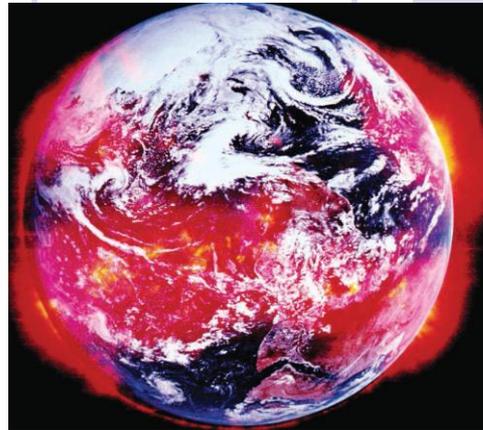




**2nd U.S.-China Symposium on CO₂ Emission Control
Science & Technology, Hangzhou China
28th -30th, May, 2008**

Status of CCS in China



Prof. Zhongyang Luo

**State Key Laboratory of Clean Energy Utilization
Zhejiang University**

29th May, 2008

⊕ Global CO₂ Emissions

Country	CO ₂ Emissions (Million Tons Carbon)			
	1990	1997	2001	2010
USA	1345	1480	1559	1800
<i>China</i>	<i>620</i>	<i>822</i>	<i>832</i>	<i>1109</i>
Former USSR	1034	646	654	825
Japan	274	297	316	334
World	5836	6175	6522	8512

Source: Energy Information Administration/International Energy Outlook 2001

⊕ Global CO₂ Emissions from Fossil Fuel Use in 2006

Country	CO ₂ Emissions (1 million metric tons CO ₂)	Percentage (%)
Total	28,160	100
USA	5,750	20.42
China	5,680	20.17
EU-15	3,330	11.72
Russia	1,620	5.75
Japan	1,210	4.3

But, according to the **Netherlands Environmental Assessment Agency**, it said that **China** produced **6,200 million tonnes of CO₂** in 2006, compared with **5,800 million tonnes** from the **USA**. **Britain** produced about **600 million tonnes**.

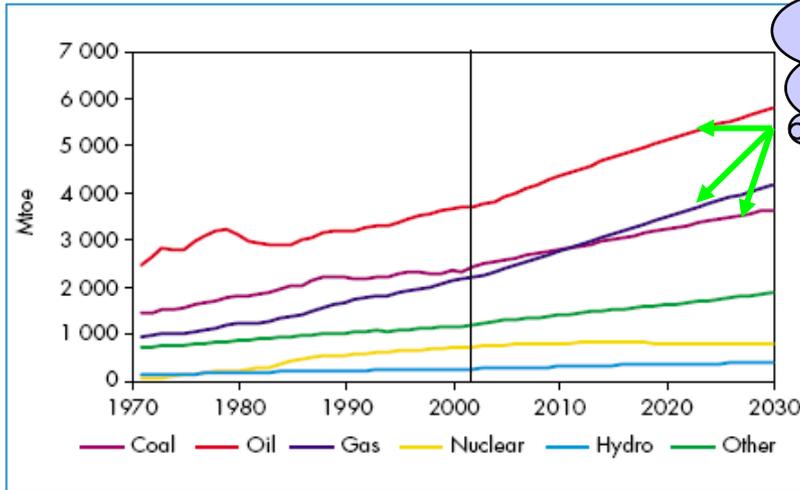
BP Statistical Review of World Energy, June 2007
<http://www.bp.com/sectiongenericarticle.do?categoryId=6914&contentId=7042803>

<http://www.mnp.nl/en/dossiers/Climatechange/moreinfo/Chinanowno1inCO2emissionsUSAinsecondposition.html>



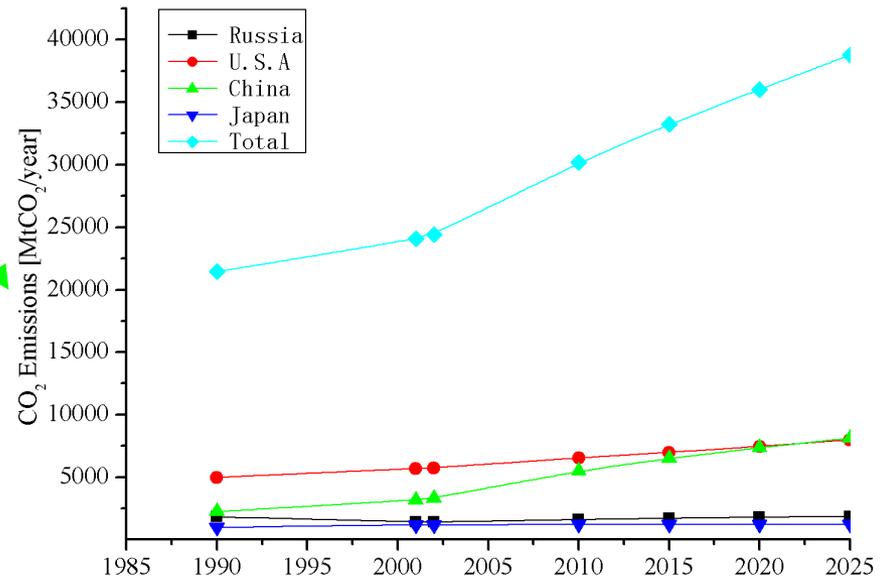
Global CO₂ Emissions in the Future

World Primary Energy Demand by Fuel



Fossil fuel will continue to dominate the global energy mix.

Global increase of total CO₂ emissions



World Primary Energy Demand by Fuel from now to 2030 (*IEA Report, 2004*)

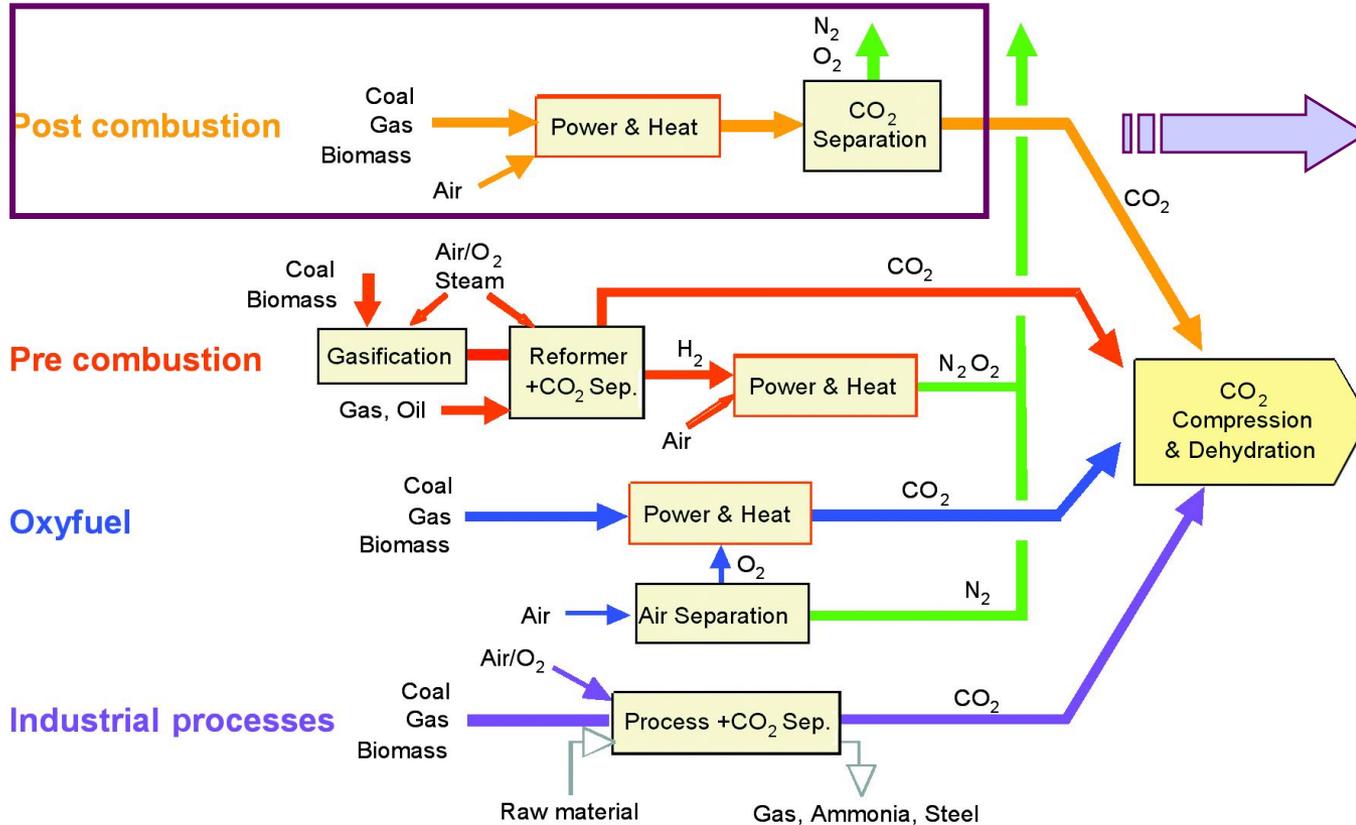
CO₂ Emission in China

Year	Total	Coal		Petroleum		Natural Gas	
	Million Tons Carbon	MtC	%	MtC	%	MtC	%
1990	620	514	82.9	98	15.8	8	1.29
1996	801	625	81.4	138	17.2	11	1.37
1997	822	661	80.4	148	18	12	1.46
2001	832	668	80.3	151	18.2	15.5	1.86
2010	1,457	1,115	76.5	277	19	65	4.46

In China, CO₂ capture from the coal-fired flue gases will be an effective way to control CO₂ emission in short term.

How to Control CO₂ Emission?

Overview of CO₂ capture processes and systems



CO₂ capture in post combustion mode might be the best method to control CO₂ emission in the short term.

Resource: IPCC Special Report about CO₂ Capture and Storage, 2001.

I Post-combustion CO₂ Capture Technologies

- **Absorption Methods**

Chemical absorption and physical absorption

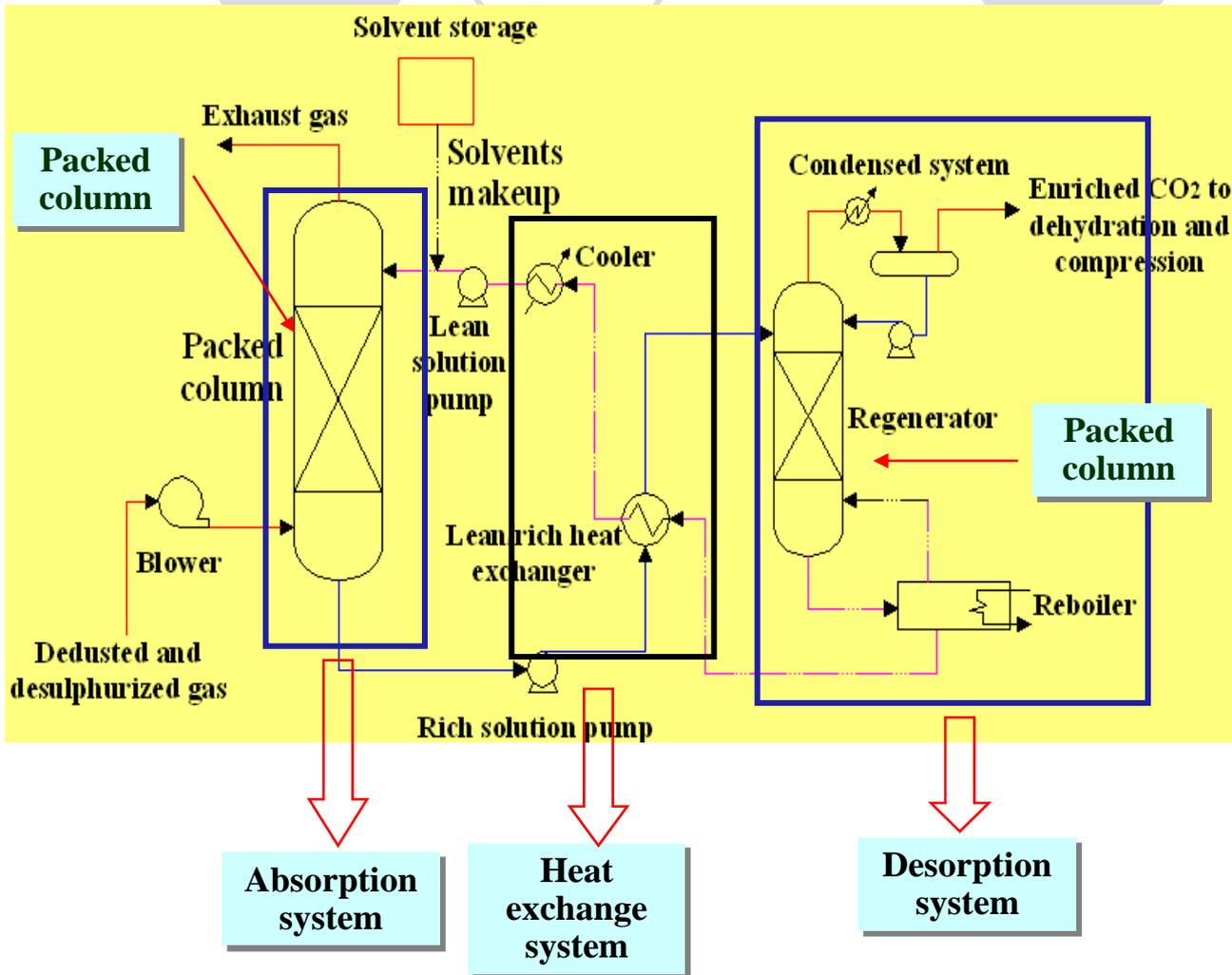
- **Membrane technologies**

Membrane separation and membrane gas absorption

- **Adsorption methods**

- **Cryogenic processes**

⊕ 1. CO₂ Capture by Chemical absorption Method



The CO₂ absorption process generally consists of (1) **an absorption unit** where CO₂ is recovered from a gas phase into a liquid solvent and (2) **a heat exchange system** where rich solution is heated by lean solution before being pumped into desorption unit, and lean solution is cooled to a specific temperature value before being pumped into absorption unit; (3) **a desorption unit** where the absorption capability of the used solvent is recovered before being reintroduced to the absorption unit.

Typical amine absorption process for CO₂ recovery from flue gas

Research on chemical absorption in China

In China, so many institutes are now paying their attention on CO₂ separation using chemical absorption method, such as Najing University, Tsinghua University, Beijing University of Chemical Technology, Zhejiang University, Tianjin University, Nanjing University of Science and Technology, Huazhong University of Science and Technology, CAS and so on.

The research keystone:

- **New Solvents** (Provide equivalent or greater CO₂ absorption rates than MEA, adequate capacity for CO₂ and reduced cost of regeneration);
- New Packing or New Absorber;
- Kinetics and Dynamics of Solvents;
- Effects of some operating conditions on CO₂ removal performance.

Research on Absorbents by Chemical Absorption in China

Institute [↵]	Reactors [↵]	Absorbents [↵]	Gas [↵]	Some Results [↵]
Zhejiang University [↵]	Stirred-cell Absorber/Packed column [↵]	Blended absorption solution MEA, MDEA, PZ, Ammonia, PG, AMP, DEA, DETA and TETA [↵]	CO ₂ +N ₂ [↵]	Absorption rate and capacity of CO ₂ can be remarkably enhanced by the system of aqueous MEA/MDEA and MDEA/PZ solutions. [↵]
Tsinghua University [↵]	Agitated kettle [↵]	Blended absorption solution of MDEA + <u>Sulfolane</u> [↵]	CO ₂ [↵]	Some kinetic data for absorption of CO ₂ with the blended solvents of MDEA and <u>sulfolane</u> aqueous solution. [↵]
	Sieve-plate reactor [↵]	Ammonia... [↵]	CO ₂ +N ₂ [↵]	Effects of operating parameters on CO ₂ removal efficiency; Analysis on dynamics. [↵]
Haerbin Engineering University [↵]	Disc column [↵]	Blended amine aqueous solution of TEA+DETA [↵]	CO ₂ [↵]	Absorption rate and CO ₂ absorption capacity of blended amines [↵]
	Stirred absorption desorption apparatus [↵]	MEA, DEA, MDEA and AMP [↵]	CO ₂ [↵]	CO ₂ absorption and desorption rate of four kinds of solvents. [↵]
	Stirred-cell reactor [↵]	Seawater and MEA, DEA, TEA, MDEA, AMP, DETA, TETA, TEPA [↵]	CO ₂ [↵]	Using seawater and amines to form solvents can improve the CO ₂ absorption performance. [↵]

Research on Absorbents by Chemical Absorption in China

Institute	Reactors	Absorbents	Flue gas	Some Results
Shanghai <u>Jiaotong</u> University	—	MDEA/ PZ and MDEA/DEA	CO ₂	CO ₂ absorption rate using MDEA/PZ or MDEA/DEA; the kinetic data of these solvents
Nanjing Agricultural University	Stirred-cell reactor	Aqueous MEA solution	CO ₂	PH was measured during the absorption of CO ₂ into aqueous of MEA at different conditions; Discussing the relationship between absorption time and depletion of amine concentration at different temperature and different pressure.
Anhui University of Science and Technology	Super-gravity bed with disks	Ammonia...	CO ₂ /N ₂	Optimum structure of super-gravity bed with disks can be used to separate CO ₂ from flue gas effectually.

⊕ Researches on CO₂ Absorption Using Chemical Absorption Method by ZJU

□ Experimental Apparatus

Structured packed column

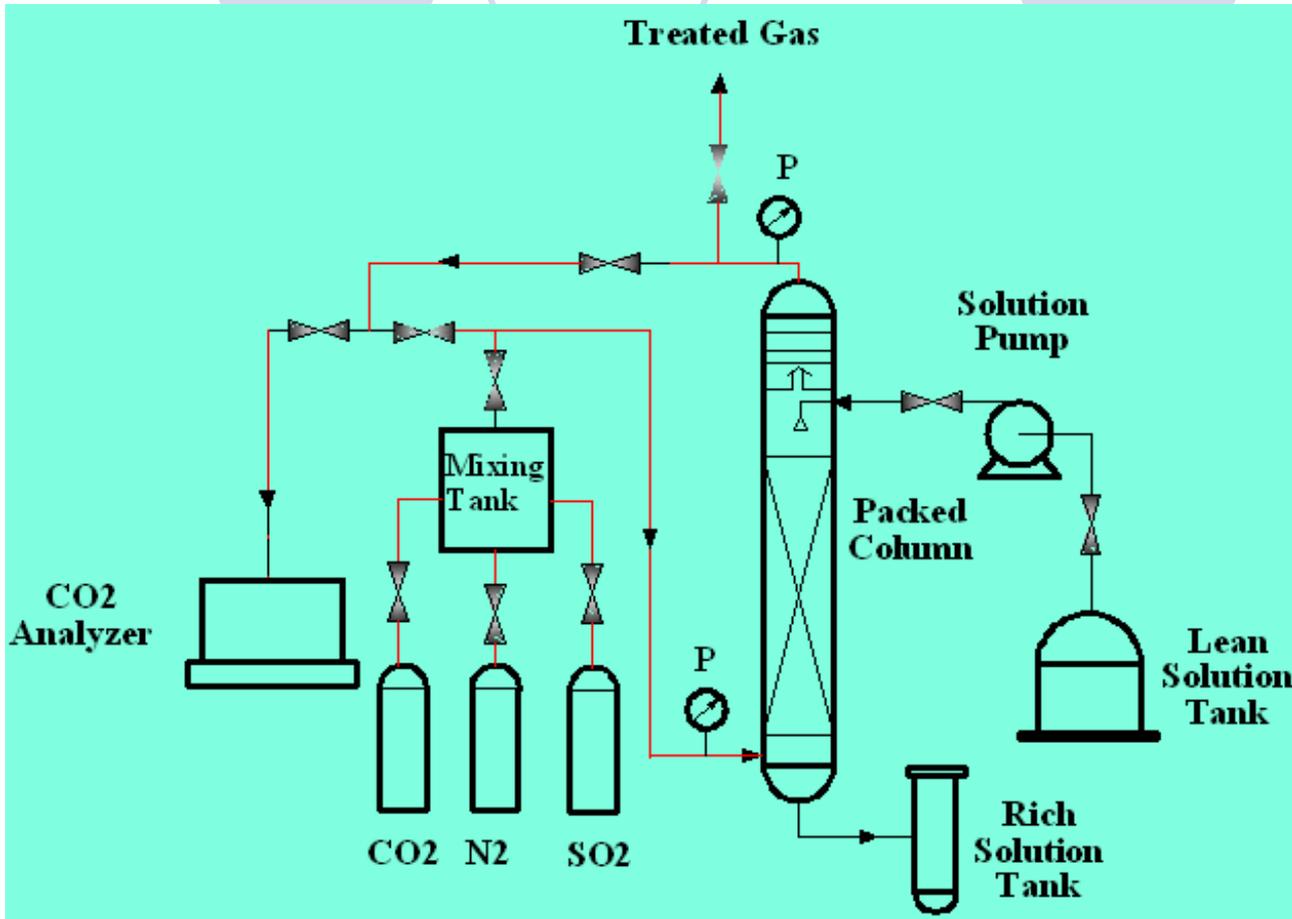
□ Absorbents

Single absorbents (MEA, DEA, AMP, PZ, MDEA, PG (Potassium Glycinate), Ammonia...); **Blended absorbents** (MEA/MDEA, MDEA/PG, MDEA/PZ, DEA/AMP...)

□ Operating Conditions

Flow rate, temperature, CO₂ concentration, SO₂ concentration, Additive concentration...

Experimental Setup

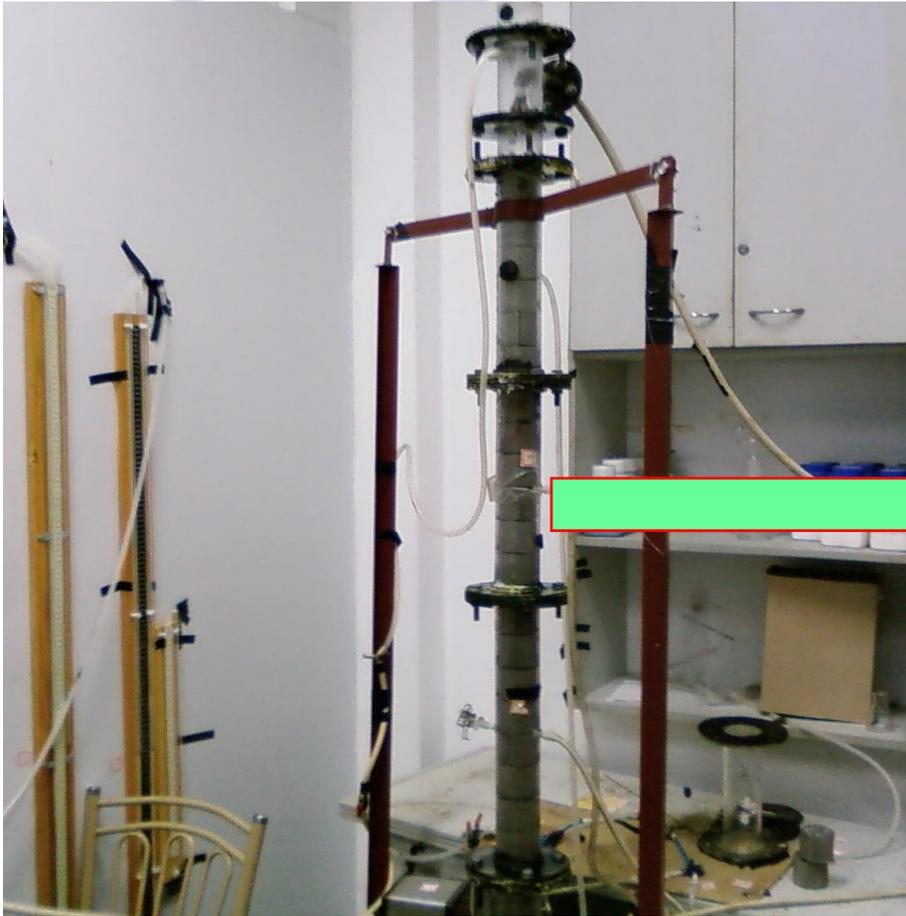


The simulated flue gas (CO₂/N₂/SO₂) was fed into the bottom of the packed column. At this point, the prepared absorbent solution from the lean solution tank was pumped to the top of the column at the desired flow rate by solution pump.

In the column, the gas and liquid phases contacted countercurrently, and CO₂ was absorbed by the absorbents. The treated gas finally left from the top of column, and CO₂-rich solution was pumped from the bottom of column to the rich solution tank.

Experimental Setup of CO₂ Absorption by Packed Column

⊕ Structured packed column



Structured packed column: Perspex glass

**Stainless steel silk
screen packing**



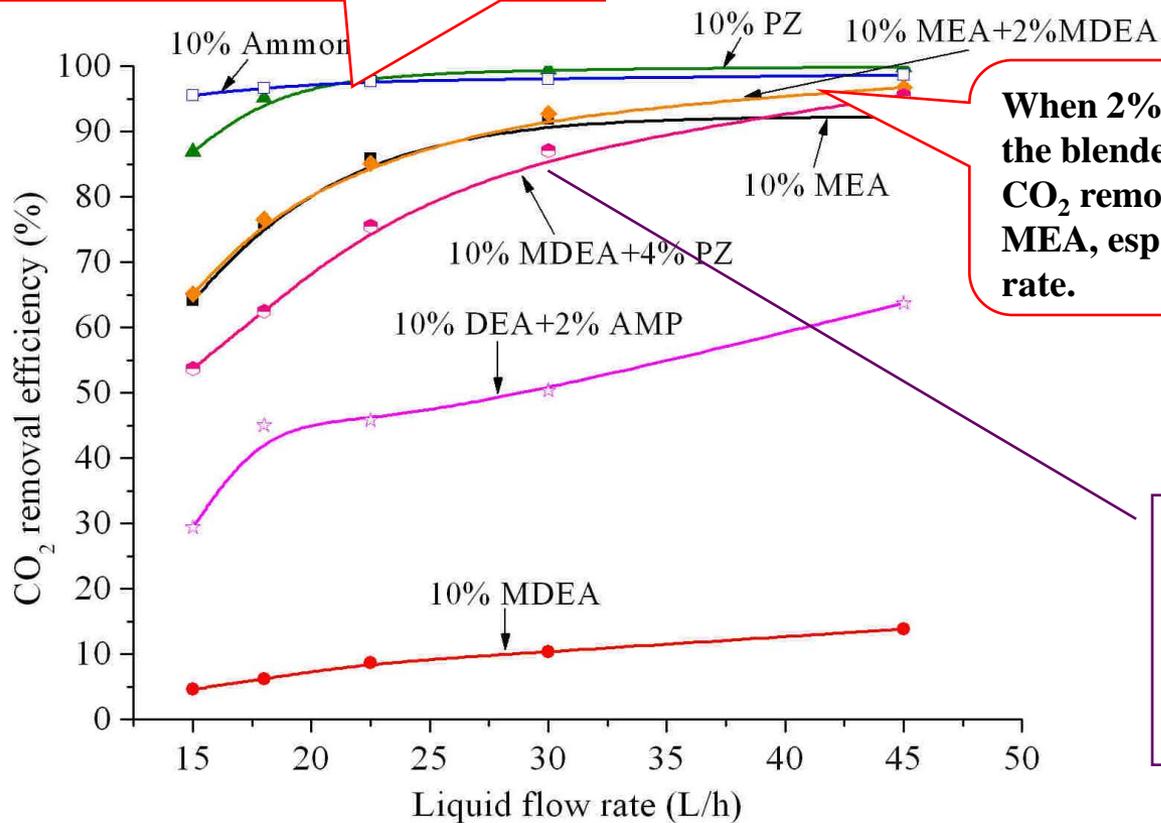
**Structured packing was placed
with each layer rotated with
 90° with respect to the
previous one.**

⊕ Operating Parameters

Item [⊖]		Characteristic Parameters [⊖]
Packed column [⊖]	Column I.D. (mm) [⊖]	58 [⊖]
	Packing height (m) [⊖]	0.6~1.35 [⊖]
	Packing size (mm) [⊖]	φ58×50 [⊖]
	Specific area (m ²) [⊖]	500, 700, 1000 [⊖]
Operating conditions [⊖]	Gas flow rate (m ³ /h) [⊖]	1.8 [⊖]
	CO ₂ concentration (%) [⊖]	12~16 [⊖]
	SO ₂ concentration (ppmv) [⊖]	0, 500, 1000, 1500 [⊖]
	Gas temperature (°C) [⊖]	Ambient [⊖]
	Gas pressure (kPa) [⊖]	~5 [⊖]
	Liquid flow rate (L/h) [⊖]	15~45 [⊖]
	Liquid temperature (°C) [⊖]	15~50 [⊖]
	Solvent concentration (w.t.%) [⊖]	5~30 [⊖]
	Solvent type [⊖]	Ammonia, MEA, PZ, MDEA, DEA, AMP [⊖]

Some Useful Results

Ammonia and PZ have the best CO₂ removal performance compared with other absorbents.

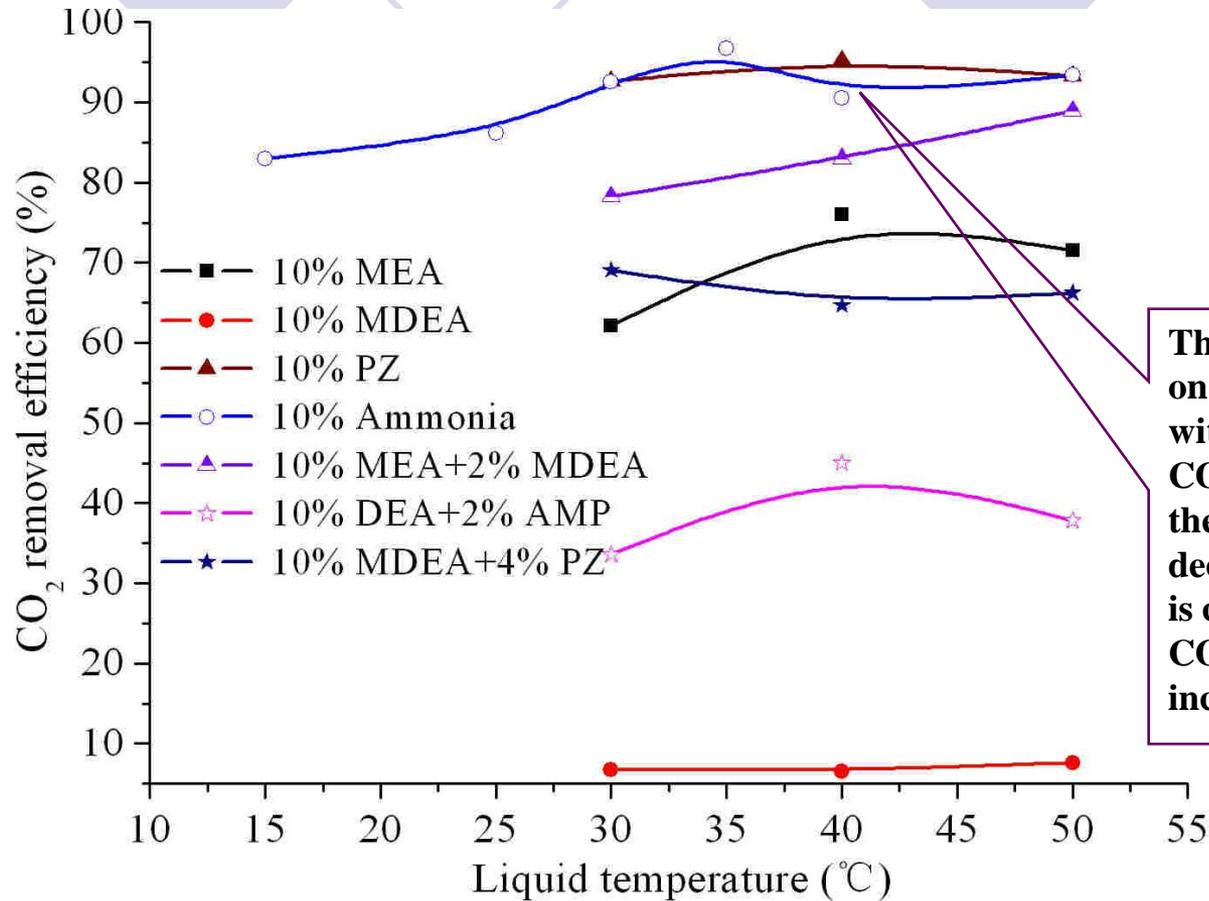


When 2% MDEA was added into MEA, the blended solvent has the slightly larger CO₂ removal performance than the single MEA, especially in the higher liquid flow rate.

Compared to the single MDEA, the CO₂ removal performance of MDEA/PZ solvent was increased greatly because of the active PZ.

Effect of liquid flow rate on CO₂ removal efficiency

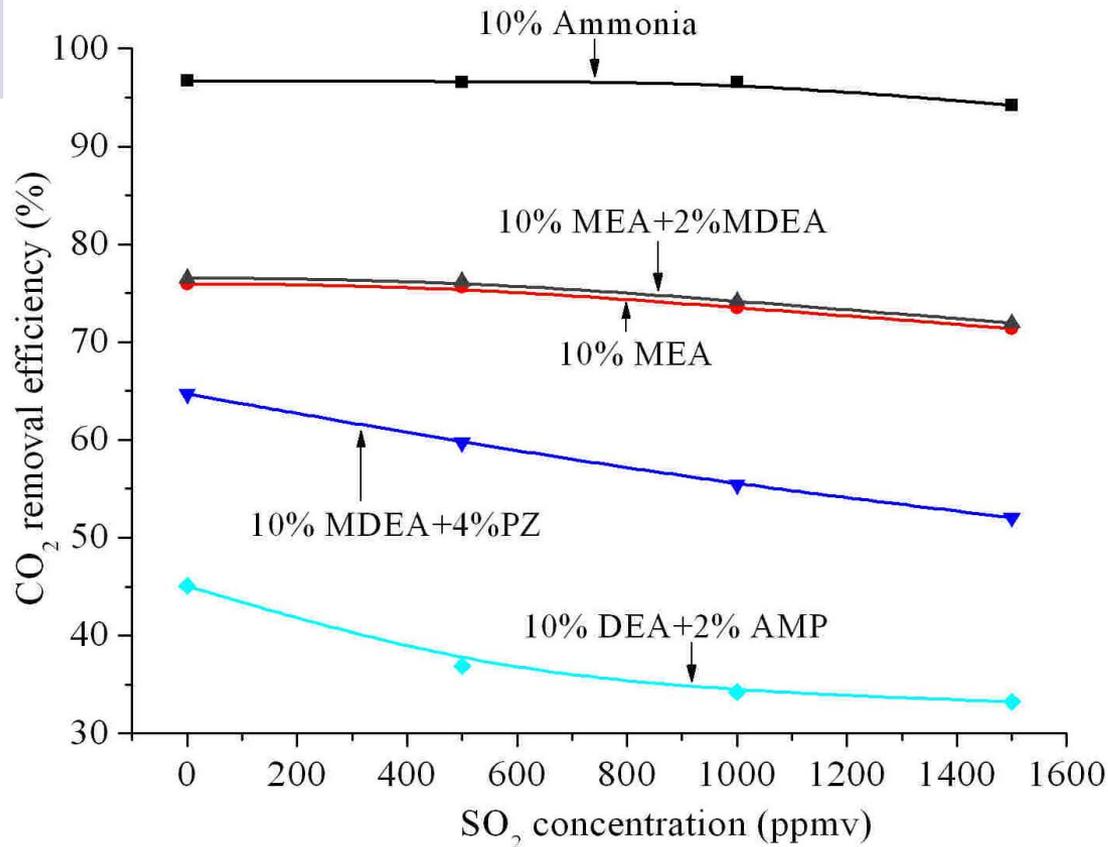
Effect of liquid temperature



The effect of liquid temperature on ammonia is not consistent with other solvents. Firstly, the CO₂ removal efficiency will reach the highest value, and then decrease. But, if the temperature is continuously increased, the CO₂ removal efficiency will increase again.

Effect of liquid temperature on CO₂ removal efficiency

Effect of SO₂



• **Results:** The effect of SO₂ concentration on CO₂ removal efficiency is negative due to the competitive reaction between absorbent and SO₂ or CO₂. (Reaction rate between SO₂ and absorbent is higher than that between CO₂ and absorbent)

• **Tendency:** The extent of negative effect of SO₂ on CO₂ removal increases with the decrease of reaction rate between absorbent and CO₂. For example, the negative effect of SO₂ on DEA/AMP > MDEA/PZ > MEA > MEA/MDEA > Ammonia.

Demonstration Plant of CO₂ Separation from Coal-fired Flue Gas in China

In **Dec. 26th, 2007**, the **first demonstration project** of CO₂ separation from coal-fired flue gas was **started at Beijing Thermal Power Plant** which is affiliated with China Huaneng Group. The demonstration project was supported by **China Huaneng Group and Commonwealth Scientific and Industrial Research Organization of Australia (CSIRO)**. The demonstration plant will be put into operation before the Opening date of Beijing Olympic Games.

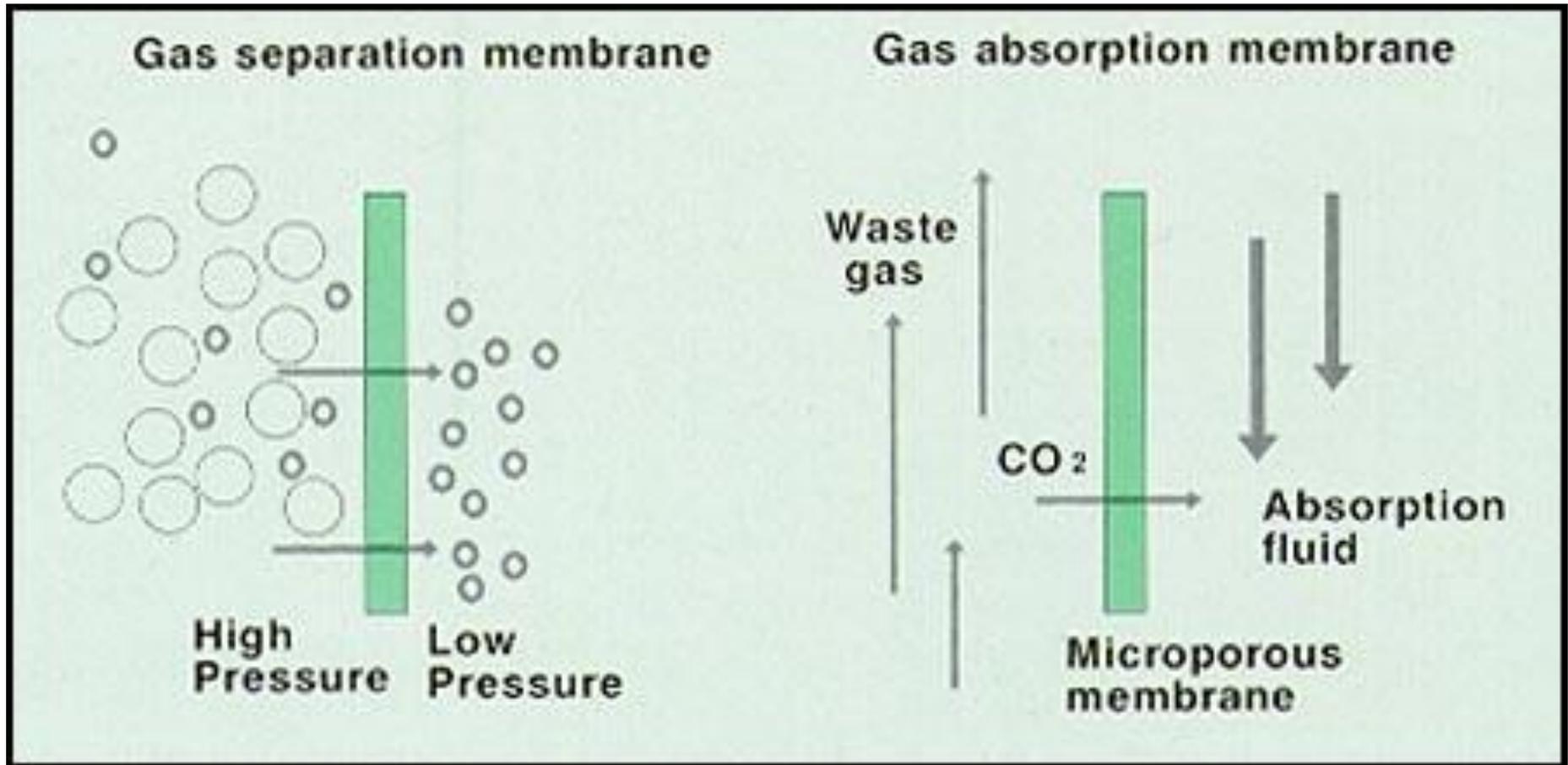


Demonstration Plant of CO₂ Separation from Flue Gas

Technological Goals:

- **CO₂ Recovery Efficiency: >85%**
- **CO₂ Yields: up to 3000 ton/year**
- **CO₂ Purity after separation and purification: >99.5%**
- **CO₂ Consumer: Food service industry.**

⊕ 2 Membrane Technology for CO₂ Separation



Research on Membrane Separation in China

As the promising CO₂ capture technology, many researchers are now devoting their enthusiasms on it, such as CAS, Dalian University of Technology, Beijing Mining Institute, Tsinghua University, Beijing University of Chemical Technology, Zhejiang University, Tianjin University, Nanjing University of Science and Technology and so on.

The research keystone:

- **New Separation Membranes** (higher CO₂ selectivity or higher hydrophobicity);
- Mass Transfer Model;
- Effects of operating conditions on CO₂ removal performance;
- **Effect of membrane wetting and plugging on CO₂ removal;**
- Technologies to prevent the wetting and plugging problems (New membrane, new solvent...).

Membrane Separation Research in China

Researchers [↵]	Membrane Modules [↵]	Solvents [↵]	Gas Phase [↵]	Main studies [↵]
Zhejiang University [↵]	Polypropylene (PP) hollow fiber membrane [↵]	Aqueous solutions of MEA, MDEA, PG and some blended amines [↵]	CO ₂ /N ₂ /O ₂ [↵]	The 5 Nm ³ /h pilot-scale experimental system for CO ₂ separation from flue gas was built and new absorbents were tested. Comparative analysis of membrane absorption and chemical absorption was also studied. [↵]
Beijing Mining Institute [↵]	Polypropylene hollow fiber membrane [↵]	Aqueous NaOH solution [↵]	CO ₂ /Air [↵]	Effects of operating conditions on the overall mass transfer coefficient were experimented, and a model to simulate the overall mass transfer coefficient was developed. [↵]
CAS [↵]	Polypropylene (PP) hollow fiber membrane... [↵]	Water, aqueous NaOH solution [↵]	CO ₂ / N ₂ [↵]	Characteristics of membrane absorption process have been experimentally studied; New membranes were also tested. [↵]

Some Researchers on CO₂ membrane absorption

Researchers [↵]	Membrane Modules [↵]	Solvents [↵]	Gas Phase [↵]	Main studies [↵]
<u>Tianjing</u> University [↵]	Polypropylene (PP) hollow fiber membrane [↵]	Water, aqueous [↵] <u>NaOH</u> , MEA and AMP solutions [↵]	CO ₂ [↵]	The mathematical model of mass transfer in the membrane module. [↵]
<u>Dalin</u> University of Science and Technology [↵]	Polypropylene hollow fiber [↵] membrane [↵]	Water, aqueous <u>NaOH</u> [↵] and K ₂ CO ₃ solutions [↵]	CO ₂ [↵]	The effects of some key operating parameters [↵]
Nanjing University of Science and Technology [↵]	Polypropylene hollow fiber [↵] membrane [↵]	Aqueous MDEA and [↵] MDEA/PZ solutions [↵]	CO ₂ / N ₂ [↵]	The effects of some key operating parameters; effect of membrane pore wetting [↵]
Beijing University of Chemical Technology [↵]	Polypropylene hollow fiber [↵] membrane [↵]	Water, aqueous <u>NaOH</u> solution [↵]	CO ₂ /Air [↵]	The effects of some key operating parameters [↵]

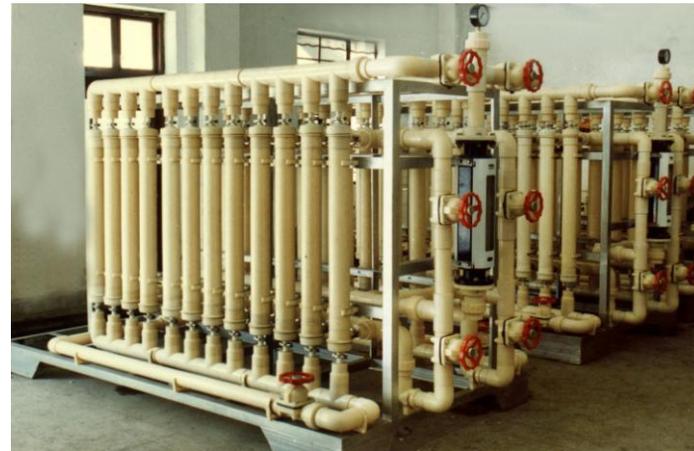
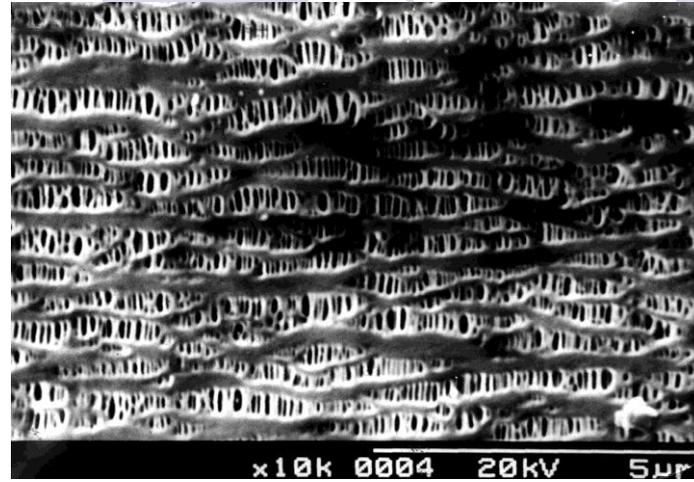
In China, most researches on CO₂ separation using membrane technology were focused on the experimental discussion and few pilot scale test facility was erected.

Membrane Gas Absorption Facility in ZJU

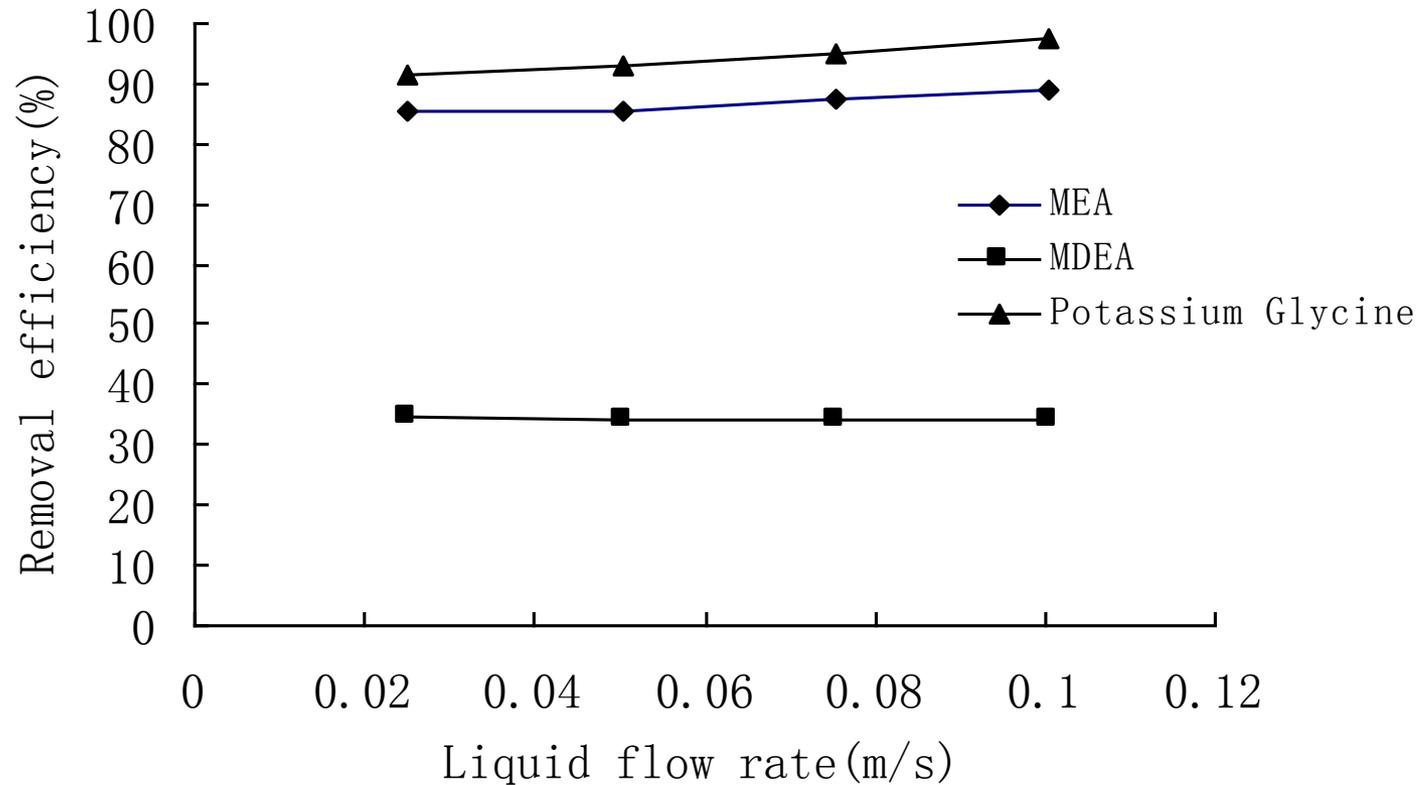


5 Nm³/h

Characteristics of PP hollow fiber membrane

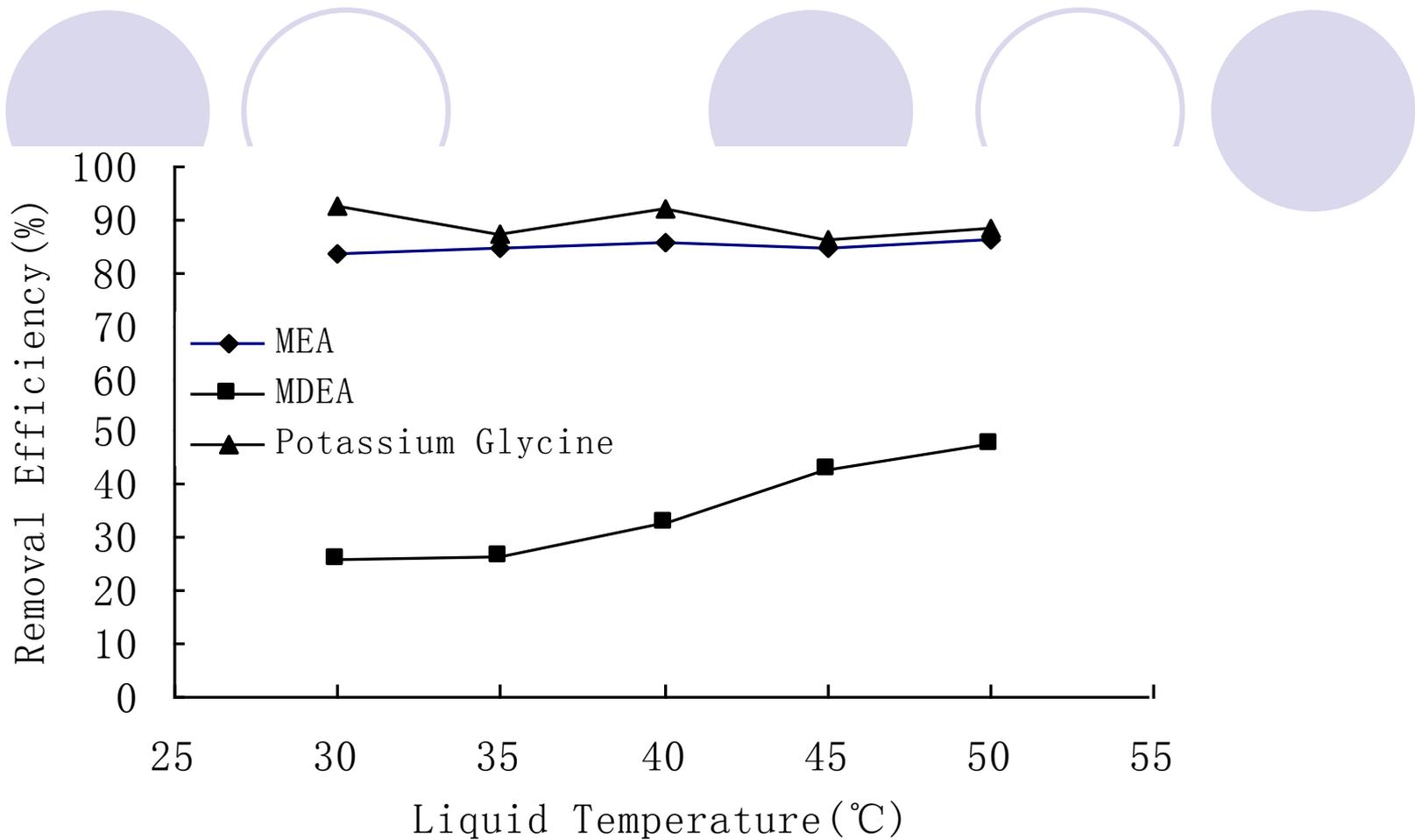


Results of CO₂ Capture from Flue Gas Using Membrane Contactors

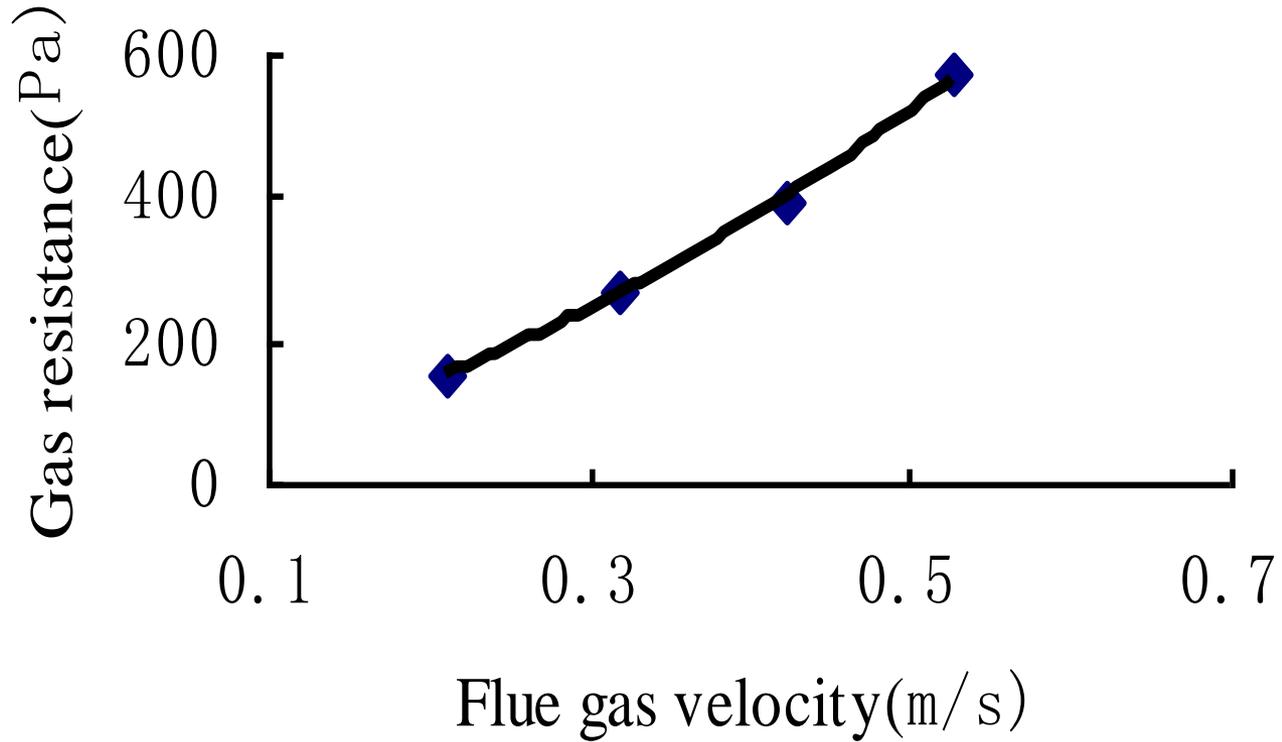
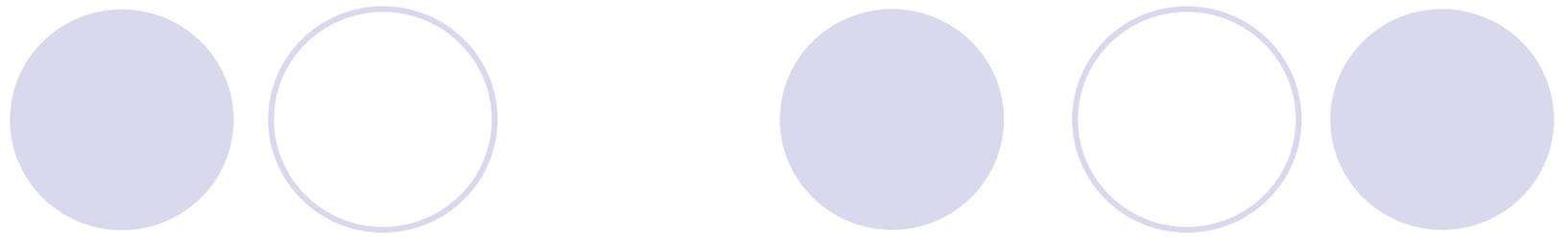


Influence of the absorption solutions on removal efficiency

(Flue gas: 0.211 m/s, absorption solution: 35 °C, 3 mol/L)

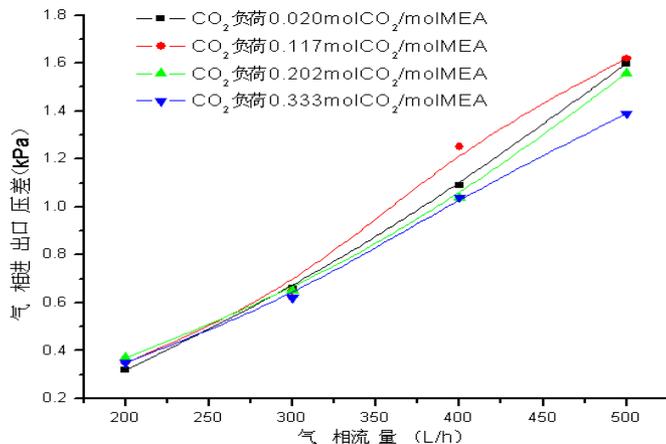
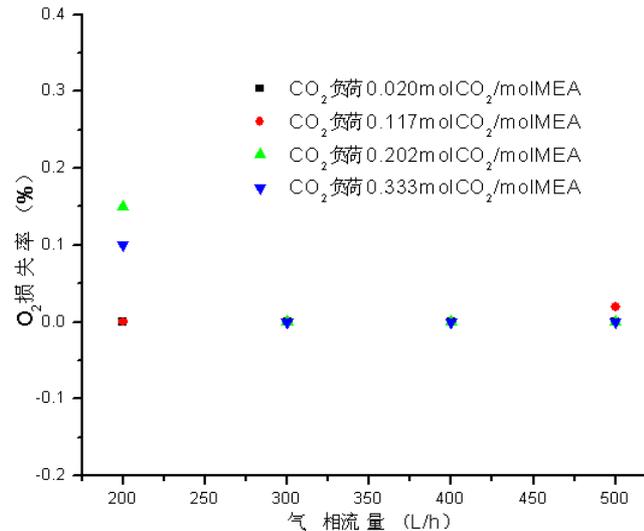
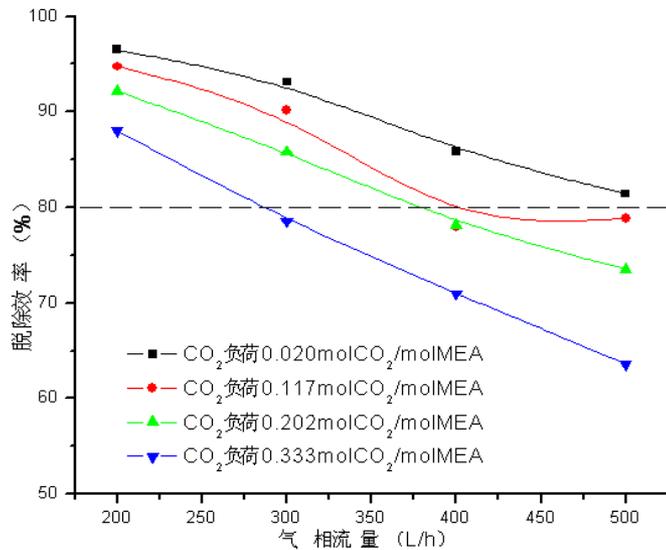


- **Influence of absorption solutions temperature on removal efficiency**
- **(Flue gas: 0.211 m/s, absorbent: 1 mol/L, 0.0503 m/s)**



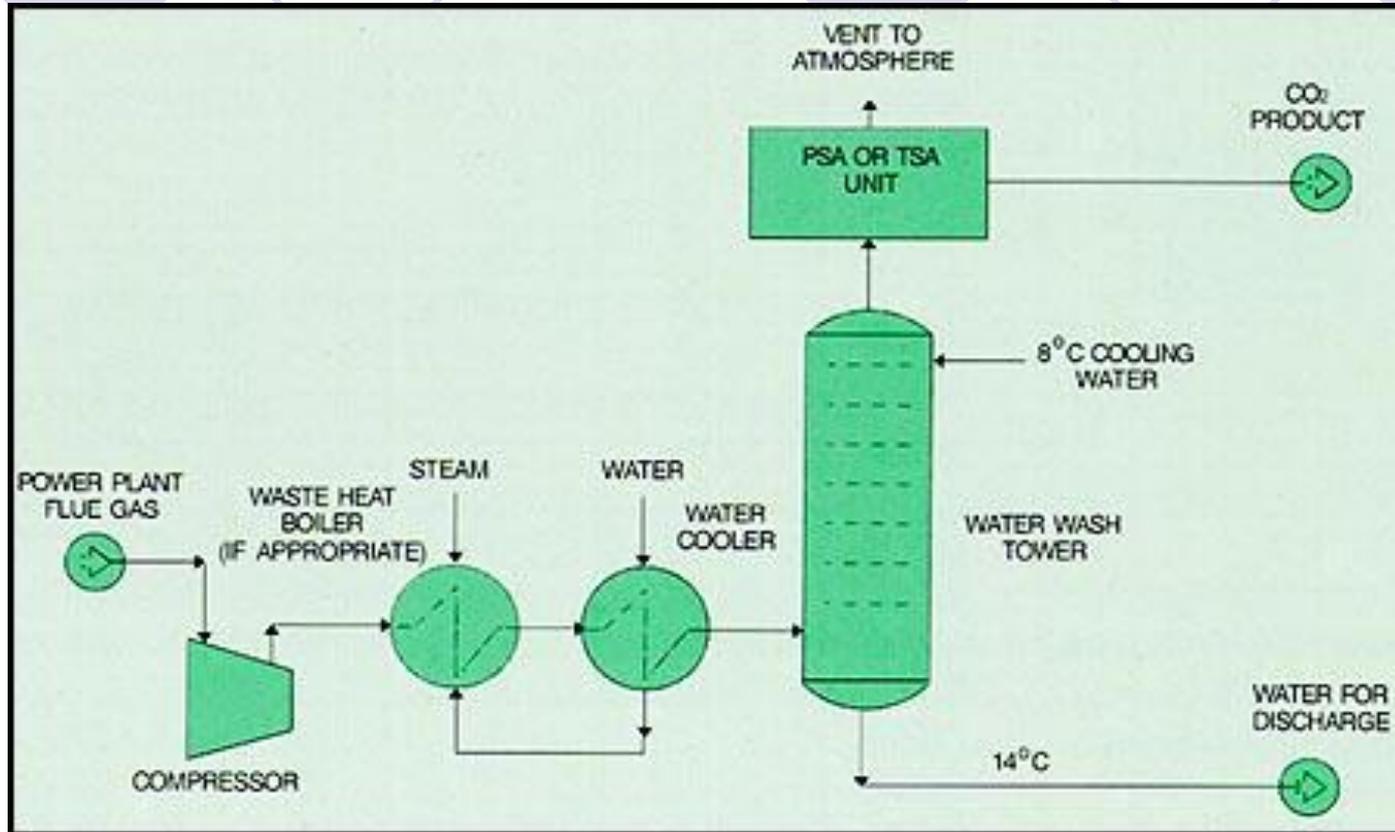
Gas pressure drop of membrane contactor

Results of CO₂ Capture from Low CO₂ Concentration Gas



CO₂ recovery efficiency, O₂ loss rate and gas pressure drop (1% CO₂ Concentration)

3 Adsorption of carbon dioxide

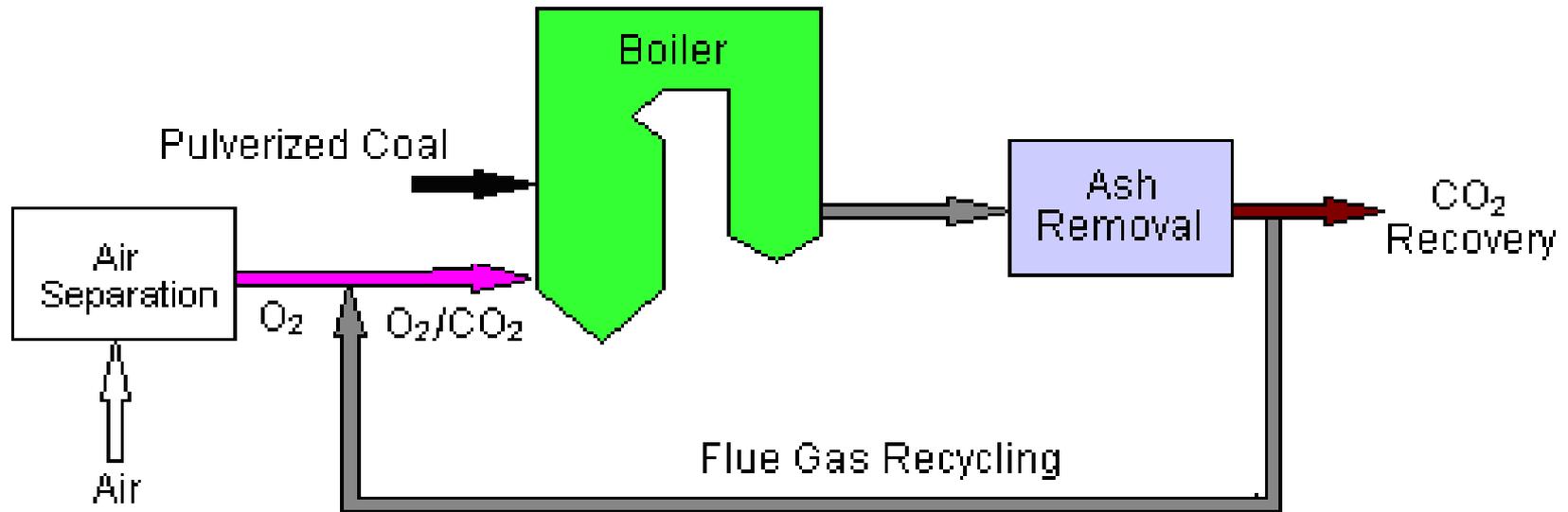


Adsorption is based on the inter-molecular attraction between the gas molecular and the active sites on the surface of adsorbent. The adsorbent is made up of some special solid, such as zeolite, active carbon and molecular sieve.

Research on CO₂ adsorption in China

- **China University of Mining Technology** prepared the **active carbon** using anthracite as the feed coal used in separating CO₂. This active carbon have a long adsorption time and a great adsorption capacity.
- *Nanjing Chemical Institute* researched the adsorption using **zeolite**, **active carbon** and **molecular sieve** in 1991. There are more than **100 apparatus** have been applied and the scale is more than 60000m³/h. The apparatuses are mainly used in purification of CO₂, H₂S, NO_x, H₂, and natural gas and removal of some toxic substance.

II Oxy-fuel Combustion

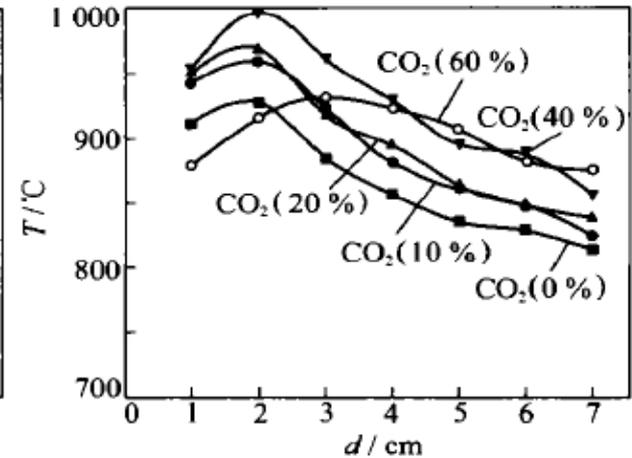
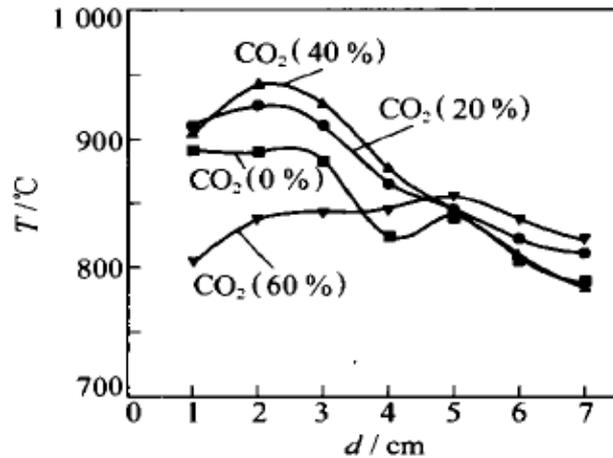
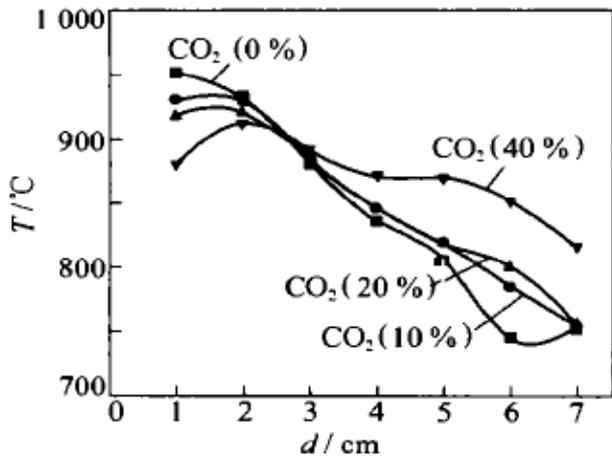


Typical Oxy-fuel combustion process

Research on Oxy-fuel Combustion in China

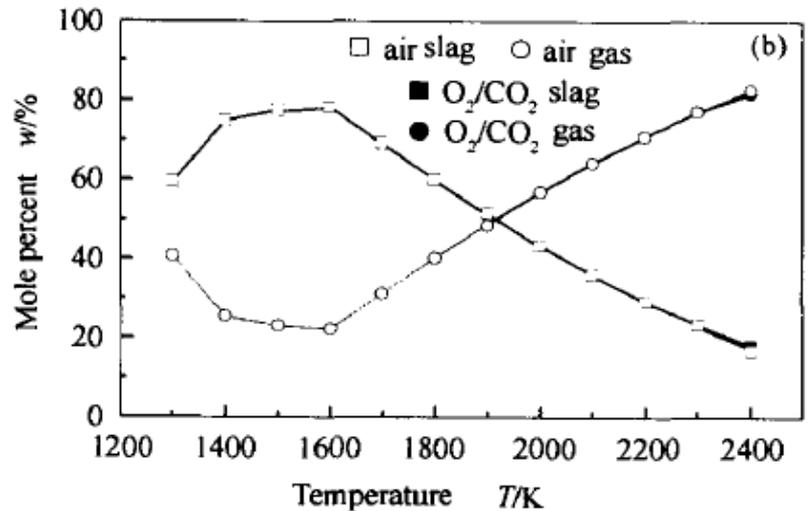
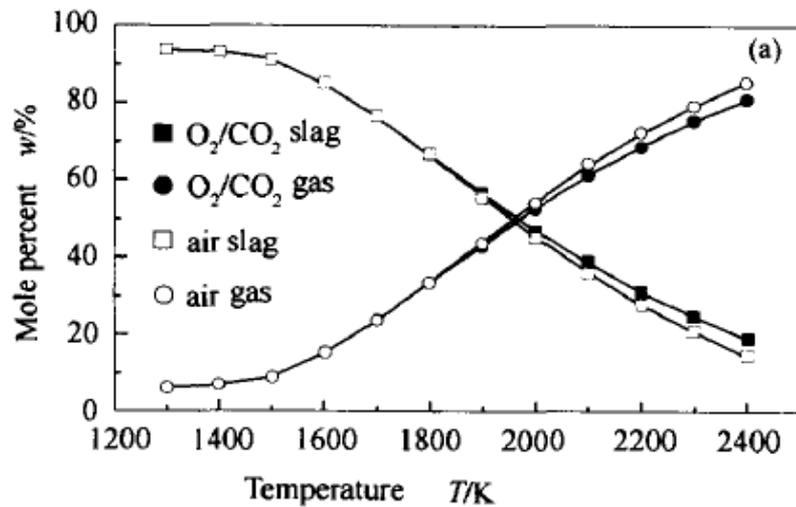
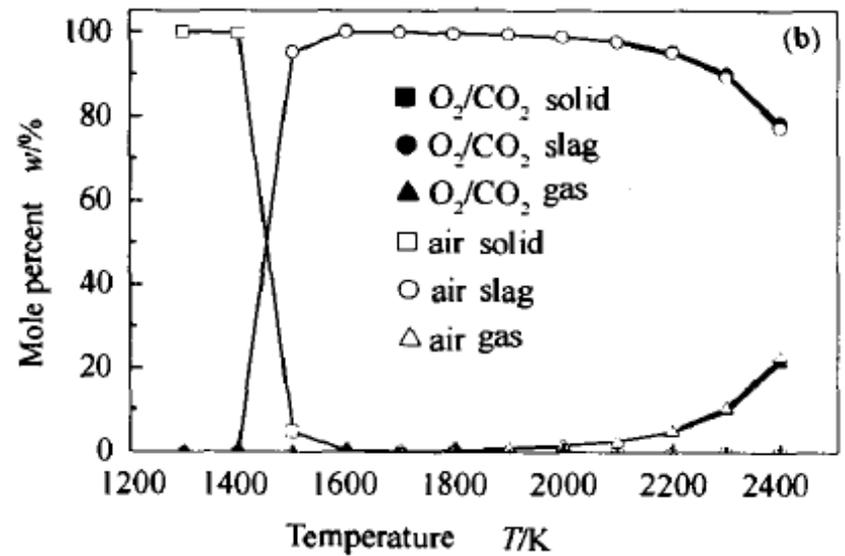
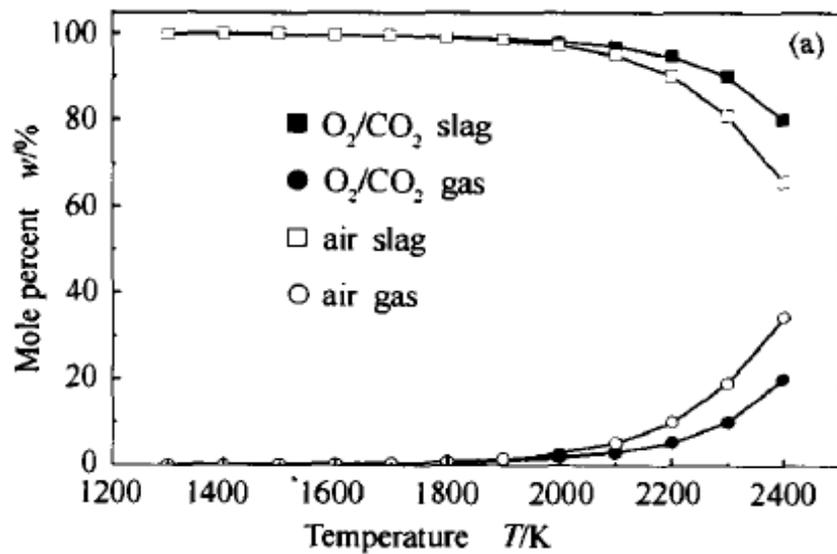
- **Southeast University:** Characteristics of Pulverized Coal Combustion in O₂/CO₂ Mixtures; Thermodynamics analysis; Pollutant Control...
- **North China Electric Power University.** Based on the experiments of characteristics of coal combustion and pollutant emissions under the oxy-fuel conditions using an entrained-flow reactor, they reported the decrease of SO₂ and NO_x emissions.
- **Huazhong University of Science and Technology** studied the desulfurization under the condition of oxy-fuel combustion in a vertical tube electrical heating reactor and a small oxy-fuel combustion test system was built.
- **Zhejiang University** studied the oxy-fuel combustion with the CFB.
- Tsinghua University
- Chongqing University
-

Oxyfuel Combustion Research by Huazhong University of Science and Technology



Effect of air excess coefficient (λ) on temperature characteristics of CH_4 flame under O_2/CO_2 atmosphere.

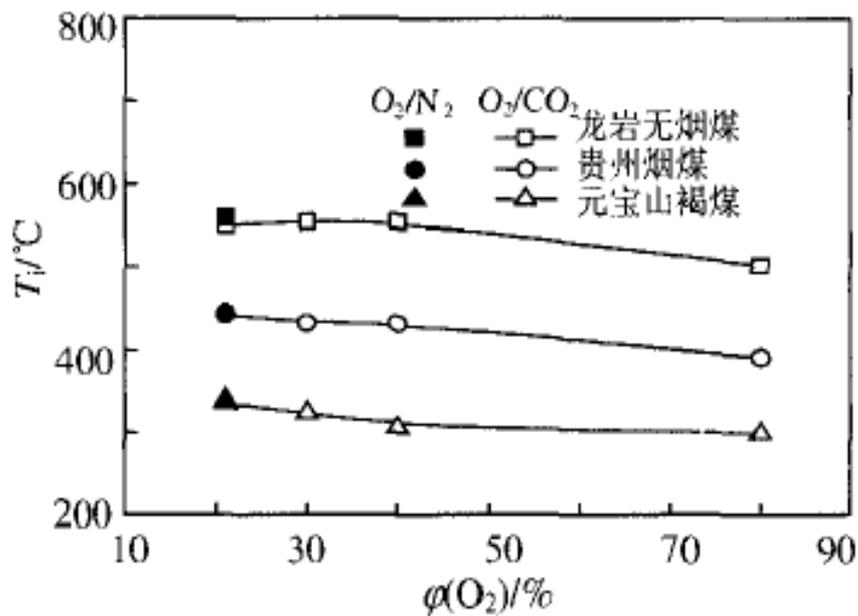
(Left: $\lambda=0.8$; Medium: $\lambda=1$; Right: $\lambda=1.1$)



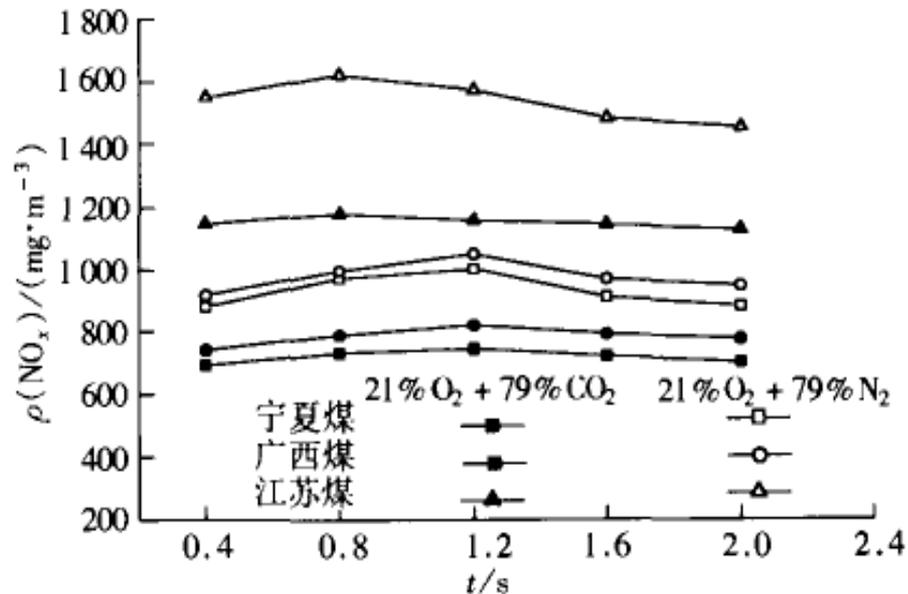
Equilibrium distributions of Fe and Na under different atmospheres.

(a: reducing condition; b: oxidizing condition)

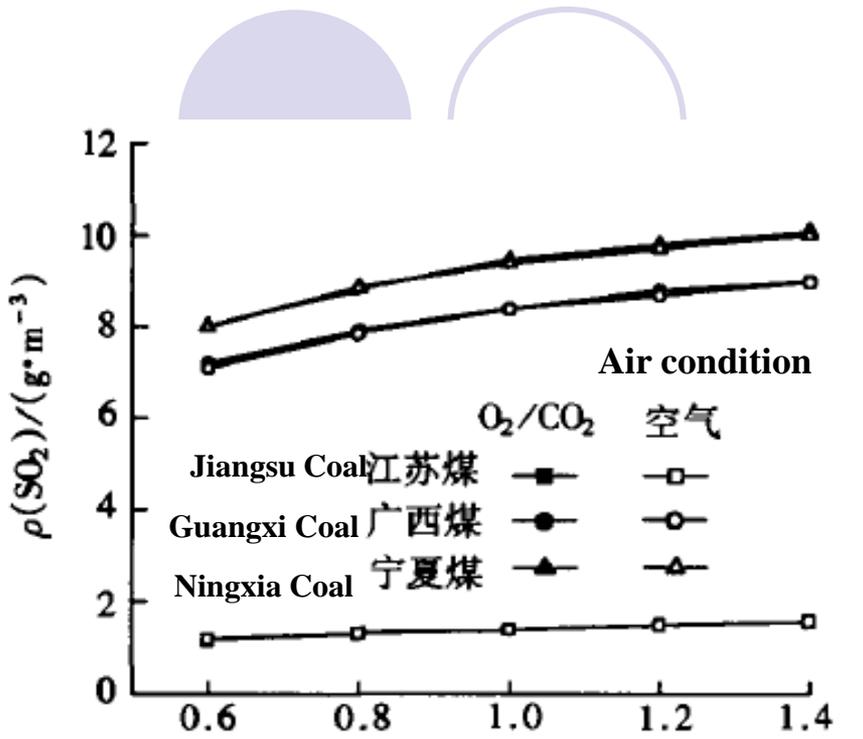
Oxyfuel Combustion Research by Southeast University and North China Electric Power University



Effect of O_2 content on burnout temperature (T)

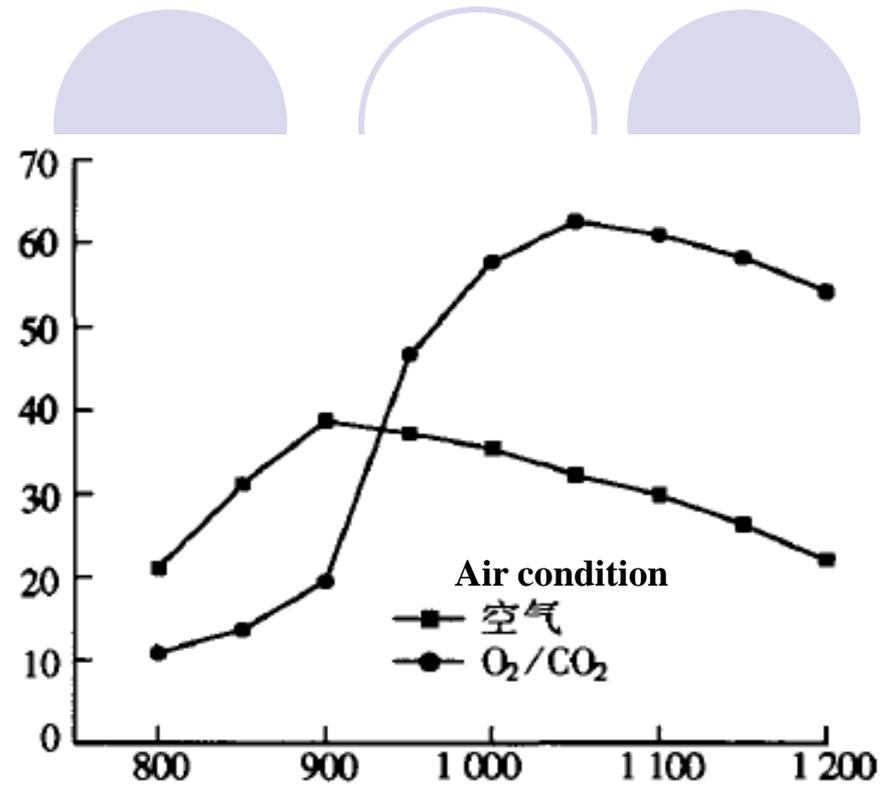


Characteristic of NO_x emission (ρ) under O_2/CO_2 condition



Chemical equivalent of fuel and O₂

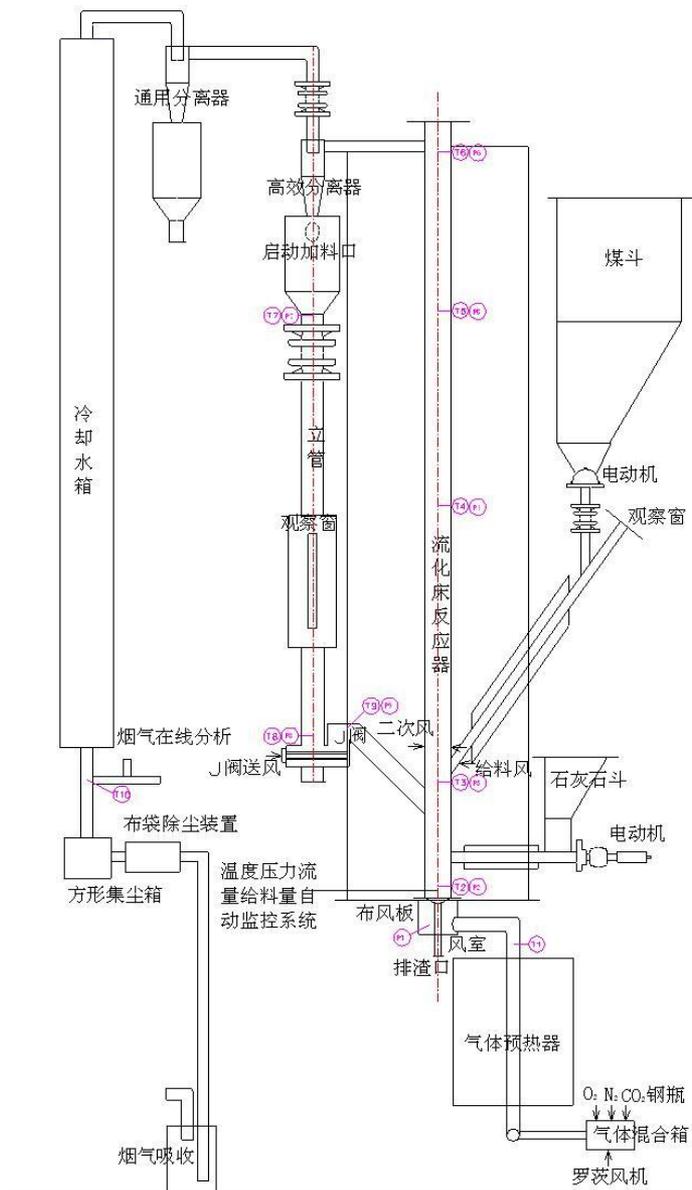
Effect of fuel/O₂ chemical equivalent on SO₂ emission



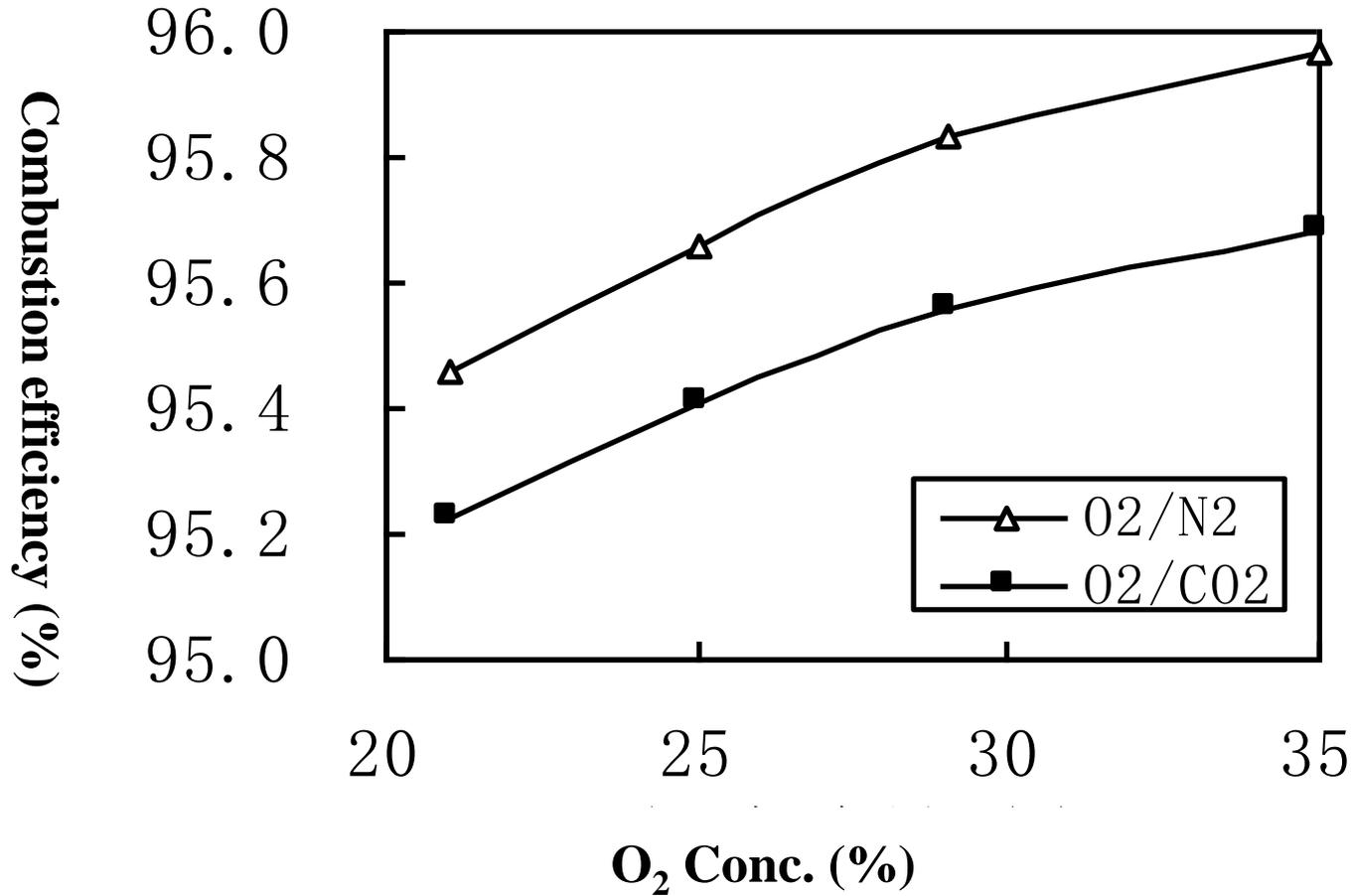
Temperature

Effect of temperature on SO₂ removal efficiency

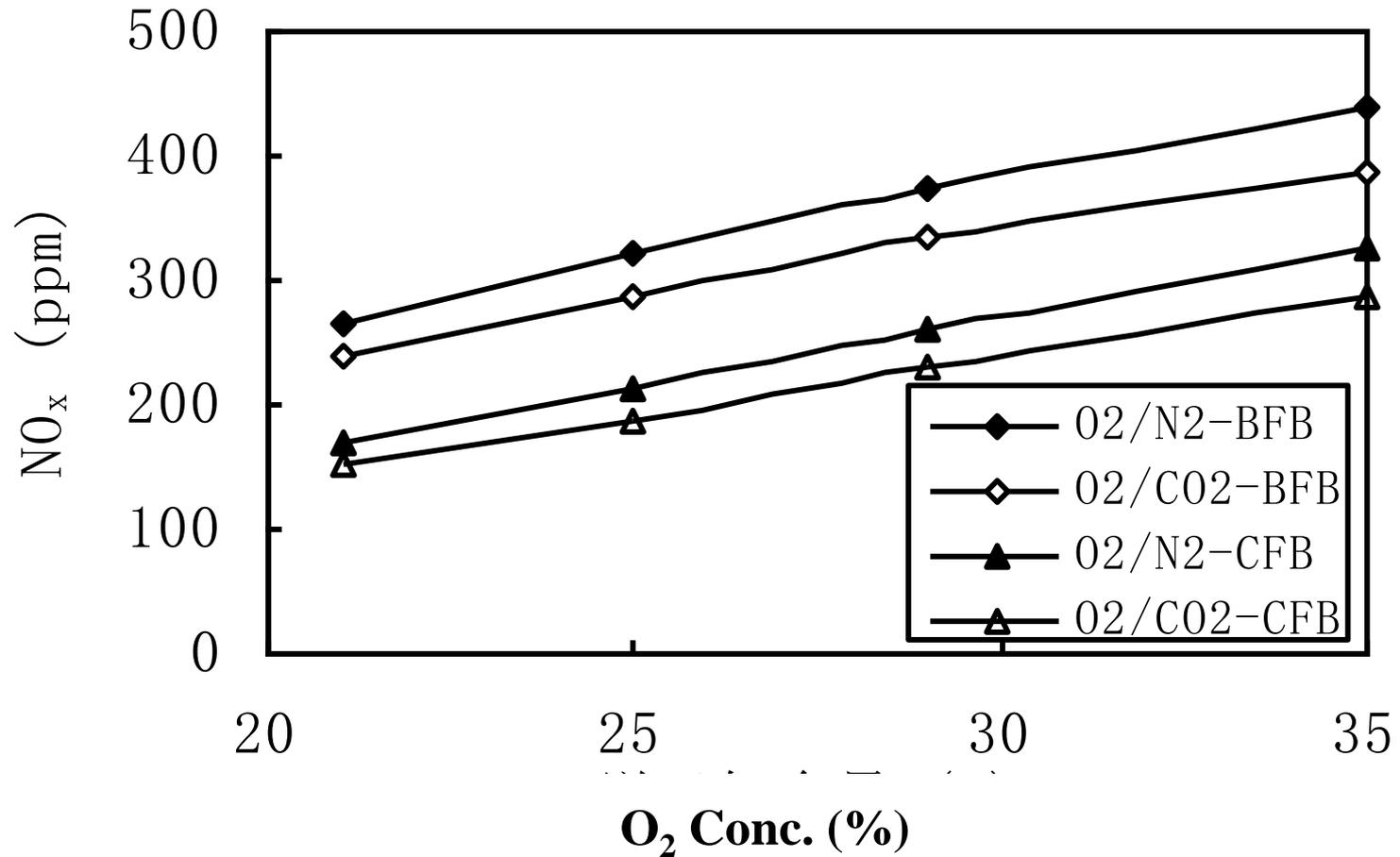
Oxyfuel Combustion Test-facility in ZJU



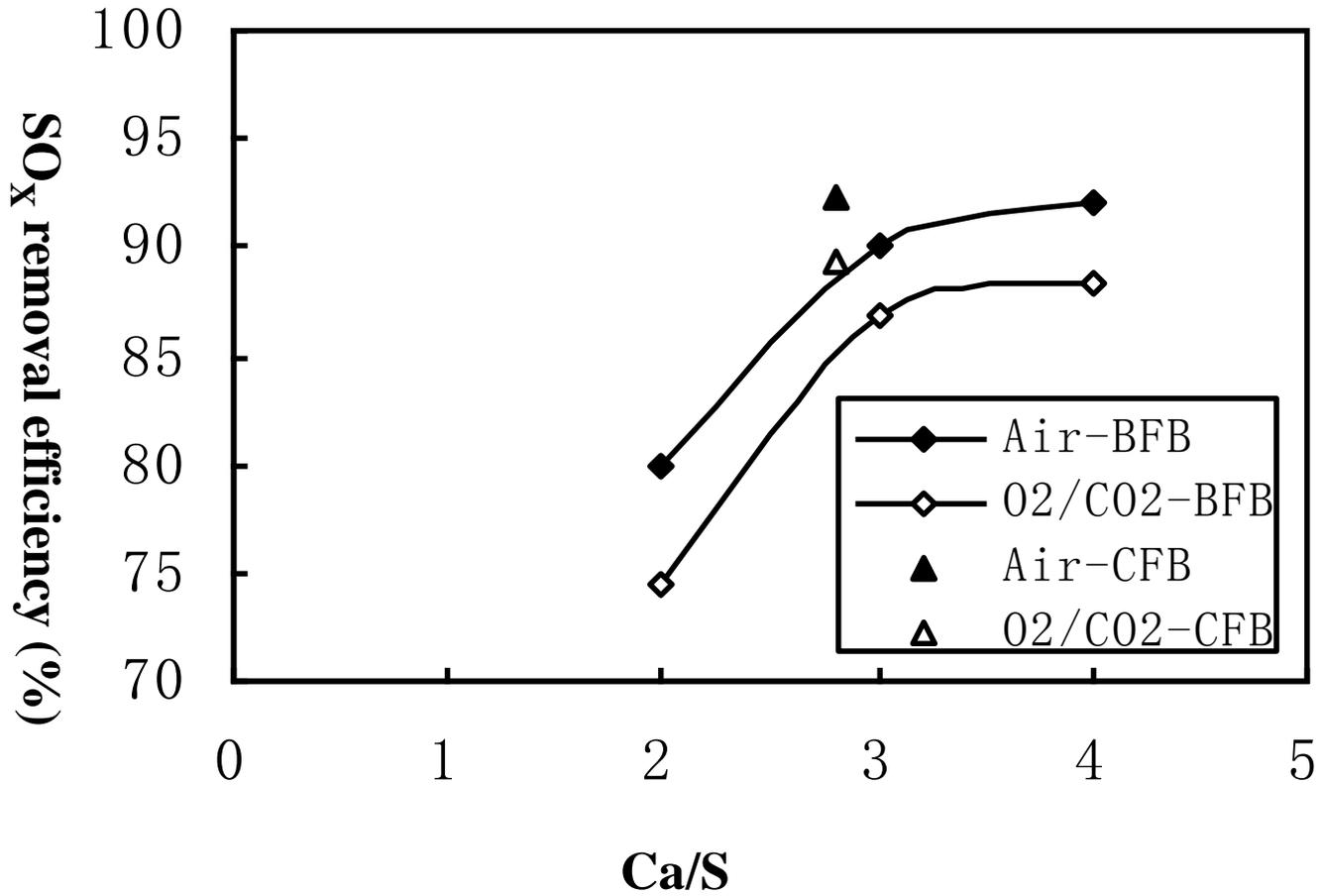
Oxy-fuel Combustion Efficiency



Emission of O₂ Content on NO_x Emission

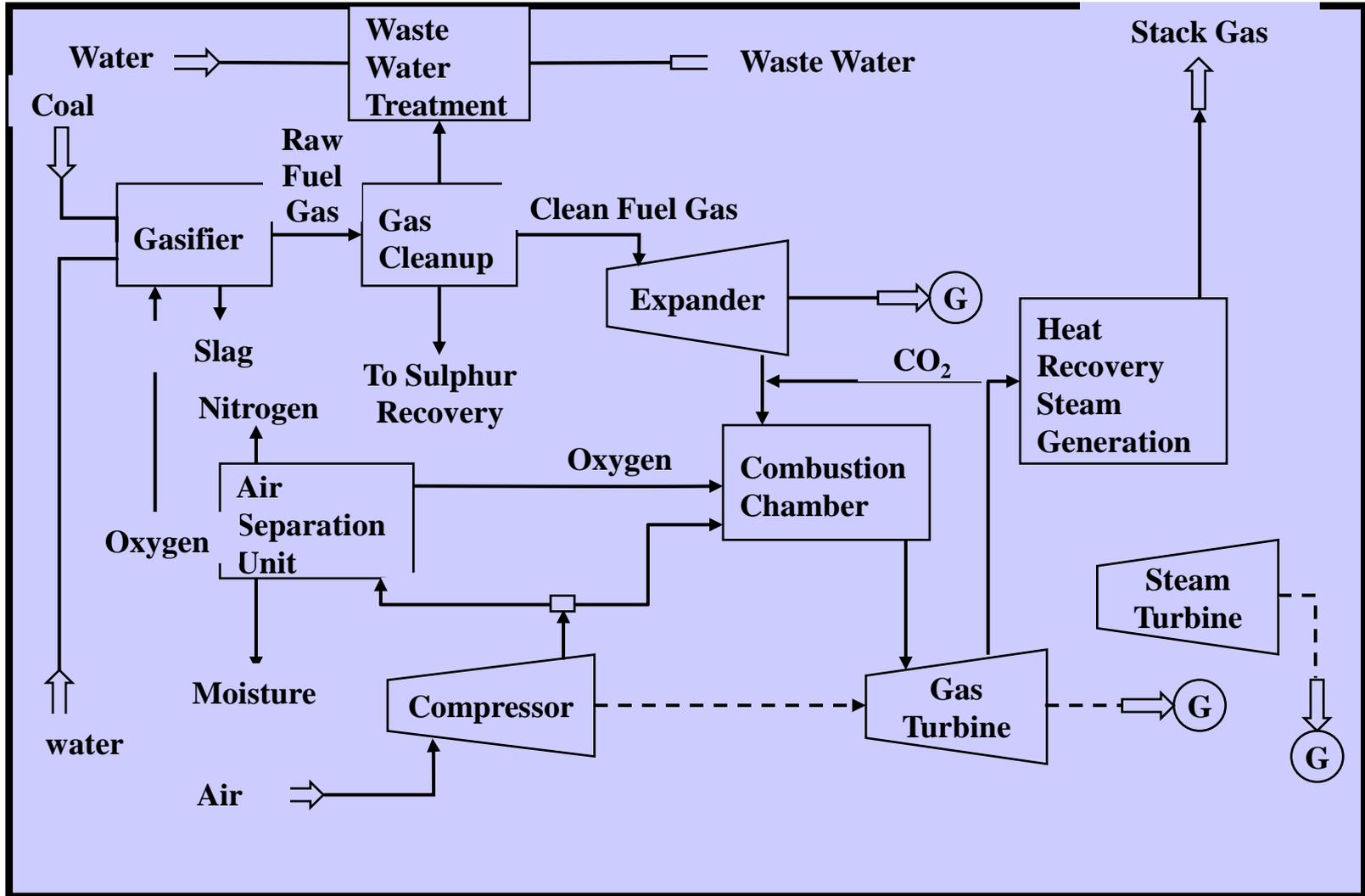


SO_x Removal Efficiency



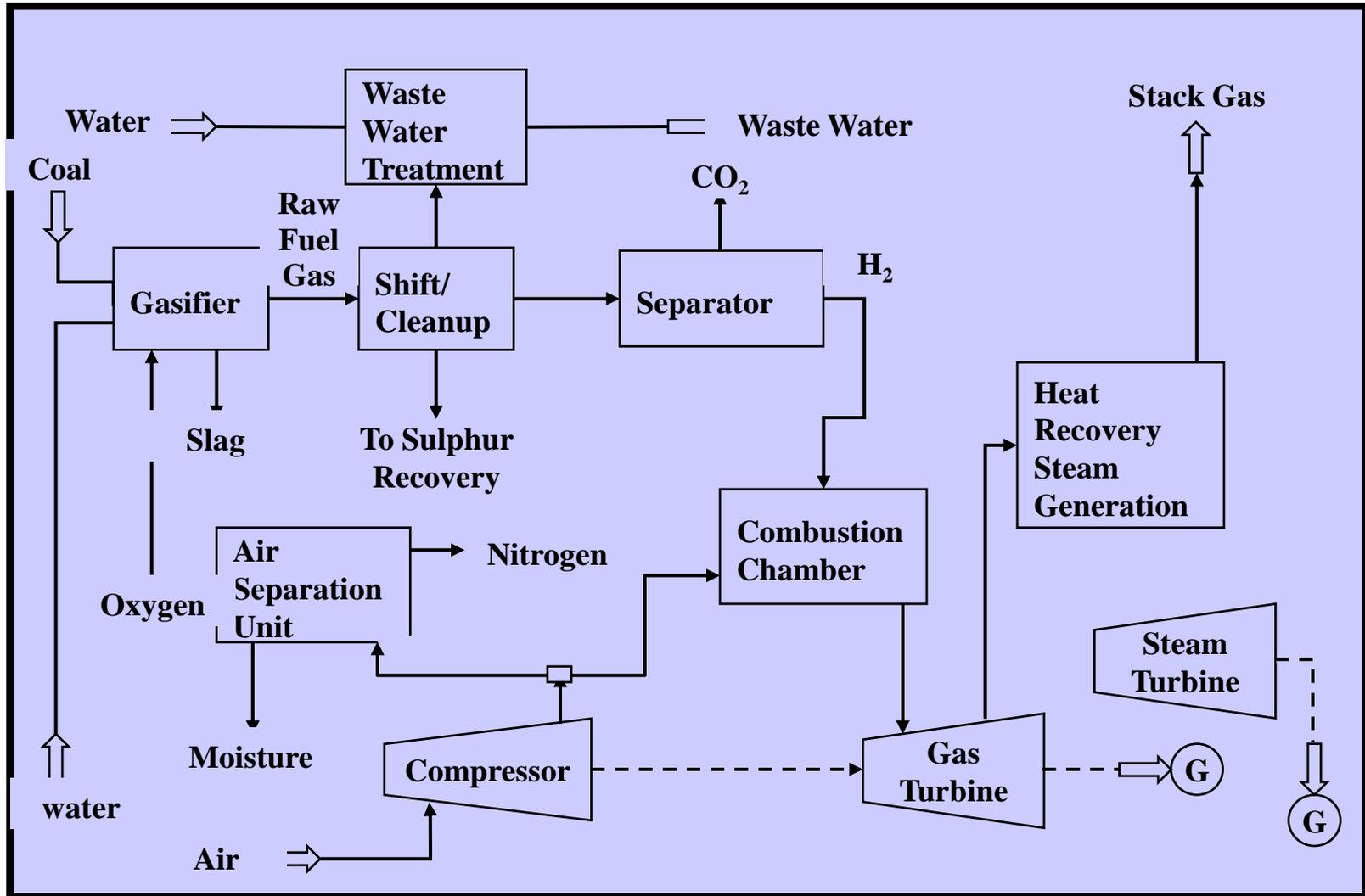
III-1 IGCC with CO₂ separation

Scheme1: IGCC with CO₂ Separation after combustion



III-2 IGCC with CO₂ separation

Scheme2: IGCC with CO₂ Separation before combustion



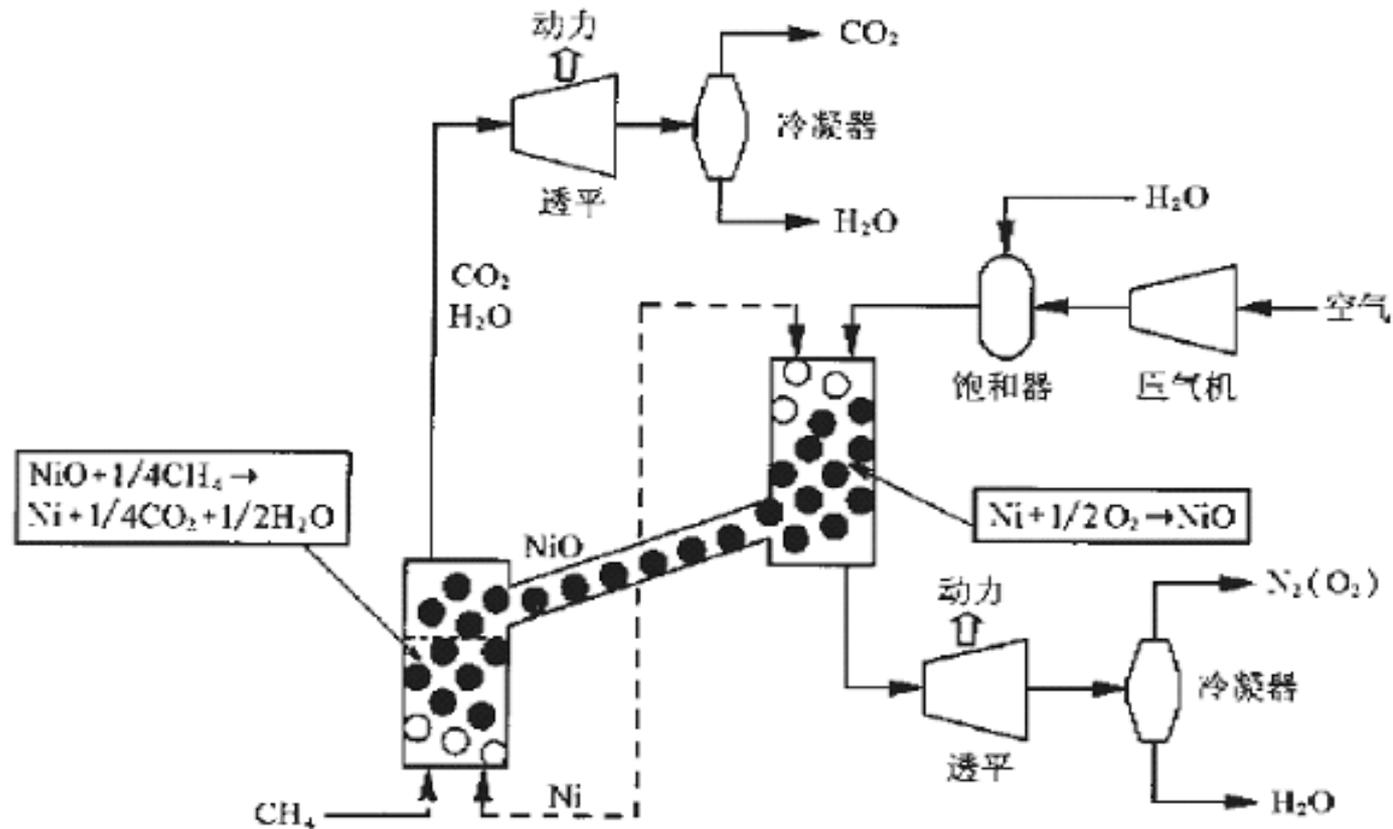
GreenGen's 400 MW Demonstration Project with H₂, Electricity Production and CO₂ separation in Tianjin



Technological goals of GreenGen

- To complete industrial demonstration project of 400MW before 2020
- Generation efficiency up to 55%-60%
- SO₂ and NO_x emission less than 20mg/Nm³
- PM_{2.5}, Hg and VOC near zero discharge
- Over 80% of CO₂ separated and treated
- Byproducts effectively utilized
- Acceptable cost

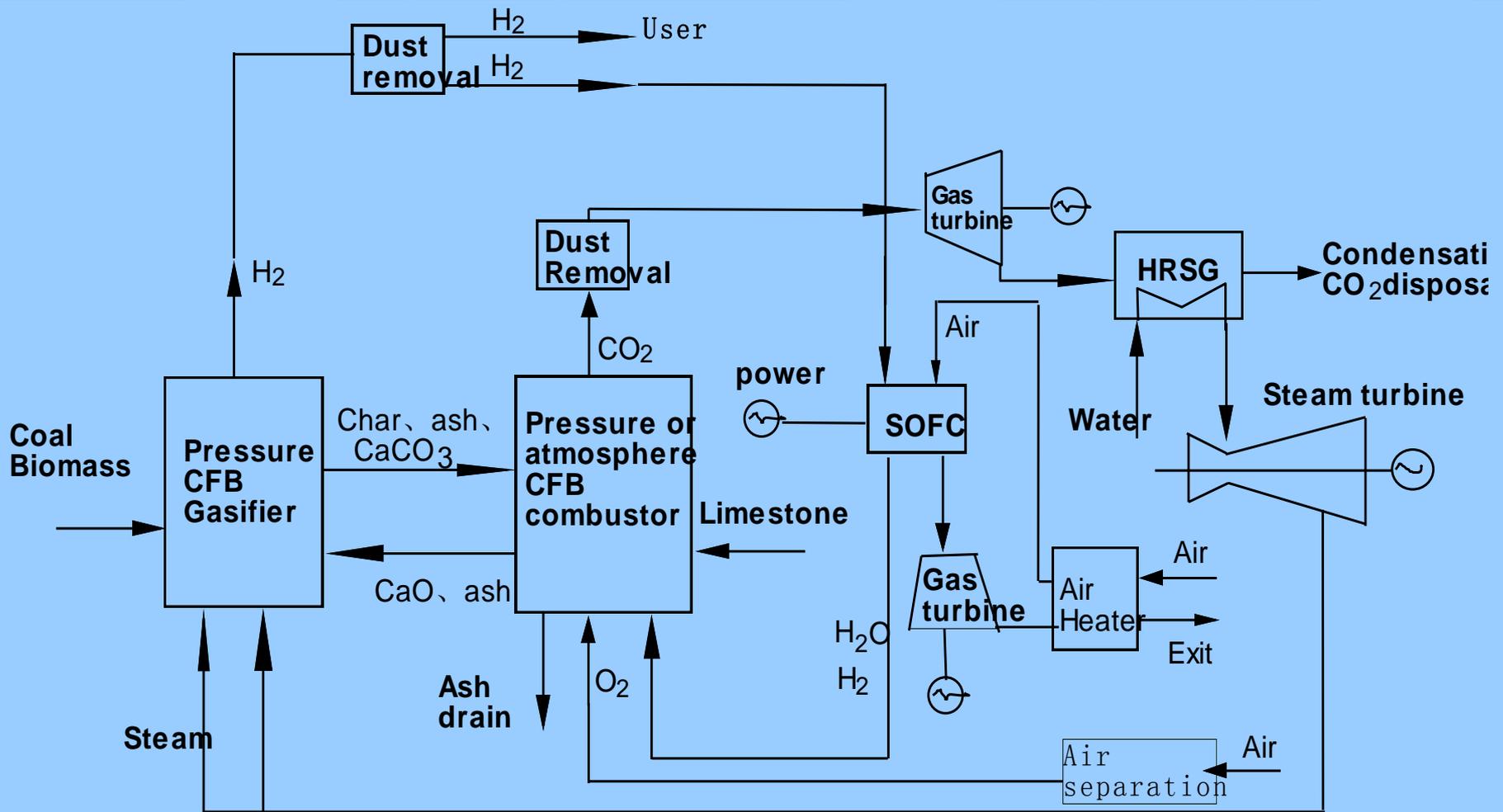
IV Chemical-looping Combustion



Research on Chemical looping Combustion system

- **Institute of Engineering Thermophysics** studied the chemical looping combustion technology combined with the saturation for air combustion turbine using $\text{NiO/NiAl}_2\text{O}_4$ as oxygen carriers.
- **Southeast University** studied the high rate power generation and CO_2 separation with gasify, chemical looping combustion and combined cycle technologies using $\text{Fe}_2\text{O}_3, \text{Fe}_3\text{O}_4$ as oxygen carriers.
- **Zhejiang University** also studied the chemical looping combustion.

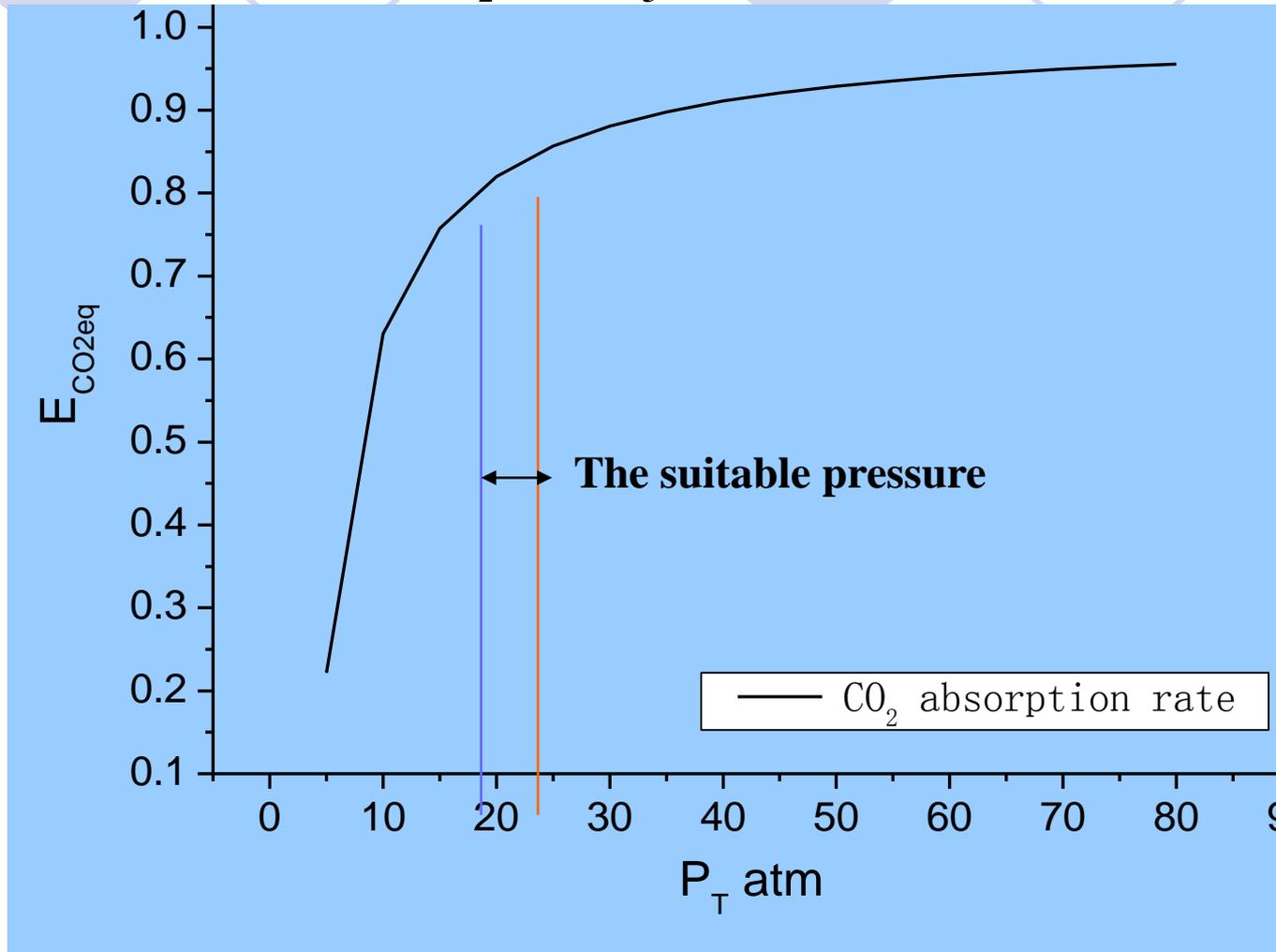
V Near Zero Emissions Coal Utilization Technology to Produce H₂ in ZJU



System Design for 400MW Plant

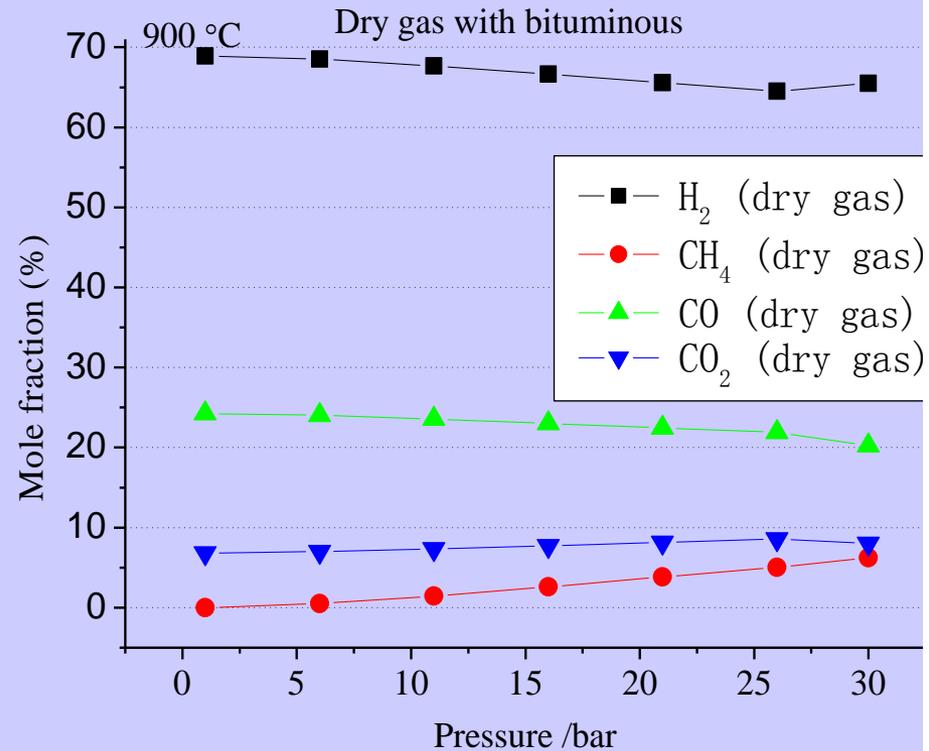
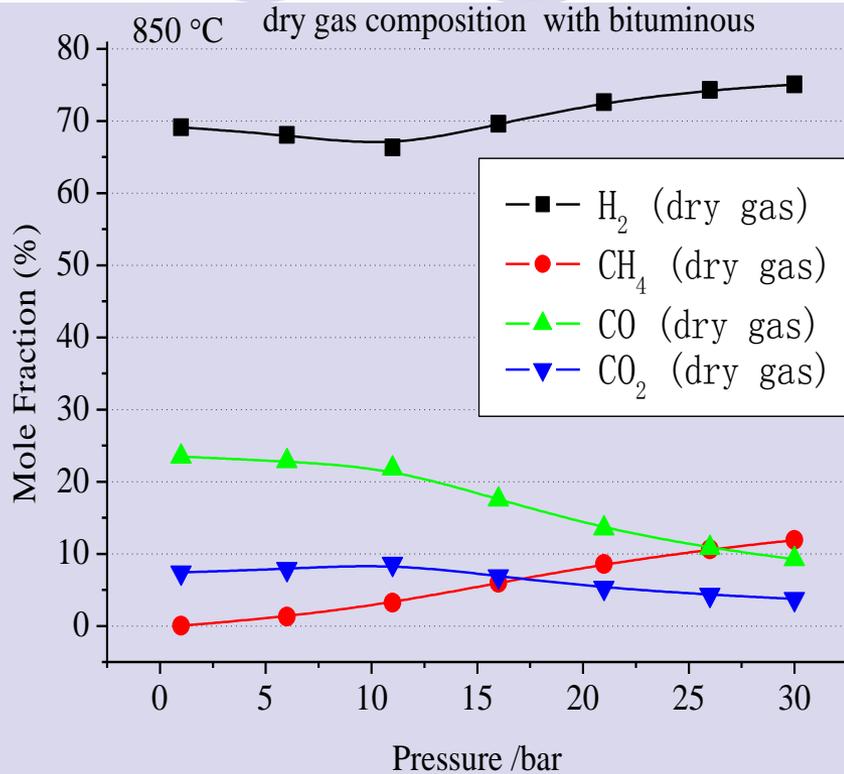
Coal feed rate (kg/s)	18.35
Operation Pressure (bar)	25
Temperature in the gasifier (K)	1123
Heat loss of the gasifier (kJ/s)	3120.93
Coal gasification ratio	0.7
Hydrogen production rate (kmol/s)	1.42
Pressure in combustor (bar)	25
Temperature in the combustor (K)	1223
Limestone supplement (kg/s)	4.25
Fuel utilization ratio in the SOFC (%)	85
Oxygen utilization ratio in the SOFC (%)	51
Power generation from the SOFC (kW)	269208.48
Efficiency of gas turbine cycle (%)	59
Power generation from the gas turbine (KW)	152115.31
Power generation from the steam turbine cycle,kw	3956.7
Power consumption of the air separation (kW)	25955
Power consumption of the air compressor, kw	99821.12
Net Power generation (kW)	413538.88
System efficiency (%)	66.52

Effect of pressure on carbonation reaction rate



Suitable pressure range may be 20-30 bars

Effect of Pressure on Dry Gas Composition

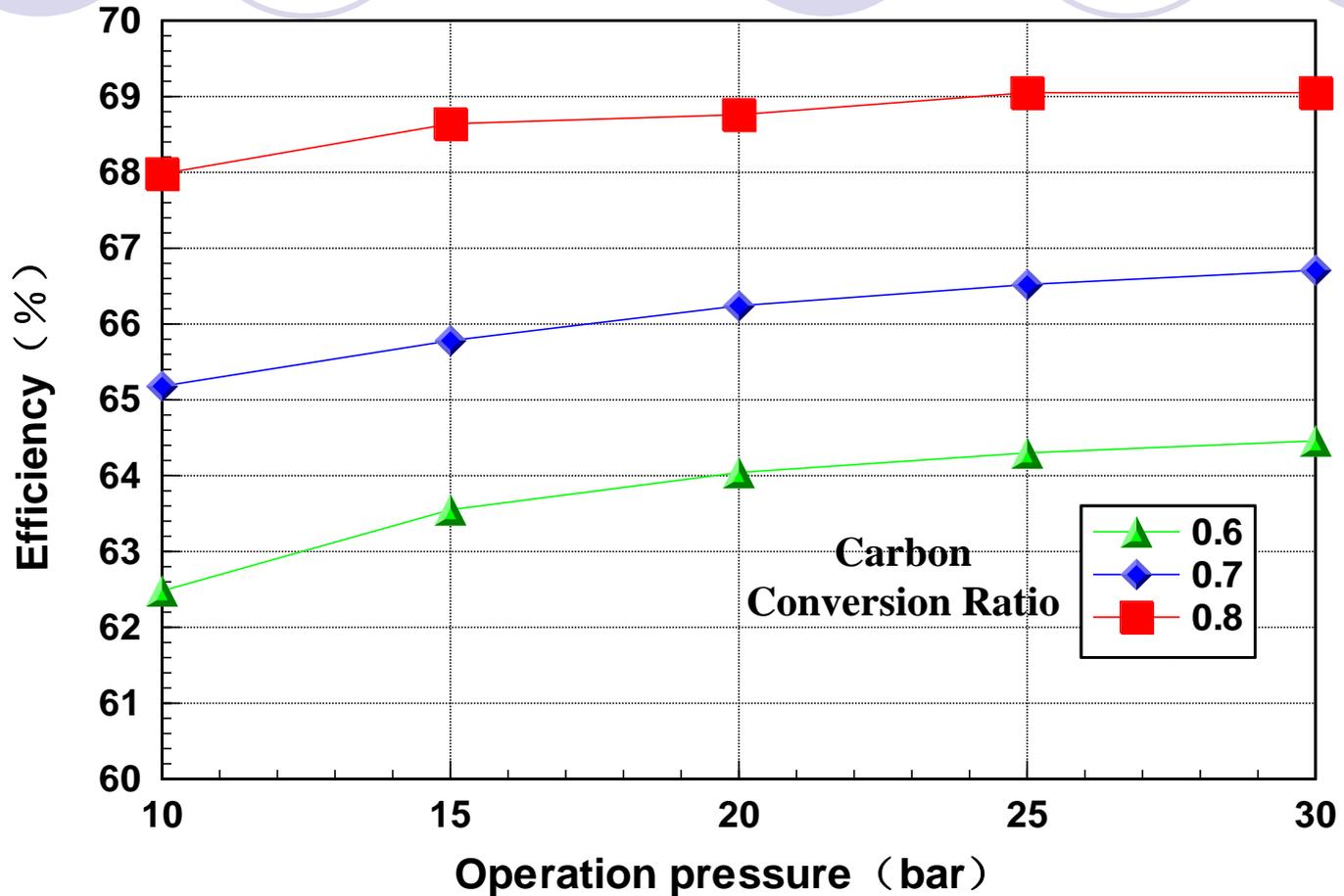


Bituminous coal

Reaction temperature: 850 °C (left) , 900 °C (right)

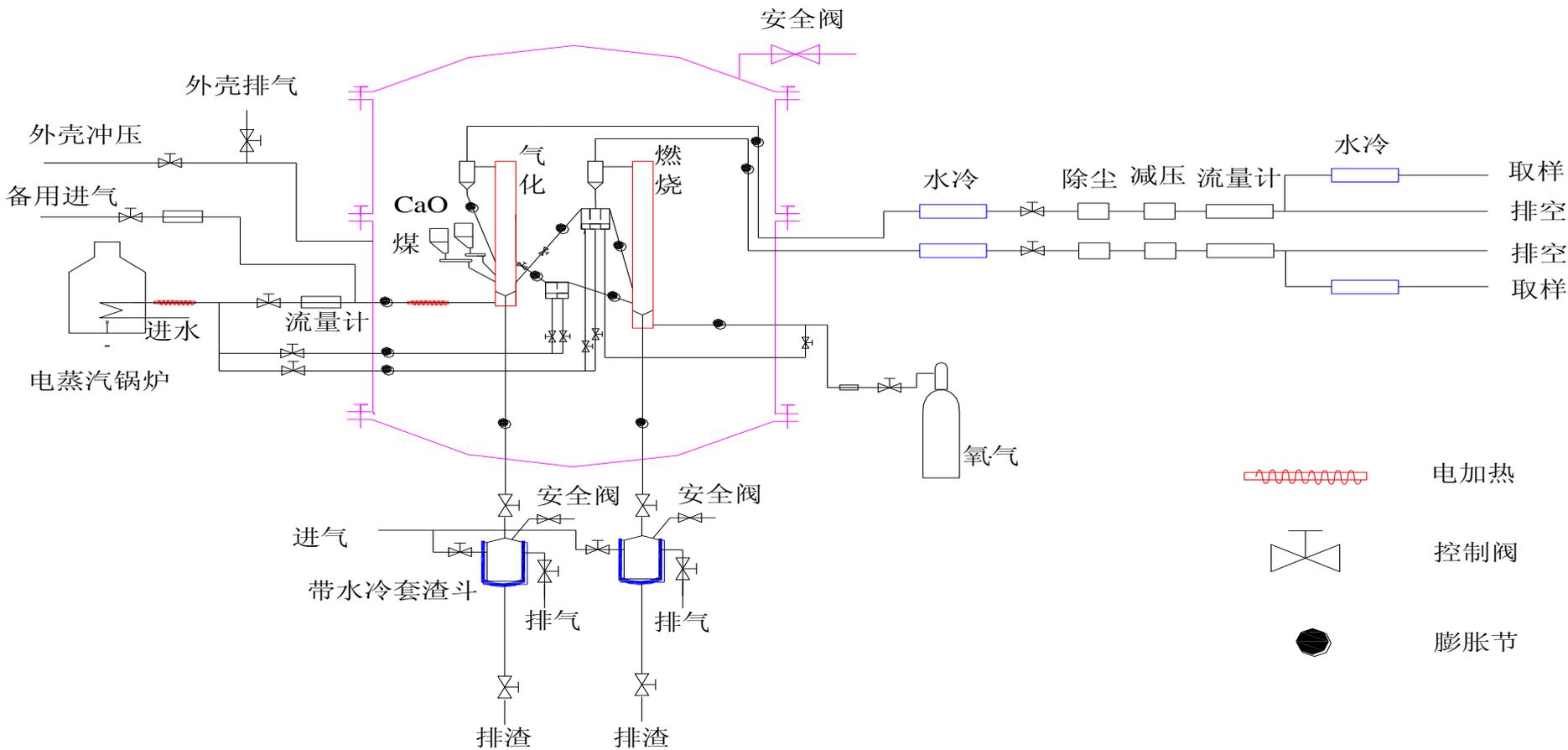
C:H₂O:CaO=1:2:1

Effect of Operation Pressure

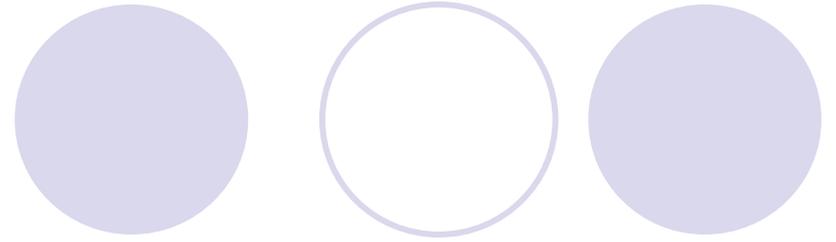
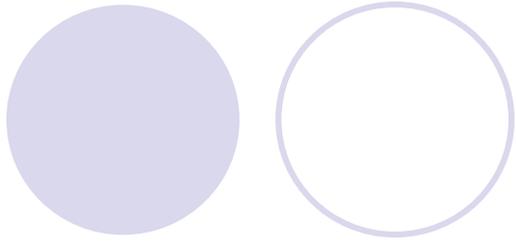


System efficiency versus operation pressure under different carbon conversion ratio

Near Zero Emissions Coal Utilization Facility in ZJU



0-30bar, 1000°C



VI CO₂ Utilization and Fixation

Characteristics of Sequestration Methods

• **Underground Geological Storage**

- **CO₂ can store underground in a long time;**
- **EOR technology can reduce the cost of CCS;**
- **Selection of suitable storage site and monitor of CO₂ leakage are essential.**

• **Ocean Storage**

- **Research phase**
- **Immature Technology**
- **Higher Cost**
- **Unclear Effect of CO₂ on Ocean Ecological Environment in long term**

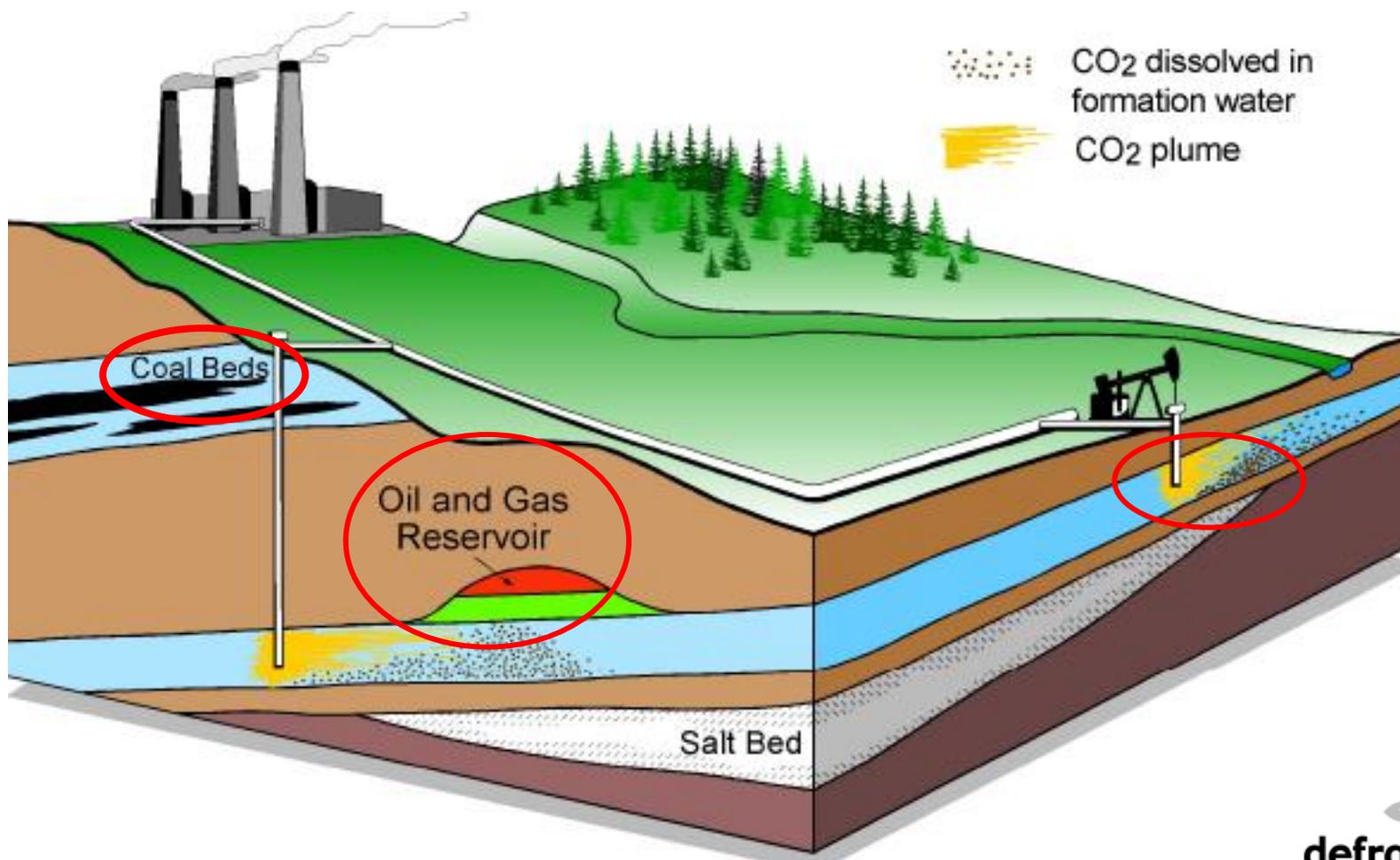
• **Mineral Carbonation**

- **Higher Cost;**
- **Immature Technology;**
- **Long Storage time.**

• **Biological Storage**

- **Short Storage time**
- **Higher Cost**

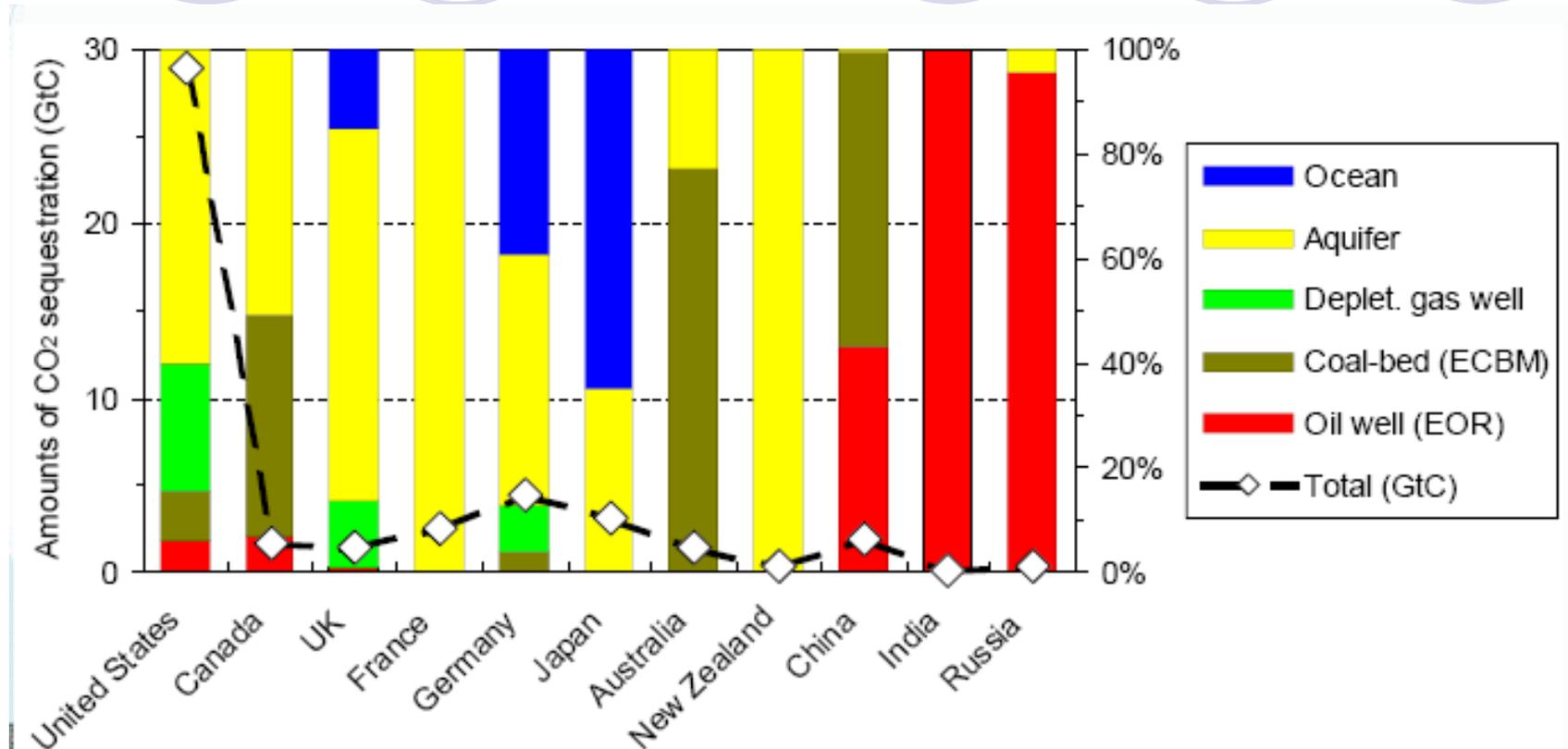
1. Underground Geological Storage in Oil & Gas Reservoirs, Coals & Saline Aquifers



1. Underground Geological Storage

- **Aquifers**
- **Oil Fields** (**CO₂-EOR**, enhanced oil recovery)
- **Natural Gas Fields** (**CO₂-EGR**, enhanced gas recovery)
- **Coal Bed** (**CO₂-ECBM**, enhanced coal beds methane)
- **CO₂ injected into depleted oil & gas fields, aquifers and coal beds which can not be exploited can be stored in a very long time, but the revenue may be low.**
- **The operations that CO₂ was directly injected into the exploiting oil or gas fields and coal seams can not only reduce the CO₂ emissions, but also increase the recovery efficiency of oil, gas and coal seam gas (Win-win situation between Revenue and environmental protection)**

Options and Deploy Scale of Underground Storage in China



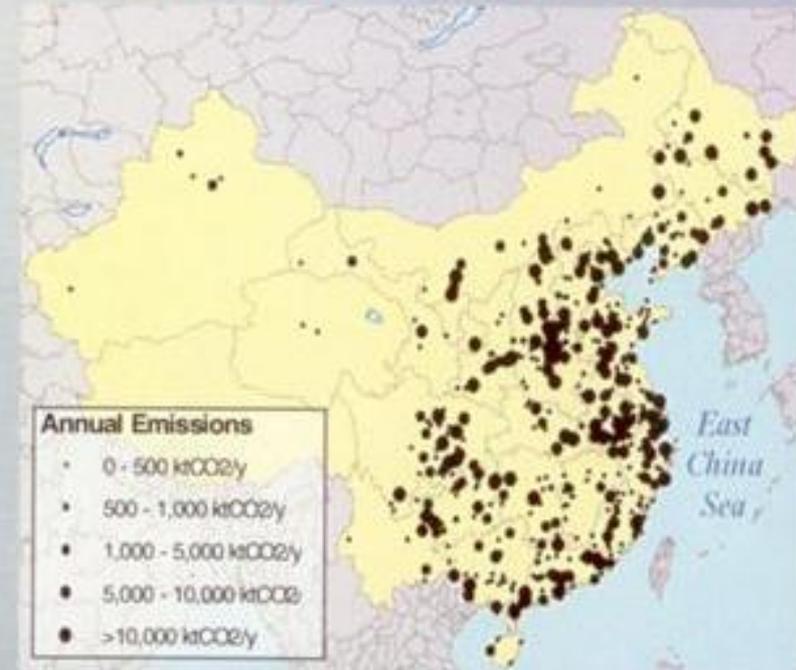
(Cited from Institute of Rock and Soil Mechanics, Chinese Academy of Sciences)

CO₂ Storage Capacity In China

Sources by type



Sources by annual emissions

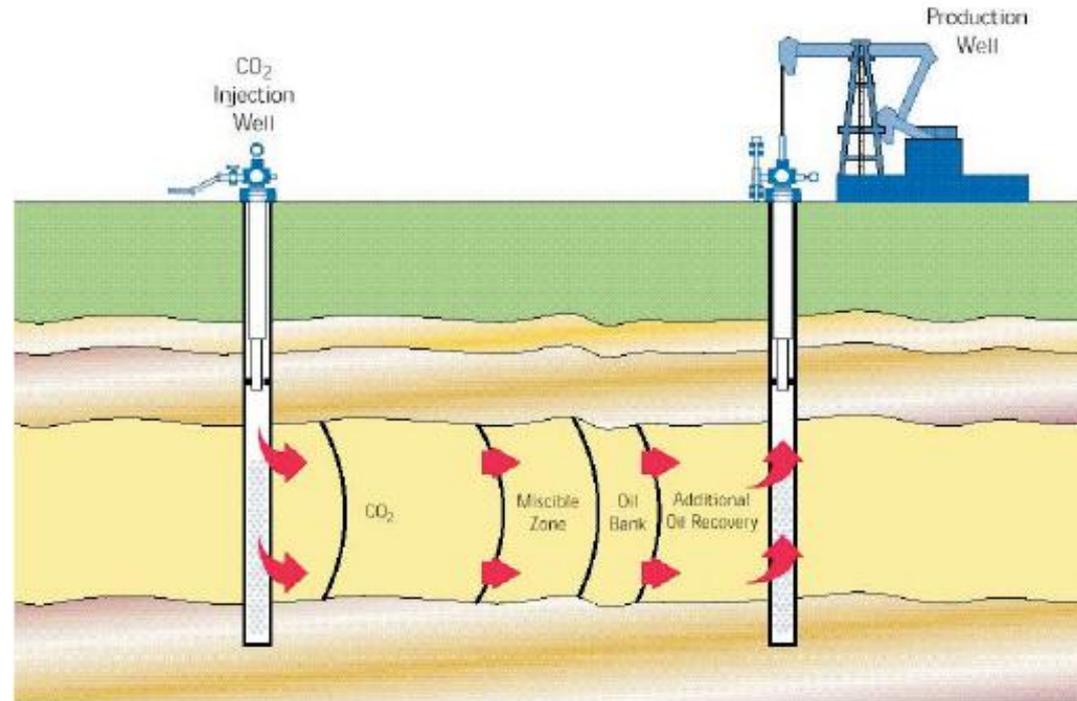


26 GtCO₂ storage capacity identified in the literature within Chinese coalbeds, depleted oil and gas fields

360 GtCO₂ storage capacity in deep saline formations in China estimated by Dooley and Friedman (2004)

CO₂ Enhanced Oil Recovery (EOR) in China

- CO₂ separated is mainly applied in enhanced oil recovery (EOR).
- EOR can not only increase the oil field production, but also prolong the oil field production period.
- Study have been made on EOR in China.



Research of CO₂-EOR in China

Project	Scale of project	Start date	Storage type
Daqing Oil Field	Pilot	1990	CO₂-EOR
Jiangsu Oil Field	Pilot	1998	CO₂-EOR
Liaohe Oil Field	Pilot	2001	CO₂-EOR
Zhongyuan Oil Field	Pilot		CO₂-EOR
Jilin Oil Field	Pilot	1994	CO₂-EOR
Shengli Oil Field	Pilot		CO₂-EOR

Application Project of CO₂ -EOR in China

- ◆ December, 2006. CO₂ was injected into the No. 9711 and 9117 wells to recover oil by **Gas Production Branch of Daqing Oil Field**.
- ◆ 12nd, June, 2007. In the Tuha oil field, **CO₂ with 90 m³** volume was successfully injected into the oil field to recover oil. The operation was lasted about **2 hours**.
- ◆ 12nd, November, 2007. In the **Kongdian Oil Field** which is affiliated with Dagang Oil Field Company, CO₂ was successfully used to enhance the oil yield at the Yang 7-13-2 well. The operation was lasted about **1.5 years**, and the oil output was increased from **2 ton/day to 10 ton/day**.
- ◆ January, 2007. The best mining technology was awarded to **Liaohu Petroleum Exploration Bureau** because of their enhanced oil recovery technology using flue gas, which can increase the oil recovery by **10%~20%**.

Application Project of CO₂ -EOR in China



August, 2007. The project, “**Research on Exploitation of Natural Gas with Higher CO₂ Concentration, CO₂ storage and Comprehensive Utilization of Resources in Jilin Oil Field**”, was started.



Jilin Oil Field, China National Petroleum Corporation

Application Project of CO₂ -EOR in China

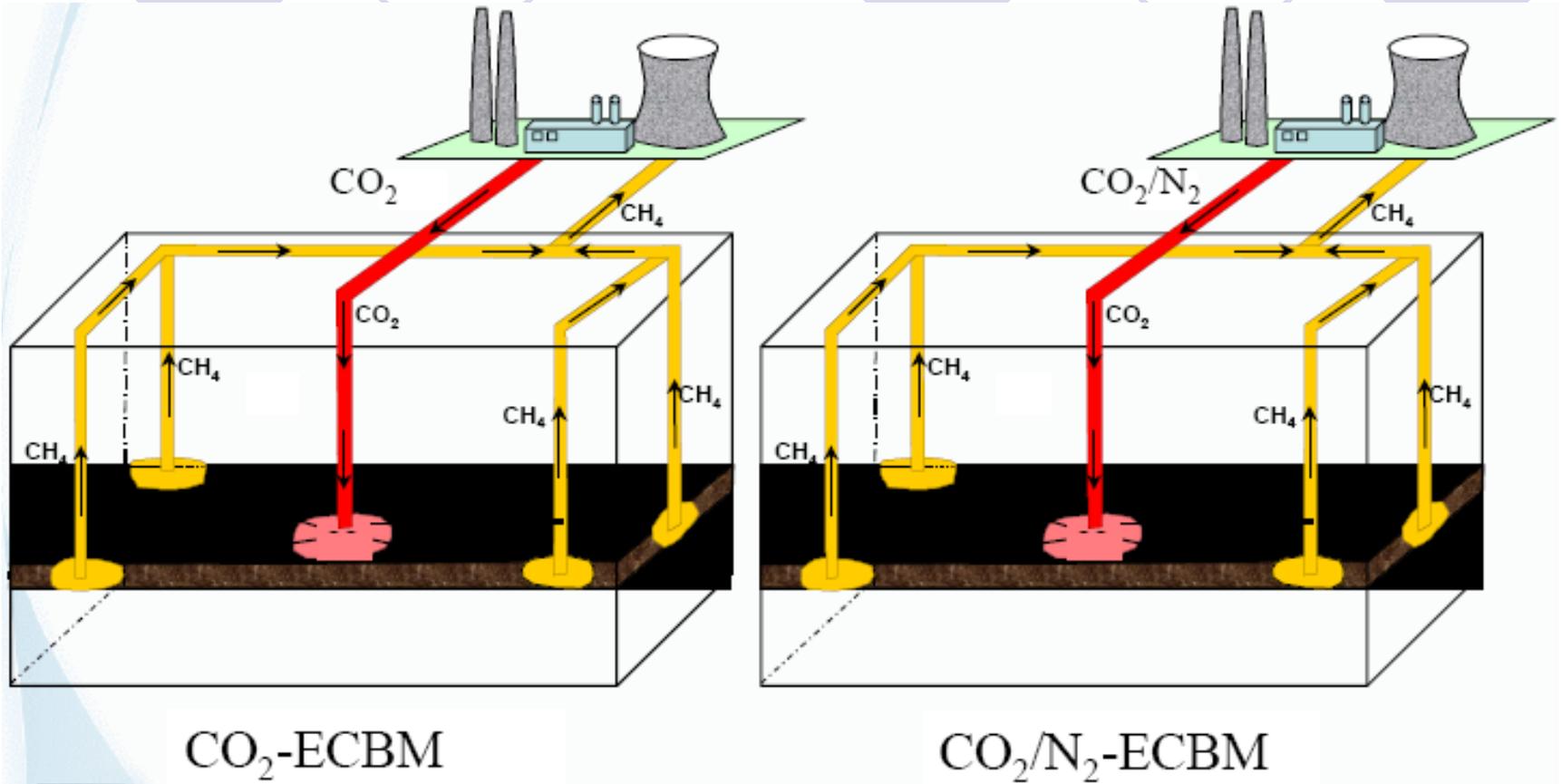


**Zhongyuan Oil Field,
SINOPEC**

February, 2008. The experimental project of CO₂-EOR in Zhongyuan Oil Field was started. The project will be implemented by Exploration and Exploitation Institute, Petroleum Production Institute and Second Petroleum Production of Zhongyuan Oil Field.

Goals: Using CO₂ to enhance the oil recovery.

CO₂-ECBM Concept



(Cited from Institute of Rock and Soil Mechanics, Chinese Academy of Sciences)

CO₂ Underground Geological Storage in China

Project	Scale of project	Start date	Storage type
Daqing Oil Field	Pilot	1990	CO₂-EOR
Jiangsu Oil Field	Pilot	1998	CO₂-EOR
Liaohu Oil Field	Pilot	2001	CO₂-EOR
Zhongyuan Oil Field	Pilot		CO₂-EOR
Jilin Oil Field	Pilot	1994	CO₂-EOR
Shengli Oil Field	Pilot		CO₂-EOR
Qinshui Coal Field	Pilot	2003	CO₂-ECBM
Pingdingshan Coal Field	Pilot	2006	G-ECBM

2. Other CO₂ Utilization Technologies

- ***Physical Applications:*** CO₂ can be used to act as (1) the refrigerating medium; (2) extinguishing agent; (3) supercritical extractant to purify the Chinese traditional medicine and spice; (4) working substance in low-temperature heat power station. CO₂ also can be used for food storage and gas shielded welding.
- ***Chemical Applications:*** CO₂ may be the perfect materials which can be used to synthesize the new chemical compounds such as urea, methanol, pure caustic soda, polycarbonate, salicylic acid and propylene carbonate, synthesize the liquid hydrocarbon combined with CH₄ and synthesize the gasoline combined with H₂. CO₂ also can be used in water treatment.
- ***Biological Fixation:*** Using some suitable algae (microalgae) and bacterium to fix CO₂.

CO₂ Productions



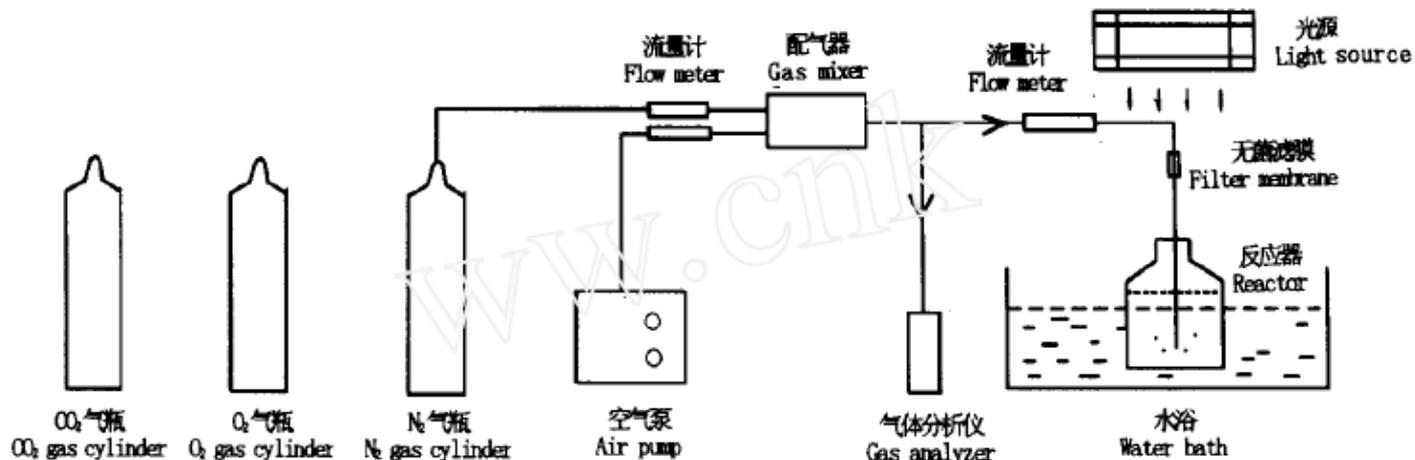
Carbon dioxide
copolymer

Ammonia
bicarbonate

Sodium
carbonate

CO₂ Fixation by Microalgae

- **Zhejiang University** have screened the blue algae and green algae under high temperature and high concentration CO₂ and studied the highly active membrane-based algae-light reactor.
- **Northeastern University** studied the CO₂ separation by algae and cultured the algae mixture samples using flue gas. They also isolated the useful algae species: ZY-1.



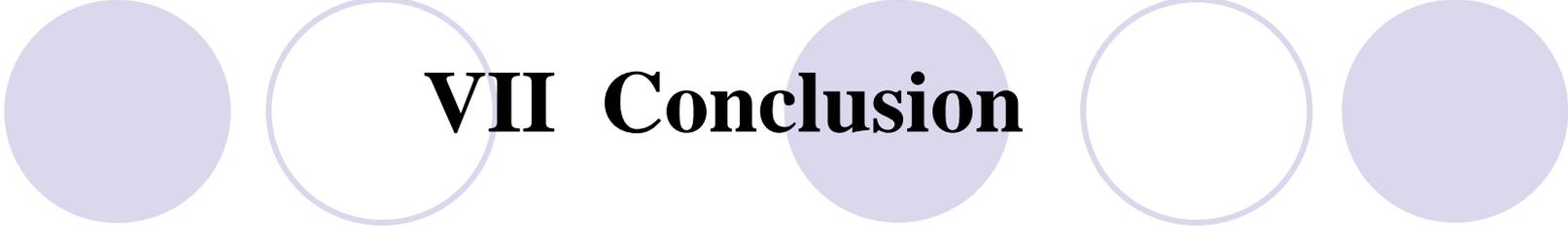
培养特性测试实验装置图

Diagram of the experimental apparatus of determining the cultural characteristics

CO₂ as Fertilizer of Plant

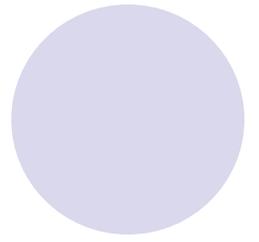
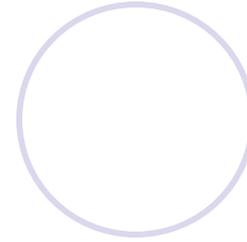
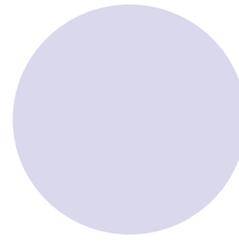
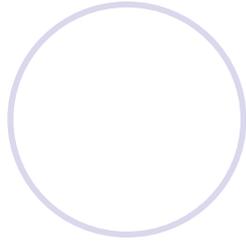
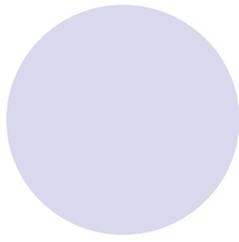


If CO₂ concentration increases from 355 ppm to 675 ppm, tomato yields are increased by about 30%, sugar beet yields are increased by about 22%.



VII Conclusion

- **Many researches on CO₂ capture have been made and many fruits have been got in China;**
- **Few demonstration projects of CCS in China;**
- **Demonstration project must be supported by government.**
- **International cooperation on CCS need to be enhanced.**



Thank you for your attentions!