

# MMV Support for SECARB Phase II Field Demonstration Projects

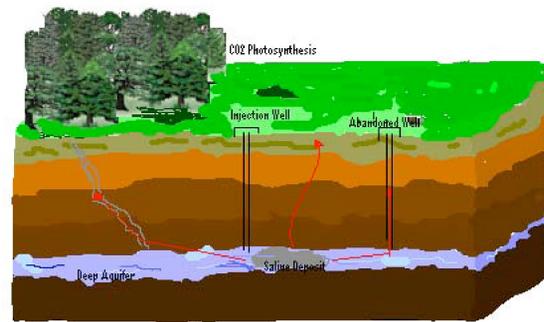


Jeff Lindner, John Luthe, F-X. Han, Teresa Leone  
Institute for Clean Energy Technology at Mississippi State University



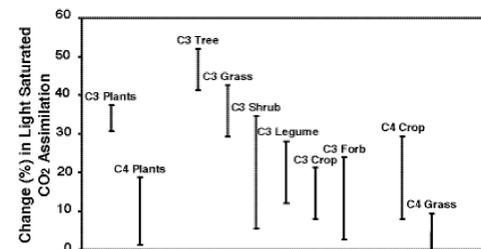
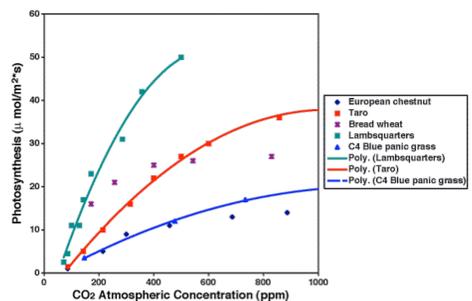
## Identify Emerging Technologies to Fill MM&V Gaps

- Review of techniques applicable to quantification of CO<sub>2</sub> (reservoir, groundwater, soil, atmosphere).
- Identification of novel measurement techniques.
- Evaluation of instrumentation required for pipelines.
- Development of an MM&V relational database.
- Background on spatial and temporal variations of CO<sub>2</sub> (atmospheric and soil gas).



## Major MM&V Challenge - Accurately Identifying and Quantifying CO<sub>2</sub> Seepage

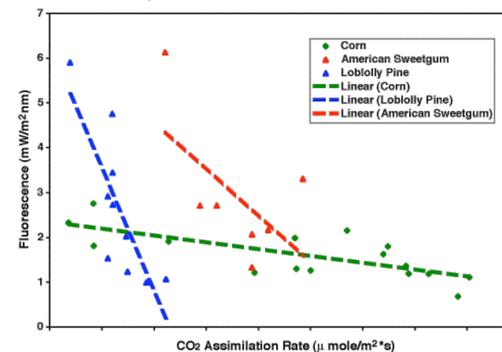
- Coefficients of variation of natural CO<sub>2</sub> flux (seasonal and diurnal variations) over limited areas have been found in the range of 25 - 85%.
- A large number of samples are required to achieve an accurate measurement with an acceptable precision.
- CRDS and tracers with portable GC, GC/MS measurement frequency will be limited until market-driven forces become active.



Photosynthetic response to elevated CO<sub>2</sub> concentrations in various C<sub>3</sub> and C<sub>4</sub> Plants; greenhouse simulations, growth chambers and open-top chambers (left panel)<sup>1,2</sup> and free air CO<sub>2</sub> enrichment (FACE) conditions (right panel).<sup>3</sup>

## Use of Fluorescence to Monitor Enhanced Plant Growth and Locate Seeps (concept in development)

- Chlorophyll fluorescence - carbon dioxide fixation
  - climate change
  - terrestrial sequestration.
- Instruments commercially available (lab and field).
- Measurements less complicated than traditional photosynthesis measurements (leaf respiration).
- Fluorescence inversely correlated to plant carbon assimilation.<sup>4,5</sup>
- More CO<sub>2</sub> incorporated into the plant reduces fluorescence as the rate of electron transport is lowered.



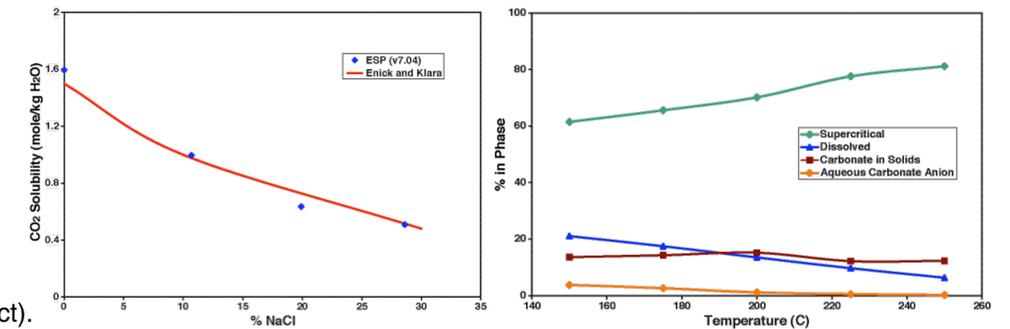
## Support Field Validation MM&V

- Down-selection of conservative tracers for ECBM (Black Warrior project).
- Physical properties<sup>6</sup> (solubility, vapor pressure, Mw, B.P., M.P.) of fluoro-carbon (and other) molecules that have been employed in oil recovery and sequestration studies were compared with anticipated seam conditions (150 psi, 50°C).
- CO<sub>2</sub> within the thin coal seam will exist in the vapor phase.
- Desired tracer has limited solubility, is already in the gas phase and will exhibit minimal adsorption on the coal.
- Down-selected tracer candidates will be analyzed using coal from actual core samples.

Tracer	MW (g/mol)	B.P. (°C)	Density (kg/L) STP	Solubility in H2O STP ln(x <sub>1</sub> ) (mole fraction)
CO <sub>2</sub>	44	-78.5	0.00198	-1.311
CH <sub>3</sub> F	34	-78.4	0.5786	-6.725
CH <sub>2</sub> F <sub>2</sub>	52	-51.6	0.909	-7.301
SF <sub>6</sub>	146	-63	0.0066	-12.186
SF <sub>5</sub>	196	-21		
C <sub>2</sub> F <sub>6</sub>	138	-79	1.59 (-78°C)	-13.786
C <sub>3</sub> F <sub>8</sub>	188	-36.7	0.0082L	-14.395 (15°C)
PCB	200	-6	1.724	-12.814
PDCB	300	45	1.62	-12.31 to -21.52 (see text)
PMCP	300	50	1.707	-12.31 to -21.52 (see text)
PMCH	350	75	1.788	-12.31 to -21.52 (see text)
PDCH	400	102	1.828	-12.31 to -21.52 (see text)
PECH	400	101.7	1.788	-12.31 to -21.52 (see text)
PTCH	450	127	1.888	-12.31 to -21.52 (see text)
IPPC	450	130		-12.31 to -21.52 (see text)
nPPCH	450	130		-12.31 to -21.52 (see text)

## Support Field Validation MM&V

- Thermodynamics - model validation activities.
- Environmental Simulation Program (ESP, OLI Systems Inc.)<sup>7</sup>
  - interphase (solid-liquid-vapor) and intraphase equilibria (redox, co-precipitation)
  - extensive unit operations.
- Model accurately predicts CO<sub>2</sub> solubility in brine and phase behavior.<sup>8</sup>
- Survey calculations based on the work of Kaszuba, et al.<sup>9</sup>
- Solids predicted (200°C, 200 bar) magnesite (MgCO<sub>3</sub>), siderite (FeCO<sub>3</sub>), CaMg<sub>3</sub>(CO<sub>3</sub>)<sub>4</sub>, SiO<sub>2</sub>, KAl<sub>3</sub>(OH)<sub>6</sub>(SO<sub>4</sub>)<sub>2</sub>.
- Magnesite, analcime (NaAlSi<sub>2</sub>O<sub>6</sub>·H<sub>2</sub>O), and smectite (clays) were observed.
- Work on incorporating kinetics into the model is in progress.



## References

1. Luo, et al. 1996. *Global Biogeochemical Cycles* 10:209-222.
2. Ghannoum, et al. 2000. *Plant Cell Environment* 23:931-942.
3. S.P. Long, et al. 2004. *Annual Review of Plant Biology* 55:591.
4. A. Freedman, et al. 2002. *Photosynthetica* 40:127.
5. G.A. Carter, et al. 1990. *Plant, Cell and Environment* 13:79.
6. CRC, NIST Chemistry WebBook, <http://webbook.nist.gov/chemistry/>
7. <http://www.olisystems.com/>
8. R.M. Enick and S.M. Klara. 1990. *Chemical Engineering Communication* 90:23.
9. J.P. Kaszuba, et al. 2003. *Applied Geochemistry* 18:1065-1080.

## Acknowledgements

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