electricity (COE) ?
  - Highest Efficiency? Lowest dispatch cost?
  - Maximum CO<sub>2</sub> capture?
  - Near Zero (Minimal) Emissions?
  - Lowest cost of product ?

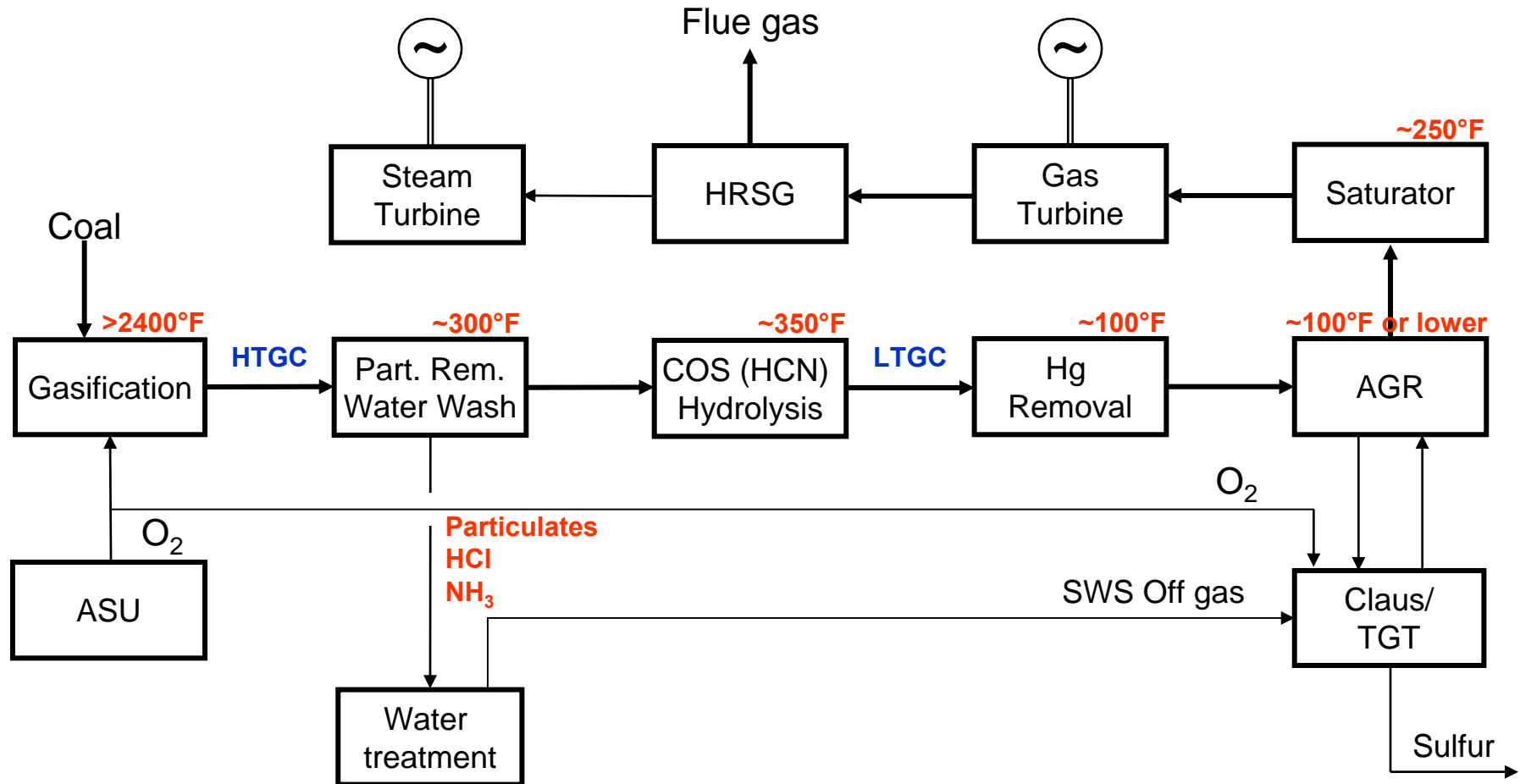
**No Single “Best” Gasifier – “Best” Depends on Project Requirements**



# IGCC/Gasification Status for CO<sub>2</sub> Capture

- GE gasifiers with full or partial water quench provide best CO<sub>2</sub> capture economics for bituminous coals
- COP partial slurry quench (PSQ) design (ala Wabash) includes some water quench and lower CH<sub>4</sub> production
- Shell gasifiers offer high efficiency over wide range of feed stocks. Water quench design is in development.
- Siemens gasifiers include partial water quench and handle wide range of feedstocks
- Selexol and Rectisol processes for CO<sub>2</sub> capture are commercial and proven
- Gas Turbine vendors currently offering gas turbines that are enabled for Hydrogen firing

# Simplified IGCC Block Flow Diagram

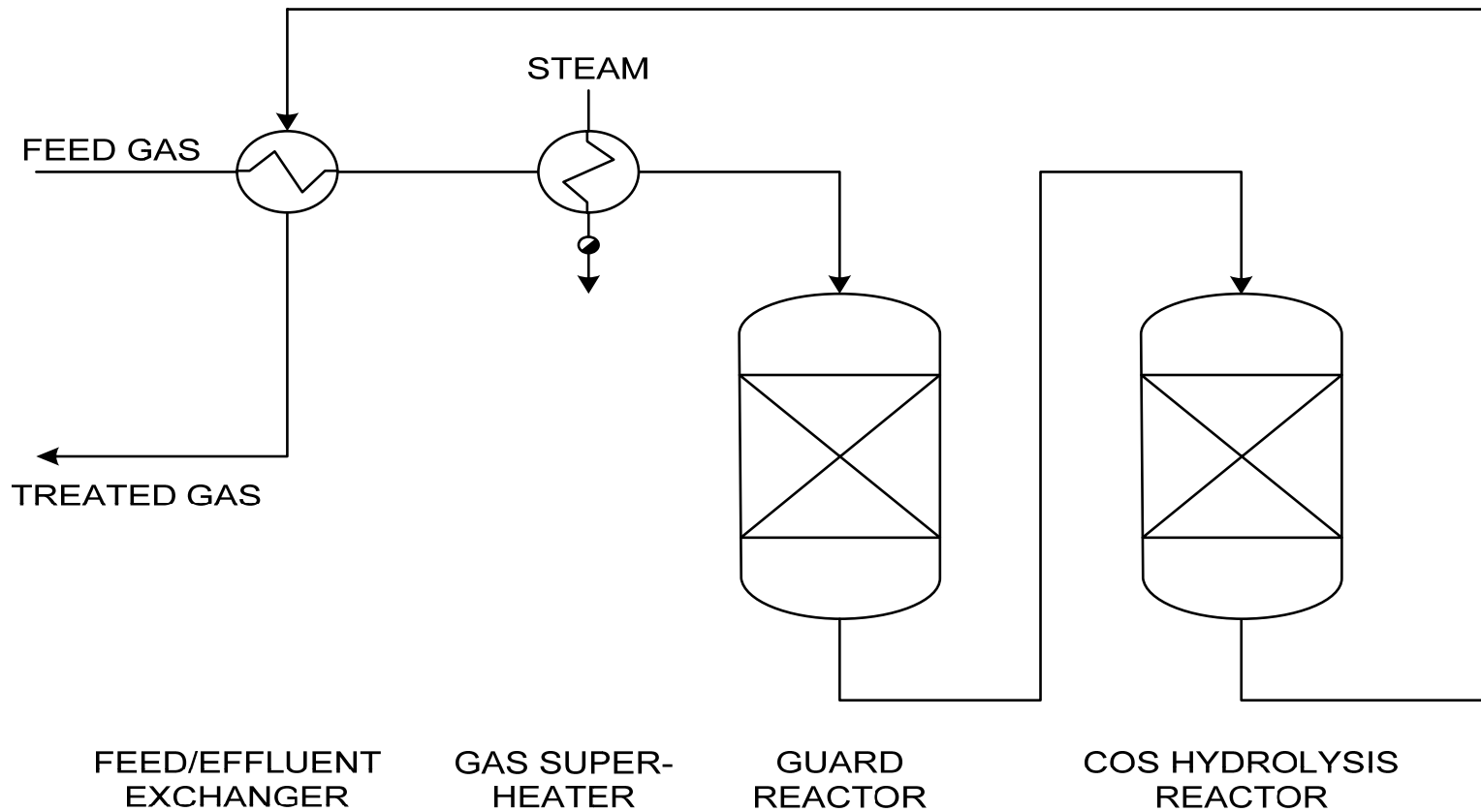


## IGCC Environmental Attributes

- Sulfur is removed (99.5-99.99%) from syngas
- NO<sub>x</sub> emissions are controlled by removal of nitrogen-containing species from the syngas and by flame temperature moderation in the gas turbine with a downstream SCR possible
- Particulates are removed from the syngas by filters and water wash prior to combustion so emissions are negligible
- Current IGCC design studies with SCR plan ~3ppmv each of SO<sub>x</sub>, NO<sub>x</sub> and CO
- Mercury and other HAP's removed from the syngas by adsorption on activated carbon bed
- Water use is lower than conventional coal
- Byproduct slag is vitreous and inert and often salable
- CO<sub>2</sub> under pressure takes less energy to remove than from PC flue gas at atmospheric pressure

# COS Hydrolysis / Simplified PFD

## Process Schematic



Source: EPRI UDBS-2007

# Water-Gas Shift Reactors

Pressure (psig)	
GE RQ	800
GE Q	1000
E-Gas FSQ	600
Shell GQ	600

H <sub>2</sub> O/CO Ratio	
GE RQ	1.3
GE Q	≥3.0
E-Gas FSQ	0.4
Shell GQ	0.1

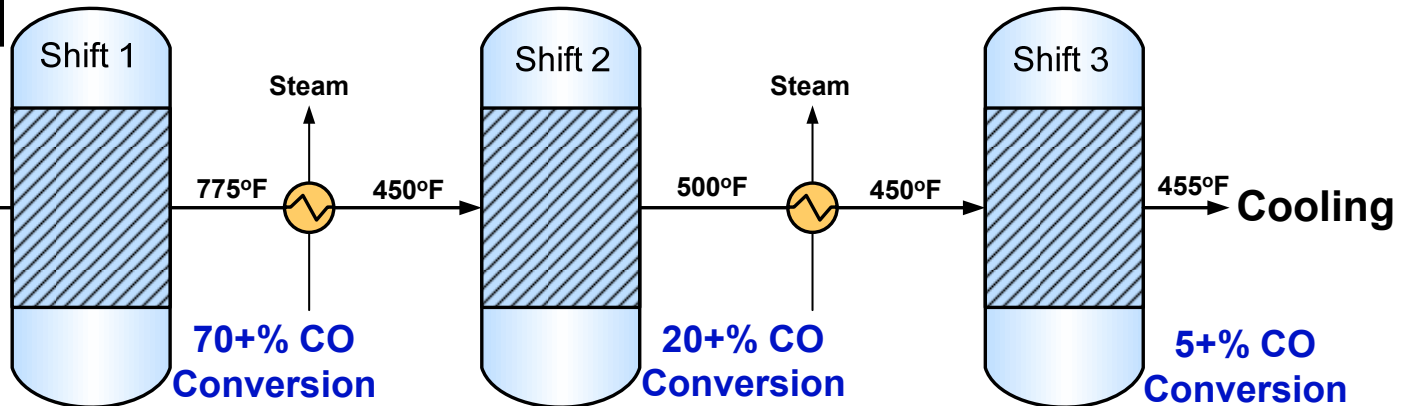
★ Shell WQ ~1.5

	Relative HP* Steam Flow
GE RQ	1.0
GE Q	0.0
E-Gas FSQ	2.0
Shell GQ	2.8

\*High Pressure Steam

## Design:

- Haldor Topsoe SSK Sulfur Tolerant Catalyst
- Up to 99% CO Conversion
- H<sub>2</sub>O/CO = 2.3 (Project Assumption)
- Overall ΔP = ~30 psia



Water-Gas Shift Reaction

★ Shell Water Quench design can substantially reduce HP Steam Flow

**Steam injection requirements have significant impact on plant performance**

# Mercury Capture in IGCC

- Capture on sulfur impregnated activated carbon
- Standard natural gas application for LNG plants
- In syngas service at Eastman Chemical Co. since 1984
- Capture rate ~94% of vapour phase mercury
- Bed life ~ 2 years
- Spent carbon to hazardous landfill



Source: Trapp, 2002

# IGCC CO<sub>2</sub> Capture Technologies

## *Current & Developmental Systems*

### Currently Available Technologies

- MDEA
- UOP **Selexol**
- Linde **Rectisol**

### Developmental Physical Solvent Absorption

- GTI & Uhde – **Morphysorb** Solvent Absorption

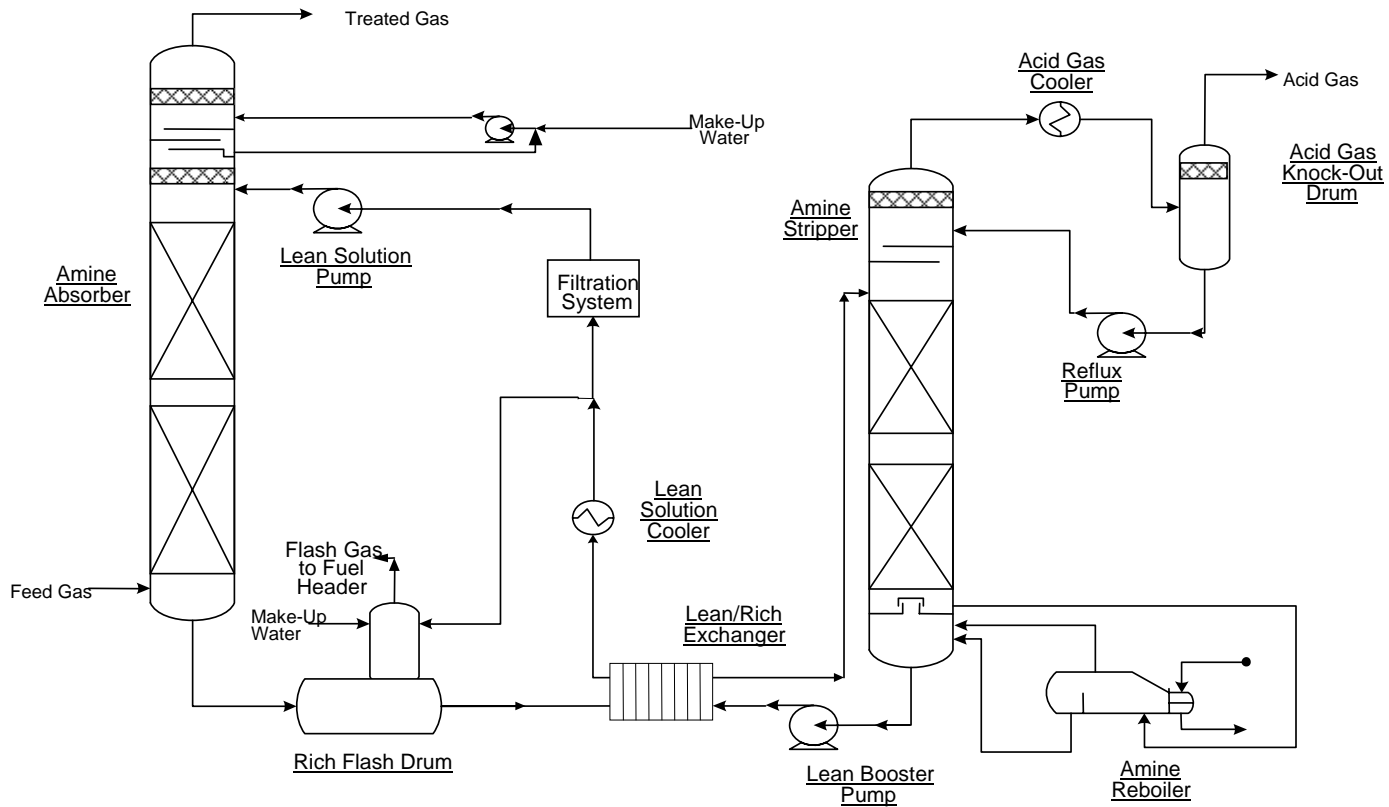
### Developmental Membrane CO<sub>2</sub>/H<sub>2</sub> Separation

- NETL & Eltron **H<sub>2</sub> Transport Membranes (HTM)**
- RTI & Air Liquide **Reverse Selective Polymeric Membranes**

### Other Developmental Technologies (NETL + Other Research Organizations)

# MDEA Process for IGCC

## Process Flow Diagram: H<sub>2</sub>S Capture Only

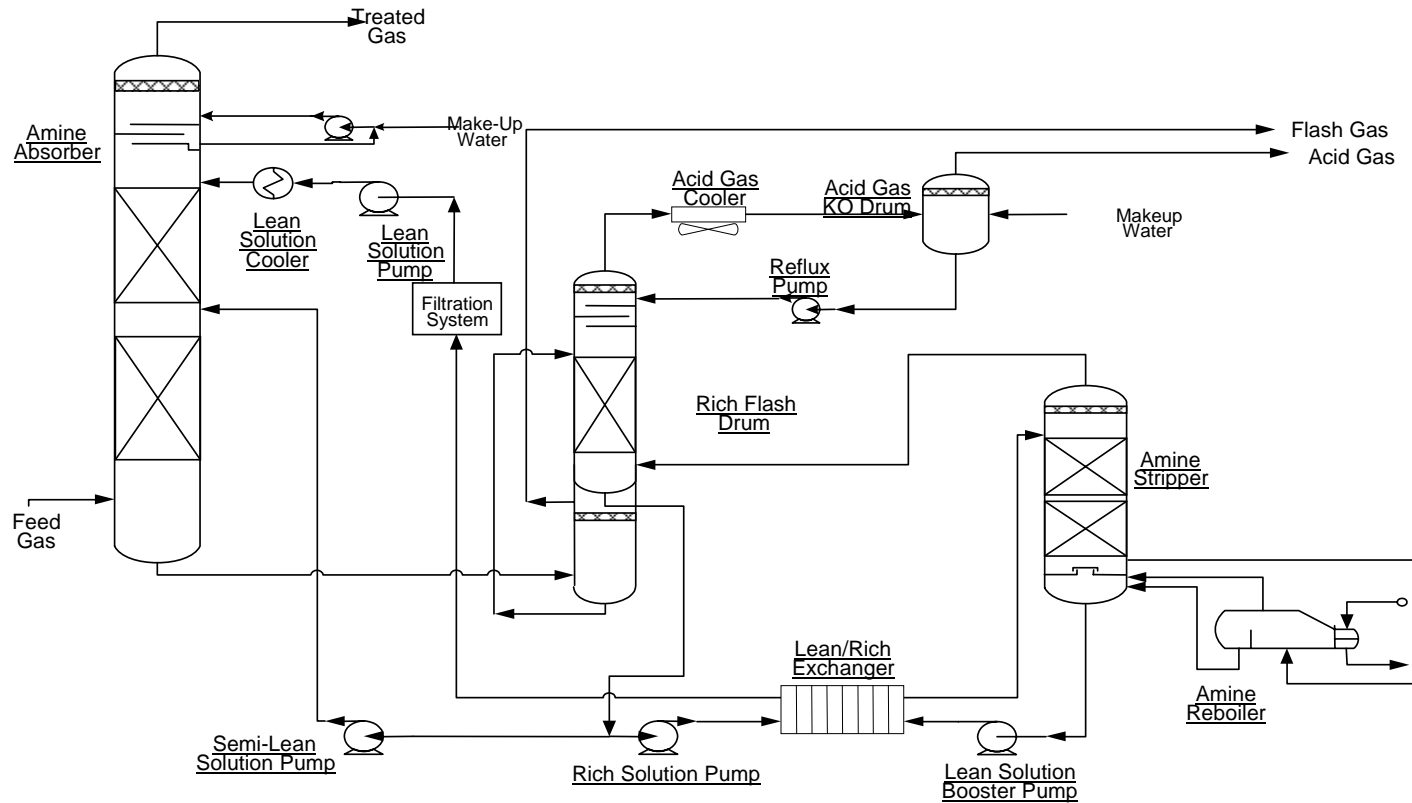


**AMINE-H<sub>2</sub>S PROCESS FLOW DIAGRAM**



# MDEA Process for IGCC

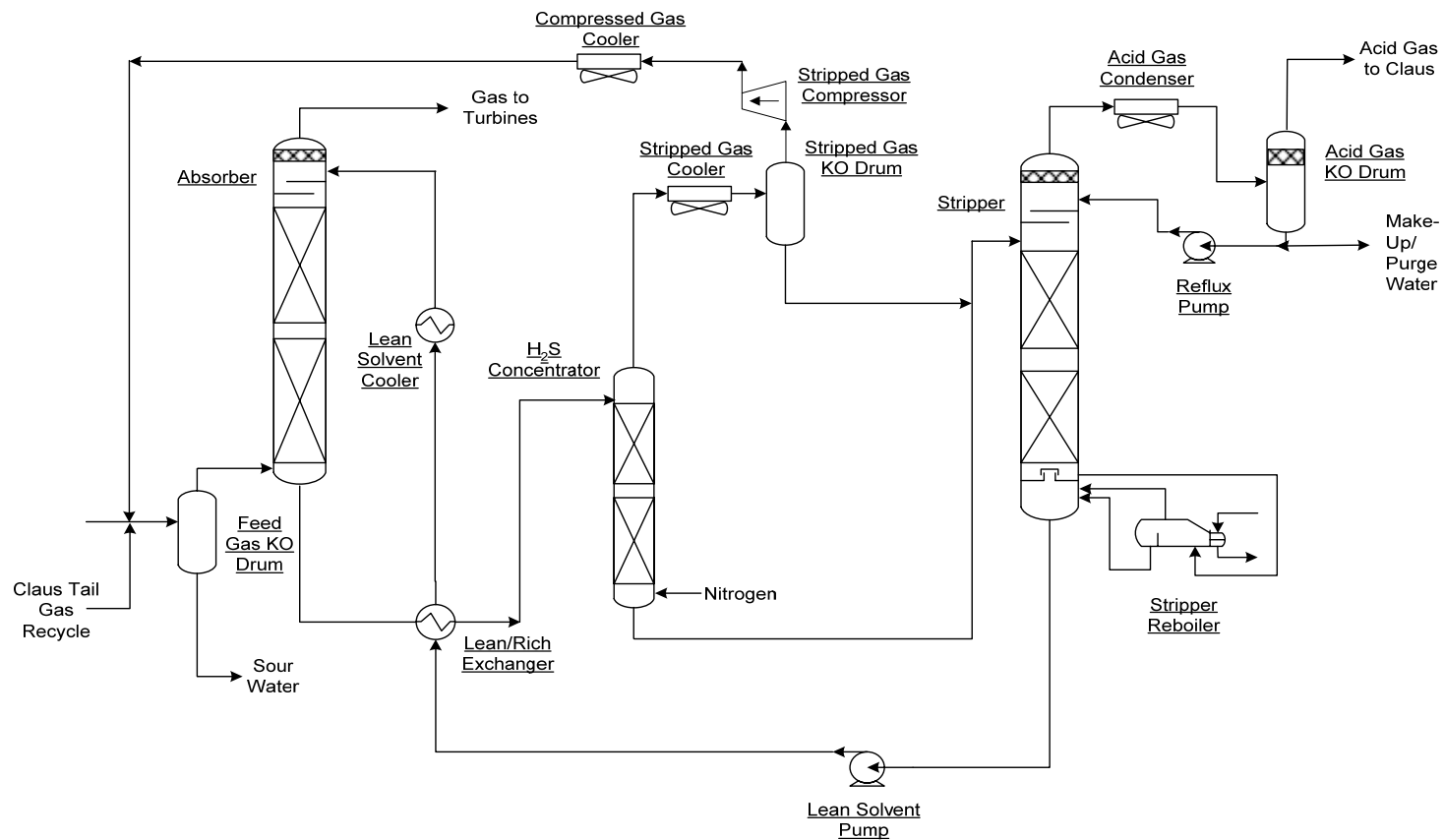
## Process Flow Diagram: H<sub>2</sub>S and CO<sub>2</sub> Capture (Low energy design)



LOW ENERGY AMINE-CO<sub>2</sub> PROCESS FLOW DIAGRAM

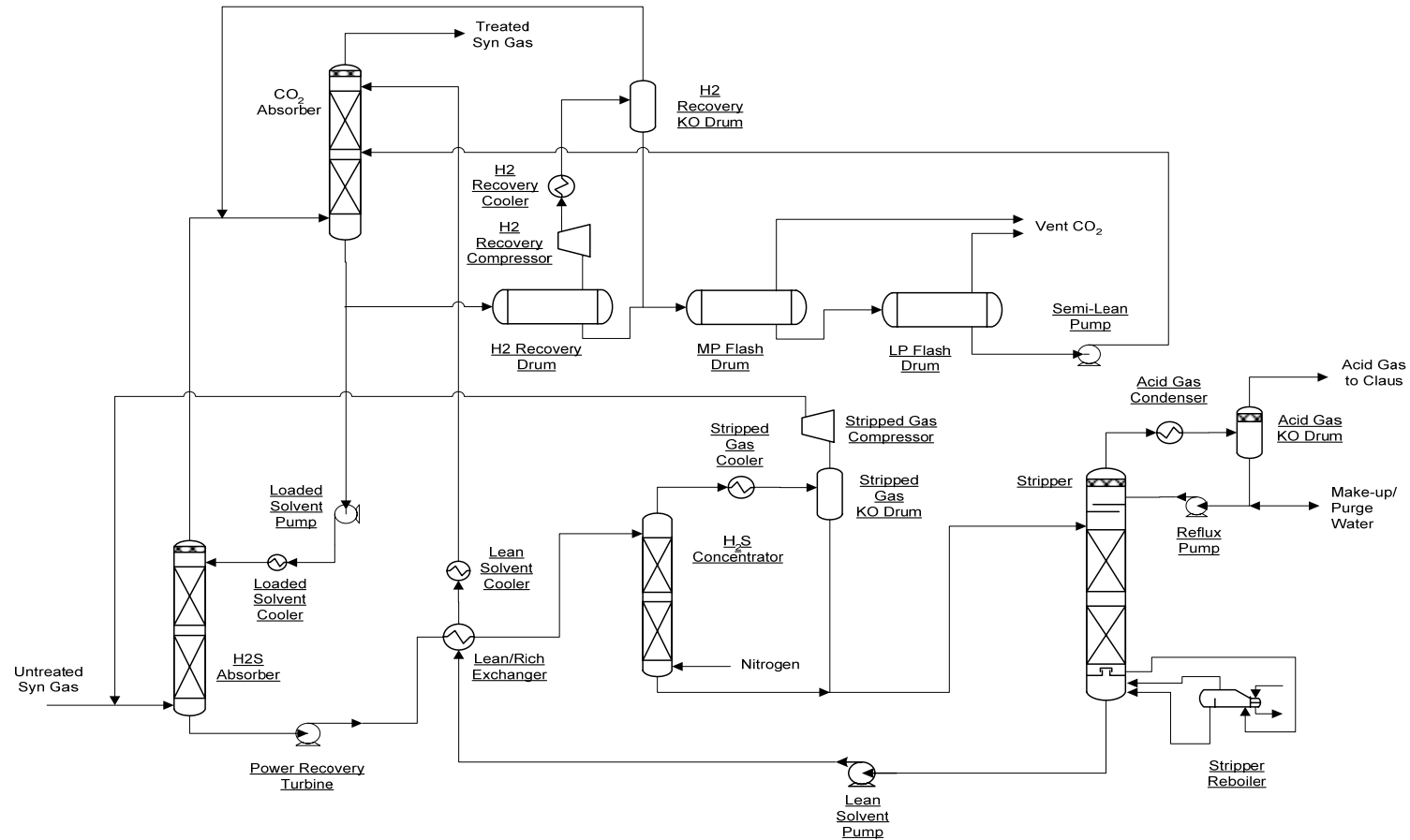
# UOP Selexol Process for IGCC

## Process Flow Diagram: H<sub>2</sub>S Capture Only



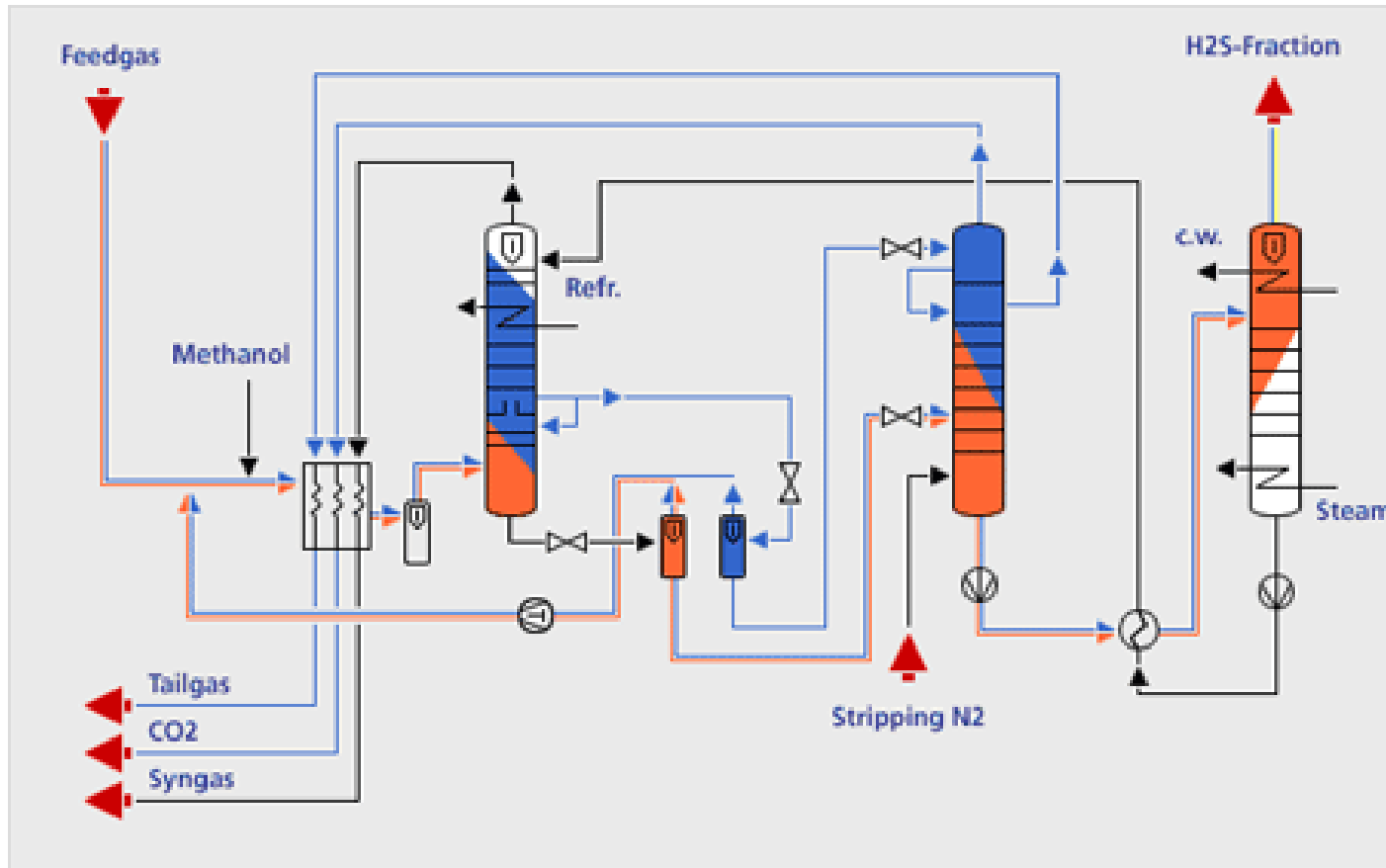
# UOP Selexol Process for IGCC

## Process Flow Diagram: H<sub>2</sub>S and CO<sub>2</sub> Capture (4 tower design)



# Linde Rectisol for IGCC

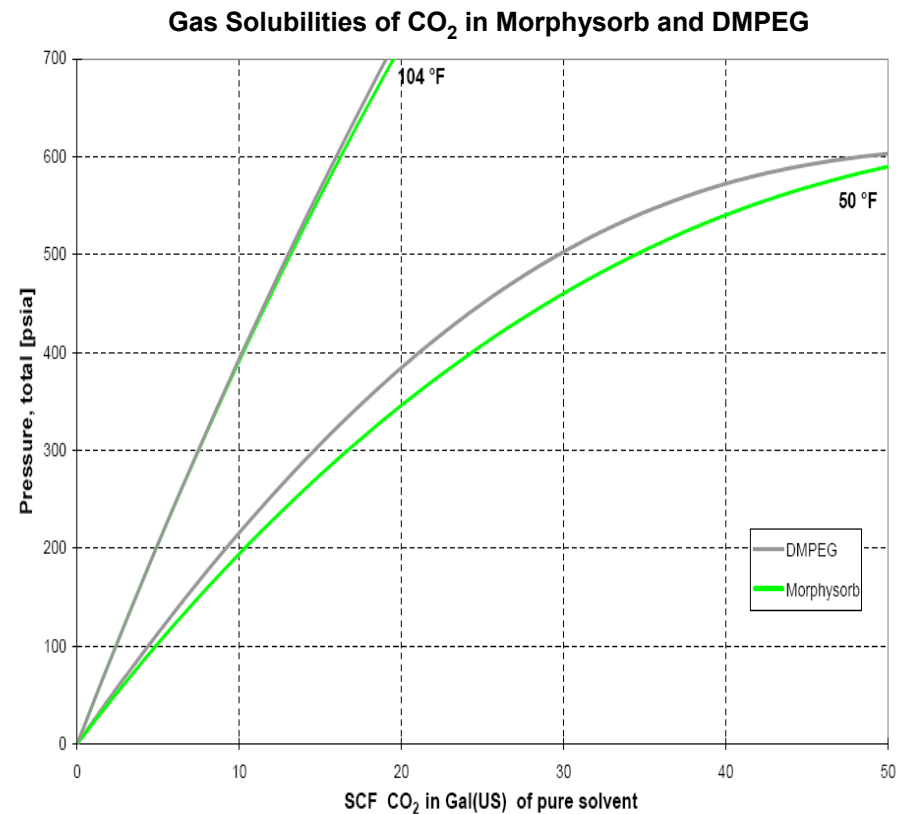
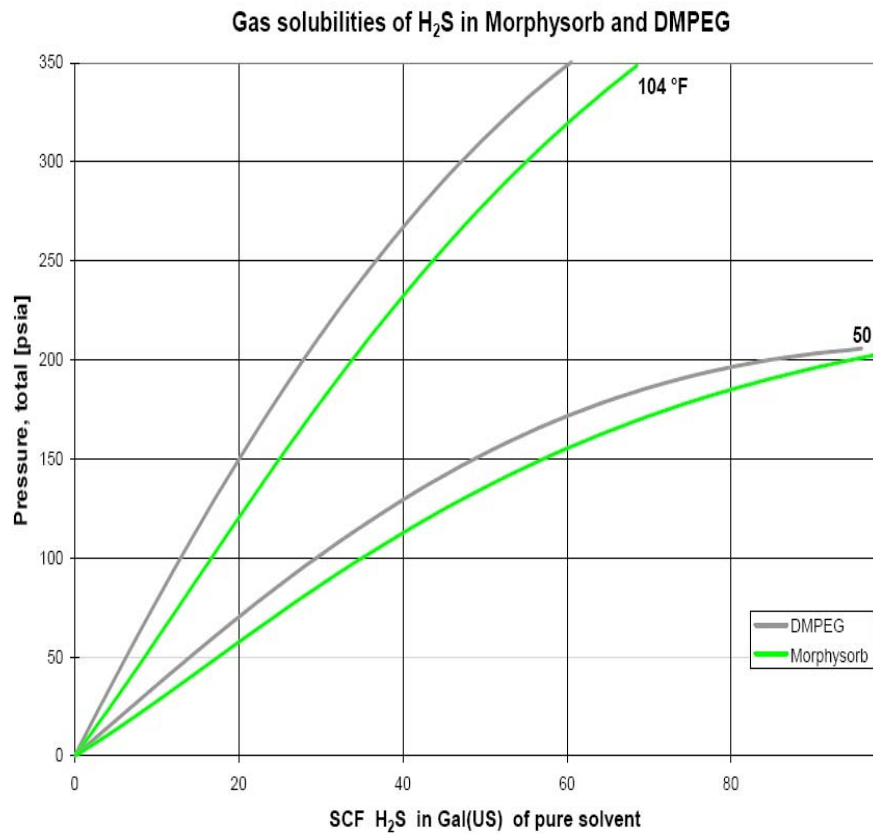
## Process Flow Diagram: H<sub>2</sub>S and CO<sub>2</sub> Capture (3 tower design)



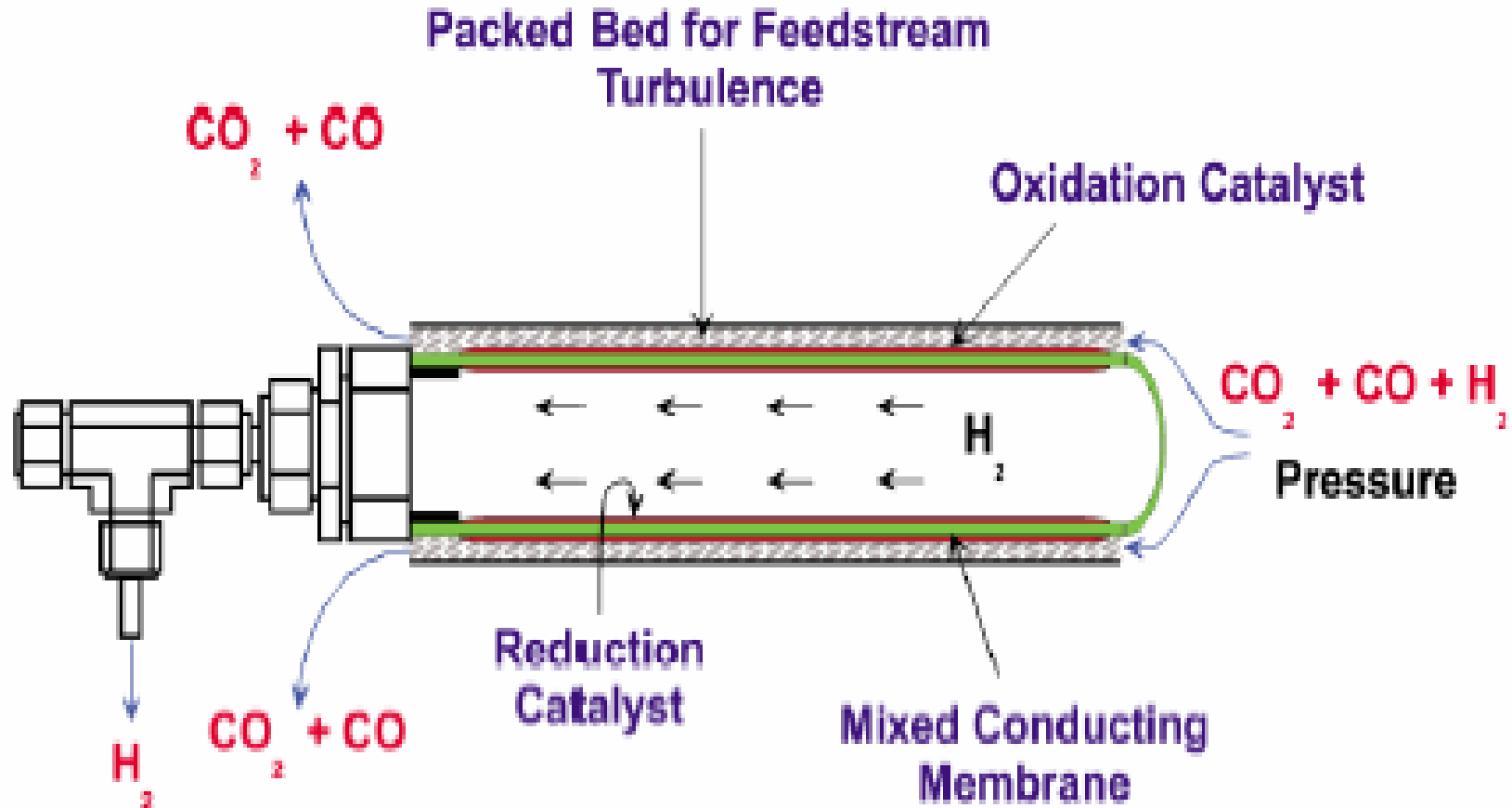
Source: [www.linde-anlagenbau.de/process\\_plants/hydrogen\\_syngas\\_plants/gas\\_processing/rectisol\\_wash.php](http://www.linde-anlagenbau.de/process_plants/hydrogen_syngas_plants/gas_processing/rectisol_wash.php)

# Morphysorb Solvent Process

## $H_2S$ & $CO_2$ Solubility – Morphysorb vs. Selexol (DMPEG)



# Hydrogen Ion-Transport Membranes *Process Drawing*



Source: Eltron 2005

# Reverse Selective Polymer Membranes Technology Description

## Opportunity

### Hydrogen production

- Steam reforming or gasification of hydrocarbon fuels, followed by water gas shift reaction

## Problem

H<sub>2</sub> product contaminated with CO<sub>2</sub> byproduct and other polar, acid gases (H<sub>2</sub>S, COS, etc.)

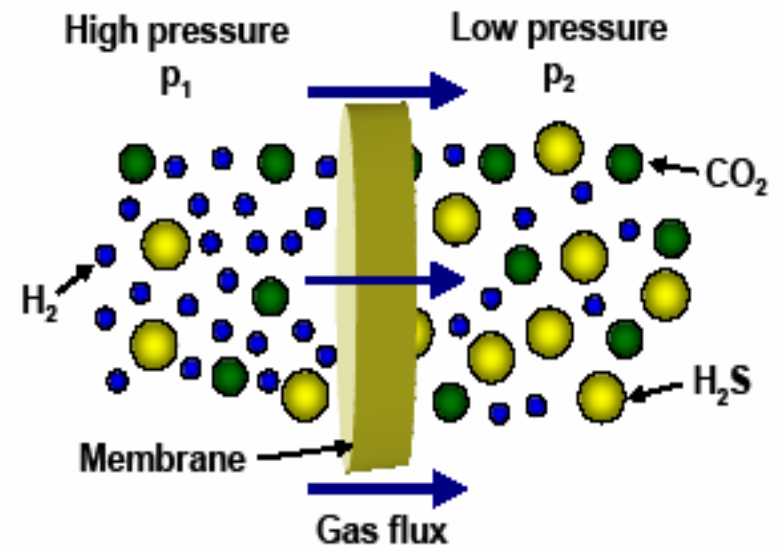
## Technical approach

Reverse-selective membranes for H<sub>2</sub> purification

- Bulk removal of acid gases (CO<sub>2</sub>, H<sub>2</sub>S)
- Selective permeation of larger acid gases over smaller H<sub>2</sub>

## Project Team

- RTI International
- The University of Texas at Austin (Academic partner)
- MEDAL, L.P./Air Liquide (Industry partner)
- DOE/NETL (Federal government support)



# Development Activity for IGCC

## *Key Long-Term Research Projects*

- **Stamet pump** for dry coal feed
- **AP ITM Oxygen** process (note: single point failure, no other ASU improvements being funded, need more processes)
- Advanced sulfur removal processes (high temperature or low energy)
- Advanced H<sub>2</sub> and **CO<sub>2</sub> separation** processes (Solvents or Membranes)
- Advanced H<sub>2</sub>-fired combustion turbines (GE and Siemens are current contractors for DOE work)
  - Results need to be extended to H-class CTs
- RamJet CO<sub>2</sub> Compressor (heat recovery improvement)
- Fuel Cell development
- FutureGen
  - Should provide first opportunity to test many of the items above at reasonable scale and realistic conditions



## R&D Work Underway

- SRI International, RTI, & NETL working to quantify the impact of contaminants on fuel cell operation
- TDA Research is researching a catalyst that can remove Arsenic at high temperatures that may also remove some Selenium
  - Testing of system at PSDF with Southern Company should be underway
  - Removal of As and Se would be beneficial for GT and WWT
- Others...?

EPRI may be able to provide support: Contact us with your ideas

# Questions and Discussion

