

In Situ Device for Real-Time Catalyst Deactivation Measurements*

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*** Patent Pending**

Project Overview

Demonstrate the viability of an in situ approach of monitoring catalyst activity in coal-fired SCR reactors

- Design and install a prototype in situ system at a host utility site
- Conduct test program to characterize deactivation rate for individual catalyst layers over two ozone operating seasons
- Utilize host unit operator's annual O&M catalyst sampling/analysis activities to validate the in situ deactivation measurements

Project funding

- U.S. Department of Energy (NETL)
- Electric Power Research Institute
- Southern Company

Catalyst Activity and Reactor Potential

Catalyst Activity (K)

- Measure of how active the catalyst material is for reducing NO_x

Reactor Potential (RP)

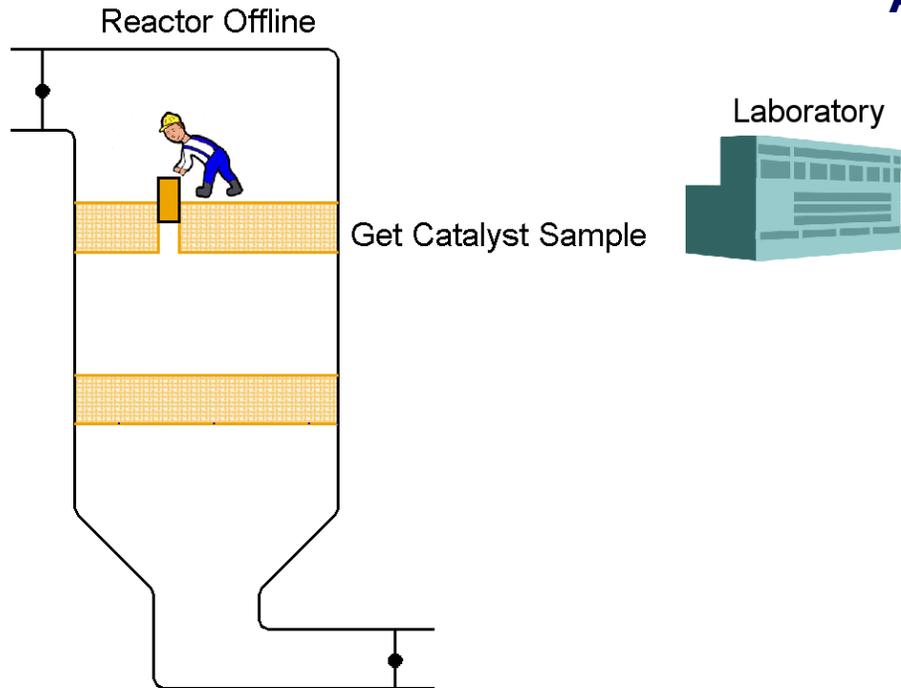
- Measure of the overall potential of the reactor to reduce NO_x

- $RP = K/A_{V,FS}$ $A_{V,FS}$ = full scale area velocity (blockage)
 $= \frac{K(1-B)}{A_{Vd}}$ B = catalyst blockage
 A_{Vd} = design area velocity (zero blockage)

- Inherently accounts for both catalyst deactivation (K/K_0) as well as catalyst layer blockage, thereby providing a true assessment of the condition of the SCR reactor
- RP is the quantity utilized by catalyst management software programs such as EPRI's CatReact product in the evaluation of various catalyst management strategies

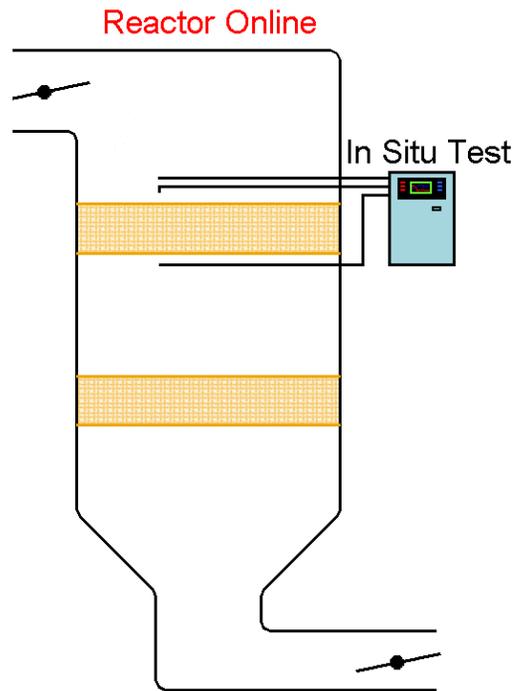
Laboratory Determination of Reactor Potential

- A catalyst sample is tested at the design A_V with $\text{NH}_3/\text{NO}_x = 1$, and then the activity determined from $K = -A_{Vd} \cdot \ln(1 - \Delta\text{NO}_x)$
- An estimate of the blockage in the full scale reactor is then used to calculate the reactor potential: $\text{RP} = \frac{K}{A_{Vd}}(1-B)$

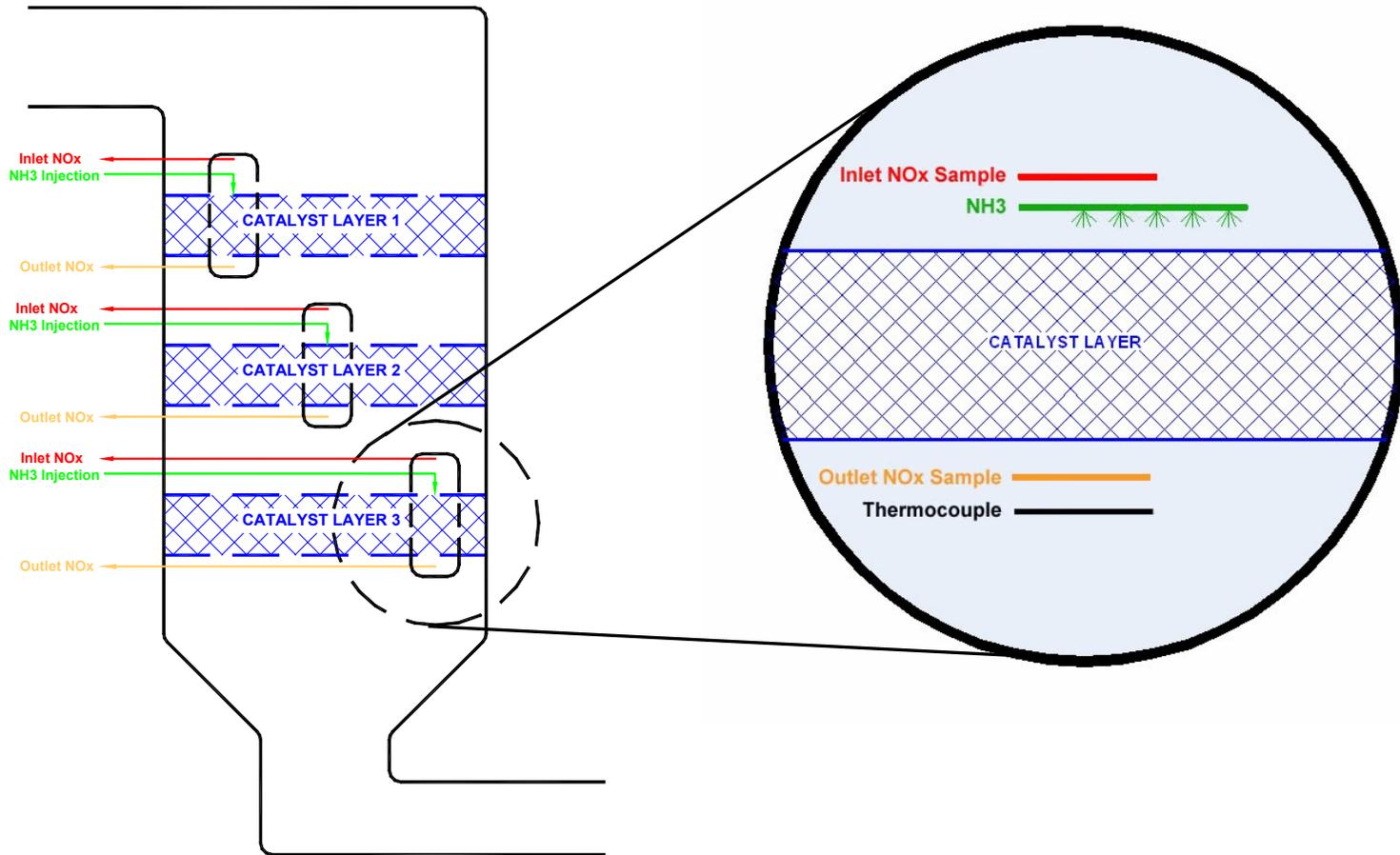


In Situ Reactor Potential

- If $RP = K/A_{V,FS}$ and $K = -A_{V,FS} \cdot \ln(1 - \Delta NO_x)$, then $RP = -\ln(1 - \Delta NO_x)$
- Thus, an in situ measurement of the maximum NO_x removal across the layer yields a direct measurement of the reactor potential



Multiple *In-Situ* Test Modules - General Approach



Host Site Description

Demonstration host site provided by Southern Company

- **Alabama Power Company's Gorgas Unit 10**
 - **700 MW**
 - **Twin-furnace, tangentially-fired**
 - **Started operation 1972**
 - **Alabama bituminous coal**
- **SCR on-line May 2002, ozone season operation**
 - **Two reactors**
 - **3 + 1 configuration**
 - **Anhydrous NH₃**
 - **Initial load: three layers of extruded honeycomb catalyst**
 - **Fourth layer of plate catalyst added prior to 2006 ozone season**

Equipment Overview

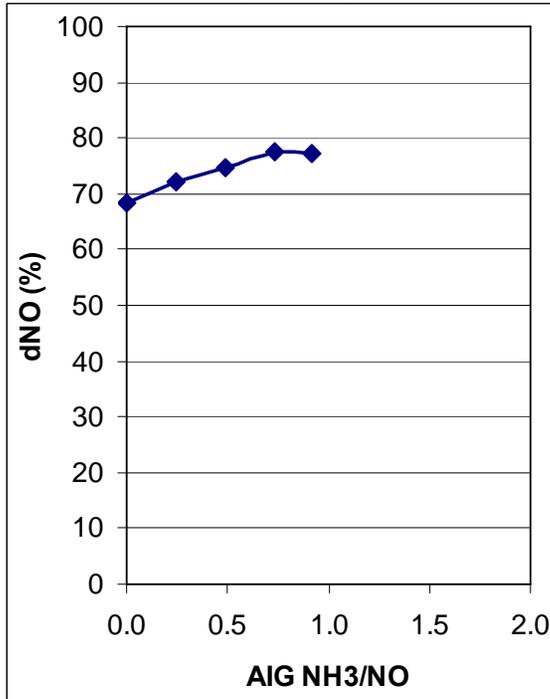
- Stationary NEMA 4 enclosure housing multi-position sampling valves
- Portable “cart” housing gas analysis instrumentation and control system



In Situ Test Protocol and Typical Test Results

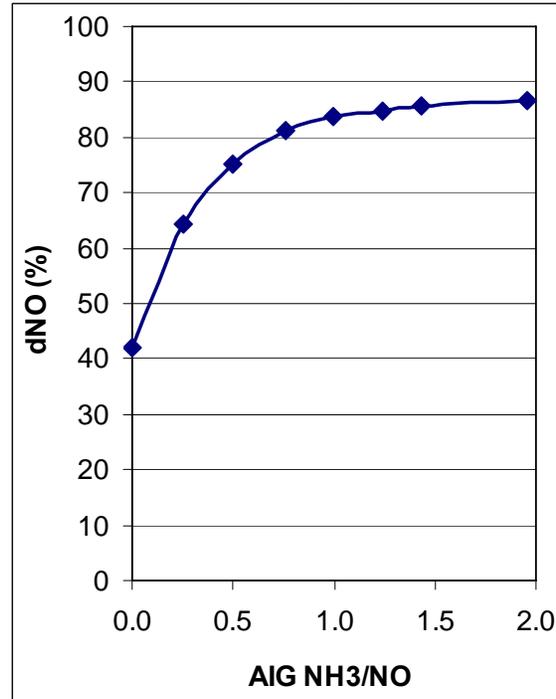
- Measure NO removal across test module without NH₃ injection
- Add NH₃ via test module AIG to point of maximum NO removal
- Calculate reactor potential from $RP = -\ln(1 - \Delta NO)$

Catalyst Layer 1



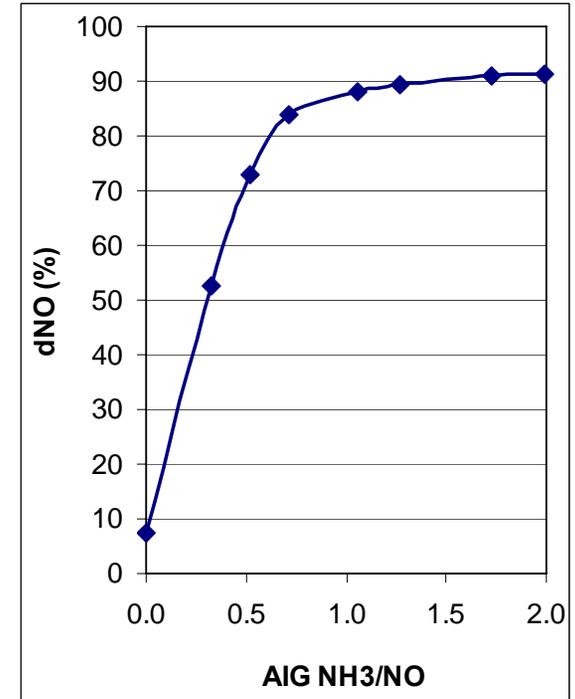
- dNO maximum = 77.3%

Catalyst Layer 2



- dNO maximum = 86.6%

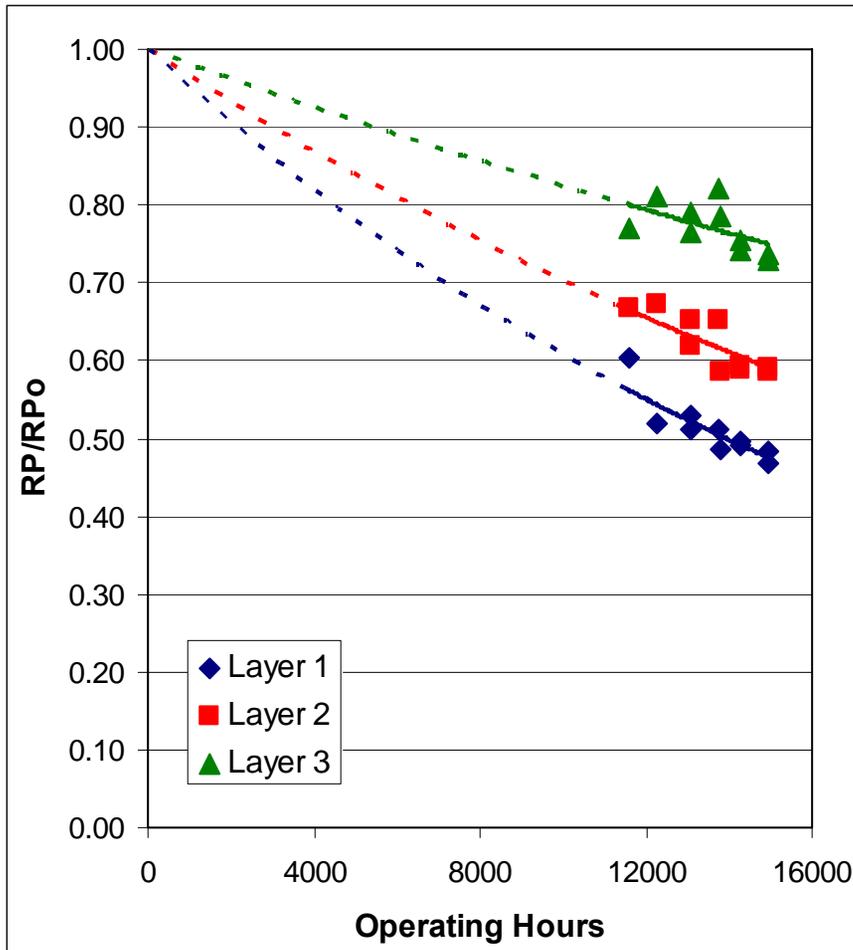
Catalyst Layer 3



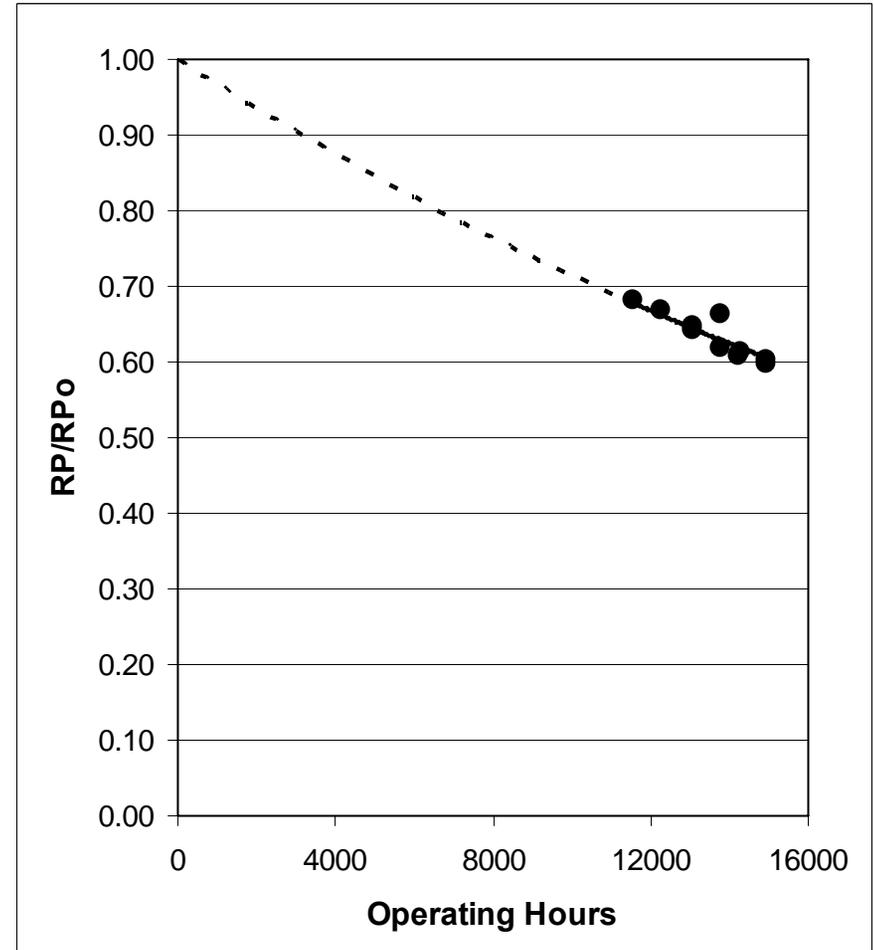
- dNO maximum = 91.3%

In Situ Reactor Potential Results - 2005

Individual Layers



Overall Reactor



Blockage Estimation for Laboratory RP Calculation

End of 2005 Ozone Season

Source of Estimate	Blockage Estimate		
	Layer 1	Layer 2	Layer 3
Plant visual inspection	20%	30%	25%
Calculate from change in layer pressure drop from new (clean)	40%	24%	n/a

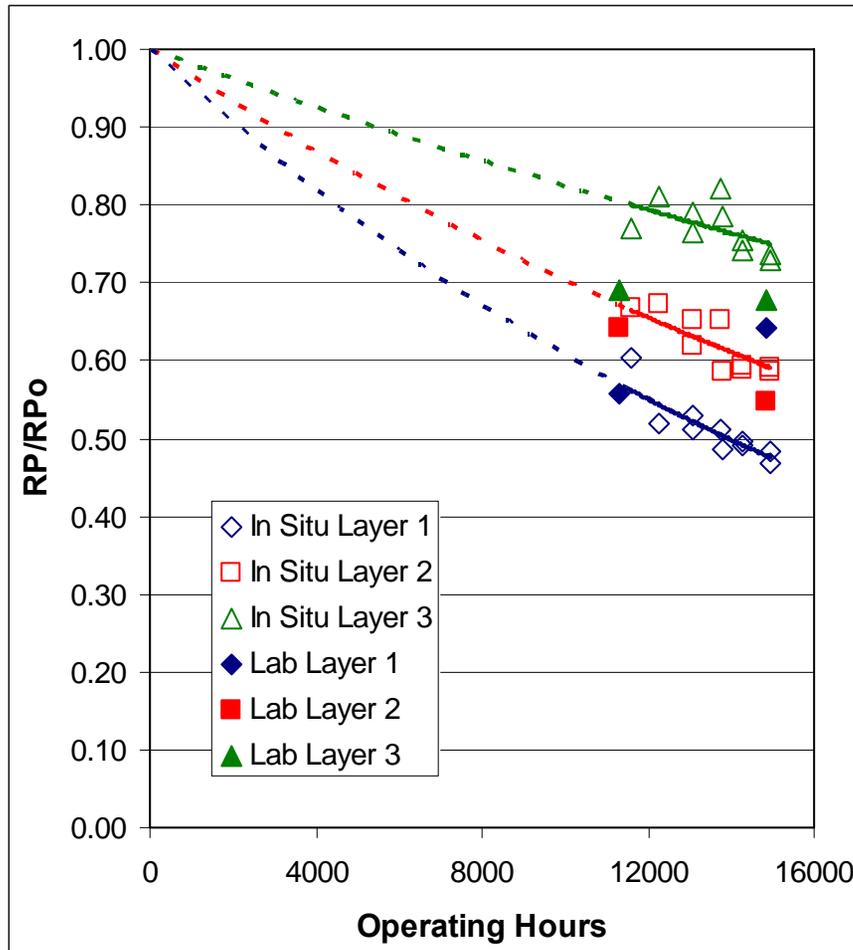
$$RP = \frac{K}{A_v} (1-B)$$

Question: What would B need to be to make the laboratory RP value equal to the in situ value?

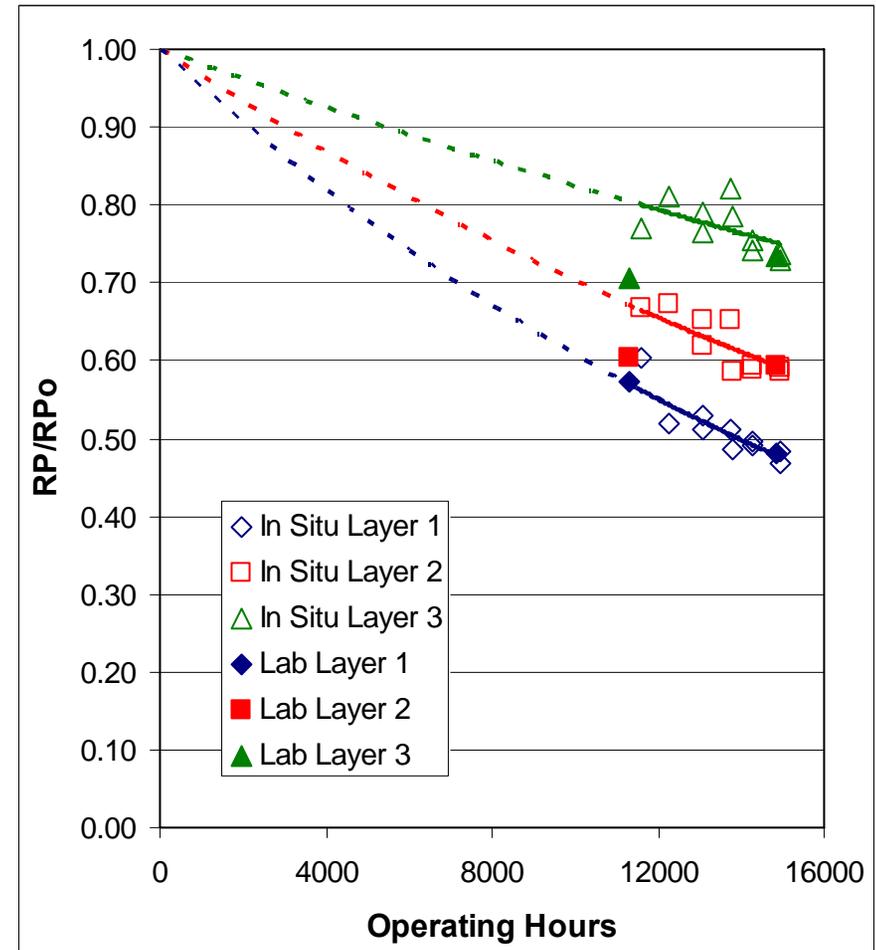
	Layer 1	Layer 2	Layer 3
Calculate blockage value from in situ RP and laboratory K/A_v	41%	24%	19%

Laboratory RP/RP₀ for Individual Catalyst Layers

Visual Blockage Estimate



Blockage Calculated from dP



Why do the In Situ and Laboratory RPs Differ?

Laboratory technique measures K at the design A_v

- Estimated blockage needed to calculate RP may be in error
- Actual flue gas flow may differ from the design flow

In Situ technique measures RP for the same piece of catalyst

- Laboratory uses a different piece for each test
- Possible variations in K across the catalyst layer

Operating Hours	K/K_0 Layer 1	K/K_0 Layer 2	K/K_0 Layer 3
11317 (end 2004)	0.71	0.75	0.79
14870 (end 2005)	0.80	0.78	0.90

In Situ tests at $NH_3/NO_x \sim 1.5$, laboratory tests at $NH_3/NO_x = 1.0$

- Increasing NO_x removal at $NH_3/NO_x > 1$ will yield higher RP, but the RP/RP_0 should be the same.

Comparison of Measurement Techniques

Laboratory	In Situ
<p><u>Pro</u></p> <ul style="list-style-type: none"> ● Accurate K determination ● Physical sample can also be analyzed for physical and chemical properties 	<p><u>Pro</u></p> <ul style="list-style-type: none"> ● Direct measurement of RP ● Larger data set ● Can test immediately after unit “upset”
<p><u>Con</u></p> <ul style="list-style-type: none"> ● Relies on an estimate of blockage to calculate RP ● Limited data set (outages only) 	<p><u>Con</u></p> <ul style="list-style-type: none"> ● Need blockage to calculate K ● ΔNO_x increase at $\text{NH}_3/\text{NO}_x > 1$ may yield slightly higher absolute RP values

The in situ technique should not be thought of as a replacement for laboratory analysis of catalyst samples, but as a companion measurement