

ALTERNATIVE CONCEPTS FOR SUPPLYING CARBON
DIOXIDE FOR ENHANCED OIL RECOVERY PROJECTS

FINAL REPORT

Date Published—July 1980

Work Performed for the Department of Energy
Under Contract No. EW-78-C-21-8333

Science Applications, Inc.
Golden, Colorado



**Bartlesville Project Office
U. S. DEPARTMENT OF ENERGY
Bartlesville, Oklahoma**

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Available from the National Technical Information Service, U.S. Department of Commerce, Springfield, Virginia 22161.

NTIS price codes

Paper copy: \$11.00

Microfiche copy: \$ 3.50

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Final Report

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U.S. DEPARTMENT OF ENERGY

TABLE OF CONTENTS

	PREFACE	iii
	LIST OF FIGURES	iv
	LIST OF TABLES	iv
I.	INTRODUCTION	1
	CO ₂ Requirements Identified Sources	4
	Status of CO ₂ Supply Projects	7
	Status of Processing Natural Gas/CO ₂ Streams	9
II.	APPROACH	12
	Approach to Analysis of Processing Alternatives	12
	Approach to Analysis of Transportation Alternatives	17
	Approach to Analysis of Changes in CO ₂ Reserves	19
III.	RESULTS	20
	Processing Alternatives	21
	A - Processes for Natural Gas Streams Containing Less than 5 Mole % CO ₂	21
	B - Processes for Natural Gas Streams Containing 5-30 Mole % CO ₂	30
	C - Processes for Natural Gas Streams Containing Greater than 30 Mole % CO ₂	30
	Transportation Alternatives	38
	Thermodynamic Properties Affecting Pipeline Design	39
	Transportation Mode Optimization	48
	Pipeline Transport - Trunk Line	49
	Truck Transport	53
	Truck vs Lateral Pipeline	59
	Impact on CO ₂ Reserves & Recoverable Oil	60
IV.	CONCLUSIONS & RECOMMENDATIONS	61
	Conclusions	61
	Conclusions from the Processing Alternatives	61
	Conclusions from the Transportation Alternatives	62
	Conclusions on CO ₂ /EOR Reserves	62
	Recommendations	63

V.	REFERENCES	64
	Figures	66-81
	Tables	82-117
	Appendices	
	A. Tractor and trailer requirements and costs for CO ₂ distribution by truck from a trunk ² pipeline	A-1
	B. Calculation of the upstream pressure and the amount of work expected out of the turboexpander	B-1

PREFACE

A study of alternative concepts for supplying carbon dioxide for enhanced oil recovery projects was undertaken by Science Applications, Inc. The work was funded under the System Support Services Contract with the Morgantown Energy Technology Center of the Department of Energy (Contract No. EW-78-C-21-8333, Task - 4). This report describes the conduct and the findings of this study. The purposes of the study were to:

- Define the CO₂ requirements for candidate EOR reservoirs.
- Generate alternative concepts for collecting, transporting, storing and processing CO₂ from naturally occurring sources.
- Compare the alternatives on the basis of cost effectiveness.
- Reestimate the magnitude of the CO₂ EOR reserve base on the results of the above findings.
- Recommend future DOE program activities.

The results of the study reported herein fulfill the above stated objectives. The results, opinions, and recommendations expressed in this report are those of Science Applications, Inc. and not necessarily those of the U.S. Department of Energy.

FIGURES

1	Gas Treating for CO ₂ Removal	66
2	Schematic Process Flow Diagram for Natural Gas Streams	67
3	Equilibrium Liquid-Vapor Ratios for CH ₄	68
4	Equilibrium Liquid-Vapor Ratios for CO ₂	69
5	Equilibrium Liquid-Vapor Ratios for CO ₂ (Calculated)	70
6	Equilibrium Liquid-Vapor Ratios for CH ₄ (Calculated)	71
7	Convergence Pressure for CH ₄ at -50°F	72
8	Convergence Pressure for CH ₄ at 0°F	73
9	Convergence Pressure for CH ₄ at 50°F	74
10	Critical Pressure vs. Critical Temperature For Mixtures of Methane and Carbon Dioxide	75
11	Carbon Dioxide Thermodynamic Properties: Pressure-Enthalpy Diagram	76
12	Initial Pressurization of CO ₂	77
13	Initial Pressurization of CO ₂ Compressor Horsepower Comparison	78
14	Temperature Distribution in 50 MMSCFD CO ₂ Pipelines	79
15	Pipeline Pressurization and Thermal Conditioning of CO ₂	80
16	Comparison of Truck and Lateral Pipeline CO ₂ Transport Costs	81

TABLES

1	Summary of CO ₂ Supply Projects	82
2	Comparison of Some Commonly Used Gas Purification Processes	83
3	Composition, Heating Values, Specific Gravity, Wobbe Number of Natural Gas Samples From Various States, Counties, and Fields in the U.S.A.	84
4	Truck Loading @ 300 PSI and 90°F	85
5	Truck Loading @ 300 PSI and 0°F	86
6	Truck Loading @ 1000 PSI and 90°F	87

TABLES (CONTINUED)

7	Pipeline @ 2000 PSI and 140°F	88
8	Total Capital and Operating Costs for Various Options	89
9	Cost of CO ₂ at the Point of Consumption	90
10	Results of Flash Calculations For: 30 Mole Percent CO ₂ in Feed	91
11	Results of Flash Calculations For: 35 Mole Percent CO ₂ in Feed	92
12	Results of Flash Calculations For: 40 Mole Percent CO ₂ in Feed	93
13	Results of Flash Calculations For: 45 Mole Percent CO ₂ in Feed	94
14	Results of Flash Calculations For: 50 Mole Percent CO ₂ in Feed	95
15	Comparison of Vapor-Liquid Equilibrium Ratio For Methane From Various Sources	96
16-21	Calculation of Heat Capacity and l/M For Various Compositions of Methane and Carbon Dioxide Under Various Conditions of Temperature and Pressure	97
22	Comparison of CO ₂ Pipeline Repressurization and Thermal Conditioning Requirements	107
23	Comparison of CO ₂ Pipeline Repressurization Power and Thermal Conditioning	108
24	Summary of Transportation Costs	109
25	Summary of CO ₂ Pipeline Designs	110
26	Trunk Pipeline Calculations	111
27	Lateral Pipeline Calculations	112
28	Summary of Tractor and Trailer Requirements For CO ₂ Transport	114
29	Development of Annual Tractor Unit Cost	115
30	Development of Annual Trailer Unit Cost	116
31	Summary of Transport Costs for Truck Delivered CO ₂ (\$/MCF) 300 PSIA, 0°F, Saturated Vapor	117

I. INTRODUCTION

The United States has reached a critical point in its energy consumption patterns. The reliance on liquid petroleum for nearly fifty percent of its energy supply and the enormous growth in energy consumption over the past two decades have outstripped the ability to produce from domestic sources. The shortfalls in domestic petroleum production have been overcome by the importation of crude oil from foreign countries, to the point where the U.S. currently imports more crude and refined products than it produces. The 1974 oil embargo by the Arab states and the 1979 supply interruptions caused by the political unrest in Iran have left their marks on the national economic and social consciousness. The high rate of crude importation has caused an unfavorable balance of payments and an unprecedented inflation rate in the U.S., and has placed the national security in serious jeopardy.

There has been a concerted effort to arrest the decline of national production. Following the 1974 Arab oil embargo, the oil industry embarked on an intensive drilling program in the lower 48 states. The active rotary rig count has averaged more than 2,000 units over that period of time, and has recently reached a 30 year record of over 2,500 units. In other words, since 1974, approximately 70% of the world's drilling resources have been devoted to improving the petroleum reserve and production picture of the continental United States. Unfortunately, an improved picture has not evolved. In 1978 the nation's drawdown of proven reserves was seven times as great as the establishment of new reserves. Texas and other critical producing states experienced record declines in production rates.

In summary, the nation is faced with three elements of a perplexing and critical energy problem:

- Our energy utilization infrastructure is heavily dependent on products derived from liquid petroleum. Significant alterations of this infrastructure (such as massive shifts to coal utilization or nuclear energy) would require huge investments in capital and elapsed time.
- Our declining petroleum production and domestic reserve situation are falling short of meeting the demand for petroleum products by a factor of two.
- The shortfall between domestic demand and supply is being accommodated by the importation of petroleum produced in foreign countries, resulting in intolerable consequences to the economic and strategic well being of the United States.

Coupled with these facts is the realization that less than one-third of the original oil-in-place is produced by presently used primary and secondary recovery techniques. It is evident that improvements in the percentage recovery of the oil-in-place could significantly improve the U.S. energy supply picture.

The production techniques that are applied after completion of secondary recovery operations or to heavy oil reservoirs is referred to as enhanced oil recovery (EOR) methods. The methods are normally characterized as thermal, chemical, or gas miscible recovery processes.

The specific resource target for EOR is 250 to 300 billion barrels of light and heavy crude left in reservoirs after the utilization of conventional production methods. This is one of the few hydrocarbon resources that can be utilized without causing massive perturbations to the energy utilization infrastructure. With successful tech-

nology development, as much as 50 billion of the 300 billion barrels could be recovered. (REF.1) As much as 21 billion of the 50 billion barrels could be recovered by the gas miscible process using carbon dioxide as the injected gas.

Despite EOR's potential, its development and implementation have been slow to materialize. Only 400,000 bbl/day of the nation's 8,400,000 bbl/day production can be attributed to the implementation of EOR processes. The vast majority of this production results from thermal recovery methods applied to heavy oil reservoirs which have experienced little to no primary production. The reasons for the slow progress of EOR implementation are multifaceted. In some cases, the technology is not sufficiently mature, in other cases the economics are unattractive at the present risk level. The implementation is also hindered by institutional constraints.

For example, in highly integrated oil companies, EOR projects must compete for investment funds with projects that vary from the acquisition of coal and uranium properties to the acquisition of electric motor manufacturing operations. Until the perceived economic risk of EOR relative to other investment options is substantially reduced, EOR's potential for reducing the nation's energy dependence on foreign sources will never be realized.

The principal risks that are associated with the CO₂ miscible process are:

- Incomplete understanding of the behavior of CO₂ after it is injected into representative reservoir fluids.
- Inability to contact all of the reservoir due to viscous fingering, poor mobility control, and gravity segregation.
- Insufficient quantities of low-cost carbon dioxide near the candidate reservoirs.

It is the last stated risk that was addressed in the study reported herein.

CO₂ Requirements - Identified Sources

Several attempts have been made since the 1974 Arab oil embargo to estimate the potential for production of petroleum in the United States by utilizing enhanced oil recovery techniques. (REF. 1-3) The most recent estimate by the Office of Technology Assessment of the U.S. Congress (REF. 1) is that under the most favorable sets of assumptions up to 51 billion barrels of additional oil could be produced using EOR techniques. The most favorable assumptions were that:

- EOR technology would improve considerably over current levels.
- The prices would be virtually unconstrained.
- The injected fluids would be available in the quantities needed at "reasonable" prices.

The study projected that up to 21.1 billion of the 51 billion barrels of oil produced would be recovered by the CO₂ miscible EOR technique. The specific technology assumption that accompanied that projection was that the oil recovery efficiency of CO₂ would be improved from the current value (approximately 10 to 7 MCF/barrel of oil recovered) to 5.1 MCF/barrel. To achieve the oil production estimate of 21.1 billion barrels a total of 107.6 trillion SCF would be required. Since approximately 25% of the CO₂ requirements could be met by reinjecting some of the CO₂ that is co-produced with the tertiary oil (also assumed in REF. 1) a net of 80.7 TSCF of CO₂ would be required for the initial injection into tertiary oil reservoirs.

The cost and availability of CO₂ from various sources within the United States have been estimated in the previously conducted studies. (REF. 4-6) On the basis of the reserves stated in these references or assuming a twenty year production of the rates reported in these references the following conclusions were drawn by the writers:

SUMMARY OF CO₂ RESERVES

Natural sources containing high concentrations of CO ₂ (>80%)	30 TSCF
Process vents containing high concentrations of CO ₂	5 TSCF
Flue gas stacks containing low concentrations of CO ₂ (5 < CO ₂ < 17%)	182 TSCF

On the surface of the figures shown in the above table one can conclude there are sufficient quantities of CO₂ available to meet the needs of future EOR projects. Unfortunately, the most voluminous source is, for the most cases, very costly. The costliness stems from the very high cost associated with separating the low concentrations of CO₂ from the products of combustion and the high cost of transporting the CO₂ from large fossil fuel burning installations to candidate petroleum reservoirs. The second cost contributor results from the fact that most large flue gas sources are electrical generating facilities and that these facilities are located near densely populated regions of the U.S. and not near oil producing regions.

Although estimates of the delivered cost vary widely the OTA study indicates that the cost of recovering CO₂ from flue gas is more than twice that required to recover CO₂ from natural reservoirs. It is reasonable to conclude

that the availability of only 35 TSCF from less costly sources could conceivably reduce the amount of oil recovered by the CO₂ miscible process by over a factor of two. It was this observation that prompted the initiation of the study reported herein. The question asked was "Is there another source of CO₂ available that is more attractive than removing CO₂ from flue gases?". The answer to the question was an assessment of the quantity of the CO₂ that could be recovered from the production and processing of natural gas.

Looking at natural gas production as a potential source of CO₂ was appealing for the following reasons:

- Even at low CO₂ concentrations the sheer magnitude of natural gas production could result in a large source of CO₂. (Example: If the known reserves of natural gas (approximately 200 TSCF) contained 5% CO₂ the results would be 10 TSCF of CO₂.)
- The cost of separating the CO₂ from the natural gas could be partially borne by the revenue generated from the sale of natural gas.
- Since natural gas and petroleum are normally produced in the same general regions the transportation distance between CO₂ sources and tertiary oil reservoirs would in most cases be less than the distance from the reservoirs to large electrical generating plants.
- The advent of significantly higher natural gas prices and the depletion of older reservoirs of "sweet" natural gas have made production from reservoirs containing higher concentrations of CO₂ and H₂S more attractive than in the past (witness recent development in the Western Overthrust Belt and Louisiana's Tuscaloosa Trend). Estimates that over 200 TSCF of natural gas could be produced from the Western Overthrust Belt which extends from

the Canadian to the Mexican border. The CO₂ content has been relatively high but there is insufficient production data to confirm the average CO₂ content. The probable reserve forecast for the Tuscaloosa Trend is over 80 TSCF. Several wells have produced in excess of 10% CO₂ at very high total flow rates. Additional drilling and production will be required to firm up both the extent of the natural gas reserves and the content of the CO₂ in the gas. Any CO₂ that would be separated from either of the plays could find application to reservoirs in relatively close proximity. They also share the advantage of new, high capacity reservoir systems whose productive lives coincide with the need for CO₂ for EOR projects.

Status of CO₂ Supply Projects

The CO₂ supply projects discussed in the following pages are those that are associated with large scale EOR projects. There have been numerous CO₂ pilots and mini tests which have been serviced with temporary CO₂ supply systems. There are nine CO₂ supply projects that have been announced in open literature. The status of these projects range from fully operational to the preliminary planning stages (Table 1). It is useful to analyze the projects to test the validity of the assumptions that have been made in previous supply studies. The first pipeline, SACROC, is the only one that is in operation. It has been in service since the early 1970's transporting a 90% CO₂, 10% CH₄ mixture from the Val Verde basin in Southerwestern Texas to the Kelly-Snyder Field (Canyon Reef Reservoir) in Scurry County, Texas and the North Crossed Field in Upton County, Texas. Virtually all of the operational experience that exists has been gained on the SACROC project. (Reference 7)

The next four projects are transporting CO₂ from large

natural reservoirs in Southern Colorado, Northeastern New Mexico and North Central Mississippi to large reservoirs in West Texas and Central Mississippi. It is interesting to note that these CO₂ producing provinces were the ones identified in Reference 5 as having the most attractive near-term utility to large CO₂ EOR projects. The producing provinces included:

- o McElmo Dome - Doe Canyon, Colorado (Shell)
- o Sheep Mountain - Dike Mountain, Colorado (ARCO)
- o Brovo Dome Unit - New Mexico (Amoco)
- o Jackson Dome - Mississippi (Shell)

The next two projects are pipelines that connect CO₂ from process vents (anhydrous ammonia plants) to smaller reservoirs. The characteristics of these lines are lower in flow rates (most anhydrous ammonia plants yield less than 50 MMSCFD) and shorter transport distances than those of the four western lines.

The eighth and ninth supply projects listed derive their CO₂ from the co-production of natural gas and CO₂ where the CO₂ concentration ranges from 15 to 66% by volume.

The trend that is developing with the announcement of these projects is probably indicative of the near future and includes the following characteristics:

- o The coupling of the least expensive naturally occurring sources having high deliverability to the large (in some cases giant) reservoirs such as Canyon Reef, Wasson, North Cowden, etc.
- o The utilization of high concentration process vent sources of CO₂ to flood reservoirs of more modest sizes that are located relatively close to the source of the CO₂.
- o The exclusive use of pipelines as the transport mode.

In all cases for which detailed information has been published, the pipelines have been designed to operate in the super-critical dense-phase-fluid thermodynamic state (i.e. the pressure is greater than the critical pressure, and the temperature is at ground ambient which is usually less than the critical temperature (88°F)).

Status of Processing Natural Gas/CO₂ Streams

There are over fifty well known gas purification processes to date and new ones are being continuously developed. The numerous choices available make the selection of a process applicable to specific situations more difficult. There are many factors that need to be considered when selecting a process for a given sweetening application. Some of these factors are:

1. The types of impurities to be removed from the gas stream
2. The concentration level of these impurities and the degree of removal desired
3. The acid gas selectivity required, if any
4. The volume of gas to be processed and the temperature-pressure conditions at which gas is available
5. The feasibility and desirability of sulfur recovery
6. Relative economic factors of the suitable pressures
7. Other factors such as, track record, ease of operation, depending on location, etc.

Each of the above factors, except #7, is further described in Ref. 15. Depending on the site specific criteria, one or two of the above factors will be critical and affect the selection of a process much more than the others. The factors mentioned in #7 are involved in the final decision making step. If these factors could be digitized and some dollar value put on it, it can be combined into the "Relative Economic Factors" (#6).

The commercially available gas purification processes can be divided into the following general categories:

- A - Processes which involve a reversible chemical reaction between the impurities and the solution used to extract it.
- B - Processes using solvents which physically absorb the impurities.
- C - Solid bed and adsorption processes, where the impurities are either selectively adsorbed on the surface area of the adsorbent, or engage in a chemical reaction by coming in contact with the reagent in the solid bed.
- D - Processes involving separation based on boiling point differences, etc.

Some of the processes which fall into the "A" category are:

Amine Processes:	MEA, DEA, TEA, MDEA, DIPA
New Amine Type Processes:	Diglycolamine, Alkanolamine, SNPA-DEA, Alkazid, ADIP, Sulfiban
Ammonia Processes:	Collin, Diamox, and Aqueous Ammonia
Carbonate Processes:	Seaboard, Vacuum Carbonate, Tripotassium Phosphate, Sodium Phenolate, Alkacid, Hot Carbonate, Catcarb, Benfield (Modified and Hipure), Giammarco-Vetrocoke

Typical processes which fall into the "B" category are:

Water Absorption
Fluor Solvent, Selexol, Rectisol, Purisol, Estrasolvan

A number of other physical solvents whose list and physical properties are readily available have been investigated by Woertz (15).

There are processes which can be considered as a hybrid between the categories "A" and "B". The most commonly known process which utilizes a solvent which is a combination of chemical reactant as well as a physical absorbent is sulfinol. Other processes, such as AMISOL can also be named.

There are processes such as: Burkheiser, Ferrox, Gluud, Manchester, Thylox, Fischer, Cataban, Konox, Perox, Lo-Cat, and others which are selective for removal of hydrogen sulfide only.

Processes which fall into the "C" category are:

Iron Oxide (sponge), Activated-Carbon, Cold Bed Adsorption (CBA), EFCO processes, and Appleby-Frodingham

In category "D" type processes, refrigeration at predetermined pressures will take place in order to separate the hydrocarbons from their impurities. Depending on the available upstream pressure, the use of turboexpansion machinery to provide the required refrigeration can be considered.

In general, the natural gases containing CO_2 can be categorized on the basis of their CO_2 content. The natural gas is divided into three categories: 1) Natural gases containing less than 5 mole percent CO_2 ; 2) Natural gases containing 5-30 mole percent CO_2 ; and 3) Natural gases containing greater than 30 mole percent CO_2 . In the section entitled "Analysis of Process Alternatives," the basis for the above categories is described in detail.

A comparison of the most common commercial gas purification processes, in as far as their limitations on the feed and product stream conditions are concerned, is given in Table 2 . Depending on specific feed stream conditions and the desired quality of the product stream, a process can be selected using Table 2, especially if the CO_2 concentration is between 5 and 30 mole percent.

II. APPROACH

The objective of this study was to explore alternative means of supplying CO₂ to enhanced oil recovery projects. The study emphasis was placed on:

- Analyzing processing alternatives for separating CO₂ from natural gas streams of varying compositions.
- Analyzing alternatives to pipeline only transport modes.
- Projecting the impact that any of the evolving concepts might have on the availability of low-cost CO₂ and hence the quantity of oil potentially recoverable by CO₂ EOR methods.

From the outset it appeared both possible and prudent to separate the "processing" and "transportation" portions of the study from each other and thus allow those areas to be worked in parallel. The transportation analysis address the transport of nearly pure (98% CO₂) carbon dioxide irrespective of its source.

Approach to Analysis of Processing Alternatives

In order to analyze processing alternatives for natural gases containing CO₂, the natural gases were divided into the following three general categories:

- 1 - Natural gas streams containing less than 5 mole % CO₂
- 2 - Natural gas streams containing greater than 5 but less than 30 mole % CO₂
- 3 - Natural gas streams containing greater than 30 mole % CO₂

The reasoning behind the above division is based on the heating value and the Wobbe Index.

The Wobbe Index is used as the most convenient index of commercial gas quality, combining measurements of calorific value and specific gravity, and indicating the range of gas qualities that can be satisfactorily handled by a predetermined orifice size.

$$\text{Wobbe Number} = \frac{\text{Calorific Value}}{\sqrt{\text{Specific Gravity}}}$$

Calorific Values, Specific Gravities and Wobbe Nos. for Typical Fuel Gases are (22):

<u>Type of Gas</u>	<u>Average Calorific Value Btu/Cuft</u>	<u>Specific Gravity</u>	<u>Wobbe Number</u>
1. Coal gas			
- horizontal retorts	560	0.472	878
- vertical retorts w/steaming	475	0.4744	690
- low temp. process	889	0.5351	1215
2. Coke oven gas	525	0.3922	838
3. Water gas			
- "blue"	295	0.5500	398
- "carburetted"	500	0.6457	622
4. Producer gas			
- from coke	132	0.8940	140
- from coal	163	0.8669	175
5. Blast furnace gas	92	1.0100	92
6. "Oil gas"			
I.C.I. "500" process	500	0.4714	728
C.R.G. Process	800 - 975	0.5614	1170
7. Natural gas			
Average "Commercial" gas	1,000 - 1,100	0.65	1302
Methane	1,100	0.5517	1481
Ethane	1,800	1.0345	1770
Propane	2,300	1.5172	1867
Butane	2,940	2.0000	2079

It is important to define Natural Gas on a commercial basis. Some twelve gas samples amongst 200 or more samples were chosen from different states and counties. The heating value for these ranged from 198 to 2323 Btu/scf. The methane, CO₂, heavier hydrocarbons, Nitrogen content of these gases are given in Table 3. The specific gravity and the Wobbe Number were calculated for these gases. For a typical unprocessed natural gas the Wobbe Number will range from 800 to 1500.

Table K shows how, on a commercial basis, natural gases are sometimes classified in the U.S. into "high inert," "high methane" and high "Btu" groups. The specific gravity, heating value, and the Wobbe for the commercial gases are given in this table.

TABLE K

Classification of Commercial Natural Gases

	<u>S. Gr.</u>	<u>Btu/Cuft</u>	<u>Wobbe No.</u>
1. High inert	.66 - .708	958 - 1051	1179 - 1249
2. High methane	.59 - .614	1008 - 1071	1312 - 1367
3. High Btu	.62 - .719	1071 - 1124	1360 - 1326

To relate the above discussions to the reasoning behind categorizing the natural gas into the a) CO₂ < 5 m%, b) 5 < CO₂ < 30 m%, and c) CO₂ > 30 m% , it is assumed that the natural gas is made up of CH₄ and CO₂. It is important to know the limits or the CO₂ content in order to classify the gas as follows:

- 1 - Natural gas containing CO₂
- 2 - Natural CO₂ source containing other impurities

Table K-1 gives Gross Heating Value, Specific Gravity, and Wobbe Number for methane-carbon dioxide mixture.

TABLE K-1

GHV, S.Gr. and Wobbe No. for CH₄ - CO₂ Mixtures

<u>CH₄ - CO₂ Mixutre</u>	<u>S. Gr.</u>	<u>GHV Btu/SCF</u>	<u>Wobbe No.</u>
95 - 5	0.6	997.5	1288
90 - 10	0.65	945.0	1174
80 - 20	0.74	840.0	973
70 - 30	0.84	735.0	801
60 - 40	0.94	650.0	651
50 - 50	1.03	525.0	516

From the results given in the above table it can be seen that the Wobbe Index reaches the minimum limit as shown in Table 3. As a result, the cut-off in the CO₂ content of a natural gas, as far as the processing is concerned, is set at 30 mole percent CO₂.

Another reason for the 30 mole % CO₂ content as a cut-off point in categorizing natural gas is that most physical absorption processes (as defined in Section I-C) are most effective economically up to about 30 mole % acid gas loading (CO₂ in this case).

Choice of solvent for gas treating is important because solvent circulation rate usually determines equipment sizes, the overall cost of the unit, and therefore the cost of the investment. Circulation rate of the solvent also determines the amounts of cooling or heating needed and sets the required horsepower for the moving machinery.

Solvent selection should be based on composition, temperature, and pressure of the feed gas plus the desired conditions of the product gas. These parameters usually determine whether a chemical or physical solvent is more economical. For a case where all parameters except the acid gas loading is the same, the circulation rate becomes directly proportional to the acid gas content of the feed. Chemical solvents can absorb acid gases without a high sensitivity to pressure, whereas the physical solvents can absorb acid gases in proportion to their partial pressures.

Chemical solvents usually consist of an aqueous solution of a reagent. Substantial amounts of energy (heat) are needed to regenerate the rich solution. Physical solvents have low heats of solution and in most cases do not require heat for regeneration.

As an example, with acid gas partial pressure at 75 psig (90 psia), capacity of physical solvents in commercial use far surpasses that of chemical solvents.

Figure 1 shows the range of treating for various processes. Assuming natural gas containing only CO₂ at an operating pressure of 1000 psig, the amine processes are economically preferred up to a partial pressure of 10 - 15 psi (1 to 1.5 CO₂ mole percent). Other chemical processes such as activated hot potassium carbonate can be considered up to a partial pressure of 50 - 75 psi (5 - 7.5 mole percent) while producing a comparable product purity. This can be considered as a main reason for drawing the line at 5 mole percent CO₂ in categorizing the natural gas containing carbon dioxide.

Natural gas streams containing greater than 30 mole % CO₂ must be purified in order to meet the Wobbe Index or other gas quality specifications, even if the product is not restricted to a certain CO₂ purity level. These types of streams require special processing. A typical schematic flow sheet is shown in Figure 2. Considering natural gases containing CO₂ as CO₂ resource for the Enhanced Oil Recovery processes, the natural gas containing more than 30% CO₂ can be considered more as a CO₂ resource where upon processing will yield a very valuable by-product namely, methane and other hydrocarbons.

Processing alternatives for natural gas can be summarized as follows:

- 1) Streams containing less than 5% mole CO₂ which can be treated using chemical solvents.
- 2) Streams containing greater than 5% mole and less than 30% CO₂ are excellent candidates for the physical solvents.
- 3) Streams containing greater than 30% CO₂ require special processing. A typical schematic flow sheet is shown in Figure 2.

Approach to Analysis of Transportation Alternatives

The transportation modes considered in previously conducted global CO₂ supply studies (References 4 and 6) were restricted to pipelines only. While it was obvious that pipelines were clearly the most efficient method of moving large quantities of CO₂ over a long period of time, it was not obvious that it was the best means of supplying small quantities for short periods of time. Specifically, the crossover point between truck, rail, and pipeline transport had never been reported in a generalized sense. (The study performed by Lawrence-Allison & Assoc. did address this issue for the Los Angeles Basin on a restrictive basis). (Reference 8) There remained the open issue; could a mixed mode transportation system consisting of a large diameter trunk line and combinations of lateral pipelines and trucks better serve the needs of more modest sized EOR projects.

This question arose from the fact that large pipelines favored long term (20 year +) investment recovery economics, and that most CO₂ injection projects required on the order of five years to complete.

The second area of concern with the results of past studies was the assumption made regarding the thermodynamic state of CO₂ for both transport and pressurization. The past studies concluded that the supercritical state was the thermodynamic state which provided the most efficient pipeline transport. The definition "super critical" merely means that the pressure of the fluid is higher than the critical pressure. At pressures over the critical pressure there can be no abrupt phase change thus ruling out the problems associated with two-phase flow. Although there are no distinct phase changes in this regime, there is a significant difference in the fluid's compressibility depending on the temperature assumed. The previous studies (Reference 6) recommended that the fluid be heated to 130° F prior to compression resulting in very high compressibility

and hence, high compressor work. The energy requirements and economics of pressurizing in the high compressibility regime were compared to those which assumed a lower temperature prior to compression.

The approach followed to assess the impact of novel methods of processing and supplying CO₂ on the amount of EOR oil was to determine the bounds of additional CO₂ that was available from natural sources (specifically natural gas CO₂ mixtures) if the novel methods were applied. Unfortunately, the assessment is tied to both the amount of new gas found and its CO₂ concentration.

The details of the methodology followed in the analysis are discussed in the next section of this report.

Approach to Analysis of Changes in CO₂ Reserves

The changes in CO₂ reserves will occur if the newly discovered natural gas sources containing 5 to 10% CO₂ are added to the CO₂ reserves. As discussed in the "Approach to Processing Alternatives," natural gases containing up to 10% CO₂ can be purified (taking the CO₂ out) using chemical or physical solvents. The economic incentives for taking small amounts of CO₂ out of the natural gas are discussed in the "Results" section. As long as the amount of CO₂ produced is in the range of 5-10 MMSCFD and the processing plant is within a 100 miles of the oil reservoir where the CO₂ can be utilized, separating the CO₂ from the hydrocarbon portion of the natural gas will have distinct economic incentives. The 5-10 MMSCFD CO₂ can be supplied from various combinations of amounts of natural gas and its CO₂ concentration. A 200 MMSCFD natural gas stream containing only 2.5 mole % CO₂ will be as attractive as a 100 MMSCFD natural gas with 5 mole % CO₂ in as far as total CO₂ content is concerned. When sufficient amounts of CO₂ can be produced, the natural gas containing only a few mole percent CO₂ can be considered a CO₂ reserve.

III. RESULTS

The results of the analysis are divided into the categories discussed in the previous section:

- Processing Alternatives
- Transportation Alternatives
- Changes in Availability of CO₂ from Natural Sources

The discussions of the results that appear in this section are divided into the same above mentioned categories.

Processing Alternatives

As described earlier, the processing of natural gas containing CO₂ was divided into three general categories.

- 1 - Natural gas streams containing less than 5 mole % CO₂
- 2 - Natural gas streams containing greater than 5 but less than 30 mole % CO₂
- 3 - Natural gas streams containing greater than 30 mole % CO₂

The results from each of the above will be discussed separately.

A. Processes for Natural Gas Streams Containing Less Than 5 Mole % CO₂

Economically, chemical type purification processes are best suited to natural gases containing less than 5 mole percent CO₂. Depending on feed composition, temperature and especially pressure, the chemical processes may have an economic edge over physical processes even up to about 10 mole percent CO₂ content in the feed.

In as far as the purpose of this study is concerned, the natural gases containing 5 - 10 mole percent CO₂ were examined as an additional CO₂ resource.

Cost of Processes Used for 5 to 10 Mole % CO₂

Natural gas streams containing 5 to 10 mole % CO₂ will need to be purified to meet pipeline quality standards. In the process of sweetening the natural gas, a high purity CO₂ mixed with a small amount of hydrocarbons will come out of the regenerator.

The pressure and temperature of the CO₂ coming out of a regenerator will depend on the type (chemical or physical) and the specific process used. Ranges of outlet pressure and temperature will be:

Outlet Pressure	Vacuum to 30 psig
Outlet Temperature	200 to 450°F

The processing cost of CO₂ will be mainly dependent upon the mode and condition of transportation. The CO₂ delivery conditions considered in this report were:

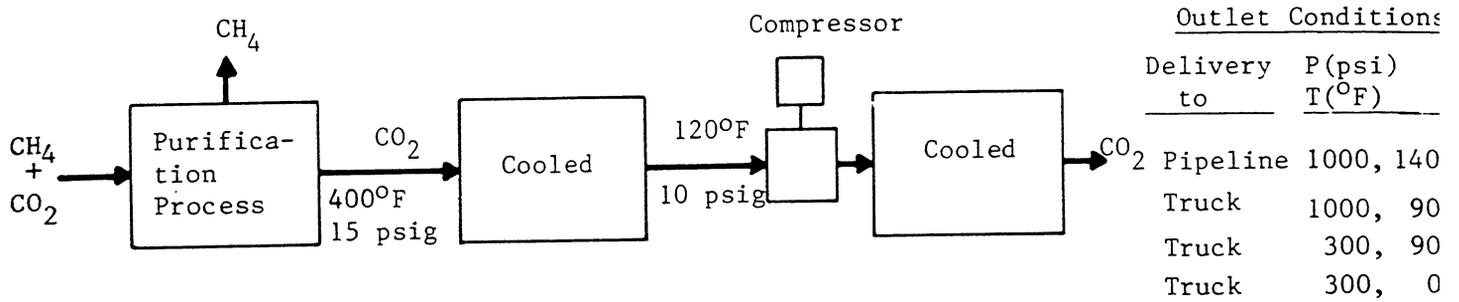
CO₂ Delivery Conditions

<u>Delivery To</u>	<u>Pressure (psi)</u>	<u>Temperature (°F)</u>
Pipeline	2000	140
Truck	1000	90
Truck	300	90
Truck	300	0

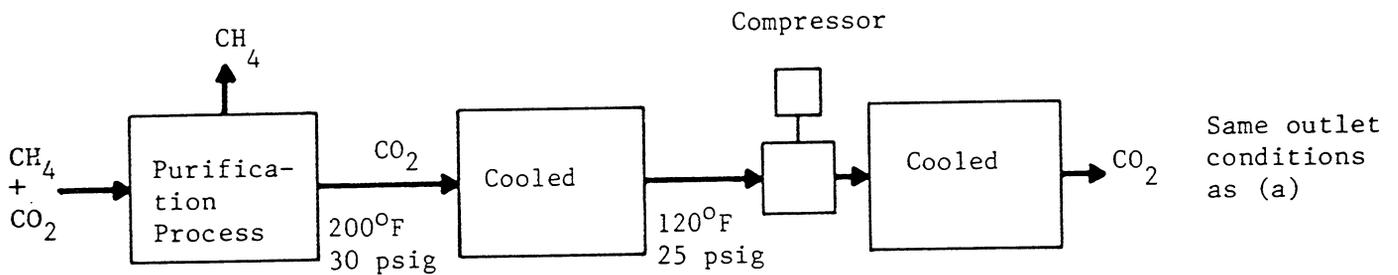
The general processing scheme required is given in Figure R-1.

Numerous costing alternatives can be considered. The following table describes the alternatives and the elements involved.

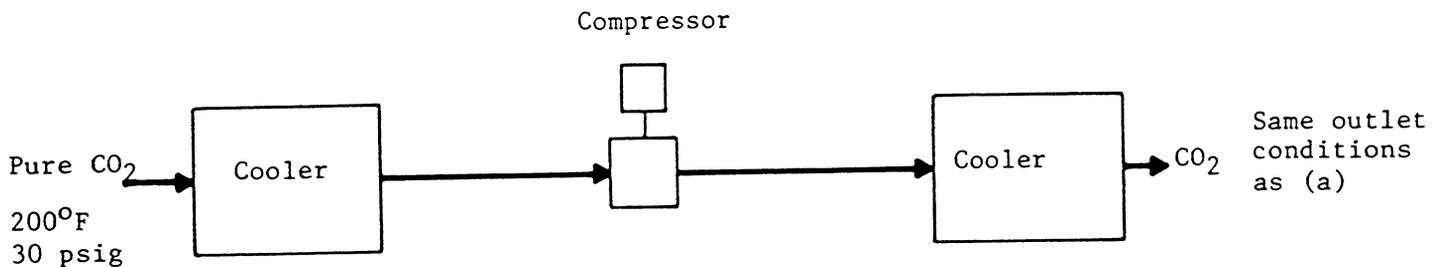
<u>Alternative</u>	Purification Cost			<u>Type of Process</u>	<u>Type of Delivery Pressure, Temp.</u>
	<u>Included</u>	<u>Adjusted</u>	<u>Ignored</u>		
A-1	yes	--	--	Chemical	Pipeline (2000, 140)
A-2	yes	--	--	Chemical	Truck (1000, 90)
A-3	yes	--	--	Chemical	Truck (300, 90)
A-4	yes	--	--	Chemical	Truck (300, 0)
B-1	yes	--	--	Physical	Pipeline (2000, 140)
B-2	yes	--	--	Physical	Truck (1000, 90)
B-3	yes	--	--	Physical	Truck (300, 90)
B-4	yes	--	--	Physical	Truck (300, 0)



a) Processing required using a chemical processing



b) Processing required using a physical processing



c) Processing required for pure CO₂ out of a sweetening plant

Figure R-1 General Processing Schemes for Delivering CO₂ to Various Transporting Conditions

As seen from the above table, there are eight alternatives when the total purification cost is included in the final cost of CO₂. Obviously this is unfair since the natural gas will be upgraded as far as its heating value is concerned. Assuming the price of natural gas depends on its Btu content, i.e., so many dollars per million Btu, then the purification cost can be adjusted such that the CO₂ product does not bear the entire purification cost. It is also possible to completely ignore the purification cost in costing the CO₂ product. This is assuming that the natural gas had to be purified for other reasons, and its CO₂ content was to be released into the atmosphere.

To determine the various cost elements the following example will serve as a basis:

<u>Natural Gas Stream</u>	<u>Conditions</u>
Amount	100 MMSCF/D
Methane	92.5 mole percent
Carbon Dioxide	7.5 mole percent
Pressure	1000 psig
Temperature	100°F
Other Impurities	Nil

The example is chosen such that either a chemical or physical solvent can be used.

To calculate the total cost of processing the above gas and delivering it in accordance to the transportation conditions required, the costing will be broken down as follows:

- Cost of Purification
- Cost of Additional Processing Required by the Transportation

Cost of Purification

For the given example, it was assumed that the natural gas product stream will be virtually free of CO₂ and the product stream will not contain any hydrocarbons. The conditions of the product streams will be:

Product Streams

	<u>Natural Gas</u>	<u>Carbon Dioxide</u>
Amount (MMSCF/D)	92.5	7.5
Pressure (psig)	985	vacuum - 30
Temperature (°F)	110	200 - 450
Impurities	Nil	Nil

As stated earlier, the natural gas in this example can be processed using either a chemical or a physical solvent. Knowing that the actual costs depend on the specific chemical or physical process, however, for simplicity, in this section the cost of a typical chemical process is compared against that of a typical physical process.

Purification Costs for Chemical and Physical Processes

	<u>Typical Chemical Process</u>	<u>Typical Physical Process</u>
Capital Investment \$/MSDCF Acid Gas	100	130
Total Operating Costs* \$/MSCFD Acid Gas	18.0	12.0

*Includes Operating Labor, Maintenance, Chemical Losses, Utilities, and Indirect Operating Costs

Cost of Additional Processing Required Before Transportation

This cost must be divided into the following:

- Cost of cooling before compression
- Cost of compression
- Cost of cooling after compression
- Cost of refrigeration when necessary

For comparison purposes, all cooling was assumed to be done by air except when cooling to lower than 140°F is required. The same cost of compression horsepower is used. The basis for cost calculations are:

- \$600/Installed Horsepower
- 9000 Btu/HP-HR
- 7.5 HP/ton of refrigeration
- 340 Days/Year
- 1979 dollars
- \$2.5/MMBtu

Basic engineering principles as outlined in references (9, 18) when used to determine heat exchanger area required, compression horsepower or refrigeration loads.

Cost of Cooling Before Compression

The cost of cooling before compression will depend on whether a chemical or physical process was used. Since the pure CO₂ gas coming out of a chemical process is at a higher temperature than that coming out of a physical process, the amount of heat to be removed will be higher for a chemical process than the physical one. The following table shows a comparison between cooling requirements and costs for a typical chemical against a typical physical process.

Comparison of Cooling Cost Before Compression

	<u>Typical Chemical Process</u>	<u>Typical Physical Process</u>
Amount CO ₂ (lbs/yr)	36,278	36,278
Temperature before cooling (°F)	450	200
Temperature after cooling (°F)	140	140
Heat Removed (MMBtu/hr)	4.53	0.848
HX surface area (sq.ft.)	1,140	488
Motors (Horsepower)	30	10
Capital Cost (1979 dollars)	25,800	11,800
Operating Cost (\$/year)	5,500	1,840

Cost of Compression, Intercooling, and After-cooling

Depending on the mode and specific conditions of transportation, the compression costs (capital and operating costs), and Intercooling and After-cooling costs (capital and operating costs) are given in Tables 4-7 for the following;

- Truck Transport at 300 psi and 90°F (Table 4)
- Truck Transport at 300 psi and 0°F (Table 5)
- Truck Transport at 1000 psi and 90°F (Table 6)
- Pipeline Transport at 2000 psi and 140°F (Table 7)

Total Capital and Operating Costs

The total capital and operating cost includes:

- Purification Cost (Chemical or Physical)
- Cooling Cost Before Compression (Chemical or Physical)
- Compression Cost, Intercooling, and After-cooling, with the following Transportation Options:
 - Truck 300 psi, 90°F
 - Truck 300 psi, 0°
 - Truck 1000 psi, 90°F
 - Pipeline 2000 Psi, 140°F

As stated earlier, other options where the purification costs are either ignored or adjusted should also be included.

Table 8 provides the total capital and operating cost matrix for the various options:

For the "Adjusted Option" the purification costs were adjusted on the following basis:

- 100 MMSCFD gas inlet to the sweetening plant
- 92.5 MMSCFD sweet gas out if processed
- Distance to be transported is 100 miles
- Btu value of the gas when it reaches the consumer will be the same whether 100 MMSCFD of sour gas or 92.5 MMSCFD of sweet gas is put through a trunkline
- Same trunkline and distance to market is assumed for the sweet and the sour gas

Trunkline transportation requirements are that the gas must be dried to less than 7 lbs/MMSCF. In both cases (sour or sweet gas) the gas must be dried to the above specification. Incentives for sweetening a natural gas containing CO₂ are:

- Less gas to be dried (92.5 vs. 100 MMSCFD) while delivering the same heating value to the consumer.
- Less gas to be transported through a trunk line

The following table shows the reductions in drying and transportation costs:

<u>Amount of Gas</u> MMSCF/D	<u>Nature</u> <u>of Gas</u>	<u>Dehydration Cost</u>		<u>Trunkline</u>
		Capital Cost (\$1000)	Operating Cost (\$1000/yr)	Transportation Cost (\$1000/yr)
925	sweet	475	157.3	3145
100	sour	<u>525</u>	<u>186.2</u>	<u>3400</u>
Savings		\$50	28.9	255

Therefore, for the "Adjusted Option" the total capital cost of the purification process (chemical or physical) will be reduced by \$50,000, while the annual operating cost is reduced by $255,000 + 28,900 = \$283,900$.

In the "Ignored Option" it is assumed that the CO₂ product will not be burdened with the purification costs. However, the capital and operating costs for the necessary pre-compression cooling will still be added.

Cost of CO₂ at the Delivery Point to the Consumer

The cost of CO₂ per MSCF at the delivery point to the consumer will include the following:

- Cost of purification
- Cost of pre-compression cooling
- Cost of compression including intercooling costs
- Cost of after-compression cooling or refrigeration (when necessary)
- Cost of transportation from the process plant to the point of consumption

The same economic criteria were used for determining the purification cost and other processing costs as those used for the transportation.

Table 9 shows the cost of CO₂ at the point of consumption for the various options given in Table 8 . The total cost was determined assuming the distance between the process plant and point of consumption to be 100 miles.

B. Process for Natural Gas Streams Containing 5-30 Mole% CO₂

For natural gas streams falling in this category the choice amongst processes will become narrower. Almost without question, physical processes will have a clear advantage. The only shortcoming some of the physical processes will have is the inability to meet strict purity requirements in the processed gas. In most cases like a "Hipure" Benfield Process, this problem is solved by adding a DEA unit to scrub the gas to the desired purity requirement.

Some of the processes which are suitable in the range of CO₂ concentration are:

- Benfield (Modified or Hipure, if necessary)
- Selexol
- SNPA-DEA
- Sulfinol

The final choice will depend on the economics and energy efficiency factor of the above processes.

Natural gases containing more than 10 mole % CO₂ are the best candidates for physical type sweetening processes. The economics compared to the economics of sweetening natural gas streams containing less than 10 mole % will be improved because both the capital and annual operating costs per MSCF of CO₂ processed will improve as the CO₂ concentration in the natural gas stream increases.

The increase in the CO₂ content of the feed gas will also improve the economics of the other processing required prior to transportation and the economics of transportation.

C. Processes for Natural Gas Streams Containing Greater Than 30 Mole % CO₂.

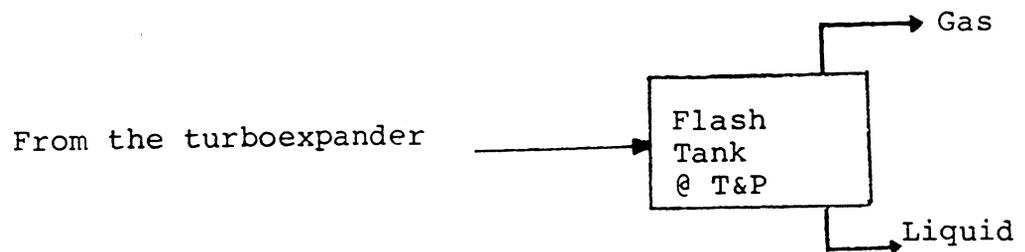
As described earlier, these types of natural gas streams should be categorized as natural CO₂ sources. The hydrocarbon content should be looked at as a welcomed by-product and a valuable one. The general processing scheme for these gases are given in Figure 2.

As shown in this figure, the gas stream must be cooled to a desired temperature in order to optimize liquid CO₂ recovery. The cooling process can be achieved a) by expanding the gas (if it is available at high enough pressure), and b) by using outside refrigeration. Economizers will be used to exchange heat between the hot streams (such as the feed) to be cooled and the cold streams which need to be heated.

It was mentioned in Part B (above) that physical sweetening processes show an economic advantage as the CO₂ concentration increases. However, at some point, like 30 mole %, the capital and operating cost of the sweetening process will be proportional to the circulation rate of the physical solvent and the circulation rate will be directly proportional to the CO₂ concentration in the sour fuel gas.

As shown in the schematic flow sheet, use of turboexpanders (when a high wellhead pressure is available) or other refrigeration means (when the wellhead pressure decreases) will produce a liquid stream with a high CO₂ content and a gas stream with a CO₂ content in an optimized range for a physical solvent sweetening process.

Assuming the natural gas to be a binary mixture of CO₂-CH₄, flash calculations can be run at different pressure and temperatures for different mixture compositions.



Moles of CO₂ in the Liquid Phase = $X_{CO_2} L$

$$X_{CO_2} L = \frac{Z_{CO_2} K_{CH_4} + K_{CO_2} (1 - K_{CH_4} - Z_{CO_2})}{(K_{CH_4} - K_{CO_2}) (1 - K_{CO_2})} \quad (1)$$

Where: X_{CO_2} = Mole fraction of CO_2 in the liquid

L = Moles of liquid

Z_{CO_2} = Mole fraction of CO_2 in the feed

K_{CH_4} = Vapor-liquid equilibrium ratio for methane

K_{CO_2} = Vapor-liquid equilibrium ratio for carbon dioxide

As seen in Equation (1) which is a result of material balance and vapor liquid equilibrium, the moles of CO_2 in the liquid phase are related to the moles of CO_2 in the feed, the equilibrium constants for CH_4 and CO_2 .

The temperature and the pressure in the flash tank following the turboexpander (or other means of refrigeration) will determine:

- Amount of liquid stream containing higher CO_2 concentration
- Amount of gas stream containing lower CO_2 concentration

As mentioned earlier, a simple computer program was written to determine the temperature and pressure (for specific feed composition) at which the amount of CO_2 in the liquid phase will be a maximum. Tables 10 to 14 show the temperature and pressure results for feed composition of 30 to 50 % CO_2 . The above tables were prepared only for temperature, pressure, and composition for which some experimental data were available.

The proper choice of temperature and pressure will depend on the well head pressure and the percent CO_2 in the fluid. The high pressure streams can be expanded through a turboexpander to drop the temperature of the fluid.

The cost of the gas processing plant will depend on the flow rate and the pressure of the feed gas. Most sweetening processes are temperature sensitive. The MEA process for example will not

be effective at all at inlet gas temperatures above 120°F. Regardless of the type of sweetening process (chemical, physical, or solid bed) the cooler the inlet gas is (within limits, i.e., prevent gas hydrates, etc.), the more efficient the separation process will be. The inlet gas pressure has the opposite effect as the inlet gas temperature. The higher the pressure (within practical limitation), the more effective the separation will be. The inlet gas pressure has a twofold effect: 1) more efficient separation process, especially physical and solid bed type processes, will be at higher pressures, 2) smaller processing equipment required (such as contact towers) to process the same amount of gas at low pressures. However, higher pressure means thicker vessels, more steel and greater cost. An optimum contact tower pressure can be determined based on vessel size, wall thickness and weight, pump horsepower for returning the solvent into the tower and other sundry equipment, such as heat exchanges, etc. The contact tower pressure will also be determined by inlet pressure required for the pipeline transporting the processed gas. Therefore, it is definitely uneconomical to process the gas at much higher pressure than required as the inlet pressure to the gas pipeline plus 10-20 psi for pressure drop in contact tower.

Natural gas reservoirs containing sufficient CO₂ will be more interesting if the wellhead pressure is higher than the pressure required for the contact tower. The higher wellhead pressure can be utilized to cool the gas using turboexpansion.

Use of Turboexpansion in Recovering CO₂ from Natural Gas

In order to carry out any sort of process calculations on a mixture of CH₄-CO₂, a reliable range of vapor-liquid equilibria (K) are required. The K data for the mixture of CH₄-CO₂ are especially critical in the design of a Turboexpander and the following separation equipment. As will be explained in the ensuing discussion, it is very difficult to find any consistent data.

Inconsistency in the Vapor-Liquid Equilibrium (K) Data for CH₄-CO₂ Mixtures

The K values for any component in a mixture are a function of temperature, pressure, and composition.

$$K = K(T, P, X_i) \quad (2)$$

considering the phase rule Equation:

$$P + F = C + 2 \quad (3)$$

for a two component mixture; i.e., CH₄-CO₂, C = 2. Hence, in a two phase region (vapor-liquid equilibrium), p = 2. Substituting the values for C and P, the degree of freedom (F) from Equation (3) is calculated to be 2. Therefore, if temperature and pressure are specific (two degrees of freedom), the K values for the CH₄-CO₂ mixture must be the same for all various compositions.

The K values for CH₄ in a 50-50 mixture of CH₄-CO₂ from Reference 19 are shown in Figures 3-5. The K values computed using the NGSPA program for a 40-60 (CH₄-CO₂) mixture are shown in Figure 6. As can be seen from these figures, the K data for CH₄ at the same T and P are not the same for the (50-50) and the (40-60) mixture.

The K data for methane at different convergence pressures of 800, 1000, 1500, 2000, 3000, and 5000 psia were obtained at temperatures of -50, 0, and 50°F and pressures of 100, 250, 500, 750, 1000, and 1500 psia. A plot of K_{CH_4} against convergence pressure is given for the applicable pressure range for each of the above temperatures (-50, 0, 50°F). These plots are shown in Figures 7, 8, and 9. As can be seen from these plots, the K_{CH_4} values are very much a function of the convergence pressure when C.P. is below 2000 psia. The convergence pressure is a way to determine the effect of composition on the K value of a component in a mixture.

An attempt was made to determine whether there was any consistency between the K_{CH_4} data in a 50-50 CH₄-CO₂ mixture experimental data, 40-60 mixture calculated values, and values obtained from the K_{CH_4} vs. convergence pressure plots.

Unfortunately, no consistency or meaningful correlation can be found. The K_{CH_4} was also calculated using a correlation given

by Canfield (23). In this reference K_{CH_4} is given as a function of T, P, and convergence pressure. The result of K_{CH_4} determined from various sources at a given T, and P are compared in Table 15. As it can be seen from the results given in Table 15, the K_{CH_4} values differ by a wide enough margin that none can be trusted.

This leads one to think that there is a very dire need for reliable and consistent vapor-liquid equilibrium data for CH_4 - CO_2 mixtures. Such data should be determined experimentally at wide composition, temperature, and pressure ranges such that the two phase (vapor-liquid) envelope can be plotted for each mixture composition. A plot of critical point locus as a function of composition for the CH_4 - CO_2 mixture can also be prepared.

Most CH_4 - CO_2 mixtures transported by pipeline or other means will have a small amount of other impurities, such as higher Hydrocarbons (ethane, propane), Nitrogen, and other gaseous elements. It will not be possible to determine all necessary K values experimentally due to infinite numbers of combinations that can exist. However, an attempt can be made to determine the effect of presences of a third element (which can be a single component or combination of two or more components) in the CH_4 - CO_2 mixture. The presence of a third element will provide extra degrees of freedom in accordance to Equation 2 and hence, make the K_{CH_4} , and K_{CO_2} a function of composition as well as temperature and pressure.

Lack of Other Physical Property Data for CO_2 - CO_4 Mixtures

- Critical Temperature and Pressure Data

A plot of critical pressure against critical temperature for different CH_4 - CO_2 mixture compositions was prepared using available experimental data references (19, 20, and 21). This plot is shown in Figure 10. As it can be seen from this figure, CH_4 - CO_2 mixtures behave very nonideally. The deviation of the critical pressure of the mixture at some points is as much as 100 percent from those which can be predicted using various mixing rules.

The critical property of the mixture is very important in determining phase behavior.

- Heat Capacity Data

No reliable data for heat capacities of the CH₄-CO₂ mixtures could be found. The heat capacities are needed in performing heat transfer calculations. The ratio of C_p/C_v is very important in calculating the outlet temperature out of the turbo-expander.

The heat capacities for the mixture at a wide range of temperatures were calculated using reduced temperature (T_r) and reduced pressure (P_r) and generalized enthalpy departure charts $(H-H^*)/T_c$. Since T_c , and P_c data for a few mixtures were available hence, Figure 10 was prepared and then used to determine T_c and P_c for other mixtures.

The values for $1/m = k/(k-1)$ were also calculated using C_p values as described above. The values for various CH₄-CO₂ mixtures at different pressures and temperatures are given in Tables 16 through 21. The $1/m$ values are extremely critical in calculating outlet temperatures out of an expander of a compressor. The $1/m$ values calculated using experimental $T_c + P_c$ values as explained varied as much as 200-300 percent from those obtained from a molecular weight correlation given in Reference (9).

- Compressibility Factor, Viscosity, and Thermal Conductivity Data

Compressibility, viscosity, and thermal conductivity data for CO₂ containing small amounts of methane (5-8%) play an important role in performing pipeline, compression, and other transportation costs.

In order to show that the type of processing shown in Figure 2 is economically more cost effective than just a physical solvent type purification process, it is important to use reliable physical properties for CH₄-CO₂ mixtures. In preparing the table given in Figure 2 it was realized that variation in equilibrium vapor-liquid data or other physical property data causes a great variation in the amount of liquid or vapor produced and their concentrations. The amounts and compositions of liquid or vapors produced are the main determining factors in the size, costs, and as a result, the optimized conditions of the process.

- Calculation of Upstream Pressure and Work Out of the Turboexpander

The procedure for calculating the upstream pressure necessary to provide the proper outlet temperature when expanded is given in Appendix B.

The amount of work which can be expected out of the turboexpander is also given in Appendix B.

TRANSPORTATION ALTERNATIVES

In this section we evaluate alternative means for transporting carbon dioxide from a source to an EOR project. Large, natural CO₂ resources appear to be inadequate to supply projected U. S. EOR demand for carbon dioxide (reference 1), but published studies of CO₂ transportation seem directed principally to high-rate supply from large, naturally-occurring resources. Furthermore, though rapid gains are being made, the U. S. petroleum industry has had very little experience building and operating carbon dioxide pipelines, and our brief review (reference 10) indicated that serious uncertainties surround the thermodynamic properties of both pure CO₂ and also CO₂-hydrocarbon mixtures. Consequently, this section examines the following basic questions:

Are pipelines being designed for the transmission of CO₂ properly accommodating the unique thermodynamic properties of CO₂?

Is the exclusive use of pipelines the optimum transport mode for all CO₂ EOR projects, regardless of injection rates and project durations?

Cost-saving innovations in CO₂ transport may permit economical utilization of smaller amounts of CO₂ than are currently judged practical, so that an expanded number of the lower-purity, natural CO₂ reservoirs may be economically extracted, and thus contribute to increased U. S. CO₂ reserves.

Thermodynamic Properties Affecting Pipeline Design

The critical design parameter for economical CO₂ pipeline design is pressure drop, which is largely dependent upon the thermodynamic properties of the transported gas. Existing CO₂ pipelines and previous studies of CO₂ transport have placed emphasis upon operation in the supercritical gas phase (References 4, 6, 8). A pressure-enthalpy (Mollier) diagram for CO₂ is shown in Figure 11. The four regions identified in the figure are the liquid two-phase, gas, and the "dense-phase-fluid," which is the description of the condition of carbon dioxide above the critical pressure of 1,071 psia. The supercritical gas condition refers to the part of the dense-phase-fluid (DPF) region in Figure 11 labeled "high compressibility." Operation in the supercritical gas phase is economically attractive because higher densities can be achieved than those possible below the critical pressure (the top of the "dome" in Figure 11), and because abrupt pressure and temperature drops will not necessarily induce two-phase flow, which is damaging to pressurization equipment (Reference 7).

Within the DPF region of CO₂ there are operating conditions of pressure and temperature that are economically more attractive than others for CO₂ pipeline design. In such a region, which is identified as "low compressibility" in Figure 11, the lines of constant-entropy ($S = \text{constant}$) become more vertical as one progresses from the higher temperature to the lower temperature of the DPF region. In intuitive terms, CO₂ in the DPF behaves more as a "liquid" than as a "gas" as the temperature is decreased at constant pressure. This characteristic is normally expressed in the equation of state by a reduction in the compressibility factor (hence, the labeling of Figure 11). The work required to compress

the DPF in the low-compressibility region is significantly less than that required in the high-compressibility region, so that one would expect the energy (and thus the cost) required to transport CO₂ as a low-compressibility DPF would be less than as a high-compressibility DPF.

The economic incentive for CO₂ pipeline operation in the low-compressibility-DPF region is a function of the following comparative design requirements for low-compressibility and high-compressibility schemes to transport a given quantity of gas:

- Costs of pressurization at compressor stations
- Costs of compressor repressurization stations
- Separation between compressor stations
- Diameter of pipeline

The optimal selection of these pipeline system design parameters depends on the quantity of CO₂ to be transported, the cost of pipe as a function of diameter, the cost of compressors as a function of horsepower (and, precisely but to a limited extent, flow rate), the cost of energy to pressurize the CO₂, and the physical properties of carbon dioxide at the various pressures and temperatures of the pipeline (e. g., at compressor inlets and outlets). In specific applications, terrain and routing also are key factors in pipeline design; for example, at the base of a steep mountain ridge, a depressurization station may be necessary to assure that static head pressure in the pipeline does not exceed the allowable stress for the pipe. While a specific design must consider such real-world constraints as terrain, to characterize the potential economic advantages of low-compressibility operation, we assumed a "flat," unconstrained pipeline route, and evaluated alternative low- and high-compressibility paths for CO₂ pipeline operation at various delivery rates.

The candidate paths for the initial pressurization of the carbon dioxide are illustrated in Figure 12 . The following assumptions are common to both cases:

200 psia and 140°F wellhead conditions
First-stage compression from wellhead conditions to 1000 psia and 80°F, preceded by cooling
Single-stage pressure ratio limited to 3:1 for high isentropic efficiency

In order to access the low-compressibility region without the aid of refrigeration, it is necessary to compress CO₂ in the gas phase to a pressure which has a reasonably high saturation temperature; this first compression stage is common for both low- and high-compressibility paths. The gas was initially cooled to reduce the compressor work (1 to 2), and compressed in the first stage to 1000 psia and 80°F (2 to 3). The gas was then intercooled (3 to 4), and compressed in the second compressor stage. In the second stage, the two paths depart. The high-compressibility path continues in the near-isentropic manner to point 5. The low-compressibility path from the departure point 4' to point 6 consists of heat transfer and condensation of the carbon dioxide. The remainder of the pressurization for this path is performed with CO₂ in the liquid and low-compressibility DPF states. The departure reduces the pressurization work for the low-compressibility path; the advantage appears in Figure 12 as the difference between $\Delta h_2'$ and that portion of Δh_2 above the point of path departure (4' to 5). As previously stated, the reason for the departure at 1000 psia is that this pressure represents the lowest at which condensation can be reasonably achieved without the aid of refrigeration.

The compressor horsepower requirements associated with

the two thermodynamic paths are shown in Figure 13 , which is a plot of compressor horsepower per MSCFD vs CO₂ wellhead pressure. Of course, in an actual application CO₂ reservoir pressure should be utilized whenever possible to reduce compressor horsepower requirements. For the range of hypothetical pressures of 400 to 1200 psia in the example, however, the Figure indicates that substantial reductions in compressor horsepower requirements can be achieved by utilizing the low-compressibility path. At low wellhead pressures, these differences become smaller, however, for the relatively longer length of the commonly shared legs decreases the difference in total work between the two paths. Nevertheless, even though the percentage energy savings for the low-compressibility path are less at lower wellhead pressure, the difference over the range of 400 to 800 psia is still 20 to 50 hp per MSCFD.

Repressurization is required for long pipelines to overcome the frictional losses encountered by the gas. The total energy requirement of repressurization is the sum of the energy required by the compression, plus any energy required for special thermal conditioning. Assuming that the pipeline can be buried sufficiently deep to eliminate temperature changes due to variations in ambient air (which, for normal soils, requires about 4-ft burial), we assumed that the ground temperature never exceeds 70°F. (Most of the Midcontinent region has temperatures significantly lower than 70°F.) As cited in reference 6 , we also assume that CO₂ must be heated to 130°F (at 1400 psia pipeline pressure) to provide a sufficiently high compressibility factor to meet gas compressor manufacturer's warranty conditions. Thus, compliance with manufacturer's warranty requirements on the compressors forces additional energy to be added to the carbon dioxide for thermal conditioning prior to repressurization. However, this thermal conditioning is only needed in the high-compressibility path.

Figure 14 shows the relative temperature changes experienced by low- and high- compressibility pipelines for 70°F ground temperature and minimum pressure of 1400 psia. The temperature of the CO₂ must be reheated to meet the compressibility requirements prior to entry into the compressor.

If pressurization is accomplished in the low-compressibility area, then the temperature of the CO₂ never departs significantly from ground temperature. For pipelines of equal diameter and capacity, the pressure loss per unit length will be lower for the lower temperature, more dense fluid.

These two approaches to repressurization can be analyzed by plotting the paths on a Mollier diagram as shown in Figure 15. The pump work required in the low-compressibility path is indicated by the Δh_p . The energy required to condition the CO₂ (i.e., heat it to 130°F) will depend upon the temperature the CO₂ had reached just prior to the compressor station. One can see that in cases where the CO₂ temperature approaches the ground temperature, substantial amounts of energy are required to reheat the fluid to the required compressor inlet conditions.

To determine the impact of the above mentioned phenomena, the operating characteristics were calculated for CO₂ pipelines of four different capacities. In all cases a maximum operating pressure of 2000 psia and a minimum operating pressure of 1400 psia were assumed. In one set of calculations, the initial conditions of 2000 psia and 80°F were assumed (low-compressibility path). In the second set of calculations, the initial conditions were assumed to be 2000 psia and 185°F (high-compressibility path). Modified Panhandle (Panhandle B) equations (REF. 9) were used to compute the pressure drop of the high-compressibility path. Both the Modified Panhandle equations and methods more suitable for product pipelines were used for the low compressibility path. The heat transfer from the CO₂ was limited by the heat transfer coefficient at the outer wall of the pipe, which was assumed to be 0.6 BTU/ft-°F-hr, a

value calculated for average soil conductivities. The pipeline diameter was held fixed for each line capacity.

The results of the pressure loss and heat transfer calculations are shown in Table 22. The pressure loss is represented by the distance between compressor stations (i.e., the line length at which the low-pressure limit had been reached - $\Delta p=600$ psi). As a result of the higher density, the low-compressibility path always resulted in longer distances between compressor stations (i.e., lower unit pressure losses) than the high-compressibility path. The differences in distances ranged from 17% for the 8-in diameter pipeline to 150% for the 20-in line. This variation in differences is caused by the fact that the smaller-diameter pipeline transfers heat more rapidly than the large-diameter pipe because of the higher surface-to-volume ratio of the smaller pipe. Additional evidence of the decreasing unit heat transfer rate with increasing pipe diameter is the higher final fluid temperatures of the larger-diameter pipes.

The compressor power and thermal conditioning requirements for the four pipelines are listed in Table 23. The significant differences in compressor power (hp/mi) results from both the closer compressor spacing and the higher power required to compress the higher temperature fluid for the high-compressibility path. The differences in compressor requirements for the two paths are substantial, ranging from a factor of 2 for the 8-in line to almost a factor of 8 for the 20-in line. The thermal conditioning requirements for the smaller pipelines are significant and tend to negate the beneficial effects that transferring heat to the ground had in reducing the pressure loss. Though thermal conditioning and compression requirements are superior for the low-compressibility designs, these differences are not representative of pipelines whose designs would be optimized for the respective fluid conditions. The higher capacity lines designed for the high-compressibility conditions would definitely require larger diameters to

arrive at more optimum operating conditions.

The differences in heat and power requirements for alternative optimized designs may be more or less than the examples in Table 23, but the differences reported there are illustrative of the improvements possible with a given line diameter operating in a low-compressibility path.

As one would expect from the decreased energy requirements of the low-compressibility path, the estimated CO₂ transportation costs are lower for the low-compressibility pipeline than for the high-compressibility case. Table 24 summarizes the investment costs for two cases. The comparison is based on the standard pipeline diameter most appropriate to low-compressibility fluid transport at each of the four line capacities of 50, /125, /250 and 500 MMSCFD. Although actual high-compressibility fluid pipelines would probably be different from the diameters chosen, the analysis is illustrative of the relative economic efficiency of the two transport alternatives.

The investment costs include pipeline and repressurization facilities. Pipeline investment was assumed to be \$20,000/in mile for both low- and high- compressibility cases. Repressurization costs were assumed to be \$540/hp for both cases. As seen in Table 24, at the lower two capacities the investment costs per mile are roughly the same. The investment for the high-compressibility transport mode increases more rapidly with capacity than does the low-compressibility investment. At 500 MMSCFD, the investment for the high-compressibility system is 27% greater than the investment for the low-compressibility system with a 20-in pipeline diameter. The difference in investments is due principally to the closer spacing of compressor stations for the high-compressibility pipeline than that required for the low-compressibility pipeline.

The difference in operating costs between the two alternatives is much more pronounced than the difference in investment. Operating costs include repressurization;

station labor and supervision; and fuel for both repressurization and revaporization (needed only for the high-compressibility transport of 50 to 125 MMSCFD). Table 24 shows that the high-compressibility system has significantly higher operating costs than the low-compressibility system. At 50 MMSCFD, the high-compressibility operating costs are 52% greater than the low-compressibility costs; and at 500 MMSCFD, the high-compressibility operating costs are 147% greater than the low-compressibility costs.

The cost analysis suggests that both investment and operating costs are lower for a dense-phase-fluid CO₂ pipeline than for a pipeline designed for a high-compressibility path. These results are only qualitative, however, since the same diameter pipe was assumed for both low- and high-compressibility cases. Though beyond the scope of this study, a more accurate economic comparison would involve selection of the cost-minimizing combination of pipe diameter, pressures, temperatures and pump stations. The analysis summarized in Table 24 is nonetheless useful as an indication of the potential economic benefits to be gained from operation of a CO₂ pipeline in an energy-conserving, low-compressibility mode. Of additional interest in this respect are the fuel costs of the two cases. Fuel cost accounts for from 29% to 43% of the operating cost for the high-compressibility pipeline, but fuel cost is only 3% to 16% of the operating cost for the dense-phase-fluid systems. Hence, the cost of the high-compressibility CO₂ transport system is more sensitive to rising energy cost than is the low-compressibility transport cost.

In conclusion, substantial reductions in compressor work and thermal conditioning can be achieved by maintaining CO₂ in the lower-temperature, dense-phase-fluid region (above the critical pressure) in pipelines.

The reductions in compressor work and thermal conditioning can translate into substantial reductions in investment and/or operating costs. These results are strongly dependent on assumptions regarding compressor operating characteristics. Specifically, the stated need to provide the compressor a fluid having a compressibility factor of 0.5 ($T = 130^{\circ}\text{F}$ at 1400 psia) or higher, places a severe thermodynamic penalty on CO_2 pressurization schemes and pipelines. The advantages of dense-phase-fluid pressurization and transport should provide the motivation to explore means of pressurizing CO_2 in the low-compressibility, more liquid-like area of the dense-phase-fluid thermodynamic region.

Two large diameter CO_2 pipeline projects have been announced since this phase of the study was conducted and reported (REF.10). The first by Shell is a pipeline that transports CO_2 from Montezuma County, New Mexico to the Denver Unit of the Wasson Field in Yoakum County, Texas (REF. 11). The line was sized at 36 in nominal diameter to transport 400 MMSCFD. The net elevation change between the origin and the terminus is approximately 3600 feet or a positive head equivalent to 1250 psia. Shell is providing for only one recompression station (at the 150-mile point) for the entire 478 mile length of the pipeline.

The second pipeline announced by ARCO transports CO_2 from Huerfano County, Colorado to the Willard Unit of the Wasson Field and the Seminole Field in Yoakum County and Gaines County, Texas, respectively (REF.12). This line will have a nominal diameter of 20 inches and will transport 300 MMSCFD. The net elevation change from the highest elevation of the line to the terminus is 4800 feet, with a positive head equivalent to 1680 psi. There will be no repressurization stations along the 420 mile-length of the pipeline.

Both of the above mentioned lines are conservatively

sized (i.e., low average velocity). The flow rate per unit flow area is 23% (ARCO) and 67% (Shell) lower than the low-compressibility path cases previously discussed (which is representative of the SACROC pipeline design). Apparently, constantly rising energy costs and the difficulty of providing energy to remote compressor locations is causing operators to adopt a more conservative (i.e., more capital-intensive) design philosophy. This observation will be discussed in greater detail in the next section.

The approach of making use of the low-compressibility thermodynamic regions of CO₂ was followed by ARCO in the design of its pipeline. ARCO's thermodynamic design assumptions are discussed in detail in REF.12 . There were insufficient details presented in REF.12 to determine the thermodynamic design assumptions made by Shell in the design of the McElmo Dome pipeline.

Transportation Mode Optimization

The relatively higher unit investment for small-capacity pipelines suggests that exclusive use of pipeline might not be optimum for the transport of CO₂ to smaller EOR reservoirs. Studies have assumed a twenty-five-year economic life for a pipeline; also, flow rates less than 25 MMSCFD have not been considered. Reference 1 suggests that CO₂ project lives of 5 to 7 years should provide a planning basis for CO₂ transport systems.

The assumption of a twenty-five year pipeline life is reasonable for long-distance, high-capacity trunk lines, such as those being planned by ARCO, Shell, and Amoco. In these cases the pipeline life coincides with expected productive life of the CO₂ natural reservoirs. It is unrealistic, however, to assume that pipelines that distribute CO₂ from the trunkline to a given field would be in use for twenty-five years. The average injection period reported (1)

indicates that five years is a more reasonable distribution system life. The injection system was projected to operate at full capacity for only two years due to the increased use of reinjected CO₂ in the latter phases of the project. A lateral pipeline or alternative transportation system that would be dedicated to a particular field would have to be selected on the basis of a five-year, rather than a twenty-year useful life.

Pipeline Transport - Trunkline

Two sets of pipelines were designed to allow the comparison of alternative transport modes. The first set was felt to be representative of trunk lines, the second set represented lateral pipelines (i.e., lines whose sole function was to supply a given project). The lines were sized to provide very low average velocities with the attendant low pressure losses. This design approach is consistent with the pipeline projects previously discussed and is founded on the following:

- The rapid rate of increase of energy costs and the large uncertainty that this causes the economics of long-term projects.
- Pipeline operation and maintenance costs are almost exclusively a function of the number of compression stations along the line.
- Since the CO₂ being transported cannot be used as a fuel for the compressor's prime mover (as in the case of a natural gas pipeline), additional compressor stations mean long distance electrical transmission lines are required.

The results of the pipeline-sizing calculations are shown in Table 25.

The cost of transporting CO₂ through the trunk lines sized as those in Table 25 were almost exclusively due to the fixed cost associated with the capital required for the construction of the pipelines. The cash flow assumptions used in computing the cost for the trunk line delivered CO₂ were as shown below:

- The line would be constructed in two years and the investments would be made at the ends of years -1 and 0 (where time "0" designates the start of the productive life of the project).
- The line would operate the first year at an average of 80% of design capacity, the second year at 93% of design capacity, and at full capacity for the remainder of its twenty-year life. This start-up profile is consistent with that planned for the Shell McElmo line. The less-than-capacity start-up is due to the time required to complete the development of the CO₂ fields.
- The line would have a zero salvage value at the end of its twenty-year productive life.

Other assumptions made in the calculation of the cost of transporting CO₂ through the trunk line were:

- Investment tax credit of 10% of the total investment.
- Investment costs of \$15,000/inch·mile and \$20,000/inch·mile for the entire pipeline system.
- Discounted cash flow rate of return (after taxes) of 15%.
- Total (state and federal) income tax rate of 50%.
- Straight-line depreciation over 20 years.
- Yearly operating cost equal to 1% of the total investment cost.

The operating cost assumption was checked for several of the lines and found to be a conservative (i.e., generous allowance) estimate of operating cost. The resulting cost of transporting CO₂ is shown in Table 26. The delivery costs for the two different pipeline construction cost assumptions bound the pipeline cost reported in Reference 6. The costs are totally dependent on the pipeline construction cost assumptions. The assumptions used represented the range of average cost of pipelines that were reported in the Oil and Gas Journal's Pipeline Cost Index adjusted by the second quarter 1979 composite cost index (Reference 13).

The long distance, high-capacity pipeline is by far the most efficient means of transporting carbon dioxide. The only issue is to provide sufficient time to recover the initial investment. For the twenty-year project life that was assumed for the trunk line application, the pipeline has no competition.

Pipeline Transport - Lateral

The principal difference in the analysis of the lateral and trunk pipeline cases is the assumed length of the project. In the OTA Study (Reference 1) and in the earlier Lewin Study (Reference 2), a universal CO₂ injection profile was developed that was felt to be representative of most candidate EOR reservoirs. The profile accounted for the reinjection of CO₂ that would be produced with the recovered oil in the latter phase of the projects. The injection profile that was developed in the two referenced studies is shown in the following table:

<u>Year</u>	<u>Purchased CO₂ % of total</u>	<u>Recycled CO₂ % of total</u>
1	20	0
2	20	0
3	16	4
4	13	7
5	6	14

The relatively short project life and the sub-capacity operation for three of the five years motivated comparison of truck transport cost and lateral pipeline cost.

The cash flow assumptions used in computing the cost of the CO₂ delivered via lateral pipelines were as follows:

- The line would be constructed in one year and the investment would be made at the end of year -1 or the start of year 0 (where time "0" designates the start of the productive life of the project).
- The line would operate the first and second year at design capacity and at 84%, 65%, and 30% of design capacity for the third, fourth and fifth years respectively.
- The line would have a zero salvage value at the end of its five-year life.

Other assumptions made in the calculation of cost of transporting CO₂ through the lateral pipeline were

- Investment tax credit of 6.66% of the total investment (owing to the 5-year production life).
- Investment costs of \$10,000/inch·mile, \$15,000/inch·mile, and \$20,000/inch·mile (depending on size) for the entire system.

- Discounted cash flow rate of return (after taxes) of 15%
- Total (State and Federal) income tax rate of 50%
- Straight-line depreciation over 5 years.
- Yearly operating cost equal to 1% of the total investment cost.

The resulting cost of delivering CO₂ is shown in Table 27. For the cases where the lateral pipeline capacity overlapped the trunk line, the cost of the lateral delivered CO₂ was 48% higher than that of the trunk line. The difference in the costs is due to both the shorter assumed project life and the fact that the lateral does not flow at design capacity for three of the five years. To separate these two effects, the cost calculations were performed assuming that the the line flowed at full capacity for all of the five years. The change in flow assumption reduced the cost of the delivered CO₂ by 18%. The difference between the trunk line delivered and a full capacity lateral-line cost was reduced to a 19% difference. This discrepancy is totally attributable to the difference between the project life assumptions of 20 years and 5 years. It should be recalled that in time-value-of-money problems the cash flows in the earlier years have a much larger impact on cost than those in later years (see Reference 14).

Truck Transport

Truck-trailer distribution of CO₂ was investigated as an alternative to pipeline laterals. Lubbock Manufacturing Company of Lubbock, TX, provided specifications and performance data on its line of insulated steel CO₂ trailers. The largest Lubbock trailer, having 5,9000-gallon capacity, was

selected for the subsequent cost analysis. Transport of pure CO₂ at 300 psia and 0°F was assumed. The pressure is limited by the weight associated with pressure vessel tankage. Trailer capacity in standard cubic feet is as follows:

$$\begin{aligned}
 \text{Trailer capacity} &= \\
 &5,900 \text{ gal} \times 0.13368 \text{ cu ft/gal} \times 4.29 \text{ lb/cu ft} \\
 &(\text{@ } 300 \text{ psia} + 0^\circ\text{F}) \div 0.11 \text{ lb/cu ft} (\text{@ std cond.}) \\
 &= 0.10246 \text{ MMSCF} (\text{@ std. conditions})
 \end{aligned}$$

* This density is the maximum that could be achieved by turboexpanding (isentropic expansion) from 1400 psia and 70°F.

If Q is the daily CO₂ injection requirement in MMSCFD, then Q/0.10246 is the number of daily trailer loads required. The total requirement for trailers, however, depends upon the total trailer cycle time, which was assumed to consist of the following elements:

Fill time (from manufacturer):	0.75 hr
Trailer hitch and turnaround:	0.15 hr
Travel time (D=one way haul, miles):	2 x D/(30 mph)hr
Drain time (Q=MMSCFD CO ₂ requirements):	<u>0.10246/Qx24 hr.</u>

$$\text{TOTAL TRAILER CYCLE TIME: } (1 + D/15 + 2.459/Q)\text{hr}$$

The above formula is based upon assumption of an average travel speed of 30 mph, irrespective of haul distance, and assumption of drain rate being a function of CO₂ demand rate only (i.e. trailers can be operated to drain rapidly enough). If a sufficient number of tractors are available to pull the tanker-trailers, and the average availability of trailers is 95%, then the total number of trailers required is as follows:

$$\begin{aligned}
T &= \text{Total Trailers} = \\
&\frac{\text{CO}_2 \text{ requirement per hour} \div \text{Trailer availability}}{\text{Trailer capacity} \div \text{Trailer cycle time, hours}} \\
&= \frac{Q/24/0.95}{0.10246/(1 + D/15 + 2.459/Q)}
\end{aligned}$$

$$T = Q(0.4281 + D/35.0416) + 1.0526 \quad (\text{Equation 1})$$

Because the truck tractors need not loiter during trailer filling and draining times, the tractor cycle time is shorter than the trailer cycle time.

$$\begin{aligned}
\text{Trailer hitch and turnaround:} & \quad 2 \times 0.25 \text{ hr} \\
\text{Travel time:} & \quad \underline{2 \times D/(30 \text{ mph}) \text{ hr}}
\end{aligned}$$

$$\text{TOTAL TRACTOR CYCLE TIME:} \quad (0.5 + D/15) \text{ hours}$$

According to various trucking companies we contacted, tractor availability cannot, on the average, exceed 64% (due to maintenance, breakdowns, etc.) so that the total number of tractors required to distribute QMMSCFD is

$$\begin{aligned}
S &= \text{Total Tractors} = \\
&\frac{\text{CO}_2 \text{ requirement per hour} \div \text{Tractor availability}}{\text{Trailer capacity} \div \text{Tractor cycle time, hours}} \\
&= \frac{Q/24/0.64}{0.10246/(0.5 + D/15)}
\end{aligned}$$

$$S = (D/23,607 + 0.3177)Q \quad (\text{Equation 2})$$

Equation 2 does not explicitly include tractor units needed to avoid trailer waiting time that could occur because fewer tractors than trailers are required. Turnaround and travel times were deliberately hedged to provide an extra tractor capacity "cushion" to simplify the calculations without explicit consideration of waiting-time problems. For large daily CO₂ requirements, the trailer waiting time should be negligible.

Table 28 presents estimates of tractor and trailer requirements for various CO₂ distribution requirements and distances from the trunk pipeline to the consumer oil fields. Equations 1 and 2 were used to estimate unit requirements under the assumption that tractor and trailer usage could be prorated among EOR projects of identical size and haul distance. (Put another way, "fractional tankers" and "fractional tractors" are charged to each CO₂ project over the haul distances and at the demands shown in Table 28.) For the 10-mile haul at a 0.2-MMSCFD demand, only about one and a third trailers are required, and these can be serviced by one tractor dedicated only 26.5% of its time. At the other extreme of haul and demand, a 10-MMSCFD CO₂ demand 100 miles from the trunk pipeline requires service by about 34 trailers and 46 tractors. Note that at haul distances and CO₂ demands beyond 25 miles and 2 MMSCFD, respectively, the tractor unit requirements exceed the trailer requirements. (These cases are enclosed by the polygon on Table 28.) This situation is due to the low availability (about 64%) of tractors relative to trailers (95%).

Based upon a brief survey of highway tanker companies and manufacturers' data, we estimate distribution costs to include the following components:

Tractor Purchase Cost: \$50,000

Trailer Purchase Cost: \$45,000

Driver Labor
(including fringe benefits): \$30,000 per year

Tractor Operation
(fuel, tires, maintenance, etc.): \$0.35/haul mile

Trailer Operation
(tires, wheels, maintenance, etc.): \$0.10/haul mile

Tractor Fixed Cost
(dispatching, taxes, insurance, etc.): \$6,000/year

Trailer Fixed Cost: \$1,000/year

The ICC permits 7-year, straight line depreciation to be taken on 90% of the investment in a tractor and trailer. Tables 29 and 30 display details of the development of annualized tractor and trailer charges as functions of one-way haul distance and daily CO₂ requirements, and for 15% discount rate and 50% effective tax rate. For a given demand (Q) and haul distance (D), the product of the total tractor requirement times the annualized cost per tractor, plus the product of the total trailer requirement times the annualized cost per trailer, is the total CO₂ distribution cost by truck. (A 340-day operating year is assumed.)

Table 31 summarizes the truck-and-trailer transport costs for truck-delivered CO₂ (Complete cost tables are presented in Appendix A) For trailers loaded with saturated-vapor CO₂ at 0°F and 300 psia, Table 31 shows that the delivered-gas cost varies from \$0.304/MCF to haul 10 MMSCFD over 5 miles, to \$3.086/MCF to haul 0.5 MMSCFD over 100 miles. Additional Costs for truck-delivered CO₂ at the transport conditions of Table 31 include turboexpansion from 1400 psia (minimum trunk pipeline pressure) and 70°F to 300 psia and 260°F for injection at the trailer delivery point. Costs of turboexpansion (less recovered work) and recompression total about \$0.07/MCF. Total truck-delivered CO₂ cost thus ranges from \$0.374/MCF to \$3.156/MCF over hauls of 5 to 100 miles and for demands of 0.5 MMSCFD to 10 MMSCFD.

Different trailer load conditions of temperature and pressure change the CO₂ costs. The 5,900-gal Lubbock trailer is rated to carry 44,000lb of commodity load. If the trailer carries saturated liquid at 0°F and 300 psia (approximately 7lb/gal), then each trailer can deliver 375,480 SCFD, a 266% increase over the saturated-vapor transport case. For 0.5 MMSCFD CO₂ and a 100-mi haul, the transport cost is \$1.066/MCF, while at 10 MMCFD and a 5-mi haul, the transport cost is \$0.100/MCF. To these costs, however, must be added the cost of refrigerating the saturated vapor at 300 psia and 0°F to saturated liquid, a change of 121 Btu/lb. The cost of refrigeration is about \$0.34/MCF, so that the total cost of transporting a saturated liquid ranges from \$0.420/MCF to \$1.486/MCF.

We also considered a CO₂ distribution by a 5,900-gal trailer rated for 1000 psia and 67°F. Such a trailer could be constructed using state-of-the-art technology in pressure vessel design, as with Kevlar wrapping on a steel shell, the design of such a trailer is beyond the scope of the present study. Assuming that the trailer cost would double (\$90,000 trailer), we estimate that truck-delivered CO₂ cost would vary between \$0.948/MCF (0.5 MMCFD and 100 miles) and \$0.096/MCF (10 MMCFD and 5 miles). Additional costs for turboexpansion and recompression we estimate at less than 5¢/MCF.

Truck vs. Lateral Pipeline

Figure 16 compares lateral-pipeline, 300-psi-truck; (saturated vapor), and 1000-psi-truck transport system costs. The saturated-vapor, 300-psi-truck system is economically inferior to the lateral pipeline at all CO₂ rates and distances. However, the 1000-psi (saturated vapor) truck is economically preferred for delivering CO₂ at distances beyond 15 miles (assuming, of course, that such a trailer could be built for \$90,000 or less).

Impact on CO₂ Reserves & Recoverable Oil

The prospects for generating substantial quantities of naturally occurring CO₂ by the co-production of natural gas-CO₂ mixtures were discussed in Section I of the report. The prospects hinged on the existence of development of efficient means of separating relatively low concentrations (5% CO₂) from natural gas streams, the development of new natural gas production regions whose productive lives would coincide with the time frame of high CO₂ demand and efficient means of transporting small quantities of CO₂ over reasonably short distances.

The processing techniques discussed in the previous sections provide a large array of methods for separating CO₂ from natural gas mixtures of varying concentrations. In most cases, the natural gas will have to be separated from the CO₂ in order to meet the natural gas market specifications. The principal cost of the CO₂ processing is merely the cost of pressurizing the CO₂ to the required transport associated with high-purity process-vent sources, such as, anhydrous ammonia plants. The only additional cost associated with co-produced CO₂ could be the cost of gathering several relatively small quantities of CO₂ from the natural gas processing site. The practice that seems to be evolving in the Overthrust Belt is to build gas processing plants in modules of 250 MMSCFD capacity. If the feed gas CO₂ concentration were 5%, then a 250 MMSCFD processing plant would provide a CO₂ source of 12.5 MMSCFD, a source equivalent to a small anhydrous ammonia plant. The need to gather many of these sources would depend on the proximity of petroleum reservoirs.

If the estimate of ultimate natural gas recovery from the Western Overthrust Belt and the Tuscaloosa Trend are realized, then 14 TSCF of additional low-cost CO₂ could be available for CO₂-flooding projects. This could result in the production of an additional 2.7 billion barrels of recovered oil under the assumptions presented in Reference 1. While the above-mentioned source does not completely resolve the CO₂ shortfall problem, it does present a source that is approximately three times the size of currently available high-purity process-vents.

IV. Conclusions and Recommendations

A. Conclusions

The following objectives were accomplished:

- Defining the CO₂ requirements for candidate EOR reservoirs.
- Generating alternative concepts for collecting, transporting, storing and processing CO₂ from naturally occurring sources and comparing them on the basis of cost effectiveness.
- Re-estimating the magnitude of the CO₂ EOR reserves based on the findings of this study.

The details for each of the above are given in the "Results" section. The conclusions derived from the study are divided into the following:

- Conclusions from the processing alternatives
- Conclusions from the transportation alternatives
- Overall effects of conclusions from processing and transportation on CO₂ - EOR reserves.

Conclusions from the Processing Alternatives

After studying about 50 different natural gas purification processes for separating CO₂, the following conclusions were reached:

- If the amount of CO₂ separated out of a natural gas exceeds 5 MMSCFD, the natural gas containing CO₂ should be sweetened even if it contains only 2-3 mole % of CO₂.
- As a rule of thumb, in separating CO₂ from natural gas, chemical solvents should be used when the CO₂ content is less than 4-5 mole %, physical solvents when the CO₂ content is 5-30 mole %, and a combination of refrigeration (using turboexpansion when high wellhead pressures are available) and physical solvents when the CO₂ content is greater than 30 mole percent.

- Cost of CO₂ prior to transportation depends on
 - mode of transportation (.30-.65 \$/MMSCF)
 - type of purification (.29-.30 \$/MSCF)
 - whether or not it includes all, adjusted capital and operating cost, or none of the purification cost (.03, 0.18, 0.21 \$/MSCF)
- Cost of processing CO₂ prior to transportation depends on the mode and conditions required at the loading point. The cost of processing compared to the cost of transportation ranges from about 10 to 75 percent, even when all of the purification costs are burdened on to the CO₂ costs.

Conclusions From The Transportation Alternatives

Various modes of transportations and combinations were studied. Conclusions are:

- Pipelines and truck-trailers at about 1000 psi have a distinct advantage over trucks at 300 psi for all distances considered.
- Trucks at 1000 psi and ordinary temperatures (90°F) have an advantage over lateral pipelines up to distances of 100 miles.
- Refrigerated trucks at low pressure (300 psi, 0°F) have an advantage over unrefrigerated low pressure trucks (300 psi, 90°F) due to a great change in density of CO₂ as a result of a phase change. Total cost of processing plus transport (100 miles) for an unrefrigerated low pressure truck is the highest (see table 9).

Conclusions on CO₂-EOR Reserves

If natural gases containing less than 10 mole percent CO₂ are processed, and trucks/trailers at 1000 psi and 90°F are used to transport it (a distance of 100 miles), the CO₂ can be made available at the point of consumption for about \$1.15/MSCF. The above cost includes all of the purification cost for 100 MMSCFD of natural gas containing 7.5 mole % CO₂.

- Considering only the natural gases from the Western Overthrust Belt and the Tuscaloosa Trend, the CO₂ reserves could be increased about 14 trillion standard cubic feet.

B. Recommendations

It is recommended that DOE include the following in their future programs:

- Experimental work determining vapor-liquid equilibrium ratios for mixtures of methane-carbon dioxide at temperatures and pressures encountered in processing the natural gas.
- Experimental work determining other physical properties, such as heat capacity, conductivity, compressibility, viscosity, etc. for CH₄-CO₂ mixtures containing other impurities, such as H₂S or others which exist in natural gases in the United States.
- High pressure truck/trailers should be studied in greater detail to determine an optimum pressure considering practical construction materials plus transportation rules and regulations.

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Fig. 1 GAS TREATING FOR CO₂ REMOVAL

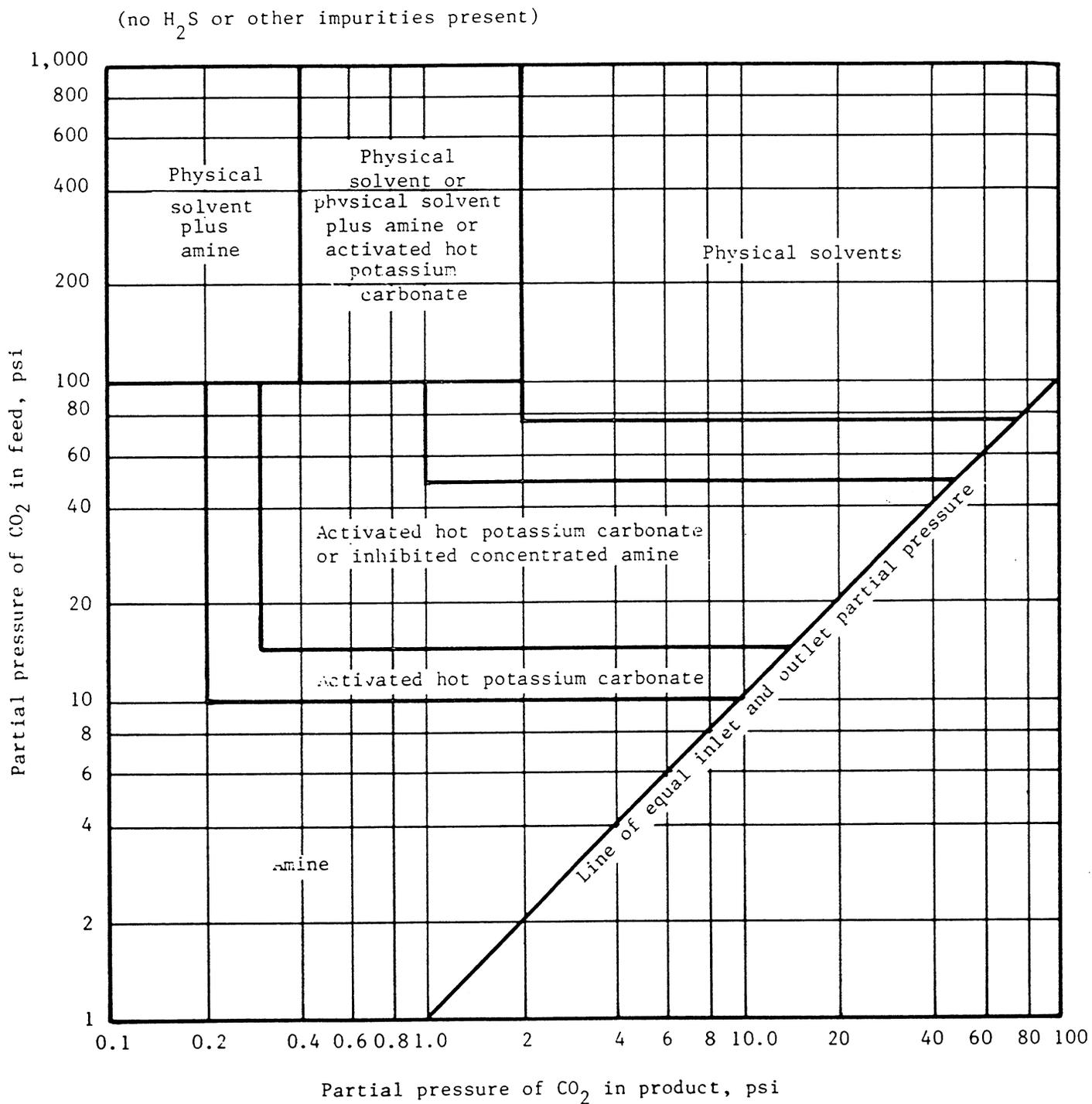
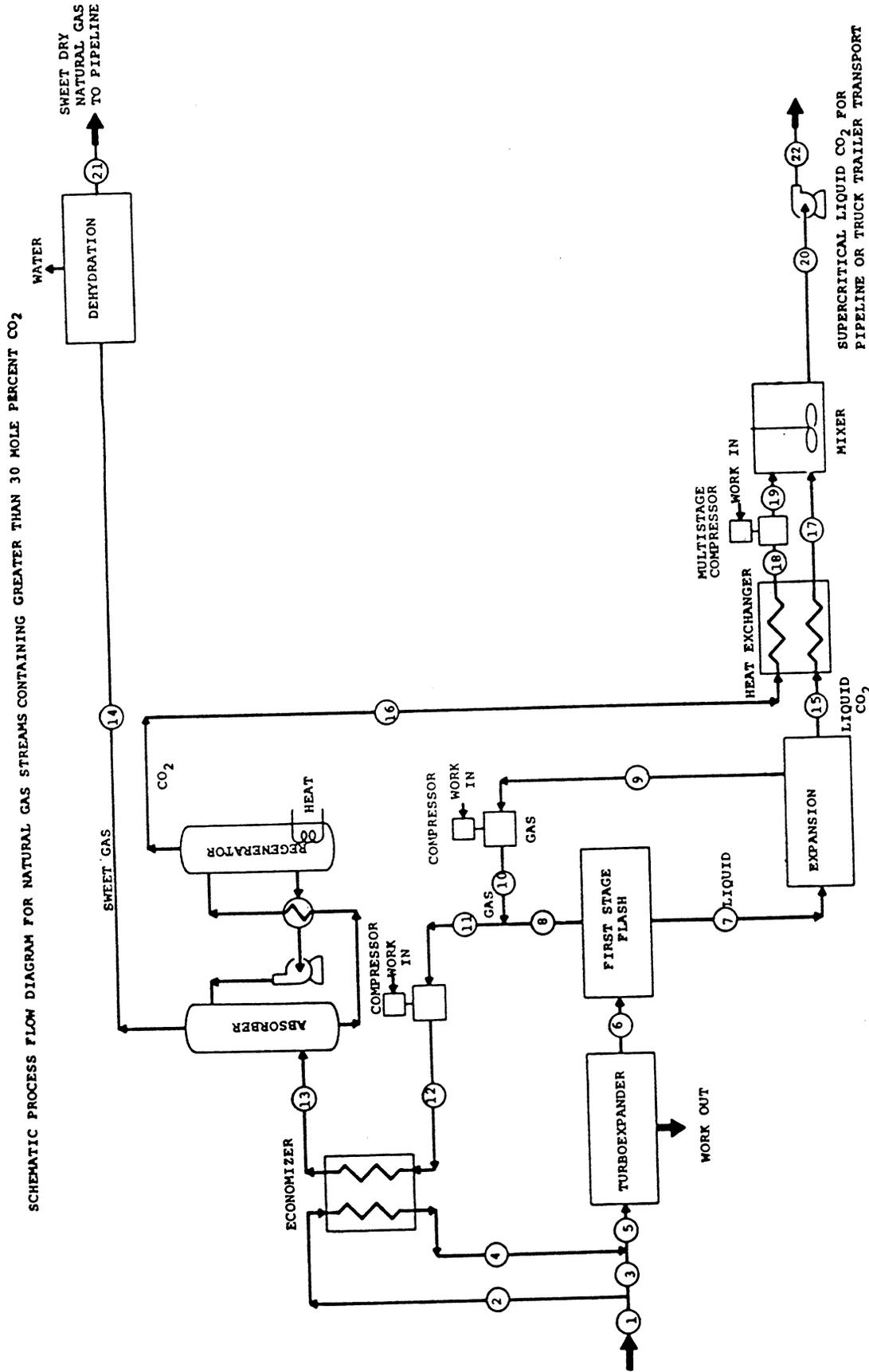


Fig. 2



Description	Stream No.																					
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22
Amount																						
CH ₄ Gas	50	50	0	50	50	50	0	19	24	24	43	43	43	0	0	0	0	0	0	0	43	7
CH ₄ Liq	0	0	0	0	0	0	1601	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CO ₂ Gas	50	50	0	50	50	50	0	8.7	12	12	20.7	20.7	20.7	0	0	0	0	20.7	0	0	0	0
CO ₂ Liq	0	0	0	0	0	0	33780	0	0	0	0	0	0	0	19863	0	20318	0	14333	34651	0	34552
Temperature, °F	110	110	0	90	90	-30	-30	-30	-30	11	-10	5	52	100	-30	400	55	45	55	110	65	65
Pressure, Psi	2093	2093	0	2088	2088	1000	1000	1000	1000	1000	1000	1110	1100	1090	750	15	745	15	750	750	1080	1000

Fig. 3 50% CH₄ - 50% CO₂

● -40F

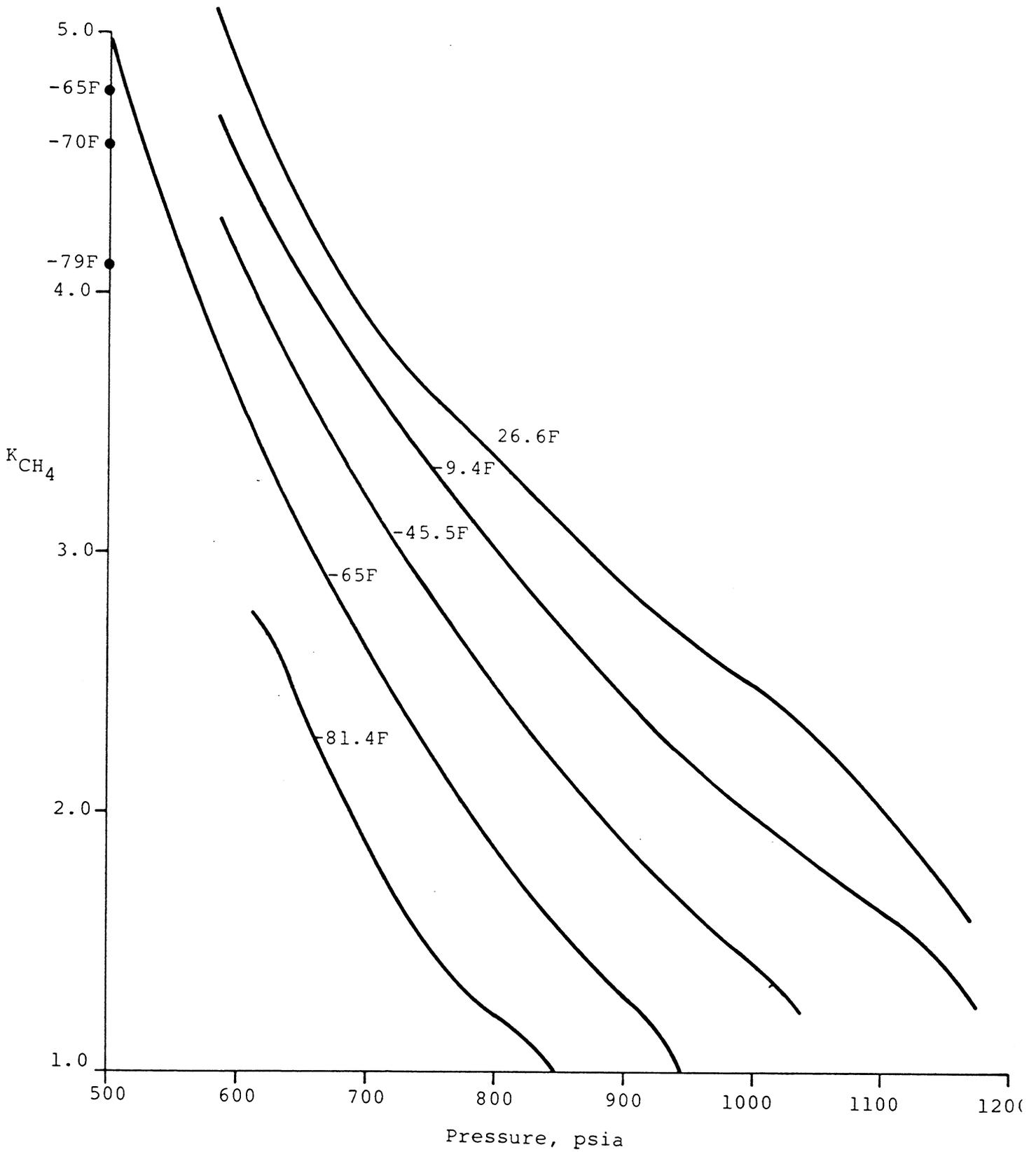


Fig. 4 50% CH₄ - 50% CO₂

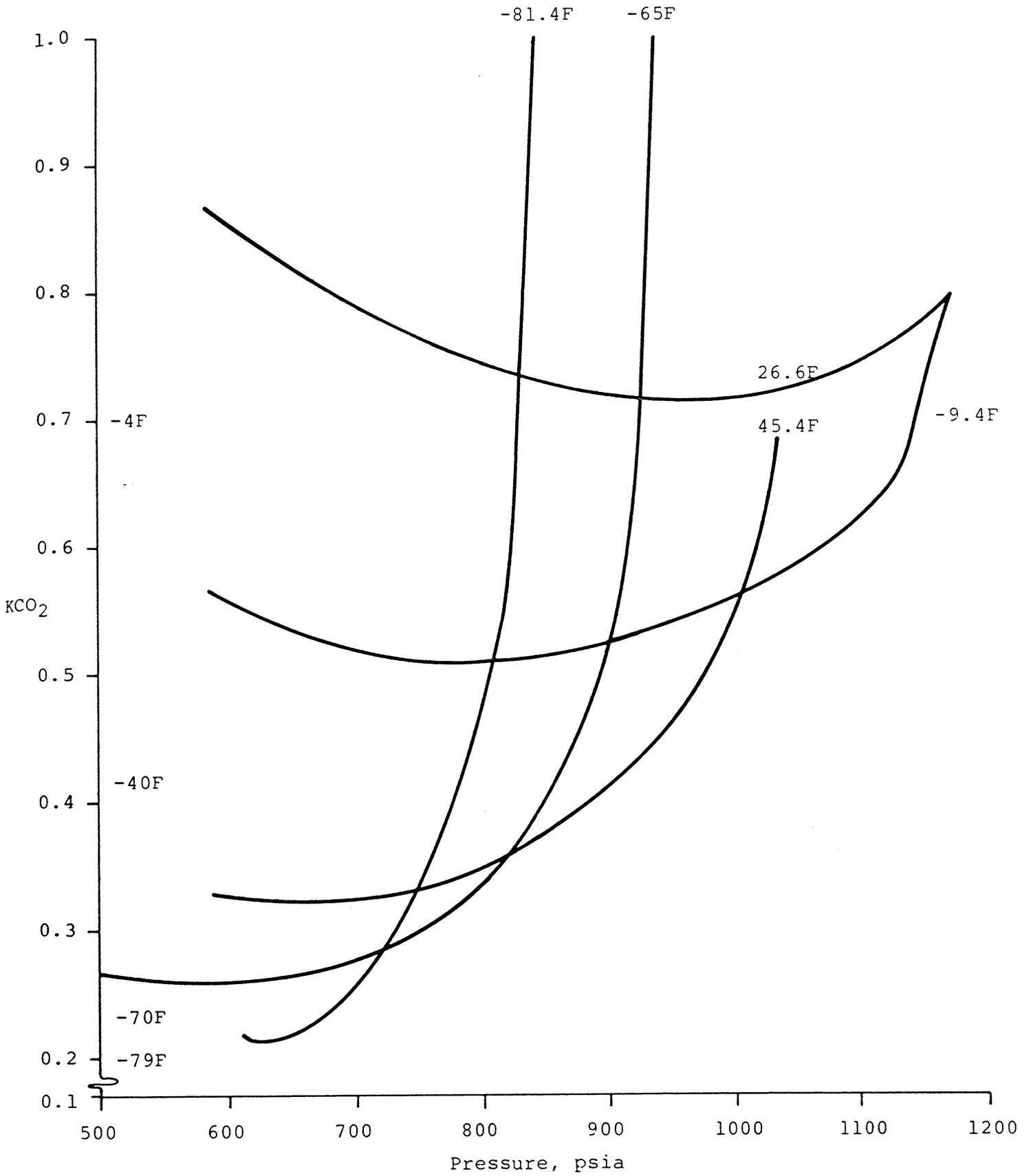


Fig. 5 40% CH₄ - 60% CO₂

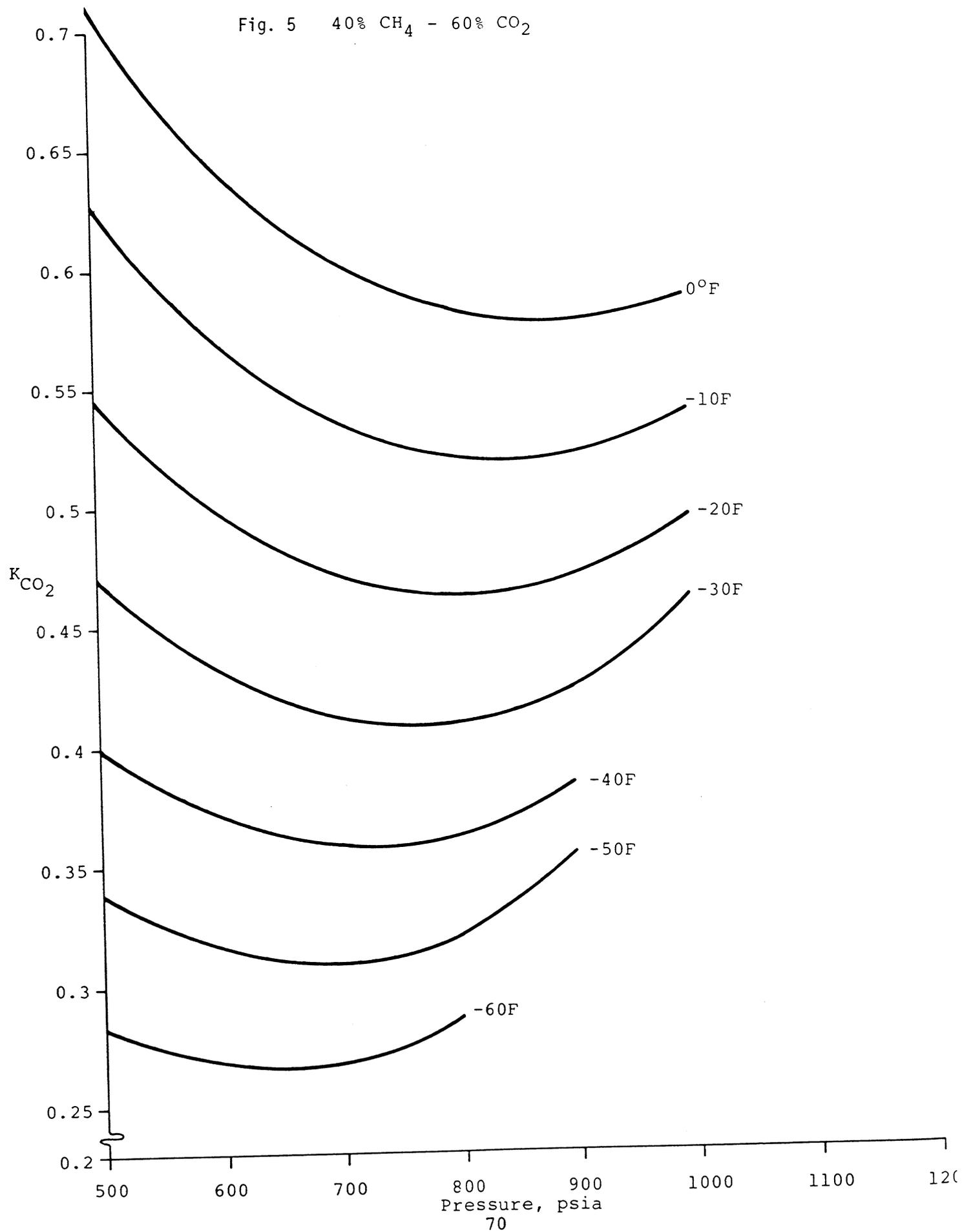


Fig. 6 40% CH₄ - 60% CO₂

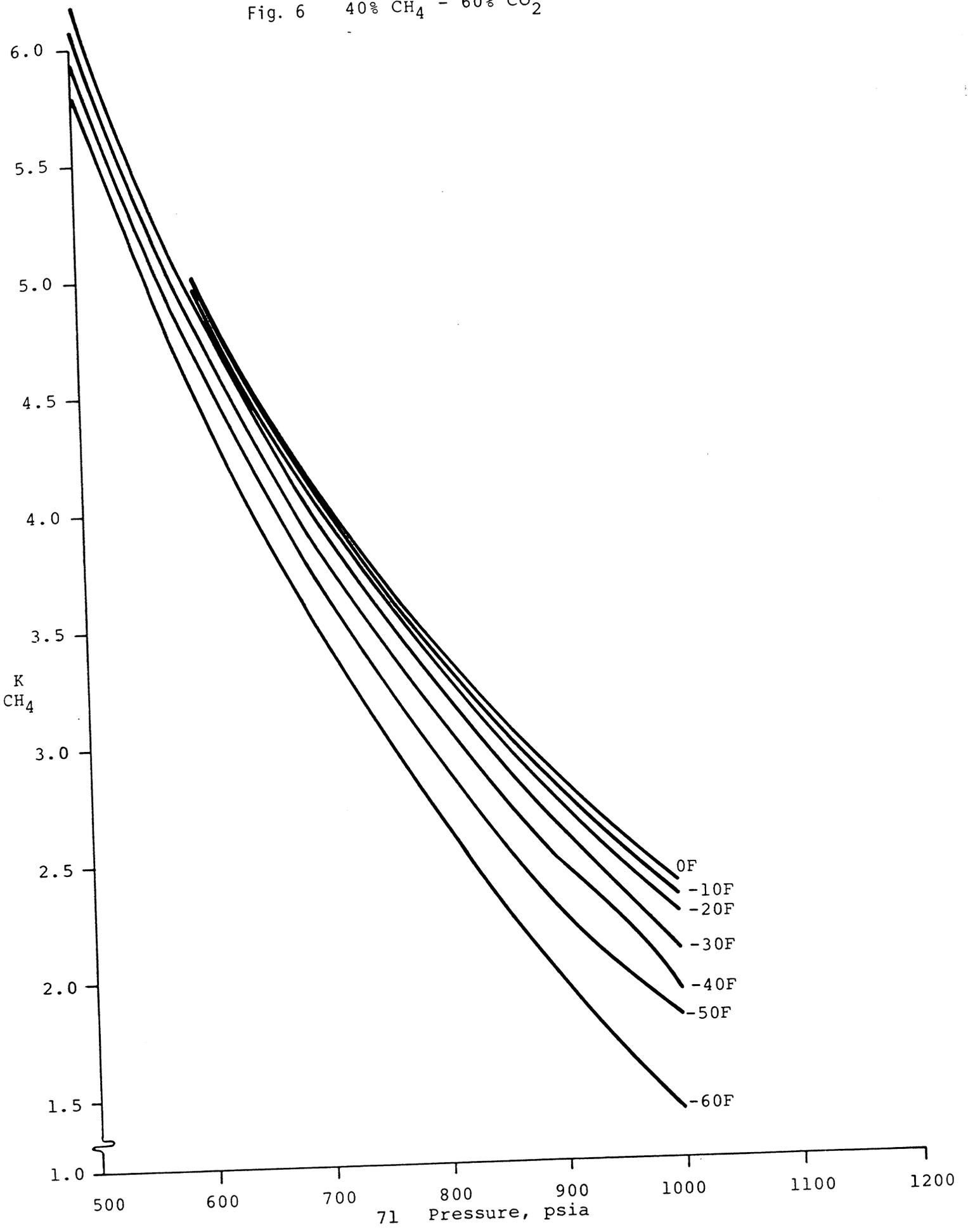


Fig. 7 -50°F for CH₄

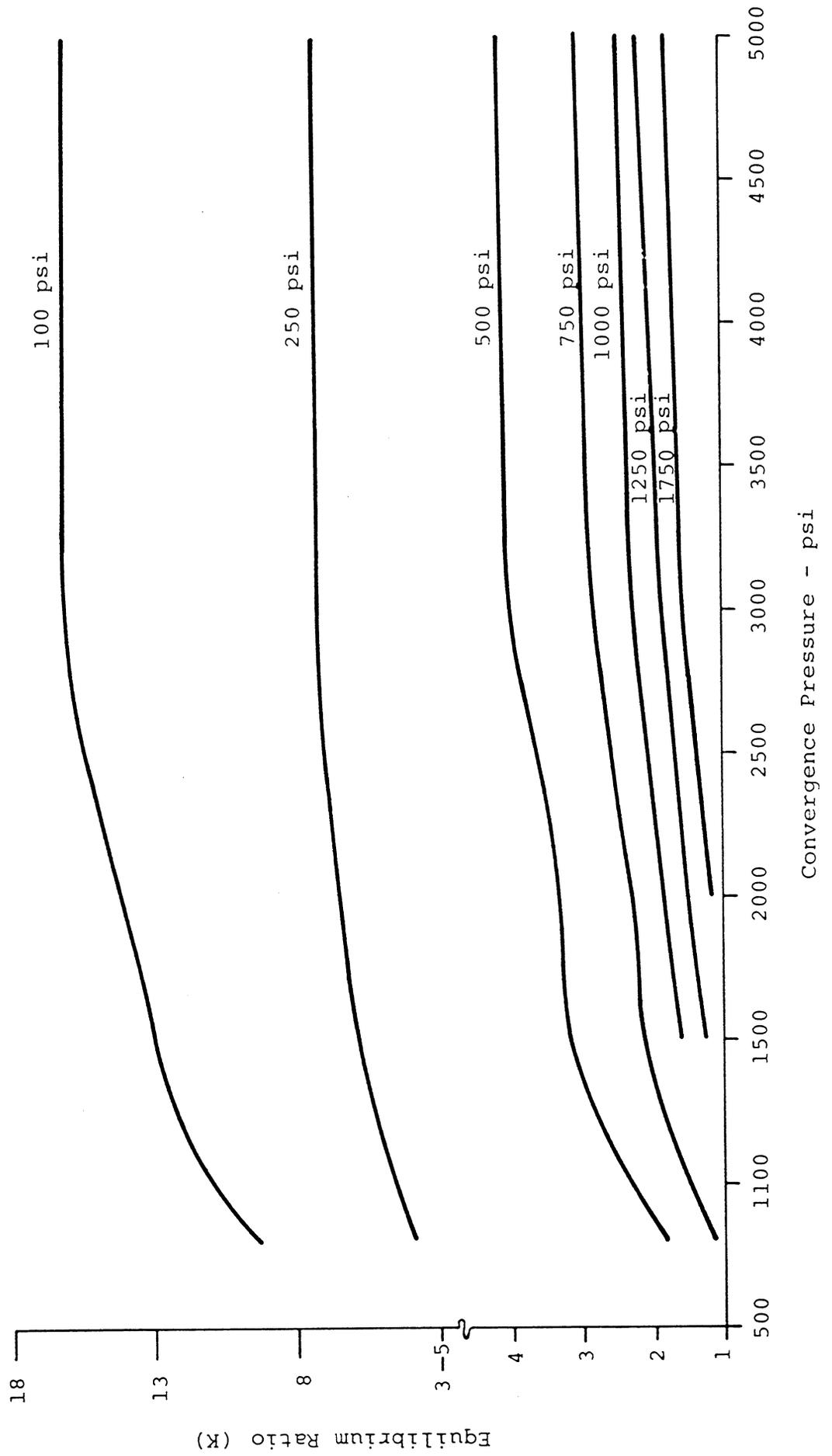


Fig. 8 0°F - K for CH₄

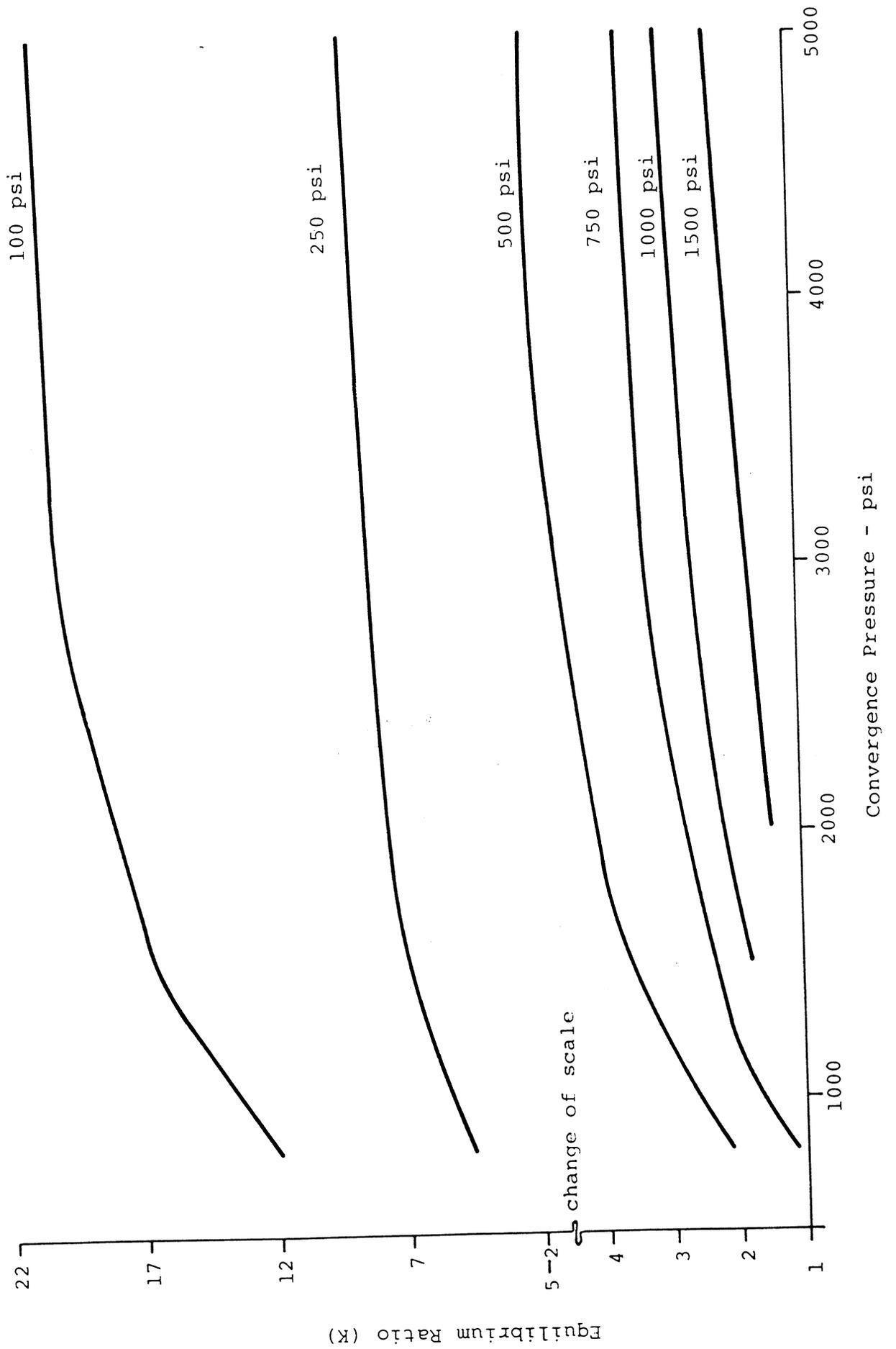


Fig. 9 +50F for CH₄

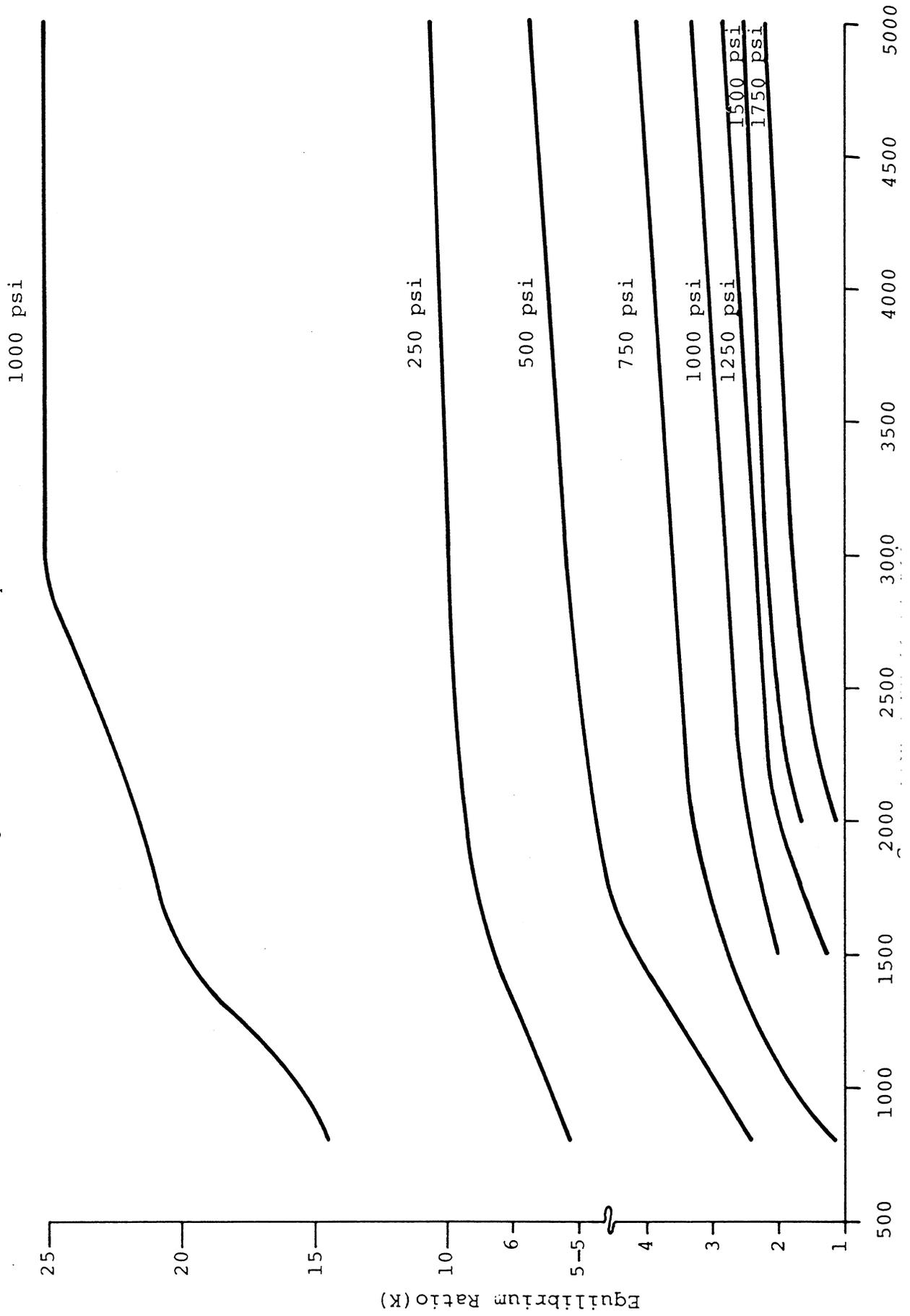


Fig. 10 Critical Pressure vs. Critical Temperature for Mixtures of Methane and Carbon Dioxide

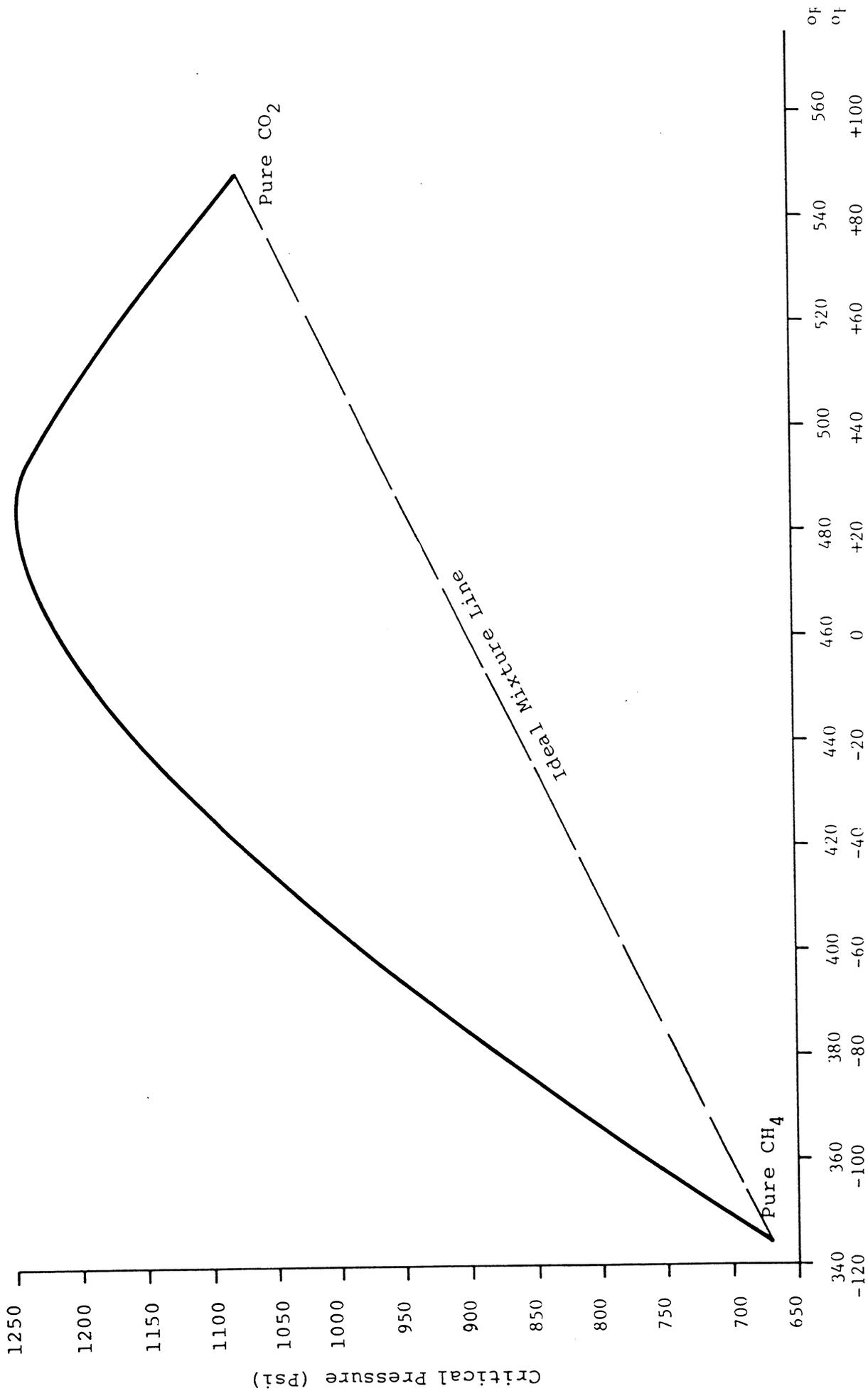


Fig. 11 CARBON DIOXIDE THERMODYNAMIC PROPERTIES
 PRESSURE-ENTHALPY (MOLLIER) DIAGRAM

Critical Pressure = 1071 PSIA
 Critical Temperature = 88°F

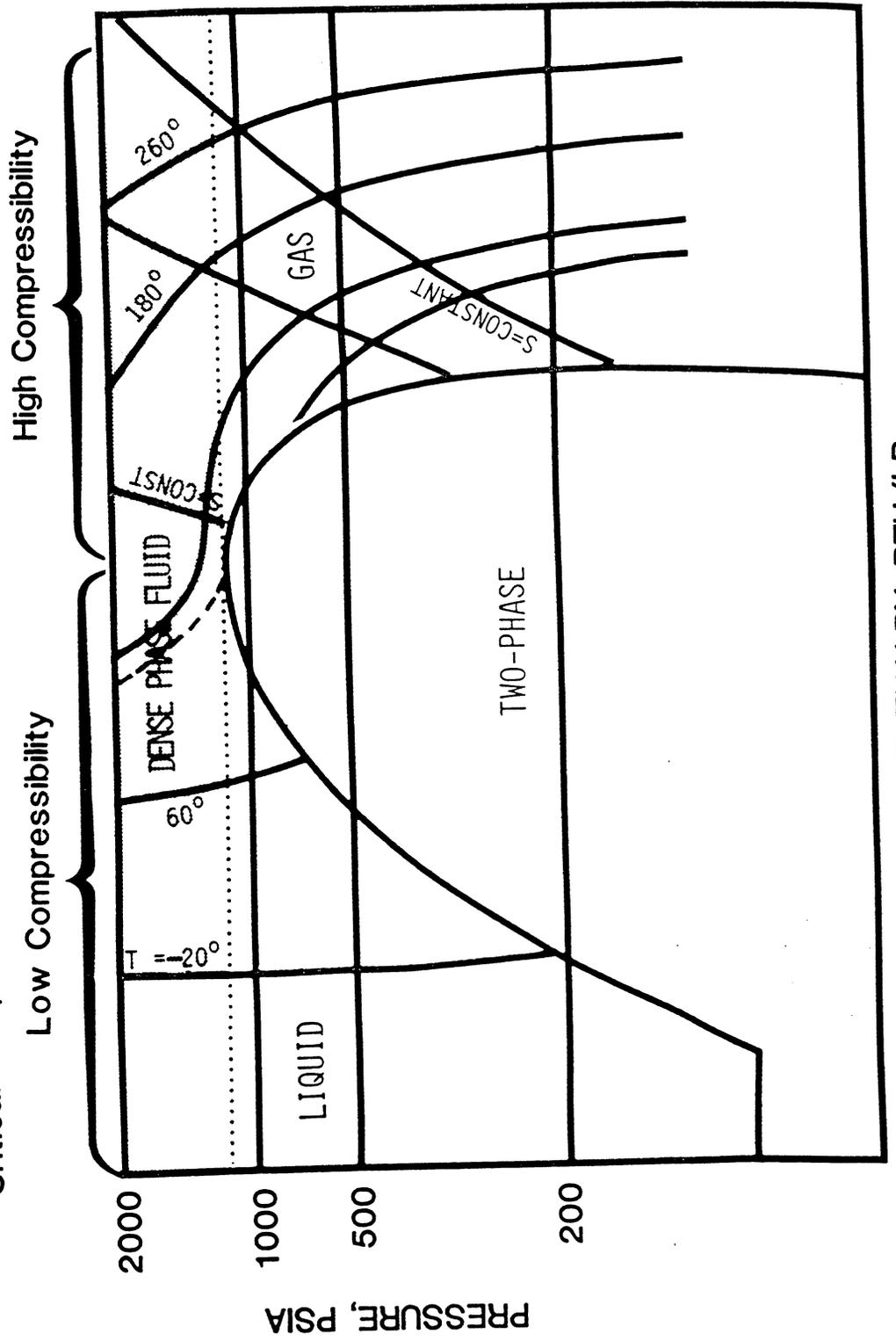


Fig. 12 INITIAL PRESSURIZATION OF CO₂

Thermodynamic Comparison

Compressor/Pump Work = $\Delta h_1 + \Delta h_2$

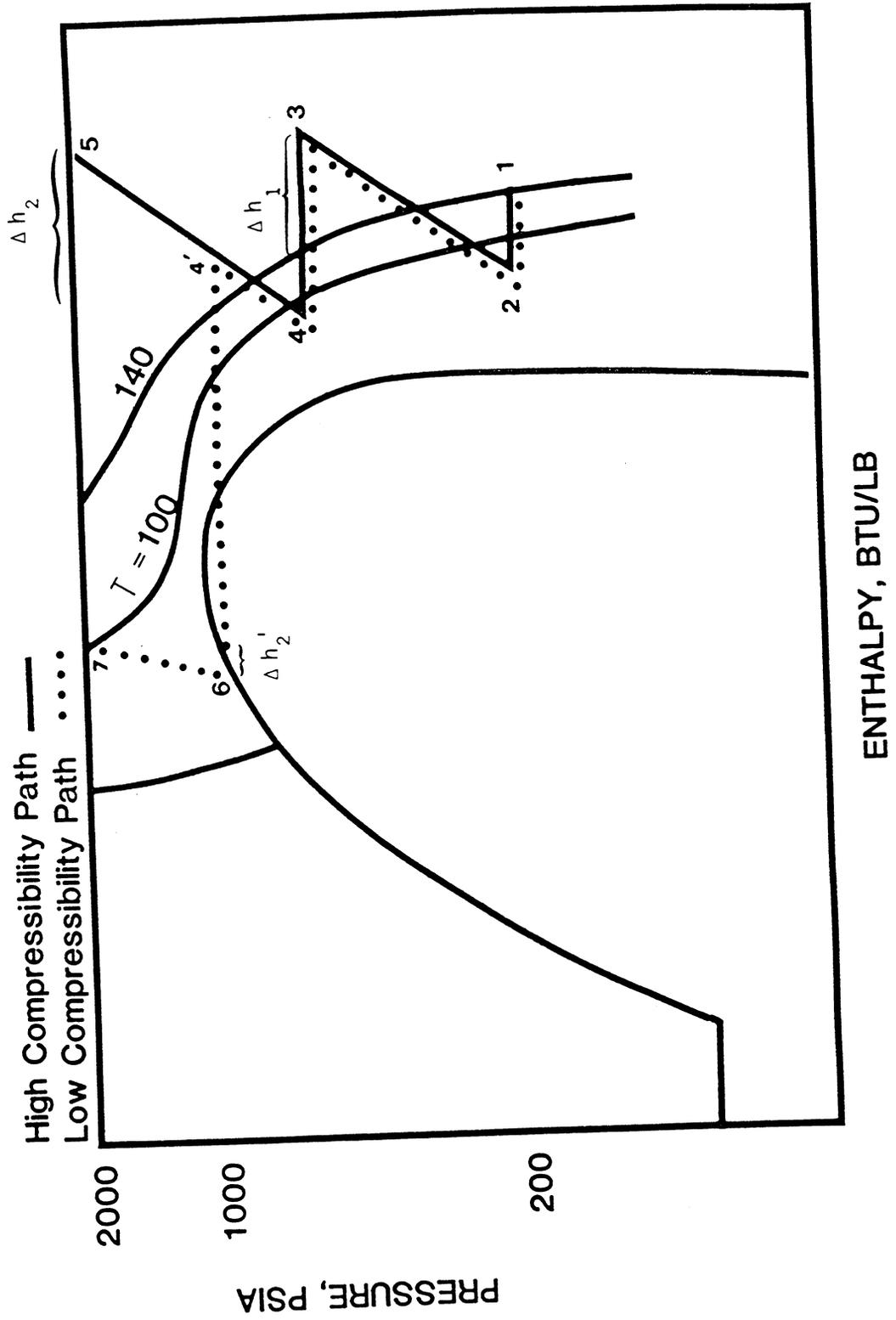


Fig. 13 INITIAL PRESSURIZATION OF CO₂
 COMPRESSOR HORSEPOWER COMPARISON

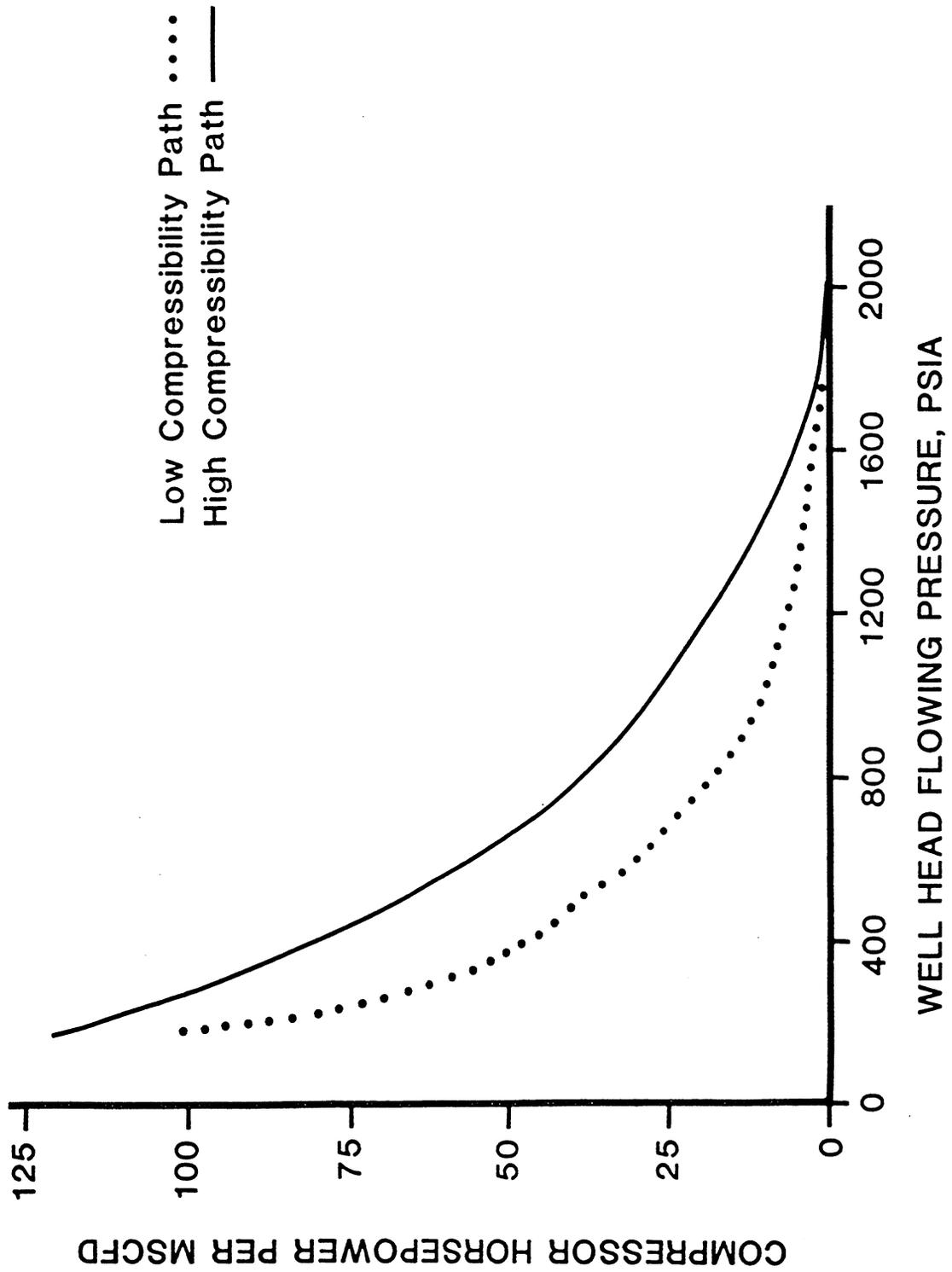


Fig. 14 TEMPERATURE DISTRIBUTION IN 50 MMSCFD CO₂ PIPELINES

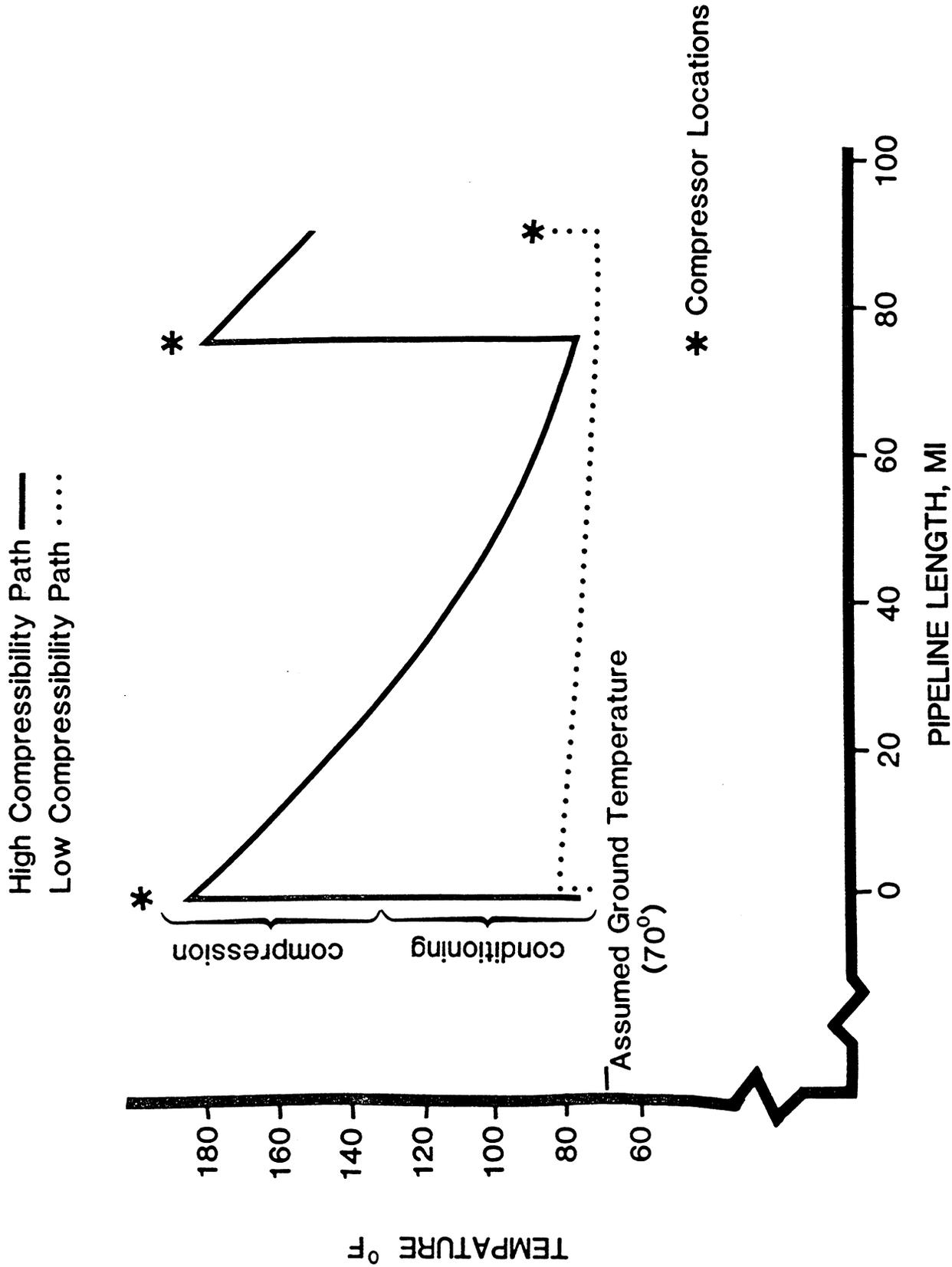


Fig. 15 PIPELINE REPRESSURIZATION & THERMAL CONDITIONING OF CO₂

Thermodynamic Comparison

High Compressibility Path —
 Low Compressibility Path ···

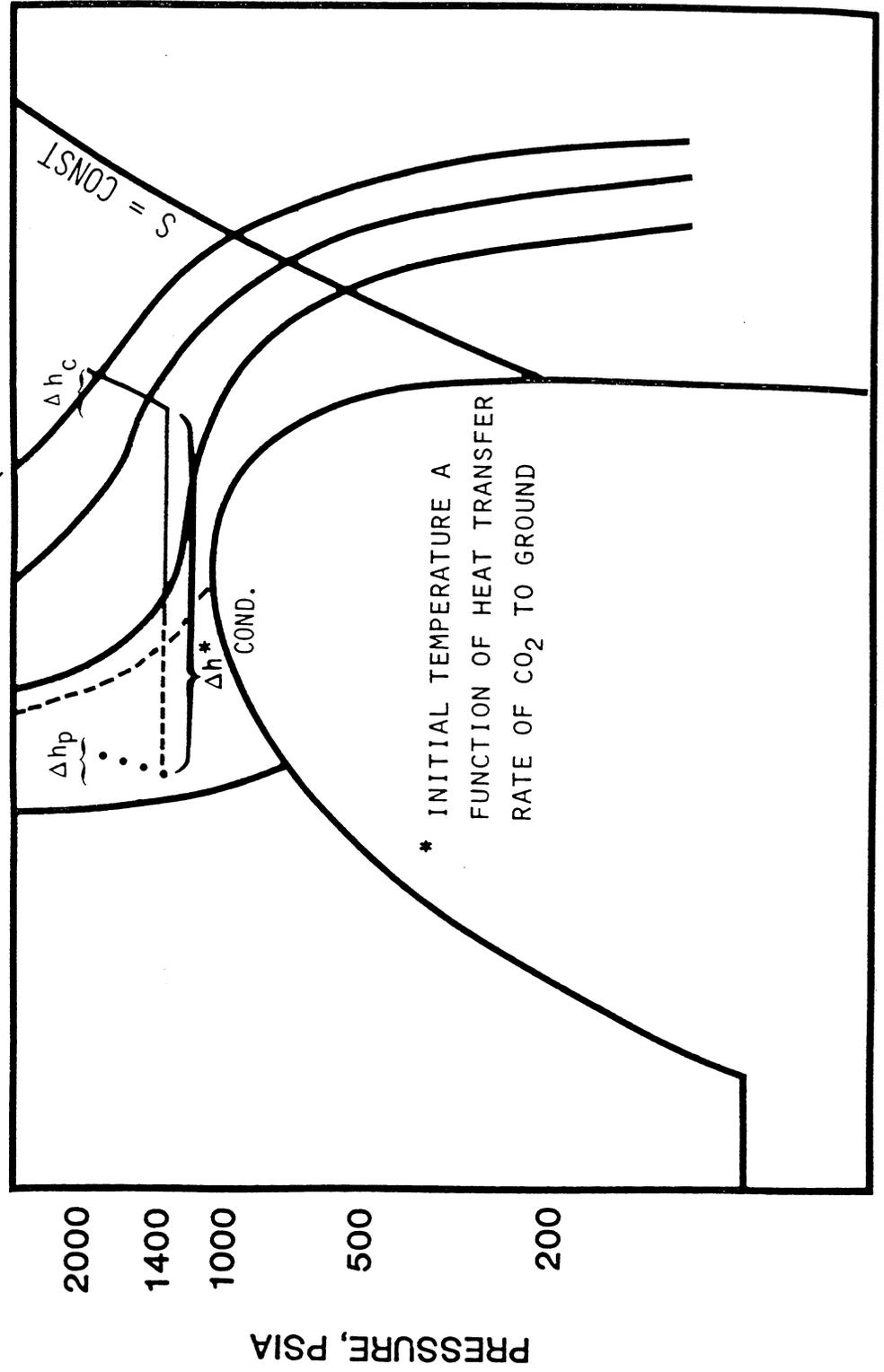
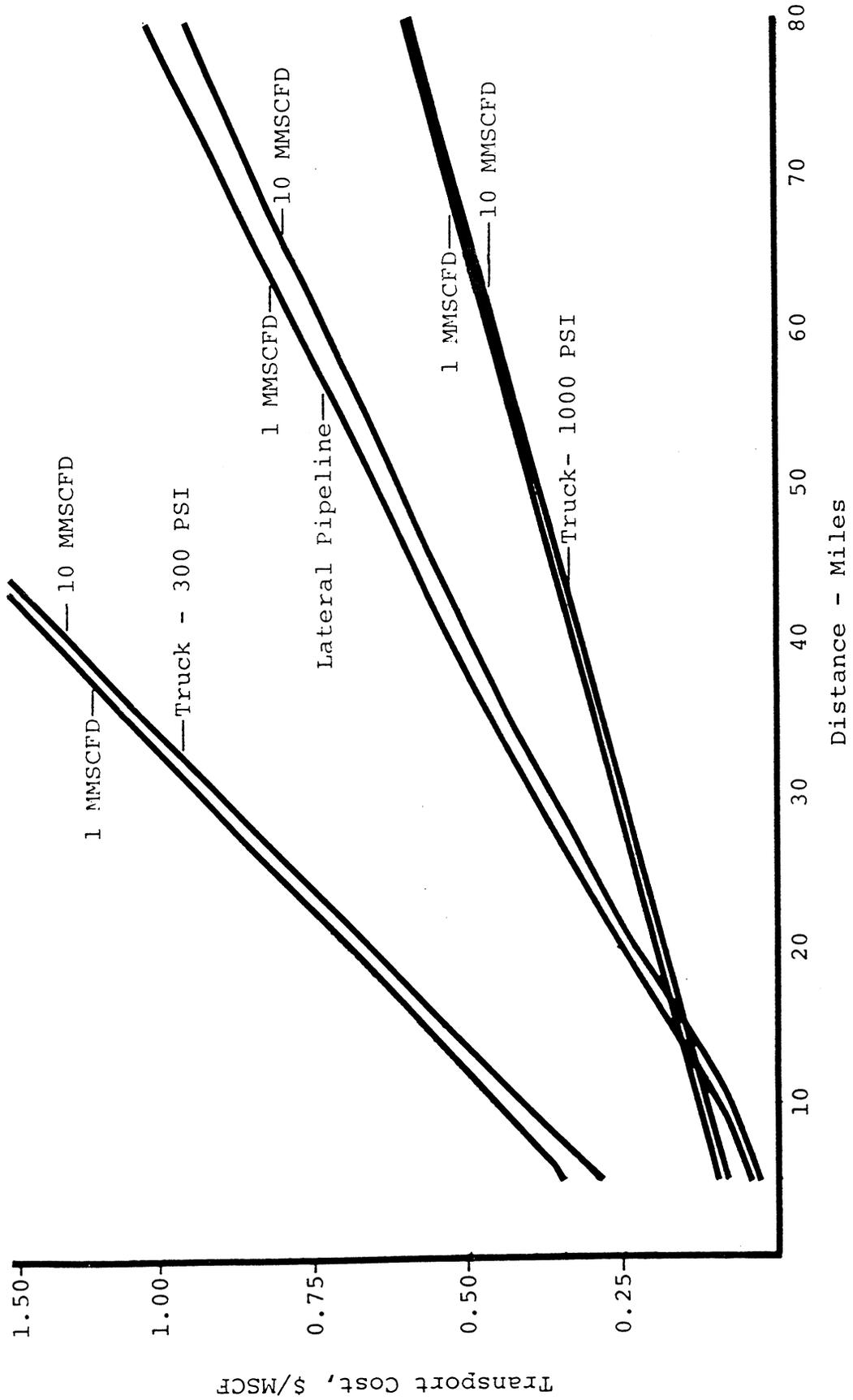


Fig. 16 COMPARISON OF TRUCK AND LATERAL PIPELINE
CO₂ TRANSPORT COST



SUMMARY OF CO₂ SUPPLY PROJECTS

	<u>Operator</u>	<u>Origin</u>	<u>CO₂ Source</u>	<u>Capacity (MMSCFD)</u>	<u>Terminus</u>	<u>Field Served</u>
2	SACROC	Terrell & Pecos Counties, TX	Val Verde Basin Natural Gas (18 to 53% CO ₂)	240	McCamey (Upton Ct) & Snyder (Scurry Ct.), TX	North Crossett & Kelly-Snyder
3	Shell	Dolores & Montezuma Ct, CO	McElmo Dome & Doe Canyon CO ₂ Fields (96-99% CO ₂)	400	Denver City (Yoakum Ct.) TX	Wasson/Denver Unit
3	ARCO	Huerfano Ct, CO	Sheep Mt & Dike Mt CO ₂ Fields (97-99% CO ₂)	300	Denver City (Yoakum Ct) & Seminole (Gains Ct) TX	Wasson/Willard Unit
4	AMACO	Union & Harding Counties, N.M.	Brovo Dome Unit CO ₂ Fields (99-100% CO ₂)	?	West Texas	Probably: North Cowden Slaughter and Leve-land
5	SHELL	Rankin Ct, MISS	Jackson Dome CO ₂ Field (98-99% CO ₂)	200	Lincoln & Pike Cts, MISS (&beyond)	Little Creek
6	TRANS- PETRO	Borger (Hutchinson Ct), TX	Cominco, Inc. Anhydrous Ammonia Plant Vent	27	Ochiltree & Hansford Cts., TX	Marmaton
7	CITY SERV. ARCO	Garfield Ct, OK	Farmlands Ind Anhydrous Ammonia Plant Vent	35	Garvin Ct., OK	E. Velma & N.E. Purdy
8	COLUMBIA GAS	Kanawa Ct., W VA	Indian Creek Natural Gas/CO ₂ Field (66% CO ₂)	30	?	?
9	CONOCO	Pecos Ct, TX	Val Verde Basin Natural Gas	20	Orla (Reeves & Culberson Cts.) TX	Ft. Geraldine

TABLE 1

Table 2 Comparison of Some Commonly Used Gas Purification Processes Considering Limitations on the Feed and the Product Streams

Process/Type	Allowable ranges of Acid Gas Concentrations in the Feed	Feed Conditions Allowable	Acid Gas Concentrations in Product	Product Conditions	Acid Gas Conditions	Special Considerations
Fluor Ecoamine Process, Diglycolamine (DGA) Sorbent	Usually 2 - 5% total acid gas - been used to 11.4% CO ₂ with nil H ₂ S and 31.7% H ₂ S with 3.9 CO ₂	< 160°F with no limit on pressure with previous design flow rates running from 3 to 550 MMSCFD	0.01 mole % CO ₂ , 0.25 grains/100 SCF H ₂ S capable of 4 ppmv H ₂ S and 50 ppmv CO ₂	90 - 220°F pressure dependent on feed, etc.	Near atmospheric pressure ambient temp.	Will remove other organic sulfurs
Fluor Solvent Process, Propylene Carbonate Sorbent	Acid gas partial pressure > 75 psi	Feed gas pressure whatever, ambient or below temperatures	< .25 grains H ₂ S/100SCF < 2% CO ₂	Subambient temperatures and atmospheric pressures	-25°F, Atmospheric pressure	Can be used for selective removal H ₂ S from CO ₂ , H ₂ S; also will remove other organic sulfur. Inert gas is injected in the stripper diluting acid gas
Benfield Processes - Hot Potassium Carbonate & High Pure Modification	5 - 50% acid gas max to date 12% H ₂ S + 42% CO ₂	Moderate to high pressures, (10 - 225 MMSCFD, 100 - 2000 psig, ambient to 400°F. Highest acid concentrations	Hot Carbonate CO ₂ < 500 ppm H ₂ S < 1 ppm Hi Pure CO ₂ < 20 ppm H ₂ S < 1 ppm	100-200 psig pressure & temperature - ambient to 400°F	Ambient Conditions	Will handle substantial H ₂ O & heavy hydrocarbons selective H ₂ S absorption capability
Union Carbide - Molecular Sieve	0.1 - 50% CO ₂ < 1 - 400 grains H ₂ S/100SCF	110 - 115°F + 200 - 260 psig or greater	< 50 ppmv CO ₂ < 1/4 grain H ₂ S/100 SCF	1 - 2 psi less than inlets and same temperature	Low pressure, probability but not necessarily atmospheric Acid gas at elevated temps.	- Regeneration gas is required so CO ₂ is not recovered in a pure stream - Molecular sieves can remove water & heavy hydrocarbons as well
Shell ADIP Disopropanolamine (DIPA)	High concentrations of acid gas, < 1 to > 50% H ₂ S < 1 to > 50% CO ₂	100 - 140°F usually < 100°F psia	< 50 ppmv H ₂ S to < 1 ppmv H ₂ S as desired variable CO ₂ recovery generally proportional to small	40 psi, 110°F	Ambient Conditions	Selective absorption of H ₂ S even in very high concentrations of CO ₂
Shell Sulfinol Sulfolane (tetrahydrothiophenedioxide) & alkanolamine	< 1 to > 50% H ₂ S < 1 to > 50% CO ₂	110°F + 1000 psia typically, pressure can vary from 22 to > 1330 psia	< 1 ppmv H ₂ S+CO ₂ < 50 ppmv CO ₂ usually < 2-3% achievable	1000 psia & 113°F	Ambient Conditions	Can be used for selective removal of H ₂ S
Selerol - Allied Chemical, dimethyl ether of polyethylene	Typical case 30% CO ₂ , 20% H ₂ S concentrations not critical	1000 psig, ambient or lower temperature	< 2 ppmv CO ₂ < 1/4 gr/100 scf	Ambient temp. 1000 psig	Ambient Conditions	Energy efficient, can be somewhat selective

Table 3 Composition, Heating Value, Specific Gravity, Wobbe Number of Natural Gas Samples from Various States, Counties, and Fields in U.S.A.

Component	Gas Sample Number											
	1	2	3	4	5	6	7	8	9	10	11	12
Methane	18.9	30.3	40.5	49.7	79.4	93.6	83.8	73.1	66.7	68.4	60.2	26.2
Ethane	0.2	1.8	2.4	3.3	0.8	2.5	8.9	7.7	12.5	7.9	15.8	15.8
Propane	0.1	0.9	1.2	1.1	0.2	1.2	2.8	6.6	8.5	10.4	12.0	26.0
N-Butane	--	0.3	0.4	0.2	T	0.3	0.6	2.6	4.3	4.0	3.7	14.2
Isobutane	--	0.2	0.3	0.3	0.1	0.4	0.5	1.1	1.8	3.0	2.8	5.5
N-Pentane	--	T	T	T	T	0.1	0.1	0.7	1.5	1.2	0.9	4.5
Isopentane	--	0.1	0.2	0.1	T	0.1	0.3	1.2	1.4	1.6	1.8	3.4
Cyclopentane	--	T	T	T	T	0.1	T	0.5	0.1	0.4	0.2	0.4
Hexanes plus	--	T	T	T	T	0.2	0.2	0.8	0.5	1.0	1.4	1.6
Nitrogen	3.0	26.2	25.7	42.3	9.4	0.7	1.1	4.3	1.0	0.4	0.3	2.1
Oxygen	0.4	--	--	--	--	--	--	0.4	0.1	0.1	T	0.3
Argon	--	0.1	0.1	0.2	0.1	--	--	T	T	--	--	--
Hydrogen	--	0.4	0.3	--	T	--	--	--	T	--	--	--
Carbon Dioxide	77.3	0.1	0.5	T	9.7	0.9	1.7	0.9	1.6	1.8	0.9	T
Helium	--	39.5	28.4	2.8	0.2	T	--	--	--	T	T	T
Heating Value Btu/Scf	198	385	518	612	827	1070	1145	1315	1479	1526	1618	2323
Spec. Gravity	1.319	0.521	0.582	0.784	0.692	0.589	0.674	1.037	0.867	0.879	0.937	1.378
Wobbe Number	172	533	679	691	994	1394	1395	1291	1588	1628	1672	1979
State	Alaska	Texas	Texas	Kansas	Colorado	Louisiana	Colorado	Colorado	Texas	Texas	Oklahoma	Oklahoma
County	Kemai Pen Borough	Potter	Potter	Kiowa	Mesa	Plaque- mines Parish	Morgan	Morgan	Colorado	Colorado	Custer	Lincoln
Field	Not Given	Cliff- side	Cliff- side	Nullin- ville	Wildcat	Bastian Bay SE	Wildcat	Renegade	Third Creek	Name Le	Weather- ford	Fallis

Table 4 TRUCK LOADING @ 300 PSI AND 90°F

Stage Number	Description	Compression		Cooling	
		Power (HP-hr)	Capital Cost (\$1000)	Heat Removed MMBtu/hr	Operating Cost (\$1000/yr)
1	Boosting press. to 50 psi	957	574	--	--
2	Cooling down to 140°F	--	--	1.85	1.38
3	Boosting press. to 150 psi	694	416	--	--
4	Cooling down to 140°F	--	--	1.45	1.38
5	Boosting press. to 300 psi	465	280	--	--
6	Cooling down to 90°F.	--	--	1.38	1.20
TOTAL		2116	1270	4.68	3.96
TOTAL (Compression + Cooling)		Capital Cost (\$1,301,000)		Annual Operating Cost (\$392,300)	

NOTE: Operating cost consists of fuel cost based on 340 days of operation per year

Table 5 TRUCK LOADING @ 300 PSI AND 0°F

<u>Stage Number</u>	<u>Description</u>	<u>Compression</u>		<u>Cooling</u>		
		Power (HP/hr)	Capital Cost (\$1000)	Operating Cost (\$1000/yr)	Heat Removed MMBtu/hr	Capital Cost \$1000
1	Boosting press. to 50 psi	957	574	175.6	--	--
2	Cooling down to 140°F	--	--	--	1.85	1.38
3	Boosting press. to 150 psi	694	416	127.3	--	--
4	Cooling down to 140°F	--	--	--	1.45	1.38
5	Boosting Press. to 300 psi	465	280	85.4	--	--
6	Cooling down to 90°F	--	--	--	1.38	1.20
7	Refrigeration to 0°F	--	--	--	5.45	624
TOTAL		2116	1270	388.3	10.13	2070.6
TOTAL (Compression + Cooling)		Capital Cost (\$3,341,000)		Annual Operating Cost (\$1,061,260)		627.96

NOTE: Operating cost consists of fuel cost based on 340 days of operation per year

Table 6 TRUCK LOADING @ 1000 PSI AND 90°F

<u>Stage Number</u>	<u>Description</u>	<u>Compression</u>			<u>Cooling</u>		
		Power HP-hr	Capital Cost (\$1000)	Operating Cost (\$1000/yr)	Heat Removed MMBtu/hr	Capital Cost (\$1000)	Operating Cost (\$1000/yr)
1	Boosting press. to 50 psi	957	574	175.6	--	--	--
2	Cooling down to 140°F	--	--	--	1.85	10.75	1.38
3	Boosting press. to 150 psi	694	416	127.3	--	--	--
4	Cooling down to 140°F	--	--	--	1.45	10.75	1.38
5	Boosting press. to 500 psi	732	440	134.4	--	--	--
6	Cooling down to 140°F	--	--	--	1.67	9.10	0.92
7	Boosting press. to 1000 psi	465	280	85.4	--	--	--
8	Cooling down to 90°F	--	--	--	2.07	12.20	2.20
TOTAL		2848	1710	522.7	7.04	42.8	5.88
TOTAL (Compression + Cooling) Capital Cost (\$1,753,000), Annual Operating Cost (\$528,600)							

NOTE: Operating cost consists of fuel cost based on 340 days of operation per year

Table 7 PIPELINE @ 2000 PSI AND 140°F

Stage Number	Description	Compression			Cooling		
		Power (HP-hr)	Capital Cost (\$1000)	Operating Cost (\$1000/yr)	Heat Removed MMBtu/hr	Capital Cost \$1000	Operating Cost (\$100/yr)
1	Boosting press. to 50 psi	957	574	175.6	--	--	--
2	Cooling down to 140°F	--	--	--	1.85	10.75	1.38
3	Boosting press. to 150 psi	694	416	127.3	--	--	--
4	Cooling down to 140°F	--	--	--	1.45	10.75	1.38
5	Boosting press. to 500 psi	732	440	134.4	--	--	--
6	Cooling down to 140°F	--	--	--	1.67	9.10	.92
7	Boosting press. to 1000 psi	465	280	85.4	--	--	--
8	Cooling down to 140°F	--	--	--	1.35	10.75	1.38
9	Boosting press. to 2000 psi	380	230	69.8	--	--	--
10	Cooling down to 140°F	--	--	--	2.29	12.2	1.84
TOTAL		3228	1940	592.5	8.61	53.55	6.90

TOTAL (Compression + Cooling) Capital Cost (\$1,994,000) Annual Operating Cost (\$599,400)

NOTE: Operating cost consists of fuel cost based on 340 days of operation per year

TABLE 8 TOTAL CAPITAL AND OPERATING COSTS FOR VARIOUS OPTIONS
(NATURAL GAS STREAMS CONTAINING 5-10% CO₂)

Purification Cost	Options				Total Capital Cost (\$1000)	Total Operating Cost (\$1000/year)
	Type of Purification	Type of Transportation	Pressure psi	Temp. of F		
Included	Chemical	• Truck	300	90	2076.80	532.80
		• Truck	300	0	4116.80	1201.76
		• Truck	1000	90	2528.80	669.10
		• Pipeline	2000	140	2769.80	739.90
	Physical	• Truck	300	90	2287.80	484.14
		• Truck	300	0	4327.80	1153.10
		• Truck	1000	90	2739.80	620.44
		• Pipeline	2000	140	2980.80	691.24
Adjusted	Chemical	• Truck	300	90	2001.00	243.40
		• Truck	300	0	4041.00	912.36
		• Truck	1000	90	2453.00	379.70
		• Pipeline	2000	140	2694.00	450.50
	Physical	• Truck	300	90	2226.00	198.40
		• Truck	300	0	4266.00	867.36
		• Truck	1000	90	2678.00	334.70
		• Pipeline	2000	140	2919.00	405.50
Ignored	Chemical	• Truck	300	90	1326.80	397.80
		• Truck	300	0	3366.80	1066.76
		• Truck	1000	90	1778.80	534.10
		• Pipeline	2000	140	2019.80	604.90
	Physical	• Truck	300	90	1312.80	394.14
		• Truck	300	0	3352.80	1063.10
		• Truck	1000	90	1764.80	530.44
		• Pipeline	2000	140	2005.80	601.24

Table 9 COST OF CO₂ AT THE POINT OF CONSUMPTION
 BASIS: DISTANCE OF 100 MILES BETWEEN PROCESS PLANT AND POINT OF CONSUMPTION

Purification Cost	Options				Cost of CO ₂ Prior to Transport (\$/MSCF)	Cost of Transport (\$/MSCF)	Cost of CO ₂ @ Delivery Point (\$/MSCF)
	Type of Purification	Type of Transportation	Pressure psi	Temp. of			
Included	Chemical	• Truck	300	90	0.30	2.65	2.95
		• Truck	300	0	0.65	0.87	1.52
		• Truck	1000	90	0.37	0.78	1.15
		• Pipeline	2000	140	0.41	1.25	1.66
Adjusted	Physical	• Truck	300	90	0.29	2.65	2.94
		• Truck	300	0	0.64	0.87	1.51
		• Truck	1000	90	0.36	0.78	1.14
		• Pipeline	2000	140	0.40	1.25	1.65
Ignored	Chemical	• Truck	300	90	0.18	2.65	2.83
		• Truck	300	0	0.53	0.87	1.40
		• Truck	1000	90	0.25	0.78	1.03
		• Pipeline	2000	140	0.29	1.25	1.54
Ignored	Physical	• Truck	300	90	0.17	2.65	2.82
		• Truck	300	0	0.52	0.87	1.39
		• Truck	1000	90	0.25	0.78	1.03
		• Pipeline	2000	140	0.28	1.25	1.53
Ignored	Chemical	• Truck	300	90	0.21	2.65	2.86
		• Truck	300	0	0.56	0.87	1.43
		• Truck	1000	90	0.29	0.78	1.07
		• Pipeline	2000	140	0.32	1.25	1.57
Ignored	Physical	• Truck	300	90	0.21	2.65	2.86
		• Truck	300	0	0.56	0.87	1.43
		• Truck	1000	90	0.28	0.78	1.06
		• Pipeline	2000	140	0.32	1.25	1.57

Table 10

MOLE PERCENT CO2 IN FEED XCO2= 0.30									
TEMPERATURE= -81.40 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
512.	2.770	0.219	0.306	0.694	0.848	0.152	0.273	0.727	0.1897
525.	2.699	0.210	0.317	0.683	0.857	0.143	0.291	0.709	0.1984
550.	2.383	0.221	0.360	0.640	0.859	0.141	0.318	0.682	0.2037
575.	2.119	0.236	0.406	0.594	0.860	0.140	0.352	0.648	0.2090
700.	1.888	0.257	0.456	0.544	0.860	0.140	0.396	0.604	0.2156
725.	1.654	0.290	0.520	0.480	0.861	0.139	0.473	0.527	0.2266
750.	1.464	0.337	0.588	0.412	0.861	0.139	0.590	0.410	0.2430
775.	1.327	0.396	0.649	0.351	0.861	0.139	0.758	0.242	0.2664****

MOLE PERCENT CO2 IN FEED XCO2= 0.30									
TEMPERATURE= -55.00 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
500.	5.042	0.267	0.154	0.846	0.774	0.226	0.119	0.881	0.1009
600.	3.651	0.261	0.213	0.782	0.796	0.204	0.166	0.834	0.1299
700.	2.643	0.277	0.306	0.694	0.808	0.192	0.215	0.785	0.1491
800.	1.858	0.335	0.437	0.563	0.811	0.189	0.297	0.703	0.1676
900.	1.294	0.520	0.620	0.380	0.803	0.197	0.563	0.437	0.2137****

Table 11

MOLE PERCENT CO2 IN FEED XCO2= 0.35									
TEMPERATURE= -81.40 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
612.	2.770	0.219	0.306	0.694	0.848	0.152	0.366	0.634	0.2537
625.	2.699	0.210	0.317	0.683	0.857	0.143	0.383	0.617	0.2617
650.	2.383	0.221	0.360	0.640	0.859	0.141	0.419	0.581	0.2679
675.	2.119	0.236	0.406	0.594	0.860	0.140	0.462	0.538	0.2744
700.	1.888	0.257	0.456	0.544	0.860	0.140	0.520	0.480	0.2829
725.	1.654	0.290	0.520	0.480	0.861	0.139	0.620	0.380	0.2971
750.	1.464	0.337	0.588	0.412	0.861	0.139	0.773	0.227	0.3185
775.	1.327	0.396	0.649	0.351	0.861	0.139	0.994	0.006	0.3492****

MOLE PERCENT CO2 IN FEED XCO2= 0.35									
TEMPERATURE= -55.00 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
500.	5.042	0.267	0.154	0.846	0.774	0.226	0.200	0.800	0.1691
600.	3.651	0.261	0.213	0.782	0.796	0.204	0.253	0.747	0.1975
700.	2.643	0.277	0.306	0.694	0.808	0.192	0.314	0.686	0.2182
800.	1.858	0.335	0.437	0.563	0.811	0.189	0.431	0.569	0.2427
900.	1.294	0.520	0.620	0.380	0.803	0.197	0.837	0.163	0.3178****

Table 12

MOLE PERCENT CO2 IN FEED		XC02= 0.40							
TEMPERATURE= -81.40 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
612.	2.770	0.219	0.306	0.694	0.848	0.152	0.458	0.542	0.3177
625.	2.699	0.210	0.317	0.683	0.857	0.143	0.476	0.524	0.3250
650.	2.383	0.221	0.360	0.640	0.859	0.141	0.519	0.481	0.3320
675.	2.119	0.236	0.406	0.594	0.850	0.140	0.572	0.428	0.3399
700.	1.888	0.257	0.456	0.544	0.860	0.140	0.643	0.357	0.3501
725.	1.654	0.290	0.520	0.480	0.861	0.139	0.766	0.234	0.3675
750.	1.464	0.337	0.588	0.412	0.861	0.139	0.956	0.044	0.3939***

MOLE PERCENT CO2 IN FEED		XC02= 0.40							
TEMPERATURE= -65.00 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
500.	5.042	0.267	0.154	0.846	0.774	0.226	0.280	0.720	0.2374
600.	3.651	0.261	0.218	0.782	0.796	0.204	0.339	0.661	0.2652
700.	2.643	0.277	0.306	0.694	0.808	0.192	0.414	0.586	0.2873
800.	1.858	0.335	0.437	0.563	0.811	0.189	0.564	0.436	0.3179***

Table 13

MOLE PERCENT CO2 IN FEED		XC02= 0.45							
TEMPERATURE= -81.40 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
612.	2.770	0.219	0.306	0.694	0.848	0.152	0.550	0.450	0.3817
625.	2.699	0.210	0.317	0.683	0.857	0.143	0.569	0.431	0.3883
650.	2.383	0.221	0.360	0.640	0.859	0.141	0.619	0.381	0.3962
675.	2.119	0.236	0.406	0.594	0.860	0.140	0.682	0.318	0.4054
700.	1.888	0.257	0.456	0.544	0.860	0.140	0.767	0.233	0.4174
725.	1.654	0.290	0.520	0.480	0.861	0.139	0.913	0.087	0.4379***

MOLE PERCENT CO2 IN FEED		XC02= 0.45							
TEMPERATURE= -65.00 F									
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID
500.	5.042	0.267	0.154	0.846	0.774	0.226	0.361	0.639	0.3056
600.	3.651	0.261	0.218	0.782	0.796	0.204	0.426	0.574	0.3328
700.	2.643	0.277	0.306	0.694	0.808	0.192	0.513	0.487	0.3565
800.	1.858	0.335	0.437	0.563	0.811	0.189	0.598	0.302	0.3930***

Table 14

MOLE PERCENT CO2 IN FEED		XCO2= 0.50								
TEMPERATURE= -31.40 F										
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2	
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID	
612.	2.770	0.219	0.306	0.694	0.848	0.152	0.642	0.358	0.4457	
625.	2.699	0.210	0.317	0.683	0.857	0.143	0.662	0.338	0.4515	
650.	2.383	0.221	0.360	0.640	0.859	0.141	0.720	0.280	0.4604	
675.	2.119	0.236	0.406	0.594	0.860	0.140	0.792	0.208	0.4708	
700.	1.888	0.257	0.456	0.544	0.860	0.140	0.890	0.110	0.4847****	

MOLE PERCENT CO2 IN FEED		XCO2= 0.50								
TEMPERATURE= -65.00 F										
PRESS.	K	K	X	X	Y	Y	MOLES	MOLES	MOLES CO2	
PSIA	CH4	CO2	CH4	CO2	CH4	CO2	LIQUID	GAS	IN LIQUID	
500.	5.042	0.267	0.154	0.846	0.774	0.226	0.442	0.558	0.3738	
600.	3.651	0.261	0.218	0.782	0.796	0.204	0.512	0.488	0.4005	
700.	2.643	0.277	0.306	0.694	0.808	0.192	0.613	0.387	0.4256	
800.	1.858	0.335	0.437	0.563	0.811	0.189	0.831	0.169	0.4682****	

Table 15 Comparison of Vapor-Liquid Equilibrium
Ratio For Mehtane (KCH₄) From Various Sources

P (PSIA)	T = -50°F			T = 0°F			T = +50°F								
	(1)	(2)	(3)	(1)	(2)	(3)	(1)	(2)	(3)						
			CP=1000 CP=2000 CP=5000			CP=1000 CP=2000 CP=3000			CP=1000 CP=2000 CP=5000						
500	NA	5.97	2.25	3.3	4.1	NA	5.93	2.6	4.0	5.0	NA	NA	2.0	4.7	5.7
750	2.70	3.26	1.5	2.3	3.0	3.43	3.75	1.7	2.8	3.6	NA	NA	1.8	3.3	4.1
1000	1.25	1.39	NA	1.85	2.45	2.15	2.37	NA	2.15	3.0	NA	NA	NA	2.45	3.3
1500	NA	0.86	NA	1.36	1.9	NA	1.47	NA	1.45	2.25	NA	NA	NA	1.65	2.5

(1) Experimental Data (50% CH₄ - 50% CO₂)

(2) Calculated Using NCPA Program (40% CH₄ - 60% CO₂)

(3) CP = Convergence Pressure

NA Not Available

Table 16

COMPOSITION:		100. %CH4	0. %CO2				
		CRITICAL TEMPERATURE =		344. RANKINE			
		CRITICAL PRESSURE =		673. PSIA			
		900.	1000.	1100.	1200.	1300.	1400.
		PSIA	PSIA	PSIA	PSIA	PSIA	PSIA
400. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.16	1.16	1.16	1.16	1.16	1.16
	(H*-H)/TC	2.80	3.25	3.65	3.95	4.30	4.70
	CP*	7.490	7.490	7.490	7.490	7.490	7.490
	CP	5.082	4.695	4.351	4.093	3.792	3.448
	1/M	2.55	2.36	2.19	2.06	1.91	1.73
420. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.22	1.22	1.22	1.22	1.22	1.22
	(H*-H)/TC	2.20	2.75	3.10	3.40	3.75	4.05
	CP*	7.629	7.629	7.629	7.629	7.629	7.629
	CP	5.827	5.377	5.090	4.844	4.558	4.312
	1/M	2.93	2.70	2.56	2.43	2.29	2.17
440. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.28	1.28	1.28	1.28	1.28	1.28
	(H*-H)/TC	2.05	2.30	2.55	2.80	3.05	3.30
	CP*	7.770	7.770	7.770	7.770	7.770	7.770
	CP	6.167	5.972	5.776	5.581	5.385	5.190
	1/M	3.10	3.00	2.90	2.80	2.71	2.61
460. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.34	1.34	1.34	1.34	1.34	1.34
	(H*-H)/TC	1.75	2.00	2.25	2.40	2.65	2.95
	CP*	7.911	7.911	7.911	7.911	7.911	7.911
	CP	6.602	6.415	6.228	6.116	5.929	5.705
	1/M	3.32	3.22	3.13	3.07	2.98	2.87
480. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.40	1.40	1.40	1.40	1.40	1.40
	(H*-H)/TC	1.55	1.75	1.95	2.10	2.30	2.50
	CP*	8.054	8.054	8.054	8.054	8.054	8.054
	CP	6.943	6.800	6.656	6.549	6.406	6.262
	1/M	3.49	3.42	3.34	3.29	3.22	3.15
500. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.45	1.45	1.45	1.45	1.45	1.45
	(H*-H)/TC	1.35	1.55	1.75	1.85	2.10	2.25
	CP*	8.198	8.198	8.198	8.198	8.198	8.198
	CP	7.269	7.132	6.994	6.925	6.753	6.650
	1/M	3.65	3.58	3.51	3.48	3.39	3.34

Table 16 Continued

COMPOSITION: 100. %CH4 0. %CO2
 CRITICAL TEMPERATURE = 344. RANKINE
 CRITICAL PRESSURE = 673. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
520. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.51	1.51	1.51	1.51	1.51	1.51
	(H*-H)/TC	1.25	1.40	1.55	1.65	1.85	2.00
	CP*	8.554	8.554	8.554	8.554	8.554	8.554
	CP	7.727	7.628	7.529	7.462	7.330	7.231
	1/M	3.88	3.83	3.78	3.75	3.68	3.63
540. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.57	1.57	1.57	1.57	1.57	1.57
	(H*-H)/TC	1.15	1.30	1.40	1.50	1.60	1.85
	CP*	8.639	8.639	8.639	8.639	8.639	8.639
	CP	7.906	7.811	7.747	7.683	7.620	7.460
	1/M	3.97	3.93	3.89	3.86	3.83	3.75
560. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.63	1.63	1.63	1.63	1.63	1.63
	(H*-H)/TC	1.05	1.15	1.25	1.35	1.50	1.65
	CP*	8.722	8.722	8.722	8.722	8.722	8.722
	CP	8.077	8.016	7.954	7.893	7.801	7.708
	1/M	4.06	4.03	4.00	3.97	3.92	3.87
580. R	PR	1.34	1.49	1.63	1.78	1.93	2.08
	TR	1.69	1.69	1.69	1.69	1.69	1.69
	(H*-H)/TC	0.95	1.00	1.10	1.20	1.35	1.50
	CP*	8.817	8.817	8.817	8.817	8.817	8.817
	CP	8.254	8.224	8.165	8.105	8.016	7.927
	1/M	4.15	4.13	4.10	4.07	4.03	3.98

Table 17

COMPOSITION: 90. %CH₄ 10. %CO₂
 CRITICAL TEMPERATURE = 362. RANKINE
 CRITICAL PRESSURE = 760. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
400. R	FR	1.18	1.32	1.45	1.58	1.71	1.84
	TR	1.10	1.10	1.10	1.10	1.10	1.10
	(H*-H)/TC	3.15	3.55	4.00	4.45	4.85	5.30
	CP*	7.524	7.524	7.524	7.524	7.524	7.524
	CP	4.674	4.312	3.904	3.497	3.135	2.728
	1/M	2.35	2.17	1.96	1.76	1.58	1.37
	420. R	FR	1.18	1.32	1.45	1.58	1.71
TR		1.16	1.16	1.16	1.16	1.16	1.16
(H*-H)/TC		2.50	2.80	3.20	3.50	3.80	4.20
CP*		7.666	7.666	7.666	7.666	7.666	7.666
CP		5.511	5.252	4.908	4.649	4.390	4.046
1/M		2.77	2.64	2.47	2.34	2.21	2.03
440. R		FR	1.18	1.32	1.45	1.58	1.71
	TR	1.22	1.22	1.22	1.22	1.22	1.22
	(H*-H)/TC	2.20	2.40	2.65	2.90	3.20	3.50
	CP*	7.808	7.808	7.808	7.808	7.808	7.808
	CP	5.998	5.834	5.628	5.422	5.176	4.929
	1/M	3.01	2.93	2.83	2.72	2.60	2.48
	460. R	FR	1.18	1.32	1.45	1.58	1.71
TR		1.27	1.27	1.27	1.27	1.27	1.27
(H*-H)/TC		1.85	2.10	2.30	2.50	2.70	3.00
CP*		7.950	7.950	7.950	7.950	7.950	7.950
CP		6.494	6.298	6.140	5.983	5.825	5.589
1/M		3.26	3.16	3.09	3.01	2.93	2.81
480. R		FR	1.18	1.32	1.45	1.58	1.71
	TR	1.33	1.33	1.33	1.33	1.33	1.33
	(H*-H)/TC	1.65	1.80	2.00	2.20	2.45	2.70
	CP*	8.093	8.093	8.093	8.093	8.093	8.093
	CP	6.849	6.736	6.585	6.434	6.246	6.057
	1/M	3.44	3.38	3.31	3.23	3.14	3.04
	500. R	FR	1.18	1.32	1.45	1.58	1.71
TR		1.38	1.38	1.38	1.38	1.38	1.38
(H*-H)/TC		1.45	1.65	1.80	1.95	2.15	2.30
CP*		8.237	8.237	8.237	8.237	8.237	8.237
CP		7.187	7.042	6.934	6.825	6.680	6.572
1/M		3.61	3.54	3.48	3.43	3.36	3.30

Table 17 Continued

COMPOSITION:		90. %CH ₄		10. %CO ₂			
		CRITICAL TEMPERATURE =				362. RANKINE	
		CRITICAL PRESSURE =				760. PSIA	
		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
520. R	PR	1.18	1.32	1.45	1.58	1.71	1.84
	TR	1.44	1.44	1.44	1.44	1.44	1.44
	(H*-H)/TC	1.50	1.60	1.75	1.90	2.10	2.30
	CP*	8.571	8.571	8.571	8.571	8.571	8.571
	CP	7.527	7.457	7.353	7.248	7.109	6.970
	1/M	3.78	3.75	3.69	3.64	3.57	3.50
	540. R	PR	1.18	1.32	1.45	1.58	1.71
TR		1.49	1.49	1.49	1.49	1.49	1.49
(H*-H)/TC		1.40	1.50	1.60	1.80	1.95	2.05
CP*		8.661	8.661	8.661	8.661	8.661	8.661
CP		7.722	7.655	7.588	7.454	7.353	7.286
1/M		3.88	3.85	3.81	3.75	3.70	3.66
560. R		PR	1.18	1.32	1.45	1.58	1.71
	TR	1.55	1.55	1.55	1.55	1.55	1.55
	(H*-H)/TC	1.30	1.40	1.50	1.65	1.80	1.90
	CP*	8.748	8.748	8.748	8.748	8.748	8.748
	CP	7.908	7.843	7.778	7.681	7.584	7.520
	1/M	3.97	3.94	3.91	3.86	3.81	3.78
	580. R	PR	1.18	1.32	1.45	1.58	1.71
TR		1.60	1.60	1.60	1.60	1.60	1.60
(H*-H)/TC		1.20	1.30	1.40	1.55	1.65	1.80
CP*		8.846	8.846	8.846	8.846	8.846	8.846
CP		8.097	8.034	7.972	7.878	7.816	7.722
1/M		4.07	4.04	4.01	3.96	3.93	3.88

Table 18

COMPOSITION: 80. %CH4 20. %CO2
 CRITICAL TEMPERATURE = 381. RANKINE
 CRITICAL PRESSURE = 890. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
400. R	PR	1.01	1.12	1.24	1.35	1.46	1.57
	TR	1.05	1.05	1.05	1.05	1.05	1.05
	(H*-H)/TC	3.15	3.75	4.50	5.25	5.95	6.55
	CP*	7.559	7.559	7.559	7.559	7.559	7.559
	CP	4.558	3.987	3.273	2.558	1.891	1.320
	1/M	2.29	2.00	1.64	1.29	0.95	0.66
	420. R	PR	1.01	1.12	1.24	1.35	1.46
TR		1.10	1.10	1.10	1.10	1.10	1.10
(H*-H)/TC		2.50	2.90	3.30	3.60	3.95	4.40
CP*		7.702	7.702	7.702	7.702	7.702	7.702
CP		5.435	5.072	4.709	4.437	4.119	3.711
1/M		2.73	2.55	2.37	2.23	2.07	1.86
440. R		PR	1.01	1.12	1.24	1.35	1.46
	TR	1.15	1.15	1.15	1.15	1.15	1.15
	(H*-H)/TC	2.10	2.35	2.65	2.95	3.20	3.65
	CP*	7.847	7.847	7.847	7.847	7.847	7.847
	CP	6.028	5.812	5.552	5.292	5.076	4.686
	1/M	3.03	2.92	2.79	2.66	2.55	2.35
	460. R	PR	1.01	1.12	1.24	1.35	1.46
TR		1.21	1.21	1.21	1.21	1.21	1.21
(H*-H)/TC		1.70	1.95	2.25	2.55	2.70	3.00
CP*		7.989	7.989	7.989	7.989	7.989	7.989
CP		6.581	6.374	6.126	5.877	5.753	5.505
1/M		3.31	3.20	3.08	2.95	2.89	2.77
480. R		PR	1.01	1.12	1.24	1.35	1.46
	TR	1.26	1.26	1.26	1.26	1.26	1.26
	(H*-H)/TC	1.55	1.80	2.00	2.25	2.45	2.65
	CP*	8.133	8.133	8.133	8.133	8.133	8.133
	CP	6.902	6.704	6.545	6.347	6.188	6.029
	1/M	3.47	3.37	3.29	3.19	3.11	3.03
	500. R	PR	1.01	1.12	1.24	1.35	1.46
TR		1.31	1.31	1.31	1.31	1.31	1.31
(H*-H)/TC		1.35	1.50	1.75	1.90	2.15	2.25
CP*		8.276	8.276	8.276	8.276	8.276	8.276
CP		7.247	7.133	6.943	6.828	6.638	6.562
1/M		3.64	3.58	3.49	3.43	3.34	3.30

Table 18 Continued

		COMPOSITION: 80. %CH ₄ 20. %CO ₂					
		CRITICAL TEMPERATURE = 381. RANKINE					
		CRITICAL PRESSURE = 890. PSIA					
		900.	1000.	1100.	1200.	1300.	1400.
		PSIA	PSIA	PSIA	PSIA	PSIA	PSIA
520. R	PR	1.01	1.12	1.24	1.35	1.46	1.57
	TR	1.36	1.36	1.36	1.36	1.36	1.36
	(H*-H)/TC	1.45	1.65	1.80	1.90	2.10	2.30
	CP*	8.588	8.588	8.588	8.588	8.588	8.588
	CP	7.526	7.379	7.269	7.196	7.049	6.903
	1/M	3.78	3.71	3.65	3.62	3.54	3.47
540. R	PR	1.01	1.12	1.24	1.35	1.46	1.57
	TR	1.42	1.42	1.42	1.42	1.42	1.42
	(H*-H)/TC	1.30	1.45	1.60	1.70	1.85	2.00
	CP*	8.682	8.682	8.682	8.682	8.682	8.682
	CP	7.765	7.659	7.553	7.483	7.377	7.271
	1/M	3.90	3.85	3.80	3.76	3.71	3.65
560. R	PR	1.01	1.12	1.24	1.35	1.46	1.57
	TR	1.47	1.47	1.47	1.47	1.47	1.47
	(H*-H)/TC	1.20	1.35	1.50	1.60	1.70	1.85
	CP*	8.774	8.774	8.774	8.774	8.774	8.774
	CP	7.958	7.856	7.753	7.685	7.617	7.515
	1/M	4.00	3.95	3.90	3.86	3.83	3.78
580. R	PR	1.01	1.12	1.24	1.35	1.46	1.57
	TR	1.52	1.52	1.52	1.52	1.52	1.52
	(H*-H)/TC	1.10	1.25	1.40	1.50	1.60	1.75
	CP*	8.875	8.875	8.875	8.875	8.875	8.875
	CP	8.152	8.053	7.955	7.889	7.824	7.725
	1/M	4.10	4.05	4.00	3.96	3.93	3.88

Table 19

COMPOSITION: 63. %CH₄ 37. %CO₂
 CRITICAL TEMPERATURE = 414. RANKINE
 CRITICAL PRESSURE = 1039. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
400. R	PR	0.87	0.96	1.06	1.15	1.25	1.35
	TR	0.97	0.97	0.97	0.97	0.97	0.97
	(H*-H)/TC	8.20	8.25	8.35	8.50	8.60	8.75
	CP*	20.463	20.463	20.463	20.463	20.463	20.463
	CP	11.976	11.925	11.821	11.666	11.562	11.407
	1/M	6.02*	5.99*	5.94*	5.86*	5.81*	5.73*
	420. R	PR	0.87	0.96	1.06	1.15	1.25
TR		1.01	1.01	1.01	1.01	1.01	1.01
(H*-H)/TC		2.85	3.70	6.00	7.00	7.45	8.70
CP*		7.765	7.765	21.191	21.191	21.191	21.191
CP		4.956	4.118	15.277	14.291	13.848	12.615
1/M		2.49	2.07	7.68*	7.18*	6.96*	6.34*
440. R		PR	0.87	0.96	1.06	1.15	1.25
	TR	1.06	1.06	1.06	1.06	1.06	1.06
	(H*-H)/TC	2.30	2.75	3.20	3.60	4.20	4.75
	CP*	7.912	7.912	7.912	7.912	7.912	7.912
	CP	5.748	5.324	4.901	4.524	3.960	3.442
	1/M	2.89	2.68	2.46	2.27	1.99	1.73
	460. R	PR	0.87	0.96	1.06	1.15	1.25
TR		1.11	1.11	1.11	1.11	1.11	1.11
(H*-H)/TC		2.00	2.30	2.60	2.90	3.15	3.50
CP*		8.056	8.056	8.056	8.056	8.056	8.056
CP		6.256	5.986	5.716	5.446	5.221	4.906
1/M		3.14	3.01	2.87	2.74	2.62	2.47
480. R		PR	0.87	0.96	1.06	1.15	1.25
	TR	1.16	1.16	1.16	1.16	1.16	1.16
	(H*-H)/TC	1.65	1.85	2.20	2.35	2.60	2.85
	CP*	8.200	8.200	8.200	8.200	8.200	8.200
	CP	6.777	6.604	6.302	6.173	5.957	5.742
	1/M	3.41	3.32	3.17	3.10	2.99	2.89
	500. R	PR	0.87	0.96	1.06	1.15	1.25
TR		1.21	1.21	1.21	1.21	1.21	1.21
(H*-H)/TC		1.40	1.60	1.80	2.00	2.25	2.45
CP*		8.343	8.343	8.343	8.343	8.343	8.343
CP		7.183	7.018	6.852	6.687	6.480	6.314
1/M		3.61	3.53	3.44	3.36	3.26	3.17

Table 19 Continued

COMPOSITION: 63. %CH4 37. %CO2
 CRITICAL TEMPERATURE = 414. RANKINE
 CRITICAL PRESSURE = 1039. PSIA

		900.	1000.	1100.	1200.	1300.	1400.
		PSIA	PSIA	PSIA	PSIA	PSIA	PSIA
520. R	PR	0.87	0.96	1.06	1.15	1.25	1.35
	TR	1.26	1.26	1.26	1.26	1.26	1.26
	(H*-H)/TC	1.50	1.65	1.85	2.05	2.20	2.40
	CP*	8.617	8.617	8.617	8.617	8.617	8.617
	CP	7.423	7.303	7.144	6.985	6.865	6.706
	1/M	3.73	3.67	3.59	3.51	3.45	3.37
540. R	PR	0.87	0.96	1.06	1.15	1.25	1.35
	TR	1.30	1.30	1.30	1.30	1.30	1.30
	(H*-H)/TC	1.40	1.50	1.75	1.90	2.00	2.20
	CP*	8.719	8.719	8.719	8.719	8.719	8.719
	CP	7.646	7.569	7.377	7.262	7.186	7.032
	1/M	3.84	3.80	3.71	3.65	3.61	3.53
560. R	PR	0.87	0.96	1.06	1.15	1.25	1.35
	TR	1.35	1.35	1.35	1.35	1.35	1.35
	(H*-H)/TC	1.30	1.40	1.60	1.70	1.80	1.95
	CP*	8.818	8.818	8.818	8.818	8.818	8.818
	CP	7.857	7.783	7.635	7.561	7.487	7.377
	1/M	3.95	3.91	3.84	3.80	3.76	3.71
580. R	PR	0.87	0.96	1.06	1.15	1.25	1.35
	TR	1.40	1.40	1.40	1.40	1.40	1.40
	(H*-H)/TC	1.15	1.30	1.45	1.55	1.60	1.80
	CP*	8.924	8.924	8.924	8.924	8.924	8.924
	CP	8.103	7.996	7.889	7.817	7.781	7.639
	1/M	4.07	4.02	3.96	3.93	3.91	3.84

Table 20

COMPOSITION: 45. %CH4 55. %CO2
 CRITICAL TEMPERATURE = 450. RANKINE
 CRITICAL PRESSURE = 1184. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
400. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	0.89	0.89	0.89	0.89	0.89	0.89
	(H*-H)/TC	10.00	10.00	10.05	10.10	10.15	10.20
	CP*	20.202	20.202	20.202	20.202	20.202	20.202
	CP	8.952	8.952	8.896	8.840	8.784	8.727
	1/M	4.50*	4.50*	4.47*	4.44*	4.41*	4.39*
420. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	0.93	0.93	0.93	0.93	0.93	0.93
	(H*-H)/TC	9.30	9.40	9.40	9.45	9.45	9.50
	CP*	20.788	20.788	20.788	20.788	20.788	20.788
	CP	10.824	10.717	10.717	10.663	10.663	10.609
	1/M	5.44*	5.39*	5.39*	5.36*	5.36*	5.33*
440. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	0.98	0.98	0.98	0.98	0.98	0.98
	(H*-H)/TC	2.50	3.35	7.85	8.00	8.15	8.30
	CP*	7.981	7.981	21.406	21.406	21.406	21.406
	CP	5.424	4.555	13.378	13.225	13.071	12.918
	1/M	2.73	2.29	6.72*	6.65*	6.57*	6.49*
460. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	1.02	1.02	1.02	1.02	1.02	1.02
	(H*-H)/TC	2.15	2.55	3.15	4.25	5.50	6.25
	CP*	8.127	8.127	8.127	8.127	8.127	22.359
	CP	6.023	5.632	5.045	3.969	2.746	16.245
	1/M	3.03	2.83	2.54	1.99	1.38	8.16*
480. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	1.07	1.07	1.07	1.07	1.07	1.07
	(H*-H)/TC	1.70	2.10	2.40	2.80	3.25	3.65
	CP*	8.271	8.271	8.271	8.271	8.271	8.271
	CP	6.677	6.302	6.021	5.646	5.224	4.849
	1/M	3.36	3.17	3.03	2.84	2.63	2.44
500. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	1.11	1.11	1.11	1.11	1.11	1.11
	(H*-H)/TC	1.60	1.80	2.05	2.40	2.65	3.00
	CP*	8.413	8.413	8.413	8.413	8.413	8.413
	CP	6.973	6.793	6.568	6.253	6.028	5.713
	1/M	3.50	3.41	3.30	3.14	3.03	2.87

Table 20 Continued

COMPOSITION:		45. %CH ₄		55. %CO ₂			
				CRITICAL TEMPERATURE = 450. RANKINE			
				CRITICAL PRESSURE = 1184. PSIA			
		900.	1000.	1100.	1200.	1300.	1400.
		PSIA	PSIA	PSIA	PSIA	PSIA	PSIA
520. R	PR	0.76	0.84	0.93	1.01	1.10	1.18
	TR	1.16	1.16	1.16	1.16	1.16	1.16
	(H*-H)/TC	1.35	1.50	1.75	2.00	2.25	2.40
	CP*	8.647	8.647	8.647	8.647	8.647	8.647
	CP	7.479	7.349	7.133	6.917	6.700	6.571
	1/M	3.76	3.69	3.58	3.48	3.37	3.30
	540. R	PR	0.76	0.84	0.93	1.01	1.10
TR		1.20	1.20	1.20	1.20	1.20	1.20
(H*-H)/TC		1.25	1.35	1.55	1.75	2.00	2.15
CP*		8.758	8.758	8.758	8.758	8.758	8.758
CP		7.716	7.633	7.466	7.299	7.091	6.966
1/M		3.88	3.84	3.75	3.67	3.56	3.50
560. R		PR	0.76	0.84	0.93	1.01	1.10
	TR	1.24	1.24	1.24	1.24	1.24	1.24
	(H*-H)/TC	1.15	1.25	1.45	1.60	1.80	1.95
	CP*	8.865	8.865	8.865	8.865	8.865	8.865
	CP	7.941	7.861	7.700	7.579	7.419	7.298
	1/M	3.99	3.95	3.87	3.81	3.73	3.67
	580. R	PR	0.76	0.84	0.93	1.01	1.10
TR		1.29	1.29	1.29	1.29	1.29	1.29
(H*-H)/TC		1.10	1.15	1.25	1.40	1.55	1.75
CP*		8.975	8.975	8.975	8.975	8.975	8.975
CP		8.122	8.083	8.006	7.889	7.773	7.618
1/M		4.08	4.06	4.02	3.96	3.91	3.83

Table 21

COMPOSITION: 28. %CH4 72. %CO2
 CRITICAL TEMPERATURE = 486. RANKINE
 CRITICAL PRESSURE = 1245. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
400. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	0.82	0.82	0.82	0.82	0.82	0.82
	(H*-H)/TC	11.20	11.25	11.25	11.30	11.30	11.30
	CP*	19.956	19.956	19.956	19.956	19.956	19.956
	CP	6.348	6.287	6.287	6.226	6.226	6.226
	1/M	3.19*	3.16*	3.16*	3.13*	3.13*	3.13*
420. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	0.86	0.86	0.86	0.86	0.86	0.86
	(H*-H)/TC	10.65	10.70	10.70	10.70	10.75	10.75
	CP*	20.407	20.407	20.407	20.407	20.407	20.407
	CP	8.084	8.026	8.026	8.026	7.968	7.968
	1/M	4.06*	4.03*	4.03*	4.03*	4.00*	4.00*
440. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	0.91	0.91	0.91	0.91	0.91	0.91
	(H*-H)/TC	9.85	9.85	9.90	9.95	9.95	10.00
	CP*	20.902	20.902	20.902	20.902	20.902	20.902
	CP	10.022	10.022	9.967	9.911	9.911	9.856
	1/M	5.04*	5.04*	5.01*	4.98*	4.98*	4.95*
460. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	0.95	0.95	0.95	0.95	0.95	0.95
	(H*-H)/TC	8.75	8.80	8.85	8.85	8.90	9.00
	CP*	21.846	21.846	21.846	21.846	21.846	21.846
	CP	12.601	12.548	12.495	12.495	12.443	12.337
	1/M	6.33*	6.31*	6.28*	6.28*	6.25*	6.20*
480. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	0.99	0.99	0.99	0.99	0.99	0.99
	(H*-H)/TC	2.15	2.60	3.50	7.25	7.55	7.75
	CP*	8.338	8.338	8.338	23.638	23.638	23.638
	CP	6.161	5.705	4.794	16.297	15.994	15.791
	1/M	3.10	2.87	2.41	8.19*	8.04*	7.94*
500. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	1.03	1.03	1.03	1.03	1.03	1.03
	(H*-H)/TC	1.85	2.25	2.65	3.20	4.00	4.65
	CP*	8.480	8.480	8.480	8.480	8.480	8.480
	CP	6.681	6.293	5.904	5.369	4.592	3.960
	1/M	3.36	3.16	2.97	2.70	2.31	1.99

Table 21 Continued

COMPOSITION: 28. %CH₄ 72. %CO₂
 CRITICAL TEMPERATURE = 486. RANKINE
 CRITICAL PRESSURE = 1245. PSIA

		900. PSIA	1000. PSIA	1100. PSIA	1200. PSIA	1300. PSIA	1400. PSIA
520. R	PR	0.72	0.80	0.88	0.96	1.04	1.12
	TR	1.07	1.07	1.07	1.07	1.07	1.07
	(H*-H)/TC	1.80	2.10	2.40	2.80	3.30	3.80
	CP*	8.676	8.676	8.676	8.676	8.676	8.676
	CP	6.994	6.714	6.433	6.059	5.592	5.125
	1/M	3.51	3.37	3.23	3.04	2.81	2.58
	540. R	PR	0.72	0.80	0.88	0.96	1.04
TR		1.11	1.11	1.11	1.11	1.11	1.11
(H*-H)/TC		1.60	1.80	2.00	2.30	2.60	3.00
CP*		8.795	8.795	8.795	8.795	8.795	8.795
CP		7.355	7.175	6.995	6.725	6.455	6.095
1/M		3.70	3.61	3.51	3.38	3.24	3.06
560. R		PR	0.72	0.80	0.88	0.96	1.04
	TR	1.15	1.15	1.15	1.15	1.15	1.15
	(H*-H)/TC	1.50	1.70	1.90	2.10	2.40	2.70
	CP*	8.909	8.909	8.909	8.909	8.909	8.909
	CP	7.607	7.434	7.260	7.087	6.826	6.566
	1/M	3.82	3.74	3.65	3.56	3.43	3.30
	580. R	PR	0.72	0.80	0.88	0.96	1.04
TR		1.19	1.19	1.19	1.19	1.19	1.19
(H*-H)/TC		1.35	1.50	1.70	1.90	2.10	2.40
CP*		9.024	9.024	9.024	9.024	9.024	9.024
CP		7.893	7.767	7.600	7.432	7.265	7.013
1/M		3.97	3.90	3.82	3.73	3.65	3.52

COMPARISON OF CO₂ PIPELINE REPRESSURIZATION AND THERMAL CONDITIONING REQUIREMENTS

Line Capacity (MMSCFD)	Nominal Diameter (in)	Low Compressibility Path			High Compressibility Path		
		Distance Between Compressor Stations (mi)	Initial Temp (F)	Final Temp (F)	Distance Between Compressor Stations (mi)	Initial Temp (F)	Final Temp (F)
50	8	88	80	75	185	77	
125	12	107	80	61	185	101	
250	16	87	80	38	185	147	
500	20	70	80	28	185	167	

NOTE: Compressor stations are located where pressure has declined from 2000 psia to 1400 psia

TABLE 22

COMPARISON OF CO₂

PIPELINE REPRESSURIZATION POWER AND THERMAL CONDITIONING

Line Capacity (MMSCFD)	Nominal Diameter (in)	Low Compressibility Path			High Compressibility Path		
		Distance Between Compressor Stations (mi)	Compressor Power (hp/mi)	Thermal Conditioning (BTU/mi/day)	Distance Between Compressor Stations (mi)	Compressor Power (hp/mi)	Thermal Conditioning (BTU/mi/day)
50	8	88	2.9	0	75	9.1	6.2x10 ⁶
125	12	107	5.9	0	61	28.0	9.6x10 ⁶
250	16	87	14.5	0	38	89.9	0
500	20	70	36.0	0	28	244.1	0

Note: Compressor stations are located where pressure has declined from 2000 psia to 1400 psia

SUMMARY OF TRANSPORTATION COSTS

Line Capacity (MMSCFD)	Nominal Diameter (in)	Investment Cost Per Mile		Annual Operating Cost Per Mile	
		Low Compressibility	High Compressibility	Low Compressibility	High Compressibility
50	8	\$162,000	\$165,000	\$11,700	\$17,800
125	12	243,000	255,000	13,900	26,900
250	16	308,000	349,000	18,200	35,600
500	20	419,000	532,000	26,000	64,100

TABLE 24

SUMMARY OF CO₂ PIPELINE DESIGNS

<u>Trunk Lines</u>	(20 Year Life)					
Flow Rate (MMSCFD)	20	50	125	250	500	
Diameter (inches)	8	10	16	22	30	
Length (miles)	50	100	200	400	500	
 <u>Laterals</u>	 (5 Year Life)					
Flow Rates (MMSCFD)	1	2	5	10	20	50
Diameter (inches)	2	3	4	5	8	10
Length (miles)	5	10	20	40	80	100

Table 25

TRUNK PIPELINE CALCULATIONS
(SUMMARY TABLE)

Flow Rate = 500 MMSCFD, D = 30 inches				
Length (Miles)	Cost	\$15,000/in·mile (\$/MSCF)	Cost	\$20,000/in·mile (\$/MSCF)
50		0.0394		0.0526
100		0.0789		0.1052
200		0.1578		0.2103
400		0.3155		0.4207
500		0.3944		0.5259
Flow Rate = 250 MMSCFD, D = 22 inches				
50		0.0578		0.0771
100		0.1157		0.1543
200		0.2314		0.3085
400		0.4628		0.6170
500		0.5784		0.7713
Flow Rate = 125 MMSCFD, D = 16 inches				
50		0.0841		0.1122
100		0.1683		0.2244
200		0.3365		0.4487
400		0.6731		0.8975
500		0.8414		1.1218
Flow Rate = 50 MMSCFD, D = 10 inches				
50		0.1315		0.1753
100		0.2629		0.3506
200		0.5259		0.7011
400		1.0517		1.4023
500		1.3146		1.7528
Flow Rate = 20 MMSCFD, D = 8 inches				
50		0.2629		0.3506
100		0.5259		0.7011
200		1.0517		1.4023
400		2.1034		2.8046
500		2.6293		3.5057

Table 26

LATERAL PIPELINE CALCULATIONS
(SUMMARY TABLE)

Flow Rate = 50 MMSCFD, D = 10 inches

Length (Miles)	Cost \$15,000/in·mile (\$/MSCF)	Cost \$20,000/in·mile (\$/MSCF)
5	0.0194	0.0259
10	0.0388	0.0517
20	0.0776	0.1035
40	0.1556	0.2070
80	0.3105	0.4140
100	0.3881	0.5175

Flow Rate = 20 MMSCFD, D = 8 inches

5	0.0388	0.0517
10	0.0776	0.1035
20	0.1552	0.2070
40	0.3105	0.4140
80	0.6210	0.8279
100	0.7762	1.0349

Flow Rate = 10 MMSCFD, D = 6 inches

5	0.0582	0.0776
10	0.1164	0.1552
20	0.2329	0.3105
40	0.4657	0.6210
80	0.9314	1.2419
100	1.1643	1.5524

Flow Rate = 5 MMSCFD, D = 4 inches

Length (Miles)	Cost \$10,000/in·mile (\$/MSCF)	Cost \$15,000/in·mile (\$/MSCF)
5	0.0259	0.0388
10	0.0517	0.0776
20	0.1035	0.1552
40	0.2070	0.3105
80	0.4140	0.6210
100	0.5174	0.7762

Table 27
(1)

LATERAL PIPELINE CALCULATIONS
(SUMMARY TABLE - CONT.)

Flow Rate = 2 MMSCFD,		D = 3 inches	
Length (Miles)	Cost \$10,000/in·mile (\$/MSCF)	Cost \$15,000/in·mile (\$/MSCF)	
5	0.0970	0.1455	
10	0.1940	0.2911	
20	0.3881	0.5820	
40	0.7762	1.1643	
80	1.5524	2.3286	
100	1.9404	2.9107	
Flow Rate = 1 MMSCFD,		D = 2 inches	
5	0.0648	0.0970	
10	0.1294	0.1940	
20	0.2587	0.3881	
40	0.5175	0.7762	
80	1.0349	1.5524	
100	1.2936	1.9405	

Table 27
(2)

SUMMARY OF TRACTOR & TRAILER REQUIREMENTS
FOR CO₂ TRANSPORT

Working Pressure = 300 psia

Haul Distances

Daily Requirement - MGCSPD	Haul Distances					
	5	10	20	40	80	100
	Tractor/ Trailers	Tractor/ Trailers	Tractor/ Trailers	Tractor/ Trailers	Tractor/ Trailers	Tractor/ Trailers
0.5	0.265/ 1.338	0.371/ 1.409	0.582/ 1.552	1.006/ 1.837	1.853/ 2.408	2.277/ 2.694
1.0	0.530/ 1.623	0.741/ 1.766	1.165/ 2.051	2.012/ 2.622	3.707/ 3.764	4.554/ 4.334
2.0	1.059 2.194	1.483/ 2.480	2.330/ 3.050	4.024/ 4.192	7.413/ 6.475	9.107/ 7.616
5.0	2.648/ 3.907	3.707/ 4.620	5.825/ 6.047	10.061/ 8.910	18.533/ 14.608	22.769/ 17.462
10.0	5.295/ 6.760	7.413/ 8.187	11.649/ 11.041	20.121/ 16.749	37.065/ 28.164	45.537/ 33.871

Table 28

Tractor Unit Investment: \$50,000

90% of original investment depreciated straight-line over
7 years to zero salvage

7-year evaluation life

15% discount rate

50% effective tax rate

10% investment tax credit

$$\text{Annual investment cost: } 50,000(0.9)(A/P_{15,7}) = 0.24036$$

\$10,800/year
(not subject to tax)

Tax deductible costs:

Fixed cost: \$6,000/year

Driver labor

(3 shifts, 4 drivers): $4 \times 30,000 = \$120,000/\text{year}$

Operation: $(\$0.35/\text{mi})(107 \text{ hr}/\text{wk})(52 \text{ wk}/\text{year})$

$$\left(\frac{D/15}{D/15 + 0.5} \right) \left(\frac{340}{365} \right) (30 \text{ mph}) = \$ \left[\frac{3628D}{D/15 + 0.5} \right] / \text{year}$$

$$\text{Total tax-deductible cost: } \$ \left[126,000 + \frac{3628D}{D/15 + 0.5} \right] / \text{year}$$

Depreciation tax savings: $50,000(0.9)(1/7)(0.5) = \$3,200/\text{year}$

$$\text{After-tax tractor unit cost: } -3,200 + \left[\frac{126,000 + \frac{3628D}{D/15 + 0.5}}{+ 10,800/0.5} \right]$$

$$= \$ \left[144,400 + \frac{3628D}{D/15 + 0.5} \right] / \text{year}$$

Table 29

Development of Annual Tractor Unit Cost

Trailer Unit Investment: \$45,000

90% of original investment depreciated straight-line over
7-years to zero salvage

7-year evaluation life

15% discount rate

50% effective tax rate

10% investment tax credit

$$\text{Annual investment cost: } 45,000(0.9) \left(\frac{A/P_{15,7}}{0.24036} \right) = \$9,700/\text{year}$$

Tax deductible costs:

Fixed costs: \$1,000/year

Operation: (\$0.10/mi) (0.95) (24 hr/day) (340 days/year)

$$\begin{aligned} & \times \left(\frac{D/15}{1 + D/15 + 2.459/Q} \right) \left(30 \text{ mph} \right) \\ & = \$ \left[\frac{1550D}{1 + D/15 + 2.459/Q} \right] / \text{year} \end{aligned}$$

$$\text{Total tax-deductible cost: } \$ \left[1,000 + \frac{1550D}{1 + D/15 + 2.459/Q} \right] / \text{year}$$

Depreciation tax savings: $45,000 (0.9) (1/7) (0.5) = \$2,900/\text{year}$

After-tax trailer unit cost:

$$\begin{aligned} & -2,900 + \left[1,000 + \frac{1550D}{1 + D/15 + 2.459/Q} \right] + 9,700/0.5 \\ & = \$ \left[17,600 + \frac{1550D}{1 + D/15 + 2.459/Q} \right] / \text{year} \end{aligned}$$

Table 30
Development of Annual Trailer Unit Cost

Table 31 Summary of Transport costs for Truck-Delivered CO₂ (\$/MCF)
 300 psia, 0°F, saturated vapor

Daily Requirement - MMSCFD	Haul Distance (mi)					
	5	10	20	40	80	100
0.5	0.407	0.548	0.830	1.394	2.522	3.086
1.0	0.353	0.494	0.776	1.340	2.468	3.032
2.0	0.325	0.466	0.748	1.312	2.440	3.004
5.0	0.309	0.450	0.732	1.296	2.424	2.988
10.0	0.304	0.445	0.727	1.291	2.419	2.982

Tractor and Trailer Requirements and Costs for CO₂ Distribution
 By Truck From A Trunk Pipeline

(Based on \$45,000 Trailer Used at 0°F and 300 psi, s.g.= 0.0688)

CO ₂ Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	CO ₂ Cost
0.50	5.00	0.265	\$166,168.00	\$43,993.10	1.338	\$18,839.73	\$25,207.44	\$0.407/Mcf
0.50	10.00	0.371	\$175,497.13	\$65,048.27	1.409	\$19,953.95	\$28,121.85	\$0.548/Mcf
0.50	15.00	0.477	\$180,680.00	\$86,103.47	1.481	\$20,960.80	\$31,036.25	\$0.689/Mcf
0.50	20.00	0.582	\$183,978.16	\$107,158.64	1.552	\$21,875.07	\$33,950.66	\$0.830/Mcf
0.50	25.00	0.688	\$186,261.53	\$128,213.84	1.623	\$22,708.99	\$36,865.06	\$0.971/Mcf
0.50	30.00	0.794	\$187,936.00	\$149,269.03	1.695	\$23,472.70	\$39,779.46	\$1.112/Mcf
0.50	35.00	0.900	\$189,216.47	\$170,324.25	1.766	\$24,174.70	\$42,693.87	\$1.253/Mcf
0.50	40.00	1.006	\$190,227.34	\$191,379.41	1.837	\$24,822.18	\$45,608.27	\$1.394/Mcf
0.50	45.00	1.112	\$191,045.69	\$212,434.59	1.909	\$25,421.26	\$48,522.66	\$1.535/Mcf
0.50	50.00	1.218	\$191,721.72	\$233,489.78	1.980	\$25,977.17	\$51,437.07	\$1.676/Mcf
0.50	55.00	1.324	\$192,289.59	\$254,545.00	2.051	\$26,494.41	\$54,351.46	\$1.817/Mcf
0.50	60.00	1.430	\$192,773.31	\$275,600.13	2.123	\$26,976.89	\$57,265.87	\$1.958/Mcf
0.50	65.00	1.536	\$193,190.34	\$296,655.38	2.194	\$27,427.99	\$60,180.27	\$2.099/Mcf
0.50	70.00	1.641	\$193,553.53	\$317,710.56	2.265	\$27,850.68	\$63,094.66	\$2.240/Mcf
0.50	75.00	1.747	\$193,872.72	\$338,765.75	2.337	\$28,247.55	\$66,009.06	\$2.381/Mcf
0.50	80.00	1.853	\$194,155.41	\$359,820.94	2.408	\$28,620.92	\$68,923.47	\$2.522/Mcf
0.50	85.00	1.959	\$194,407.56	\$380,876.13	2.479	\$28,972.79	\$71,837.86	\$2.663/Mcf
0.50	90.00	2.065	\$194,633.84	\$401,931.31	2.551	\$29,304.98	\$74,752.27	\$2.804/Mcf
0.50	95.00	2.171	\$194,838.03	\$422,986.44	2.622	\$29,619.10	\$77,666.66	\$2.945/Mcf
0.50	100.00	2.277	\$195,023.25	\$444,041.75	2.694	\$29,916.57	\$80,581.05	\$3.086/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
1.00	5.00	0.530	\$166,168.00	\$87,986.20	1.623	\$19,643.59	\$31,889.16	\$0.353/Mcf
1.00	10.00	0.741	\$175,497.13	\$130,096.55	1.766	\$21,356.97	\$37,718.00	\$0.494/Mcf
1.00	15.00	0.953	\$180,680.00	\$172,206.94	1.909	\$22,814.17	\$43,546.83	\$0.635/Mcf
1.00	20.00	1.165	\$183,978.16	\$214,317.28	2.051	\$24,068.66	\$49,375.65	\$0.776/Mcf
1.00	25.00	1.377	\$186,261.53	\$256,427.69	2.194	\$25,159.99	\$55,204.48	\$0.917/Mcf
1.00	30.00	1.589	\$187,936.00	\$298,538.06	2.337	\$26,118.04	\$61,033.29	\$1.058/Mcf
1.00	35.00	1.800	\$189,216.47	\$340,648.50	2.480	\$26,965.83	\$66,862.11	\$1.199/Mcf
1.00	40.00	2.012	\$190,227.34	\$382,758.81	2.622	\$27,721.35	\$72,690.91	\$1.340/Mcf
1.00	45.00	2.224	\$191,045.69	\$424,869.19	2.765	\$28,398.88	\$78,519.72	\$1.481/Mcf
1.00	50.00	2.436	\$191,721.72	\$466,979.56	2.908	\$29,009.92	\$84,348.52	\$1.622/Mcf
1.00	55.00	2.648	\$192,289.59	\$509,090.00	3.050	\$29,563.79	\$90,177.33	\$1.763/Mcf
1.00	60.00	2.859	\$192,773.31	\$551,200.25	3.193	\$30,068.16	\$96,006.13	\$1.904/Mcf
1.00	65.00	3.071	\$193,190.34	\$593,310.75	3.336	\$30,529.38	\$101,834.94	\$2.045/Mcf
1.00	70.00	3.283	\$193,553.53	\$635,421.13	3.478	\$30,952.75	\$107,663.73	\$2.186/Mcf
1.00	75.00	3.495	\$193,872.72	\$677,531.50	3.621	\$31,342.76	\$113,492.53	\$2.327/Mcf
1.00	80.00	3.707	\$194,155.41	\$719,641.88	3.764	\$31,703.20	\$119,321.34	\$2.468/Mcf
1.00	85.00	3.918	\$194,407.56	\$761,752.25	3.906	\$32,037.30	\$125,150.13	\$2.609/Mcf
1.00	90.00	4.130	\$194,633.84	\$803,862.63	4.049	\$32,347.86	\$130,978.94	\$2.750/Mcf
1.00	95.00	4.342	\$194,838.03	\$845,972.88	4.192	\$32,637.27	\$136,807.69	\$2.891/Mcf
1.00	100.00	4.554	\$195,023.25	\$888,083.50	4.334	\$32,907.63	\$142,636.50	\$3.032/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
1.50	5.00	0.794	\$166,168.00	\$131,979.28	1.909	\$20,207.09	\$38,570.90	\$0.334/Mcf
1.50	10.00	1.112	\$175,497.13	\$195,144.84	2.123	\$22,288.45	\$47,314.18	\$0.475/Mcf
1.50	15.00	1.430	\$180,680.00	\$258,310.41	2.337	\$23,988.53	\$56,057.44	\$0.616/Mcf
1.50	20.00	1.747	\$183,978.16	\$321,475.94	2.551	\$25,403.32	\$64,800.70	\$0.757/Mcf
1.50	25.00	2.065	\$186,261.53	\$384,641.50	2.765	\$26,599.07	\$73,543.94	\$0.898/Mcf
1.50	30.00	2.383	\$187,936.00	\$447,807.06	2.979	\$27,622.99	\$82,287.16	\$1.039/Mcf
1.50	35.00	2.700	\$189,216.47	\$510,972.75	3.193	\$28,509.64	\$91,030.41	\$1.180/Mcf
1.50	40.00	3.018	\$190,227.34	\$574,138.25	3.407	\$29,284.89	\$99,773.61	\$1.321/Mcf
1.50	45.00	3.336	\$191,045.69	\$637,303.75	3.621	\$29,968.48	\$108,516.83	\$1.462/Mcf
1.50	50.00	3.654	\$191,721.72	\$700,469.38	3.835	\$30,575.78	\$117,260.03	\$1.603/Mcf
1.50	55.00	3.971	\$192,289.59	\$763,635.00	4.049	\$31,118.87	\$126,003.23	\$1.744/Mcf
1.50	60.00	4.289	\$192,773.31	\$826,800.50	4.263	\$31,607.43	\$134,746.41	\$1.885/Mcf
1.50	65.00	4.607	\$193,190.34	\$889,966.25	4.477	\$32,049.28	\$143,489.63	\$2.026/Mcf
1.50	70.00	4.924	\$193,553.53	\$953,131.75	4.691	\$32,450.81	\$152,232.81	\$2.167/Mcf
1.50	75.00	5.242	\$193,872.72	\$1,016,297.25	4.905	\$32,817.30	\$160,976.03	\$2.308/Mcf
1.50	80.00	5.560	\$194,155.41	\$1,079,462.75	5.119	\$33,153.14	\$169,719.25	\$2.449/Mcf
1.50	85.00	5.877	\$194,407.56	\$1,142,628.50	5.333	\$33,462.03	\$178,462.44	\$2.590/Mcf
1.50	90.00	6.195	\$194,633.84	\$1,205,794.00	5.547	\$33,747.08	\$187,205.63	\$2.731/Mcf
1.50	95.00	6.513	\$194,838.03	\$1,268,959.50	5.761	\$34,010.95	\$195,948.81	\$2.872/Mcf
1.50	100.00	6.831	\$195,023.25	\$1,332,125.25	5.975	\$34,255.92	\$204,692.00	\$3.013/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
2.00	5.00	1.059	\$166,168.00	\$175,972.41	2.194	\$20,624.00	\$45,252.65	\$0.325/Mcf
2.00	10.00	1.483	\$175,497.13	\$260,193.09	2.480	\$22,951.90	\$56,910.38	\$0.466/Mcf
2.00	15.00	1.906	\$180,680.00	\$344,413.88	2.765	\$24,799.25	\$68,568.06	\$0.607/Mcf
2.00	20.00	2.330	\$183,978.16	\$428,634.56	3.050	\$26,300.94	\$80,225.75	\$0.748/Mcf
2.00	25.00	2.753	\$186,261.53	\$512,855.38	3.336	\$27,545.67	\$91,883.41	\$0.889/Mcf
2.00	30.00	3.177	\$187,936.00	\$597,076.13	3.621	\$28,594.21	\$103,541.06	\$1.030/Mcf
2.00	35.00	3.601	\$189,216.47	\$681,297.00	3.906	\$29,489.54	\$115,198.69	\$1.171/Mcf
2.00	40.00	4.024	\$190,227.34	\$765,517.63	4.192	\$30,262.97	\$126,856.31	\$1.312/Mcf
2.00	45.00	4.448	\$191,045.69	\$849,738.38	4.477	\$30,937.79	\$138,513.91	\$1.453/Mcf
2.00	50.00	4.871	\$191,721.72	\$933,959.13	4.763	\$31,531.75	\$150,171.53	\$1.594/Mcf
2.00	55.00	5.295	\$192,289.59	\$1,018,180.00	5.048	\$32,058.55	\$161,829.16	\$1.735/Mcf
2.00	60.00	5.719	\$192,773.31	\$1,102,400.50	5.333	\$32,528.96	\$173,486.72	\$1.876/Mcf
2.00	65.00	6.142	\$193,190.34	\$1,186,621.50	5.619	\$32,951.60	\$185,144.38	\$2.017/Mcf
2.00	70.00	6.566	\$193,553.53	\$1,270,842.25	5.904	\$33,333.38	\$196,801.94	\$2.158/Mcf
2.00	75.00	6.989	\$193,872.72	\$1,355,063.00	6.189	\$33,679.95	\$208,459.53	\$2.299/Mcf
2.00	80.00	7.413	\$194,155.41	\$1,439,283.75	6.475	\$33,995.97	\$220,117.16	\$2.440/Mcf
2.00	85.00	7.837	\$194,407.56	\$1,523,504.50	6.760	\$34,285.31	\$231,774.75	\$2.581/Mcf
2.00	90.00	8.260	\$194,633.84	\$1,607,725.25	7.046	\$34,551.21	\$243,432.34	\$2.722/Mcf
2.00	95.00	8.684	\$194,838.03	\$1,691,945.75	7.331	\$34,796.41	\$255,089.91	\$2.863/Mcf
2.00	100.00	9.107	\$195,023.25	\$1,776,167.00	7.616	\$35,023.23	\$266,747.50	\$3.004/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
2.50	5.00	1.324	\$166,168.00	\$219,965.50	2.480	\$20,944.94	\$51,934.41	\$0.320/Mcf
2.50	10.00	1.853	\$175,497.13	\$325,241.38	2.836	\$23,448.47	\$66,506.59	\$0.461/Mcf
2.50	15.00	2.383	\$180,680.00	\$430,517.31	3.193	\$25,392.60	\$81,078.72	\$0.602/Mcf
2.50	20.00	2.912	\$183,978.16	\$535,793.25	3.550	\$26,945.98	\$95,650.83	\$0.743/Mcf
2.50	25.00	3.442	\$186,261.53	\$641,069.25	3.906	\$28,215.66	\$110,222.91	\$0.884/Mcf
2.50	30.00	3.971	\$187,936.00	\$746,345.13	4.263	\$29,272.86	\$124,794.95	\$1.025/Mcf
2.50	35.00	4.501	\$189,216.47	\$851,621.13	4.620	\$30,166.79	\$139,367.00	\$1.166/Mcf
2.50	40.00	5.030	\$190,227.34	\$956,897.00	4.977	\$30,932.57	\$153,939.03	\$1.307/Mcf
2.50	45.00	5.560	\$191,045.69	\$1,062,173.00	5.333	\$31,595.91	\$168,511.06	\$1.448/Mcf
2.50	50.00	6.089	\$191,721.72	\$1,167,449.00	5.690	\$32,176.07	\$183,083.06	\$1.589/Mcf
2.50	55.00	6.619	\$192,289.59	\$1,272,725.00	6.047	\$32,687.79	\$197,655.09	\$1.730/Mcf
2.50	60.00	7.148	\$192,773.31	\$1,378,000.75	6.403	\$33,142.48	\$212,227.06	\$1.871/Mcf
2.50	65.00	7.678	\$193,190.34	\$1,483,277.00	6.760	\$33,549.20	\$226,799.13	\$2.012/Mcf
2.50	70.00	8.207	\$193,553.53	\$1,588,552.50	7.117	\$33,915.13	\$241,371.06	\$2.153/Mcf
2.50	75.00	8.737	\$193,872.72	\$1,693,828.75	7.474	\$34,246.14	\$255,943.06	\$2.294/Mcf
2.50	80.00	9.266	\$194,155.41	\$1,799,104.75	7.830	\$34,546.98	\$270,515.00	\$2.435/Mcf
2.50	85.00	9.796	\$194,407.56	\$1,904,380.75	8.187	\$34,821.62	\$285,087.00	\$2.576/Mcf
2.50	90.00	10.325	\$194,633.84	\$2,009,656.75	8.544	\$35,073.32	\$299,659.00	\$2.717/Mcf
2.50	95.00	10.855	\$194,838.03	\$2,114,932.50	8.901	\$35,304.84	\$314,231.06	\$2.858/Mcf
2.50	100.00	11.384	\$195,023.25	\$2,220,208.50	9.257	\$35,518.52	\$328,803.00	\$2.999/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Trailers Required	Tractor Unit Cost Annual	Tractor Cost Per Year	Trailers Required	Trailer Unit Cost Annual	Trailer Cost Per Year	C02 Cost
3.00	5.00	1.589	\$166,168.00	\$263,958.56	2.765	\$21,199.63	\$58,616.17	\$0.316/Mcf
3.00	10.00	2.224	\$175,497.13	\$390,289.69	3.193	\$23,834.08	\$76,102.80	\$0.457/Mcf
3.00	15.00	2.859	\$180,680.00	\$516,620.81	3.621	\$25,845.66	\$93,589.38	\$0.598/Mcf
3.00	20.00	3.495	\$183,978.16	\$642,951.88	4.049	\$27,431.91	\$111,075.91	\$0.739/Mcf
3.00	25.00	4.130	\$186,261.53	\$769,283.00	4.477	\$28,714.83	\$128,562.39	\$0.880/Mcf
3.00	30.00	4.766	\$187,936.00	\$895,614.13	4.905	\$29,773.84	\$146,048.84	\$1.021/Mcf
3.00	35.00	5.401	\$189,216.47	\$1,021,945.50	5.333	\$30,662.85	\$163,535.34	\$1.162/Mcf
3.00	40.00	6.036	\$190,227.34	\$1,148,276.50	5.761	\$31,419.75	\$181,021.78	\$1.303/Mcf
3.00	45.00	6.672	\$191,045.69	\$1,274,607.50	6.189	\$32,071.96	\$198,508.19	\$1.444/Mcf
3.00	50.00	7.307	\$191,721.72	\$1,400,938.75	6.618	\$32,639.79	\$215,994.63	\$1.585/Mcf
3.00	55.00	7.943	\$192,289.59	\$1,527,270.00	7.046	\$33,138.61	\$233,481.03	\$1.726/Mcf
3.00	60.00	8.578	\$192,773.31	\$1,653,601.00	7.474	\$33,580.30	\$250,967.44	\$1.867/Mcf
3.00	65.00	9.213	\$193,190.34	\$1,779,932.50	7.902	\$33,974.13	\$268,453.81	\$2.008/Mcf
3.00	70.00	9.849	\$193,553.53	\$1,906,263.50	8.330	\$34,327.48	\$285,940.19	\$2.149/Mcf
3.00	75.00	10.484	\$193,872.72	\$2,032,594.50	8.758	\$34,646.29	\$303,426.63	\$2.290/Mcf
3.00	80.00	11.120	\$194,155.41	\$2,158,925.50	9.186	\$34,935.38	\$320,912.94	\$2.431/Mcf
3.00	85.00	11.755	\$194,407.56	\$2,285,257.00	9.614	\$35,198.73	\$338,399.31	\$2.572/Mcf
3.00	90.00	12.390	\$194,633.84	\$2,411,588.00	10.042	\$35,439.63	\$355,885.75	\$2.713/Mcf
3.00	95.00	13.026	\$194,838.03	\$2,537,919.00	10.470	\$35,660.84	\$373,372.19	\$2.854/Mcf
3.00	100.00	13.661	\$195,023.25	\$2,664,250.50	10.898	\$35,864.66	\$390,858.50	\$2.995/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
3.50	5.00	1.853	\$166,168.00	\$307,951.69	3.050	\$21,406.66	\$65,297.93	\$0.314/Mcf
3.50	10.00	2.595	\$175,497.13	\$455,337.88	3.550	\$24,142.19	\$85,699.03	\$0.455/Mcf
3.50	15.00	3.336	\$180,680.00	\$602,724.25	4.049	\$26,202.92	\$106,100.03	\$0.596/Mcf
3.50	20.00	4.077	\$183,978.16	\$750,110.38	4.549	\$27,811.13	\$126,500.98	\$0.737/Mcf
3.50	25.00	4.818	\$186,261.53	\$897,497.00	5.048	\$29,101.12	\$146,901.91	\$0.878/Mcf
3.50	30.00	5.560	\$187,936.00	\$1,044,883.25	5.547	\$30,158.84	\$167,302.78	\$1.019/Mcf
3.50	35.00	6.301	\$189,216.47	\$1,192,269.50	6.047	\$31,041.84	\$187,703.66	\$1.160/Mcf
3.50	40.00	7.042	\$190,227.34	\$1,339,655.75	6.546	\$31,790.12	\$208,104.50	\$1.301/Mcf
3.50	45.00	7.784	\$191,045.69	\$1,487,042.25	7.046	\$32,432.31	\$228,505.31	\$1.442/Mcf
3.50	50.00	8.525	\$191,721.72	\$1,634,428.50	7.545	\$32,989.48	\$248,906.13	\$1.583/Mcf
3.50	55.00	9.266	\$192,289.59	\$1,781,814.75	8.044	\$33,477.48	\$269,306.94	\$1.724/Mcf
3.50	60.00	10.008	\$192,773.31	\$1,929,201.25	8.544	\$33,908.43	\$289,707.75	\$1.865/Mcf
3.50	65.00	10.749	\$193,190.34	\$2,076,588.00	9.043	\$34,291.78	\$310,108.56	\$2.006/Mcf
3.50	70.00	11.490	\$193,553.53	\$2,223,973.50	9.543	\$34,635.00	\$330,509.25	\$2.147/Mcf
3.50	75.00	12.232	\$193,872.72	\$2,371,360.50	10.042	\$34,944.09	\$350,910.06	\$2.288/Mcf
3.50	80.00	12.973	\$194,155.41	\$2,518,746.50	10.541	\$35,223.88	\$371,310.88	\$2.429/Mcf
3.50	85.00	13.714	\$194,407.56	\$2,666,133.00	11.041	\$35,478.38	\$391,711.69	\$2.570/Mcf
3.50	90.00	14.455	\$194,633.84	\$2,813,519.50	11.540	\$35,710.84	\$412,112.50	\$2.711/Mcf
3.50	95.00	15.197	\$194,838.03	\$2,960,905.50	12.040	\$35,924.01	\$432,513.25	\$2.852/Mcf
3.50	100.00	15.938	\$195,023.25	\$3,108,292.00	12.539	\$36,120.20	\$452,914.00	\$2.993/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
4.00	5.00	2.118	\$166,168.00	\$351,944.81	3.336	\$21,578.27	\$71,979.70	\$0.312/Mcf
4.00	10.00	2.965	\$175,497.13	\$520,386.19	3.907	\$24,394.02	\$95,295.25	\$0.453/Mcf
4.00	15.00	3.812	\$180,680.00	\$688,827.75	4.477	\$26,491.86	\$118,610.69	\$0.594/Mcf
4.00	20.00	4.660	\$183,978.16	\$857,269.13	5.048	\$28,115.30	\$141,926.06	\$0.735/Mcf
4.00	25.00	5.507	\$186,261.53	\$1,025,710.75	5.619	\$29,408.92	\$165,241.38	\$0.876/Mcf
4.00	30.00	6.354	\$187,936.00	\$1,194,152.25	6.190	\$30,463.96	\$188,556.66	\$1.017/Mcf
4.00	35.00	7.201	\$189,216.47	\$1,362,594.00	6.760	\$31,340.84	\$211,871.97	\$1.158/Mcf
4.00	40.00	8.048	\$190,227.34	\$1,531,035.25	7.331	\$32,081.19	\$235,187.22	\$1.299/Mcf
4.00	45.00	8.896	\$191,045.69	\$1,699,476.75	7.902	\$32,714.58	\$258,502.44	\$1.440/Mcf
4.00	50.00	9.743	\$191,721.72	\$1,867,918.25	8.473	\$33,262.63	\$281,817.63	\$1.581/Mcf
4.00	55.00	10.590	\$192,289.59	\$2,036,360.00	9.043	\$33,741.50	\$305,132.81	\$1.722/Mcf
4.00	60.00	11.437	\$192,773.31	\$2,204,801.00	9.614	\$34,163.52	\$328,448.06	\$1.863/Mcf
4.00	65.00	12.284	\$193,190.34	\$2,373,243.00	10.185	\$34,538.23	\$351,763.25	\$2.004/Mcf
4.00	70.00	13.132	\$193,553.53	\$2,541,684.50	10.756	\$34,873.17	\$375,078.44	\$2.145/Mcf
4.00	75.00	13.979	\$193,872.72	\$2,710,126.00	11.326	\$35,174.36	\$398,393.63	\$2.286/Mcf
4.00	80.00	14.826	\$194,155.41	\$2,878,567.50	11.897	\$35,446.65	\$421,708.88	\$2.427/Mcf
4.00	85.00	15.673	\$194,407.56	\$3,047,009.00	12.468	\$35,694.01	\$445,024.06	\$2.568/Mcf
4.00	90.00	16.521	\$194,633.84	\$3,215,450.50	13.039	\$35,919.70	\$468,339.13	\$2.709/Mcf
4.00	95.00	17.368	\$194,838.03	\$3,383,891.50	13.609	\$36,126.48	\$491,654.38	\$2.850/Mcf
4.00	100.00	18.215	\$195,023.25	\$3,552,334.00	14.180	\$36,316.60	\$514,969.50	\$2.991/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
4.50	5.00	2.383	\$166,168.00	\$395,937.88	3.621	\$21,722.82	\$78,661.47	\$0.310/Mcf
4.50	10.00	3.336	\$175,497.13	\$585,434.50	4.263	\$24,603.71	\$104,891.48	\$0.451/Mcf
4.50	15.00	4.289	\$180,680.00	\$774,931.25	4.905	\$26,730.38	\$131,121.34	\$0.592/Mcf
4.50	20.00	5.242	\$183,978.16	\$964,427.75	5.547	\$28,364.72	\$157,351.16	\$0.733/Mcf
4.50	25.00	6.195	\$186,261.53	\$1,153,924.75	6.190	\$29,659.96	\$183,580.94	\$0.874/Mcf
4.50	30.00	7.148	\$187,936.00	\$1,343,421.50	6.832	\$30,711.72	\$209,810.59	\$1.015/Mcf
4.50	35.00	8.101	\$189,216.47	\$1,532,918.00	7.474	\$31,582.76	\$236,040.31	\$1.156/Mcf
4.50	40.00	9.055	\$190,227.34	\$1,722,414.50	8.116	\$32,315.97	\$262,269.94	\$1.297/Mcf
4.50	45.00	10.008	\$191,045.69	\$1,911,911.50	8.758	\$32,941.66	\$288,499.50	\$1.438/Mcf
4.50	50.00	10.961	\$191,721.72	\$2,101,408.00	9.400	\$33,481.87	\$314,729.19	\$1.579/Mcf
4.50	55.00	11.914	\$192,289.59	\$2,290,905.00	10.042	\$33,953.00	\$340,958.81	\$1.720/Mcf
4.50	60.00	12.867	\$192,773.31	\$2,480,401.50	10.684	\$34,367.50	\$367,188.44	\$1.861/Mcf
4.50	65.00	13.820	\$193,190.34	\$2,669,898.50	11.326	\$34,735.00	\$393,418.00	\$2.002/Mcf
4.50	70.00	14.773	\$193,553.53	\$2,859,395.00	11.968	\$35,063.07	\$419,647.56	\$2.143/Mcf
4.50	75.00	15.726	\$193,872.72	\$3,048,892.00	12.610	\$35,357.73	\$445,877.19	\$2.284/Mcf
4.50	80.00	16.679	\$194,155.41	\$3,238,388.50	13.253	\$35,623.84	\$472,106.75	\$2.425/Mcf
4.50	85.00	17.632	\$194,407.56	\$3,427,885.00	13.895	\$35,865.35	\$498,336.38	\$2.566/Mcf
4.50	90.00	18.586	\$194,633.84	\$3,617,382.50	14.537	\$36,085.52	\$524,565.88	\$2.707/Mcf
4.50	95.00	19.539	\$194,838.03	\$3,806,878.50	15.179	\$36,287.07	\$550,795.50	\$2.848/Mcf
4.50	100.00	20.492	\$195,023.25	\$3,996,375.50	15.821	\$36,472.27	\$577,025.13	\$2.989/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
5.00	5.00	2.648	\$166,168.00	\$439,931.00	3.907	\$21,846.27	\$85,343.25	\$0.309/Mcf
5.00	10.00	3.707	\$175,497.13	\$650,482.75	4.620	\$24,781.02	\$114,487.70	\$0.450/Mcf
5.00	15.00	4.766	\$180,680.00	\$861,034.63	5.333	\$26,930.60	\$143,632.00	\$0.591/Mcf
5.00	20.00	5.825	\$183,978.16	\$1,071,586.50	6.047	\$28,572.93	\$172,776.25	\$0.732/Mcf
5.00	25.00	6.884	\$186,261.53	\$1,282,138.50	6.760	\$29,868.61	\$201,920.41	\$0.873/Mcf
5.00	30.00	7.943	\$187,936.00	\$1,492,690.25	7.474	\$30,916.91	\$231,064.53	\$1.014/Mcf
5.00	35.00	9.002	\$189,216.47	\$1,703,242.25	8.187	\$31,782.51	\$260,208.63	\$1.155/Mcf
5.00	40.00	10.061	\$190,227.34	\$1,913,794.00	8.901	\$32,509.34	\$289,352.69	\$1.296/Mcf
5.00	45.00	11.120	\$191,045.69	\$2,124,346.00	9.614	\$33,128.30	\$318,496.69	\$1.437/Mcf
5.00	50.00	12.179	\$191,721.72	\$2,334,898.00	10.327	\$33,661.73	\$347,640.75	\$1.578/Mcf
5.00	55.00	13.238	\$192,289.59	\$2,545,450.00	11.041	\$34,126.23	\$376,784.75	\$1.719/Mcf
5.00	60.00	14.297	\$192,773.31	\$2,756,001.50	11.754	\$34,534.34	\$405,928.75	\$1.860/Mcf
5.00	65.00	15.356	\$193,190.34	\$2,966,554.00	12.468	\$34,895.74	\$435,072.75	\$2.001/Mcf
5.00	70.00	16.415	\$193,553.53	\$3,177,105.00	13.181	\$35,218.02	\$464,216.81	\$2.142/Mcf
5.00	75.00	17.474	\$193,872.72	\$3,387,657.50	13.895	\$35,507.20	\$493,360.75	\$2.283/Mcf
5.00	80.00	18.533	\$194,155.41	\$3,598,209.50	14.608	\$35,768.14	\$522,504.69	\$2.424/Mcf
5.00	85.00	19.592	\$194,407.56	\$3,808,761.50	15.322	\$36,004.78	\$551,648.75	\$2.565/Mcf
5.00	90.00	20.651	\$194,633.84	\$4,019,313.50	16.035	\$36,220.36	\$580,792.75	\$2.706/Mcf
5.00	95.00	21.710	\$194,838.03	\$4,229,865.00	16.748	\$36,417.57	\$609,936.63	\$2.847/Mcf
5.00	100.00	22.769	\$195,023.25	\$4,440,417.00	17.462	\$36,598.67	\$639,080.63	\$2.988/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
5.50	5.00	2.912	\$166,168.00	\$483,924.06	4.192	\$21,952.89	\$92,025.00	\$0.308/Mcf
5.50	10.00	4.077	\$175,497.13	\$715,531.00	4.977	\$24,932.91	\$124,083.94	\$0.449/Mcf
5.50	15.00	5.242	\$180,680.00	\$947,138.25	5.761	\$27,101.08	\$156,142.69	\$0.590/Mcf
5.50	20.00	6.407	\$183,978.16	\$1,178,745.00	6.546	\$28,749.38	\$188,201.34	\$0.731/Mcf
5.50	25.00	7.572	\$186,261.53	\$1,410,352.50	7.331	\$30,044.77	\$220,259.91	\$0.872/Mcf
5.50	30.00	8.737	\$187,936.00	\$1,641,959.25	8.116	\$31,089.63	\$252,318.44	\$1.013/Mcf
5.50	35.00	9.902	\$189,216.47	\$1,873,566.50	8.901	\$31,950.24	\$284,376.94	\$1.154/Mcf
5.50	40.00	11.067	\$190,227.34	\$2,105,173.50	9.685	\$32,671.38	\$316,435.38	\$1.295/Mcf
5.50	45.00	12.232	\$191,045.69	\$2,336,780.50	10.470	\$33,284.41	\$348,493.81	\$1.436/Mcf
5.50	50.00	13.396	\$191,721.72	\$2,568,387.50	11.255	\$33,811.95	\$380,552.31	\$1.577/Mcf
5.50	55.00	14.561	\$192,289.59	\$2,799,995.00	12.040	\$34,270.71	\$412,610.69	\$1.718/Mcf
5.50	60.00	15.726	\$192,773.31	\$3,031,601.50	12.825	\$34,673.34	\$444,669.13	\$1.859/Mcf
5.50	65.00	16.891	\$193,190.34	\$3,263,209.50	13.609	\$35,029.52	\$476,727.50	\$2.000/Mcf
5.50	70.00	18.056	\$193,553.53	\$3,494,816.50	14.394	\$35,346.86	\$508,785.88	\$2.141/Mcf
5.50	75.00	19.221	\$193,872.72	\$3,726,423.50	15.179	\$35,631.38	\$540,844.13	\$2.282/Mcf
5.50	80.00	20.386	\$194,155.41	\$3,958,030.00	15.964	\$35,887.94	\$572,902.63	\$2.423/Mcf
5.50	85.00	21.551	\$194,407.56	\$4,189,637.50	16.748	\$36,120.45	\$604,961.00	\$2.564/Mcf
5.50	90.00	22.716	\$194,633.84	\$4,421,245.00	17.533	\$36,332.15	\$637,019.38	\$2.705/Mcf
5.50	95.00	23.881	\$194,838.03	\$4,652,852.00	18.318	\$36,525.70	\$669,077.63	\$2.846/Mcf
5.50	100.00	25.046	\$195,023.25	\$4,884,459.00	19.103	\$36,703.35	\$701,136.00	\$2.987/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
6.00	5.00	3.177	\$166,168.00	\$527,917.13	4.477	\$22,045.93	\$98,706.77	\$0.307/Mcf
6.00	10.00	4.448	\$175,497.13	\$780,579.38	5.333	\$25,064.48	\$133,680.16	\$0.448/Mcf
6.00	15.00	5.719	\$180,680.00	\$1,033,241.63	6.190	\$27,247.97	\$168,653.34	\$0.589/Mcf
6.00	20.00	6.989	\$183,978.16	\$1,285,903.75	7.046	\$28,900.81	\$203,626.44	\$0.730/Mcf
6.00	25.00	8.260	\$186,261.53	\$1,538,566.00	7.902	\$30,195.48	\$238,599.41	\$0.871/Mcf
6.00	30.00	9.531	\$187,936.00	\$1,791,228.25	8.758	\$31,237.03	\$273,572.38	\$1.012/Mcf
6.00	35.00	10.802	\$189,216.47	\$2,043,891.00	9.614	\$32,093.08	\$308,545.25	\$1.153/Mcf
6.00	40.00	12.073	\$190,227.34	\$2,296,553.00	10.470	\$32,809.13	\$343,518.13	\$1.294/Mcf
6.00	45.00	13.343	\$191,045.69	\$2,549,215.00	11.326	\$33,416.92	\$378,491.00	\$1.435/Mcf
6.00	50.00	14.614	\$191,721.72	\$2,801,877.50	12.182	\$33,939.30	\$413,463.88	\$1.576/Mcf
6.00	55.00	15.885	\$192,289.59	\$3,054,540.00	13.039	\$34,393.06	\$448,436.63	\$1.717/Mcf
6.00	60.00	17.156	\$192,773.31	\$3,307,202.00	13.895	\$34,790.91	\$483,409.38	\$1.858/Mcf
6.00	65.00	18.427	\$193,190.34	\$3,559,865.00	14.751	\$35,142.59	\$518,382.25	\$1.999/Mcf
6.00	70.00	19.698	\$193,553.53	\$3,812,527.00	15.607	\$35,455.67	\$553,355.00	\$2.140/Mcf
6.00	75.00	20.968	\$193,872.72	\$4,065,189.00	16.463	\$35,736.20	\$588,327.75	\$2.281/Mcf
6.00	80.00	22.239	\$194,155.41	\$4,317,851.00	17.319	\$35,988.98	\$623,300.63	\$2.422/Mcf
6.00	85.00	23.510	\$194,407.56	\$4,570,514.00	18.175	\$36,217.96	\$658,273.38	\$2.563/Mcf
6.00	90.00	24.781	\$194,633.84	\$4,823,176.00	19.031	\$36,426.33	\$693,246.00	\$2.704/Mcf
6.00	95.00	26.052	\$194,838.03	\$5,075,838.00	19.888	\$36,616.77	\$728,218.88	\$2.845/Mcf
6.00	100.00	27.322	\$195,023.25	\$5,328,501.00	20.744	\$36,791.48	\$763,191.63	\$2.986/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
6.50	5.00	3.442	\$166,168.00	\$571,910.38	4.763	\$22,127.82	\$105,388.55	\$0.306/Mcf
6.50	10.00	4.818	\$175,497.13	\$845,627.63	5.690	\$25,179.56	\$143,276.41	\$0.447/Mcf
6.50	15.00	6.195	\$180,680.00	\$1,119,345.25	6.618	\$27,375.86	\$181,164.03	\$0.588/Mcf
6.50	20.00	7.572	\$183,978.16	\$1,393,062.50	7.545	\$29,032.19	\$219,051.53	\$0.729/Mcf
6.50	25.00	8.949	\$186,261.53	\$1,666,780.00	8.473	\$30,325.89	\$256,938.94	\$0.870/Mcf
6.50	30.00	10.325	\$187,936.00	\$1,940,497.75	9.400	\$31,364.29	\$294,826.31	\$1.011/Mcf
6.50	35.00	11.702	\$189,216.47	\$2,214,215.00	10.328	\$32,216.18	\$332,713.63	\$1.152/Mcf
6.50	40.00	13.079	\$190,227.34	\$2,487,932.00	11.255	\$32,927.66	\$370,600.94	\$1.293/Mcf
6.50	45.00	14.455	\$191,045.69	\$2,761,649.50	12.182	\$33,530.81	\$408,488.19	\$1.434/Mcf
6.50	50.00	15.832	\$191,721.72	\$3,035,367.50	13.110	\$34,048.62	\$446,375.38	\$1.575/Mcf
6.50	55.00	17.209	\$192,289.59	\$3,309,085.00	14.037	\$34,498.01	\$484,262.69	\$1.716/Mcf
6.50	60.00	18.586	\$192,773.31	\$3,582,802.00	14.965	\$34,891.68	\$522,149.75	\$1.857/Mcf
6.50	65.00	19.962	\$193,190.34	\$3,856,520.50	15.892	\$35,239.41	\$560,037.00	\$1.998/Mcf
6.50	70.00	21.339	\$193,553.53	\$4,130,237.00	16.820	\$35,548.80	\$597,924.13	\$2.139/Mcf
6.50	75.00	22.716	\$193,872.72	\$4,403,954.00	17.747	\$35,825.84	\$635,811.25	\$2.280/Mcf
6.50	80.00	24.092	\$194,155.41	\$4,677,672.00	18.675	\$36,075.36	\$673,698.50	\$2.421/Mcf
6.50	85.00	25.469	\$194,407.56	\$4,951,390.00	19.602	\$36,301.27	\$711,585.63	\$2.562/Mcf
6.50	90.00	26.846	\$194,633.84	\$5,225,108.00	20.530	\$36,506.77	\$749,472.75	\$2.703/Mcf
6.50	95.00	28.223	\$194,838.03	\$5,498,825.00	21.457	\$36,694.51	\$787,359.88	\$2.844/Mcf
6.50	100.00	29.599	\$195,023.25	\$5,772,542.00	22.385	\$36,866.69	\$825,247.13	\$2.985/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
7.00	5.00	3.707	\$166,168.00	\$615,903.38	5.048	\$22,200.45	\$112,070.33	\$0.306/Mcf
7.00	10.00	5.189	\$175,497.13	\$910,675.75	6.047	\$25,281.05	\$152,872.63	\$0.447/Mcf
7.00	15.00	6.672	\$180,680.00	\$1,205,448.50	7.046	\$27,488.21	\$193,674.69	\$0.588/Mcf
7.00	20.00	8.154	\$183,978.16	\$1,500,220.75	8.045	\$29,147.26	\$234,476.59	\$0.729/Mcf
7.00	25.00	9.637	\$186,261.53	\$1,794,994.00	9.043	\$30,439.83	\$275,278.44	\$0.870/Mcf
7.00	30.00	11.120	\$187,936.00	\$2,089,766.50	10.042	\$31,475.27	\$316,080.19	\$1.011/Mcf
7.00	35.00	12.602	\$189,216.47	\$2,384,539.00	11.041	\$32,323.37	\$356,881.94	\$1.152/Mcf
7.00	40.00	14.085	\$190,227.34	\$2,679,311.50	12.040	\$33,030.74	\$397,683.56	\$1.293/Mcf
7.00	45.00	15.567	\$191,045.69	\$2,974,084.50	13.039	\$33,629.74	\$438,485.25	\$1.434/Mcf
7.00	50.00	17.050	\$191,721.72	\$3,268,857.00	14.037	\$34,143.50	\$479,286.88	\$1.575/Mcf
7.00	55.00	18.533	\$192,289.59	\$3,563,629.50	15.036	\$34,589.00	\$520,088.50	\$1.716/Mcf
7.00	60.00	20.015	\$192,773.31	\$3,858,402.50	16.035	\$34,979.00	\$560,890.13	\$1.857/Mcf
7.00	65.00	21.498	\$193,190.34	\$4,153,176.00	17.034	\$35,323.27	\$601,691.75	\$1.998/Mcf
7.00	70.00	22.980	\$193,553.53	\$4,447,947.00	18.033	\$35,629.39	\$642,493.38	\$2.139/Mcf
7.00	75.00	24.463	\$193,872.72	\$4,742,721.00	19.031	\$35,903.38	\$683,294.75	\$2.280/Mcf
7.00	80.00	25.946	\$194,155.41	\$5,037,493.00	20.030	\$36,150.05	\$724,096.38	\$2.421/Mcf
7.00	85.00	27.428	\$194,407.56	\$5,332,266.00	21.029	\$36,373.28	\$764,898.00	\$2.562/Mcf
7.00	90.00	28.911	\$194,633.84	\$5,627,039.00	22.028	\$36,576.27	\$805,699.63	\$2.703/Mcf
7.00	95.00	30.394	\$194,838.03	\$5,921,811.00	23.027	\$36,761.65	\$846,501.00	\$2.844/Mcf
7.00	100.00	31.876	\$195,023.25	\$6,216,584.00	24.026	\$36,931.62	\$887,302.63	\$2.985/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
7.50	5.00	3.971	\$166,168.00	\$659,896.50	5.334	\$22,265.30	\$118,752.09	\$0.305/Mcf
7.50	10.00	5.560	\$175,497.13	\$975,724.25	6.404	\$25,371.24	\$162,468.88	\$0.446/Mcf
7.50	15.00	7.148	\$180,680.00	\$1,291,552.00	7.474	\$27,587.68	\$206,185.34	\$0.587/Mcf
7.50	20.00	8.737	\$183,978.16	\$1,607,379.75	8.544	\$29,248.88	\$249,901.69	\$0.728/Mcf
7.50	25.00	10.325	\$186,261.53	\$1,923,207.50	9.614	\$30,540.25	\$293,617.94	\$0.869/Mcf
7.50	30.00	11.914	\$187,936.00	\$2,239,035.50	10.684	\$31,572.91	\$337,334.13	\$1.010/Mcf
7.50	35.00	13.502	\$189,216.47	\$2,554,863.50	11.754	\$32,417.55	\$381,050.25	\$1.151/Mcf
7.50	40.00	15.091	\$190,227.34	\$2,870,691.00	12.825	\$33,121.21	\$424,766.31	\$1.292/Mcf
7.50	45.00	16.679	\$191,045.69	\$3,186,518.50	13.895	\$33,716.48	\$468,482.44	\$1.433/Mcf
7.50	50.00	18.268	\$191,721.72	\$3,502,347.00	14.965	\$34,226.62	\$512,198.44	\$1.574/Mcf
7.50	55.00	19.856	\$192,289.59	\$3,818,175.00	16.035	\$34,668.66	\$555,914.50	\$1.715/Mcf
7.50	60.00	21.445	\$192,773.31	\$4,134,003.00	17.105	\$35,055.39	\$599,630.38	\$1.856/Mcf
7.50	65.00	23.033	\$193,190.34	\$4,449,831.00	18.175	\$35,396.58	\$643,346.38	\$1.997/Mcf
7.50	70.00	24.622	\$193,553.53	\$4,765,658.00	19.246	\$35,699.82	\$687,062.25	\$2.138/Mcf
7.50	75.00	26.210	\$193,872.72	\$5,081,486.00	20.316	\$35,971.12	\$730,778.38	\$2.279/Mcf
7.50	80.00	27.799	\$194,155.41	\$5,397,314.00	21.386	\$36,215.27	\$774,494.38	\$2.420/Mcf
7.50	85.00	29.387	\$194,407.56	\$5,713,142.00	22.456	\$36,436.14	\$818,210.38	\$2.561/Mcf
7.50	90.00	30.976	\$194,633.84	\$6,028,970.00	23.526	\$36,636.91	\$861,926.13	\$2.702/Mcf
7.50	95.00	32.564	\$194,838.03	\$6,344,798.00	24.596	\$36,820.23	\$905,642.25	\$2.843/Mcf
7.50	100.00	34.153	\$195,023.25	\$6,660,626.00	25.666	\$36,988.25	\$949,358.25	\$2.984/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Trailers Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
8.00	5.00	4.236	\$166,168.00	\$703,889.63	5.619	\$22,323.57	\$125,433.89	\$0.305/Mcf
8.00	10.00	5.930	\$175,497.13	\$1,040,772.38	6.760	\$25,451.91	\$172,065.09	\$0.446/Mcf
8.00	15.00	7.625	\$180,680.00	\$1,377,655.50	7.902	\$27,676.38	\$218,696.03	\$0.587/Mcf
8.00	20.00	9.319	\$183,978.16	\$1,714,538.25	9.043	\$29,339.27	\$265,326.81	\$0.728/Mcf
8.00	25.00	11.014	\$186,261.53	\$2,051,421.50	10.185	\$30,629.41	\$311,957.50	\$0.869/Mcf
8.00	30.00	12.708	\$187,936.00	\$2,388,304.50	11.326	\$31,659.49	\$358,588.06	\$1.010/Mcf
8.00	35.00	14.402	\$189,216.47	\$2,725,188.00	12.468	\$32,500.95	\$405,218.63	\$1.151/Mcf
8.00	40.00	16.097	\$190,227.34	\$3,062,070.50	13.609	\$33,201.24	\$451,849.06	\$1.292/Mcf
8.00	45.00	17.791	\$191,045.69	\$3,398,953.50	14.751	\$33,793.16	\$498,479.56	\$1.433/Mcf
8.00	50.00	19.486	\$191,721.72	\$3,735,836.50	15.892	\$34,300.03	\$545,109.88	\$1.574/Mcf
8.00	55.00	21.180	\$192,289.59	\$4,072,720.00	17.034	\$34,738.98	\$591,740.38	\$1.715/Mcf
8.00	60.00	22.875	\$192,773.31	\$4,409,602.00	18.175	\$35,122.78	\$638,370.63	\$1.856/Mcf
8.00	65.00	24.569	\$193,190.34	\$4,746,486.00	19.317	\$35,461.23	\$685,001.13	\$1.997/Mcf
8.00	70.00	26.263	\$193,553.53	\$5,083,369.00	20.458	\$35,761.91	\$731,631.50	\$2.138/Mcf
8.00	75.00	27.958	\$193,872.72	\$5,420,252.00	21.600	\$36,030.80	\$778,262.00	\$2.279/Mcf
8.00	80.00	29.652	\$194,155.41	\$5,757,135.00	22.741	\$36,272.70	\$824,892.13	\$2.420/Mcf
8.00	85.00	31.347	\$194,407.56	\$6,094,018.00	23.883	\$36,491.48	\$871,522.75	\$2.561/Mcf
8.00	90.00	33.041	\$194,633.84	\$6,430,901.00	25.024	\$36,690.30	\$918,152.88	\$2.702/Mcf
8.00	95.00	34.735	\$194,838.03	\$6,767,783.00	26.166	\$36,871.77	\$964,783.38	\$2.843/Mcf
8.00	100.00	36.430	\$195,023.25	\$7,104,668.00	27.307	\$37,038.07	\$1,011,413.63	\$2.984/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
8.50	5.00	4.501	\$166,168.00	\$747,882.75	5.904	\$22,376.20	\$132,115.66	\$0.304/Mcf
8.50	10.00	6.301	\$175,497.13	\$1,105,820.75	7.117	\$25,524.49	\$181,661.31	\$0.445/Mcf
8.50	15.00	8.101	\$180,680.00	\$1,463,758.75	8.330	\$27,755.97	\$231,206.72	\$0.586/Mcf
8.50	20.00	9.902	\$183,978.16	\$1,821,697.00	9.543	\$29,420.21	\$280,751.88	\$0.727/Mcf
8.50	25.00	11.702	\$186,261.53	\$2,179,635.50	10.756	\$30,709.11	\$330,297.00	\$0.868/Mcf
8.50	30.00	13.502	\$187,936.00	\$2,537,574.00	11.969	\$31,736.77	\$379,842.00	\$1.009/Mcf
8.50	35.00	15.303	\$189,216.47	\$2,895,512.00	13.181	\$32,575.32	\$429,386.94	\$1.150/Mcf
8.50	40.00	17.103	\$190,227.34	\$3,253,449.50	14.394	\$33,272.55	\$478,931.75	\$1.291/Mcf
8.50	45.00	18.903	\$191,045.69	\$3,611,388.00	15.607	\$33,861.41	\$528,476.63	\$1.432/Mcf
8.50	50.00	20.704	\$191,721.72	\$3,969,326.00	16.820	\$34,365.36	\$578,021.63	\$1.573/Mcf
8.50	55.00	22.504	\$192,289.59	\$4,327,265.00	18.033	\$34,801.51	\$627,566.25	\$1.714/Mcf
8.50	60.00	24.304	\$192,773.31	\$4,685,203.00	19.246	\$35,182.69	\$677,111.13	\$1.855/Mcf
8.50	65.00	26.105	\$193,190.34	\$5,043,141.00	20.458	\$35,518.67	\$726,656.00	\$1.996/Mcf
8.50	70.00	27.905	\$193,553.53	\$5,401,080.00	21.671	\$35,817.05	\$776,200.75	\$2.137/Mcf
8.50	75.00	29.705	\$193,872.72	\$5,759,018.00	22.884	\$36,083.79	\$825,745.50	\$2.278/Mcf
8.50	80.00	31.505	\$194,155.41	\$6,116,956.00	24.097	\$36,323.69	\$875,290.38	\$2.419/Mcf
8.50	85.00	33.306	\$194,407.56	\$6,474,895.00	25.310	\$36,540.59	\$924,835.13	\$2.560/Mcf
8.50	90.00	35.106	\$194,633.84	\$6,832,832.00	26.523	\$36,737.66	\$974,379.75	\$2.701/Mcf
8.50	95.00	36.906	\$194,838.03	\$7,190,770.00	27.735	\$36,917.48	\$1,023,924.38	\$2.842/Mcf
8.50	100.00	38.707	\$195,023.25	\$7,548,710.00	28.948	\$37,082.25	\$1,073,469.25	\$2.983/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
9.00	5.00	4.766	\$166,168.00	\$791,875.75	6.190	\$22,423.98	\$138,797.44	\$0.304/Mcf
9.00	10.00	6.672	\$175,497.13	\$1,170,869.00	7.474	\$25,590.15	\$191,257.59	\$0.445/Mcf
9.00	15.00	8.578	\$180,680.00	\$1,549,862.50	8.758	\$27,827.77	\$243,717.38	\$0.586/Mcf
9.00	20.00	10.484	\$183,978.16	\$1,928,855.50	10.042	\$29,493.09	\$296,177.00	\$0.727/Mcf
9.00	25.00	12.390	\$186,261.53	\$2,307,849.50	11.326	\$30,780.77	\$348,636.50	\$0.868/Mcf
9.00	30.00	14.297	\$187,936.00	\$2,686,843.00	12.611	\$31,806.18	\$401,095.94	\$1.009/Mcf
9.00	35.00	16.203	\$189,216.47	\$3,065,836.00	13.895	\$32,642.05	\$453,555.31	\$1.150/Mcf
9.00	40.00	18.109	\$190,227.34	\$3,444,829.00	15.179	\$33,336.48	\$506,014.56	\$1.291/Mcf
9.00	45.00	20.015	\$191,045.69	\$3,823,823.00	16.463	\$33,922.58	\$558,473.88	\$1.432/Mcf
9.00	50.00	21.921	\$191,721.72	\$4,202,816.00	17.747	\$34,423.85	\$610,933.13	\$1.573/Mcf
9.00	55.00	23.828	\$192,289.59	\$4,581,810.00	19.032	\$34,857.47	\$663,392.13	\$1.714/Mcf
9.00	60.00	25.734	\$192,773.31	\$4,960,803.00	20.316	\$35,236.27	\$715,851.38	\$1.855/Mcf
9.00	65.00	27.640	\$193,190.34	\$5,339,797.00	21.600	\$35,570.03	\$768,310.63	\$1.996/Mcf
9.00	70.00	29.546	\$193,553.53	\$5,718,790.00	22.884	\$35,866.34	\$820,769.88	\$2.137/Mcf
9.00	75.00	31.453	\$193,872.72	\$6,097,784.00	24.168	\$36,131.14	\$873,228.88	\$2.278/Mcf
9.00	80.00	33.359	\$194,155.41	\$6,476,777.00	25.453	\$36,369.23	\$925,688.13	\$2.419/Mcf
9.00	85.00	35.265	\$194,407.56	\$6,855,770.00	26.737	\$36,584.45	\$978,147.25	\$2.560/Mcf
9.00	90.00	37.171	\$194,633.84	\$7,234,765.00	28.021	\$36,779.95	\$1,030,606.63	\$2.701/Mcf
9.00	95.00	39.077	\$194,838.03	\$7,613,757.00	29.305	\$36,958.30	\$1,083,065.50	\$2.842/Mcf
9.00	100.00	40.984	\$195,023.25	\$7,992,751.00	30.589	\$37,121.68	\$1,135,524.50	\$2.983/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
9.50	5.00	5.030	\$166,168.00	\$835,869.00	6.475	\$22,467.55	\$145,479.22	\$0.304/Mcf
9.50	10.00	7.042	\$175,497.13	\$1,235,917.25	7.831	\$25,649.82	\$200,853.81	\$0.445/Mcf
9.50	15.00	9.054	\$180,680.00	\$1,635,966.00	9.186	\$27,892.88	\$256,228.03	\$0.586/Mcf
9.50	20.00	11.067	\$183,978.16	\$2,036,014.50	10.542	\$29,559.07	\$311,602.06	\$0.727/Mcf
9.50	25.00	13.079	\$186,261.53	\$2,436,063.00	11.897	\$30,845.56	\$366,976.00	\$0.868/Mcf
9.50	30.00	15.091	\$187,936.00	\$2,836,111.50	13.253	\$31,868.87	\$422,349.88	\$1.009/Mcf
9.50	35.00	17.103	\$189,216.47	\$3,236,160.50	14.608	\$32,702.27	\$477,723.63	\$1.150/Mcf
9.50	40.00	19.115	\$190,227.34	\$3,636,209.00	15.964	\$33,394.13	\$533,097.25	\$1.291/Mcf
9.50	45.00	21.127	\$191,045.69	\$4,036,257.50	17.319	\$33,977.69	\$588,470.88	\$1.432/Mcf
9.50	50.00	23.139	\$191,721.72	\$4,436,306.00	18.675	\$34,476.53	\$643,844.50	\$1.573/Mcf
9.50	55.00	25.151	\$192,289.59	\$4,836,354.00	20.030	\$34,907.85	\$699,218.13	\$1.714/Mcf
9.50	60.00	27.164	\$192,773.31	\$5,236,403.00	21.386	\$35,284.50	\$754,591.75	\$1.855/Mcf
9.50	65.00	29.176	\$193,190.34	\$5,636,452.00	22.741	\$35,616.24	\$809,965.38	\$1.996/Mcf
9.50	70.00	31.188	\$193,553.53	\$6,036,500.00	24.097	\$35,910.66	\$865,339.00	\$2.137/Mcf
9.50	75.00	33.200	\$193,872.72	\$6,436,550.00	25.453	\$36,173.72	\$920,712.50	\$2.278/Mcf
9.50	80.00	35.212	\$194,155.41	\$6,836,598.00	26.808	\$36,410.18	\$976,086.25	\$2.419/Mcf
9.50	85.00	37.224	\$194,407.56	\$7,236,647.00	28.164	\$36,623.88	\$1,031,459.75	\$2.560/Mcf
9.50	90.00	39.236	\$194,633.84	\$7,636,696.00	29.519	\$36,817.94	\$1,086,833.00	\$2.701/Mcf
9.50	95.00	41.248	\$194,838.03	\$8,036,744.00	30.875	\$36,994.97	\$1,142,206.75	\$2.842/Mcf
9.50	100.00	43.260	\$195,023.25	\$8,436,792.00	32.230	\$37,157.10	\$1,197,580.25	\$2.983/Mcf

C02 Demand (MMcfd)	Haul Distance (miles)	Tractors Required	Tractor Annual Unit Cost	Tractor Cost Per Year	Trailers Required	Trailer Annual Unit Cost	Trailer Cost Per Year	C02 Cost
10.00	5.00	5.295	\$166,168.00	\$879,862.00	6.760	\$22,507.44	\$152,160.97	\$0.304/Mcf
10.00	10.00	7.413	\$175,497.13	\$1,300,965.50	8.187	\$25,704.29	\$210,450.00	\$0.445/Mcf
10.00	15.00	9.531	\$180,680.00	\$1,722,069.25	9.614	\$27,952.20	\$268,738.69	\$0.586/Mcf
10.00	20.00	11.649	\$183,978.16	\$2,143,173.00	11.041	\$29,619.07	\$327,027.19	\$0.727/Mcf
10.00	25.00	13.767	\$186,261.53	\$2,564,277.00	12.468	\$30,904.42	\$385,315.56	\$0.868/Mcf
10.00	30.00	15.885	\$187,936.00	\$2,985,380.50	13.895	\$31,925.77	\$443,603.81	\$1.009/Mcf
10.00	35.00	18.003	\$189,216.47	\$3,406,484.50	15.322	\$32,756.88	\$501,891.94	\$1.150/Mcf
10.00	40.00	20.121	\$190,227.34	\$3,827,588.00	16.749	\$33,446.38	\$560,180.00	\$1.291/Mcf
10.00	45.00	22.239	\$191,045.69	\$4,248,692.00	18.175	\$34,027.61	\$618,468.13	\$1.432/Mcf
10.00	50.00	24.357	\$191,721.72	\$4,669,796.00	19.602	\$34,524.23	\$676,756.25	\$1.573/Mcf
10.00	55.00	26.475	\$192,289.59	\$5,090,900.00	21.029	\$34,953.45	\$735,044.25	\$1.714/Mcf
10.00	60.00	28.593	\$192,773.31	\$5,512,003.00	22.456	\$35,328.13	\$793,332.13	\$1.855/Mcf
10.00	65.00	30.711	\$193,190.34	\$5,933,108.00	23.883	\$35,658.04	\$851,620.13	\$1.996/Mcf
10.00	70.00	32.829	\$193,553.53	\$6,354,210.00	25.310	\$35,950.74	\$909,908.00	\$2.137/Mcf
10.00	75.00	34.947	\$193,872.72	\$6,775,315.00	26.737	\$36,212.21	\$968,196.13	\$2.278/Mcf
10.00	80.00	37.065	\$194,155.41	\$7,196,419.00	28.164	\$36,447.18	\$1,026,484.00	\$2.419/Mcf
10.00	85.00	39.183	\$194,407.56	\$7,617,523.00	29.590	\$36,659.49	\$1,084,772.00	\$2.559/Mcf
10.00	90.00	41.301	\$194,633.84	\$8,038,627.00	31.017	\$36,852.27	\$1,143,060.00	\$2.700/Mcf
10.00	95.00	43.419	\$194,838.03	\$8,459,730.00	32.444	\$37,028.09	\$1,201,347.75	\$2.841/Mcf
10.00	100.00	45.537	\$195,023.25	\$8,880,834.00	33.871	\$37,189.09	\$1,259,635.75	\$2.982/Mcf

APPENDIX B

In order to determine the upstream pressure necessary to provide the necessary cooling through expansion, the following procedures should be followed:

The outlet conditions from the turboexpander should be determined from a plot of critical pressure vs. critical temperature for CO₂ - CH₄ composition of the feed gas.

This plot shown in Figure () gives the critical pressure vs. critical temperature for CO₂ - CH₄ mixtures. For the CO₂-CH₄ mixtures of interest (% CO₂ greater than 30 but less than 80) the critical point can be determined from this figure. In order to obtain the most CO₂ in the liquid stream out of the liquid-vapor separator, the outlet pressure out of the turboexpander should be higher than the critical pressure for the feed, and the outlet temperature should be 10, 20, or 30 degrees lower than the critical temperature of the mixture. The limit on how much the outlet temperature should be lower than the critical temperature depends the amount of CH₄ desired in the CO₂ stream and also the temperature at which solid CO₂ may appear (which can cause severe operational problems).

• Once the outlet pressure and temperature (P₂ and T₂) are set and since the inlet temperature (T₁) is known, Figures B-1 through B-4 can be used to calculate the upstream pressure as follows:

- STEP 1. Calculate (T₁ - T₂) and from figure B-1. Knowing T₁ determine the factor A
- STEP 2. From Figure B-2, using A determined in step 1, and a proper efficiency, determine the factor B. The efficiency is varied from 60 to 90%. The efficiency depends on a few factors, among them the amount of liquid produced.
- STEP 3. From Figure B-3, using B obtained in Step 2 and the proper l/m, determine the factor C. As explained in the main text:

$$l/m = k/(k-1)$$

$$\text{when, } k = C_p/C_v$$

$$\text{and, } l/m = C_p/(C_p - C_v)$$

l/m values depend on methane-carbon dioxide mixture composition, temperature, and pressure. The l/m values can be obtained from Tables

STEP 4. From Figure B-4, using the factor C obtained in Step 3 and knowing the outlet pressure (P_2), determine the upstream pressure P_1 .

Example:

Data: Desired outlet conditions $T_2 = -35^{\circ}\text{F}$, $P_2 = 1000$ psi
 Inlet Temperature $T_1 = 100^{\circ}\text{F}$
 Calculate: Upstream pressure $P_1 = ?$

The solution is sketched through Figures B-1 to B-4.

<u>Figure No.</u>	<u>Known Factors</u>	<u>Obtained Factor</u>
B-1	$T_1 - T_2 = 135$, $T_1 = 100$	$A = 0.24$
B-2	$A = 0.24$, $E = 0.7$	$B = 0.34$
B-3	$B = 0.34$, $1/m = 2.5$	$C = 2.8$
B-4	$C = 2.8$, $P_2 = 1000$	$P_1 = 2750$

Upstream Pressure = 2750 Psi

Procedure for Calculating Work Out of the Turboexpander

The procedure for calculating the amount of work expected out of a turboexpander is described through the use of an example:

- Basis: 1 lb mole of feed 60 mole% methane
 40 mole % carbon dioxide

Outlet temperature (T_2) = -35°F
 Outlet Pressure (P_2) = 500 psi

Data: Inlet temperature (T_1) = 100°F
 Inlet Pressure (P_1) = 1400 psi

STEP 1. Knowing $T_2 = -35^{\circ}\text{F}$, and $P_2 = 500$ psi, the liquid mole fraction L is obtained from Figure B-5.
 $L = 0.406$

STEP 2. Knowing $T_1 = 100^{\circ}\text{F}$, and $P_1 = 1400$ psi, the inlet enthalpy (H_{in}) is obtained from Figure B-6

$H_{in} = 92$ Btu/lb

STEP 3. Knowing $T_2 = -35^{\circ}\text{F}$, $P_2 = 500 \text{ psi}$, and the liquid fraction of the outlet stream $L = 0.406$, the enthalpy of the outlet stream (H_{out}) is determined from Figure B-7.

$$H_{\text{out}} = 12 \text{ Btu/lb}$$

STEP 4. Knowing $H_{\text{in}} = 92 \text{ Btu/lb}$ (Step 2) and $H_{\text{out}} = 12 \text{ Btu/lb}$ (step 3), the change in enthalpy or theoretical work is obtained from Figure B-8.

$$\text{Theoretical Work} = 80 \text{ Btu/lb}$$

STEP 5. Knowing theoretical work = 80 Btu/lb (Step 4), and an efficiency of 70%, the actual work is obtained from Figure B-9

$$\text{Actual Work} = 0.022 \text{ Horsepower-Hour/lb}$$

B-1 A Plot of Difference in Inlet (T_1) and Outlet (T_2) Temperatures vs. An Intermediate Variable (A) for Various Values of the Inlet Temperature (T_1)

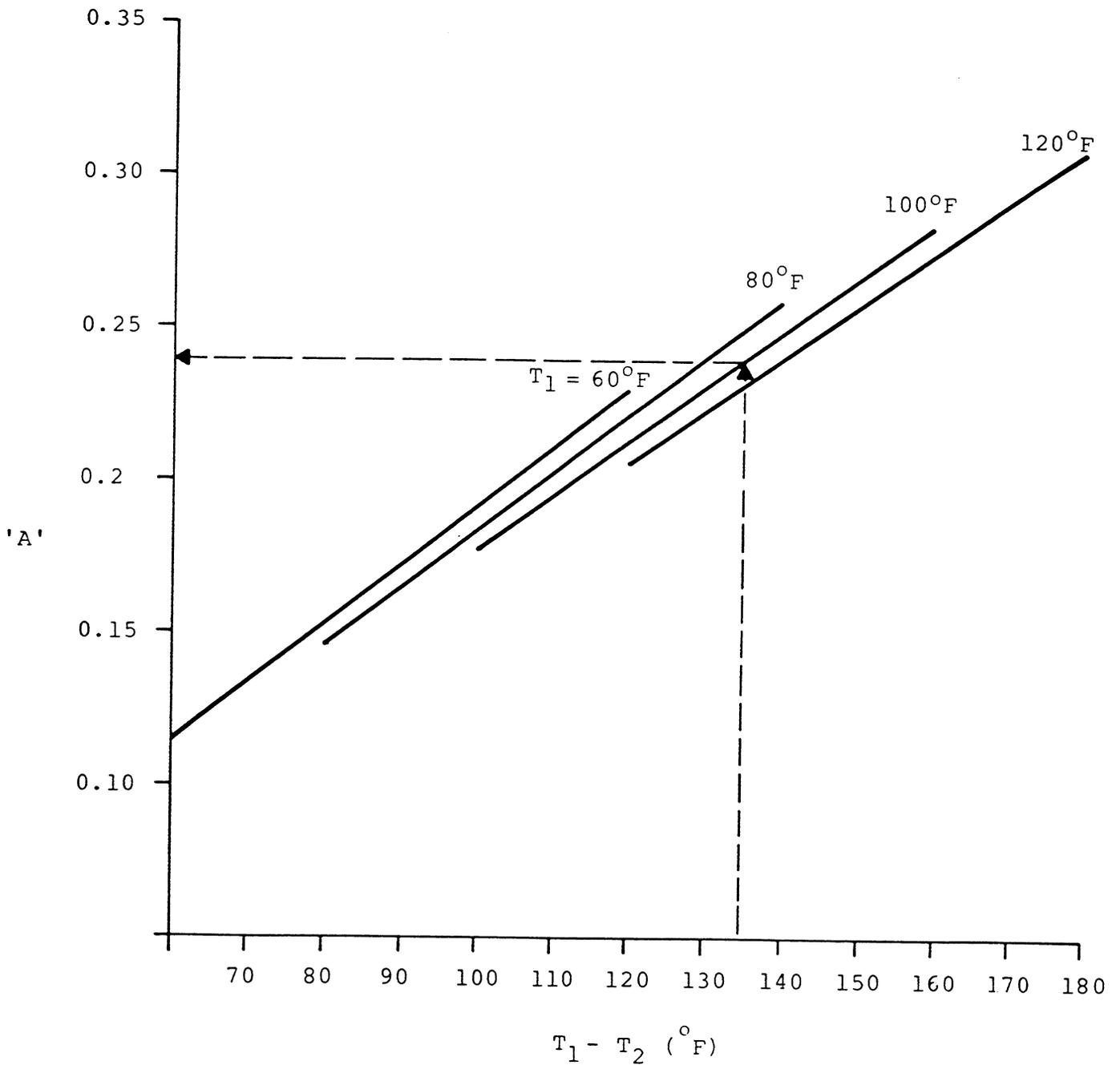


Figure B-2 A Plot of B (An Intermediate Variable) vs. A (Determined from Figure)
for Various Efficiencies of 60-90%

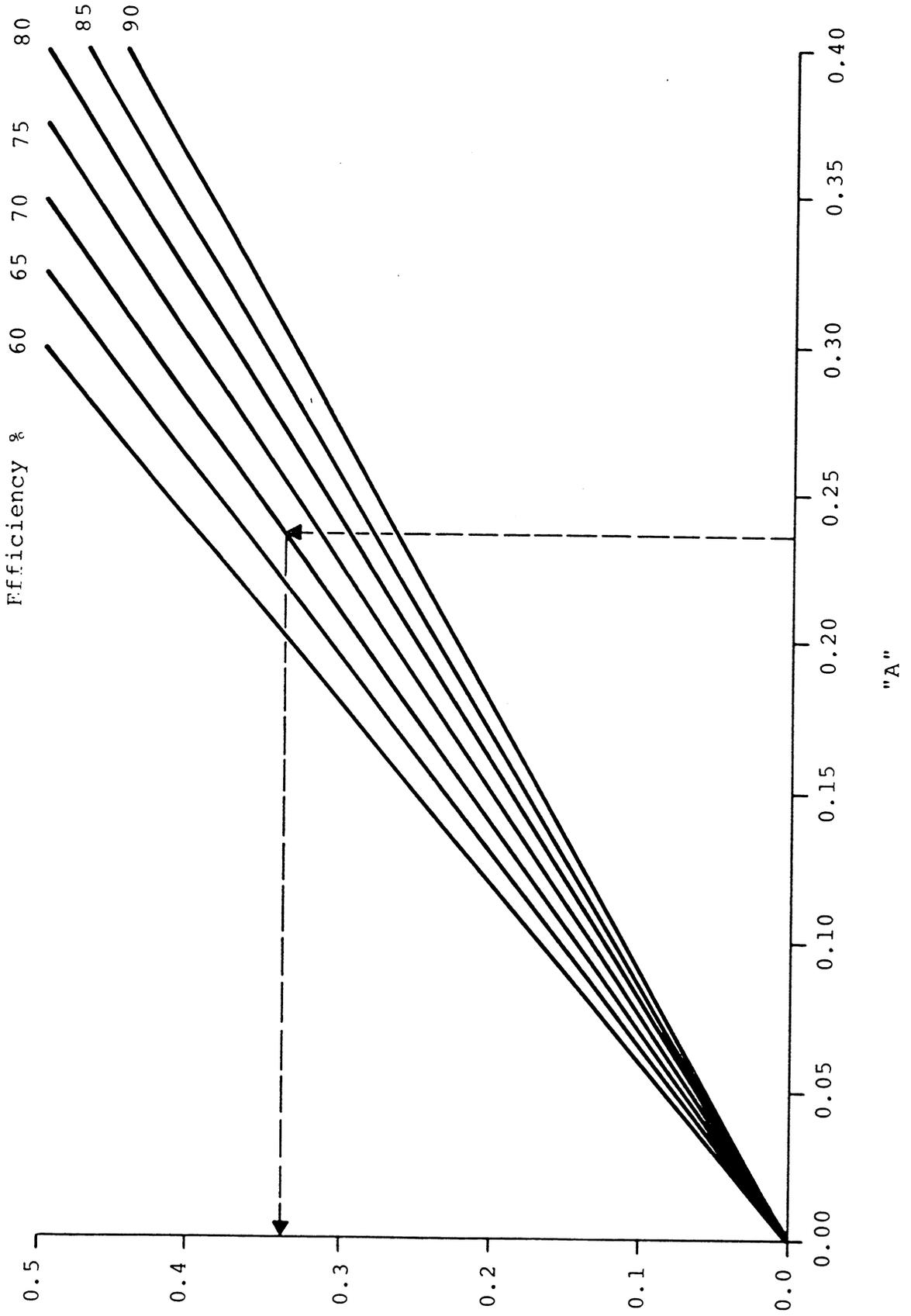


Figure B-3 A plot of C (An Intermediate Variable) vs. B (Determined from Figure) for Various $1/m$ Values of 1.0 - 8.0

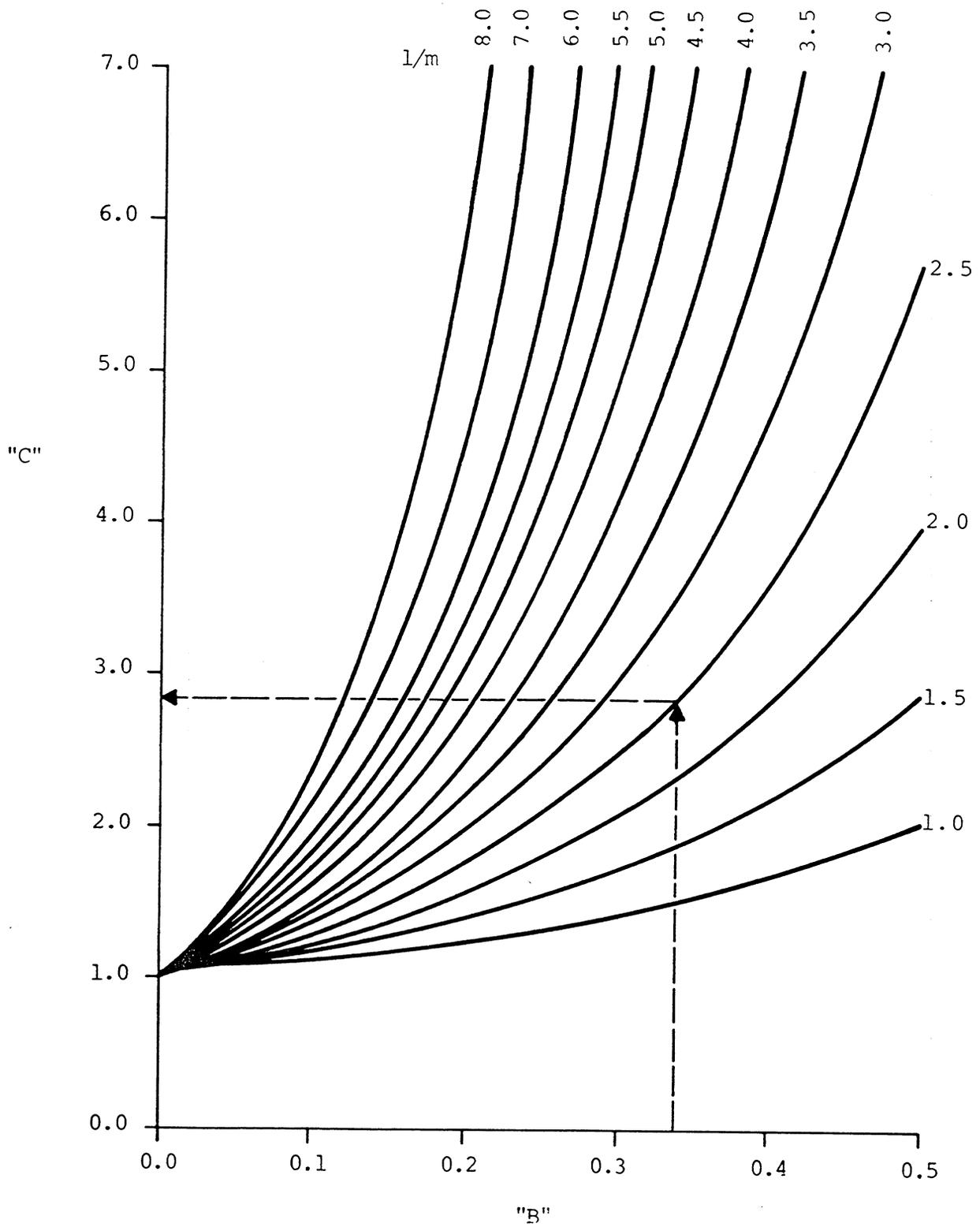


Figure B-4 A Plot of C (Determined from Figure) vs. Inlet Pressure (P_1) for Various Outlet Pressures (P_2) 500-1000 psia

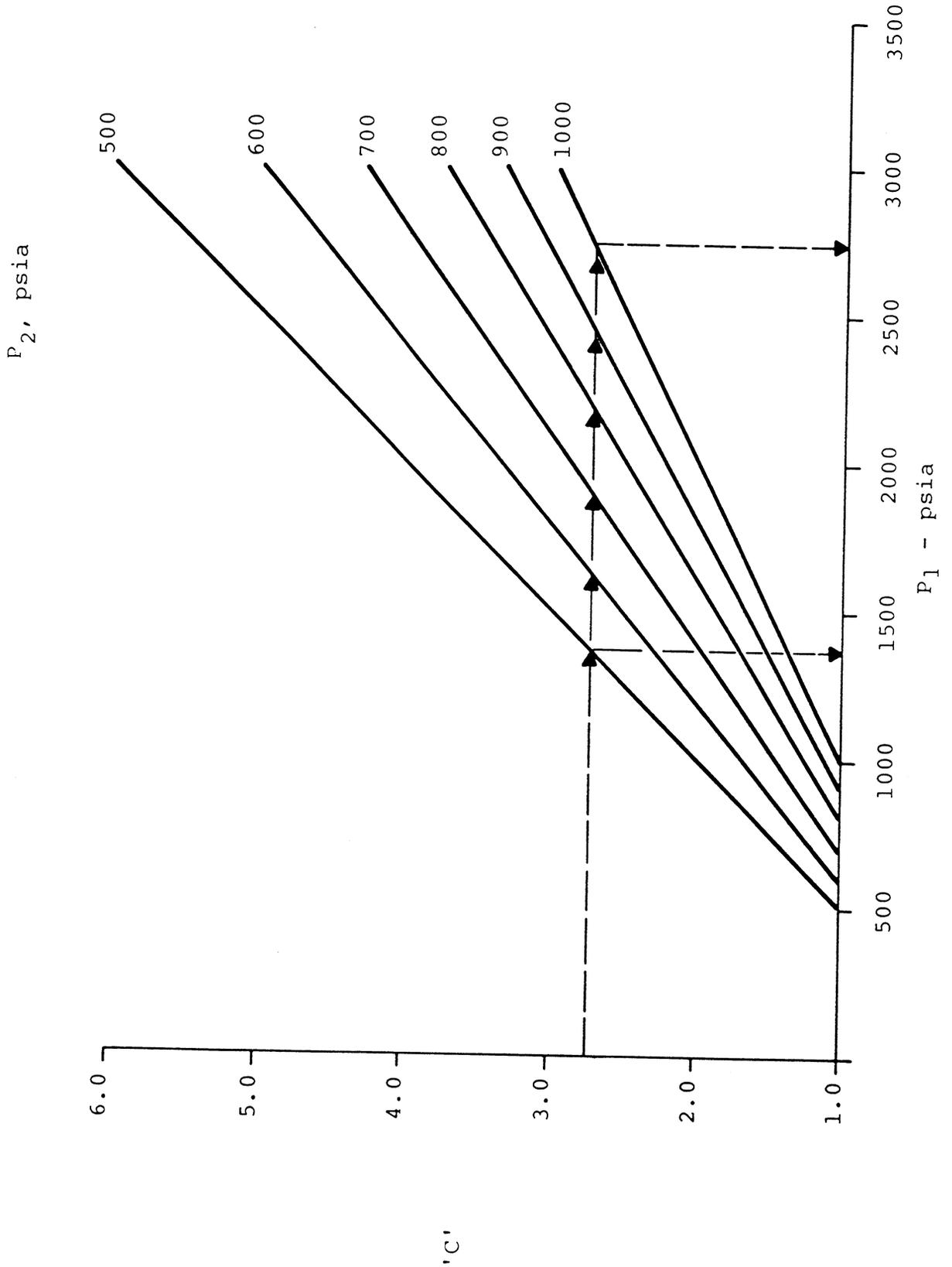


Figure B-5 Liquid Mole Fraction vs. Turboexpander Outlet Temperature (T_2) at Pressures of 500-1000 psia for a Mixture of 60% methane, 40% CO_2

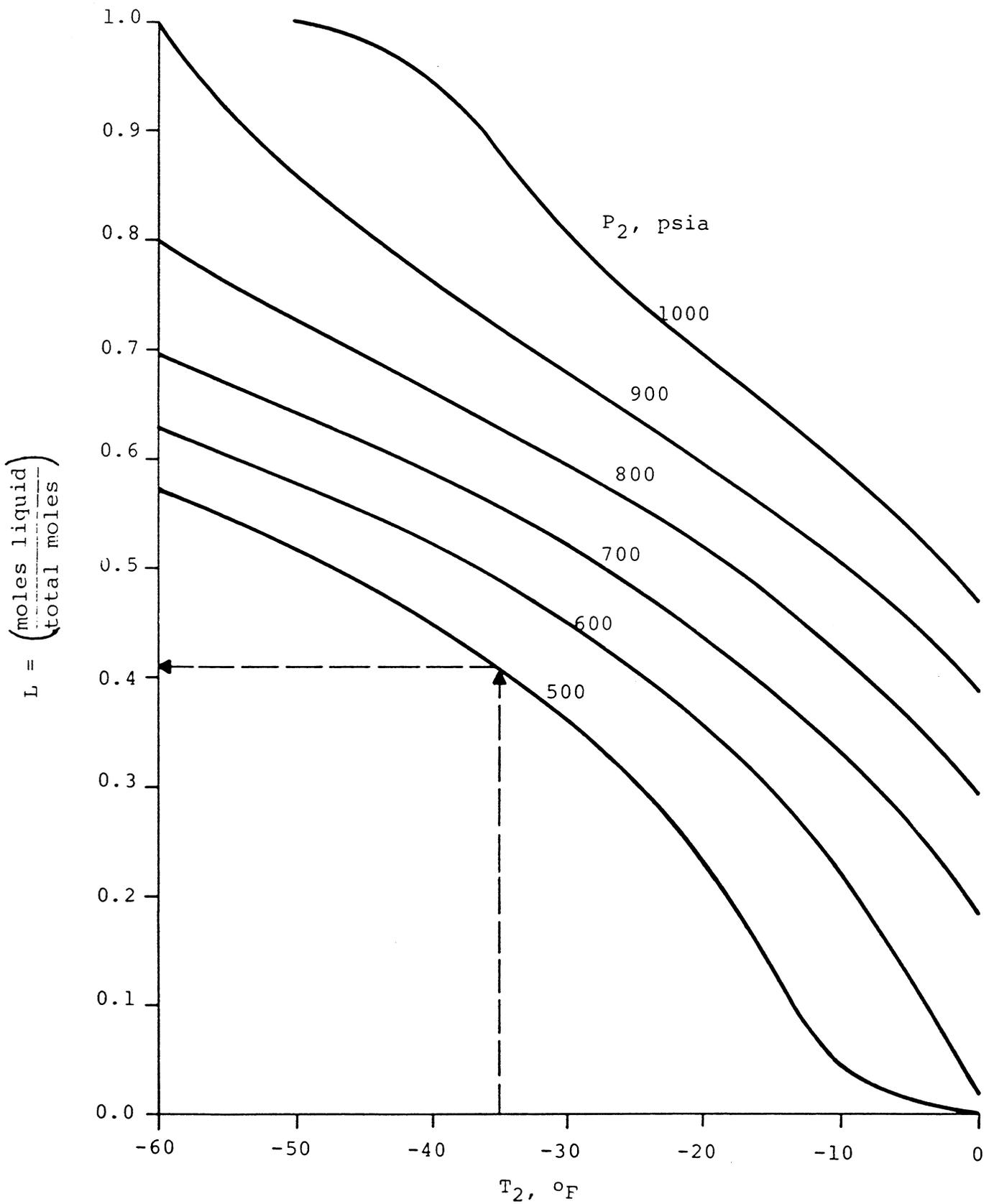


Figure B-6 Enthalpy of the Inlet Stream to the Turboexpander vs. Inlet Pressure at Inlet Temperatures of 60-120F

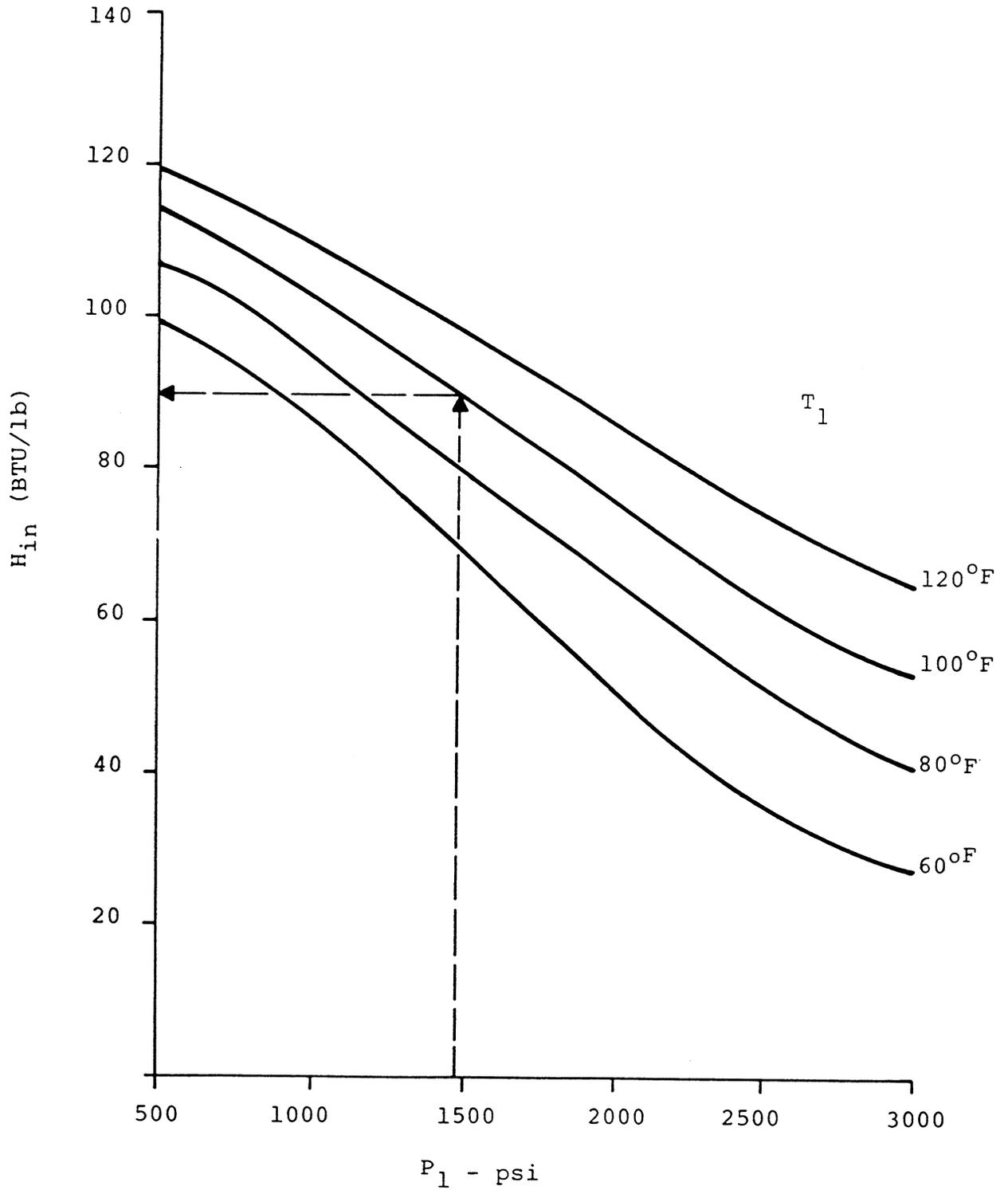
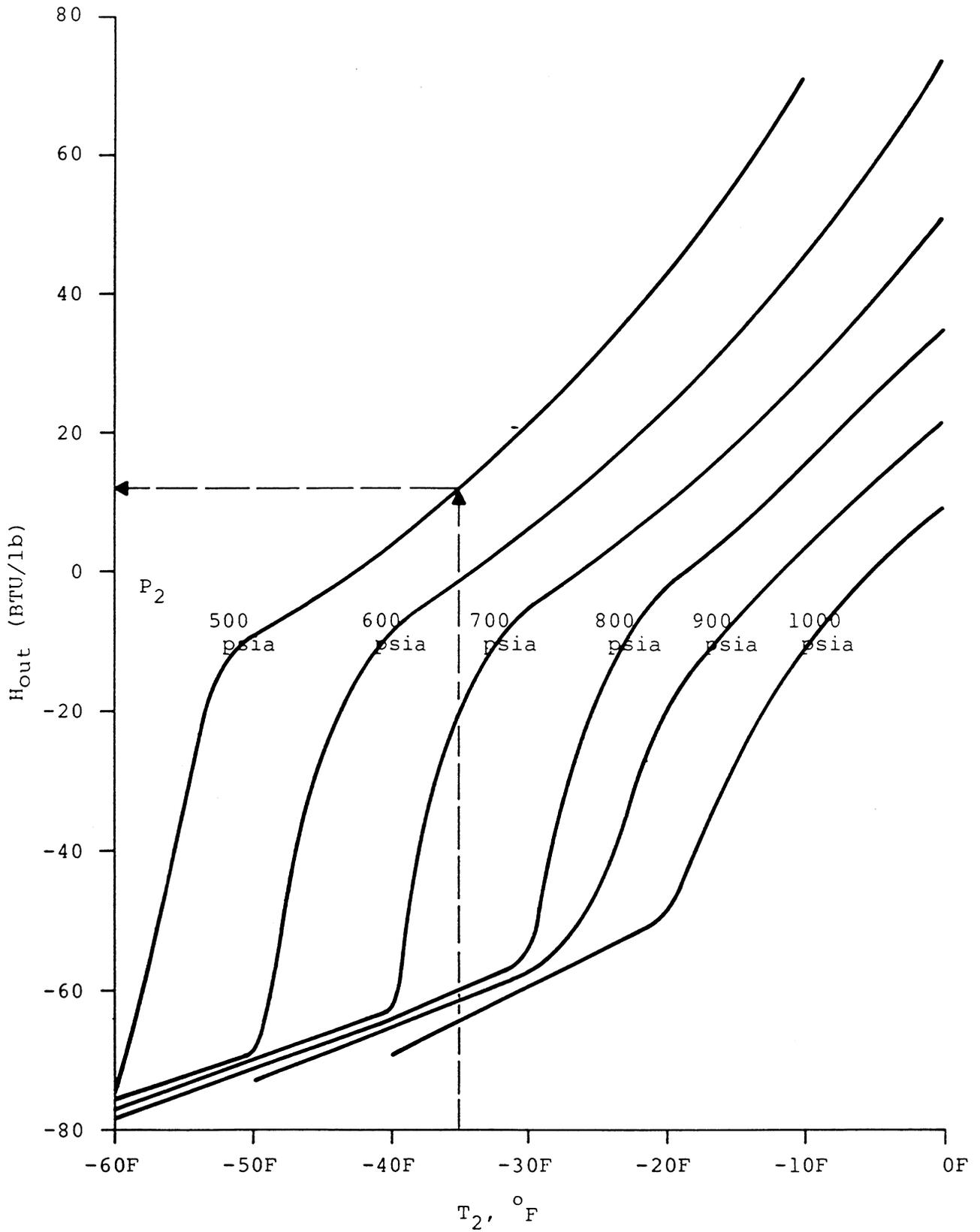


Figure B-7 Enthalpy of the Outlet Stream from the Turboexpander vs. Outlet Temperature at Outlet Pressures of 500-1000 psia



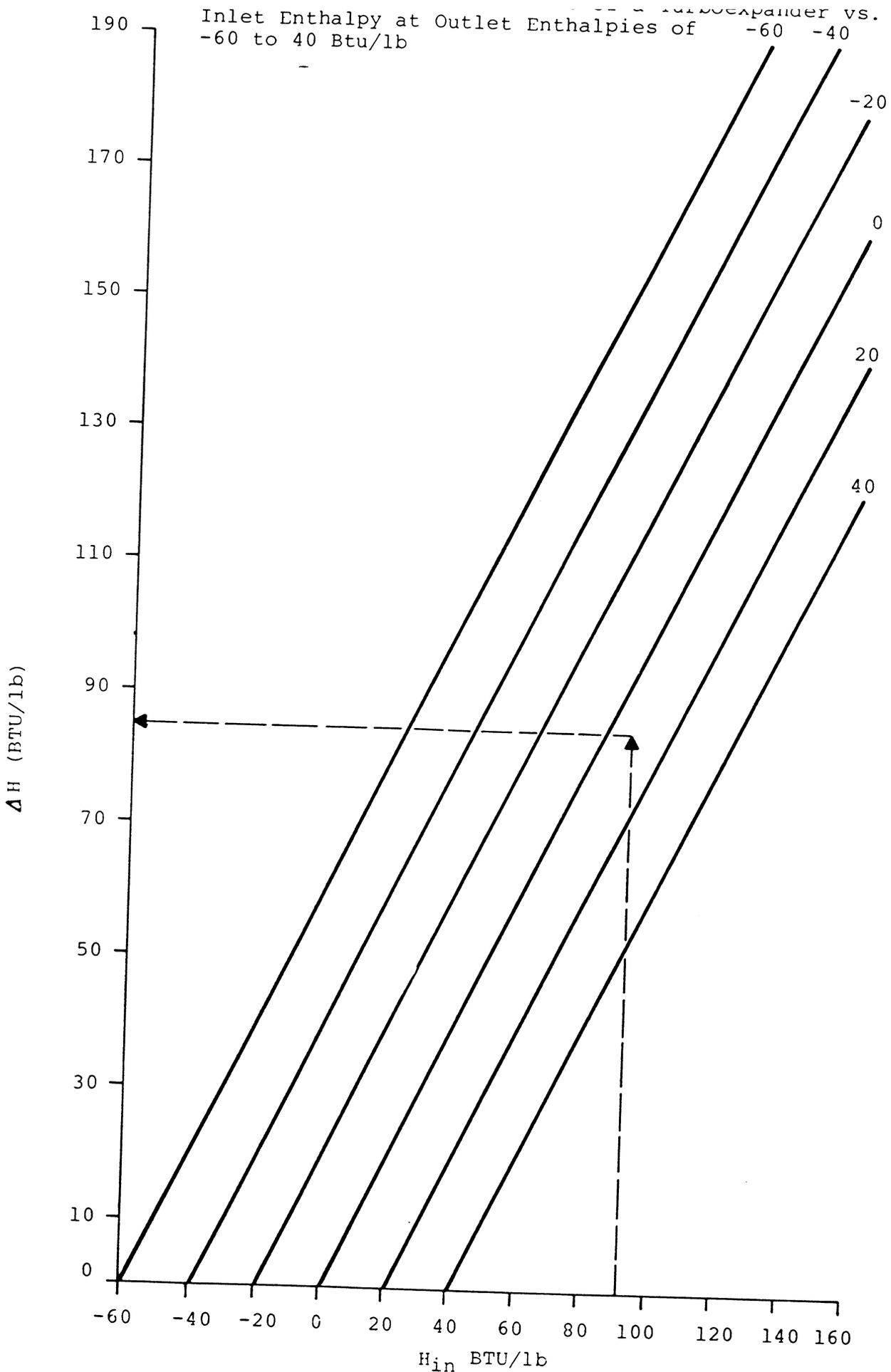


Figure B-9 Actual Work out of a Turboexpander vs. Theoretical Work for Efficiencies of 60-90%

